

## Supporting information

### Observation of delayed fluorescence/room-temperature phosphorescence emissions in organic small-molecule emitters, their properties, and electroluminescent performance

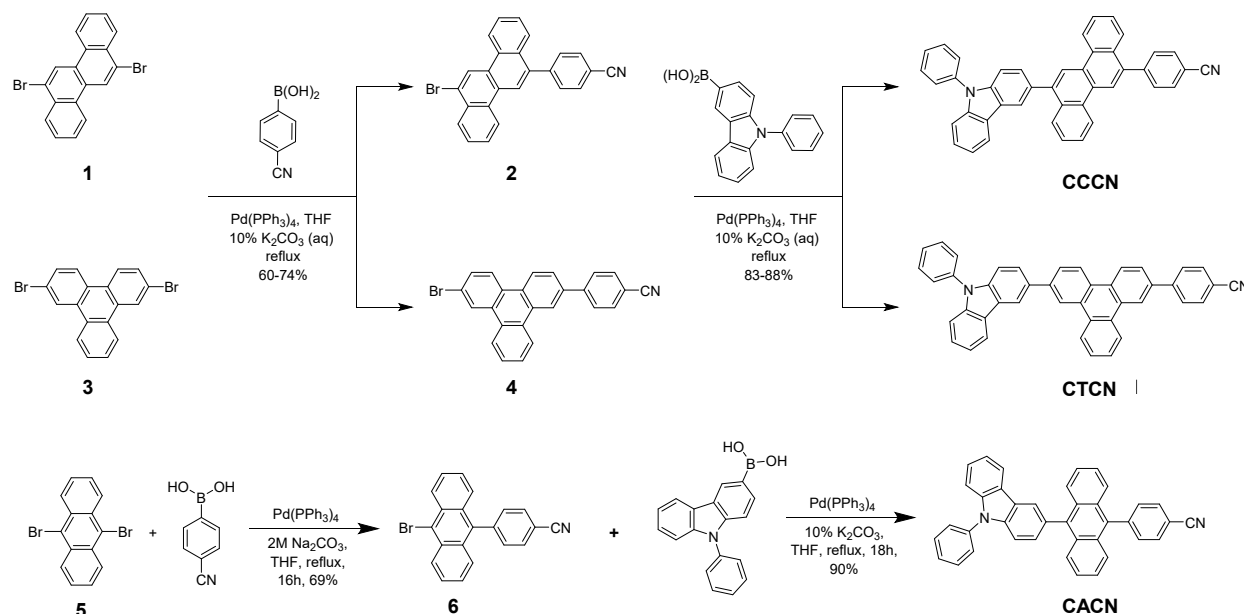
Nuttapong Chantanop,<sup>a</sup> Phattananawee Nalaoh,<sup>a</sup> Wijitra Waengdongbung,<sup>a</sup> Rattanasiri Wannapakdee,<sup>a</sup> Patteera Funchien,<sup>a</sup> Suwapat Kongsabay,<sup>a</sup> Taweesak Sudyoosuk<sup>b</sup> and Vinich Promarak<sup>a\*</sup>

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#### 1. Materials synthesis



**Scheme S1** The synthesis of the designed molecules.

#### 4-(12-Bromochrysen-6-yl)benzonitrile (**2**)

A mixture of 6,12-dibromochrysene (700 mg, 1.80 mmol), 4-cyanophenylboronic acid (88 mg, 0.6 mmol), 10%  $\text{K}_2\text{CO}_3$  (aq) (10 mL), and  $\text{Pd(PPh}_3)_4$  (35 mg, 5 mol%) in dried THF (60 mL) was degassed with  $\text{N}_2$  for 10 min. The reaction mixture was stirred at reflux under  $\text{N}_2$  for 18 h. After being cooled to room temperature, the mixture was diluted with water (50 mL) and extracted with DCM (3 x 50 mL). The combined organic layer was washed with water (50 mL), brine (2 x 50 mL), dried over anhydrous  $\text{Na}_2\text{SO}_4$ , filtered, and concentrated in vacuo to dryness. The crude was purified by column chromatography on silica gel eluting with a gradient of dichloromethane/hexane mixture (0% to 50%) to obtain light yellow solids (147 mg, 60%). M.p. = 241–245 °C;  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  9.09 (1H, s, Ar-H), 8.80 (1H, d,  $J$  = 8.4 Hz, Ar-H), 8.78 – 8.73 (1H, m, Ar-H), 8.56 (1H, s, Ar-H), 8.50 – 8.45 (1H, m, Ar-H), 7.92 – 7.84 (m, 3H), 7.81 – 7.72 (m, 5H), 7.62 (1H, ddd,  $J$  = 8.1, 6.8, 1.2 Hz, Ar-H);  $^{13}\text{C}$  NMR (151 MHz,  $\text{CDCl}_3$ )  $\delta$  145.9, 137.8, 132.3, 131.4, 130.9, 130.8, 130.4, 130.0, 128.7, 128.3, 127.8, 127.8,

127.3, 127.2, 127.1, 126.2, 125.3, 123.6, 123.5, 123.3, 122.3, 118.8, 111.6; HRMS APCI/Q-TOF ( $m/z$ ): calcd 409.0289 for  $C_{25}H_{14}BrN$ ; found 409.9745 [ $M^+$ ].

#### 4-(7-Bromotriphenylen-2-yl)benzonitrile (4)

It was synthesized similarly to compound **2** using 2,7-dibromotriphenylene and obtained as white solids (270 mg, 74%). M.p. = 287–290 °C;  $^1H$  NMR (600 MHz,  $CDCl_3$ )  $\delta$  8.83 (1H, s, Ar-H), 8.79 (1H, d,  $J$  = 1.9 Hz, Ar-H), 8.73 – 8.70 (1H, m, Ar-H), 8.68 (1H, d,  $J$  = 8.5 Hz, Ar-H), 8.62 – 8.57 (1H, m, Ar-H), 8.52 (1H, d,  $J$  = 8.7 Hz, Ar-H), 7.90 – 7.87 (3H, m, Ar-H), 7.82 (1H, d,  $J$  = 8.0 Hz, Ar-H), 7.78 (1H, dd,  $J$  = 8.6, 1.9 Hz, Ar-H), 7.76 – 7.69 (2H, m, Ar-H);  $^{13}C$  NMR (151 MHz,  $CDCl_3$ )  $\delta$  145.4, 138.1, 132.8, 131.8, 130.5, 129.8, 129.3, 129.0, 128.1, 128.0, 127.9, 126.4, 126.2, 125.2, 124.2, 123.6, 123.3, 122.2, 118.9, 111.3; HRMS APCI/Q-TOF ( $m/z$ ): calcd 409.0289 for  $C_{25}H_{14}BrN$ ; found 409.9523 [ $M^+$ ].

#### 4-(10-Bromoanthracen-9-yl)benzonitrile (6)

It was synthesized similarly to compound **2** using 9,10-dibromoanthracene and obtained as yellow solids (338 mg, 69%).  $^1H$  NMR (600 MHz,  $CDCl_3$ )  $\delta$  8.63 (d,  $J$  = 8.9 Hz, 2H), 7.92 – 7.88 (m, 2H), 7.63 – 7.60 (m, 2H), 7.57 – 7.52 (m, 2H), 7.49 (d,  $J$  = 8.8 Hz, 2H), 7.43 – 7.40 (m, 2H);  $^{13}C$  NMR (151 MHz,  $CDCl_3$ )  $\delta$  143.8, 135.1, 132.3, 132.1, 130.5, 130.2, 128.2, 127.1, 126.4, 126.3, 124.0, 118.7, 112.0. HRMS APCI/Q-TOF ( $m/z$ ): calcd for  $C_{25}H_{14}BrN$  357.0153; found: 358.0198 [ $M + H$ ] $^+$ .

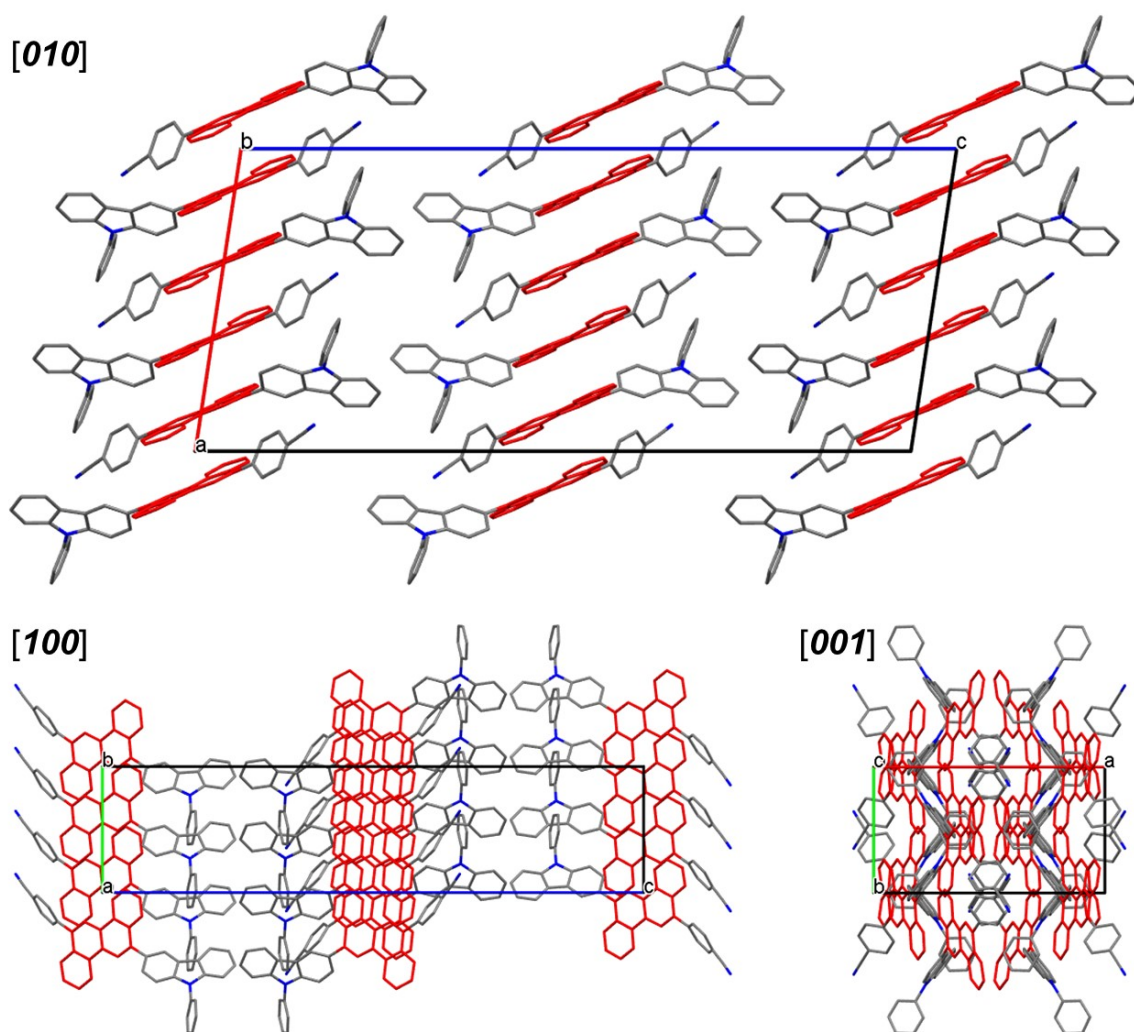
## 2. Single crystal XRD data

All crystallographic information (CIF) data, including CCDC numbers 1984569 and 2012094 for CCCN and CTCN, were deposited and obtained with no cost at <https://www.ccdc.cam.ac.uk/>.

**Table S1** Crystal structures information.

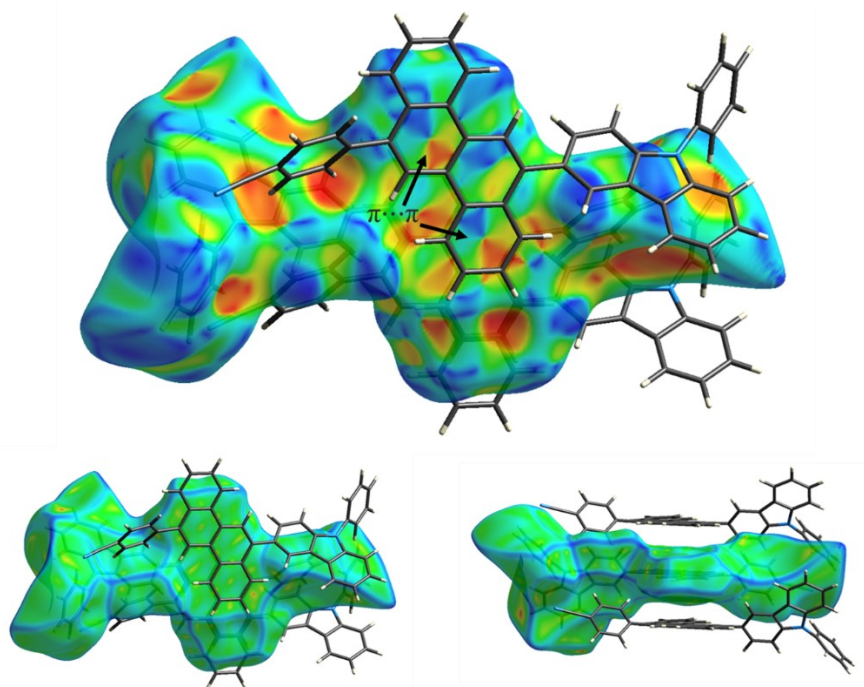
Compound	CCCN	CTCN
CCDC Deposit Number	1984569	2012094
Empirical formula	$C_{43}H_{26}N_2$	$C_{43}H_{26}N_2$
Formula weight	570.66	570.66
Temperature/K	110	195
Crystal system	monoclinic	monoclinic
Space group	$C2/c$	$P21/n$
a/Å	17.234(2)	14.7177(4)
b/Å	9.2814(11)	8.9166(2)
c/Å	40.346(5)	22.1216(6)
$\alpha/^\circ$	90	90
$\beta/^\circ$	98.618(4)	99.1280(10)
$\gamma/^\circ$	90	90
Volume/Å <sup>3</sup>	6380.7(13)	2866.29(13)
Z	8	4
$\rho_{calc}/cm^3$	1.188	1.322
$\mu/mm^{-1}$	0.069	0.590
F(000)	2384.0	1192.0
Crystal size/mm <sup>3</sup>	0.38 × 0.30 × 0.08	0.58 × 0.12 × 0.05

Radiation	MoK $\alpha$ ( $\lambda$ = 0.71073)	CuK $\alpha$ ( $\lambda$ = 1.54178)
2 $\Theta$ range for data collection/ $^{\circ}$	4.782 to 52.742	6.75 to 140.138
Index ranges	-21 $\leq$ h $\leq$ 21, -11 $\leq$ k $\leq$ 11, -50 $\leq$ l $\leq$ 50	-17 $\leq$ h $\leq$ 14, -10 $\leq$ k $\leq$ 10, -26 $\leq$ l $\leq$ 26
Reflections collected	70682	43547
Independent reflections	6531 [ $R_{\text{int}}$ = 0.0711, $R_{\text{sigma}}$ = 0.0339]	5421 [ $R_{\text{int}}$ = 0.0570, $R_{\text{sigma}}$ = 0.0284]
Data/restraints/parameters	6531/0/406	5421/0/407
Goodness-of-fit on $F^2$	1.100	1.024
Final R indexes [ $I \geq 2\sigma(I)$ ]	$R_1$ = 0.0563, $wR_2$ = 0.1296	$R_1$ = 0.0424, $wR_2$ = 0.1068
Final R indexes [all data]	$R_1$ = 0.0716, $wR_2$ = 0.1363	$R_1$ = 0.0542, $wR_2$ = 0.1141

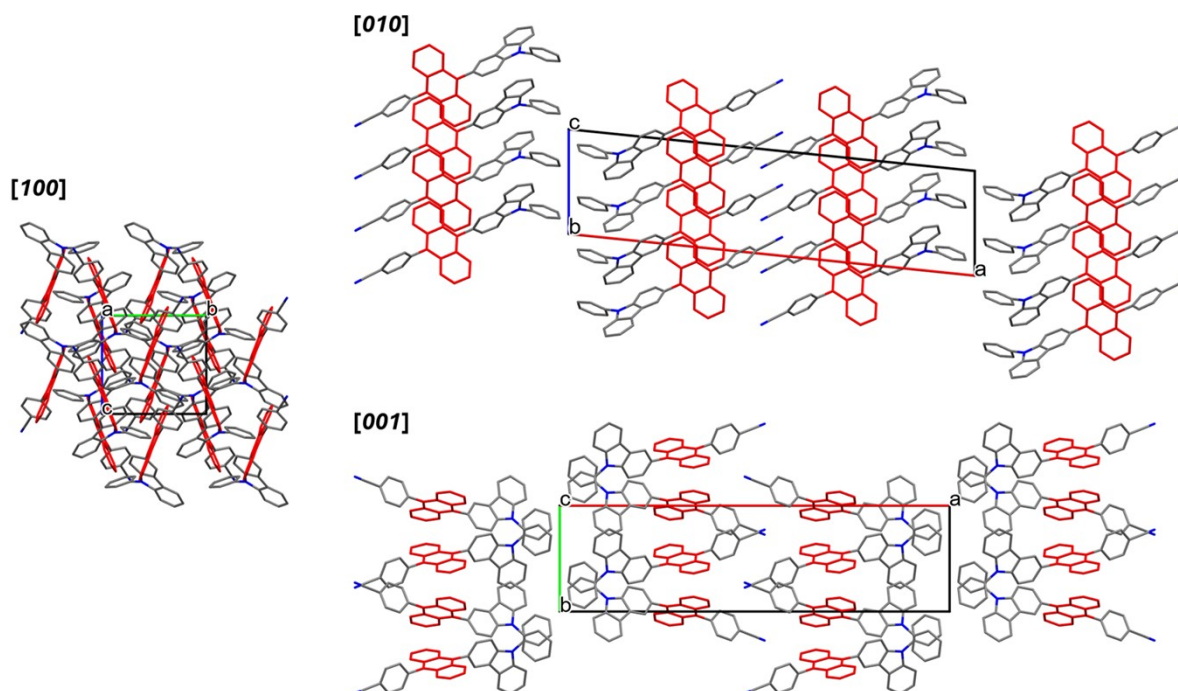


**Fi**

**g. S1.** Crystal packing of CCCN along [100], [010], and [001] directions with red-highlighted chrysene structure. All hydrogen atoms were omitted for clarity.

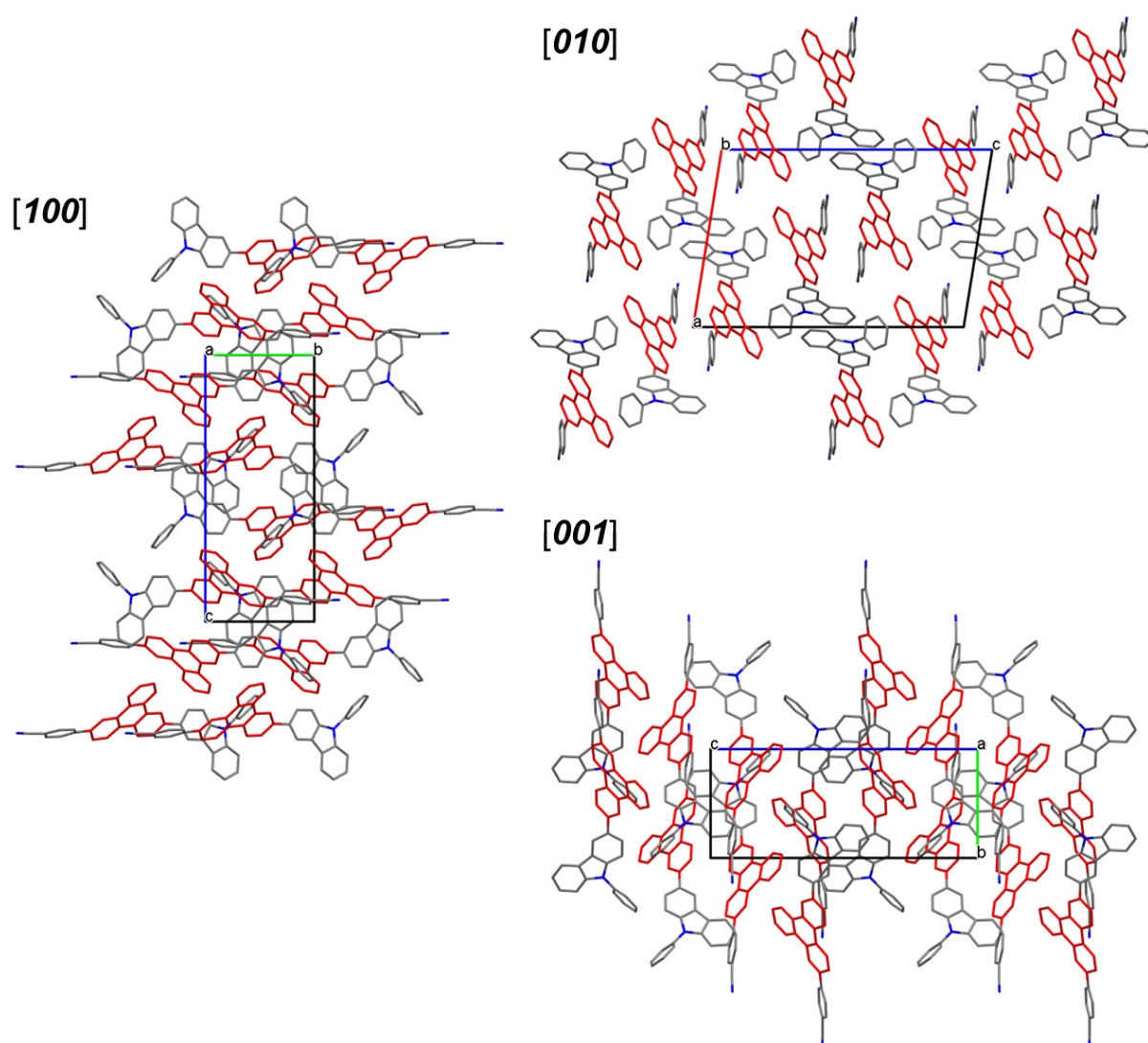


**Fig. S2** Hirshfeld Surface Analysis of **3CzCsCN** displayed shape index (top) and curvedness (bottom-left: the top view, bottom-right: the side view). The blue and red triangles on the shape index surface indicated  $\pi$ - $\pi$  interaction in the 6-membered-ring subunits of chrysene (highlighted with the arrow line). A flat green surface of curvedness indicated the intermolecular  $\pi$ - $\pi$  interaction, while the blue-line edge crossed over the subunit of the chrysene moiety that presented no interplane  $\pi$ -stacking.



**Fig. S3** Crystal packing of **CzAnthBN** from reference<sup>1</sup> along  $[100]$ ,  $[010]$ , and  $[001]$  directions with red-highlighted anthracene structure. All hydrogen atoms were omitted for clarity.

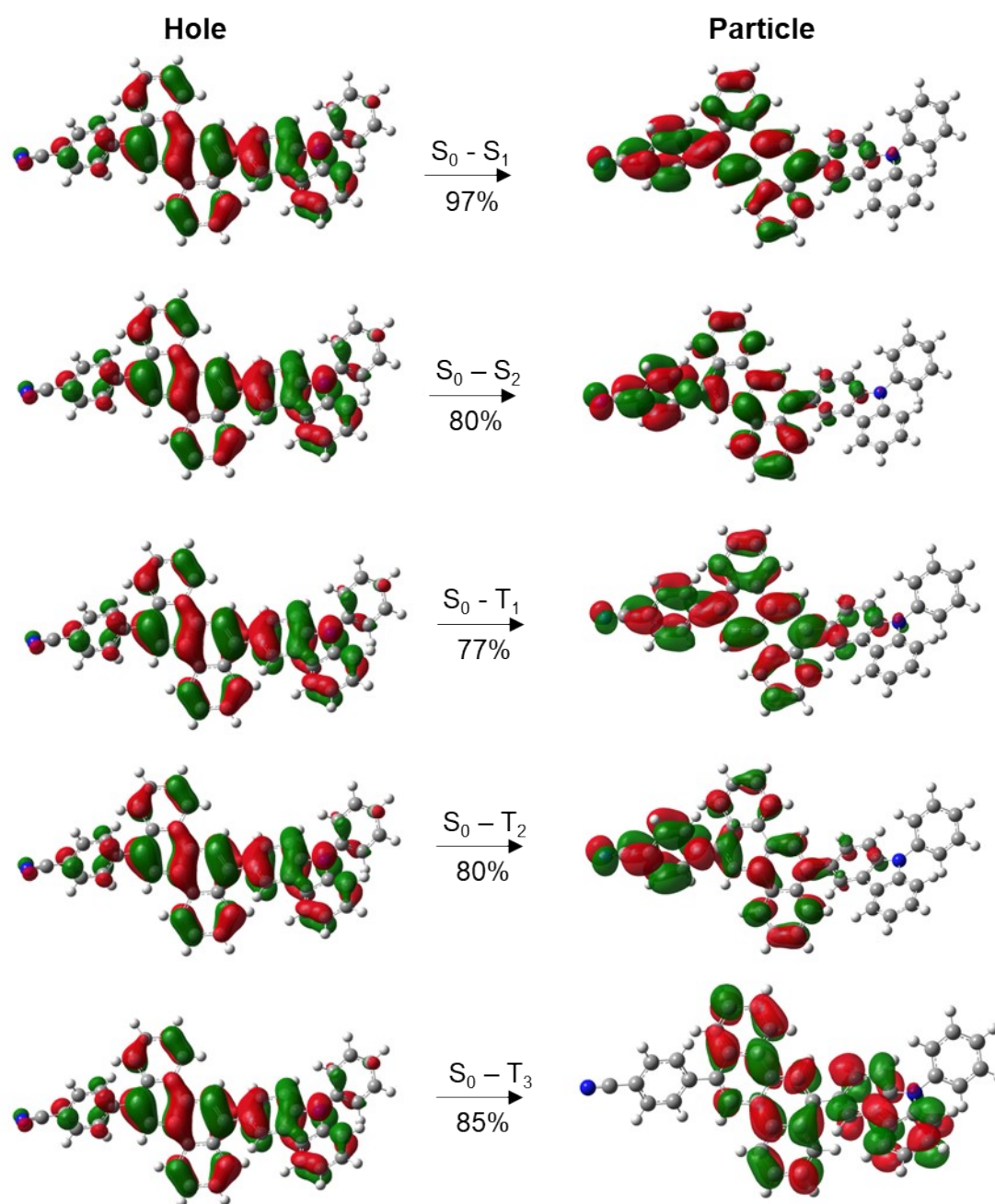
<sup>1</sup>Journal of Materials Chemistry C, 2019, 7, 1014, DOI: 10.1039/C8TC05707A



**Fig. S4** Crystal packing of CTCN along  $[100]$ ,  $[010]$ , and  $[001]$  directions with red-highlighted triphenylene structure. All hydrogen atoms were omitted for clarity.

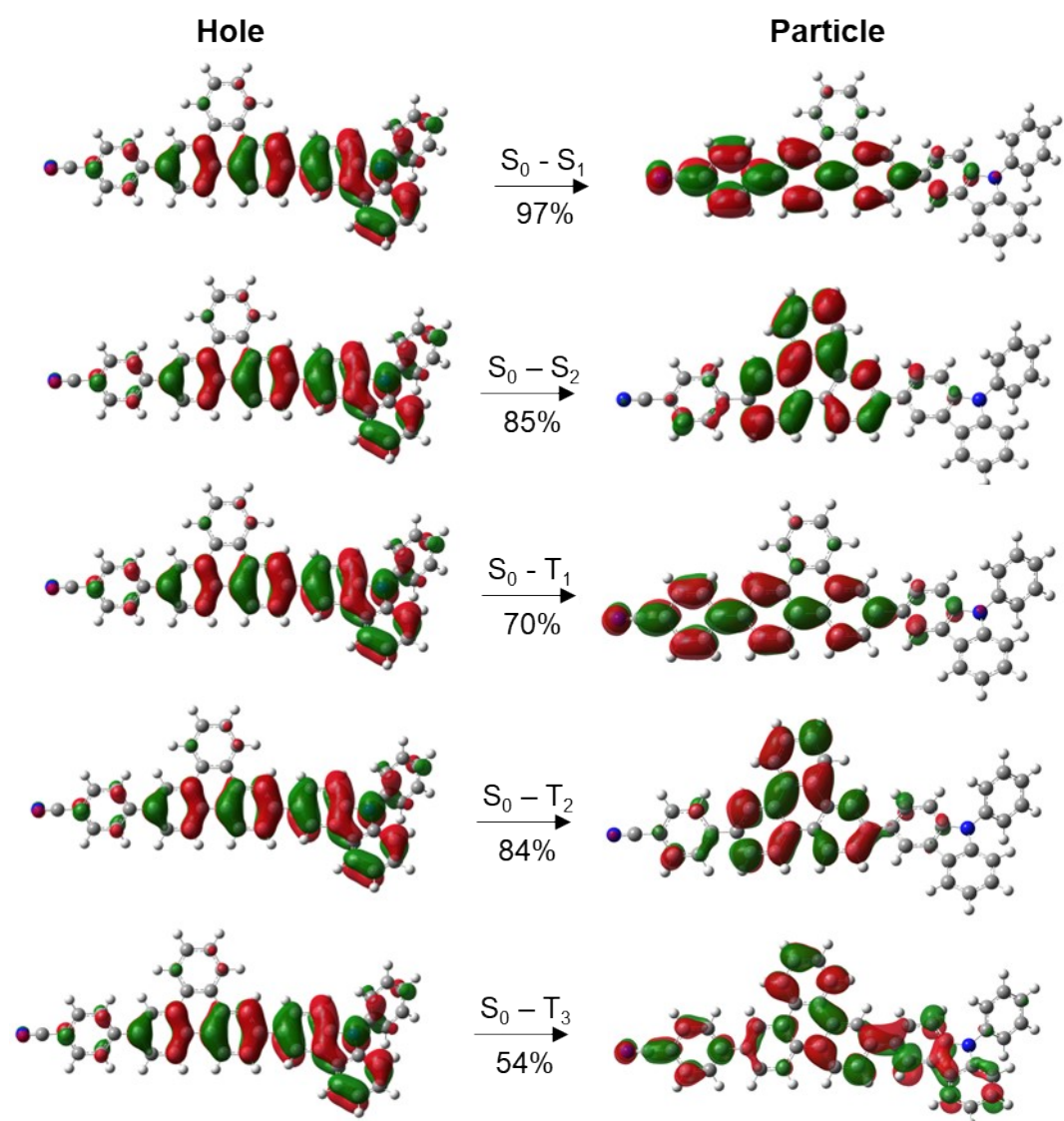


### 3. DFT calculation results



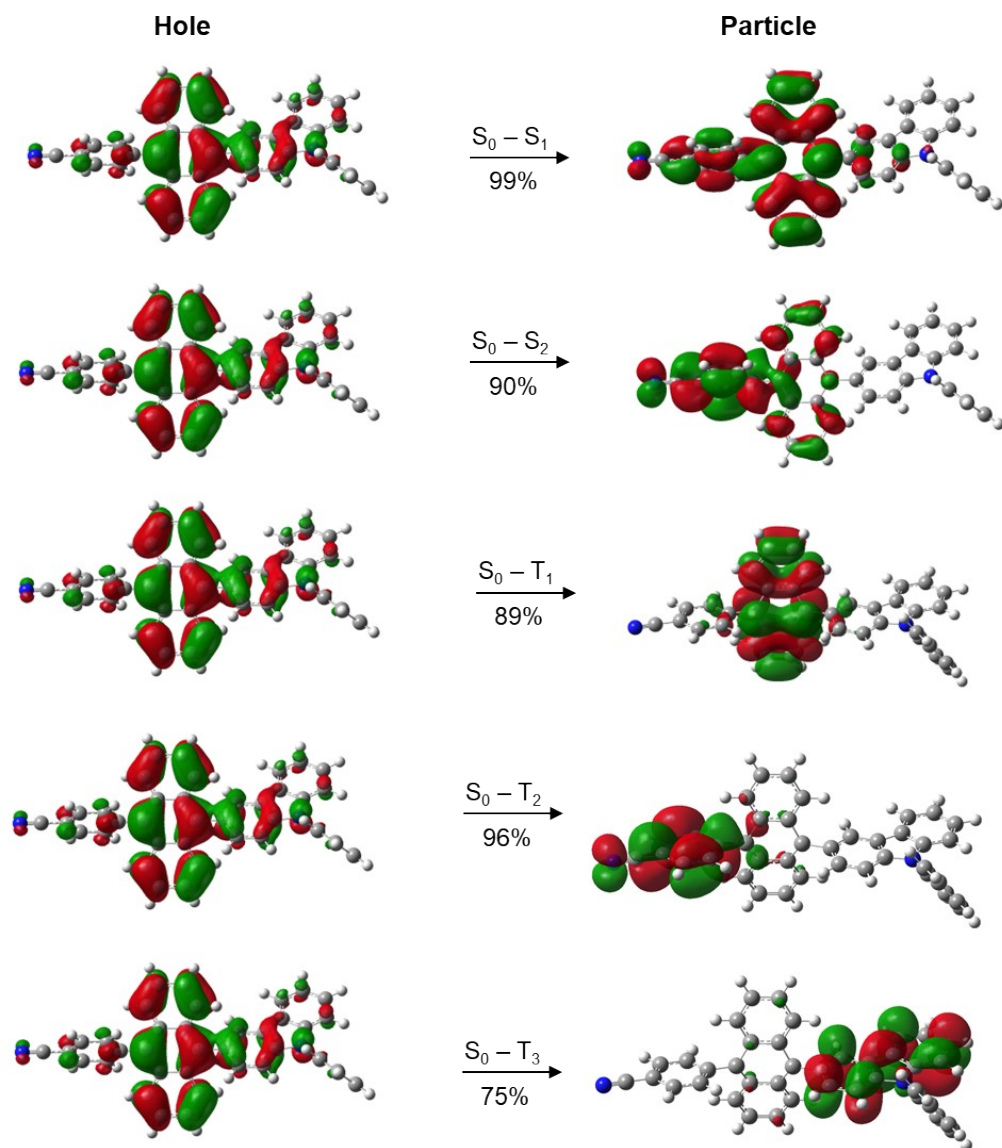
State	Energy (eV)	Osc. Strength	Transition character	Sm index
S1	2.69	0.5674	LE+CT	0.44675
S2	3.31	0.0248	LE+CT	0.31347
T1	1.48	-	LE+CT	0.61379
T2	2.72	-	LE+CT	0.46677
T3	2.88	-	LE+CT	0.36178

**Fig. S5** NTOs and the excited states energies of CCCN calculated by the TD-B3LYP/6-31G(d,p) function.



State	Energy (eV)	Osc. Strength	Transition character	Sm index
S1	2.85	0.7212	LE+CT	0.25433
S2	3.41	0.0586	LE+CT	0.53270
T1	1.92	-	LE+CT	0.46628
T2	2.69	-	LE+CT	0.40238
T3	3.03	-	LE+CT	0.43569

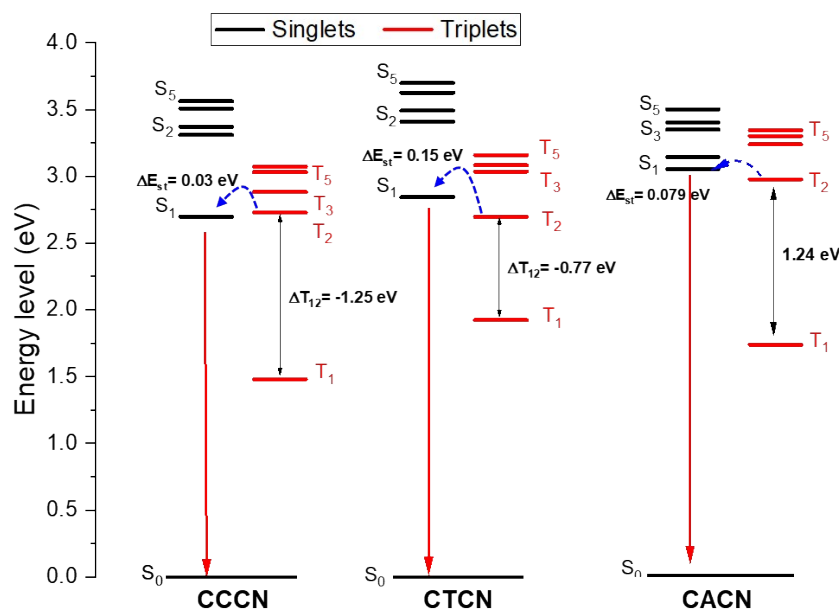
**Fig. S6** NTOs and the excited states energies of CTCN calculated by the TD-B3LYP/6-31G(d,p) function.



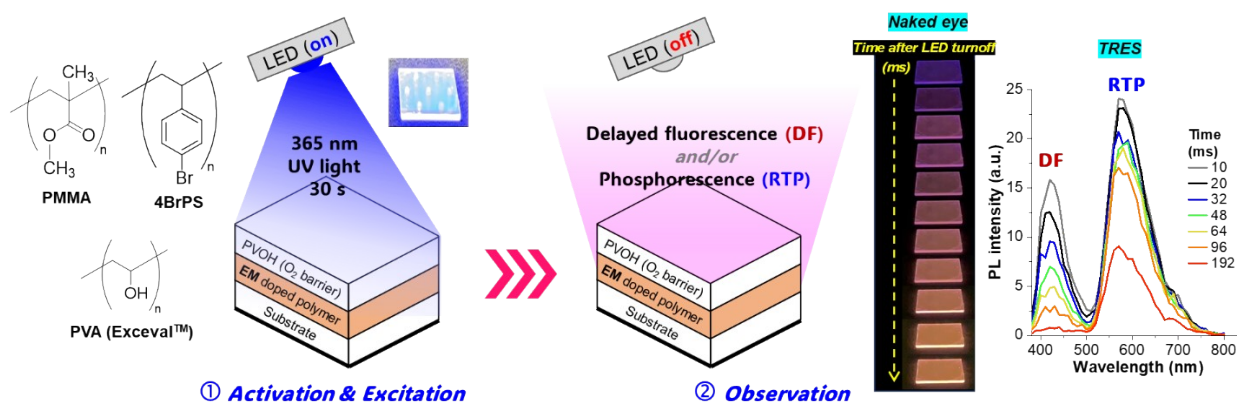
State	Energy (eV)	Osc. Strength	Transition character	Sm index
S1	3.05	0.4561	LE+CT	0.60781
S2	3.14	0.0019	LE+CT	0.21827
T1	1.73	-	LE+CT	0.71428
T2	2.97	-	LE+CT	0.27335
T3	3.23	-	LE+CT	0.42218

**Fig. S7** NTOs and the excited states energies of CACN calculated by the TD-B3LYP/6-31G(d,p) function.

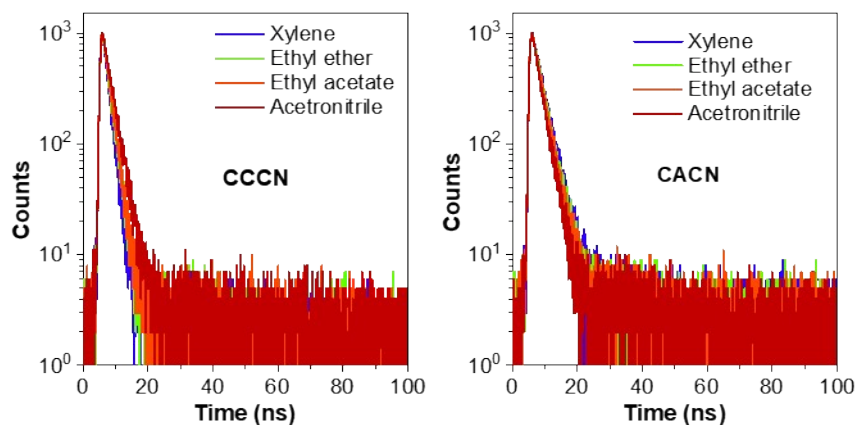




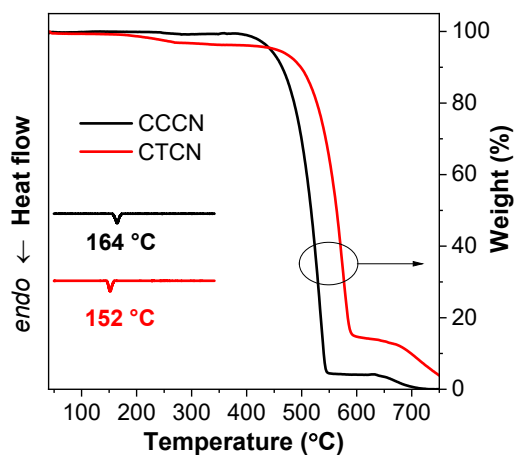
**Fig. S8** The energy diagrams of single excited states and triplet excited states calculated by TD-DFT B3LYP/6-31G(d,p).



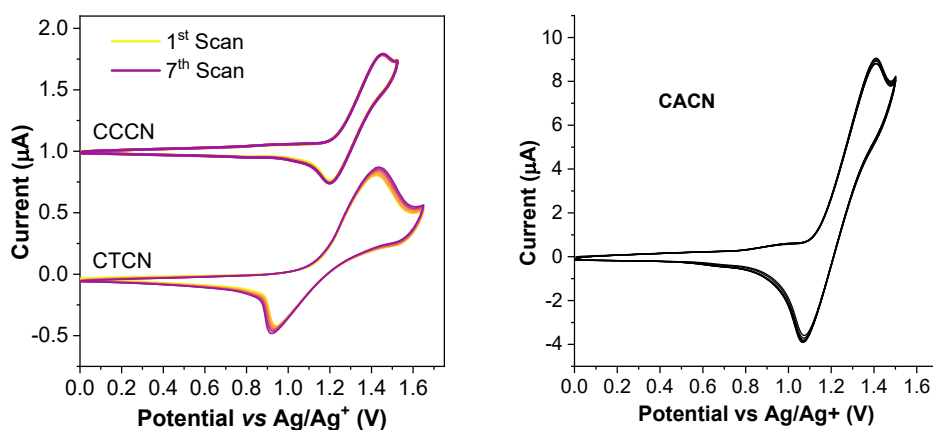
**Fig. S9** Experimental scheme for the observation of delayed fluorescence (DF)/room-temperature phosphorescence (RTP) emissions in solid films at ambient conditions.



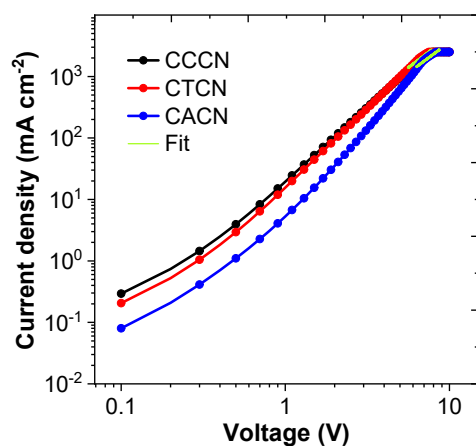
**Fig. S10** Transient PL decay spectra of CCCN and CACN in various solvents.



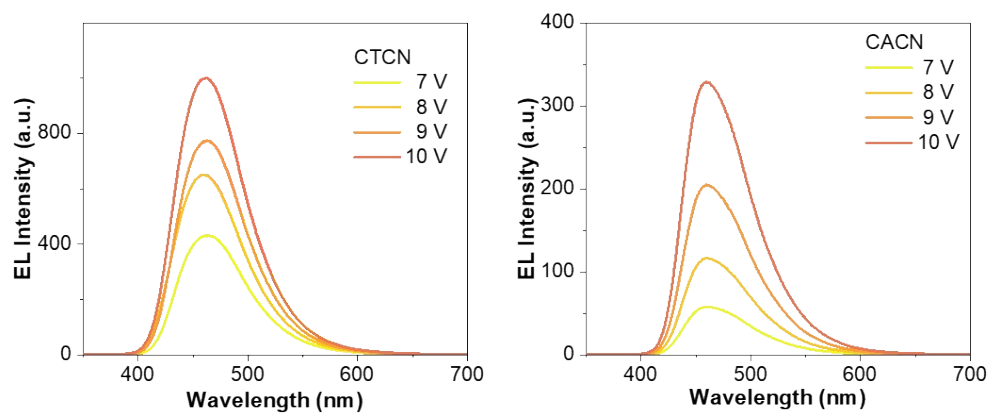
**Fig. S11** DSC and TGA thermograms of CCCN and CTCN measured at a heating rate of 10 °C/min under N<sub>2</sub> flow.



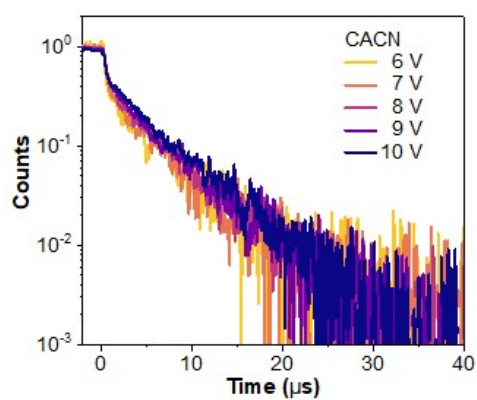
**Fig. S12** CV voltammograms analyzed in dry DCM containing *n*-Bu<sub>4</sub>NPF<sub>6</sub> as a supporting electrolyte at a scan rate of 50 mV s<sup>-1</sup> under Ar atmosphere.



**Fig. S13** *I-V* plots of the hole-only device (HOD).



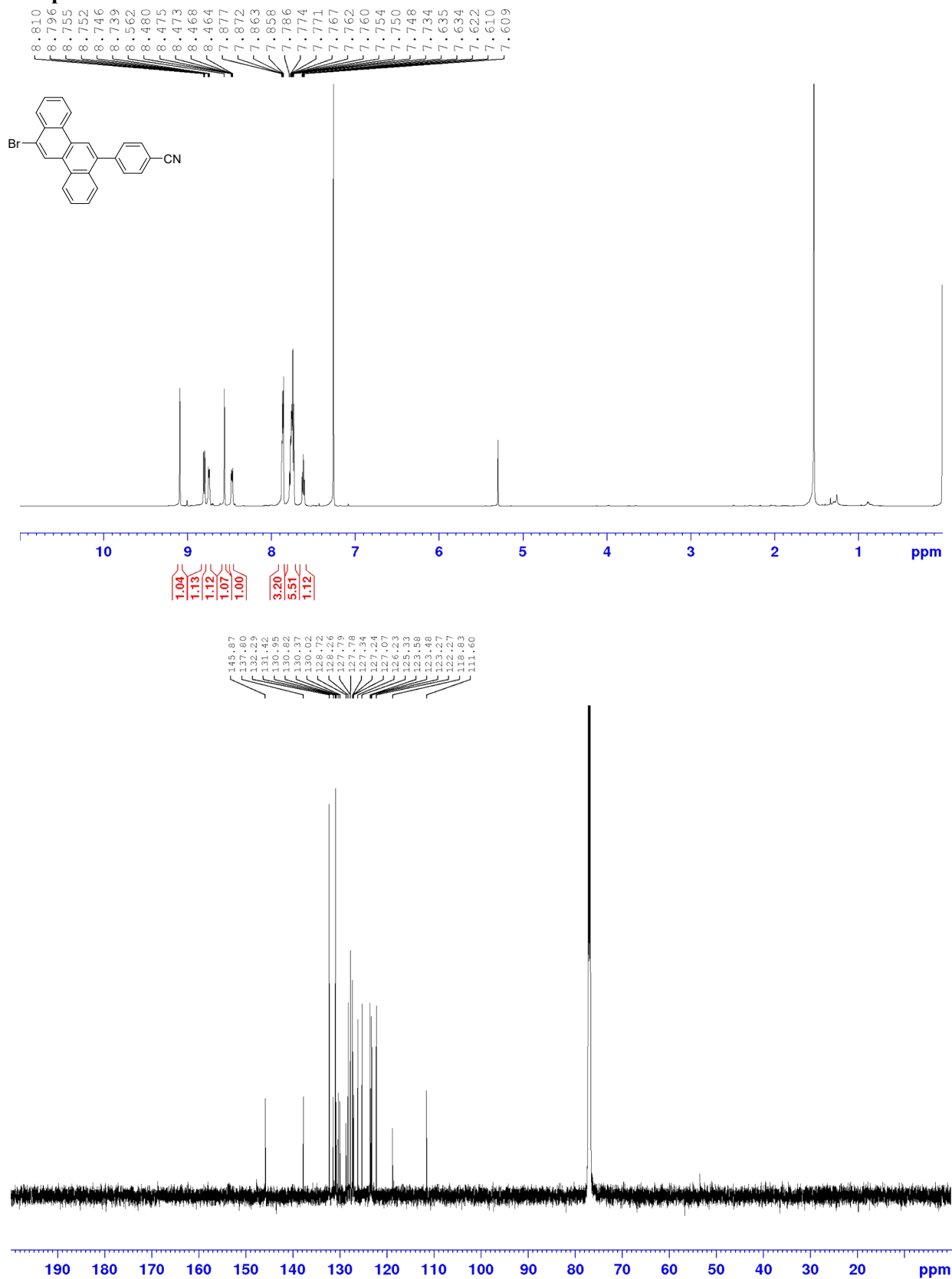
**Fig. S14** EL spectra at different voltages of CTCN- and CACN-based OLEDs.

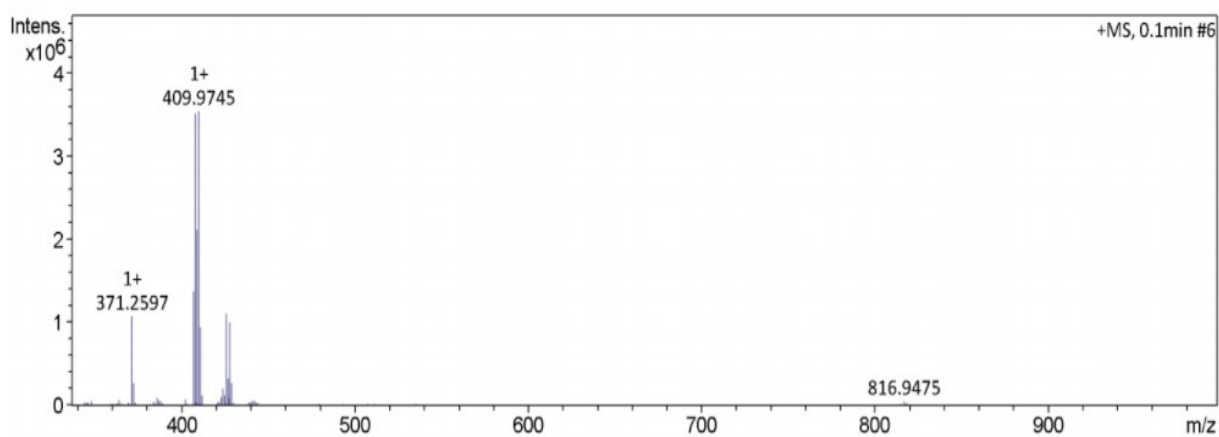


**Fig. S15** Transient EL spectra at different voltage of CACN-based device.

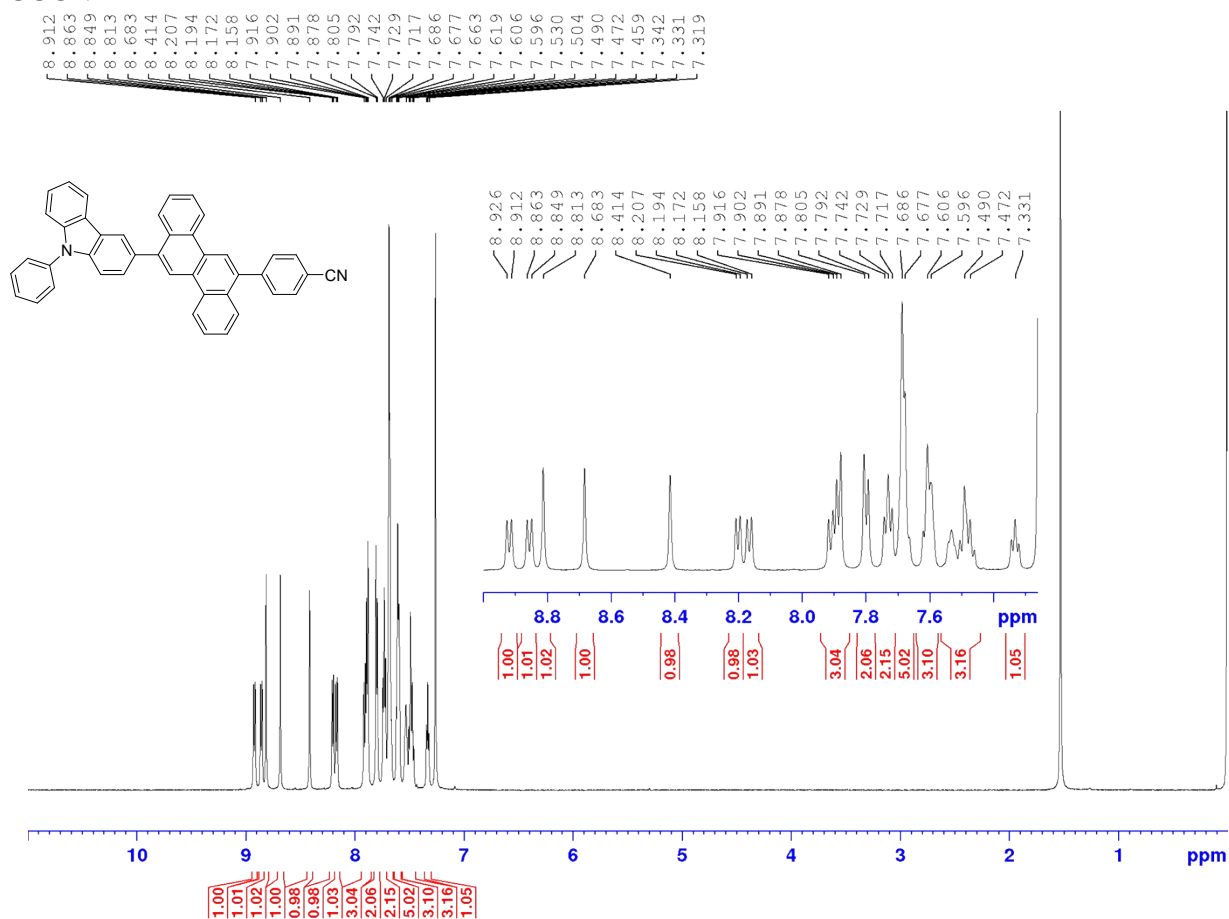
**Fig. S16** Copies of 600 MHz NMR ( $^1\text{H}/^{13}\text{C}$ ) spectra in  $\text{CDCl}_3$  and HRMS mass spectra of the synthesized compounds.

**Compound 2**

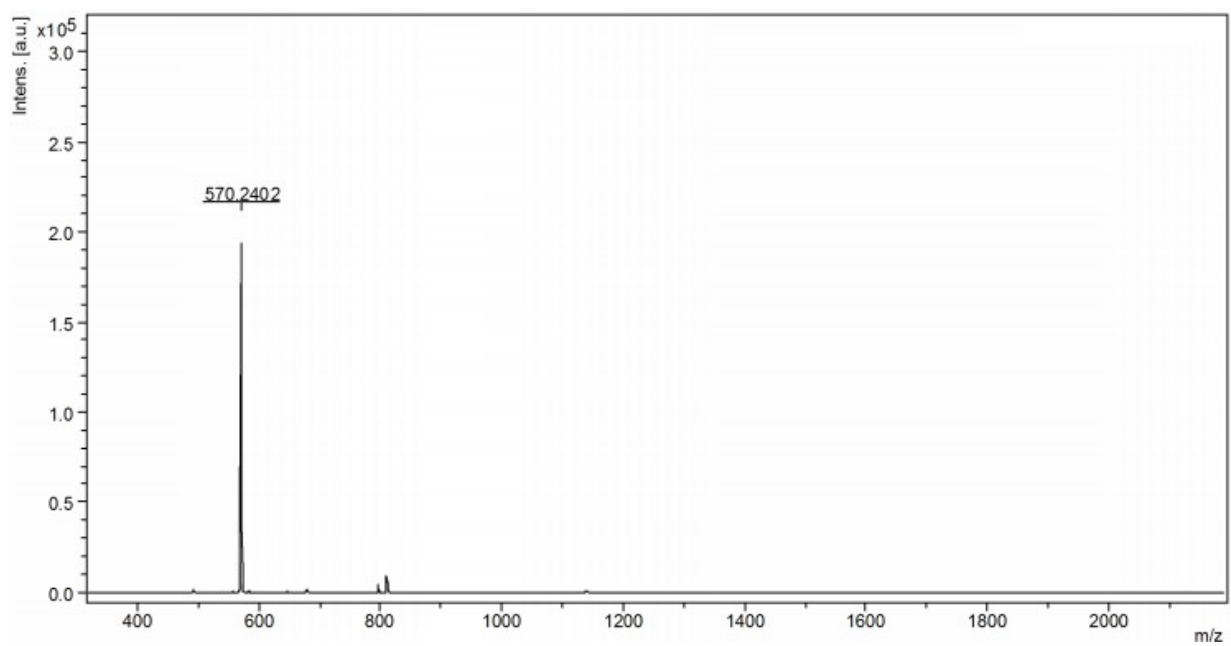
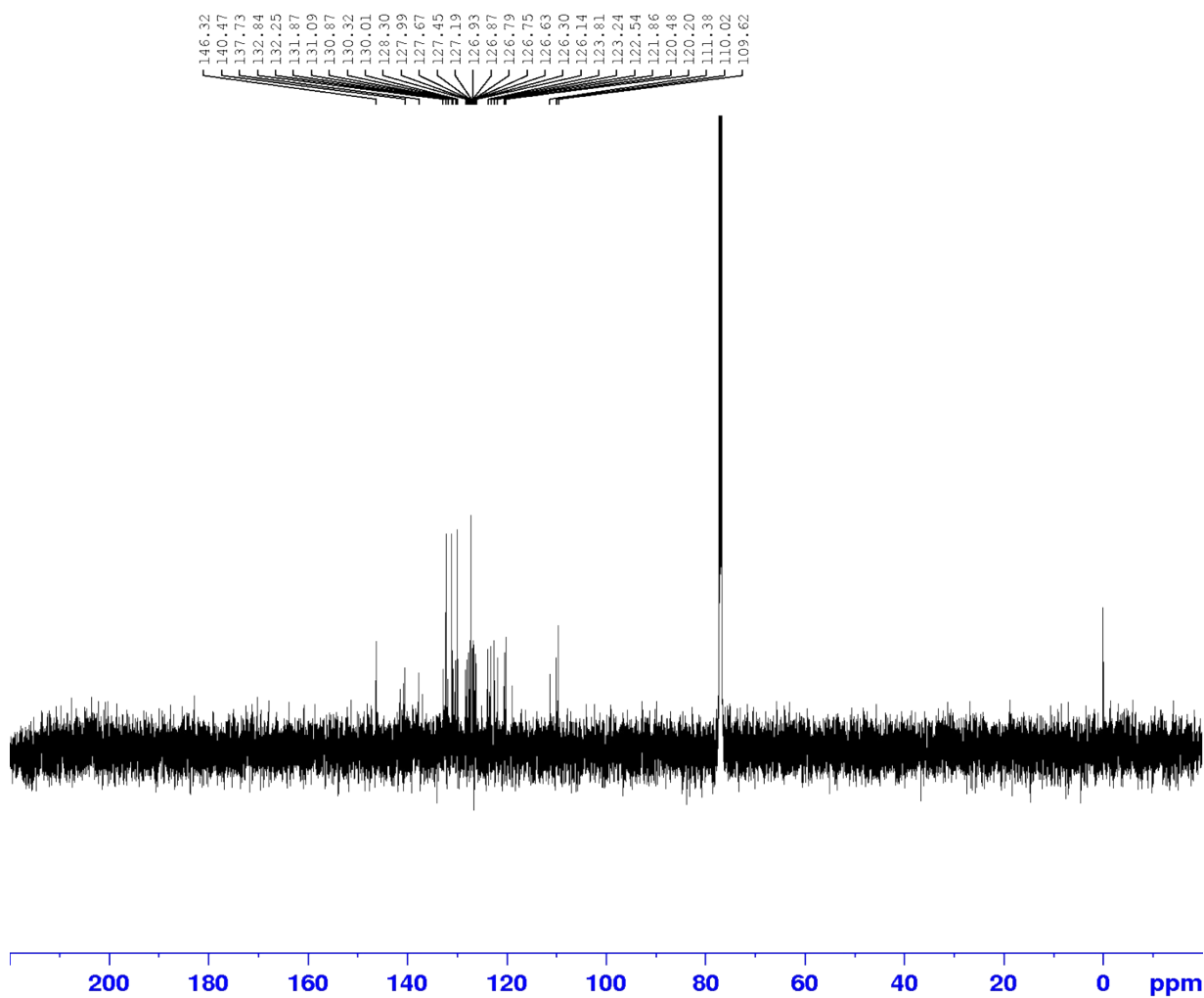




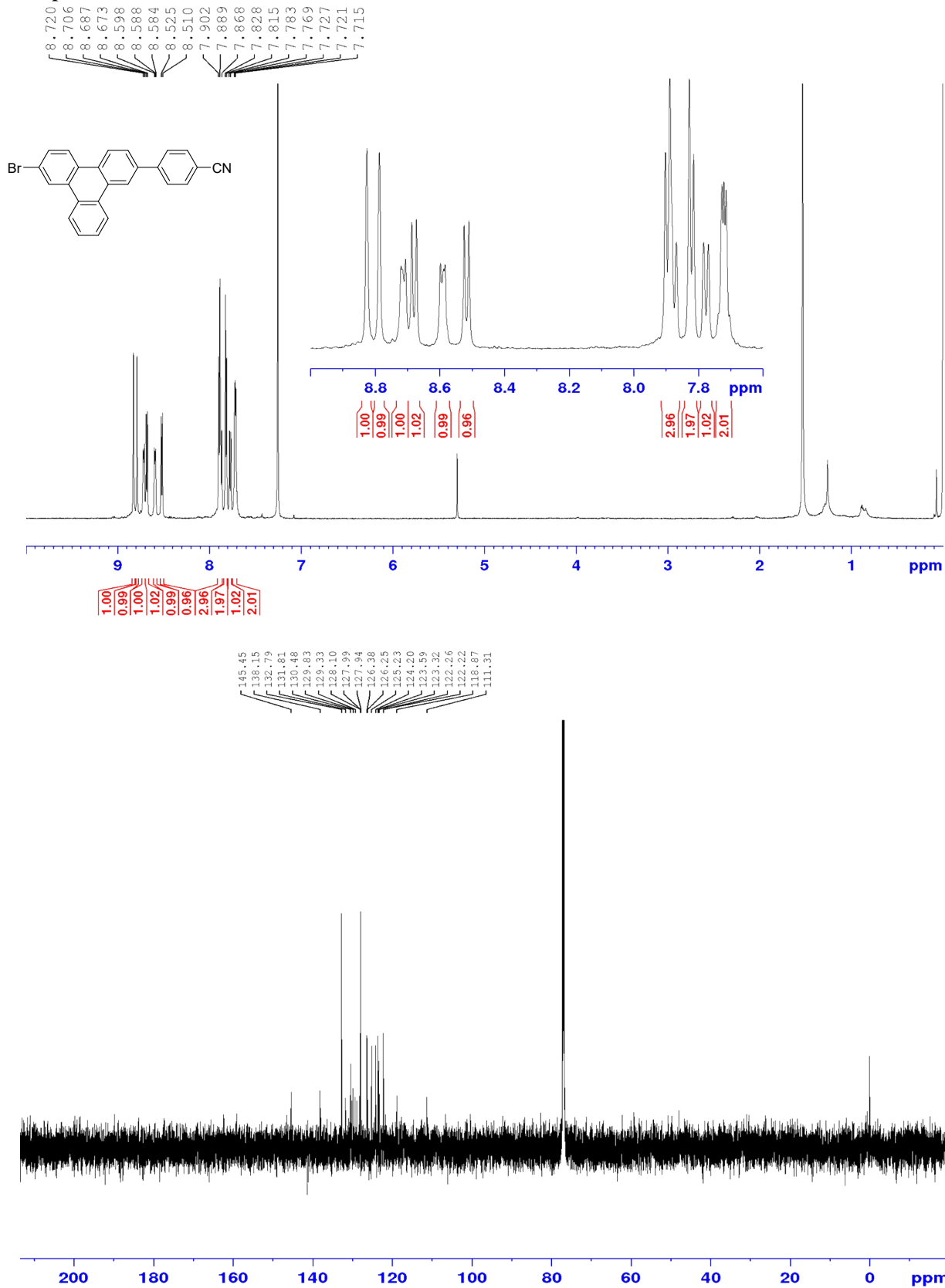
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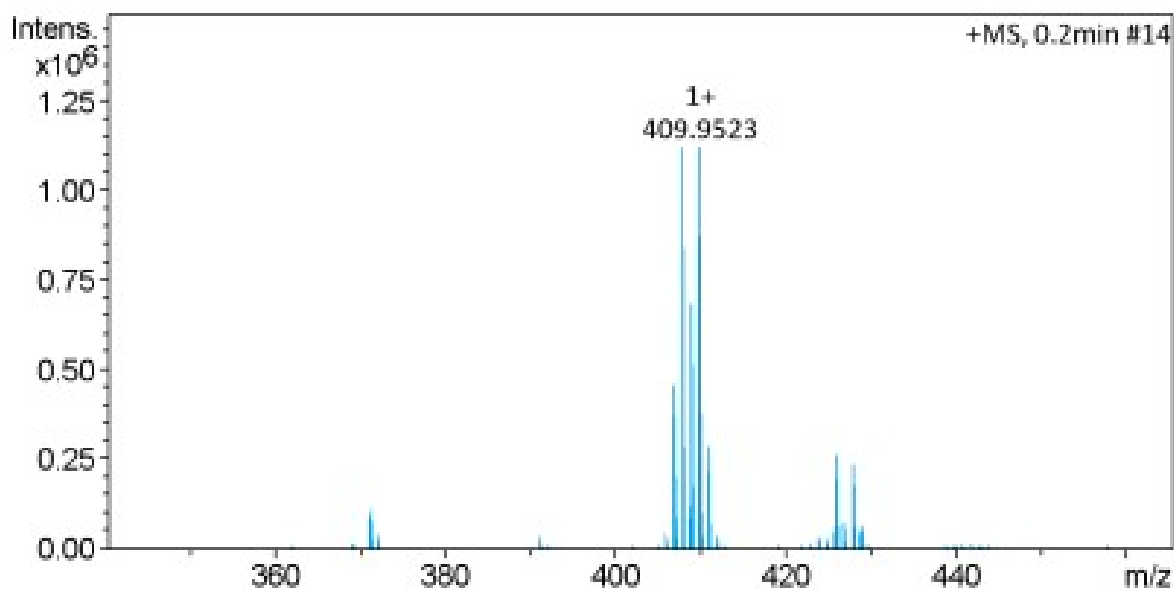




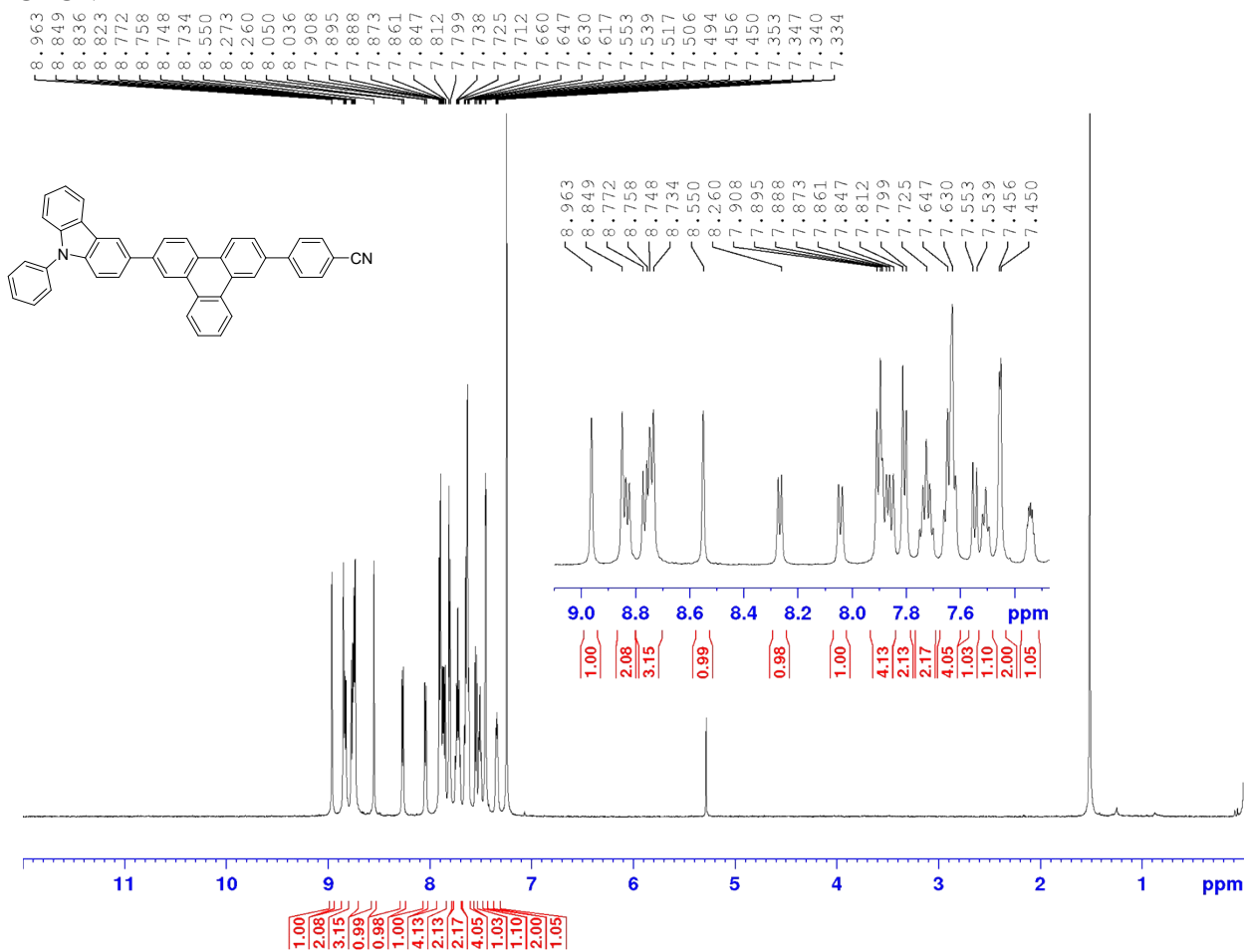


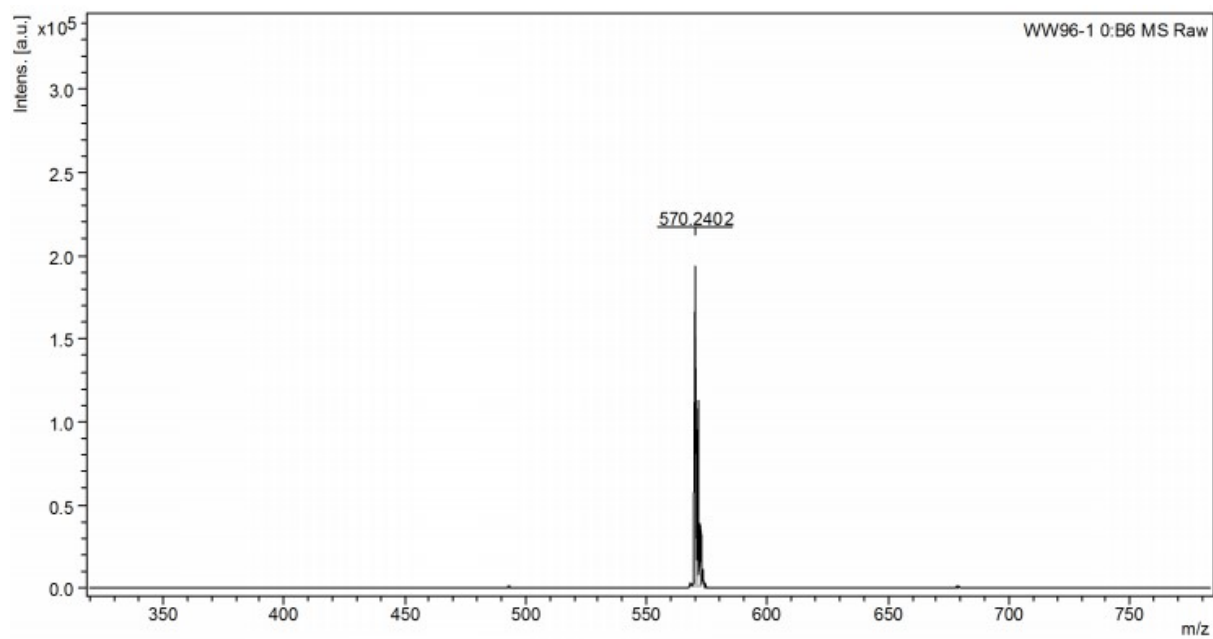
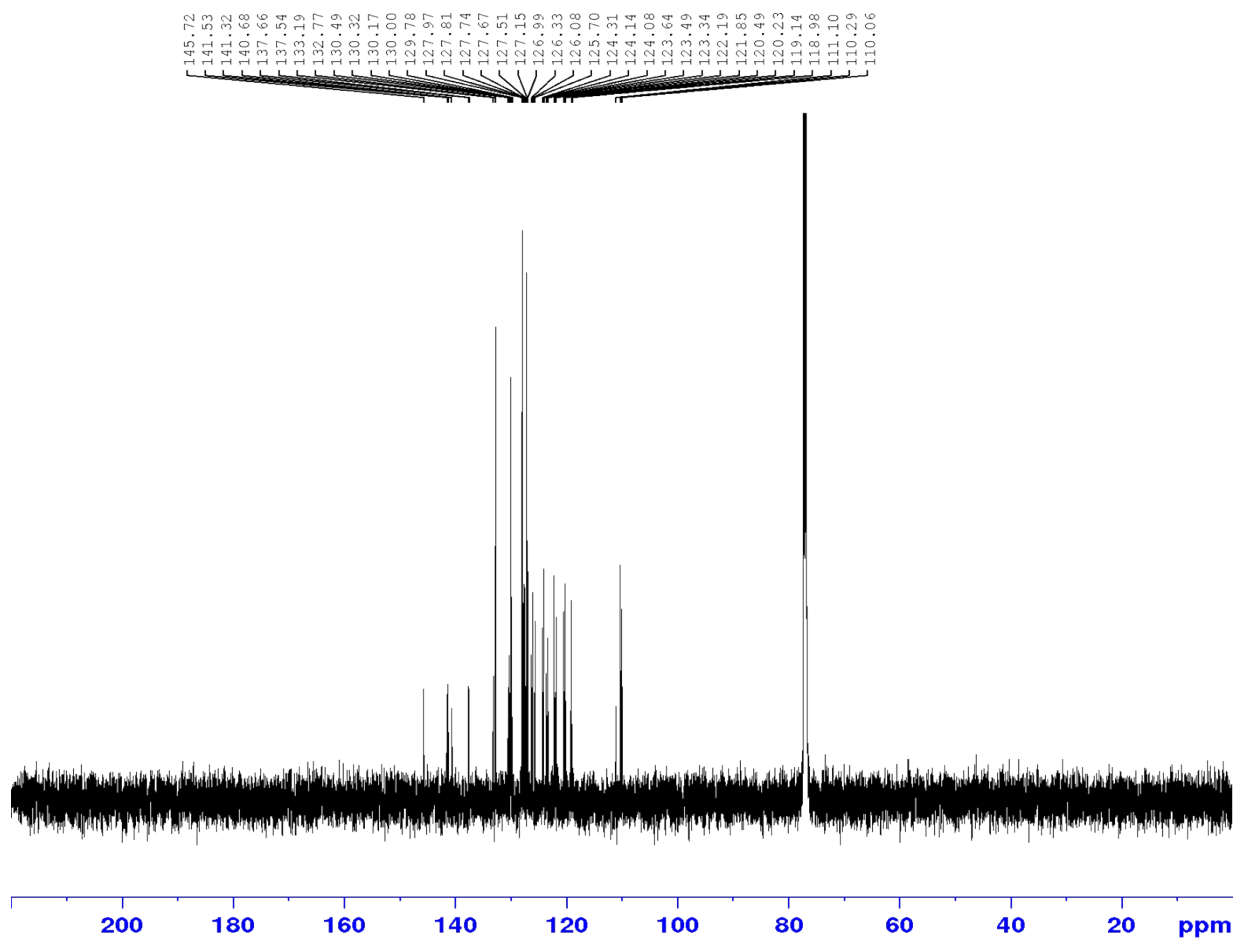
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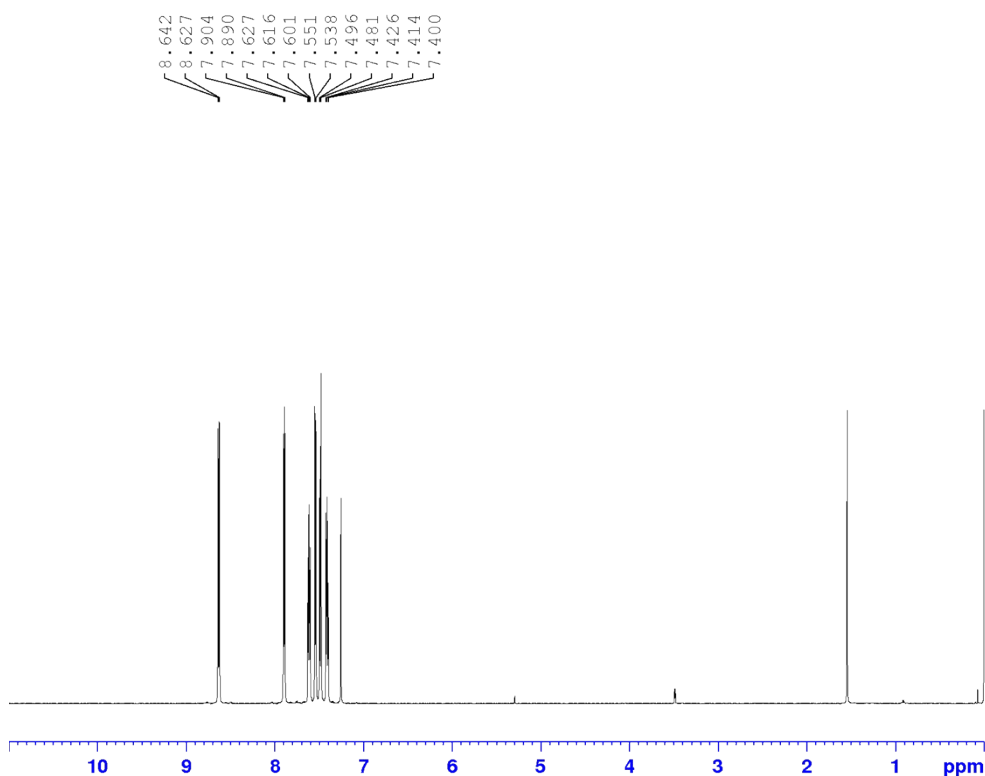


CTCN





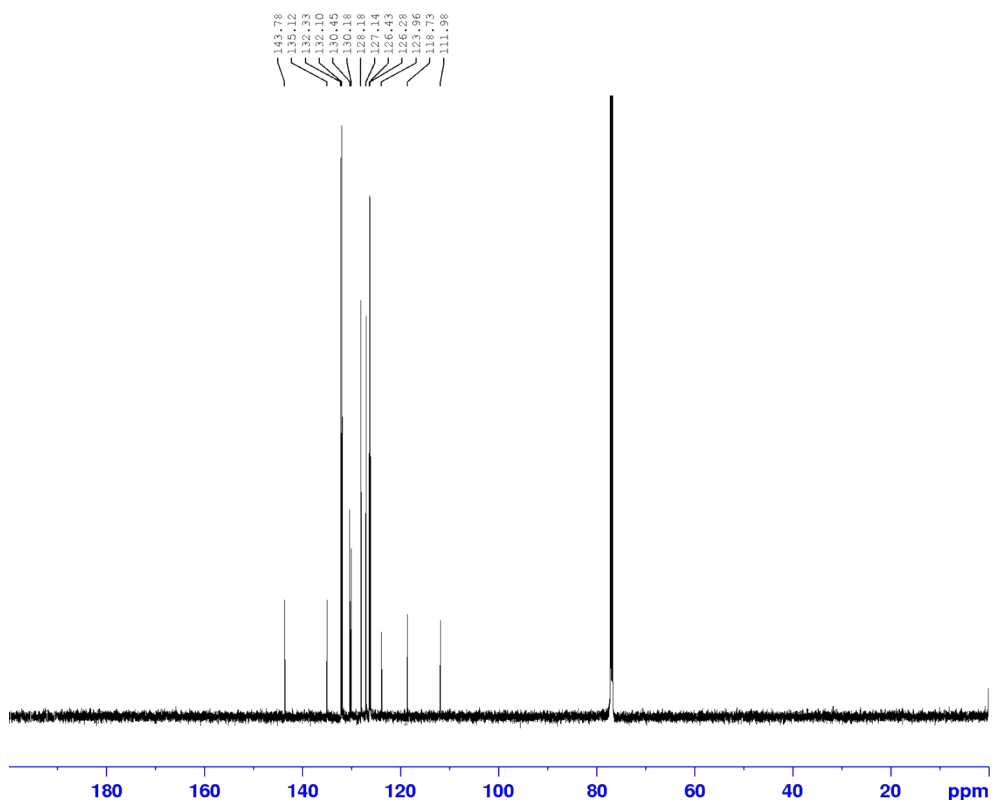
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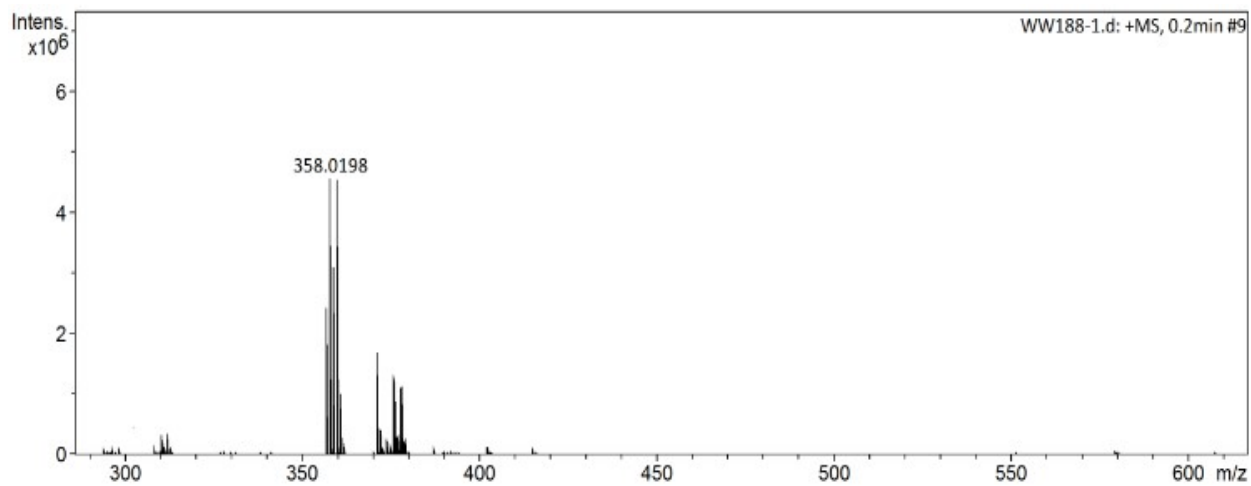


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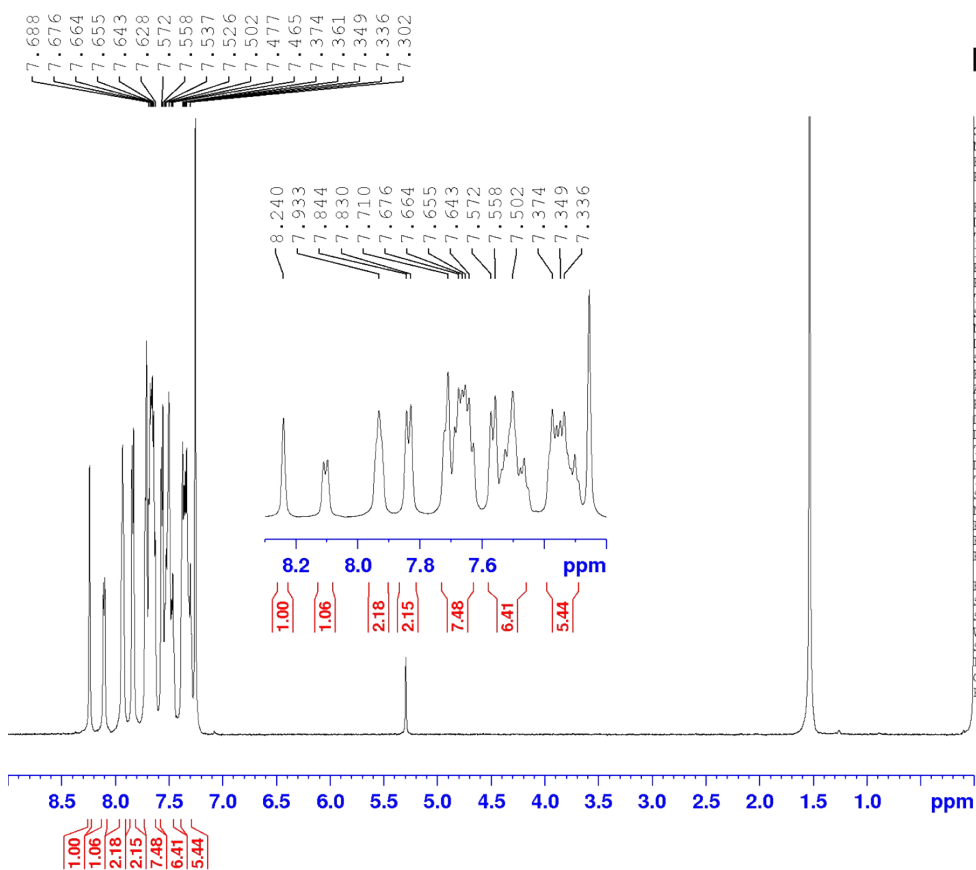
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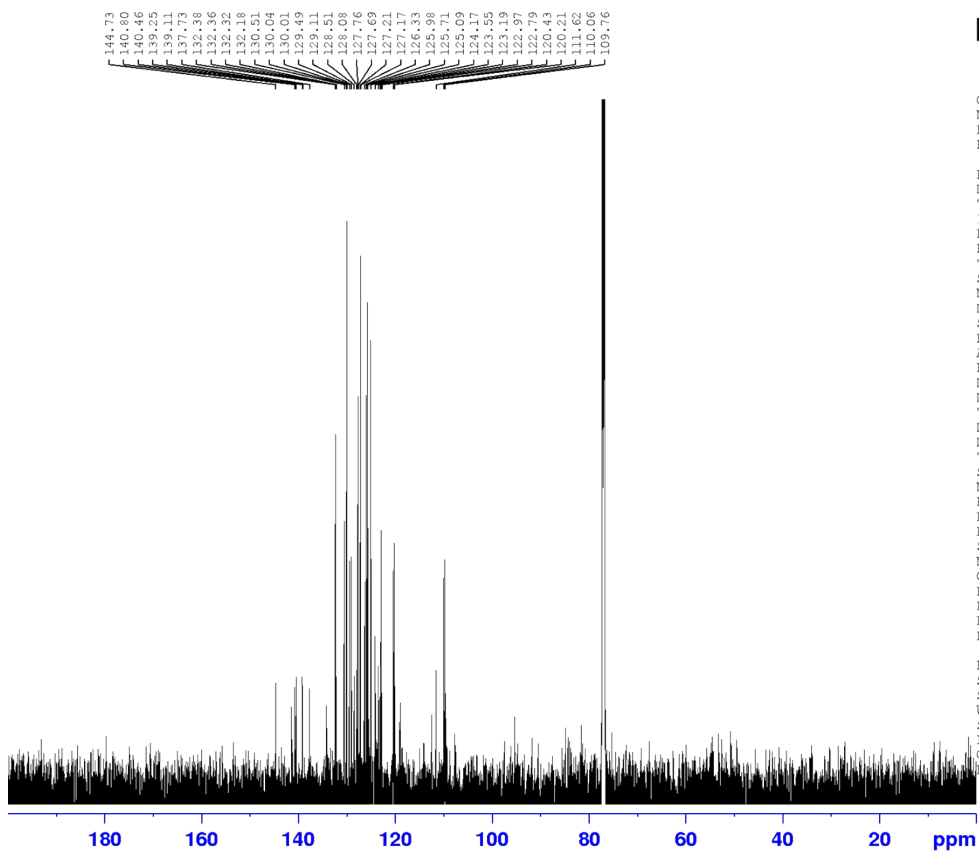
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Comment 1

Comment 2

