

## SUPPORTING INFORMATION

### **Spectroscopic Investigation of Phenazine-Based Donor–Acceptor Dyads with Thermally Activated Delayed Fluorescence: Heavy-Halogen Effects on Photosensitizing Properties**

Estera Hoffman,<sup>1</sup> Daria Grzywacz,<sup>2</sup> Michał Mońka,<sup>1</sup> Piotr Bojarski,<sup>1</sup> Illia E. Serdiuk<sup>1\*</sup>

<sup>1</sup> Faculty of Mathematics, Physics and Informatics, University of Gdańsk, Wita Stwosza 57, 80-308 Gdańsk, Poland

<sup>2</sup> Faculty of Chemistry, University of Gdańsk, Wita Stwosza 63, 80-308 Gdańsk, Poland

\*Corresponding author. E-mail: illia.serdiuk@ug.edu.pl, phone + 48 58 523 22 44

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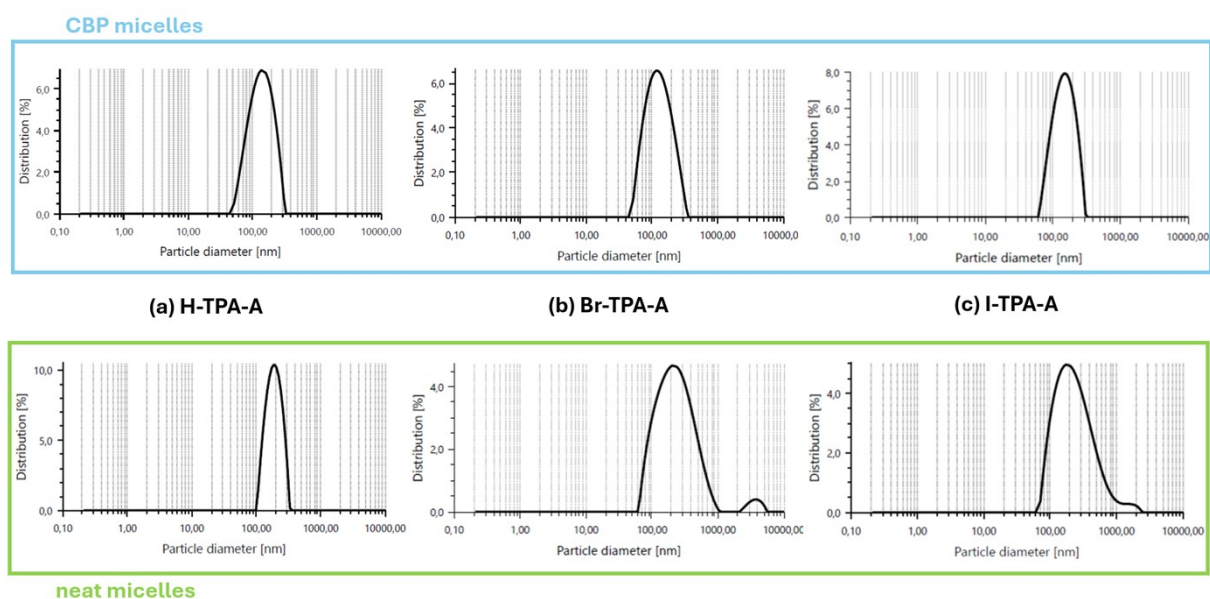
**Section S1.** DLS measurements

**Section S2.** Photoluminescence measurements

**Section S3.** Calculation results

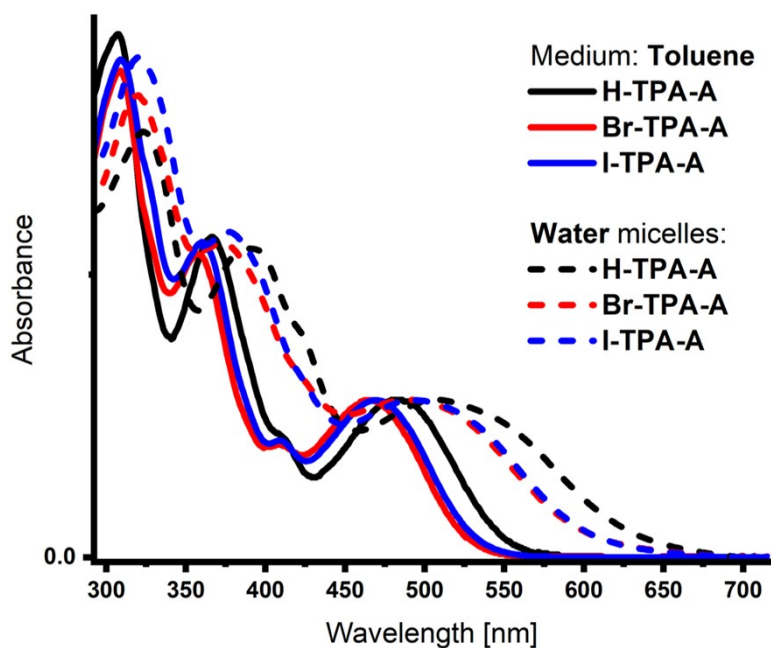
**Section S4.** NMR, MALDI-TOF spectra of synthesized compounds

## Section S1. DLS measurements

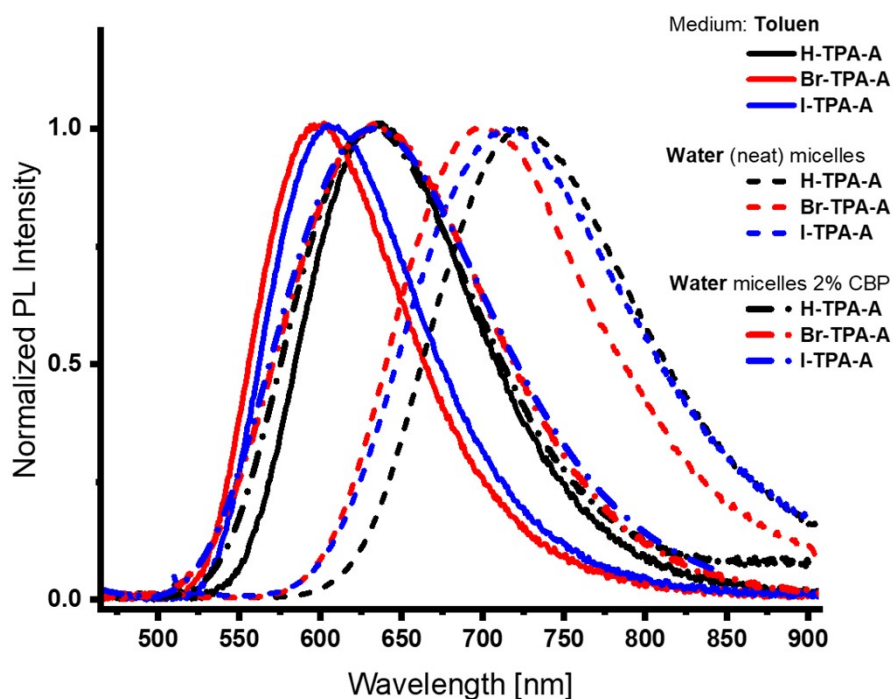


## Section S2. Photoluminescence measurements

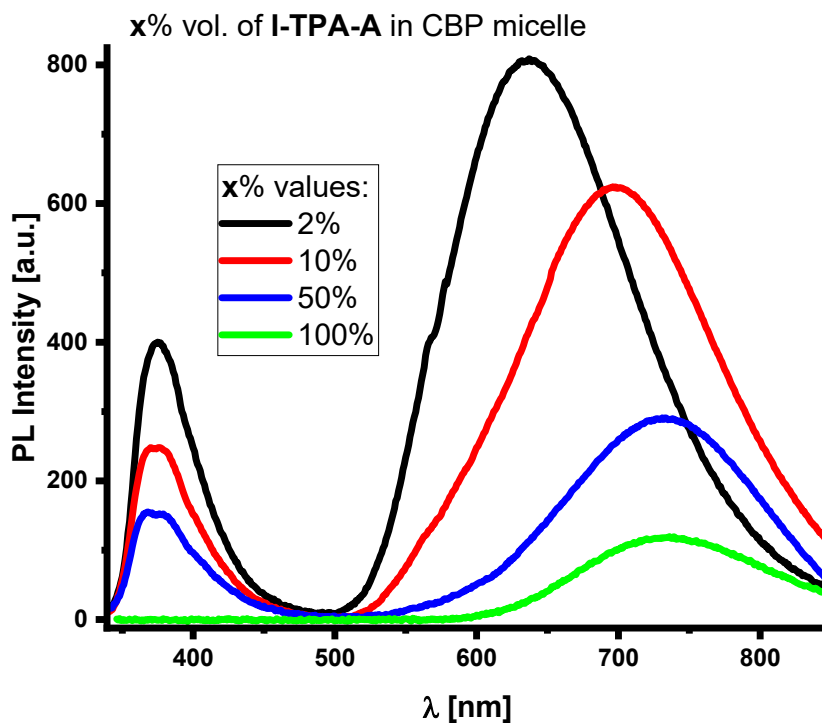
**Figure S1.** Particle size distributions based on DLS measurements for R-TPA-A for CBP micelles (blue frame) and neat micelles (green frame).



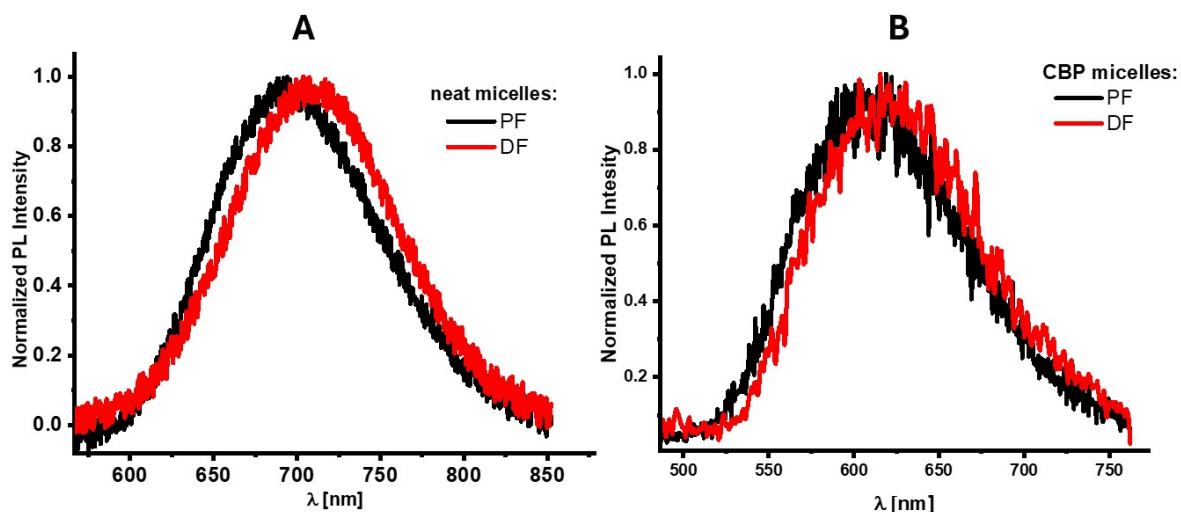
**Figure S2.** Absorption spectra of (A) R-TPA-A in toluene solutions vs. micellar solution in water.



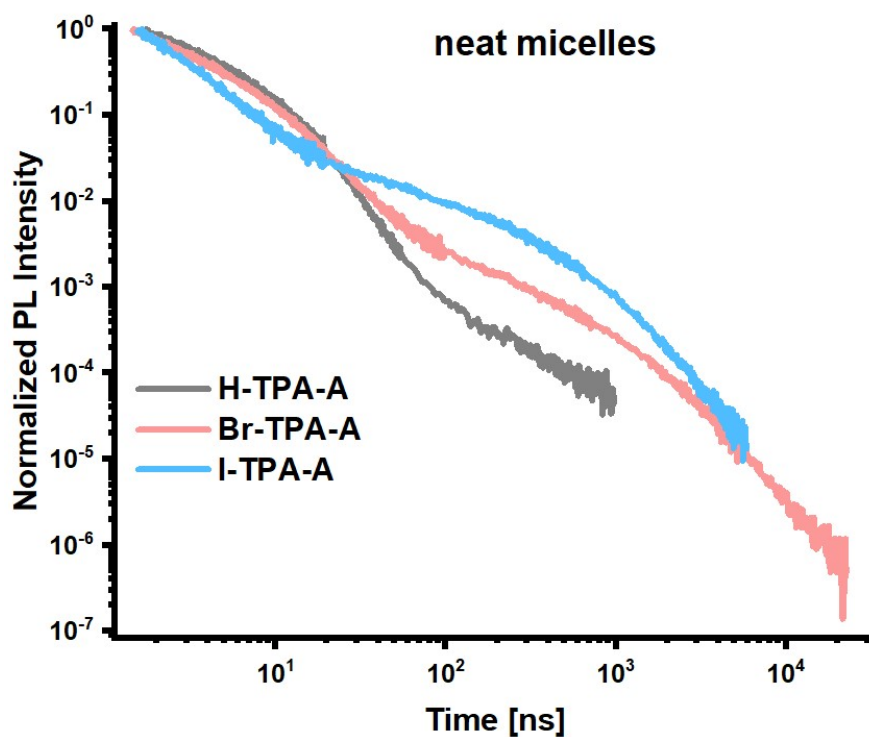
**Figure S3.** Emission spectra of **R-TPA-A** in different media: toluene solutions and water micelles with and without CBP.



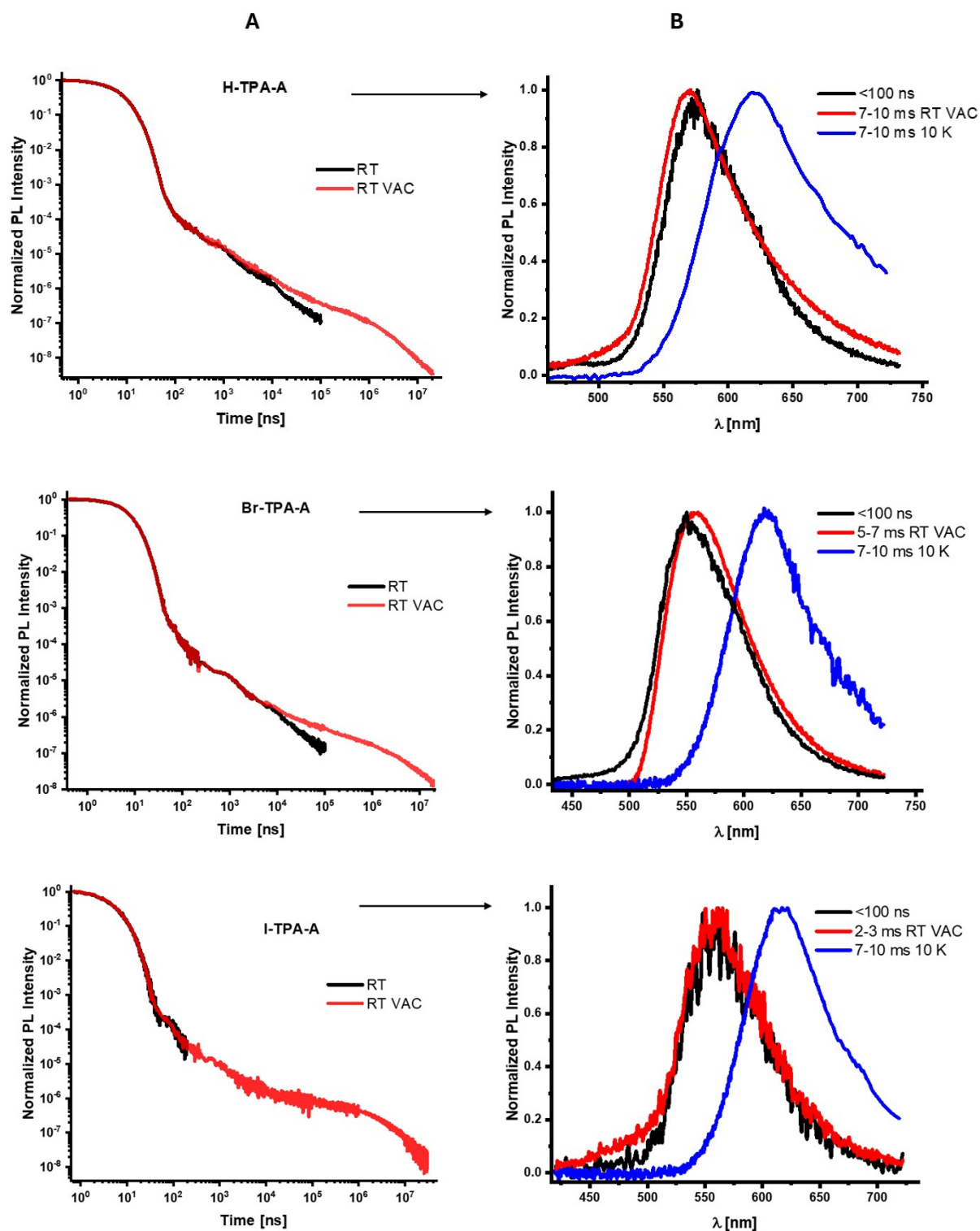
**Figure S4.** Dependence of emission spectral intensity on molar percentage of **I-TPA-A** in micelle.



**Figure S5.** Emission spectra of **Br-TPA-A** in prompt fluorescence (PF) and delayed fluorescence (DF) regime of (A) neat micelles and (B) CBP micelles.



**Figure S6.** The PL Decays of **R-TPA-A** without CBP in micelle (neat micelles).



**Figure S7.** The PL decays of R-TPA-A in ZNX films (A) and PL spectra for different time regime (B). Blue line represents phosphorescence detected in 10 K.

### Section S3. Calculation results

**Table S1.** Energy levels [eV] of HOMO and LUMO in  $S_0$ .

	structure	HOMO	LUMO
<b>H-TPA-A</b>	monomer	-6.75	-2.64
	dimer 1	-6.32	-2.65
<b>Br-TPA-A</b>	monomer	-6.86	-2.68
	dimer 1	-6.80	-2.94
<b>I-TPA-A</b>	monomer	-6.63	-2.66
	dimer 1	-6.56	-2.90

**Table S2.** The emission lifetimes, rate constants and decay parameters of **R-TPA-A** in ZNX films.

condi- -tions	structure	PLQY [%]	$\lambda_{\text{max, em}}$ [nm]	$\tau_{\text{PF}}$ [ns]	$\tau_{\text{DF}}$ [ $\mu\text{s}$ ]	$k_r$ [ $\cdot 10^7 \text{s}^{-1}$ ]	$k_{\text{ISC}}$ [ $\cdot 10^7 \text{s}^{-1}$ ]	$\phi_{\text{ISC}}$ [%]	$k_{\text{nr}}$ [ $\cdot 10^7 \text{s}^{-1}$ ]	$k_{\text{rISC}}$ [ $\cdot 10^2 \text{s}^{-1}$ ]	$\Delta E_{\text{ST}}$ [eV]
ZNX film	H-TPA-A	99	570	7.1	3505	13	0.8	6	0.1	3.0	0.121
	Br-TPA-A	99	555	5.4	5045	16	2.1	11	0.2	2.2	0.196
	I-TPA-A	67	560	4.7	5385	10	6.9	32	4.8	2.7	0.193

**Table S3.** Variants of dimers with Gibbs energy of formation ( $\Delta G$ ) for **R-TPA-A**.

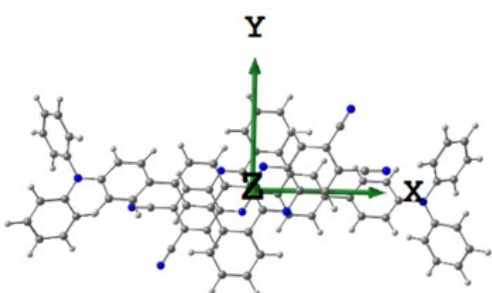
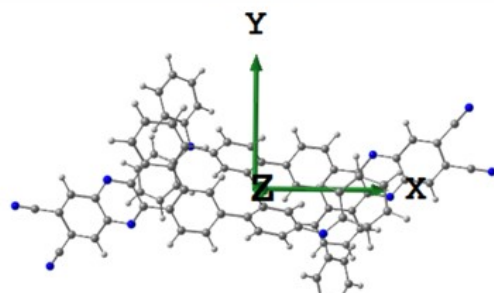
H-TPA-A	dimer 1	dimer 2
structure		
$\Delta G$	-0.613	-0.065

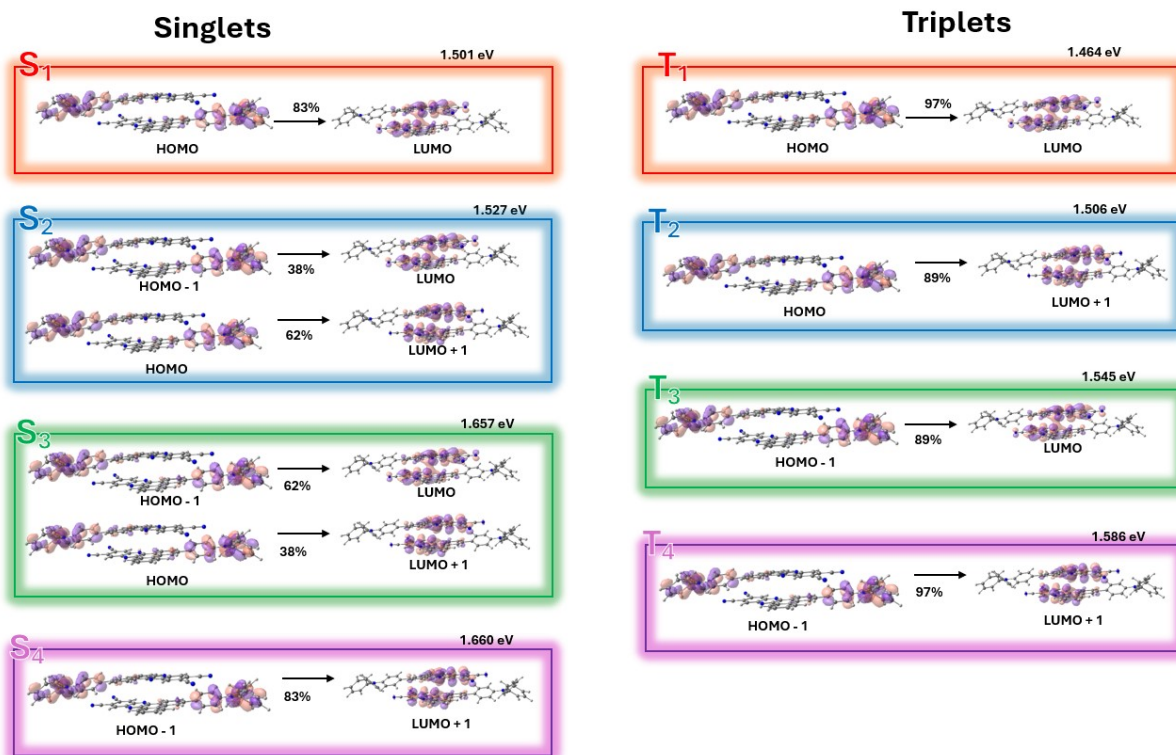
Table S3 continued

	dimer 1	dimer 2
<b>Br-TPA-A</b>		
<b>structure</b>		
<b><math>\Delta G</math></b>	-0.612	-0.205
<b>I-TPA-A</b>		
<b>structure</b>		
<b><math>\Delta G</math></b>	-0.674	-0.233

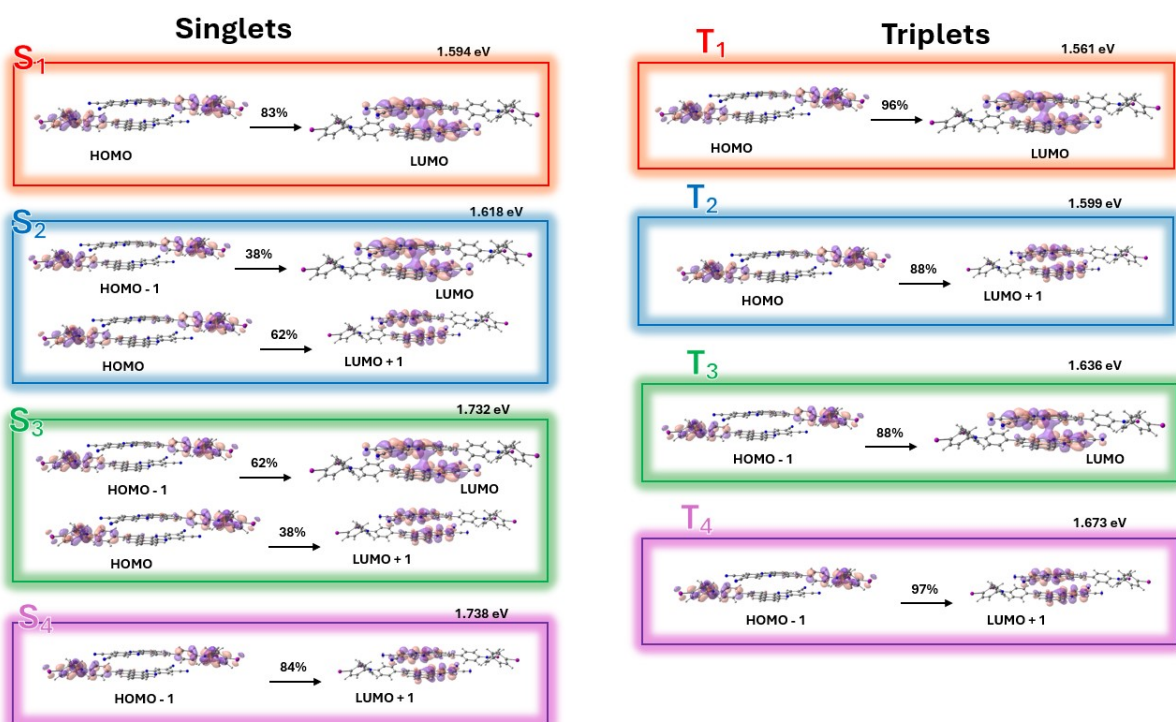
Table S3 continued

Br-TPA-A	dimer 3	dimer 4	dimer 5
structure			
$\Delta G$	-0.577	-0.218	-0.282
I-TPA-A	dimer 3	dimer 4	dimer 5
structure			
$\Delta G$	-0.531	-0.560	-0.336

\*Gibbs energy of dimer formation [eV]



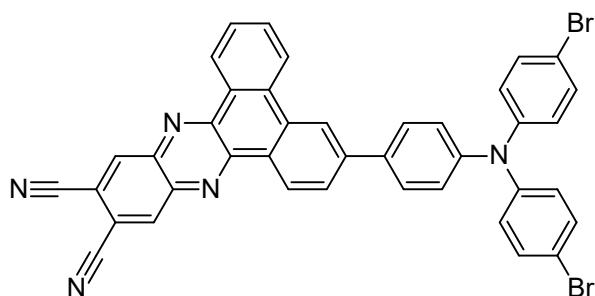
**Figure S8.** MO (molecule orbitals) for singlet ( $S_1$ - $S_4$ ) and triplet ( $T_1$ - $T_4$ ) states of **H-TPA-A**.



**Figure S9.** MO for singlet ( $S_1$ - $S_4$ ) and triplet ( $T_1$ - $T_4$ ) states of **I-TPA-A**.

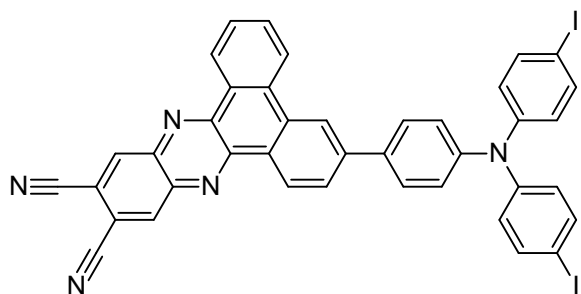
#### Section S4. NMR, MALDI-TOF spectra of synthesized compounds

##### 3-(4-(bis(4-bromophenyl)amino)phenyl)dibenzo[a,c]phenazine-11,12-dicarbonitrile (**Br-TPA-A**)

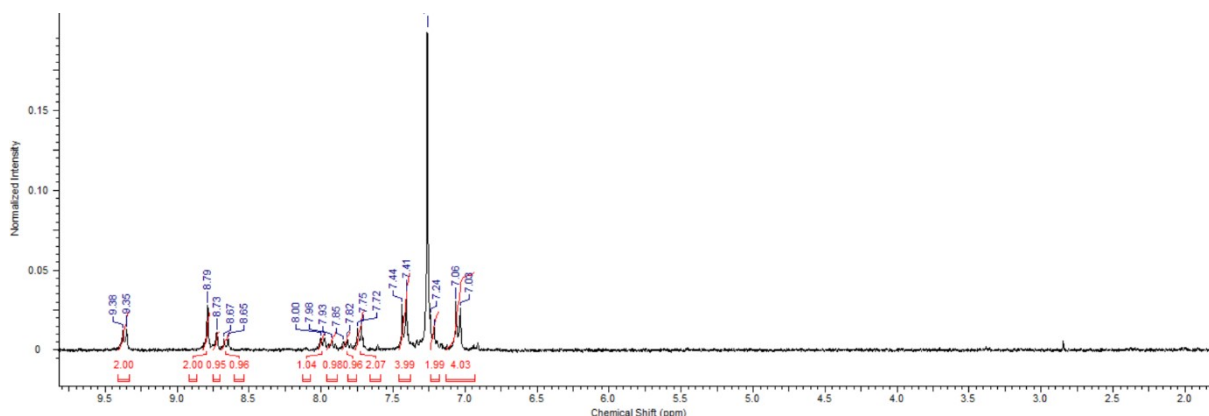


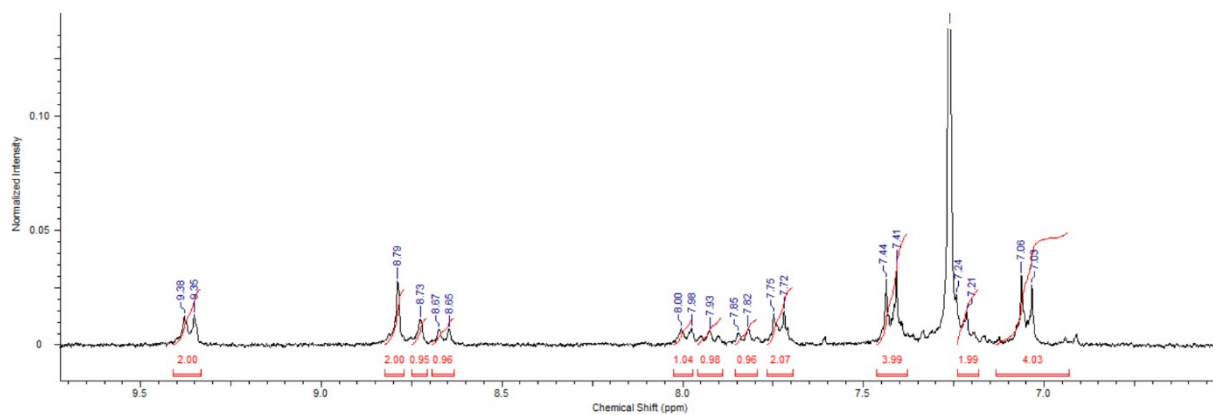
Red solid (55 mg, 87%).  $^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3-d$ ): d ppm 7.05 (d,  $J=8.80$  Hz, 4 H), 7.21 (m, 2 H), 7.42 (d,  $J=8.89$  Hz, 4 H), 7.73 (d,  $J=8.62$  Hz, 2 H), 7.83 (td,  $J=7.98$ , 0.82 Hz, 1 H), 7.93 (td,  $J=8.16$ , 0.82 Hz, 1 H), 7.99 (d,  $J=8.53$  Hz, 1 H), 8.66 (d,  $J=8.34$  Hz, 1 H), 8.73 (s, 1 H), 8.79 (s, 2 H), 9.37 (d,  $J=8.25$  Hz, 2 H); MALDI-TOF-MS:  $m/z$  calcd for  $\text{C}_{40}\text{H}_{21}\text{Br}_2\text{N}_5$  731.45, found 731.0  $[M]^+$ . Anal. Calcd., %: C 65.68; H 2.89; N 9.57. Found, %: C 65.59; H 2.92; N 9.56.

##### 3-(4-(bis(4-iodophenyl)amino)phenyl)dibenzo[a,c]phenazine-11,12-dicarbonitrile (**I-TPA-A**)

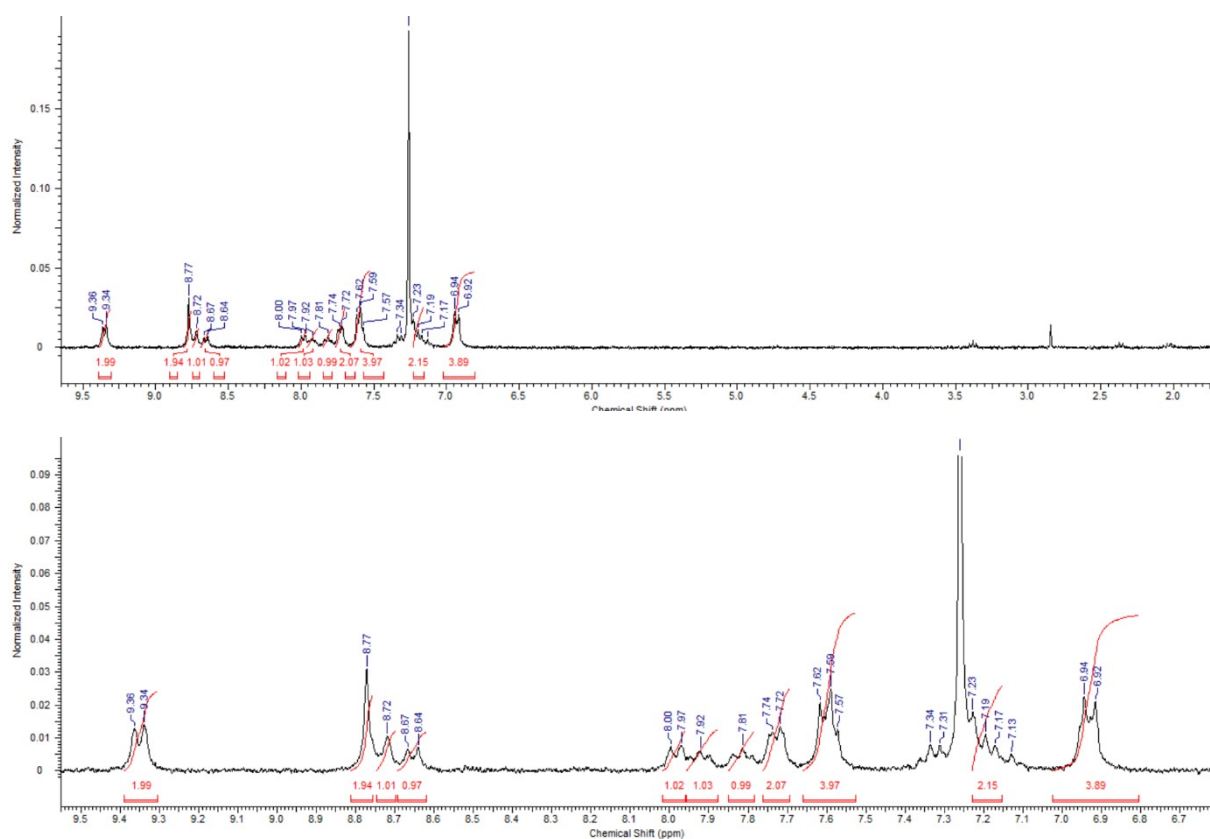


Dark red solid (57 mg, 79%).  $^1\text{H NMR}$  (500 MHz,  $\text{CDCl}_3-d$ ): d ppm 6.93 (d,  $J=8.53$  Hz, 4 H), 7.21 (m, 2 H), 7.60 (d,  $J=8.34$  Hz, 4 H), 7.73 (d,  $J=5.78$  Hz, 2 H), 7.81 (t,  $J=7.43$ , 1 H), 7.92 (t,  $J=7.70$ , 1 H), 7.98 (d,  $J=8.53$  Hz, 1 H), 8.65 (d,  $J=8.07$  Hz, 1 H), 8.72 (br. s., 1 H), 8.77 (s, 2 H), 9.35 (d,  $J=7.15$  Hz, 2 H); MALDI-TOF-MS:  $m/z$  calcd for  $\text{C}_{40}\text{H}_{21}\text{I}_2\text{N}_5$  825.45, found 827.0  $[M+1]^+$ . Anal. Calcd., %: C 58.20; H 2.56; N 8.48. Found, %: C 58.01; H 2.61; N 8.42.

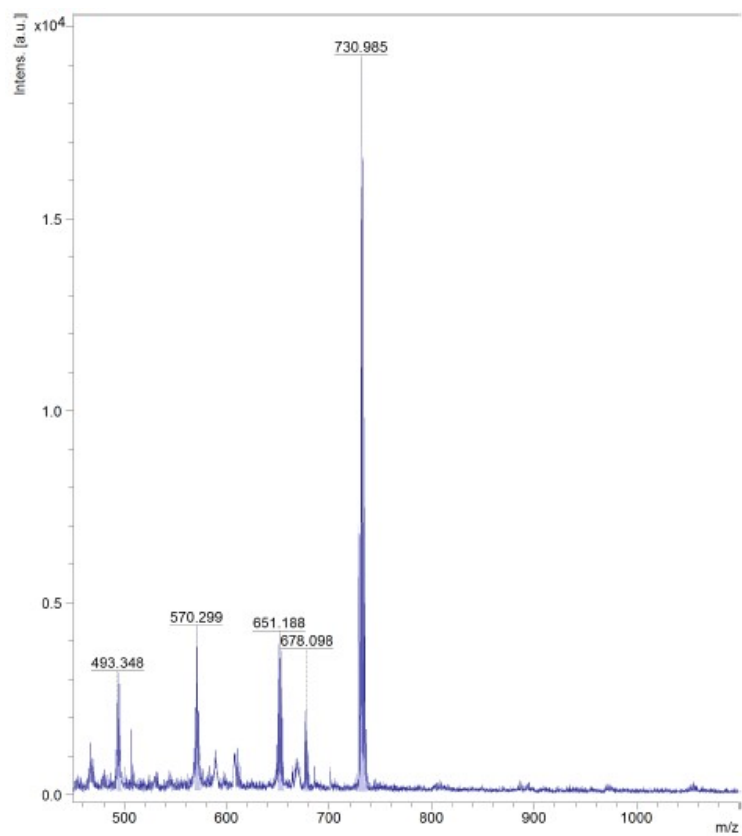




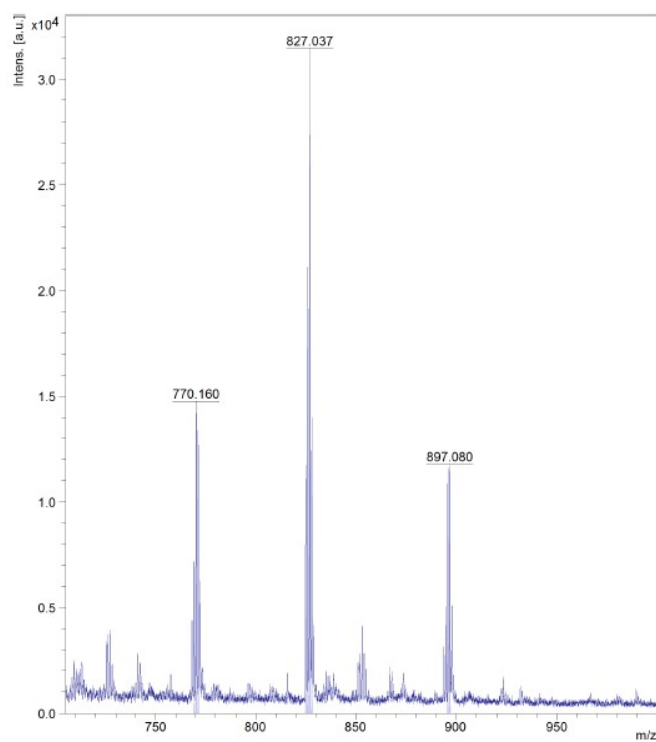
**Figure S10.** <sup>1</sup>H NMR spectra of 3-(4-(bis(4-bromophenyl)amino)phenyl)dibenzo[*a,c*]-phenazine-11,12-dicarbonitrile (**Br-TPA-A**) in CDCl<sub>3</sub>.



**Figure S11.** <sup>1</sup>H NMR spectra of 3-(4-(bis(4-iodophenyl)amino)phenyl)dibenzo[*a,c*]-phenazine-11,12-dicarbonitrile (**I-TPA-A**) in CDCl<sub>3</sub>.



**Figure S12.** MALDI-TOF mass spectrum of **Br-TPA-A**.



**Figure S13.** MALDI-TOF mass spectrum of **I-TPA-A**.