

Electronic Supplementary Information (ESI)

for the article

Exploitation of printed circuit board feeds for high-purity copper recovery using hydrometallurgy: process and life cycle assessment considerations

N. Kelly^{1*#}, P. Boelens^{1,2*#}, A.M. Parvez^{1,3}, S. Kutschke¹, D. Ebert¹, B.M. Guy¹, R. Möckel¹, M.H. Aamir¹, C. Sanchez-Garrido¹, U. Fischer¹, M. Sajjad¹, A.D. Renno¹, L. Ott⁴, F. Ellinger⁴, A.B. Patil^{1,5}, J. Gutzmer¹

Table S1: Energy and resources provider

Energy and Resources	Provider
Electricity	Electricity, medium voltage APOS, S – NO
Hydrogen peroxide	market for hydrogen peroxide, without water, in 50% solution state APOS, S
Nitrogen	market for nitrogen, liquid nitrogen, liquid APOS, S
Sulfuric acid	market for sulfuric acid sulfuric acid APOS, S
Sodium hydroxide	market for sodium hydroxide APOS, S
Water, decarbonised	market for water, decarbonised water, decarbonised APOS, S

Table S2: Concentration of selected metal ions (in mg·L⁻¹) in leach solutions for size fraction 630µm-1mm. c(H₂SO₄) = 3 mol·L⁻¹, c(H₂O₂) = 3 wt% (addition in 60 min), S/L = 1:10, t = 12 h., r.t.; c(HNO₃) = 3 mol·L⁻¹, S/L = 1:3, t = 61 h, r.t.; BDL: below detection limit.

Leaching system	Sulfuric acid	Nitric acid
Metal ion	Concentration in mg·L ⁻¹	Concentration in mg·L ⁻¹
Cu ²⁺	32600	96100
Fe ³⁺	7.51	19.4
Ni ²⁺	841	2550
Au ³⁺	BDL	0.397
Al ³⁺	527	2000

Ca ²⁺	265	2800
Ba ²⁺	<1	59
Si ⁴⁺	21.5	14.7

Table S3: Impact assessment results, expressed per kg of copper, without economic allocation

Category	Unit	Impact assessment results	
		H ₂ SO ₄	HNO ₃
agricultural land occupation (ALOP)	m ² a	1.87E+00	1.13E+00
climate change (GWP100)	kg CO ₂ -Eq	4.03E+01	2.77E+01
fossil depletion (FDP)	kg oil-Eq	1.39E+01	7.01E+00
freshwater ecotoxicity (FETPinf)	kg 1,4-DCB-Eq	9.34E+00	7.94E+00
freshwater eutrophication (FEP)	kg P-Eq	3.19E-02	2.17E-02
human toxicity (HTPinf)	kg 1,4-DCB-Eq	7.82E+01	6.29E+01
ionising radiation (IRPHE)	kg U235-Eq	4.33E+00	1.64E+00
marine ecotoxicity (METPinf)	kg 1,4-DCB-Eq	8.48E+00	7.20E+00
marine eutrophication (MEP)	kg N-Eq	6.08E-02	3.62E-02
metal depletion (MDP)	kg Fe-Eq	1.94E+01	1.67E+01
natural land transformation (NLTP)	m ²	9.22E-03	6.93E-03
ozone depletion (ODPinf)	kg CFC-11-Eq	1.20E-05	2.13E-06
particulate matter formation (PMFP)	kg PM10-Eq	1.97E-01	1.36E-01
photochemical oxidant formation (POFP)	kg NMVOC	2.04E-01	1.26E-01
terrestrial acidification (TAP100)	kg SO ₂ -Eq	6.11E-01	5.19E-01
terrestrial ecotoxicity (TETPinf)	kg 1,4-DCB-Eq	1.80E-02	1.19E-02
urban land occupation (ULOP)	m ² a	1.22E+00	9.02E-01
water depletion (WDP)	m ³	6.17E-01	4.53E-01

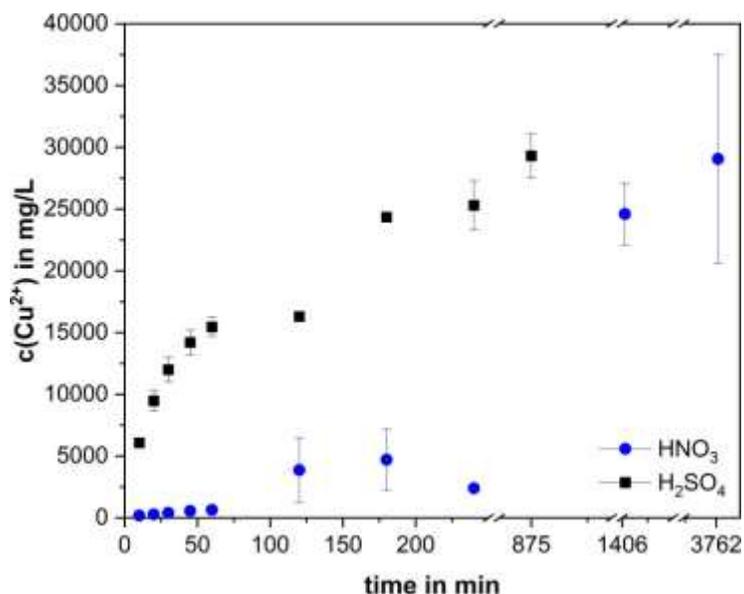


Figure S1: Leaching of copper (concentration in $\text{mg}\cdot\text{L}^{-1}$) from PCB material ($630\ \mu\text{m} - 1\ \text{mm}$) in dependence on reaction time. $c(\text{H}_2\text{SO}_4) = 3\ \text{mol}\cdot\text{L}^{-1}$, $c(\text{H}_2\text{O}_2) = 3\ \text{wt}\%$; $c(\text{HNO}_3) = 3\ \text{mol}\cdot\text{L}^{-1}$, $S/L = 1:10$, $t = 10\ \text{min}-3765\ \text{min}$, r.t.

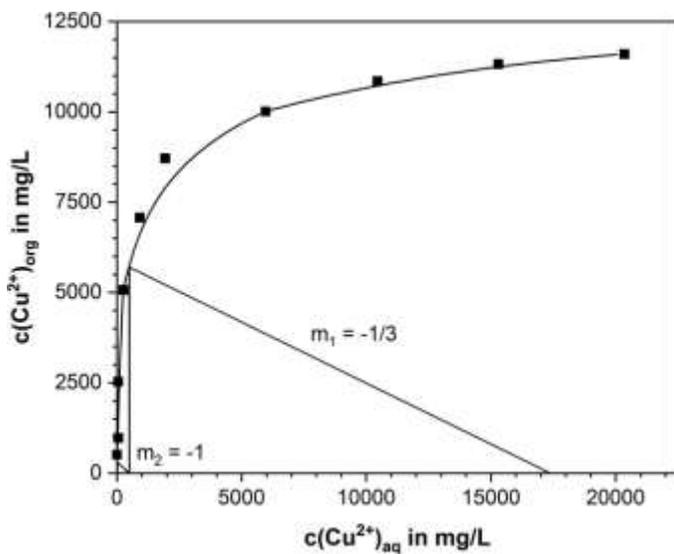


Figure S2: The McCabe-Thiele diagram for $\text{CuSO}_4\text{-NiSO}_4\text{-Na}_2\text{SO}_4\text{-H}_2\text{SO}_4/\text{LIX 84-I-kerosene}$. $\text{Cu(II)}_{\text{ini}} = 0.5\text{-}31.9\ \text{g/L}$, $\text{Ni(II)}_{\text{ini}} = 214\ \text{mg/L}$, $c(\text{Na}_2\text{SO}_4) = 0.5\ \text{mol}\cdot\text{L}^{-1}$, $\text{pH}_{\text{eq}} = 1.35 \pm 0.15$, 20 vol% LIX 84-I, $t = 15\ \text{min}$, r.t.. Error bars represent mean \pm half range ($n=2$).

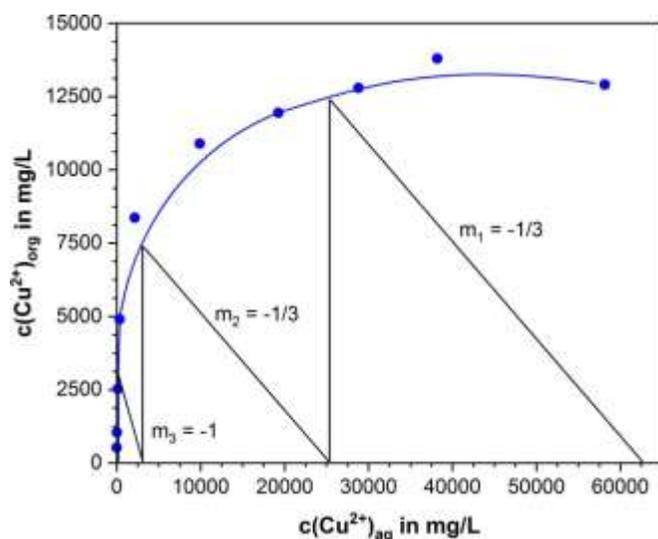


Figure S3: The McCabe-Thiele diagram for $\text{Cu}(\text{NO}_3)_2\text{-Ni}(\text{NO}_3)_2\text{-NaNO}_3\text{-HNO}_3/\text{LIX 84-I-kerosene}$. $\text{Cu}(\text{II})_{\text{ini}} = 0.5\text{-}71.1$ g/L, $\text{Ni}(\text{II})_{\text{ini}} = 743$ mg/L, $c(\text{NaNO}_3) = 1$ mol·L⁻¹, $\text{pH}_{\text{eq}} = 1.35 \pm 0.15$, 20 vol% LIX 84-I, $t = 15$ min, r.t.. Error bars represent mean \pm half range ($n=2$).

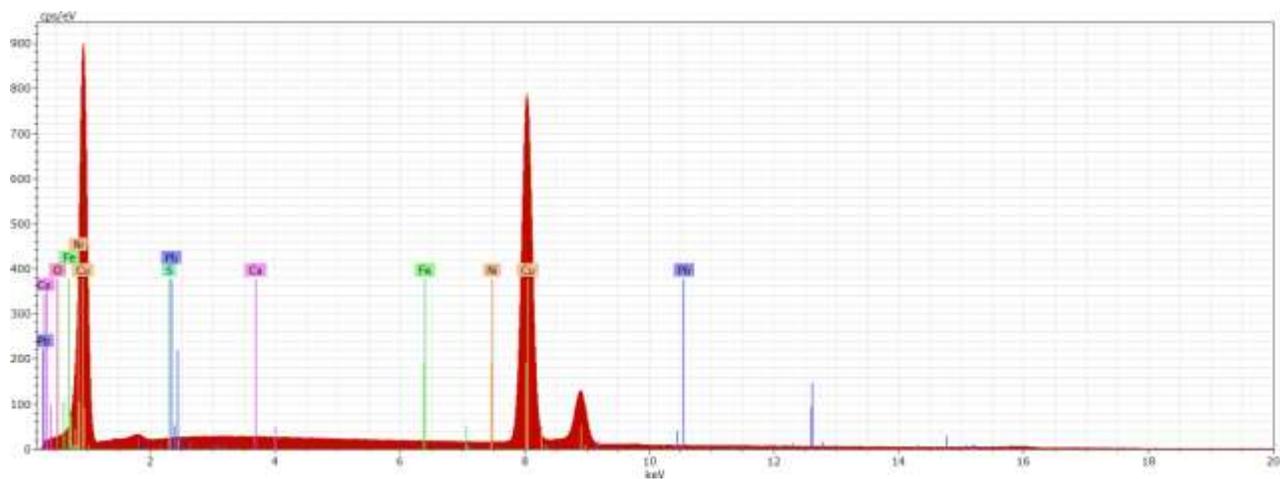


Figure S4: EDX data for the deposited Cu samples showing high purity of Cu