

[Electronic Supplementary Information]

Title: **ψ -BCN Monolayers as Emerging 2D Materials: Effects of Hydrogen Passivation on Structure, Stability, and Functionality**

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- Figure S1.** Initial and optimized configuration with different ratio of B, C, and N atoms with relative energy (ΔE)
- Figure S2.** Hydrogen passivation on multiple different atomic sites (a) H-passivation on B and C atoms (b) H-passivation on B and N atoms
- Figure S3.** Polar plots of (a) directional Young's modulus and (b) directional Poisson's ratio for hydrogenated ψ -BCN monolayers in comparison with pristine ψ -BCN.
- Figure S4.** Total potential energy fluctuation of the ψ -BCN monolayer at 300 K, 600 K, and 1000 K during the AIMD simulations. The right panels display the final structural snapshots at each temperature, confirming the thermal stability and structural integrity of the monolayer.
- Figure S5.** Total potential energy fluctuation of the hydrogenated ψ -BCN monolayer at 1000 K during the AIMD simulations. The right panels display the final structural snapshots of each hydrogenated structure, confirming the thermal stability and structural integrity of the monolayer.
- Figure S6.** Electron localization function (ELF) maps of ψ -BCN monolayers: (a) pristine ψ -BCN, (b) ψ -B_HCN, (c) ψ -BC_HN, and (d) ψ -BCN_H. Nitrogen atoms are shown in blue, boron in green, and carbon in grey. The color scale (0.0-1.0) indicates the degree of electron localization, with higher values representing stronger localization. Hydrogenation at C and B maintains the overall ELF topology, whereas hydrogenation at N leads to pronounced localization around the N-H bond and noticeable perturbation of nearby bonding regions.

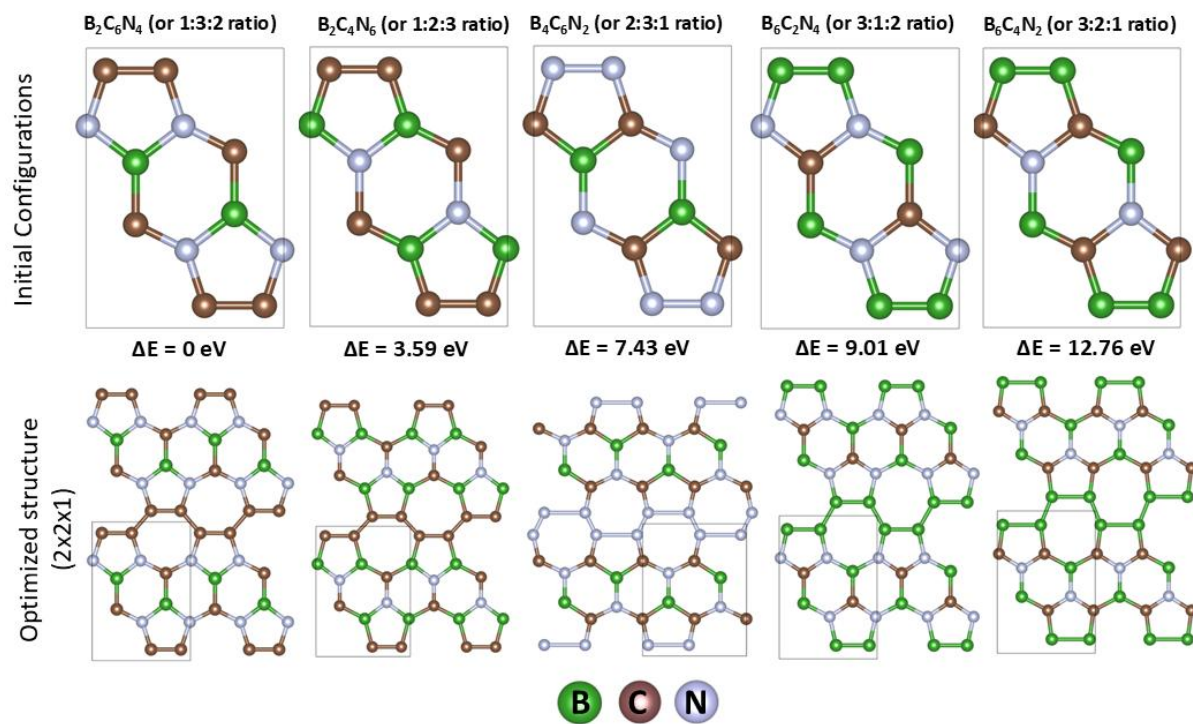


Figure S1

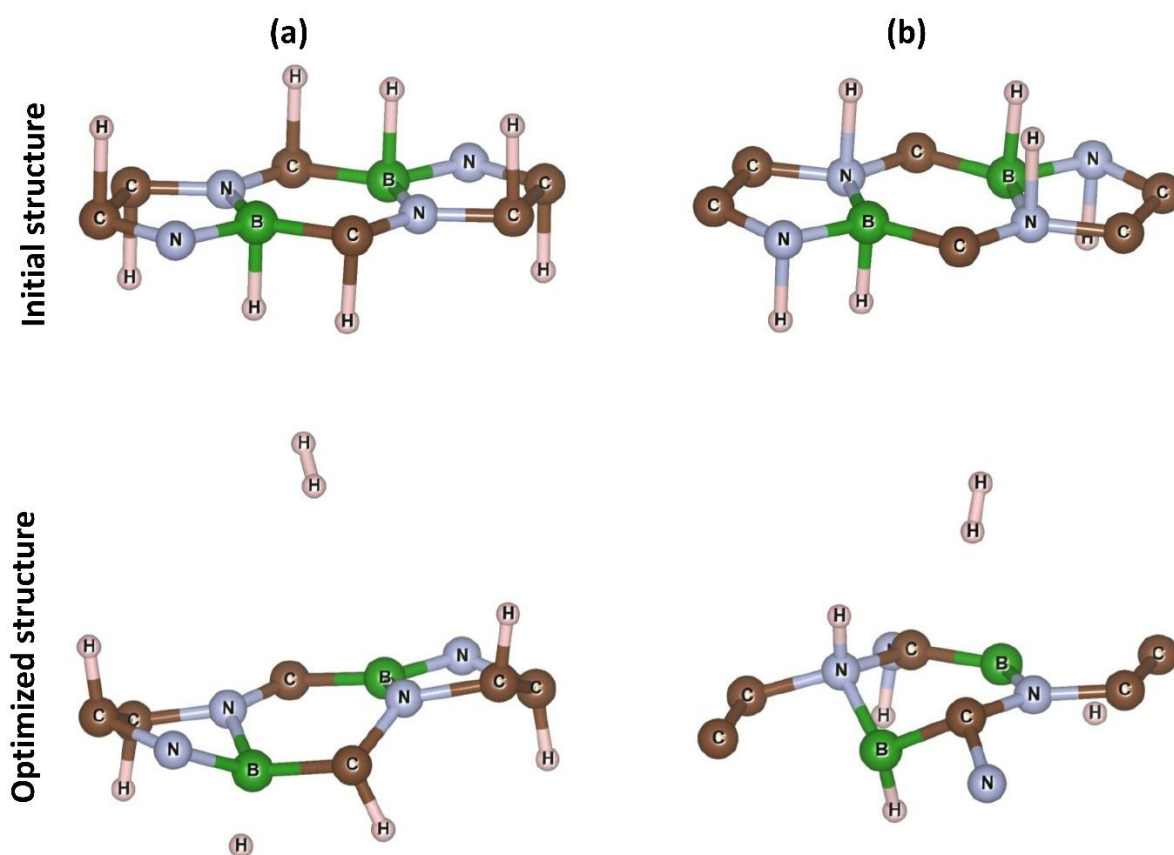


Figure S2

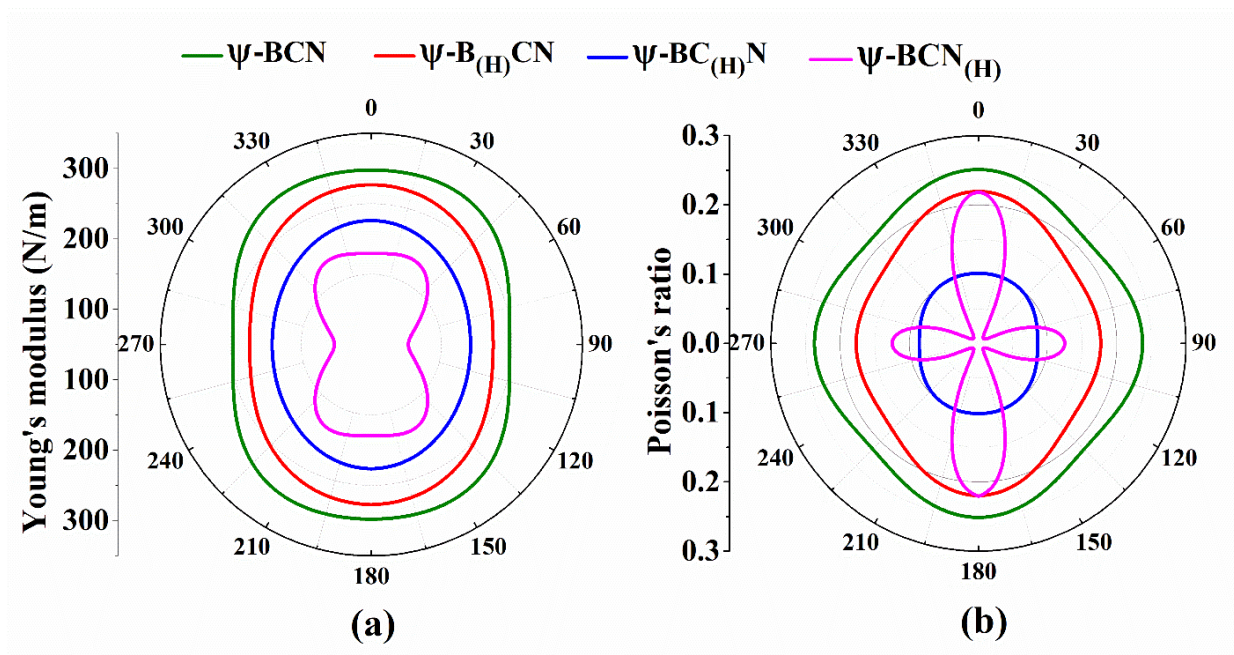


Figure S3

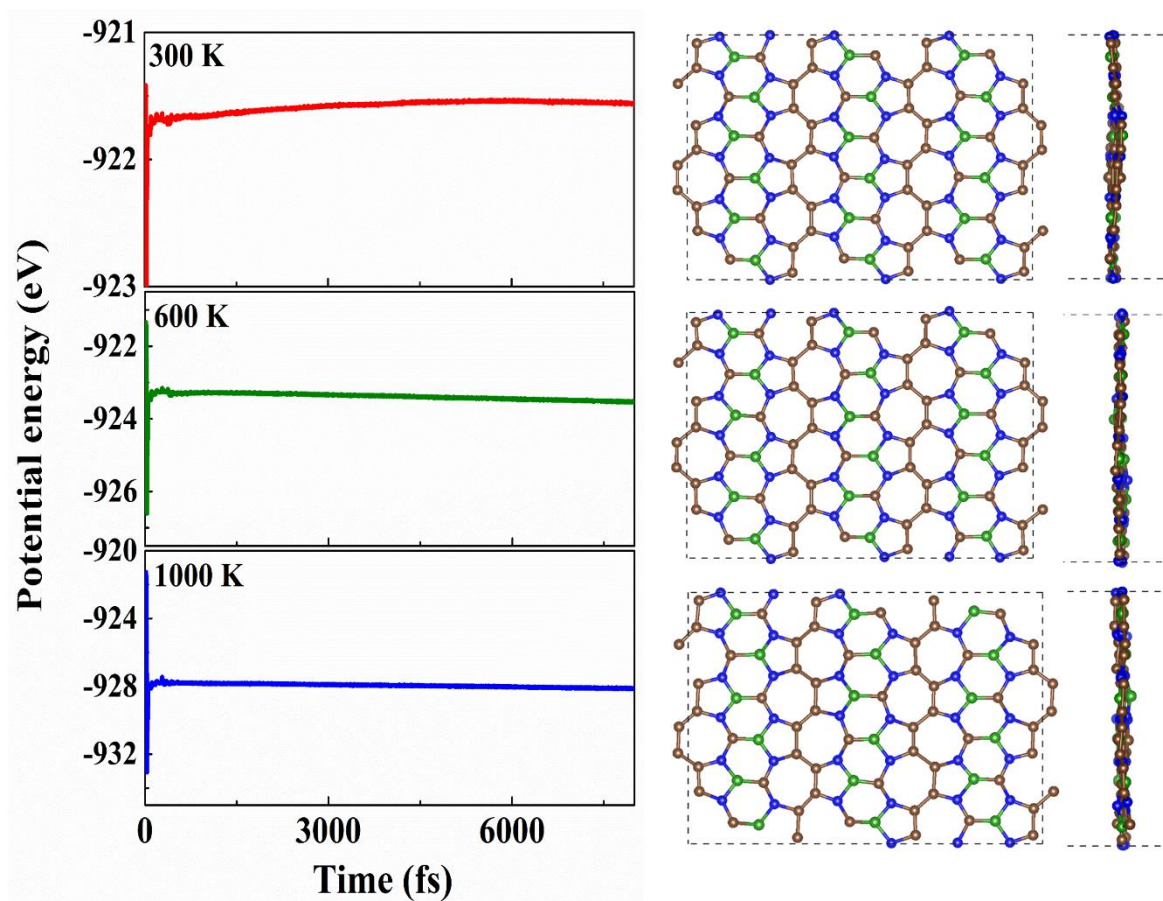


Figure S4

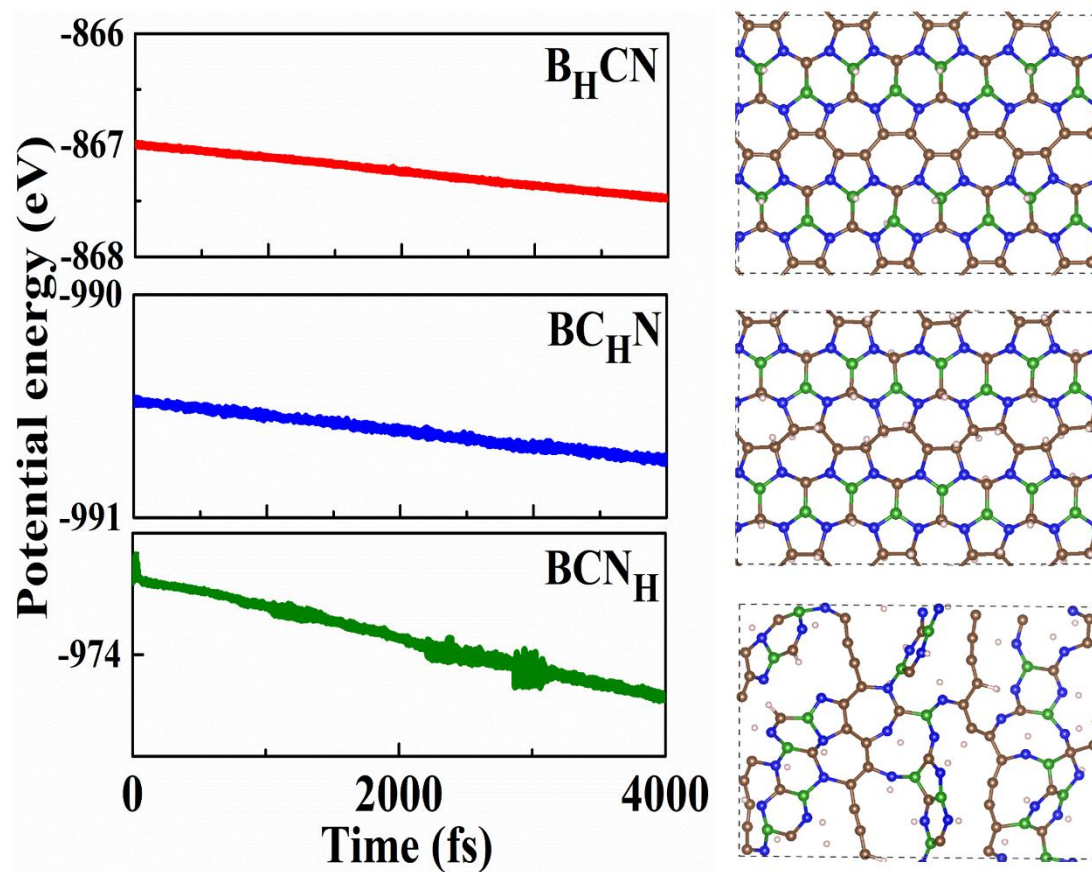


Figure S5

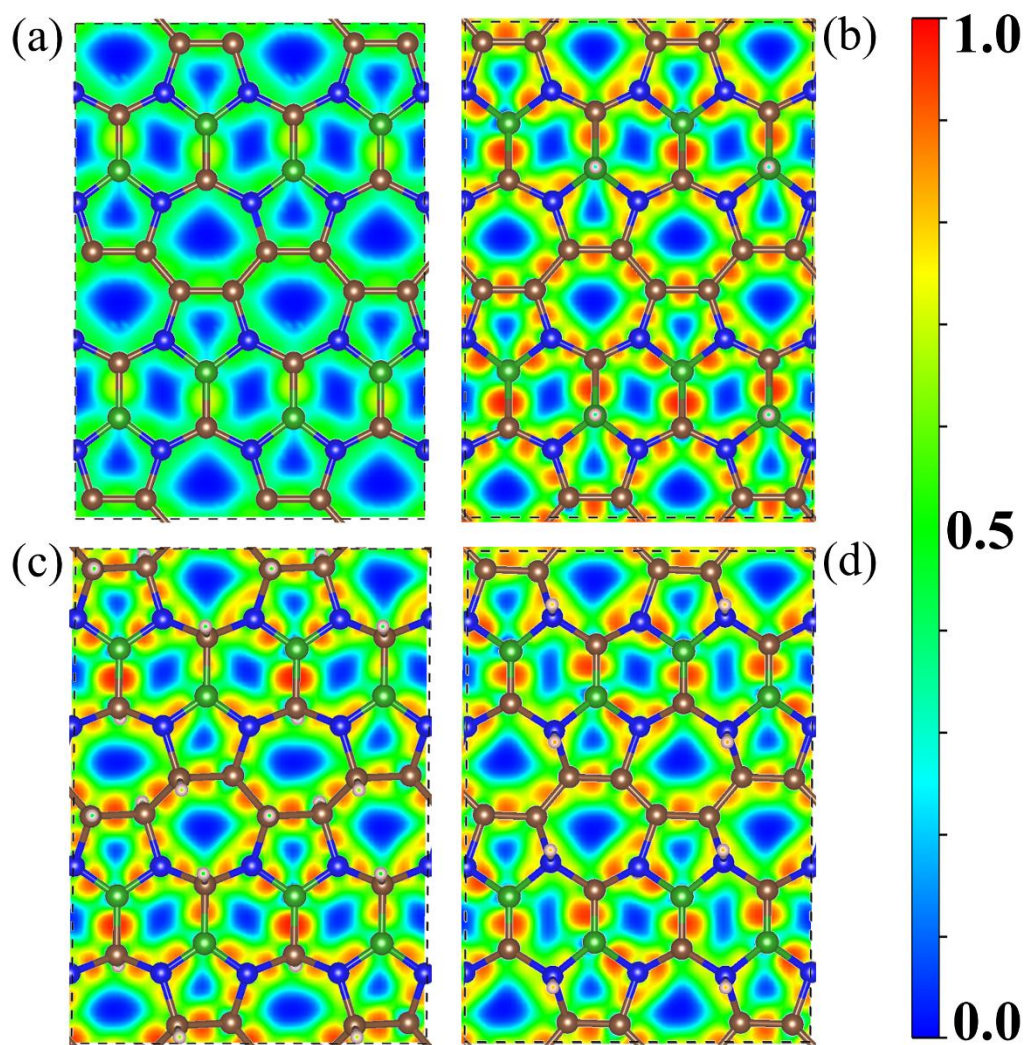


Figure S6