

Supplementary information

In-situ fabrication of N-doped iron-based Fenton-like catalyst (α -Fe₂O₃/N-C_{HCP}) with superior activity toward methylene blue degradation

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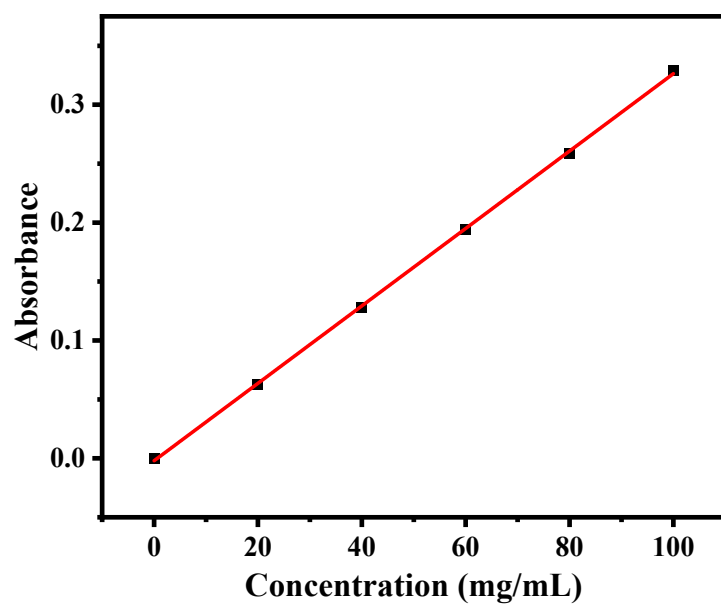


Figure S1. Calibration curve of methylene blue (MB) at 664 nm over a concentration range of 0-100 mg L⁻¹.

Table S1. Performance comparison of various Fenton-type catalysts for MB degradation.

Catalyst	Reaction Condition	TOC removal (%)	Stability	Reaction time (min)	Removal rate (%)	Ref.
α -Fe ₂ O ₃ /N-C _{HCP} -2	[MB] ₀ = 100 mg L ⁻¹ , [catalyst] = 0.40 g L ⁻¹ , [H ₂ O ₂] = 50 mM, pH = 7.3.	51.2	4 runs	15	100	This work
mFe-MPB	[MB] ₀ = 20 mg L ⁻¹ , [catalyst] = 0.1 g L ⁻¹ , [H ₂ O ₂] = 12 mM, pH = 7.	-	5 runs	30	100	1
Fe/C-700	[MB] ₀ = 10 mg L ⁻¹ , [catalyst] = 1.00 g L ⁻¹ , [H ₂ O ₂] = 7 mM, pH = 7.	58	5 runs	120	100	2
Fe ₃ O ₄ @NC-20	[MB] ₀ = 50 mg L ⁻¹ , [catalyst] = 0.75g L ⁻¹ , [H ₂ O ₂] = 239 mM, pH = 7.3.	59	--	50	100	3
Fe ₃ O ₄ /usGO	[MB] ₀ = 5 mg L ⁻¹ , [catalyst] = 2.92 g L ⁻¹ , [H ₂ O ₂] = 10 mM, pH = 3.	--	5 runs	120	100	4
Solid iron wastes (SIW)	[MB] ₀ = 10 mg L ⁻¹ , [catalyst] = 1.00 g L ⁻¹ , [PS] = 4 mM, pH = 7	72	5 runs	60	100	5
NC-CuFe-2	[MB] ₀ = 100 mg L ⁻¹ , [catalyst] = 0.50g L ⁻¹ , [H ₂ O ₂] = 0.3 M, pH = 7.		5 runs	180	98	6
C-Fe ₂ O ₃ -2	[MB] ₀ = 50 mg L ⁻¹ , [catalyst] = 100 mg, [H ₂ O ₂] = 0.3 M, pH = 7.3.	-	5 runs	420	96	7
MnO ₂ /Fe ₃ O ₄ /diatomite	[MB] ₀ = 10 mg L ⁻¹ , [catalyst] = 0.10g L ⁻¹ , [PMS] = 0.30 g/L, pH = 6.	57.2	5 runs	45	100	8

SUST/Fe-BTC	[MB] ₀ = 1.6 mmol/L, [catalyst] = 0.6 g L ⁻¹ , [H ₂ O ₂] = 7.4 mM, pH = 4.	55	3	60	100	9
Fe, N-CDs	[MB] ₀ = 20 mg L ⁻¹ , [catalyst] = 0.5 mg mL ⁻¹ , [H ₂ O ₂] = 147 mM, pH = 8.	-	-	60	97	10
Fe ₃ O ₄ @C	[MB] ₀ = 10 mg L ⁻¹ , [catalyst] = 0.5g L ⁻¹ , pH = 4.	67.5	-	80	97	11
Fe-GMCA	[MB] ₀ = 10 mg L ⁻¹ , [catalyst] = 0.02g L ⁻¹ , pH = 3, applied current, 15 Ma.	-	10 runs	60	99	12
Yolk-shell Fe ₃ O ₄ @MOF-5	[MB] ₀ = 50 mg L ⁻¹ , [catalyst] = 1.0 g L ⁻¹ , [H ₂ O ₂] = 30 mM, pH = 4.	60	5 runs	60	100	13
NH ₂ -MIL- 88B(Fe)	[MB] ₀ = 20 mg L ⁻¹ , [catalyst] = 0.2 g L ⁻¹ , [H ₂ O ₂] = 200 mM, pH = 5.6.	54.6	5 runs	45	100	14

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