

Electronic Supplementary Information (ESI)

Metal oxide hollow heterojunctions derived from bimetal-organic frameworks for efficient urea photosynthesis

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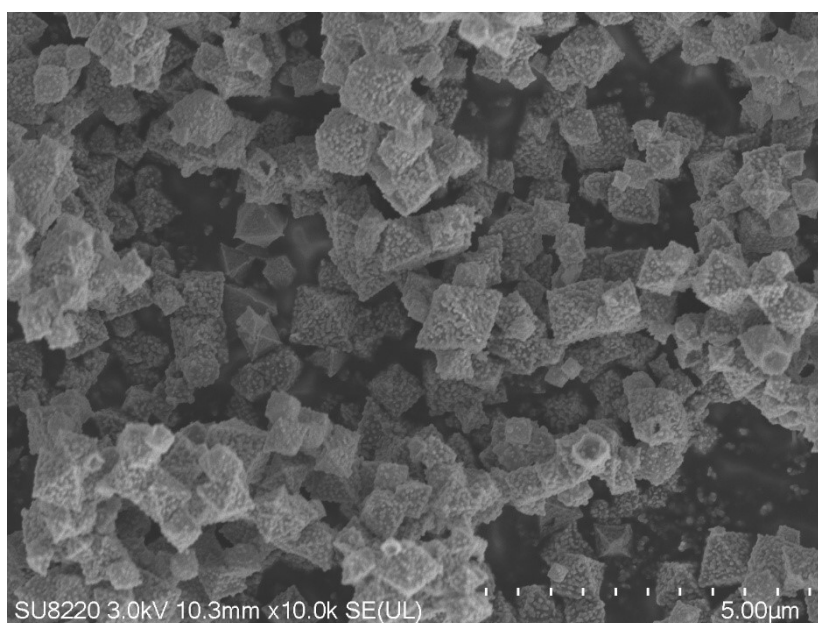
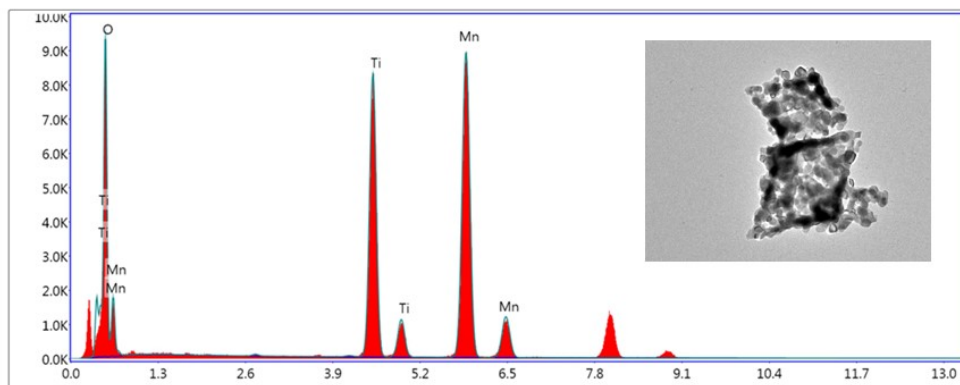


Fig. S1. SEM of TMO-7

eZAF Smart Quant Results

Element	Weight %	Atomic %	Net Int.	Error %	Kratio	Z	R	A	F
OK	89.28	96.39	769.5	5.09	0.50	1.01	0.98	0.55	1.00
TiK	5.23	1.89	1510.2	3.80	0.06	0.90	1.10	1.33	1.02
MnK	5.49	1.73	1845	2.53	0.06	0.90	1.12	1.21	1.00



Lsec: 66.0 0 Cnts 0.000 keV Det: Apollo XLT2 SUTW Det

Fig. S2. TEM and EDX of TMO-8

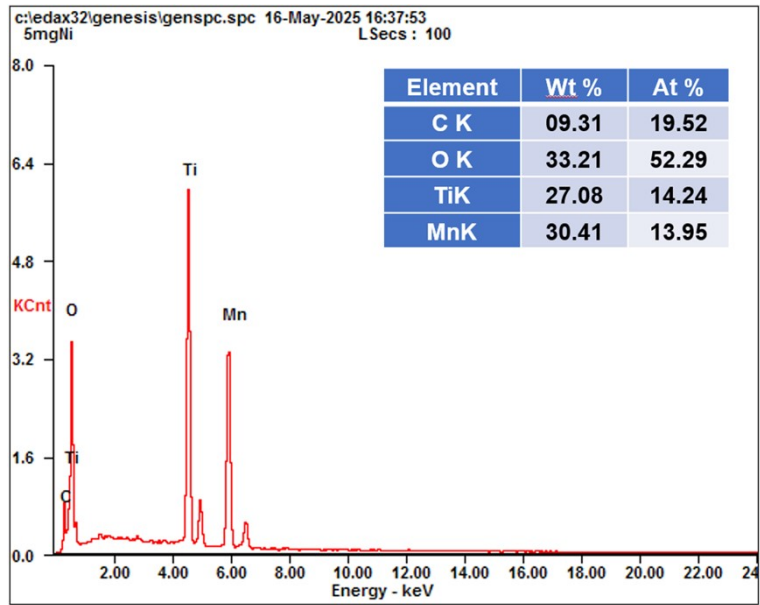


Fig. S3. EDX of TMO-7

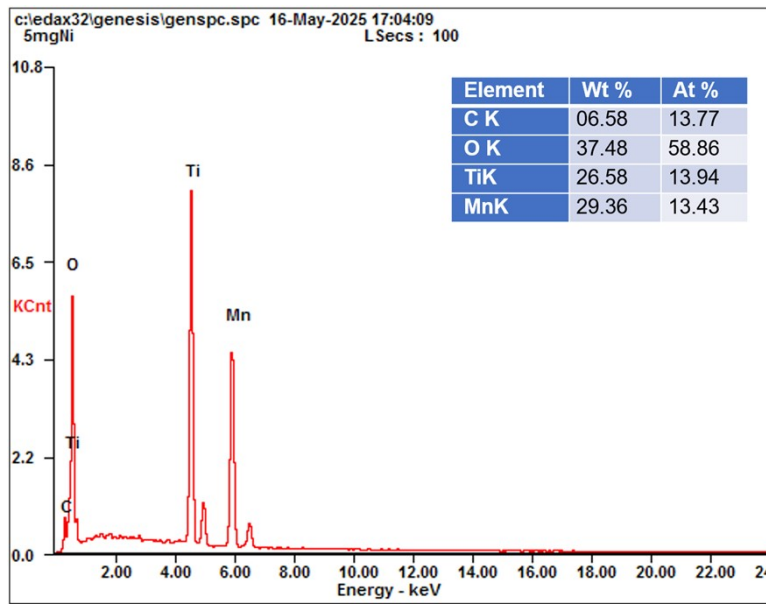


Fig. S4 EDX of TMO-8

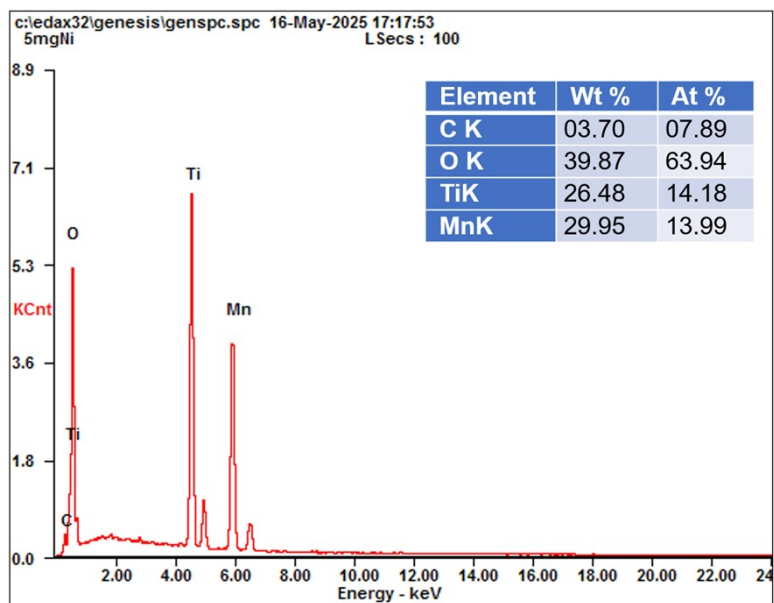


Fig. S5. EDX of TMO-9

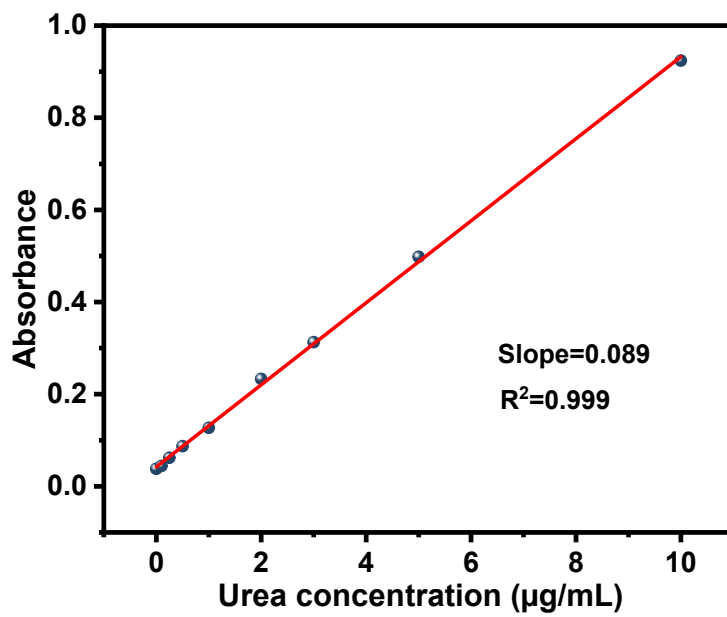


Fig.S6. standard curve for urea quantitative analysis.

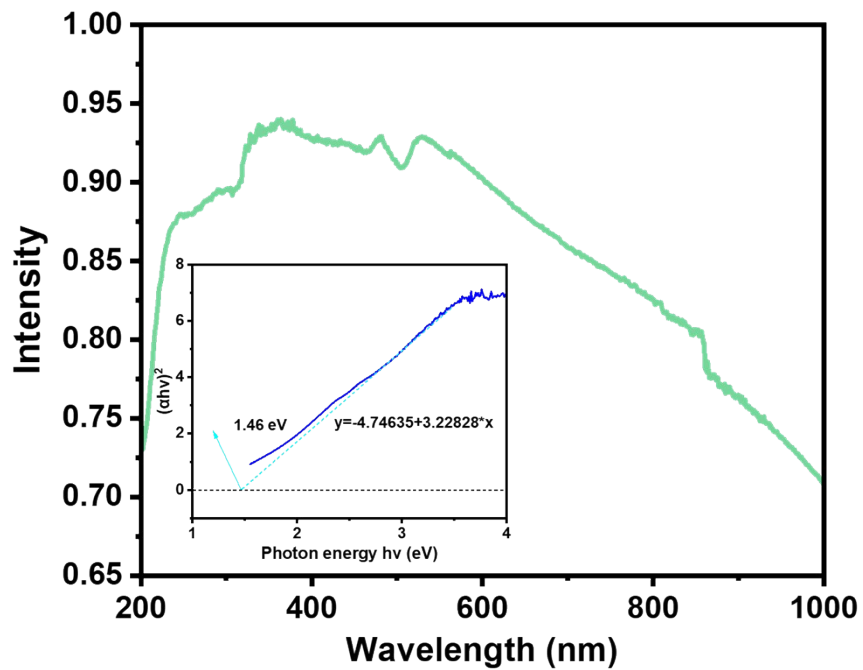


Fig. S7. The light absorbance and band gap.

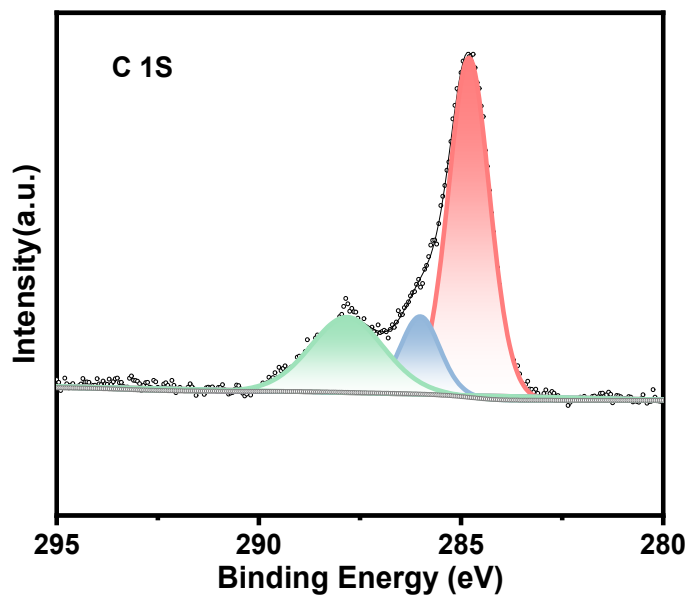


Fig. S8. C1S X-ray photoelectron spectroscopy for the TMO-8 after cycling tests.

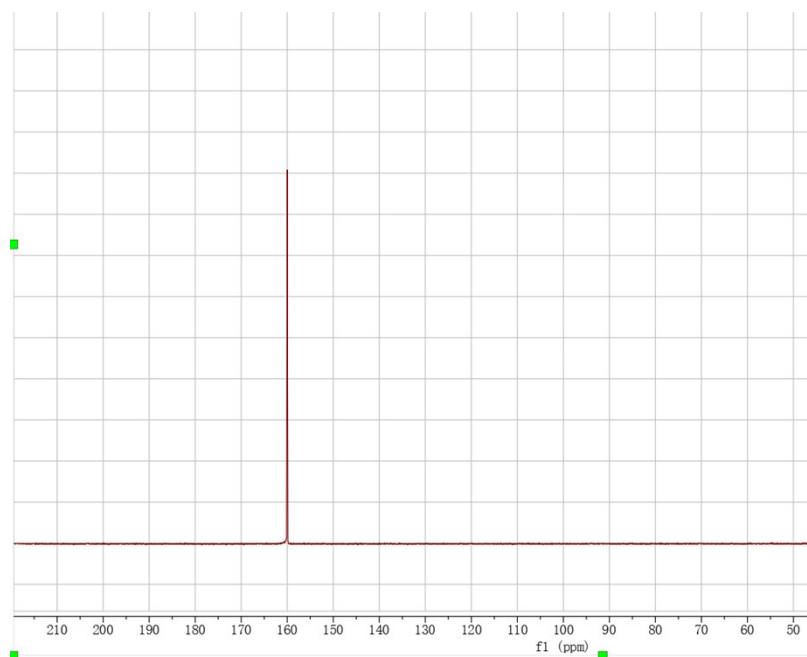


Fig. S9. ^{13}C urea standard sample.

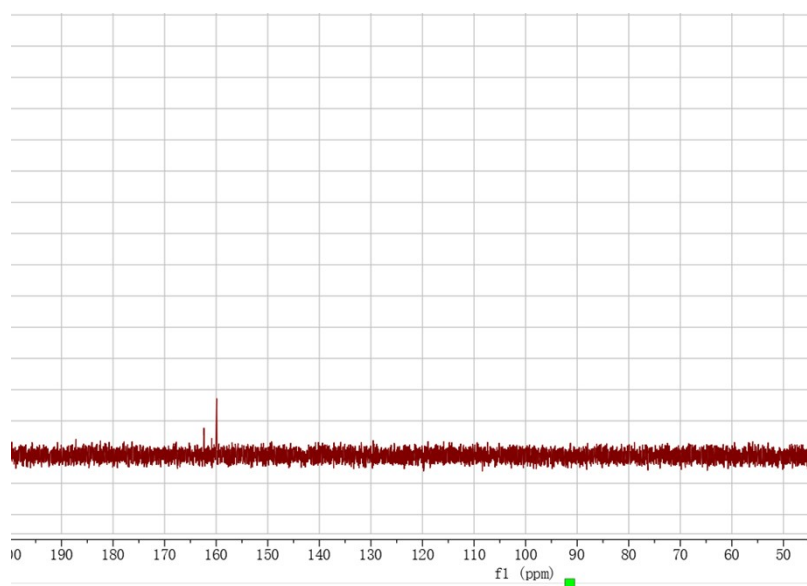


Fig. S10. ^{13}C urea measurement of the sample.

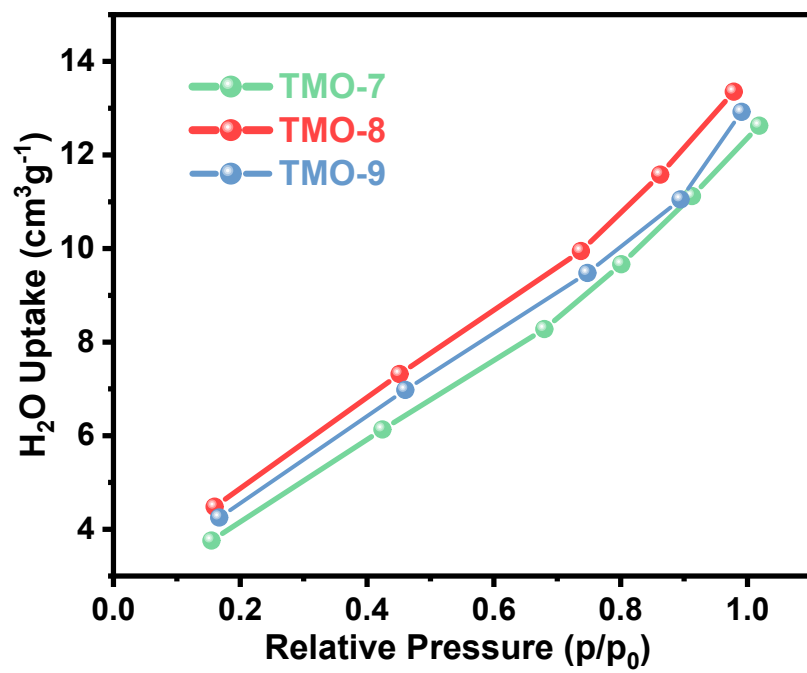


Fig. S11. H₂O gas adsorption–desorption isotherms at 298 K for TMO-x

Table S1. Comparison of photocatalytic urea synthesis performance over TMO-8 with recently reported representative photocatalysts.

Photocatalysts	Light source	Catalyst loading	Urea production	Selectivity	N source	References
TiO ₂ /Mn ₂ O ₃	300 W Xe lamp	0.17 mg/mL	93 μmol g ⁻¹ h ⁻¹	100%	NH ₃	This work
Ru-TiO ₂	420 nm monochromatic light	-	24.95 μmol h ⁻¹ g ⁻¹	-	N ₂	1
Cs ₂ CuBr ₄ /TiO _x -Ar	LED lamp (95.34 mW cm ⁻²) 365 nm	-	~0.17 μg h ⁻¹ L ⁻¹	88.45%	NO ₃ ⁻	2
Cu SA-TiO ₂	monochromatic light	-	432.12 μg g ⁻¹	-	N ₂	3
CeO ₂ -500	300 W Xe lamp	-	15.5 μg h ⁻¹	-	N ₂	4
SrTiO ₃ -FeS-CoWO ₄	300 W Xe lamp with 420 nm filter	80 mg/mL -400 mg/mL	8054.2 μg h ⁻¹ g ⁻¹	-	N ₂	5
Ni ₁ -CdS/WO ₃	300 W Xe lamp	-	78 μM h ⁻¹	-	N ₂	6
Pd-CeO ₂	300 W Xe lamp	-	9.2 μmol h ⁻¹ g ⁻¹	-	N ₂	7
SiW ₆ Mo ₆ @MIL-101(Cr)	300 W Xe lamp	-	1148 mg h ⁻¹ g ⁻¹	100%	N ₂	8
Ti ³⁺ -TiO ₂ /Fe-CNTs	300 W Hg lamp	-	710 μmol L ⁻¹ g ⁻¹	-	N ₂	9
NCP/ZIS	300 W Xe lamp	-	19.6 μmol g ⁻¹ h ⁻¹	-	N ₂	10

References

1. C. Shi, K. Xia, L. Zhang, M. Guo, X. Guan, C. Gu, X. Yang, Y. Wang, X. Liu and X. Ding, *Adv. Energy Mater.*, 2024, **14**, 2400201.
2. H. Sun, Z. Lin, R. Tang, Y. Liang, S. Zou, X. Zhang, K. Chen, R. Zheng and J. Huang, *Appl. Catal., B: Environ. Energy*, 2025, **360**, 124511.
3. D. Li, Y. Zhao, Y. Miao, C. Zhou, L.-P. Zhang, L.-Z. Wu and T. Zhang, *Adv. Mater.*, 2022, **34**, 2207793.
4. S. Yang, W. Zhang, G. Pan, J. Chen, J. Deng, K. Chen, X. Xie, D. Han, M. Dai and L. Niu, *Angew. Chem., Int. Ed.*, 2023, **62**, e202312076.
5. M. I. Ahmad, Y. Liu, Y. Wang, P. Cao, H. Yu, H. Li, S. Chen and X. Quan, *Angew. Chem., Int. Ed.*, 2025, **64**, e202419628.
6. Y. Zhang, Y. Sun, Q. Wang, Z. Zhuang, Z. Ma, L. Liu, G. Wang, D. Wang and X. Zheng, *Angew. Chem., Int. Ed.*, 2024, **63**, e202405637.
7. S. Yang, J. Deng, J. Chen, Q. Tan, T. Liu, K. Chen, D. Han, Y. Ma, M. Dai and L. Niu, *Catal. Sci. Technol.*, 2023, **13**, 1855–1865.
8. S. Su, X. Li, W. Ding, Y. Cao, S. Yuan, Z. Liu, Y. Yang, Y. Ding and M. Luo, *J. Mater. Chem. A*, 2024, **12**, 15300–15310.
9. H. Maimaiti, B. Xu, J.-y. Sun and L.-r. Feng, *ACS Sustainable Chem. Eng.*, 2021, **9**, 6991–7002.
10. H. Liang, X. Song, Y. Zhang, Y. Wu, B. Zhao, Y. Liu, Y. Jin, L. Sheng, M. Zhao, J. Liu and Z. Li, *J. Alloys Compd.*, 2023, **934**, 167884.