

Faraday Discussions

Supporting Information

Evolution of Size-Selected Pt Cluster Catalysts on Prototypical Oxide Supports

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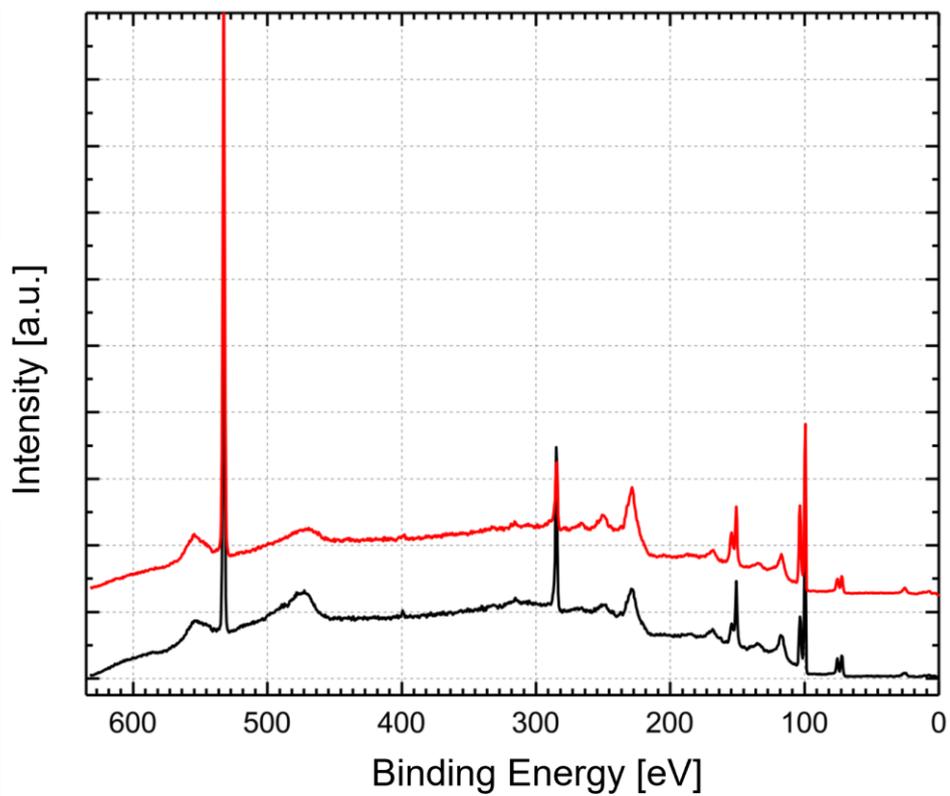
Survey scans of a representative SiO₂/Si sample

Figure S1. Survey scans of a representative Pt₂₀/SiO₂/n-Si sample as introduced into the NAP-XPS after transport (black) and after some oxidative treatment (red), recorded with a photon energy of 735 eV. For clarity, the scans are shown offset.

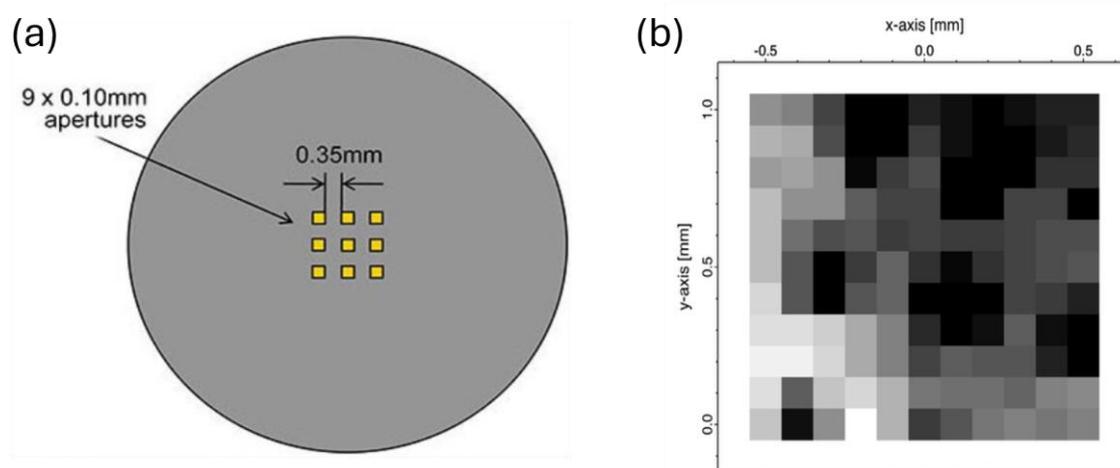
Pt distribution on suspended and Si-supported regions of a Si_3N_4 TEM grid

Figure S2. Pt distribution on a Si_3N_4 TEM grid. (a) Layout of the nine apertures in the PELCO silicon nitride TEM grid by TedPella, covered with an 8 nm thick Si_3N_4 film [Image from http://www.tedpella.com/grids_html/silicon-nitride-details.htm, 01-Feb-2017]. (b) The maximum signal of the Pt 4f peak was recorded at different points, moving in 0.1 mm steps in the $1 \times 1 \text{ mm}^2$ central region of the TEM grid, upon which a coverage of 1 atom/nm^2 of Pt_3 clusters was deposited. To rule out background-induced effects, several scans across the peaks in the Pt 4f region were also recorded (not shown), indicating that the background level was comparable throughout the measurement region. Darker colors in the 2D plot indicate lower Pt signal, lighter gray/white colors indicate higher intensity. Eight of the nine holes in a tilted square arrangement are clearly visible as dark regions, i.e. areas with a reduced amount of Pt. This Pt distribution results from charging of the suspended Si_3N_4 membrane areas during Pt_n^+ cluster deposition, leading to a deflection of the incoming positively charged clusters, while the charges can be neutralized effectively on those areas of the Si_3N_4 film supported by bulk Si, where the higher coverage is found. The HAADF-STEM images shown in the main manuscript were recorded in the regions of suspended nitride, while all XPS data is from supported areas.

Fe 3p XPS energy calibration

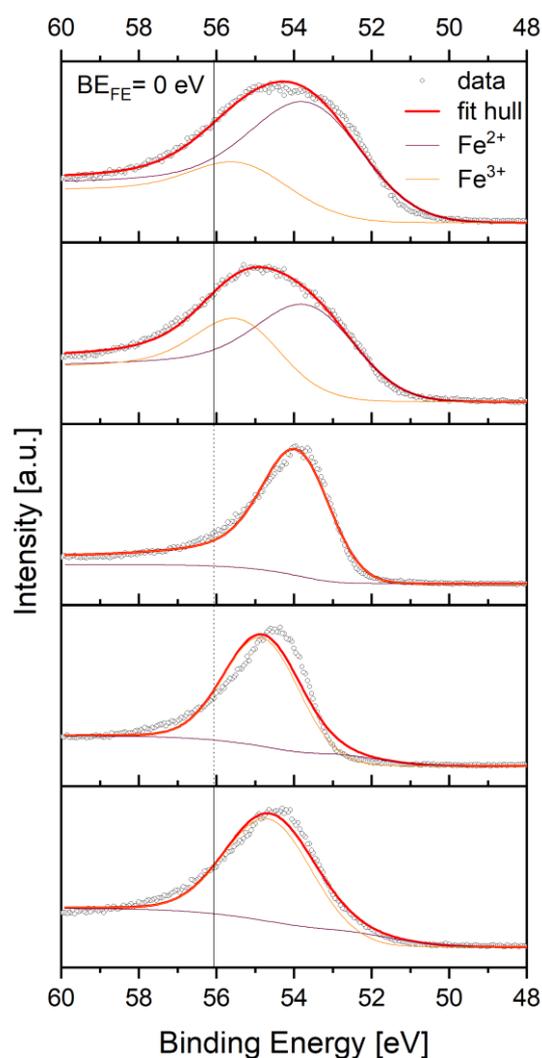


Figure S3. Uncalibrated Fe 3p spectra of Pt₁₀/Fe₃O₄(001) at 573 K during (top panel) and after 0.1 mbar H₂ (second panel), during (third panel) and after 0.1 mbar O₂ (fourth panel), and in 0.1 mbar H₂ after all treatments (bottom panel), corresponding to the measurements shown in Fig. 4e-f in the main manuscript. The spectra are fitted with two Doniach-Šunjić functions with an asymmetry parameter fixed at 0.15, and a distance between the two components fixed at 1.8 eV. The width of the Fe³⁺ feature is fixed at 0.84 times the width of the Fe²⁺ function, to best represent all spectra. In contrast to the Fermi edge, which disappears in oxidative conditions (see Fig. S4), the Fe³⁺ peak is always present and can thus be used for binding energy referencing. We shift all spectra such that the peak of the orange component is at 56.05 eV (vertical line, dashed where no Fermi edge could be measured), putting the Fermi edge position at 0.0 eV on average.

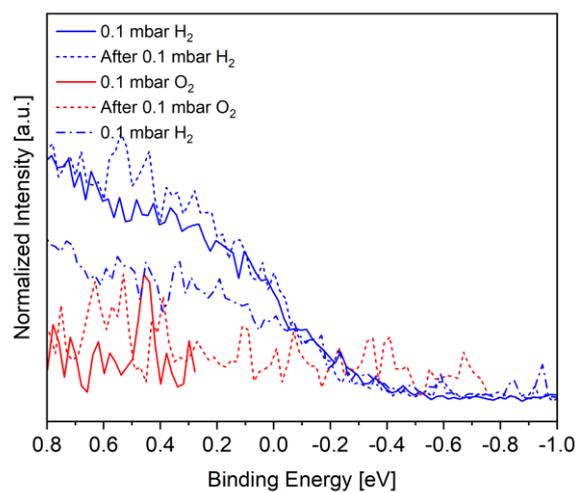
Fermi edge of Pt₁₀/Fe₃O₄(001) in redox conditions

Figure S4. Fe 3p calibrated Fermi edges of Pt₁₀/Fe₃O₄(001) at 573 K during and in between redox treatments in 0.1 mbar H₂ and 0.1 mbar O₂ at 573 K. Oxidative conditions eliminate the Fermi edge and the highest lying states are shifted to more positive binding energies.

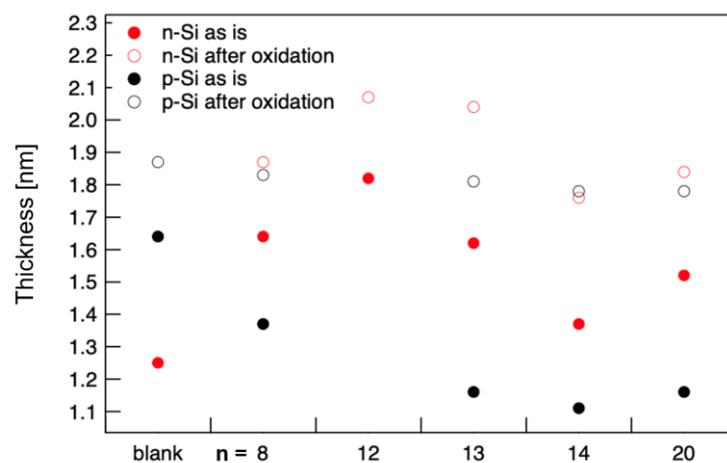
Thickness of the silica film on SiO₂/Si during oxidative treatment

Figure S5. Thickness of the SiO₂ layer on the examined samples before (filled circles) and after (open circles) the cleaning procedure, both on n-type (red) and p-type (black) Si substrates. The oxide thickness t_{ox} was calculated by $t_{\text{ox}} = \lambda_{\text{ox}} \ln [1 + (I_{\text{ox}} \lambda_{\text{M}} N_{\text{M}}) / (I_{\text{M}} \lambda_{\text{ox}} N_{\text{ox}})]$; with I_x , λ_x , and N_x being the intensity, inelastic mean free path, and atomic density of the metal (M) or oxide (ox), respectively.

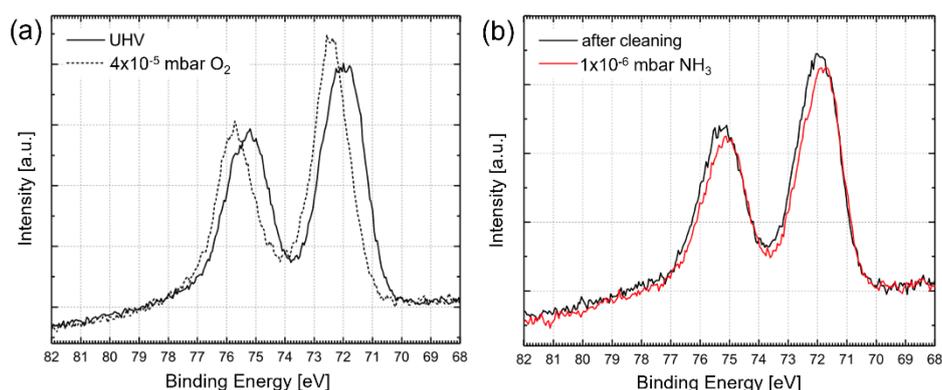
Response of Pt 4f of Pt_n/SiO₂/n-Si to various gas atmospheres

Figure S6. (a) Pt 4f spectra of Pt₄₀/SiO₂/n-Si in different gas atmospheres at RT: (a) as introduced into UHV (solid line) and in 4×10^{-5} mbar O₂ (dashed line), and (b) after cleaning (black) and during dosing of 1×10^{-6} mbar NH₃ (red), following previous cleaning in O₂, exposure to the reaction mixture at RT and heating to 423 K in NH₃.

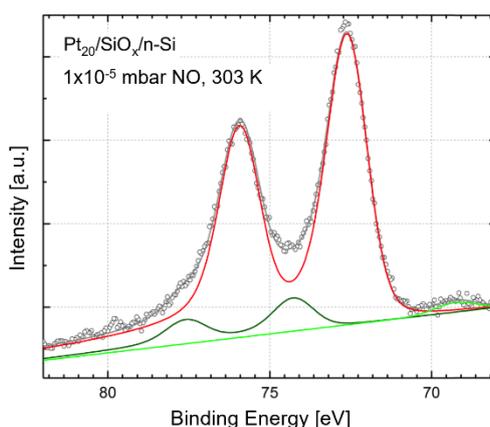


Figure S7. Pt 4f spectrum of Pt₂₀/SiO_x/n-Si in 1×10^{-5} mbar NO at 303 K on a sample that was previously exposed to reaction atmosphere and heated to 423 K in NO, including a fit with two components.

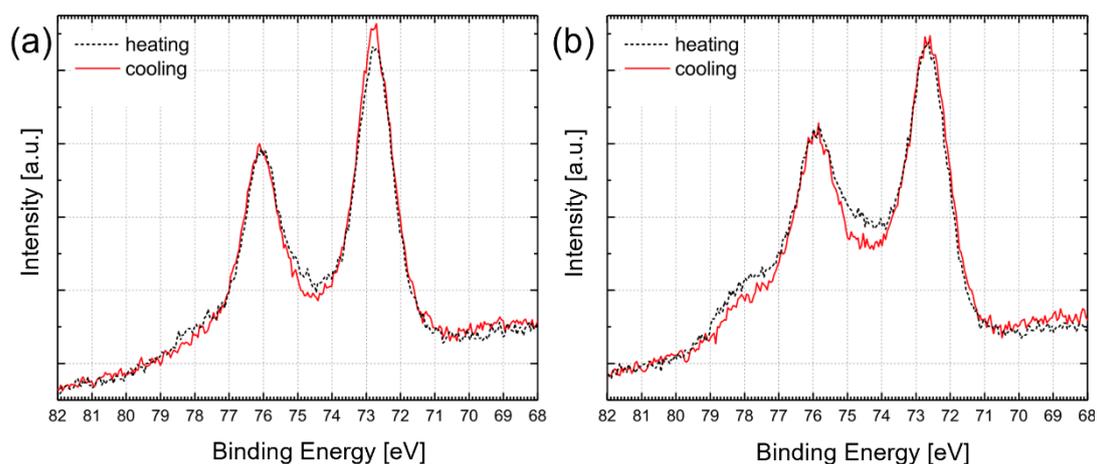


Figure S8. Comparison of the Pt 4f spectrum of (a) Pt₂₀/SiO₂/p-Si and (b) Pt₂₀/SiO₂/n-Si after cleaning and in ammonia oxidation conditions, i.e. 4×10^{-5} mbar O₂ and 1×10^{-5} mbar NH₃ at 303 K. The black dashed curves were recorded at 303 K during a heating ramp up to 423 K and the red curves at 303 K during the cooling ramp back down to RT. The oxidized species discussed in the main manuscript are clearly visible on the n-type sample and only form to a much smaller degree during the heating ramp on the p-type sample (not at all during the cool-down).