

Supplementary Information

In-Situ Carbon Recovery from Refractory Organics in Wastewater: A Critical

Review

Ning Li^a, Rui Wang^a, Jianhui Zhao^{b,*}, Xiaoming Peng^c, Wenchao Peng^d, Beibei Yan^a, Guanyi

Chen^{a,e,*}

^a National Engineering Laboratory of Petrochemical Industrial water Treatment, School of Environmental Science and Engineering, Tianjin University, Tianjin 300072, China

^b Tianjin Key Laboratory of Aquatic Science and Technology, School of Environmental and Municipal Engineering, Tianjin Chengjian University, Tianjin 300384, China

^c School of Civil Engineering and Architecture, East China Jiaotong University, Nanchang 330013, China

^d School of Chemical Engineering and Technology, Tianjin University, Tianjin 300050, China

^e School of Mechanical Engineering, Tianjin University of Commerce, Tianjin 300134, China

* Corresponding Authors:

zhaojianhui@tcu.edu.cn; chen@tju.edu.cn (G. Chen)

Table S1 Summary of representative systems for *in situ* carbon resource recovery of refractory organics

Treatment technology	Contaminant	Conditions	Degradation efficiency (%)	Product	Productivity	Selectivity (%)	Refs.
Photocatalysis	Carbamazepine (CBZ)	[CBZ]=0.02 g/L, [BiOBr/BiVO ₄]=0.30 g/L, [NaOH]=0.01 M, wavelength range 350 nm $\leq \lambda \leq$ 780 nm, light intensity 160 mW/cm ² , t=4 h, T=6°C	86.12	CO	2.56 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	100	1
	Methylene Blue (MB)	[MB]=0.008 g/L, [5%GQDs/V-TiO ₂]=1.00 g/L, wavelength range 320 nm $\leq \lambda \leq$ 780 nm, light intensity 160 mW/cm ² , t=8 h, T=6°C	99	CH ₃ OH C ₂ H ₅ OH CH ₄	13.24 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$ 5.65 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$ 0.445 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	68.5 29.2 2.30	2
	Rhodamine (RhB)	[RhB]=0.2 mM, [1%rGO/SrTi _{0.95} Fe _{0.05} O _{3-σ}]=1.00 g/L, wavelength range 320 nm $\leq \lambda \leq$ 780 nm, light intensity 160 mW/cm ² , t=6 h, T=6°C	90	CH ₃ OH C ₂ H ₅ OH	10.76 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$ 6.14 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	63.7 36.3	3
	RhB	[RhB]=0.01 g/L, [Ti ₃ C ₂ T _x /TiO ₂]=0.5 g/L, wavelength range 320 nm $\leq \lambda \leq$ 780 nm, light intensity 160 mW/cm ² , t=1 h	77.60	CO	10.02 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	/	4
	Phenol	[Phenol]=0.02 g/L, [Co/In ₂ O ₃ -550]=0.2 g/L, [[Ru(bpy) ₃]Cl ₂ ·6H ₂ O]=0.2 g/L, wavelength range 200 nm $\leq \lambda \leq$ 1200 nm, light intensity 100 mW/cm ² , t=8 h, T=25°C	30	CO	0.9 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	/	5
	RhB	[RhB]=0.01 g/L, [g-C ₃ N ₄ /Bi/BiVO ₄]=1.00 g/L, wavelength range 420 nm $\leq \lambda \leq$ 780 nm, light intensity 160 mW/cm ² , t=1 h	30	CO	0.63 $\mu\text{mol}\cdot\text{g}^{-1}\cdot\text{h}^{-1}$	/	6
Electrocatalysis -AOPs	4-nitrophenol (4-NP)	[4-NP]=0.01 g/L, [PMS]=0.5 g/L, anode=Co0.5, cathode=flower-like CuO, U=0.8 V, Na ₂ SO ₄ (anolyte) = 0.1 M, KHCO ₃ (catholyte) = 0.1 M, t=2 h, T=25°C	99	CH ₃ OH C ₂ H ₅ OH	49.15 $\mu\text{mol}\cdot\text{L}^{-1}\cdot\text{h}^{-1}$ 20.48 $\mu\text{mol}\cdot\text{L}^{-1}\cdot\text{h}^{-1}$	65.7 27.4	7
	Salicylic acid (SA)	[SA]=0.05 g/L, [PS]=0.02M, anode=MWCNT-CF, cathode=sFe-Cu@MOF, I = 0.025 A, [Na ₂ SO ₄ (anolyte)]=0.1 M, [KHCO ₃ (catholyte)]=0.1 M, t=3 h, T=25°C	99.1±0.9	HCOOH	8.00 $\mu\text{mol}\cdot\text{L}^{-1}\cdot\text{h}^{-1}$	/	8

Table S1 (continued)

	4-NP	[4-NP]=0.01 g/L, [PMS]=0.5 g/L, anode=Co0.5@C, cathode=SnO ₂ /CC, U=1.3 V, [Na ₂ SO ₄ (anolyte)] = 0.1 M, [KHCO ₃ (catholyte)] = 0.1 M, t=1 h, T=25°C	99.0	HCOOH	39.19 μmol·L ⁻¹ ·h ⁻¹	/	9
	Phenol	[Phenol]=0.01 g/L, [Na ₂ SO ₄]=0.1M, anode=Pd-TiO ₂ , cathode=Pd-Cu, wavelength range 200 nm≤λ≤1200 nm, light intensity 100 mW/cm ² , current density=10 mW/cm ² , t=8 h, T=25°C	87	CO	60.59 μmol·L ⁻¹ ·h ⁻¹	99.8	10
Photo-electrocatalysis	Methyl orange (MO)	[MO]=0.02 g/L, anode=TiO ₂ -NT, cathode=Ti ₃ C ₂ , pH(anode)=7.0, pH(cathode)=1.0, wavelength range λ≤254 nm, light intensity 3.60 mW/cm ² , t=5 h, T=25°C	74.8	CH ₄	5.2 μmol·g ⁻¹ ·h ⁻¹	77.6	11
	Phenol	[Phenol]=0.01 g/L, [KOH]=0.5M, anode=Pd QDs@TiO ₂ NTPC, cathode=Co ₃ O ₄ NWs, wavelength range 200 nm≤λ≤1200 nm, light intensity 100 mW/cm ² , U=3.0V, t=8 h, T=25°C	84.7	CO HCOOH	2.39 μmol·L ⁻¹ ·h ⁻¹ 43.42 μmol·L ⁻¹ ·h ⁻¹	5.2 94.8	12
Piezocatalysis-AOPs	Phenol	[Phenol]=0.01 g/L, [PMS]=3 g/L, [Co ₃ S ₄ /MoS ₂]=0.6 g/L, ultrasonic frequency=40 kHz, t=4 h, T=20°C	99.90	CO	17.33 μmol·g ⁻¹ ·h ⁻¹	74.5	13

Table S2 Cost assessment of each technology

Technology	Category	Item	Unit-price		Dosage		Total (USD)	Cost of category (USD)	Total TOC eliminated (mg)	Cost of category (USD/mg TOC)
			Value	Unit	Value	Unit				
Photocatalysis ¹	Electricity	Ultrasonic stirrer	0.07	USD/kW·h	0.0004	kW·h	0.00003	0.3503	0.8000	0.4378
		Drying oven	0.07	USD/kW·h	0.0032	kW·h	0.00022			
		Centrifugal machine	0.07	USD/kW·h	0.0002	kW·h	0.00001			
		Oil pump	0.07	USD/kW·h	0.4000	kW·h	0.02800			
		Cryogenic circulating pump	0.07	USD/kW·h	3.4000	kW·h	0.23800			
	Xe lamp	0.07	USD/kW·h	1.20000	kW·h	0.08400				
	Reagent	Bi(NO ₃) ₃ ·5H ₂ O	0.44	USD/g	0.0116	g	0.00508	0.6988	0.8735	
		Nitric acid	0.20	USD/mL	0.0600	ml	0.00985			
		NH ₄ VO ₃	0.50	USD/g	0.0014	g	0.00072			
		KBr	0.10	USD/g	0.0017	g	0.00018			
NaOH		1.71	USD/g	0.4000	g	0.68300				
Electricity	Drying oven	0.07	USD/kW·h	4.8400	kW·h	0.33880	0.4236	1.0277		
	Tube furnace	0.07	USD/kW·h	1.2000	kW·h	0.08400				
	Ultrasonic stirrer	0.07	USD/kW·h	0.0100	kW·h	0.00070				
	Electrochemical reactor	0.07	USD/kW·h	0.0016	kW·h	0.00013				
	Electrocatalysis-AOPs ⁷	Co(NO ₃) ₂ ·6H ₂ O	3.18	USD/g	0.0582	g			0.18510	
Reagent	Ethylene glycol	0.04	USD/mL	1.0000	mL	0.04380	2.9972	7.2713		
	Urea	0.04	USD/g	0.0500	g	0.00209				
	NH ₄ F	0.16	USD/g	0.0185	g	0.00295				

Table S2 (continued)

Technology	Category	Item	Unit-price		Dosage		Total (USD)	Cost of category (USD)	Total TOC eliminated (mg)	Cost of category (USD/mg TOC)
			Value	Unit	Value	Unit				
Photo-electrocatalysis ¹⁰	Electricity	Cetyltrimethylammonium bromide	0.43	USD/g	0.0050	g	0.00215	0.4278	0.6612	0.6470
		Conductive glass	0.27	USD/g	0.1650	g	0.04469			
		NaOH	1.71	USD/g	0.6400	g	1.09280			
		(NH ₄) ₂ S ₂ O ₈	0.07	USD/g	0.1096	g	0.00787			
		Na ₂ WO ₄ ·2H ₂ O	0.67	USD/g	0.0211	g	0.01412			
		Sodium dodecyl sulfate	1.88	USD/g	0.0461	g	0.08656			
		Copper mesh	1.60	USD/g	0.0669	g	0.10682			
		Acetone	0.06	USD/mL	10.0000	mL	0.55947			
		Ethanol	0.04	USD/mL	10.0000	mL	0.43943			
		Na ₂ SO ₄	0.43	USD/g	0.8520	g	0.36960			
		KHCO ₃	0.06	USD/g	0.6010	g	0.03934			
		PMS	0.24	USD/g	0.0060	g	0.00143			
		Ultrasonic stirrer	0.07	USD/kW·h	0.0107	kW·h	0.00075			
		Drying oven	0.07	USD/kW·h	2.5950	kW·h	0.18165			
		Tube furnace	0.07	USD/kW·h	0.3000	kW·h	0.02100			
		Electrochemical reactor	0.07	USD/kW·h	0.0040	kW·h	0.00028			
		Magnetic stirrers	0.07	USD/kW·h	0.4002	kW·h	0.02801			
	Oil pump	0.07	USD/kW·h	0.4000	kW·h	0.00280				
	Xe lamp	0.07	USD/kW·h	2.4000	kW·h	0.16800				
	Reagent	Ti	1.47	USD/g	0.5063	g	0.74630	8.3275		12.5946
Acetone		0.06	USD/mL	10.0000	mL	0.55947				

Table S2 (continued)

Technology	Category	Item	Unit-price	Dosage	Total	Cost of	Total TOC	Cost of category
------------	----------	------	------------	--------	-------	---------	-----------	------------------

		Value	Unit	Value	Unit	(USD)	category (USD)	eliminated (mg)	(USD/mg TOC)
		0.04	USD/ml	10.0000	mL	0.43943			
		0.04	USD/ml	10.000	mL	0.43800			
		0.16	USD/g	0.0333	g	0.00531			
		1.19	USD/g	0.0800	g	0.09512			
		739.90	USD/g	0.0009	g	0.67375			
		0.58	USD/g	0.0300	g	0.01727			
		1.60	USD/g	0.0669	g	0.10682			
		0.39	USD/mL	6.0000	mL	2.33958			
		0.38	USD/mL	6.0000	mL	2.29023			
		0.43	USD/g	1.4205	g	0.61621			
		0.07	USD/kW·h	0.0002	kW·h	0.00001			
		0.07	USD/kW·h	0.8060	kW·h	0.05642			
	Electricity	0.07	USD/kW·h	0.0022	kW·h	0.00015	0.4346		1.5188
		0.07	USD/kW·h	0.4000	kW·h	0.02800			
		0.07	USD/kW·h	1.6000	kW·h	0.11200			
		0.07	USD/kW·h	3.4000	kW·h	0.23800		0.2861	
Piezocatalysis- AOPs ¹³		0.40	USD/g	0.0120	g	0.00475			
		2.36	USD/g	0.0085	g	0.02003			
	Reagent	3.18	USD/g	0.0040	g	0.01272	0.0746		0.2608
		0.57	USD/mL	0.0023	mL	0.00129			
		0.24	USD/g	0.1500	g	0.03585			

Table S3 Cost of each technology for *in situ* utilization of carbon resource

Technologies	Cost_{EE} (\$/mg TOC)	Cost_{Reagent} (\$/mg TOC)	Cost_{Total} (\$/mg TOC)
Photocatalysis	0.44	0.87	1.31
Electrocatalysis- AOPs	1.03	7.27	8.30
Photo- electrocatalysis	0.65	12.59	13.24
Piezocatalysis- AOPs	1.52	0.26	1.78

Table S4. Life cycle inventory for each technology normalized to the functional unit.

Technology	Stage	Category	Item	Value	Unit	
Photocatalysis ¹	Catalyst preparation	Input	Bi(NO ₃) ₃ ·5H ₂ O	0.0145	g	
			Nitric acid	0.0048	g	
			Ultrapure water	1.5759	g	
			NH ₄ VO ₃	0.0018	g	
			KBr	0.0021	g	
			Ultrasonic stirrer	0.0005	kw·h	
			Drying oven	0.0040	kw·h	
			Centrifugal machine	0.0003	kw·h	
			Waste liquid discharge	1.5898	g	
			CBZ	0.0025	g	
	Catalytic reaction	Input	Output	Ultrapure water	125.0000	g
				NaOH	0.5000	g
				Ar	133.8000	g
				Xe lamp	1.5000	kw·h
				Cryogenic circulating	4.2500	kw·h
				Oil pump	0.5000	kw·h
				Waste liquid discharge	125.4969	g
				Emission to air:Ar	133.8000	g
				Co(NO ₃) ₂ ·6H ₂ O	0.1412	g
				Ethylene glycol	2.7001	g
Electrocatalysis-AOPs ⁷	Catalyst preparation	Input	Urea	0.1213	g	
			NH ₄ F	0.0449	g	
			Cetyltrimethylammonium bromide	0.0121	g	
			Conductive glass	0.4003	g	
			NaOH	1.5526	g	
			(NH ₄) ₂ S ₂ O ₈	0.2659	g	
			Na ₂ WO ₄ ·2H ₂ O	0.0512	g	
			Sodium dodecyl sulfate	0.1118	g	
			Copper mesh	0.1623	g	
			Acetone	19.1897	g	

Table S4 (*continued*)

Technology	Stage	Category	Item	Value	Unit	
Photo-electrocatalysis ¹⁰	Catalytic reaction	Output	Ethanol	19.1412	g	
			Ultrapure water	184.8617	g	
			Waste liquid discharge	228.5143	g	
			4-NP	0.0015	g	
			Na ₂ SO ₄	2.0670	g	
		Input	KHCO ₃	1.4580	g	
			PMS	0.0146	g	
			Electrochemical reactor	0.0039	kw·h	
			Waste liquid discharge	294.6616	g	
			Ti	0.7657	g	
		Catalyst preparation	Output	Acetone	11.9631	g
				Ethanol	11.9328	g
				Ethylene glycol	16.4398	g
				NH ₄ F	0.0504	g
				Polyvinylpyrrolidone	0.1210	g
	Input		PdCl ₂	0.0014	g	
			NaI	0.0454	g	
			Copper mesh	0.1012	g	
			HCl	0.5522	g	
			H ₃ PO ₄	13.2184	g	
	Catalytic reaction		Output	Ultrapure water	201.1494	g
				Ultrasonic stirrer	0.0162	kw·h
				Drying oven	3.9247	kw·h
				Tube furnace	0.4537	kw·h
				Electrochemical reactor	0.0042	kw·h
		Magnetic stirrers		0.0003	kw·h	
		Waste liquid discharge		266.9239	g	
		Na ₂ SO ₄		2.1484	g	
		Phenol		0.0015	g	
		N ₂		3.3727	g	
Catalytic reaction	Input	Ultrapure water	151.2402	g		
		Oil pump	0.6050	kw·h		
			Electrochemical reactor	0.0018	kw·h	

Table S4 (continued)

Technology	Stage	Category	Item	Value	Unit	
Piezocatalysis- AOPs ¹³	Catalyst preparation	Output	Xe lamp	3.6298	kw·h	
			Magnetic stirrers	0.6050	kw·h	
			Waste liquid discharge	153.3878	g	
			Emission to air: CO	0.0023	g	
			Emission to air: N ₂	3.3727	g	
			H ₂₄ Mo ₇ N ₆ O ₂₄ ·4H ₂ O	0.0419	g	
			C ₂ H ₅ NS	0.0297	g	
		Input	Co(NO ₃) ₂ ·6H ₂ O	0.0140	g	
			H ₄ N ₂ ·H ₂ O	0.0783	g	
			Ultrapure water	25.8235	g	
			Magnetic stirrers	0.0007	kw·h	
			Drying oven	2.8172	kw·h	
			Water bath stirrer	0.0077	kw·h	
			Waste liquid discharge	25.9567	g	
	Catalytic reaction	Output	Emission to air: NH ₃	0.0021	g	
			Emission to air: SO ₂	0.0024	g	
			PMS	0.5243	g	
			Ar	374.1349	g	
			Phenol	0.0017	g	
			Ultrapure water	174.7641	g	
		Input	Oil pump	1.3981	kw·h	
			Ultrasonic generator	5.5925	kw·h	
			Cryogenic circulating pump	11.8840	kw·h	
			Output	Waste liquid discharge	175.2898	g
				Emission to air:Ar	374.1349	g

Table S5. Results for the different systems/stages per functional unit (CML-IA baseline (World 2000))

Impact category	Unit	Photocatalysis			Electrocatalysis			Photo-electrocatalysis			Piezocatalysis		
		Total	Preparation	Reaction	Total	Preparation	Reaction	Total	Preparation	Reaction	Total	Preparation	Reaction
Abiotic depletion	kg Sb eq	3.51×10^{-6}	1.26×10^{-7}	3.39×10^{-6}	2.33×10^{-5}	2.33×10^{-5}	2.22×10^{-9}	5.79×10^{-6}	3.22×10^{-6}	2.57×10^{-6}	1.06×10^{-5}	1.64×10^{-6}	8.99×10^{-6}
Abiotic depletion (fossil fuels)	MJ	5.57×10^1	3.29	5.24×10^1	1.32×10^2	1.32×10^2	3.43×10^{-2}	7.85×10^1	3.88×10^1	3.97×10^1	1.46×10^2	2.56×10^1	1.21×10^2
Global warming (GWP 100a)	kg CO ₂ eq	6.22	3.01×10^{-1}	5.92	1.48×10^1	1.48×10^1	3.88×10^{-3}	8.71	4.22	4.49	1.34×10^1	2.35	1.11×10^1
Ozone layer depletion (ODP)	kg CFC-11 eq	6.35×10^{-8}	1.52×10^{-8}	4.83×10^{-8}	1.28×10^{-7}	1.28×10^{-7}	3.16E-11	7.90×10^{-8}	4.24×10^{-8}	3.66×10^{-8}	6.38×10^{-7}	1.14×10^{-7}	5.24×10^{-7}
Human toxicity	kg 1,4-DB eq	2.21	1.14×10^{-1}	2.10	5.32	5.32	1.37×10^{-3}	3.14	1.55	1.59	5.45	9.37×10^{-1}	4.51
Fresh water aquatic ecotoxicity	kg 1,4-DB eq	2.47	1.23×10^{-1}	2.35	5.88	5.87	1.54×10^{-3}	3.47	1.69	1.78	7.81	1.26	6.55
Marine aquatic ecotoxicity	kg 1,4-DB eq	1.00×10^4	4.68×10^2	9.55×10^3	2.38×10^4	2.38×10^4	6.25	1.40×10^4	6.76×10^3	7.24×10^3	2.19×10^4	3.79×10^3	1.81×10^4
Terrestrial ecotoxicity	kg 1,4-DB eq	2.27×10^{-2}	6.03×10^{-4}	2.21×10^{-2}	5.52×10^{-2}	5.52×10^{-2}	1.45×10^{-5}	3.23×10^{-2}	1.56×10^{-2}	1.68×10^{-2}	6.33×10^{-2}	9.39×10^{-3}	5.39×10^{-2}
Photochemical oxidation	kg C ₂ H ₄ eq	1.04×10^{-3}	5.17×10^{-5}	9.89×10^{-4}	2.52×10^{-3}	2.52×10^{-3}	6.47×10^{-7}	1.51×10^{-3}	7.61×10^{-4}	7.50×10^{-4}	2.34×10^{-3}	4.07×10^{-4}	1.93×10^{-3}
Acidification	kg SO ₂ eq	2.78×10^{-2}	1.31×10^{-3}	2.64×10^{-2}	6.62×10^{-2}	6.62×10^{-2}	1.73×10^{-5}	3.91×10^{-2}	1.91×10^{-2}	2.00×10^{-2}	5.91×10^{-2}	1.03×10^{-2}	4.88×10^{-2}
Eutrophication	kg PO ₄ ³⁻ eq	6.69×10^{-3}	6.35×10^{-4}	6.05×10^{-3}	1.53×10^{-2}	1.53×10^{-2}	3.96×10^{-6}	9.10×10^{-3}	4.52×10^{-3}	4.59×10^{-3}	2.83×10^{-2}	4.94×10^{-3}	2.34×10^{-2}

References:

1. L.-L. Zheng, L. Tian, L.-S. Zhang, J. Yu, Y. Chen, Q. Fu, X.-Z. Liu, D.-S. Wu and J.-P. Zou, *ACS Catalysis*, 2024, **14**, 2134-2143.
2. J.-P. Zou, D.-D. Wu, J. Luo, Q.-J. Xing, X.-B. Luo, W.-H. Dong, S.-L. Luo, H.-M. Du and S. L. Suib, *ACS Catalysis*, 2016, **6**, 6861-6867.
3. W.-H. Dong, D.-D. Wu, J.-M. Luo, Q.-J. Xing, H. Liu, J.-P. Zou, X.-B. Luo, X.-B. Min, H.-L. Liu, S.-L. Luo and C.-T. Au, *Journal of Catalysis*, 2017, **349**, 218-225.
4. X. Wu, J. Li, R. Zhong, S. Liu, F. Huang, Q. Yang, B. Zhang, X. Wang and F. Zhang, *Advanced Functional Materials*, 2024, **34**, 2405868.
5. Q. Liang, S. Zhao, Z. Li, Z. Wu, H. Shi, H. Huang and Z. Kang, *ACS Applied Materials & Interfaces*, 2021, **13**, 40754-40765.
6. Q. Xie, W. He, S. Liu, C. Li, J. Zhang and P. K. Wong, *Chinese Journal of Catalysis*, 2020, **41**, 140-153.
7. J.-P. Zou, Y. Chen, S.-S. Liu, Q.-J. Xing, W.-H. Dong, X.-B. Luo, W.-L. Dai, X. Xiao, J.-M. Luo and J. Crittenden, *Water Research*, 2019, **150**, 330-339.
8. M. Priyadarshini, A. Ahmad, S. Yadav and M. M. Ghangrekar, *Chemical Engineering Journal*, 2023, **478**, 147343.
9. M. Zhu, L. Zhang, S. Liu, D. Wang, Y. Qin, Y. Chen, W. Dai, Y. Wang, Q. Xing and J. Zou, *Chinese Chemical Letters*, 2020, **31**, 1961-1965.
10. R. Cui, Q. Shen, C. Guo, B. Tang, N. Yang and G. Zhao, *Applied Catalysis B: Environmental*, 2020, **261**, 118253.

11. J. Zhang, S. Lv, J. Zheng, P. Yang, J. Li, X. Wu, T. Jin, C. Cheng, Y. Song and L. Li, *ACS Sustainable Chemistry & Engineering*, 2020, **8**, 11133-11140.
12. Q. Shen, S. Xu, R. Cui, C. Guo and G. Zhao, *Chemical Engineering Journal*, 2023, **471**, 144422.
13. M. Ran, H. Xu, Y. Bao, Y. Zhang, J. Zhang and M. Xing, *Angewandte Chemie International Edition*, 2023, **62**, e202303728.