

Supplementary Information

**Ce-Accelerated Spontaneous Redox Synthesis
of NiFe-based Electrocatalysts for Efficient
Oxygen Evolution Reaction**

In-Kyoung Ahn*¹

¹Low Carbon Energy Group, Korea Institute of Industrial Technology (KITECH), Ulsan
44413, Republic of Korea

*Correspondence to ikahn@kitech.re.kr (I.-K. Ahn).

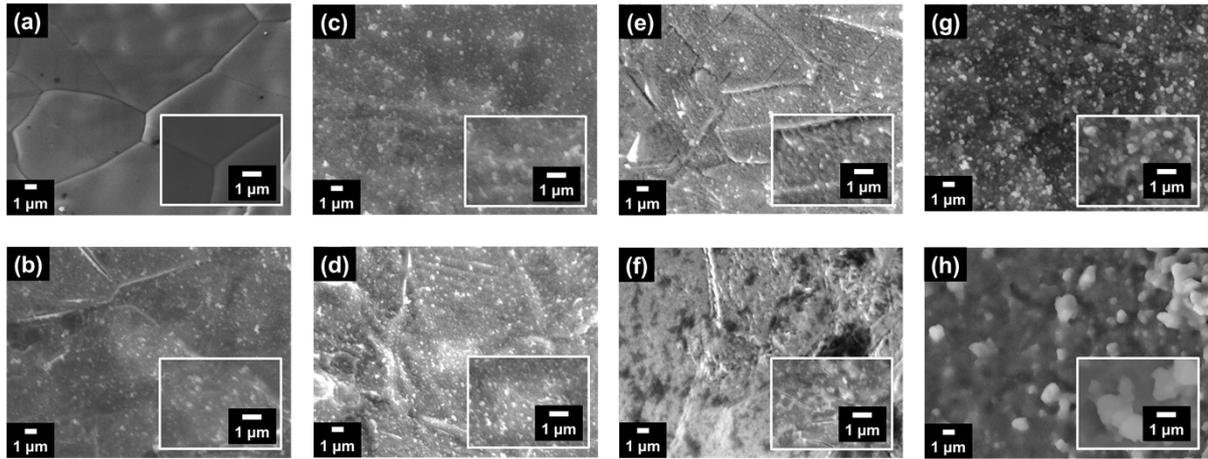


Fig.S1 Low-magnification SEM images with high-magnification insets of (a) nickel foam, (b) Fe-only/NF, (c) FeCe5/NF, (d) FeCe10/NF, (e) FeCe20/NF, (f) FeCe40/NF, (g) Ce-only40/NF, and (h) Ce-only100/NF.

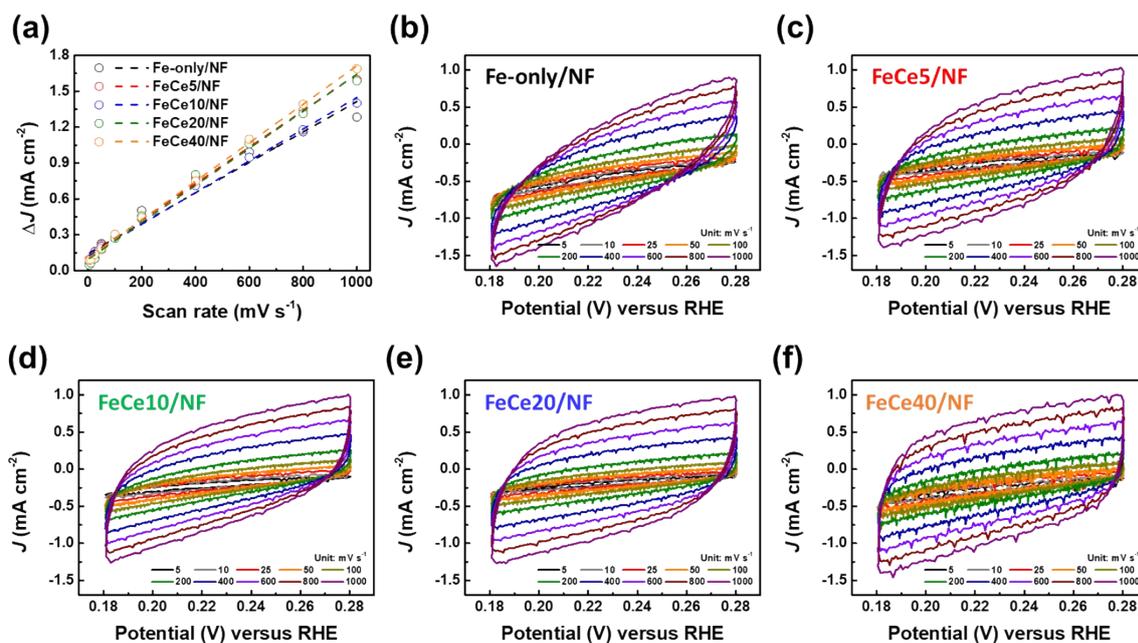


Fig.S2 ECSA analysis of synthesized catalysts to evaluate morphological characteristics. (a) Double-layer capacitance (C_{dl}) measurements derived from linear fits of charging current density vs. scan rate for various NiFe-based electrocatalysts. (b-f) Cyclic voltammetry (CV) curves recorded in the non-Faradaic region for (b) Fe-only/NF, (c) FeCe5/NF, (d) FeCe10/NF, (e) FeCe20/NF, and (f) FeCe40/NF.

The ECSA was estimated by measuring the double-layer capacitance (C_{dl}) in the non-Faradaic region. CV was performed at various scan rates within the potential range of 0.18 to 0.28 V (vs. RHE). The ΔJ ($J_a - J_c$) at 0.23 V (vs. RHE) was plotted against the scan rate (v), where the slope of the resulting linear fit corresponds to twice the C_{dl} ^{S1-S3}. The C_{dl} was calculated using the following equation:

$$C_{dl} = 1/2 \times d(\Delta J)/dv$$

The ECSA was subsequently determined by normalizing the C_{dl} with the specific capacitance (C_s) of a smooth planar surface^{S4-S6}:

$$\text{ECSA} = C_{dl}/C_s$$

According to literature, the C_s value for transition metal-based catalysts in 1.0 M KOH electrolyte typically ranges from 20 to 60 $\mu\text{F cm}^{-2}$. For the purposes of this calculation, a C_s value of 40 $\mu\text{F cm}^{-2}$ was adopted to evaluate and compare the effective surface areas of the electrodes^{S4-S6}.

Table S1. Comparison of ECSA of prepared electrocatalysts.

Electrocatalysts	Slope	C_{dl}	ECSA
Fe-only/NF	1.27 mF cm ⁻²	0.64 mF cm ⁻²	15.88 cm ² _{ECSA}
FeCe5/NF	1.52 mF cm ⁻²	0.76 mF cm ⁻²	19.00 cm ² _{ECSA}
FeCe10/NF	1.33 mF cm ⁻²	0.67 mF cm ⁻²	16.63 cm ² _{ECSA}
FeCe20/NF	1.55 mF cm ⁻²	0.78 mF cm ⁻²	19.38 cm ² _{ECSA}
FeCe40/NF	1.61 mF cm ⁻²	0.81 mF cm ⁻²	20.13 cm ² _{ECSA}

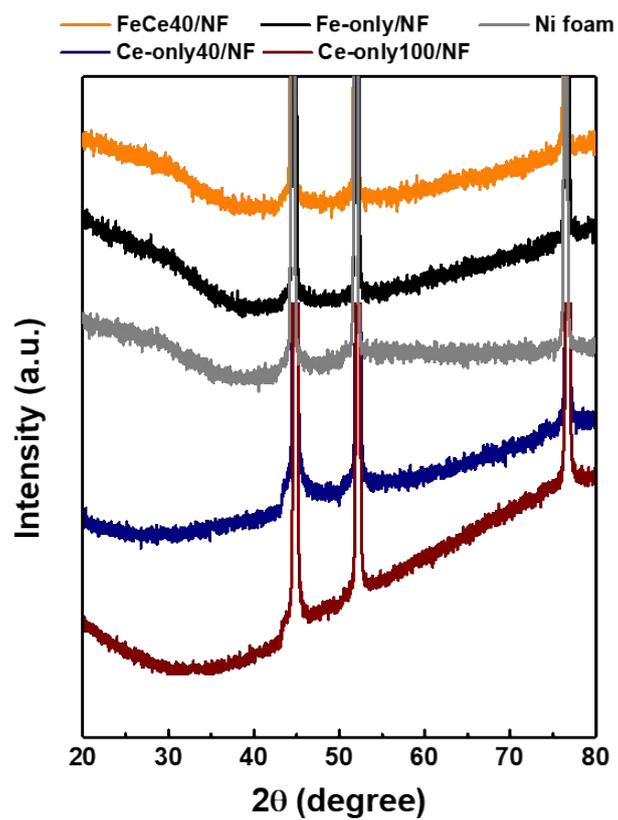


Fig.S3 XRD analysis of the as-prepared electrocatalysts.

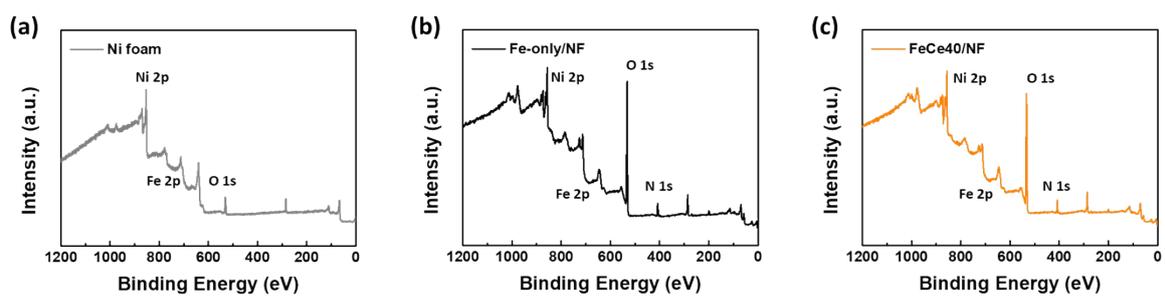


Fig.S4 XPS survey (broad scan) spectra for chemical characterization of (a) Ni foam, (b) Fe-only/NF, and (c) FeCe40/NF.

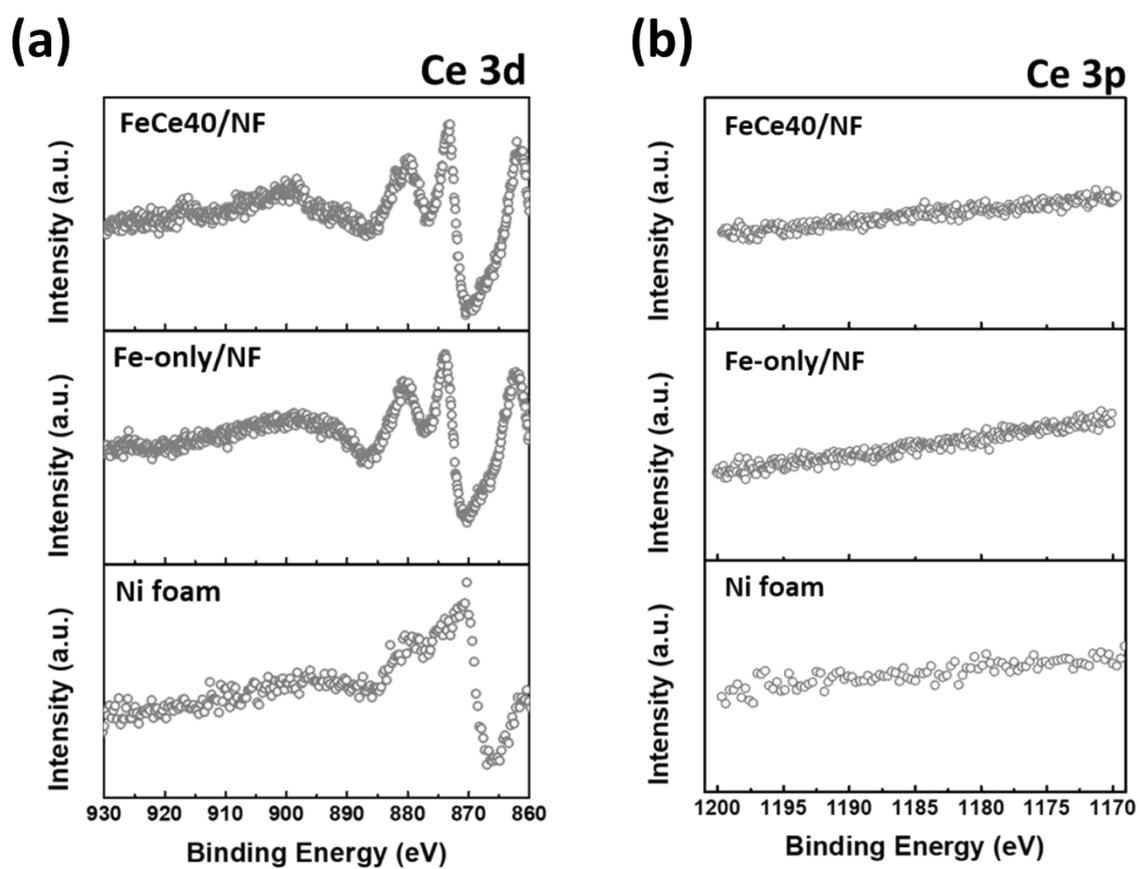


Fig.S5 High-resolution XPS spectra of the (a) Ce 3d and (b) Ce 3p region for the Fe-Ce based electrocatalysts.

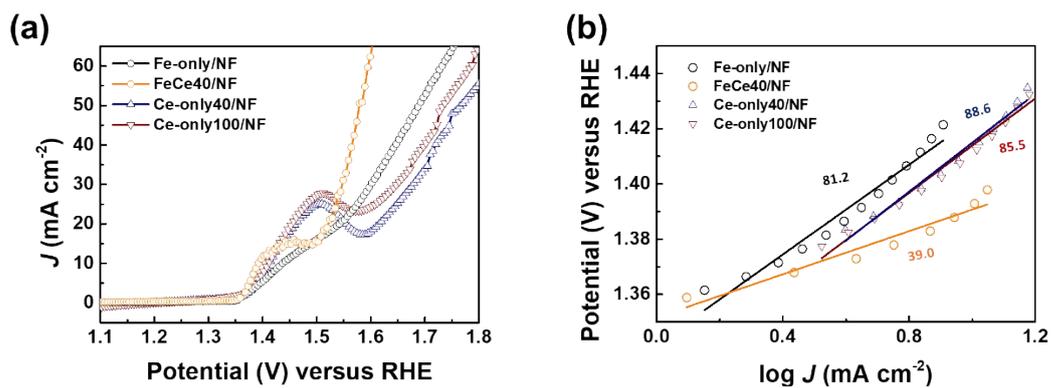


Fig.S6 Comparison of electrochemical OER performance for Fe-only/NF, Ce-only/NF, and FeCe40/NF electrodes. (a) LSV curves recorded in 1.0 M KOH. (b) Corresponding Tafel plots derived from LSV.

Table S2. Equivalent circuit parameters of prepared electrocatalysts.

Electrocatalysts	R_e	R_{ct}	C
Fe-only/NF	5.1 Ω	5.16 Ω	0.032 F
FeCe40/NF	2.1 Ω	4.27 Ω	0.035 F

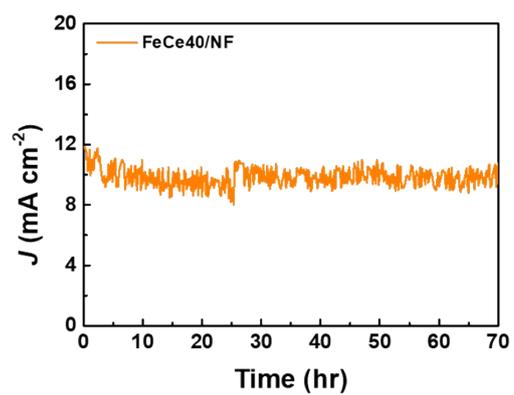


Fig.S7 Long-term stability test of the FeCe40/NF via chronoamperometry at a fixed overpotential of 270 mV (vs. RHE).

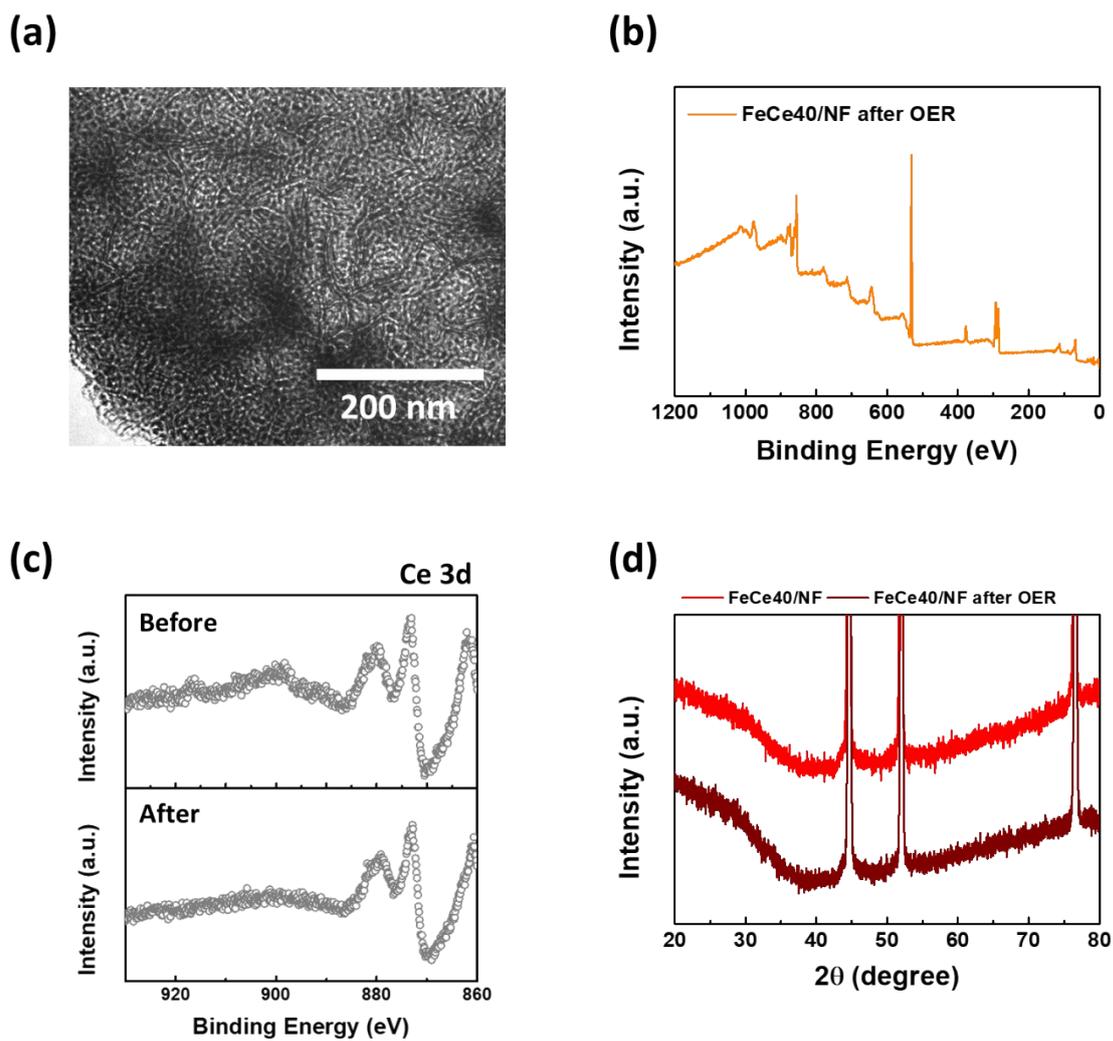


Fig.S8 Post-OER characterization of the electrocatalyst. (a) TEM image after the OER test, (b) XPS survey spectrum after the OER test, (c) comparison of Ce 3d XPS spectra before and after the OER test, and (d) comparison of XRD patterns before and after the OER test.

Table S3. Comparison of OER catalytic activity with previous reports of NiFe-based electrocatalysts.

Catalyst	Synthesis route	Electrolyte	Overpotential (mV)	Current density (mA cm ⁻²)	Tafel slopes (mV dec ⁻¹)	Ref.
FeCe40/NF	Spontaneous redox reaction	1.0 M KOH	349 mV	50 mA cm ⁻²	39 mV dec ⁻¹	Current work
			297 mV	10 mA cm ⁻²		
NiFe-LDH/SnS	Hydrothermal	1.0 M KOH	310	10	53.6	S7
NiFe-LDH/Ti ₃ C ₂	Hydrothermal	1.0 M KOH	334	10	55	S8
NiFe-LDH/Ni(OH) ₂ /NF	Microwave	1.0 M KOH	292	10	84	S9
Ni ₃ Fe-LDH	Coprecipitation	1.0 M KOH	270	10	43	S10
NiFe-OOH _{OV}	Plasma Exfoliation	1.0 M KOH	270	10	38	S11
Ni ₂ Fe-Sodium Dodecyl Sulfonate-LDH	Synthetic Intercalation	1.0 M KOH	289	10	38	S12
LDH/oGSH	Composite	0.1 M KOH	350	10	54	S13
Ni ₃ FeAl _x -LDH	Cationic vacancies	1.0 M KOH	304	20	50	S14
NiFe LDHs-3	Plasma Exfoliation	1.0 M KOH	278	10	87	S15

Dodecyl Sulfate- /NiFe LDH	Anion exchange	1.0 M KOH	300	10	36	S16
NiFe-LDH-15h	Strain tuning	1.0 M KOH	270	10	36	S17
NiFe-Dodecyl Sulfate	Anion exchange	1.0 M KOH	317	10	80	S18
NiFeLa-8 LDH	Electrodeposition	1.0 M KOH	399	50	73	S19
Co ³⁺ -NiFe-LDH	Ni and Fe substitution	1.0 M KOH	305	10	59	S20

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