

Supporting Information

Giant Electric Field Dependent Hole Mobility of CsPbBr₃ Nanocrystals Films

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Part 1 Experiment and measurements

Materials

N-hexane, chlorobenzene1-octadecene (ODE, 90%), oleic acid (OA, 90%), oleylamine (OLA, 90%), lead(II) oxide (PbO, 99.9%), n-tetradecylamine (TDA, 90%), Didodecyltrimethylammonium bromide (DDAB, 99%), cesium carbonate (Cs₂CO₃, 99%), Molybdenum trioxide (MoO₃) bromobenzaldehyde (C₇H₅BrO), were purchased from Aladdin, China. (LiF, 99.85%, Alfa Aesar), Poly(3,4-ethylenedioxythiophene):poly(styrene sulfonate) (PEDOT: PSS), poly [(9,9-dioctylfluorenyl-2,7-diyl)-alt-(4,4'-(N-(4-butylphenyl) (TFB), 1,3,5-tris(1-phenyl-1H-benzimidazol-2-yl) benzene (TPBI) were purchased from Xi'an Polymer Light Technology Corp.

Synthesis of CsPbBr₃ nanocrystal.

This method is based on the previous report by the author.¹ A mixture of 18 mL ODE, 2 mL OA, 0.46 g Cs₂CO₃ was degassed with vacuum (140 °C, 1 hour). And 0.046 g PbO, 0.120 g C₇H₅BrO, 1 mL OA and 10 mL of ODE were added to the flask and heated to 220 °C. Subsequently, 0.3 mL OLA was injected at 120-250°C to obtain different size CsPbBr₃ (for 15 minutes continuously). Then, 0.5 mL of precursor was rapidly injected into the flasks, and the reaction mixture was cooled rapidly by ice bath. The resulting solution was centrifuged twice at 6000 rpm and then added with hexane to obtain the twelve-faced CsPbBr₃ nanocrystals (NCs).

Single hole-carrier device fabrication.

The ITO-coated glass substrates were cleaned sequentially using detergent, acetone, and isopropanol in an ultrasonic bath for 30 min each. After cleaning, they were treated with ozone plasma for 5 min. Next, a PEDOT:PSS solution was spin-coated onto the ITO substrates at 6,000 rpm for 60 s and then annealed at 150 °C for 15 min. The coated substrates were then transferred into a nitrogen-filled glovebox for the deposition of subsequent layers. The TFB layer was spin-coated at 2,000 rpm for 30 s from an 10 mg ml⁻¹ chlorobenzene solution, followed by baking at 150 °C for 30 min. Next, the QD layer (20 mg ml⁻¹) was spin-coated at 2,000 rpm for 30 s, followed by baking at 80 °C for 5 min. MoO₃ (5 nm) and Al (80 nm) were deposited by thermal evaporation under vacuum of about 2.0 × 10⁻⁴ Pa. The structure of hole-carrier device is ITO/PEDOT:PSS/TFB/QDs/MoO₃/Al (device A).

Part 2 Monte Carlo Simulation and Wentzel-Kramers-Brillouin

2.1 Monte Carlo Simulation

This study uses a three-dimensional Miller-Abrahams method to simulate the hopping transport

process of carriers in CsPbBr₃ NCs films. The simulated system consists of a bottom electrode, an active layer, and a top electrode, with overall dimensions of $L_x \times L_y \times L_z$, where the xyz values are adjustable ($L_x = L_y = 200 \text{ nm}$, $L_z = 320 \text{ nm}$). The nanocrystal array is arranged in a simple cubic (SC) or face-centered cubic (FCC) lattice, with anisotropic spacing set as $\Delta x = \Delta y = \Delta z$, satisfying the physical constraints $\Delta x > 2R$, $\Delta y > 2R$, $\Delta z > 2R$ (R is the radius of the CsPbBr₃ NCs). The electrode region has a thickness $d = 10 \text{ nm}$, located at the top and bottom surfaces of the film. The energy of the CsPbBr₃ NCs in the active layer follows a Gaussian distribution $\varepsilon_i \sim N(0, \sigma^2)$, where the energetic disorder σ is calculated using a size-correction formula:

$$\sigma = \sigma_{\text{base}} \cdot \frac{R_{\text{base}}}{R}$$

Here, σ_{base} is 0.05 eV. Carrier transport follows the hopping model, and the hopping rate from position i to j is given by:²

$$r_{ij} = v_0 \exp\left(-\frac{2r_{ij}}{\xi}\right) \cdot \Gamma(\Delta E_{ij})$$

Here, v_0 is attempt frequency (10^{12} Hz), r_{ij} is distance of nanocrystals, the effective energy difference ΔE_{ij} includes both the energy difference and the contribution from the applied electric field:³

$$\Delta E_{ij} = (\varepsilon_j - \varepsilon_i) + V \cdot \frac{z_j - z_i}{L_z}$$

Heat accessory factor $\Gamma(\Delta E)$:

$$\Gamma(\Delta E) = \begin{cases} 1 & \Delta E \leq 0 \\ \exp\left(-\frac{\Delta E}{k_B T}\right) & \Delta E > 0 \end{cases}$$

Here, V is applied voltage, T is 300 K.

The simulation starts from the center of the bottom electrode, searching for nearby quantum dots within a radius of $4 \cdot \max(\Delta x, \Delta y, \Delta z)$. Each hopping process consists of three steps: (1) Calculate the hopping rates r_{ij} for all neighbors and the total rate $R_{\text{total}} = \sum r_{ij}$; (2) Randomly select a target site according to the probability r_{ij}/R_{total} ; (3) Calculate the hopping time $\Delta t = -\ln(U)/R_{\text{total}}$ (U is a uniformly distributed random number between 0 and 1). The simulation termination conditions include: (a) the electron reaches the top electrode; (b) the total rate falls below 10^{-30} s^{-1} ; (c) the number of hops exceeds the threshold $N_{\text{max}} = 10000$. The transmission path is visualized through a 3D trajectory, with the trajectory color mapping time evolution information.

This model quantitatively reveals the synergistic mechanism of inter-dot spacing, energy disorder, and temperature under electric field regulation, providing a theoretical tool for the optimization of perovskite optoelectronic devices.

2.2 WKB tunneling:

The WKB approximation, as a semiclassical method for dealing with one-dimensional barrier tunneling problems, has a core model expression in the exponential form of the tunneling probability. A research method for carrier dynamics in quantum dot devices based on the Wentzel-Kramers-Brillouin (WKB) tunneling model. This model provides a quantitative analysis of carrier transport in nanocrystals.⁴

$$T \approx \exp\left(-\frac{2}{\hbar} \int_{x_1}^{x_2} \sqrt{2m^* (V(x) - E)} dx\right)$$

Here, $E(x)$ is barrier potential height, m^* denotes the carrier effective mass, x_1 and x_2 denote the start and end points of the barrier, respectively and \hbar the reduced Planck constant. Eq.1 indicates the relationship of the tunneling efficiency (Γ) and energy barrier. This expression reveals the intrinsic characteristic of the tunneling probability exponentially decaying with barrier parameters, as well as the coupling relationship between effective mass and barrier parameters.

Part 3 Tables.

Table S1 Average hole mobility of CsPbBr₃ NCs.

Size (nm)	Hole mobility (cm ² /Vs) ($\times 10^{-9}$)			Average mobility (cm ² /Vs) ($\times 10^{-9}$)	Standard Deviation
9	0.38	0.43	0.37	0.39	0.43
12	0.78	0.81	0.78	0.78	0.81
14.5	0.92	0.95	0.9	0.93	0.95
17	1.03	1.0	1.08	1.03	1.0
22	1.84	1.80	1.89	1.85	1.80
26	2.61	2.51	2.55	2.62	2.51

Table S2 The comparison of CsPbBr₃ QD mobility data measured in other articles.

Materials	ligand	Carrier Mobility (m ² V ⁻¹ s ⁻¹)	Method	References
CsPbBr ₃ QD (26 nm)	OA/OLA (18C)	$\mu_e 5.4 \times 10^{-13} - 4.36 \times 10^{-12}$ $\mu_h 3.9 \times 10^{-13} - 4.0 \times 10^{-11}$	SCLC	This work
CsPbBr ₃ QD	OA	$\mu_e 8.14 \times 10^{-11}$	SCLC	Adv. Optical Mater. 2025, 13, 2403039
CsPbBr ₃ QD	Butylamine (4C)	$\mu_h 7 \times 10^{-11}$	FET	ACS Energy Lett. 2019, 4, 534–541
CsPbAgBr ₃ QD	Butylamine (4C)	$\mu_h 8 \times 10^{-8}$	FET	ACS Energy Lett. 2019, 4, 534–541
CsPbBr ₃ QD	oleyl amine /oleic acid (18C)	$\mu_h 6.5 \times 10^{-14}$	SCLC	Adv. Optical Mater. 2025, 13, e01361
CsPbBr ₃ QD	DDAB (12C)	$\mu_e 1.5 \times 10^{-10}$	SCLC	Adv. Optical Mater. 2025, 13, e01361
Unspecified QDs	/	$\mu_{h-e} 1 \times 10^{-7} - 1 \times 10^{-10}$	Numerical simulation	J. Appl. Phys. 126, 035704 (2019)
CsPbBr ₃ QD	OA/OLA (18C)	$\mu_e 2.1 \times 10^{-4}$ $\mu_h 6.9 \times 10^{-3}$	Monte Carlo simulation	ACS Appl. Mater. Interfaces 2021, 13, 44742–44750
CsPbBr ₃ QD	Mercaptopropionic acid (3C)	$\mu_e 5 \times 10^{-7}$	SCLC	Nanomaterials 2020, 10, 1297;
CsPbBr ₃ QD	IPABr/NaBr (1C)	$\mu_{e-h} 1 \times 10^{-6}$	SCLC	Nature Nanotechnology volume 15, pages668–674 (2020)
Materials	ligand	Carrier Mobility (m ² V ⁻¹ s ⁻¹)	Method	References
CsPbBr ₃ QD	OA/OLA (18C)	$\mu_h 6.1 \times 10^{-9}$	SCLC	J. Mater. Chem. C, 2021, 9, 11324–11330
CsGaPbBr ₃ QD	OA/OLA (18C)	$\mu_h 3.6 \times 10^{-8}$	SCLC	J. Mater. Chem. C, 2021, 9, 11324–11330

CsPbBr₃ QD	THPABr (4C+Benzene)	$\mu_h 4.79 \times 10^{-8}$ $\mu_e 4.34 \times 10^{-9}$	SCLC	Chemical Engineering Journal 501 (2024) 157596
CsPbBr₃ QD	PPABr (4C+Benzene)	$\mu_h 6.94 \times 10^{-8}$ $\mu_e 7.32 \times 10^{-9}$	SCLC	Chemical Engineering Journal 501 (2024) 157596
CsPbBr₃ QD	4-CH ₃ -PPABr (5C+Benzene)	$\mu_h 9.36 \times 10^{-8}$ $\mu_e 6.82 \times 10^{-9}$	SCLC	Chemical Engineering Journal 501 (2024) 157596
CsPbBr₃ QD	4-F-PPABr (5C+Benzene)	$\mu_h 6.9 \times 10^{-8}$ $\mu_e 8.4 \times 10^{-9}$	SCLC	Chemical Engineering Journal 501 (2024) 157596
CsPbBr₃ QD	PEABr/MBABr (3C+Benzene)	$\mu_h 3.5 \times 10^{-7}$	SCLC	Nature volume 612, pages679–684 (2022)

Table S3 Total time and jumps for 10000 Monte Carlo Simulation runs.

Size (nm)	Total Time (s)	Jumps
8	3.07×10^{-11}	95
10	8.34×10^{-12}	32
12	6.20×10^{-12}	25
14	5.47×10^{-12}	22
16	5.41×10^{-12}	21
18	4.71×10^{-12}	20
20	4.19×10^{-12}	17
22	4.09×10^{-12}	16
24	3.69×10^{-12}	14
26	3.33×10^{-12}	15
30	2.76×10^{-12}	11

Part 4 Pictures.

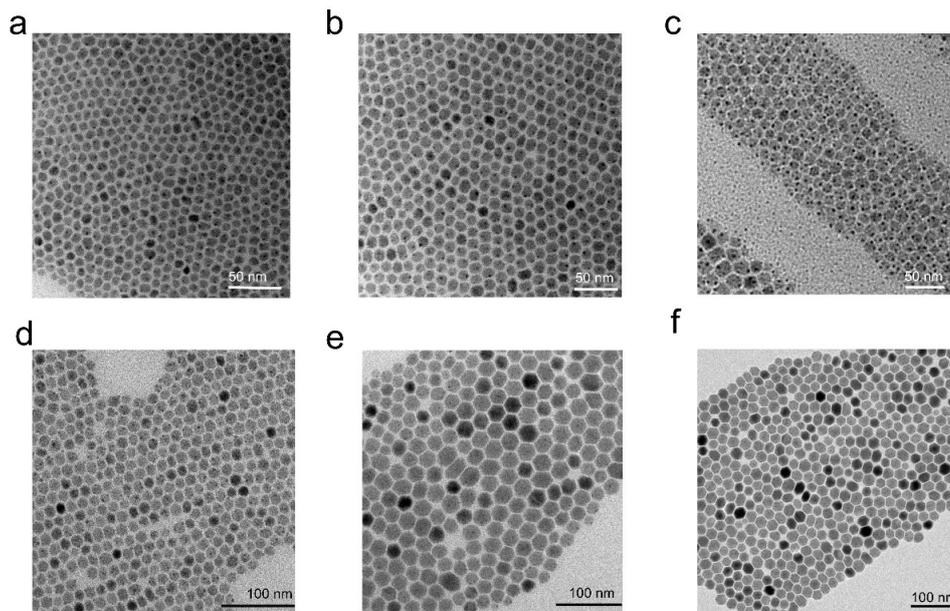


Figure S1 TEM image of CsPbBr₃ NCs. (a) 9 nm. (b) 12 nm. (c) 14.5 nm. (d) 17 nm. (e) 22 nm. (f) 26 nm.

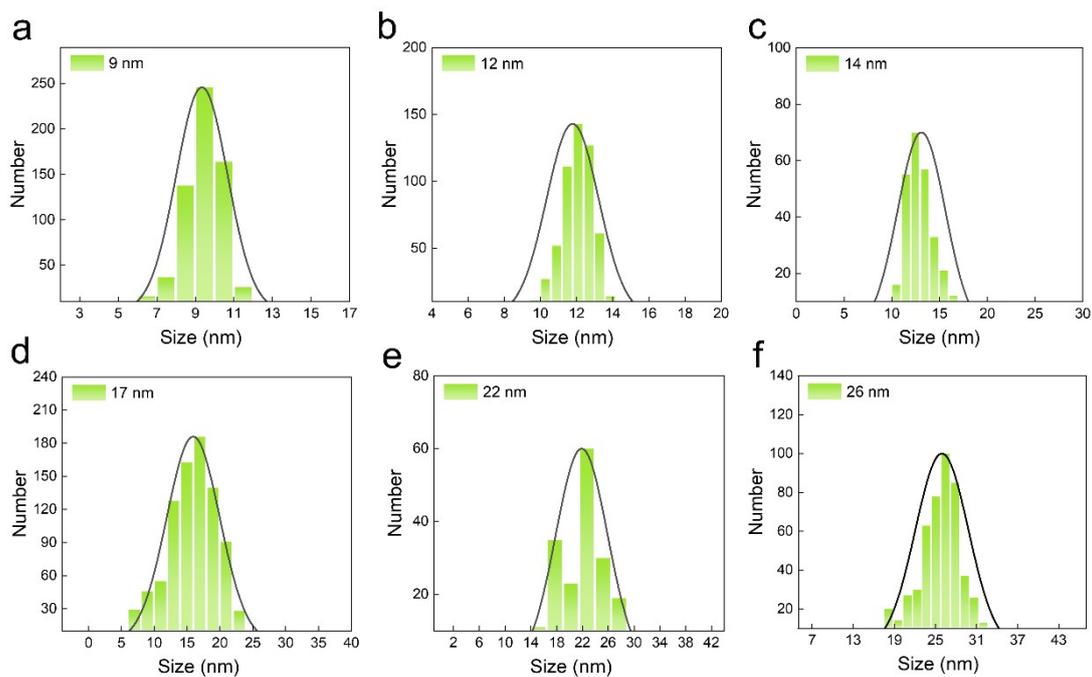


Figure S2 The histogram of different CsPbBr₃ NCs size. (a) 9 nm. (b) 12 nm. (c) 14.5 nm. (d) 17 nm. (e) 22 nm. (f) 26 nm.

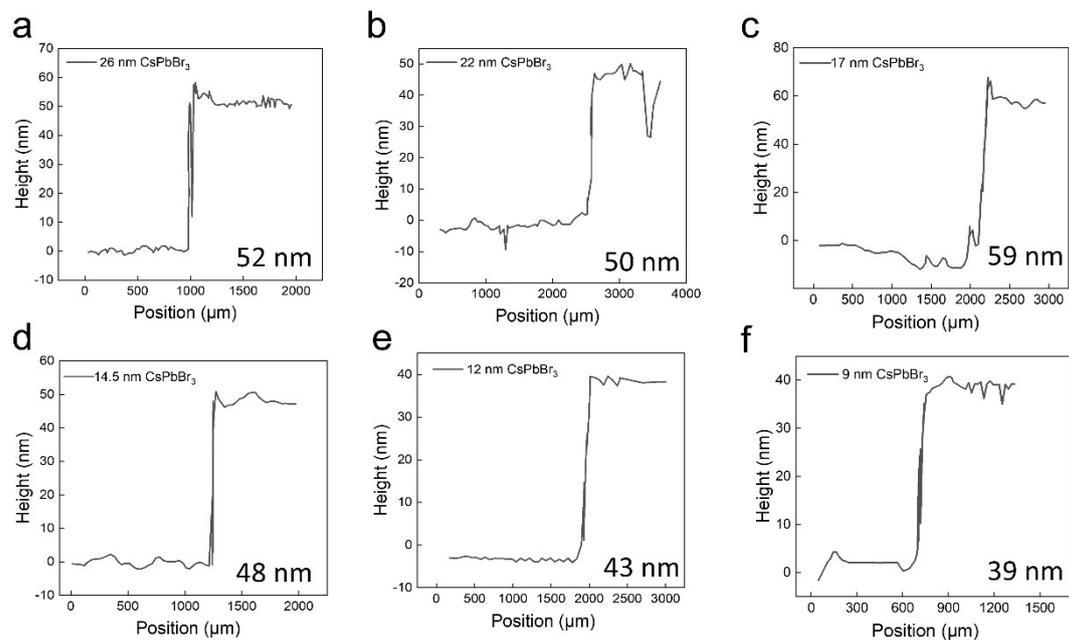


Figure S3 The thickness of different CsPbBr₃ NCs size. (a) 52 nm. (b) 50 nm. (c) 59 nm. (d) 48 nm. (e) 43 nm. (f) 39 nm.

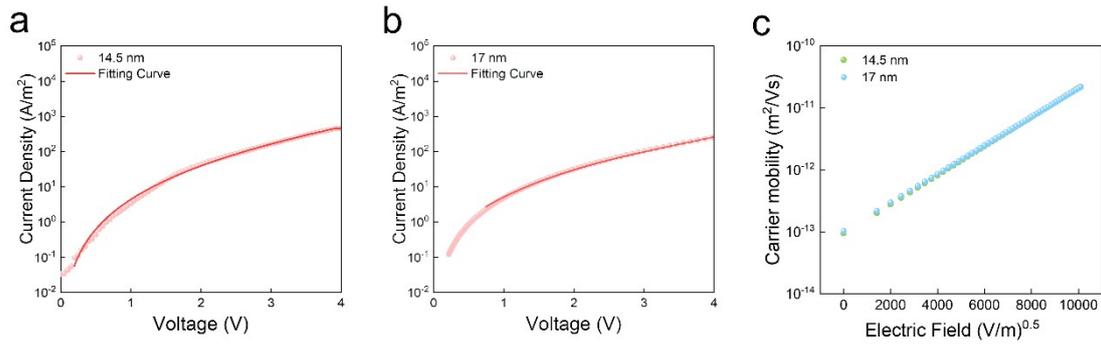


Figure S4 (a) The hole-only J-V curves of CsPbBr₃ NCs with 14.5 nm and corresponding fitting using NS-SCLC. (b) The hole-only J-V curves of CsPbBr₃ NCs with 17 nm and corresponding fitting using NS-SCLC. (c) The electric field dependent carrier mobility of different size CsPbBr₃ NCs.

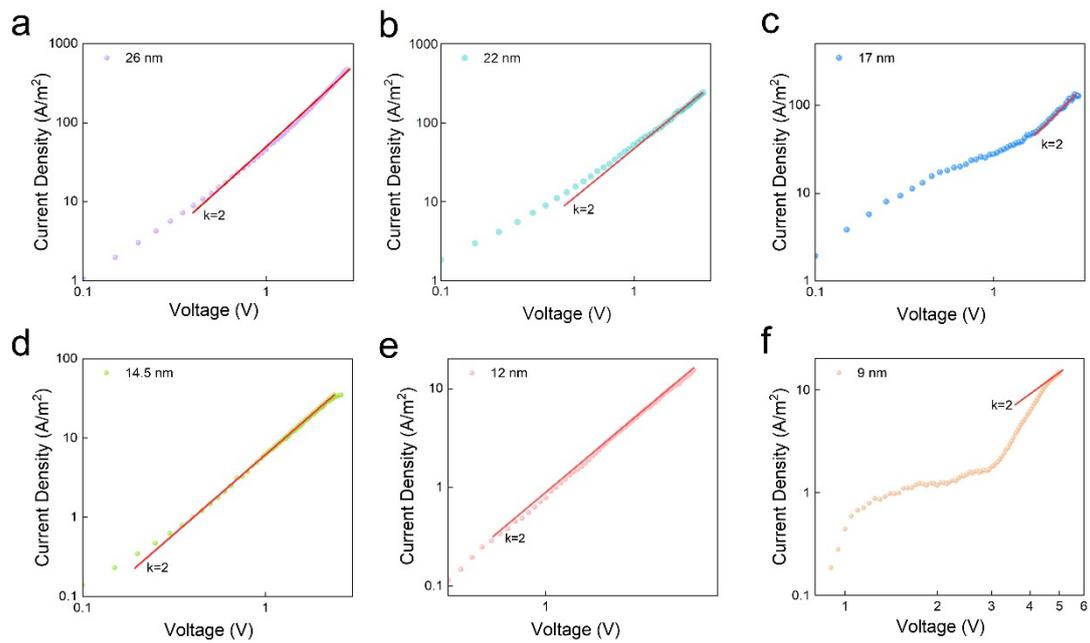


Figure S5 The electron-only J-V curves of CsPbBr₃ NCs and corresponding fitting using NS-SCLC. (a) 26 nm. (b) 22 nm. (c) 17 nm. (d) 14.5 nm. (e) 12 nm. (f) 9 nm.

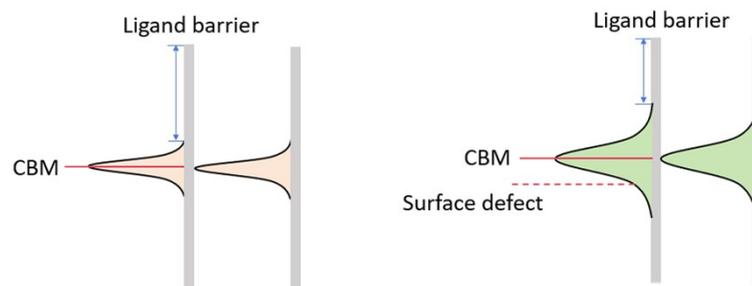


Figure S6 Charge transport processes in CsPbBr₃ NCs.

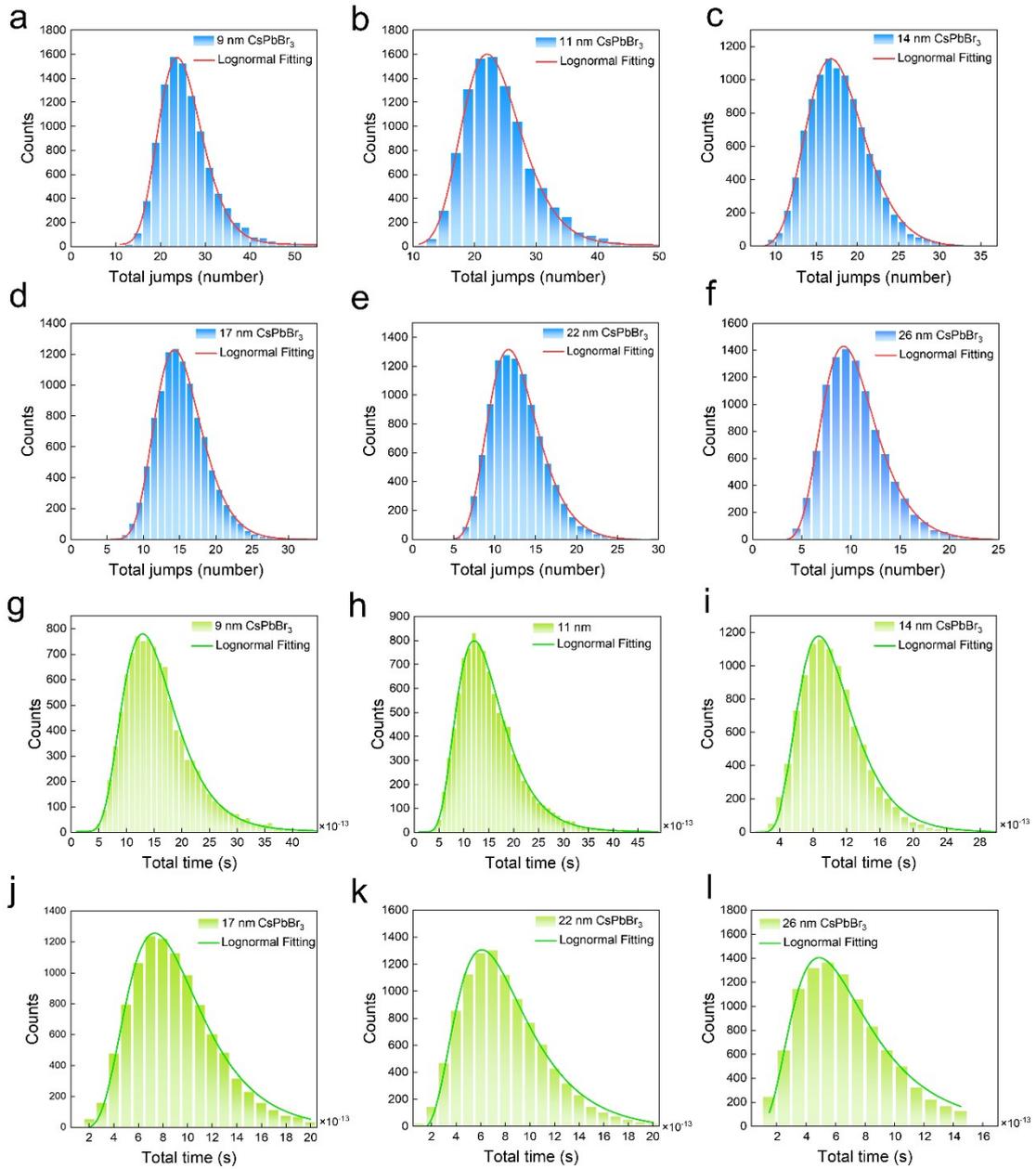


Figure S7 These data are the results of averaging 10000 simulation runs.

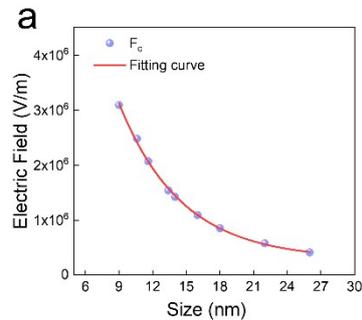


Figure S8 Applying an electric field is necessary to overcome charging energy effects in different size CsPbBr₃ NCs.

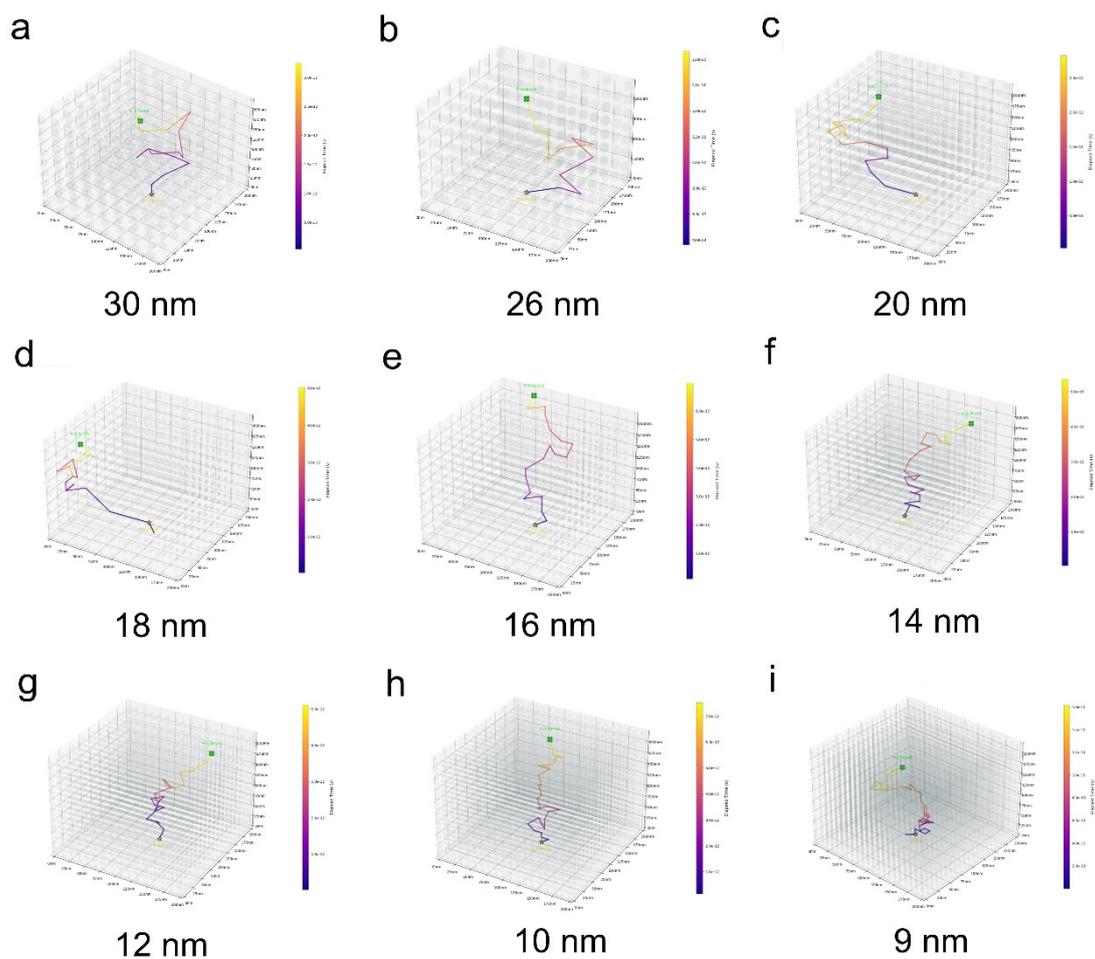


Figure S9 Carrier trajectories under 20 V for 50 Monte Carlo simulations. (a) 30 nm. (b) 26 nm. (c) 20 nm. (d) 18 nm. (e) 16 nm. (f) 14 nm. (g) 12 nm. (h) 10 nm. (i) 9 nm.

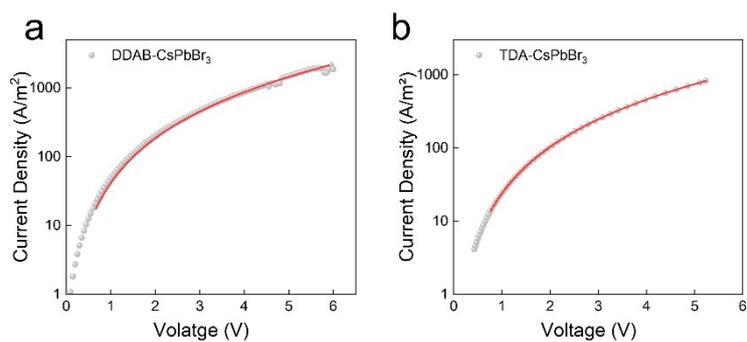


Figure S10 (a) The J-V curves of CsPbBr₃ NCs with DDAB ligand and corresponding fitting using NS-SCLC. (b) The J-V curves of CsPbBr₃ NCs with TDA ligand and corresponding fitting using NS-SCLC.

Reference

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