

Supporting Information For

**“Structurally Robust Quaterphenyl-Dicarbonitrile 2D MOF Nanopores on Cu(111) for Cobalt Spin-Coordination Motifs”**

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Figure S1

### Cleaning process and atomically flat Cu surface

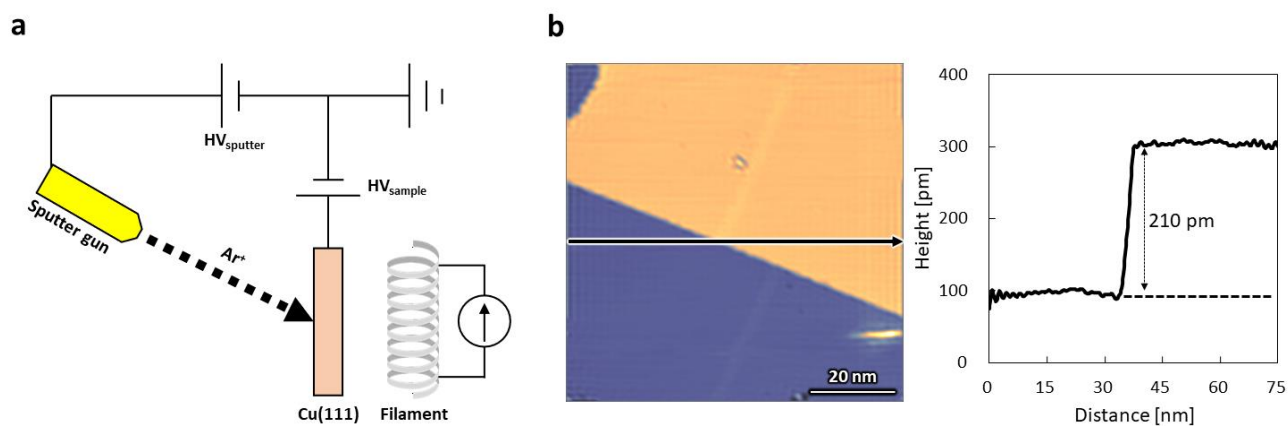


Figure S1. (a) Schematic of the sputtering–annealing cycle used for Cu(111) surface preparation. (b) STM topographic image of the cleaned Cu(111) surface ( $80 \times 80 \text{ nm}^2$ ,  $-2 \text{ V}$ ,  $300 \text{ pA}$ ). The line profile reveals a step height of approximately 210 pm, consistent with a single atomic step of Cu(111).

Figure S2

### NMR measurements of chemically synthesized Ph<sub>4</sub>DN molecules

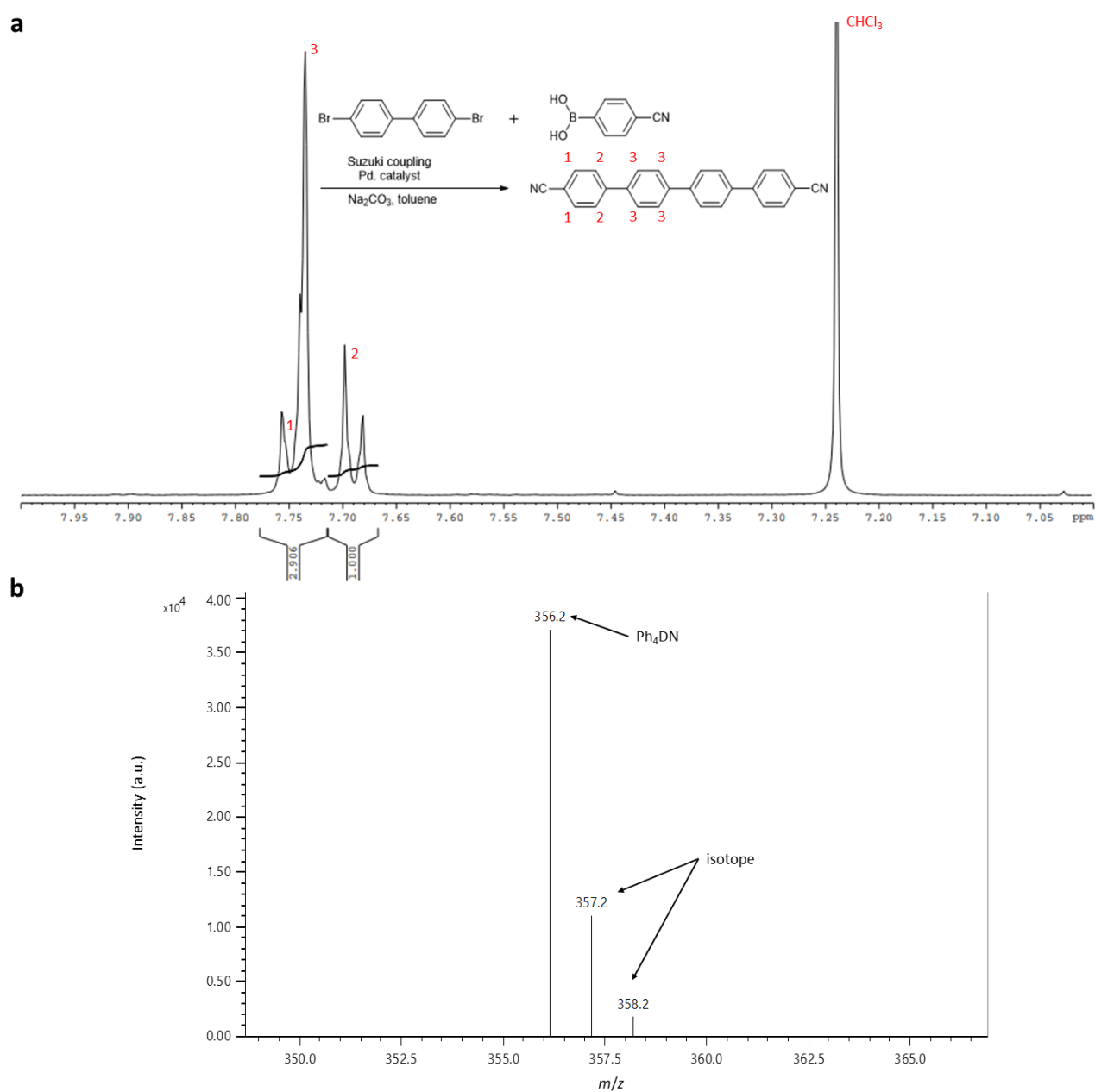


Figure S2. (a) <sup>1</sup>H NMR spectrum of Ph<sub>4</sub>DN (500MHz, CDCl<sub>3</sub>):  $\delta$ (ppm) = 7.75 (d,  $J$  = 9.0 Hz, 4H), 7.74 (s, 8H), 7.69 (d,  $J$  = 8.5 Hz, 4H). (b) Low-resolution field desorption mass spectrum of Ph<sub>4</sub>DN. The peak at  $m/z$  = 356.2 corresponds to  $[C_{26}H_{16}N_2]^+ = 356.13$ .

Figure S3

### Ph<sub>4</sub>DN molecule model and home-built evaporator

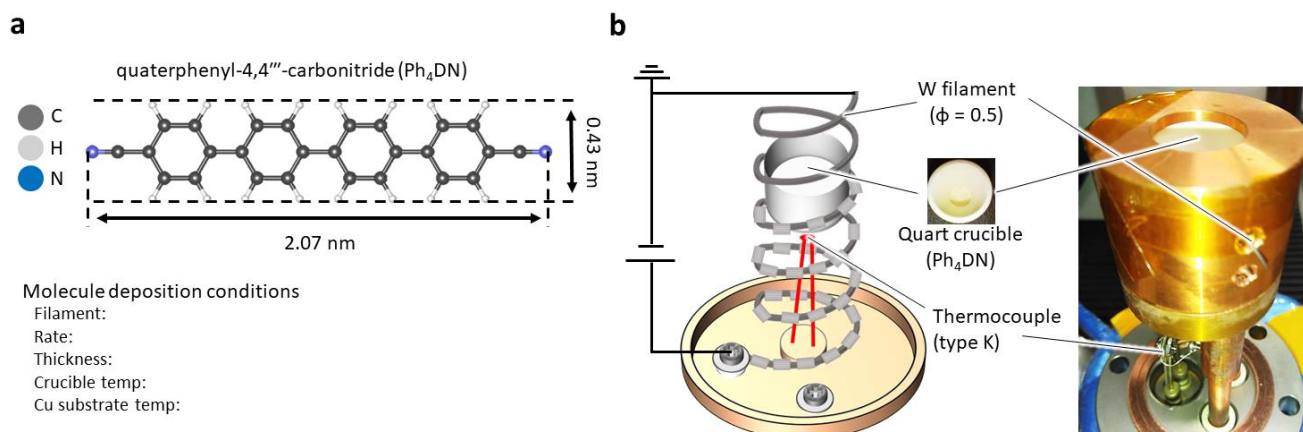


Figure S3. (a) Molecular structure of Ph<sub>4</sub>DN and deposition conditions. (b) Schematic illustration and photo image of the molecular evaporator.

Figure S4

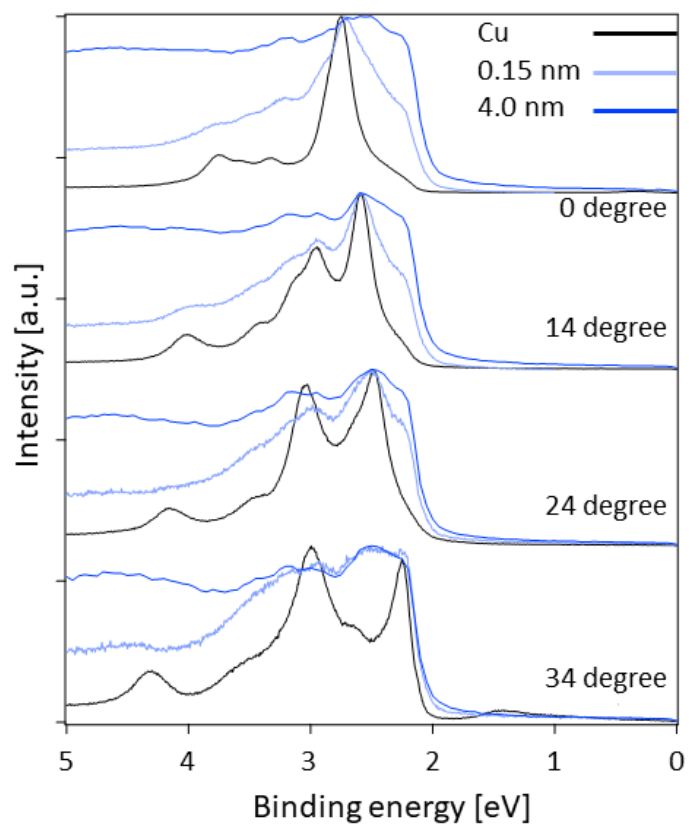
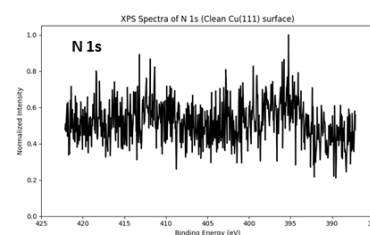
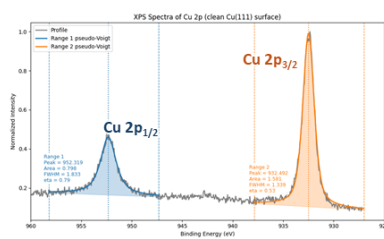
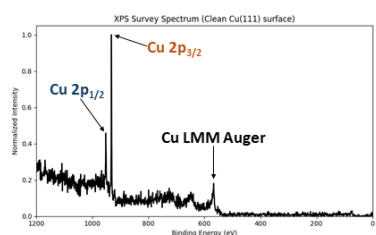


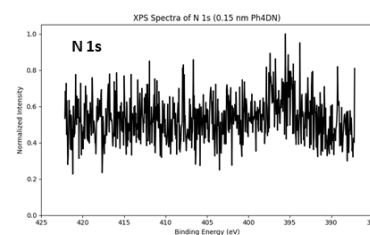
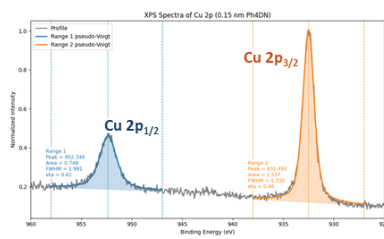
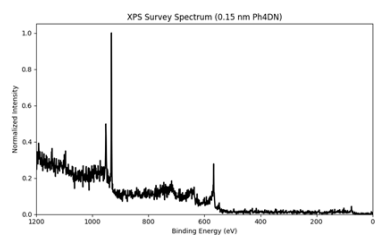
Figure S4. ARPES spectra at different emission angles for bare Cu(111), 0.15 nm Ph<sub>4</sub>DN, and 4.0 nm Ph<sub>4</sub>DN on Cu(111).

Figure S5

XPS measurements  
Pristine Cu(111)



0.15 nm Ph<sub>4</sub>DN/Cu(111)



4.0 nm Ph<sub>4</sub>DN/Cu(111)

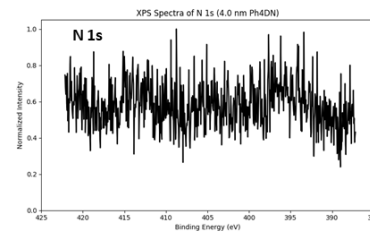
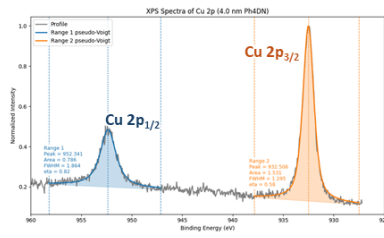
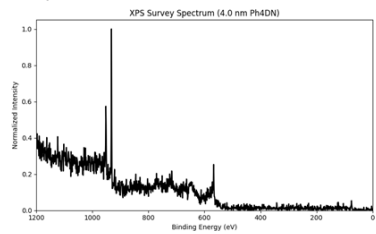


Figure S5. XPS measurements for bare Cu(111) (top row), 0.15 nm Ph<sub>4</sub>DN (middle row), and 4.0 nm Ph<sub>4</sub>DN (bottom row) on Cu(111), where the binding energy regions of Cu 2p and N 1s are measured (middle and right panels). The Cu 2p spectra show no discernable change upon the MOF formation and N 1s are likely under the detection limit in this measurement.

## Supplementary Note 1 for the Comparison of DFT and PFP

DFT calculations were conducted to evaluate the predictive accuracy of the machine-learned interatomic potential (MLIP) used in this study, PFP<sup>1</sup>. The exchange-correlation functional was treated using the generalized gradient approximation (GGA) with the Perdew–Burke–Ernzerhof (PBE) formulation<sup>2</sup>. A plane-wave basis set was employed to construct the wave functions, and core electrons were described using the projector augmented-wave (PAW) method<sup>3,4</sup>. The number of valence electrons was set to 11 for Cu, 5 for N, 4 for C, and 1 for H. The energy cutoffs for the wave function and the augmented charge density were 400 eV and 2400 eV, respectively. Atomic charges were estimated using the Bader analysis based on the electron density obtained from the DFT calculations<sup>5–8</sup>. All calculations were performed using a single  $\Gamma$ -point. The DFT calculations were carried out with the VASP code<sup>9–12</sup>.

Figure S6 presents the results of geometry optimization performed without the Cu(111) surface. The initial structure, shown in Fig. S6a, is an ideal honeycomb lattice. Both DFT and PFP calculations yielded optimized structures in which the Ph<sub>4</sub>DN molecules adopt a three-dimensionally distorted configuration with noticeable twisting. In the PFP-optimized structure, the molecules exhibit more pronounced twisting compared to the DFT result and lie relatively closer to a flat plane. Despite minor differences between the DFT and PFP results, the initially ideal 2D network underwent significant distortion during the optimization process. Furthermore, the number of valence electrons for Cu in the optimized structures was approximately 10, with no significant

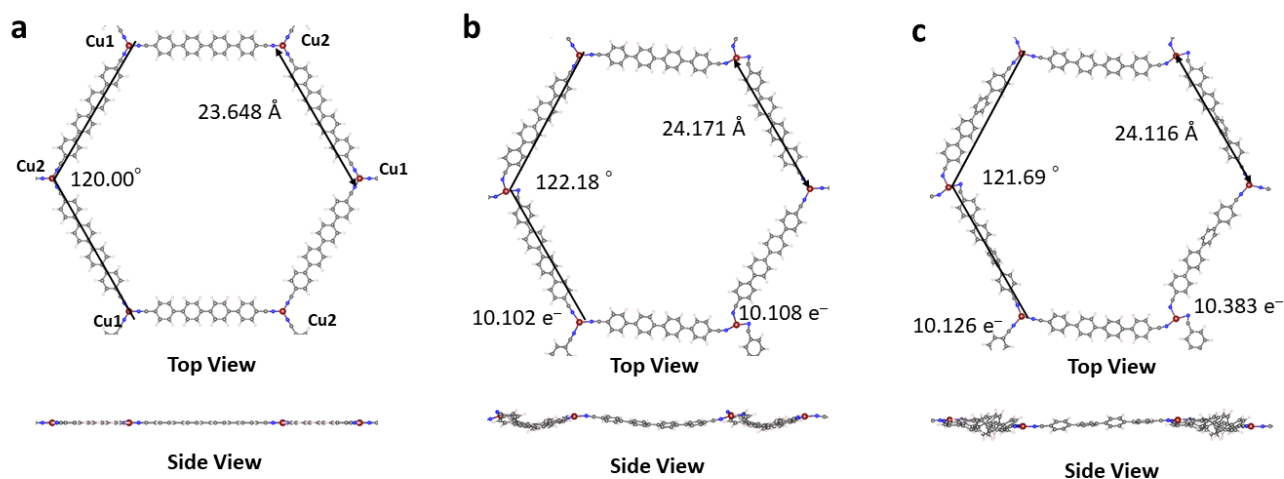


Figure S6. (a) Initial structure of Cu-Ph<sub>4</sub>DN MOF, and optimized structures by (b) DFT and (c) PFP. The valence electron numbers are also shown.

difference observed between the DFT and PFP calculations.

To assess the accuracy of energy estimation for models including the Cu substrate, the interaction energy between Cu(111) and the 2D MOF was compared. The Cu(111) surface was modeled as a monoatomic layer, and the structures of both the Cu(111) and 2D MOF were kept fixed. Figure S7 shows the calculated potential energy curves, clearly demonstrating that the PFP method accurately reproduces the DFT results for the 2D MOF on Cu(111).

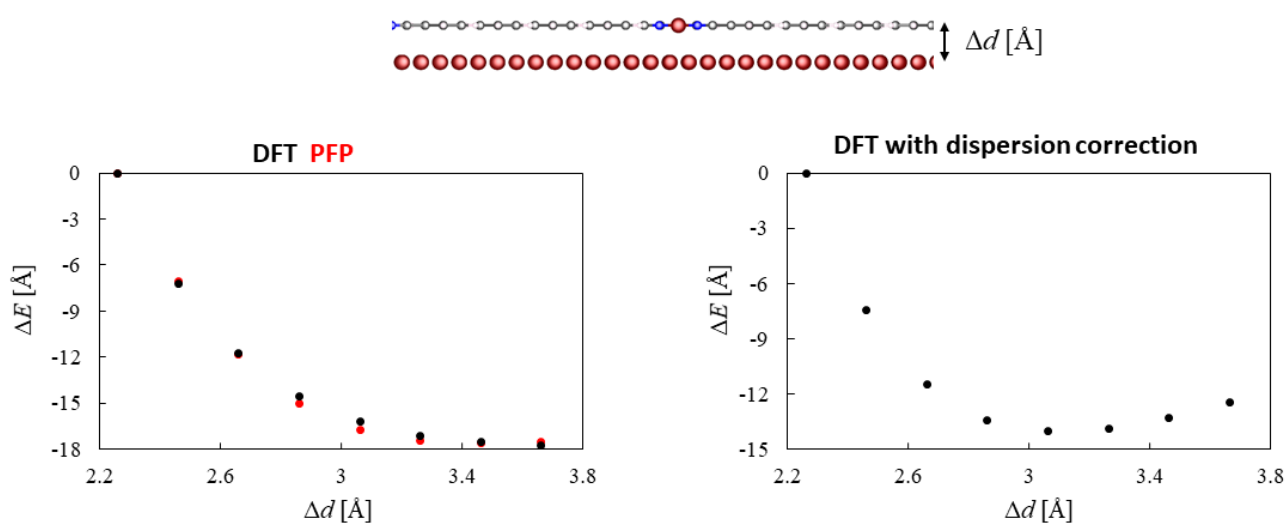


Figure S7. Total energy difference ( $\Delta E$ ) from the model structure, where the distance between Cu-Ph<sub>4</sub>DN and Cu(111) is 2.26 Å. The results of DFT (black dots) and PFP (red dots) are in good agreement. The results shown in the left panel do not include any dispersion force correction, i.e., the CRYSTAL\_U0 mode and pure PBE functional were used for PFP and DFT, respectively.

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(12) Kresse, G.; Furthmüller, J. Efficiency of *Ab-Initio* Total Energy Calculations for Metals and Semiconductors Using a Plane-Wave Basis Set. *Computational Materials Science* **1996**, *6* (1), 15–50. [https://doi.org/10.1016/0927-0256\(96\)00008-0](https://doi.org/10.1016/0927-0256(96)00008-0).

Figure S8

### Gaussian calculations of Co and Fe adatom coordination structures with $C_2$ , $C_3$ , and $C_4$ symmetries under various charge states

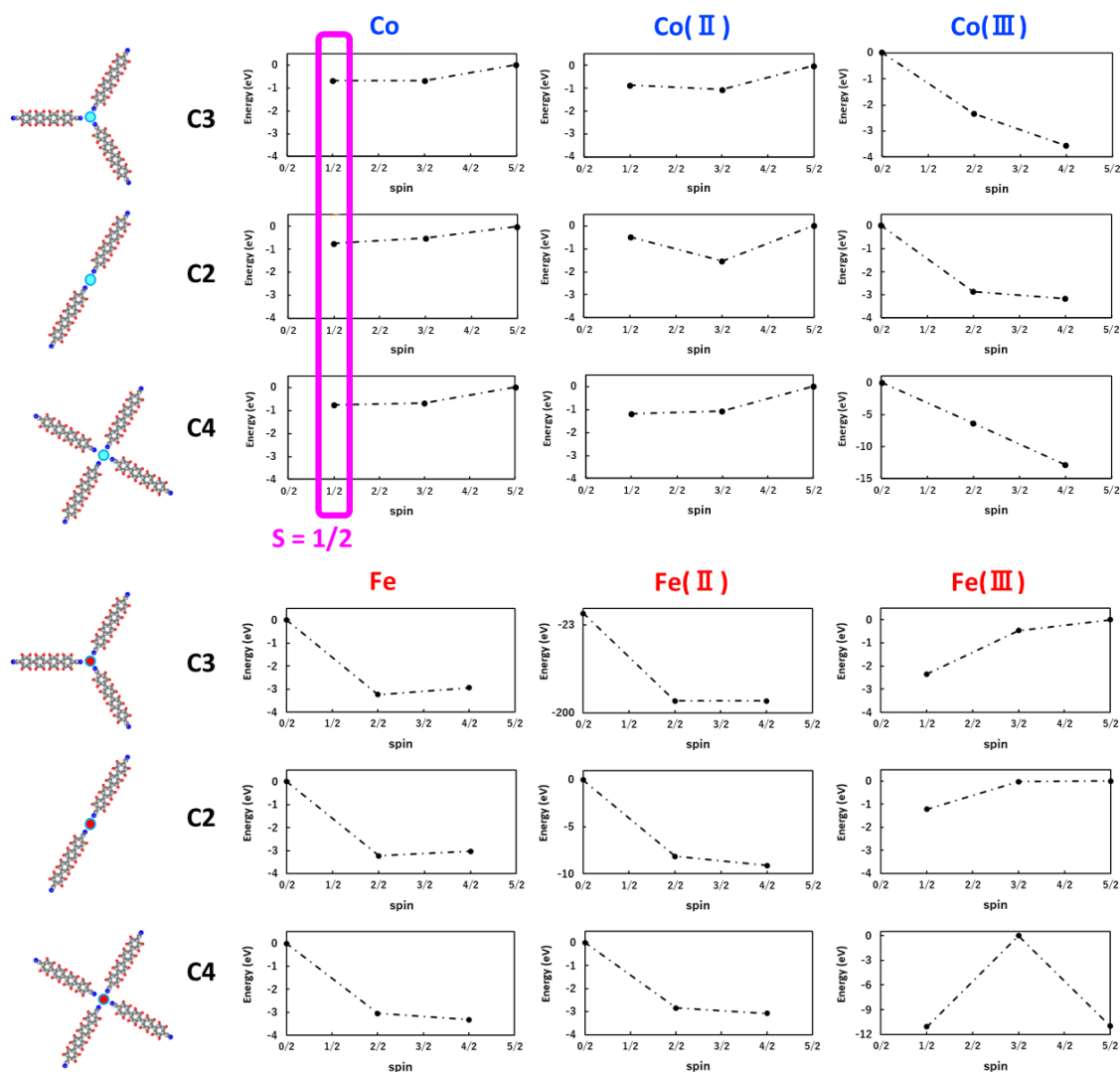


Figure S8. Relative energy differences calculated using DFT (Gaussian 16) for  $Ph_4DN$  coordination to Co and Fe atoms with  $C_2$ ,  $C_3$ , and  $C_4$  symmetries. The used functional and basis sets were B3LYP and SDDALL, respectively. The effective core potentials for the SDDALL were used.

## Supplementary Note 2

We initially investigated a shorter linear precursor composed of three benzene rings, 1,4-Di(4-pyridyl)benzene (DPB), and observed the formation of a 2D MOF on a Cu(111) surface. However, the DPB molecules primarily formed  $C_2$ -symmetric linear chains and a self-assembled monolayer (SAM) without well-defined nanopores (see Supplementary, Fig. S9). Furthermore, DPB began to sublime at approximately 320 K under UHV, just above room temperature, making it difficult to achieve a consistent and controllable deposition rate.

Figure S9

### STM measurements of DPB molecules on Cu(111)

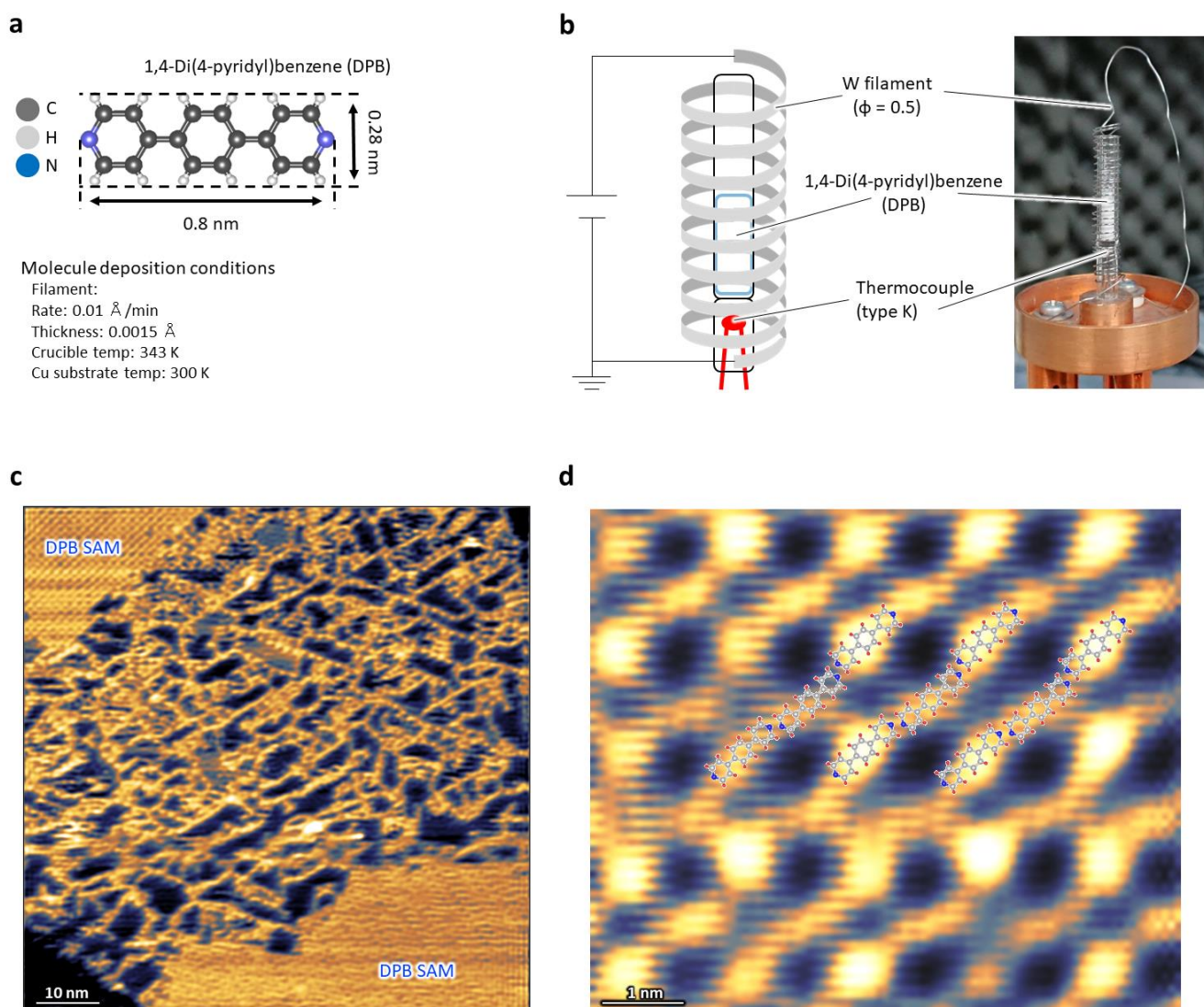


Figure S9. (a) Molecular structure of the DPB molecule and deposition conditions. (b) Schematic illustration and photo image of our home-built molecular evaporator. (c) STM topographic image of the DPB molecules on Cu(111), in which 2D SAM area and one-dimensional networks are observed ( $80 \times 80 \text{ nm}^2$ , -2.0 V, 50 pA). (d) Magnified STM topographic image of DPB SAM structure on Cu(111) ( $7.3 \times 6.2 \text{ nm}^2$ , -2.0 V, 100 pA).

Figure S10

STM topographic images at various voltages of 2D SAM films on Cu(111)

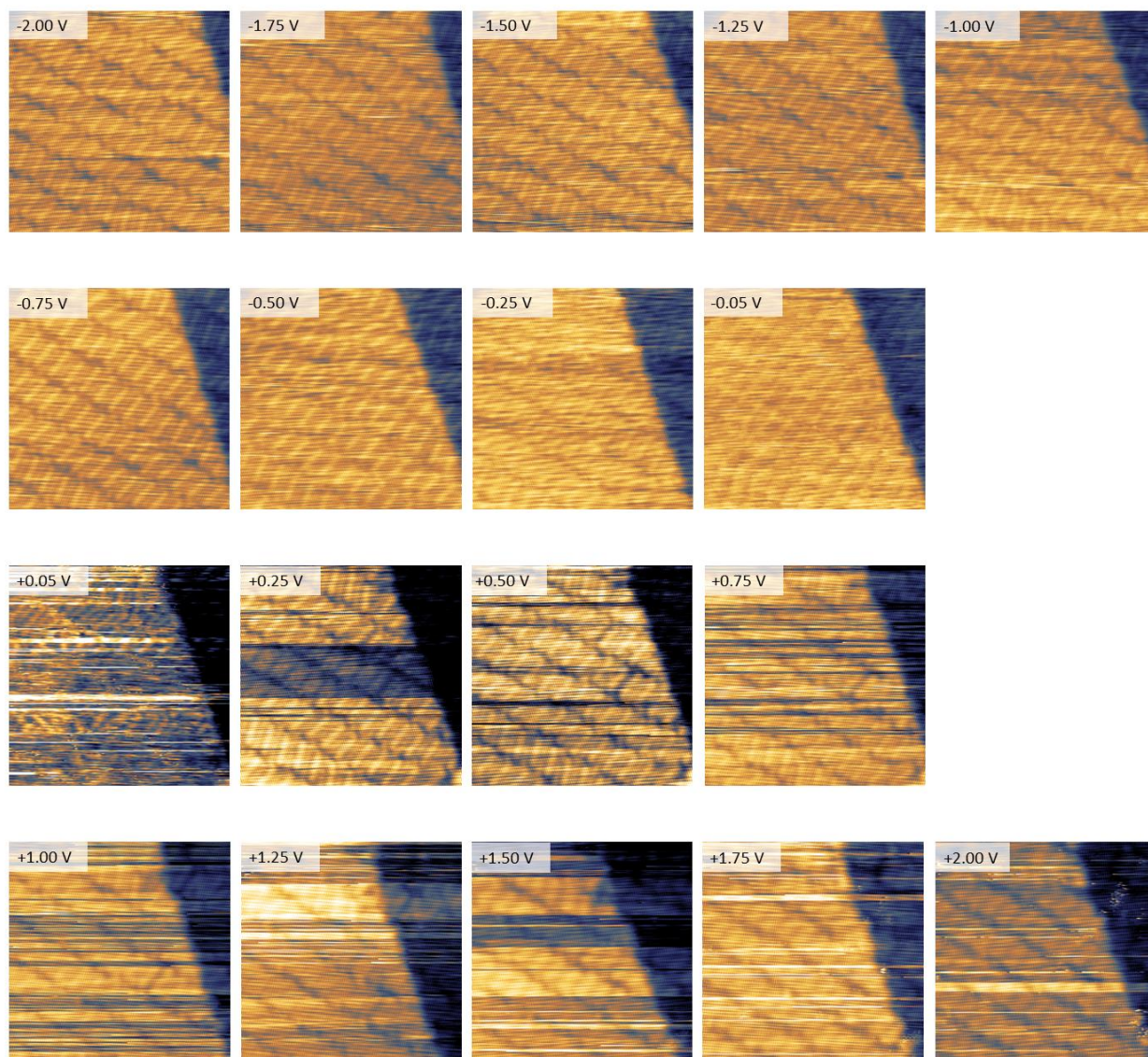


Figure S10. STM topographic images ( $14 \times 14 \text{ nm}^2$ ,  $I_t = 50 \text{ pA}$ ) of the Ph<sub>4</sub>DN SAM region at various sample bias voltages between -2 V and +2 V. Within the HOMO-LUMO gap, the molecular patterns are not clearly visible.

Figure S11

**1D Chains on Cu(111) : STM and STS  $dI/dV$  maps**

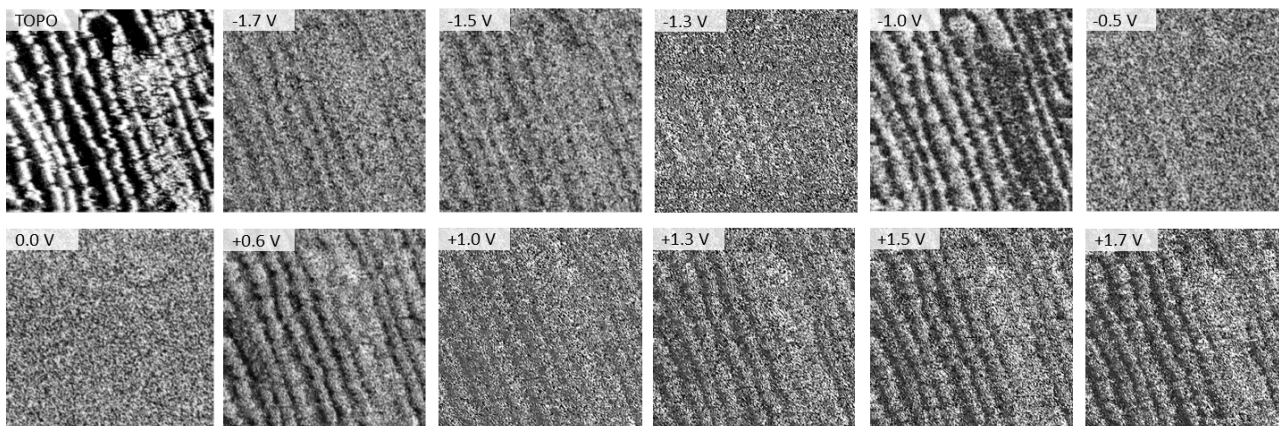


Figure S11. Scanning tunneling spectroscopy mapping on the 1D chain area of  $\text{Ph}_4\text{DN} / \text{Cu}(111)$ , where simultaneously obtained STM topographic image ( $20 \times 20 \text{ nm}^2$ ,  $-1.8 \text{ V}$ ,  $50 \text{ pA}$ ) and  $dI/dV$  maps at  $-1.7$ ,  $-1.5$ ,  $-1.3$ ,  $-1.0$ ,  $-0.5$ ,  $0.0$ ,  $+0.6$ ,  $+1.0$ ,  $+1.3$ ,  $+1.5$ , and  $+1.7 \text{ V}$ .

Figure S12

### STM topographic images and STS $dI/dV$ maps of 2D quasi-honeycomb MOFs on Cu(111)

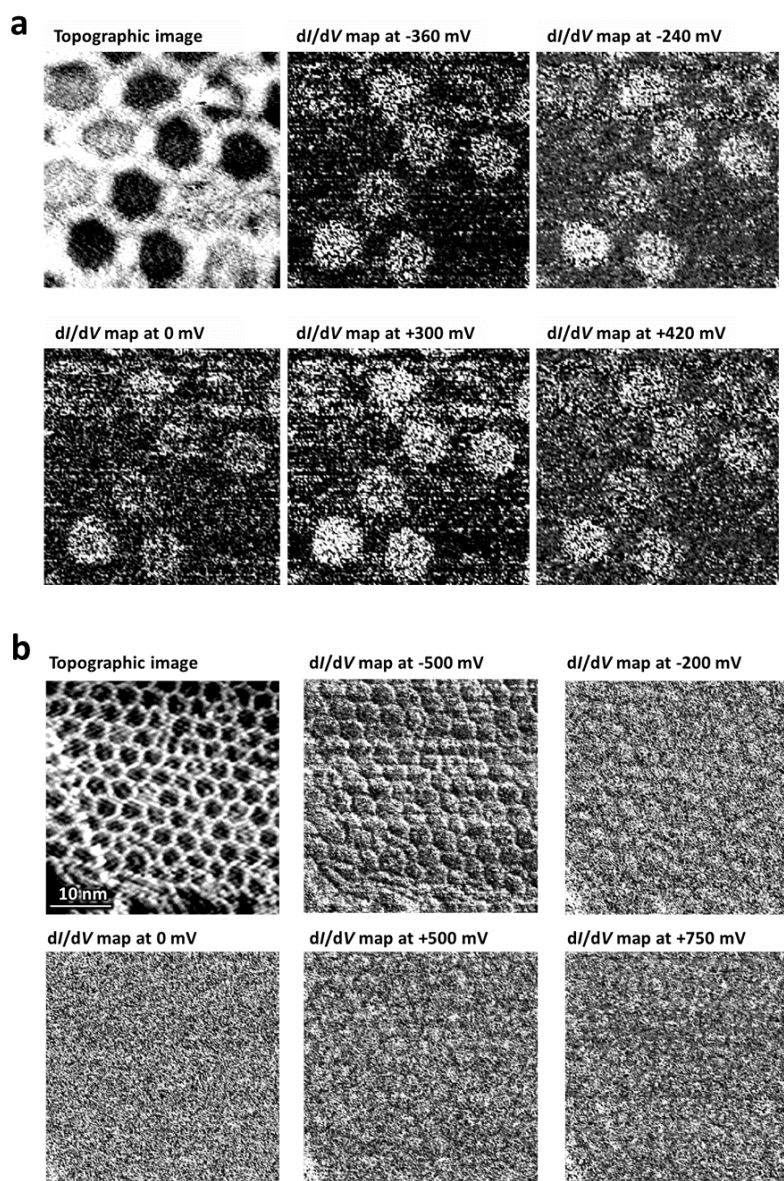


Figure S12. Electronic structures of the 2D quasi-honeycomb networks on Cu(111). (a) STM topographic image ( $14 \times 14 \text{ nm}^2$ ,  $-0.75 \text{ V}$ ,  $50 \text{ pA}$ ) and simultaneously obtained  $dI/dV$  maps at  $-360$ ,  $-240$ ,  $0$ ,  $+300$ , and  $+420 \text{ mV}$ . (b) STM topographic image ( $40 \times 40 \text{ nm}^2$ ,  $-0.75 \text{ V}$ ,  $50 \text{ pA}$ ) and simultaneously obtained  $dI/dV$  maps at  $-500$ ,  $-200$ ,  $0$ ,  $+500$ , and  $+750 \text{ mV}$ .

Figure S13

### 2D SAM film and 1D chains of Ph<sub>4</sub>DN on Cu(111)

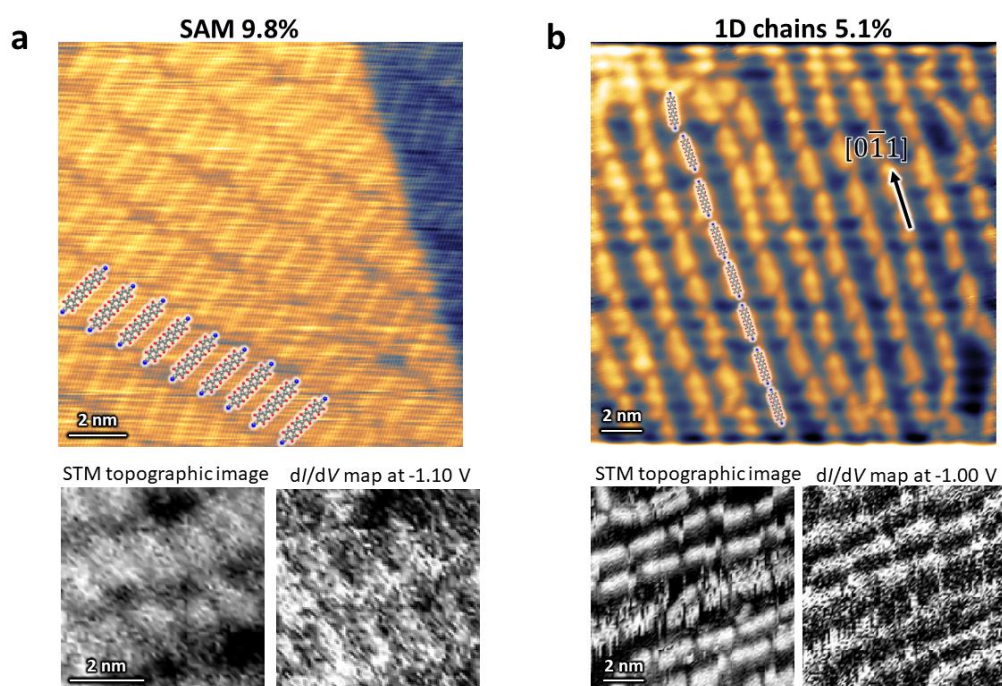


Figure S13. (a) STM topographic image of Ph<sub>4</sub>DN SAM array ( $14 \times 14 \text{ nm}^2$ ,  $V_s = -1.0 \text{ V}$ ,  $I_t = 50 \text{ pA}$ ), and the magnified STM image and the simultaneously obtained  $dI/dV$  map at  $-1.1 \text{ V}$  (lower panels,  $6 \times 6 \text{ nm}^2$ ). (b) STM topographic image of Ph<sub>4</sub>DN 1D chains ( $20 \times 20 \text{ nm}^2$ ,  $20 \text{ pA}$ ,  $-1.8 \text{ V}$ ), and the magnified STM image and the simultaneously obtained  $dI/dV$  map at  $-1.0 \text{ V}$  (lower panels,  $10 \times 10 \text{ nm}^2$ ).

Figure S14

**STM topographic images of phase boundaries  
between 2D MOFs, 1D chains, and SAM regions**

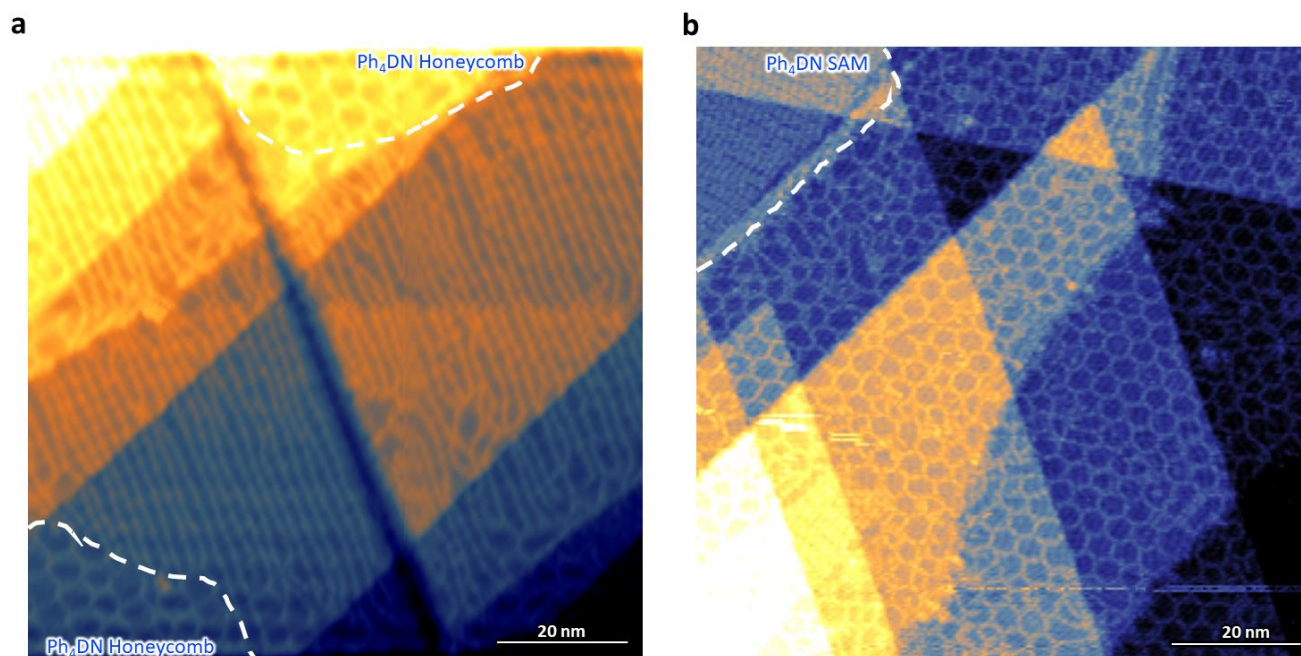


Figure S14. STM topographic images (78 K) obtained on Cu(111) after the Ph<sub>4</sub>DN deposition in UHV at 300 K. In (a), a two-dimensional honeycomb network region coexists with a one-dimensional chain region, and the two merge at their boundary ( $80 \times 80 \text{ nm}^2$ ,  $-1.8 \text{ V}$ ,  $50 \text{ pA}$ ). In (b), the honeycomb network transitions smoothly into the SAM region ( $95 \times 95 \text{ nm}^2$ ,  $0.75 \text{ V}$ ,  $50 \text{ pA}$ ).

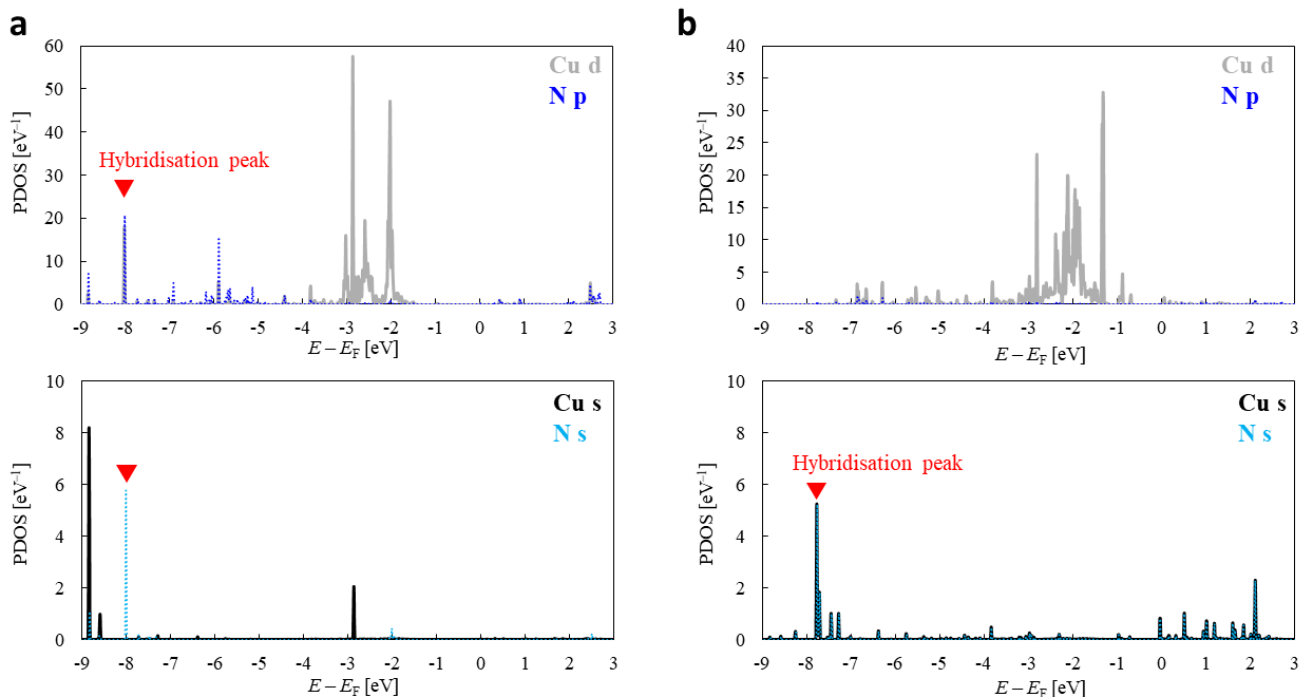
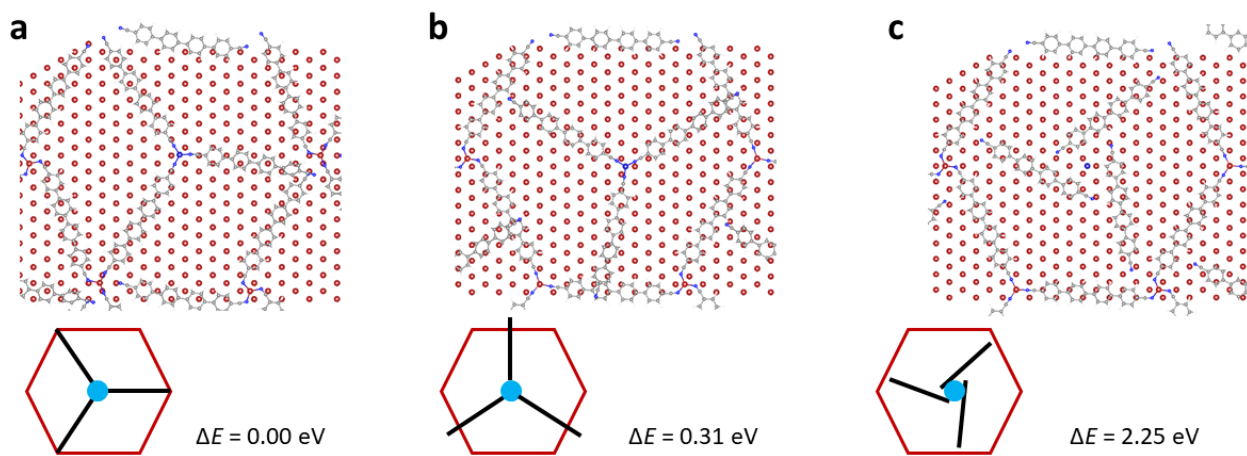
**Figure S15**

Figure S15. Projected density of states (PDOS) of the 2D MOF on Cu(111) model whose structure was optimised by MLIP and shown in Fig. 1d. (a) PDOS results of Cu in  $C_3$  junction and coordinated N, and (b) those of Cu in chiral junction and coordinated N.  $E_F$  represents the Fermi energy of calculated model. Upper panels show the components of Cu  $3d$  and N  $2p$ , while lower panels show those of Cu  $4s$  and N  $2s$ . We found Cu  $3d$  bands at  $-1.5 < E - E_F < -3.5$  eV, and hybridisation peaks between Cu and N. For  $C_3$  junction (a), Cu  $3d$ , N  $2s$ , and N  $2p$  are hybridised and the corresponding peak is found at  $E - E_F = -8$  eV, while for chiral junction, Cu  $4s$  and N  $2s$  hybridisation is found at  $E - E_F = -7.8$  eV.

Figure S16

With a Co atom



Without a Co atom

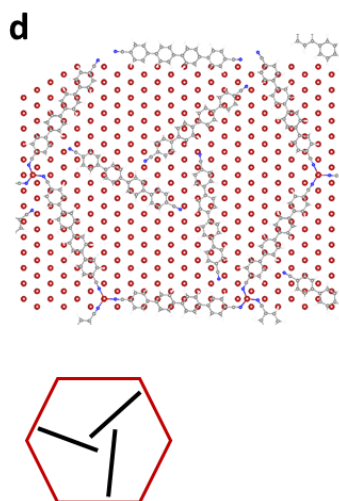


Figure S16. Optimized structures of additional Ph<sub>4</sub>DN models (a–c) with a Co adatom and (d) without. Brown hexagons represent schematic views.  $\Delta E$  denotes the energy difference of each Co-adatom model relative to its most stable configuration.