

# Structural and Electronic Properties of MXene Flakes: From Edge Effects to Bandgap Evolution

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**Table S1.** Cleavage energies,  $E_{\text{Cleavage}}$ , of  $M_2C$  MXenes obtained from nanoribbons exposing each surface. Data is disclosed in two sections, separating ABC and ABA stacking for pristine MXenes.  $(10\bar{1}0)$ ,  $(11\bar{2}0)_{\text{Non-polar}}$ , and  $(11\bar{2}0)_{\text{Polar}}$  surfaces are shown for ABC stacking and  $(10\bar{1}0)$  and  $(11\bar{2}0)$  for ABA. All values are given in  $\text{J}/\text{m}^2$ .

ABC	Sc <sub>2</sub> C	Ti <sub>2</sub> C	V <sub>2</sub> C	Cr <sub>2</sub> C	ABA	Cr <sub>2</sub> C
$(10\bar{1}0)$	7.30	9.77	14.00	16.76	$(10\bar{1}0)$	11.25
$(11\bar{2}0)_{\text{Non-polar}}$	6.56	8.70	11.08	8.96	$(11\bar{2}0)$	9.99
$(11\bar{2}0)_{\text{Polar}}$	8.76	11.83	16.02	20.77		
	Y <sub>2</sub> C	Zr <sub>2</sub> C	Nb <sub>2</sub> C			Mo <sub>2</sub> C
$(10\bar{1}0)$	5.78	8.91	12.23		$(10\bar{1}0)$	9.88
$(11\bar{2}0)_{\text{Non-polar}}$	5.38	7.94	9.45		$(11\bar{2}0)$	9.95
$(11\bar{2}0)_{\text{Polar}}$	6.83	10.26	13.65			
		Hf <sub>2</sub> C	Ta <sub>2</sub> C			W <sub>2</sub> C
$(10\bar{1}0)$		9.77	14.34		$(10\bar{1}0)$	12.14
$(11\bar{2}0)_{\text{Non-polar}}$		8.43	10.29		$(11\bar{2}0)$	12.35
$(11\bar{2}0)_{\text{Polar}}$		11.21	15.33			

**Table S2.** Regression data of Figures 3 and S2 corresponding to differences in energy per stoichiometric unit between flakes and periodic slabs of  $M_2C/M_2CO_2$  MXenes vs. number of stoichiometric units of flakes to the power of  $-1/2$ . Each MXene is referred to its own slab surface and data for Cr with ABA structure is in parenthesis and italics.

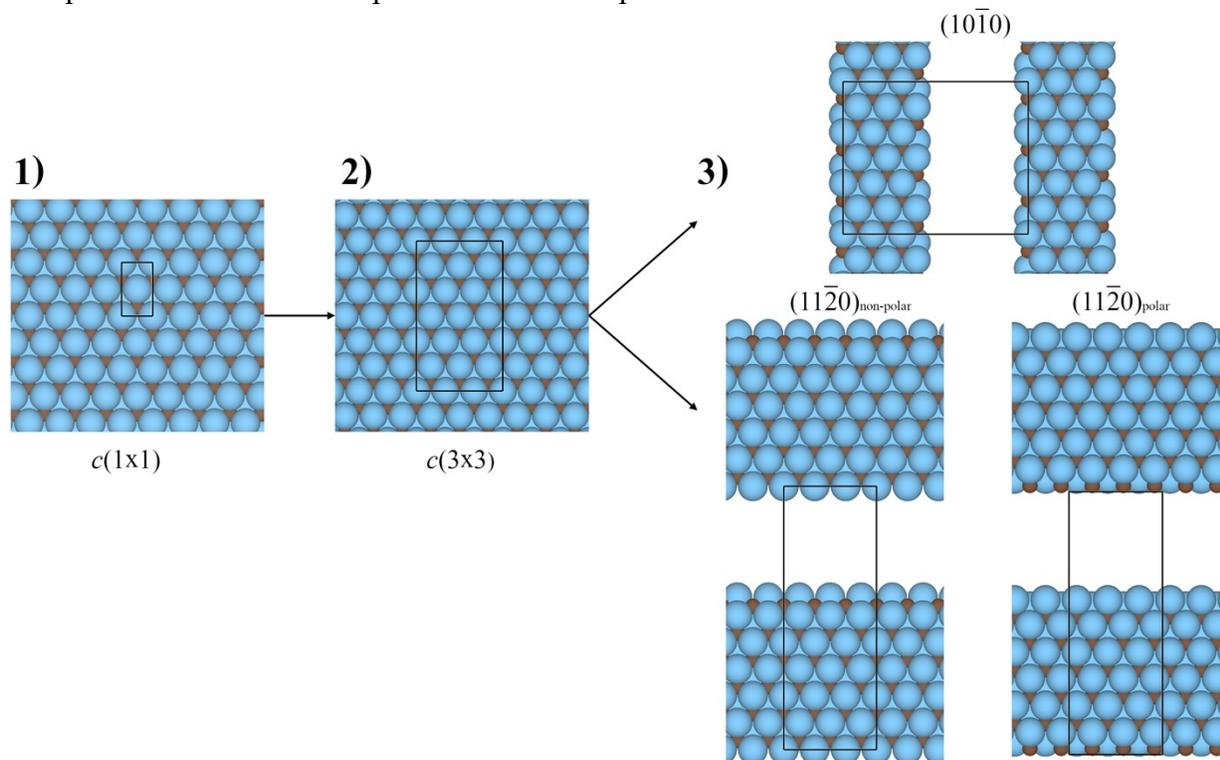
	<b>Sc<sub>2</sub>C</b>	<b>Ti<sub>2</sub>C</b>	<b>V<sub>2</sub>C</b>	<b>Cr<sub>2</sub>C</b>
<b>Intercept</b>	0.000	-0.007	-0.028	-0.051 ( <i>0.036</i> )
<b>Slope</b>	5.561	6.692	7.107	6.489 ( <i>7.657</i> )
<b>R</b>	0.999	0.999	0.999	0.999 ( <i>0.967</i> )
	<b>Y<sub>2</sub>C</b>	<b>Zr<sub>2</sub>C</b>	<b>Nb<sub>2</sub>C</b>	<b>Mo<sub>2</sub>C</b>
<b>Intercept</b>	0.005	-0.012	-0.047	-0.053
<b>Slope</b>	4.922	7.121	7.720	8.888
<b>R</b>	0.999	0.999	0.999	0.999
		<b>Hf<sub>2</sub>C</b>	<b>Ta<sub>2</sub>C</b>	<b>W<sub>2</sub>C</b>
<b>Intercept</b>	—	-0.020	-0.050	-0.028
<b>Slope</b>	—	7.465	8.727	10.905
<b>R</b>	—	0.999	0.999	0.995
	<b>Sc<sub>2</sub>CO<sub>2</sub></b>	<b>Ti<sub>2</sub>CO<sub>2</sub></b>	<b>V<sub>2</sub>CO<sub>2</sub></b>	<b>Cr<sub>2</sub>CO<sub>2</sub></b>
<b>Intercept</b>	-0.114	0.289	0.039	-0.059 ( <i>0.182</i> )
<b>Slope</b>	7.401	10.284	7.668	4.287 ( <i>10.506</i> )
<b>R</b>	0.998	0.996	0.999	0.978 ( <i>0.998</i> )
	<b>Y<sub>2</sub>CO<sub>2</sub></b>	<b>Zr<sub>2</sub>CO<sub>2</sub></b>	<b>Nb<sub>2</sub>CO<sub>2</sub></b>	<b>Mo<sub>2</sub>CO<sub>2</sub></b>
<b>Intercept</b>	0.050	0.062	0.212	-0.154
<b>Slope</b>	5.585	13.256	9.443	13.348
<b>R</b>	0.988	0.999	0.997	0.994
		<b>Hf<sub>2</sub>CO<sub>2</sub></b>	<b>Ta<sub>2</sub>CO<sub>2</sub></b>	<b>W<sub>2</sub>CO<sub>2</sub></b>
<b>Intercept</b>	—	-0.102	0.215	0.090
<b>Slope</b>	—	15.469	10.906	12.663
<b>R</b>	—	0.996	0.998	0.997

**Table S3.**  $M_2C$  and  $M_2CO_2$  MXenes calculated parameters  $d(CC)$ ,  $d(M_{top}M_{top})$ ,  $d(M_{bot}M_{bot})$ ,  $h(M_{top}M_{bot})$ , and  $h(O_{top}O_{bot})$  compared with periodic cell parameter  $a$ , and height  $h$  between metal (or oxygen for  $M_2CO_2$  MXenes) layers of the periodic systems obtained by Ontiveros *et al.*<sup>20</sup> All data correspond to the biggest calculated flake and is shown in Å. For Cr, ABA structures are shown in parenthesis and italics.

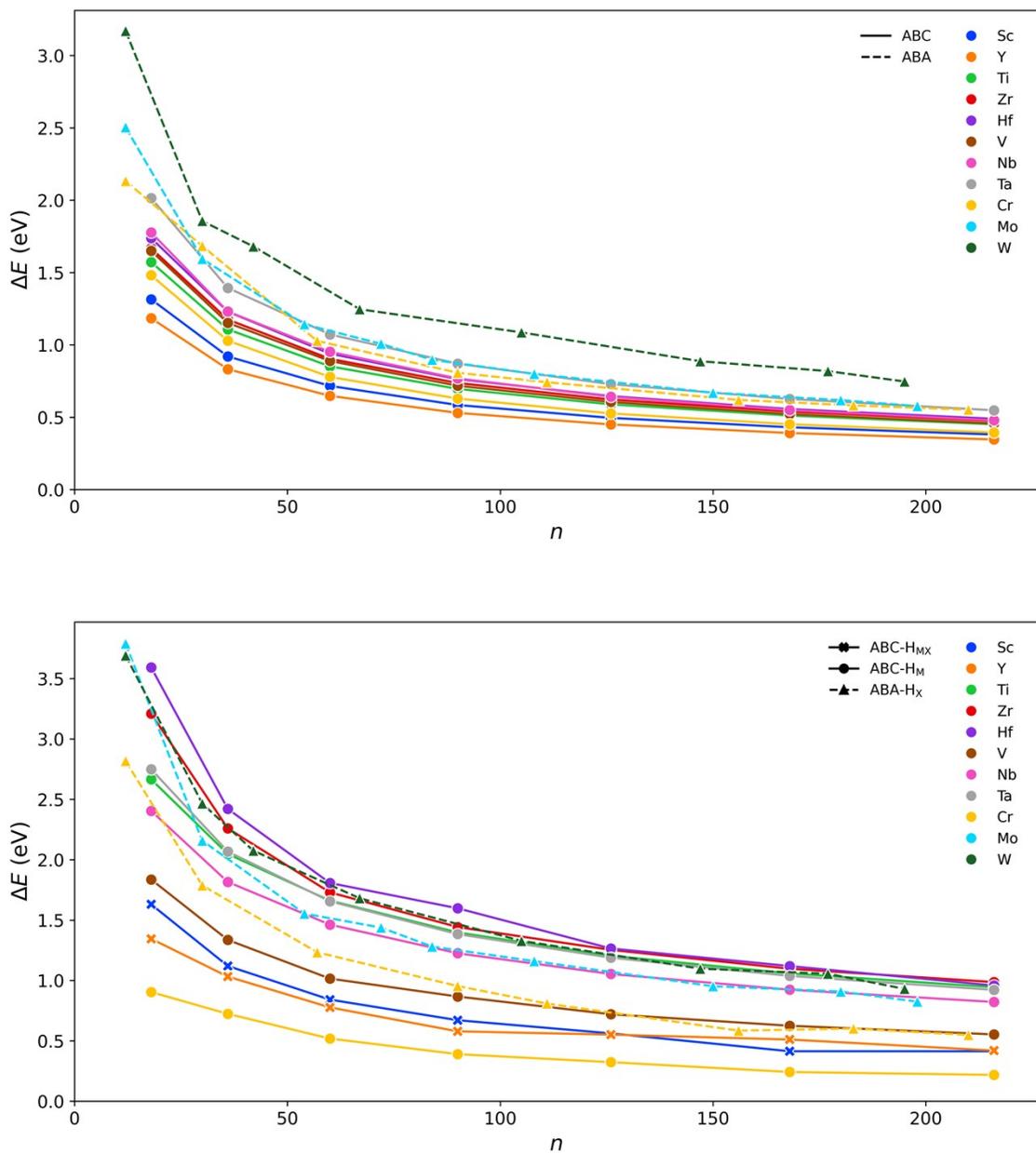
$M_2C$	$d(CC)$	$d(M_{top}M_{top})$	$d(M_{bot}M_{bot})$	Slab $a$	$h(M_{top}M_{bot})$	Slab $h$
Sc <sub>2</sub> C	3.293	3.287	3.287	3.328	2.441	2.409
Y <sub>2</sub> C	3.527	3.533	3.533	3.588	2.700	2.661
Ti <sub>2</sub> C	3.036	3.014	2.997	3.075	2.305	2.250
Zr <sub>2</sub> C	3.267	3.244	3.244	3.274	2.542	2.544
Hf <sub>2</sub> C	3.220	3.194	3.194	3.212	2.553	2.544
V <sub>2</sub> C	2.906	2.862	2.863	2.900	2.165	2.174
Nb <sub>2</sub> C	3.133	3.095	3.095	3.130	2.368	2.388
Ta <sub>2</sub> C	3.099	3.059	3.059	3.084	2.428	2.436
Cr <sub>2</sub> C	2.860 ( <i>2.673</i> )	2.794 ( <i>2.623</i> )	2.792 ( <i>2.621</i> )	2.807 ( <i>2.638</i> )	2.029 ( <i>2.478</i> )	2.112 ( <i>2.521</i> )
Mo <sub>2</sub> C	2.852	2.836	2.840	2.850	2.621	2.742
W <sub>2</sub> C	2.866	2.833	2.833	2.847	2.741	2.788

$M_2CO_2$	$d(CC)$	$d(M_{top}M_{top})$	$d(M_{bot}M_{bot})$	Slab $a$	$h(O_{top}O_{bot})$	Slab $h$
Sc <sub>2</sub> CO <sub>2</sub>	3.458	3.327	3.488	3.405	3.653	3.892
Y <sub>2</sub> CO <sub>2</sub>	3.721	3.614	3.789	3.702	3.653	4.067
Ti <sub>2</sub> CO <sub>2</sub>	3.058	3.045	3.046	3.011	4.449	4.467
Zr <sub>2</sub> CO <sub>2</sub>	3.316	3.322	3.322	3.303	4.628	4.697
Hf <sub>2</sub> CO <sub>2</sub>	3.278	3.300	3.291	3.372	4.533	4.379
V <sub>2</sub> CO <sub>2</sub>	2.919	2.913	2.913	2.875	4.476	4.456
Nb <sub>2</sub> CO <sub>2</sub>	3.143	3.140	3.141	3.132	4.666	4.722
Ta <sub>2</sub> CO <sub>2</sub>	3.131	3.134	3.134	3.113	4.681	4.700
Cr <sub>2</sub> CO <sub>2</sub>	2.859 ( <i>2.771</i> )	2.850 ( <i>2.658</i> )	2.852 ( <i>2.659</i> )	2.860 ( <i>2.650</i> )	4.494 ( <i>4.872</i> )	4.477 ( <i>4.933</i> )
Mo <sub>2</sub> CO <sub>2</sub>	2.865	2.867	2.872	2.870	5.123	5.274
W <sub>2</sub> CO <sub>2</sub>	2.895	2.877	2.876	2.879	5.070	5.297

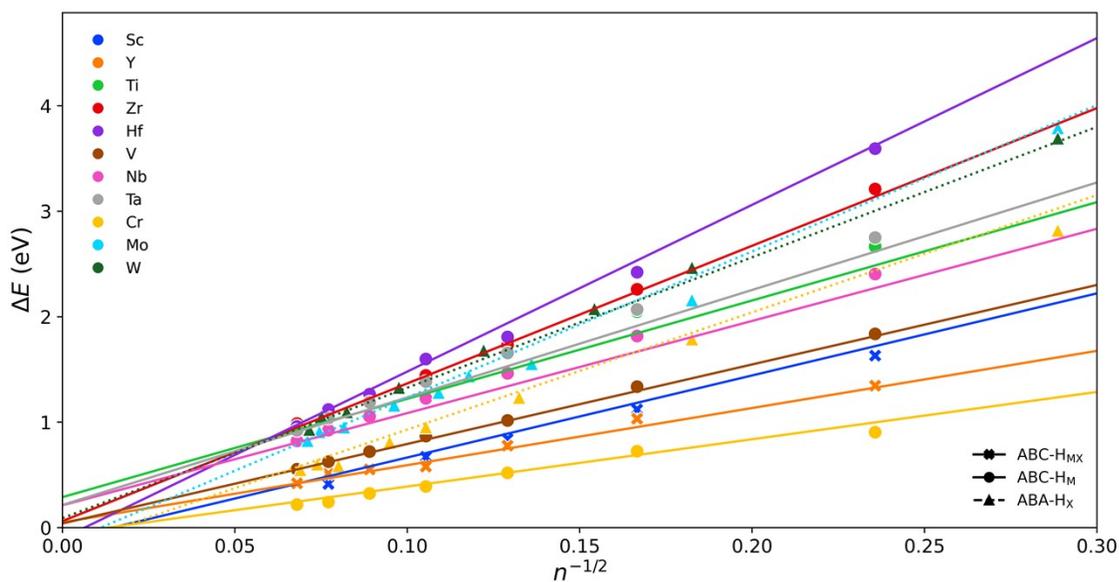
**Figure S1.** Cleavage energy calculation workflow along the three main steps: 1) Optimisation of the orthorhombic  $c(1\times 1)$  cell. 2) Replication of the  $c(1\times 1)$  cell to obtain a  $c(3\times 3)$  supercell. 3) Addition of 10 Å vacuum to the  $c(3\times 3)$  supercell in the direction normal to the desired surface. The periodic unit cells are represented from a top view and indicated with black lines.



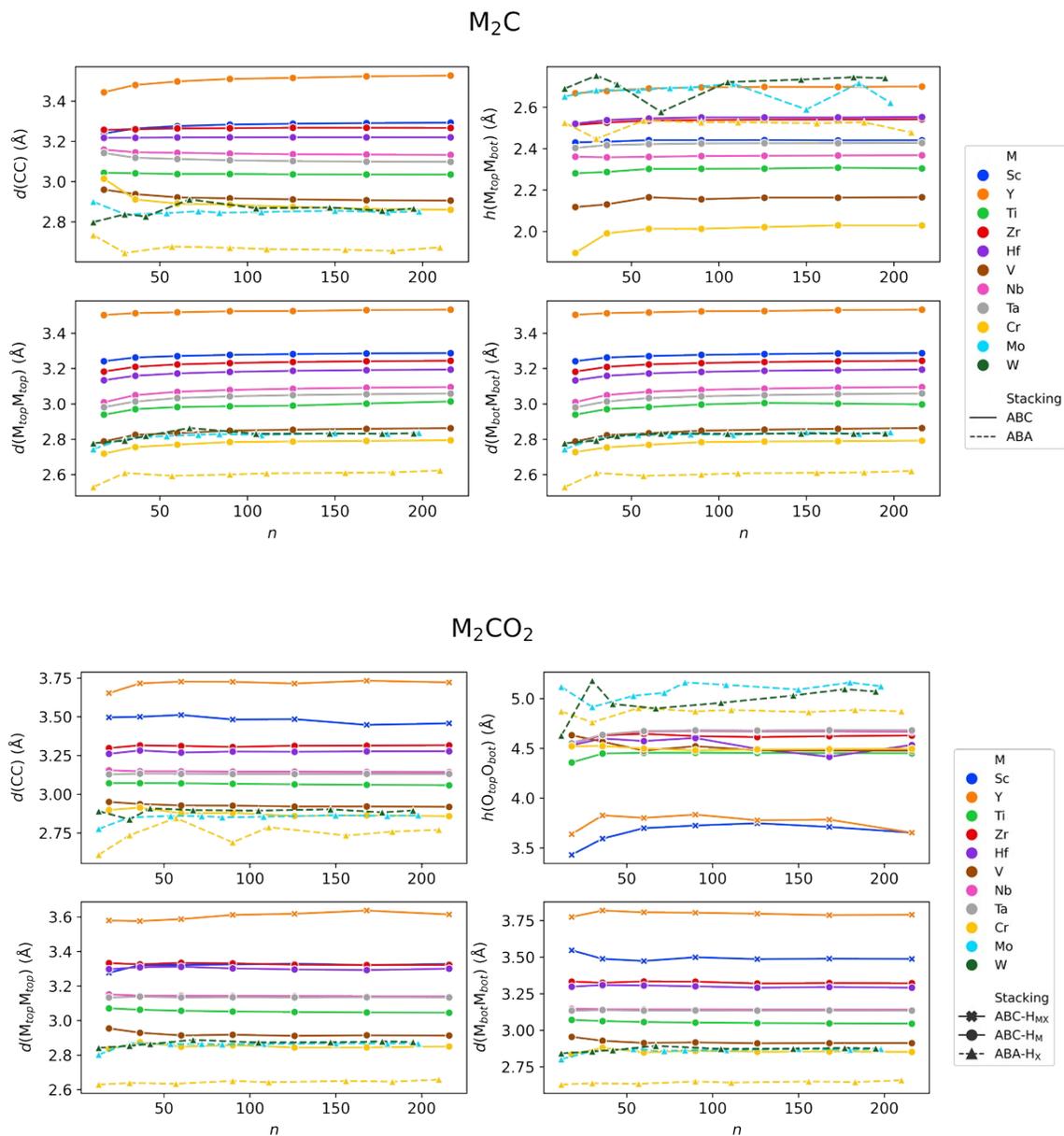
**Figure S2.** Differences in energy per stoichiometric unit between flakes and periodic slabs of  $M_2C$  (top) and  $M_2CO_2$  (bottom) MXenes vs. number of stoichiometric units of flakes. Metals are differentiated by colors, and Stacking-Site is shown in solid lines and crosses for ABC- $H_{MX}$ , solid lines and circles for ABC- $H_M$ , and dotted lines and triangles for ABA- $H_X$ . Each MXene is referred to its own slab surface.

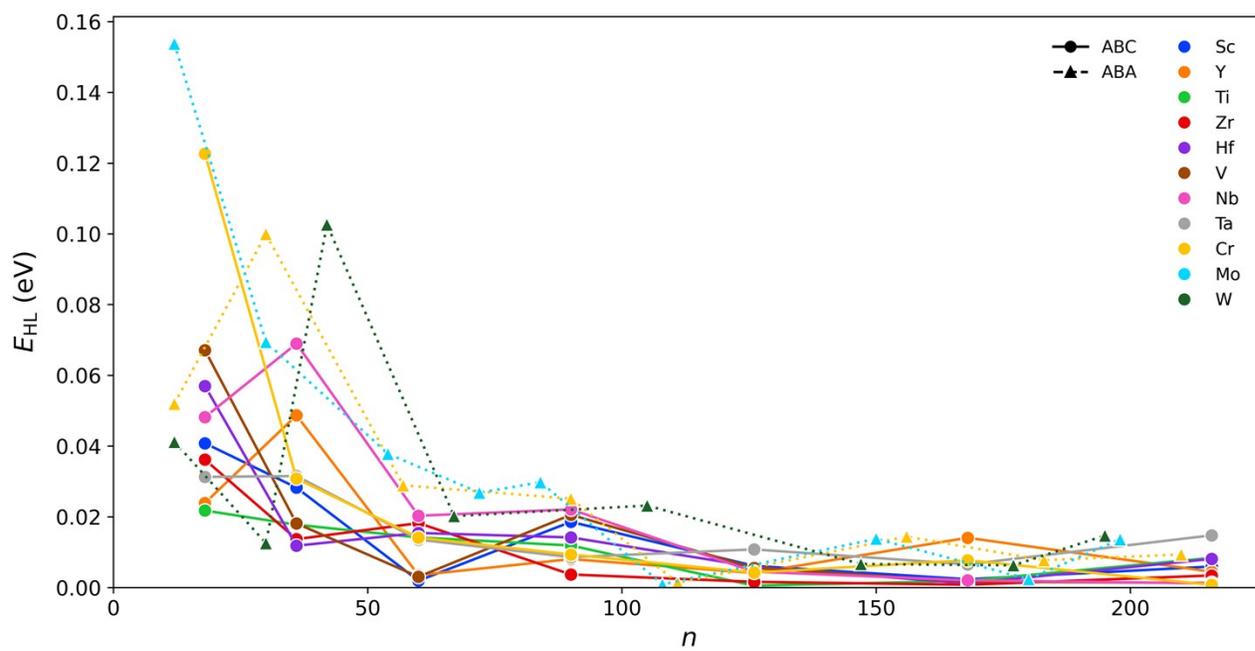


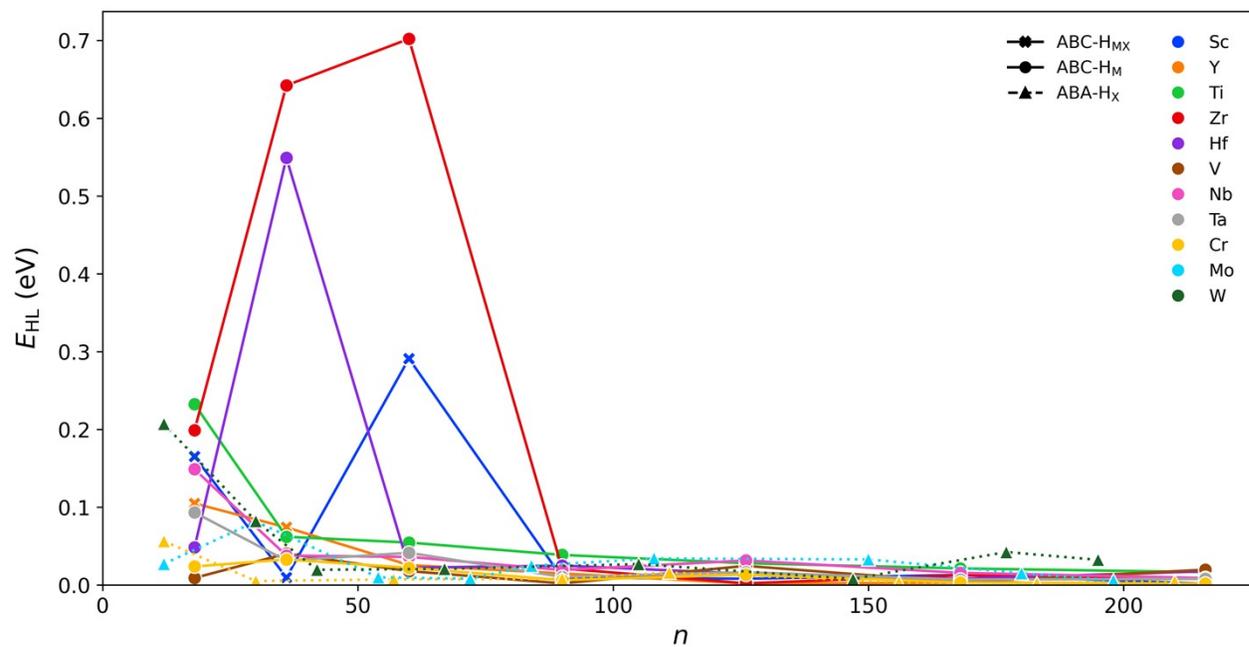
**Figure S3.** Differences in energy per stoichiometric unit between flakes and periodic slabs of  $M_2CO_2$  MXenes vs. number of stoichiometric units of flakes to the power of  $-1/2$ . Metals are differentiated by colors, and Stacking-Site is shown in solid lines and crosses for ABC- $H_{MX}$ , solid lines and circles for ABC- $H_M$ , and dotted lines and triangles for ABA- $H_X$ . Each MXene is referred to its own slab surface.



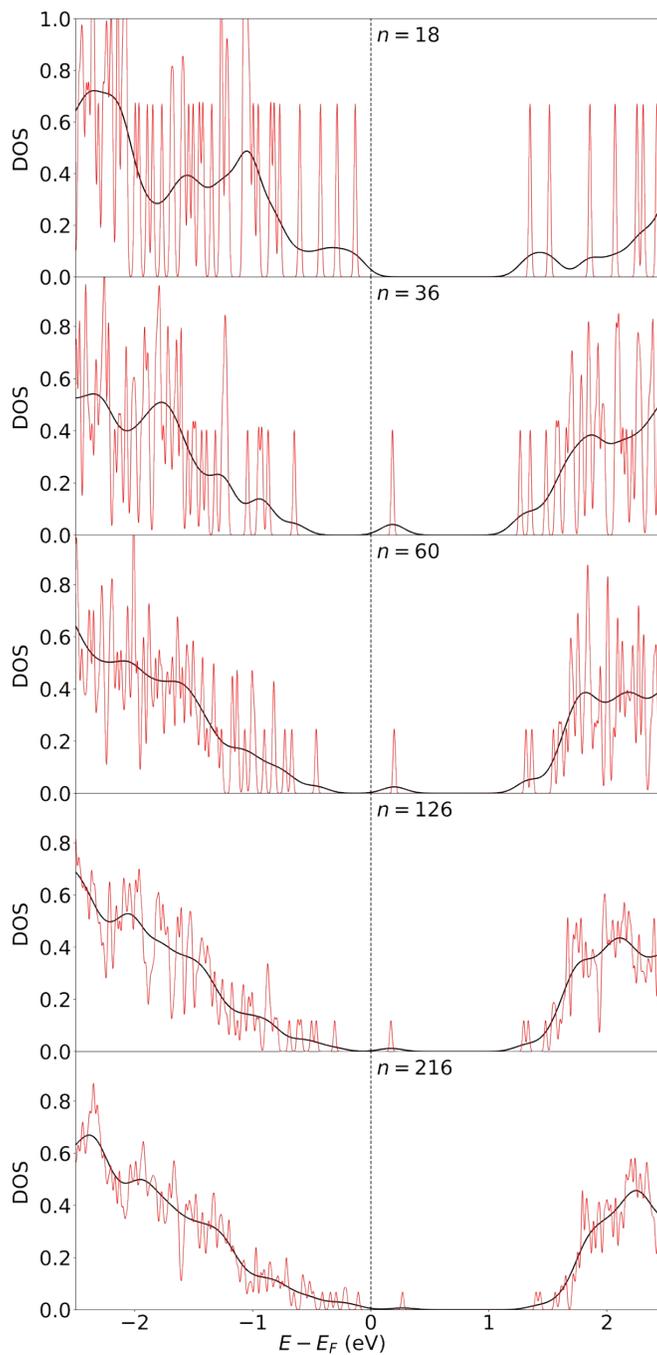
**Figure S4.** Evolution of  $d(\text{CC})$ ,  $d(\text{M}_{\text{top}}\text{M}_{\text{top}})$ ,  $d(\text{M}_{\text{bot}}\text{M}_{\text{bot}})$ , and  $h(\text{M}_{\text{top}}\text{M}_{\text{bot}})$  structural parameters on  $\text{M}_2\text{C}$  (top) and  $d(\text{CC})$ ,  $d(\text{M}_{\text{top}}\text{M}_{\text{top}})$ ,  $d(\text{M}_{\text{bot}}\text{M}_{\text{bot}})$ , and  $h(\text{O}_{\text{top}}\text{O}_{\text{bot}})$  on  $\text{M}_2\text{CO}_2$  (bottom) MXenes for different flake sizes. All values are given in Å.



**Figure S5.** Evolution of the HOMO-LUMO gap,  $E_{HL}$ , for  $M_2C$  MXenes, calculated at PBE level.

**Figure S6.** Evolution of the HOMO-LUMO gap,  $E_{HL}$ , for  $M_2CO_2$  MXenes calculated at PBE level.

**Figure S7.** PBE0 total DOS of  $(\text{Ti}_2\text{CO}_2)_{18}$ ,  $(\text{Ti}_2\text{CO}_2)_{36}$ ,  $(\text{Ti}_2\text{CO}_2)_{60}$ ,  $(\text{Ti}_2\text{CO}_2)_{126}$ , and  $(\text{Ti}_2\text{CO}_2)_{216}$  flakes represented with 0.1 and 0.01 eV smearing in black and red lines, respectively. All densities within the displayed energy window have been normalized to 1. Fermi level has been set to 0 on the total DOS and it is shown in black.



**Figure S8.** Corrected VBM and CBM energetic alignment for the largest  $\text{Sc}_2\text{CO}_2$ ,  $\text{Y}_2\text{CO}_2$ ,  $\text{Ti}_2\text{CO}_2$ ,  $\text{Zr}_2\text{CO}_2$  and  $\text{Hf}_2\text{CO}_2$  flakes. VBM energy corresponds to the  $E_{\text{F}}^{\text{c}}$  energy of each flake while CBM refers to  $E_{\text{F}}^{\text{c}} + E_{\text{g}}^{\text{c}}$  (see text). Hydrogen reduction,  $\text{H}^+/\text{H}_2$ , and water oxidation,  $\text{H}_2\text{O}/\text{O}_2$ , potentials are defined in dashed lines.

