

Supporting information

Bifunctional Group Distribution and Gradient Structure Resulting from the Anionic Copolymerization of Styrene and Vinylbenzyl N,N-Allylmethylamine

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Figures S1-S5

Figure S1



Figure S1. The GC spectrum of VBAMA isomers mixture synthesized in this work

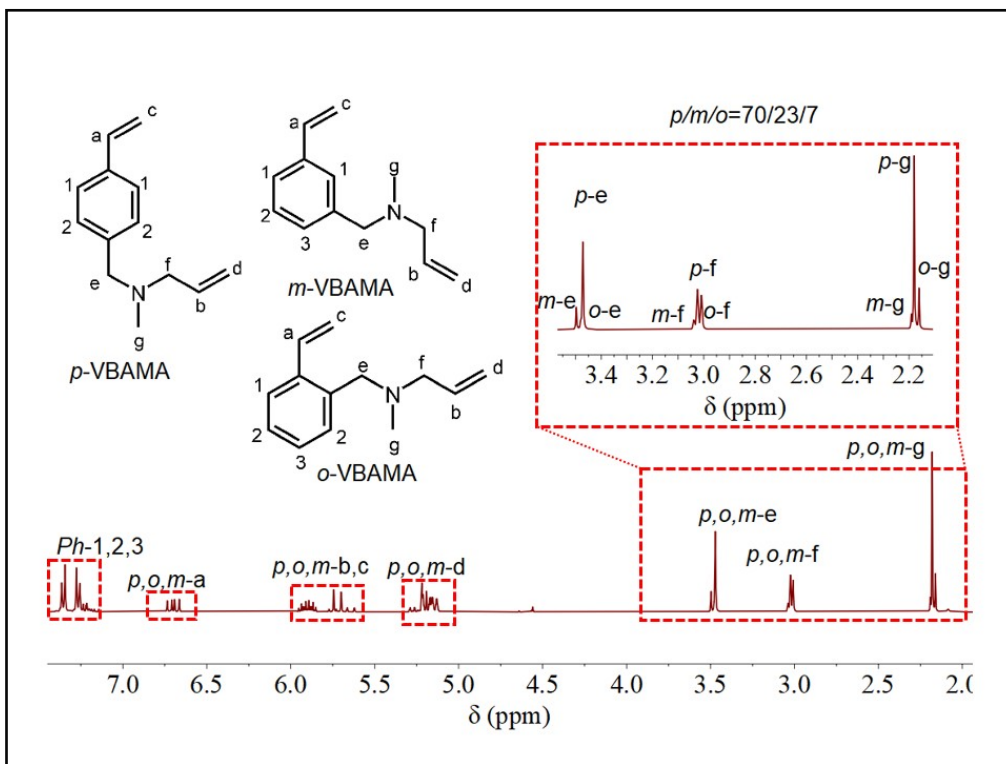


Figure S2. The ^1H NMR spectrum of VBAMA isomers in CDCl_3

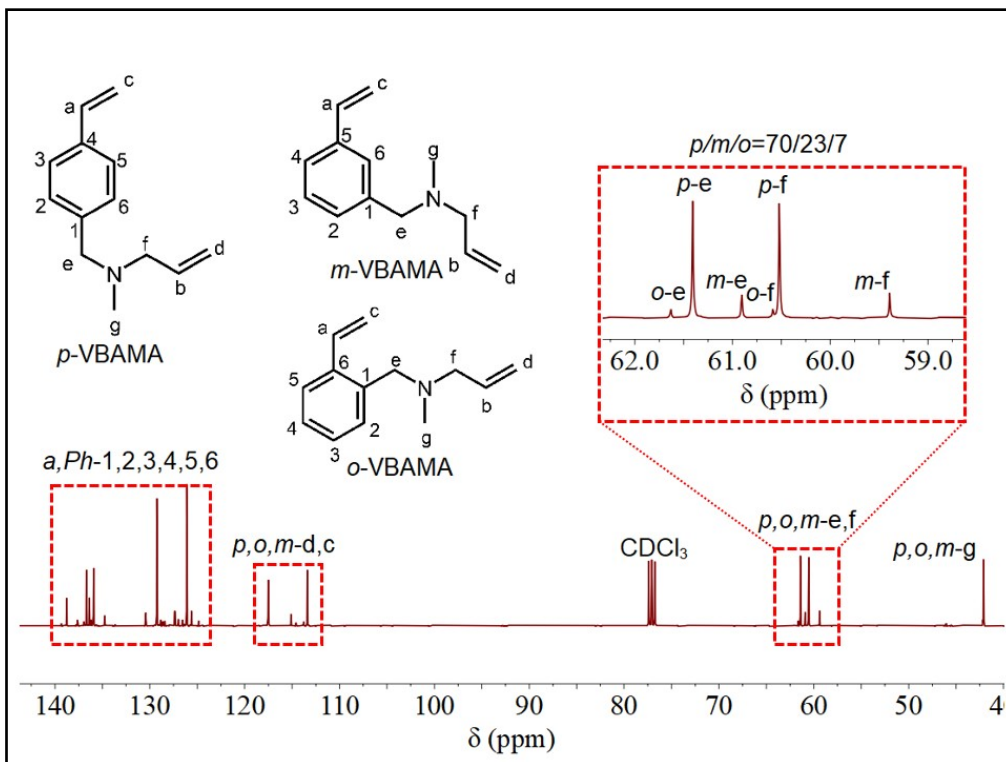


Figure S3. The ^{13}C NMR spectrum of VBAMA isomers in CDCl_3

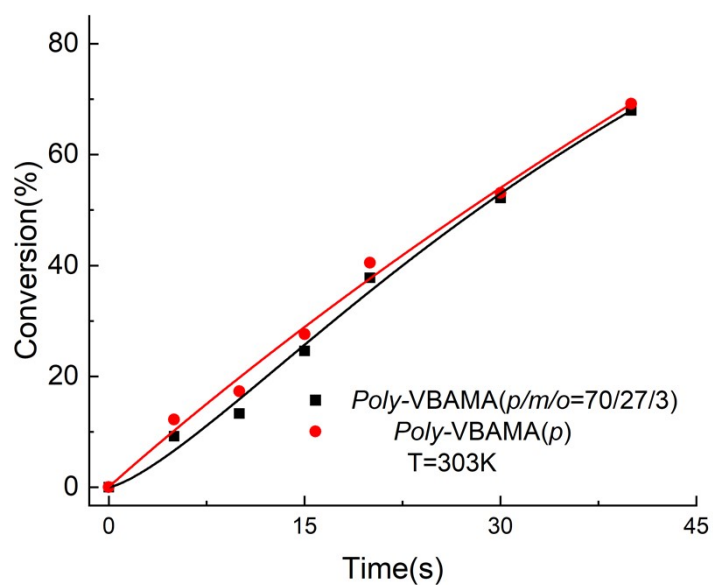


Figure S4. The homopolymerization conversion of VBAMA isomers (red line is para-VBAMA and black line is VBAMA mixture isomers) at 303K

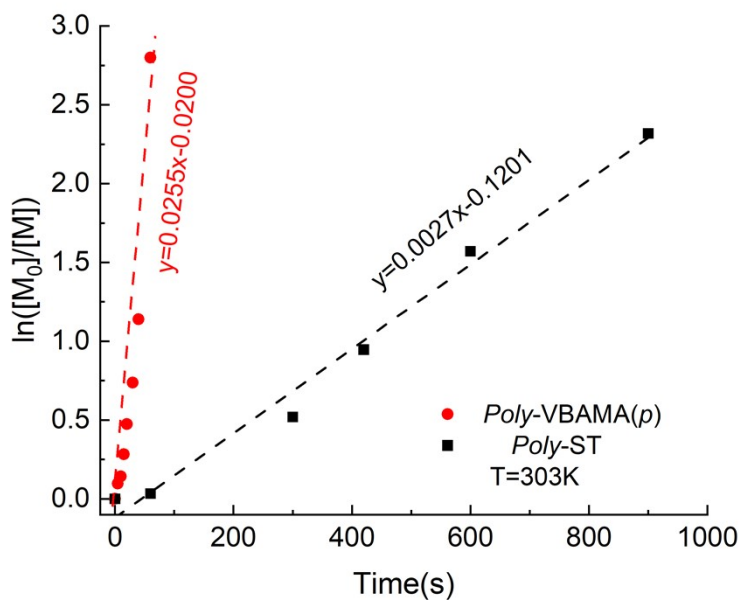


Figure S5. The first-order homopolymerization reaction curves (red line is VBAMA mixture isomers and black line is ST) at 303K

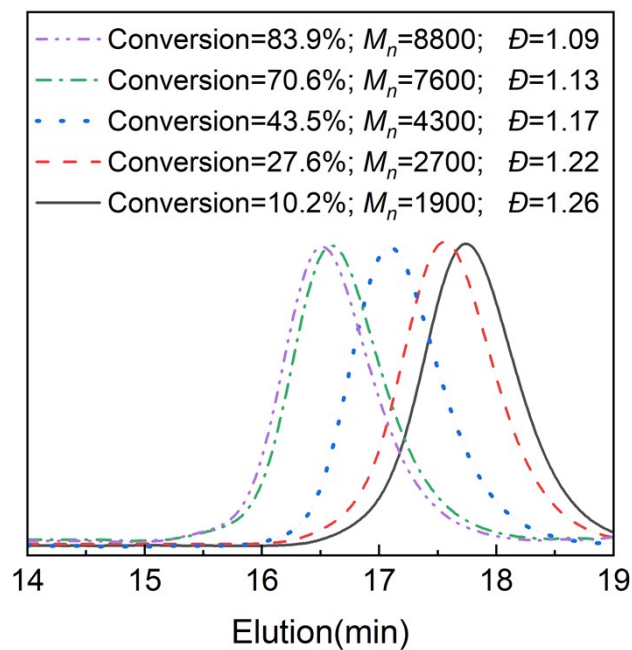


Figure S6. The SEC curves tracking the copolymerization process (ST/VBAMA=4/1 at 303K)

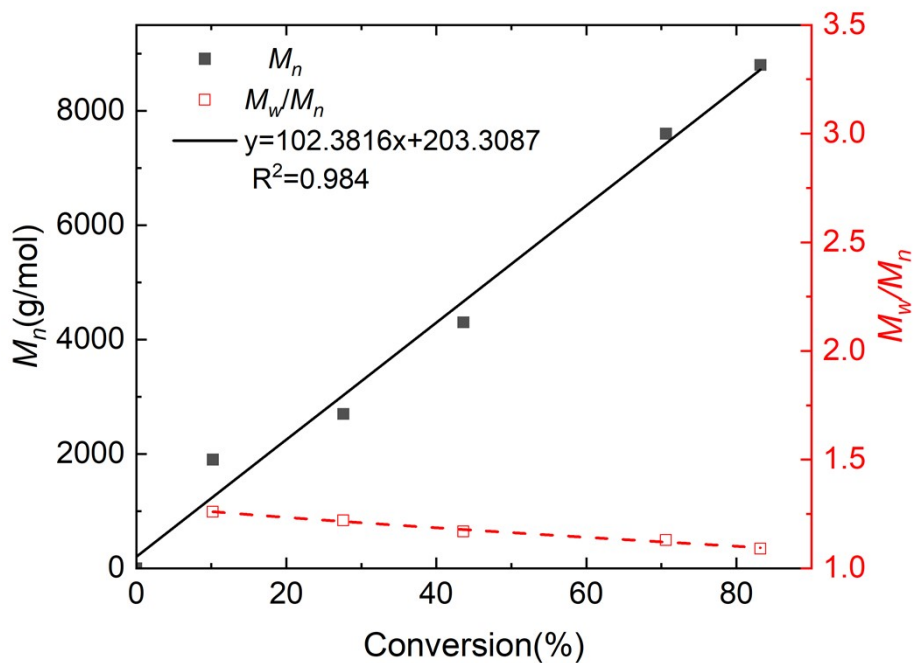


Figure S7. The copolymer M_n vs conversion (black line) and the M_w/M_n vs conversion (red line) (ST/VBAMA=4/1 at 303K)

Table S1. The key parameters used to determine the reactivity ratios of VBAMA and ST

Comonomers		$T(^{\circ}\text{C})$	$f(\text{M}_1)$	$f(\text{M}_2)$	Conversion ^a	$F(\text{M}_1)/F(\text{M}_2)$ ^b
M_1	M_2					
VBAMA	ST	30	80.0%	20.0%	9.65%	87.25%/12.75%
VBAMA	ST	30	75.0%	25.0%	8.62%	83.50%/16.50%
VBAMA	ST	30	66.7%	33.3%	7.38%	79.43%/20.57%
VBAMA	ST	30	50.0%	50.0%	2.53%	64.35%/35.65%
VBAMA	ST	30	33.3%	66.7%	6.73%	55.77%/44.23%
VBAMA	ST	30	25.0%	75.0%	13.22%	44.99%/55.01%
VBAMA	ST	30	20.0%	80.0%	10.26%	38.80%/61.20%

^a determined by weighting method;

^b determined by HNMR data