

**Bifunctional Biomass-Derived Additive for Constructing a Hybrid SEI Layer to  
Enhance the Cycling Stability of Aqueous Zinc-Ion Batteries Under High  
Current Densities**

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Figure S1: Electrolytes prepared with different concentrations of FA.

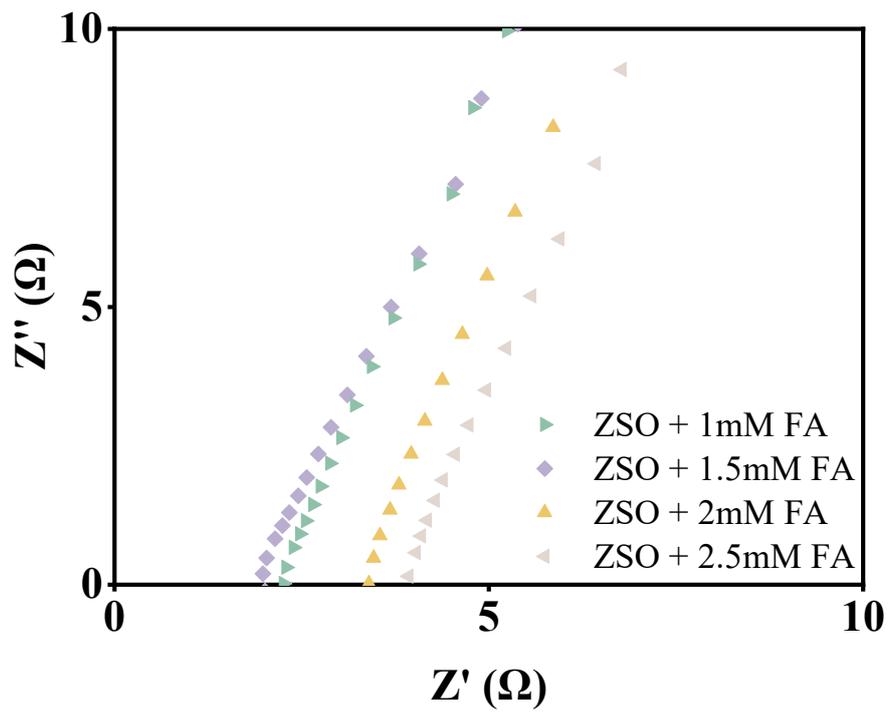


Figure S2: Nyquist plots of electrolytes with different concentrations of FA.

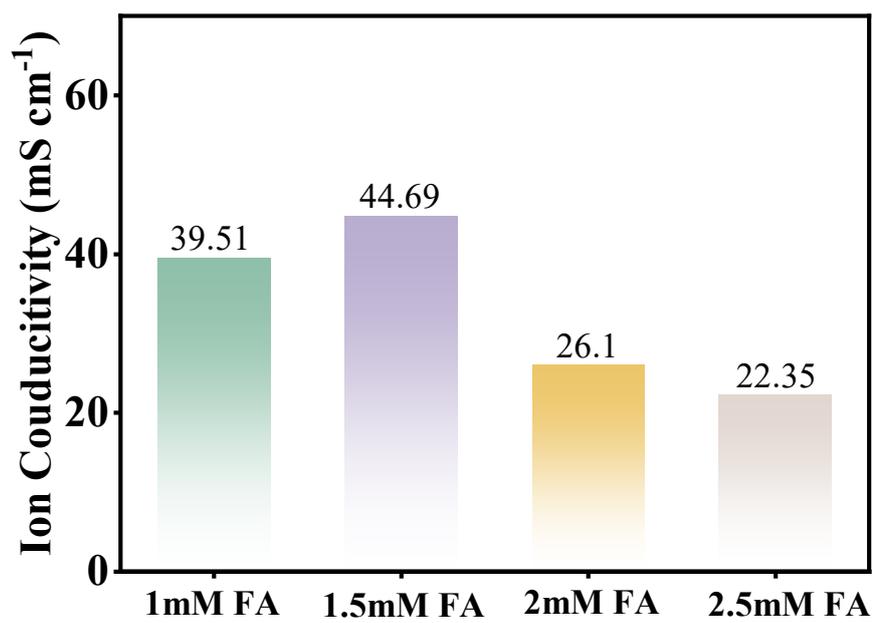


Figure S3: Corresponding conductivity.

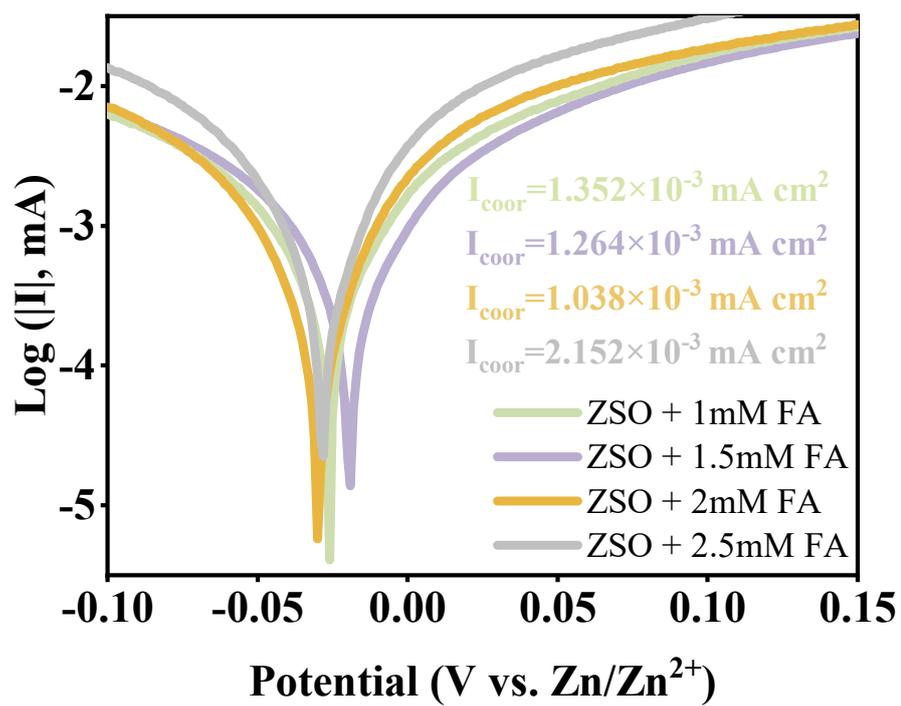


Figure S4: Tafel plots of electrolytes with different FA concentrations.

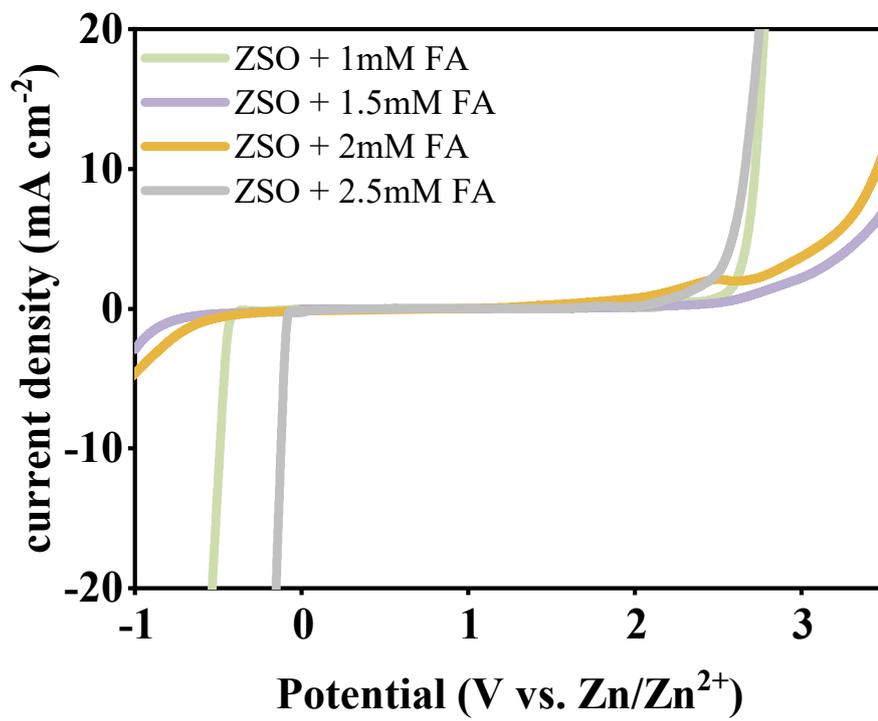


Figure S5: LSV curves of electrolytes with different FA concentrations.

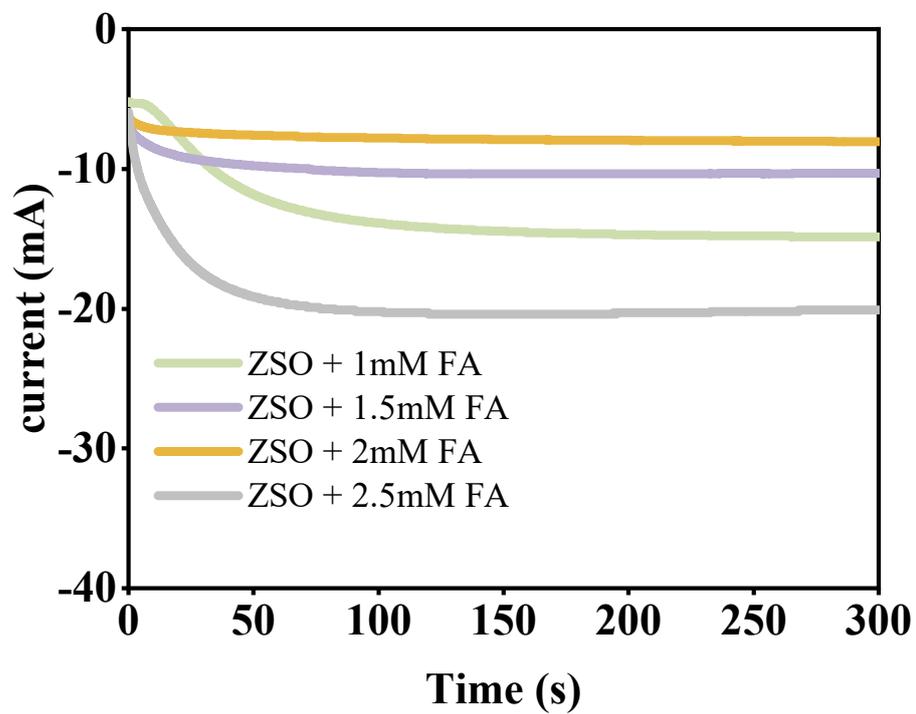


Figure S6: CA curves of electrolytes with different FA concentrations.

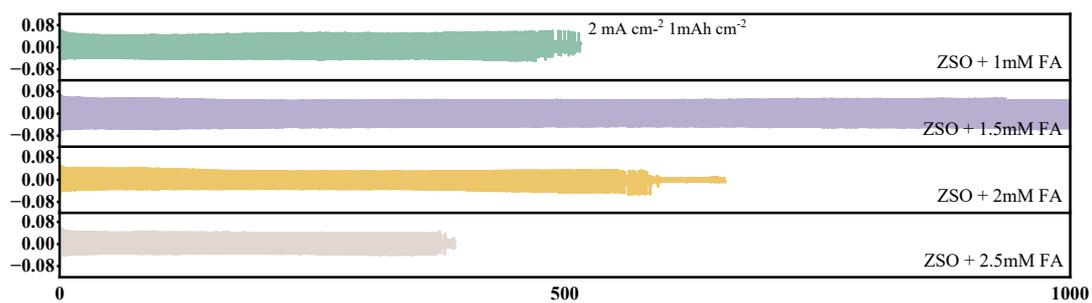


Figure S7: Cycling performance tests of Zn//Zn symmetric cells assembled with electrolytes of different FA concentrations.

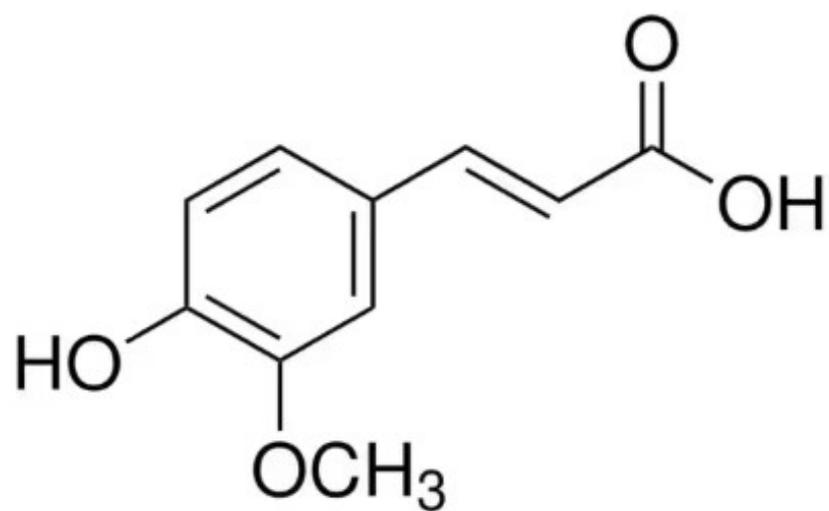


Figure S8: Chemical structure of FA.

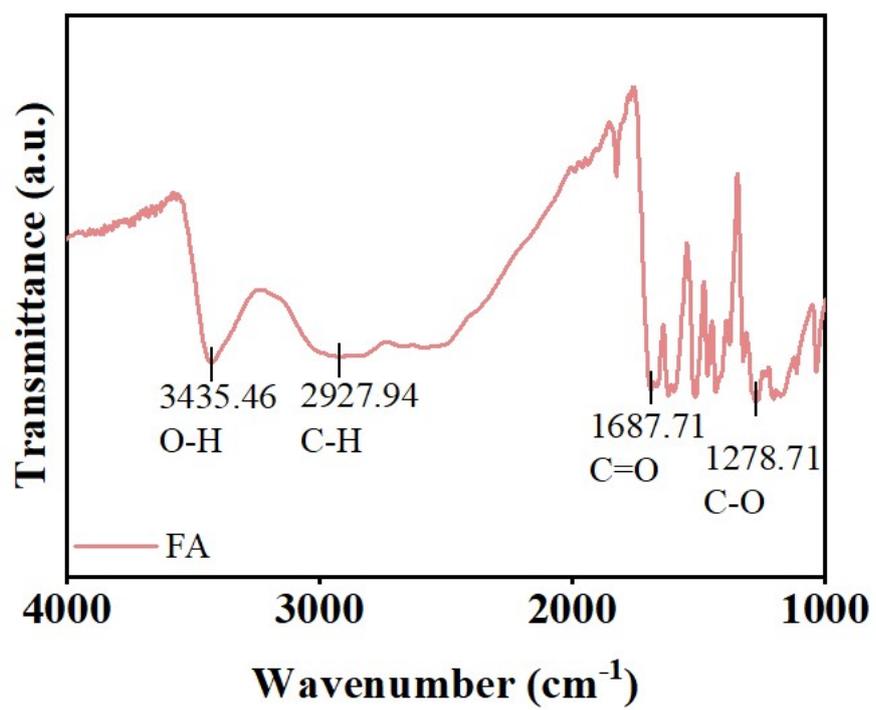


Figure S9: FTIR spectrum of FA.

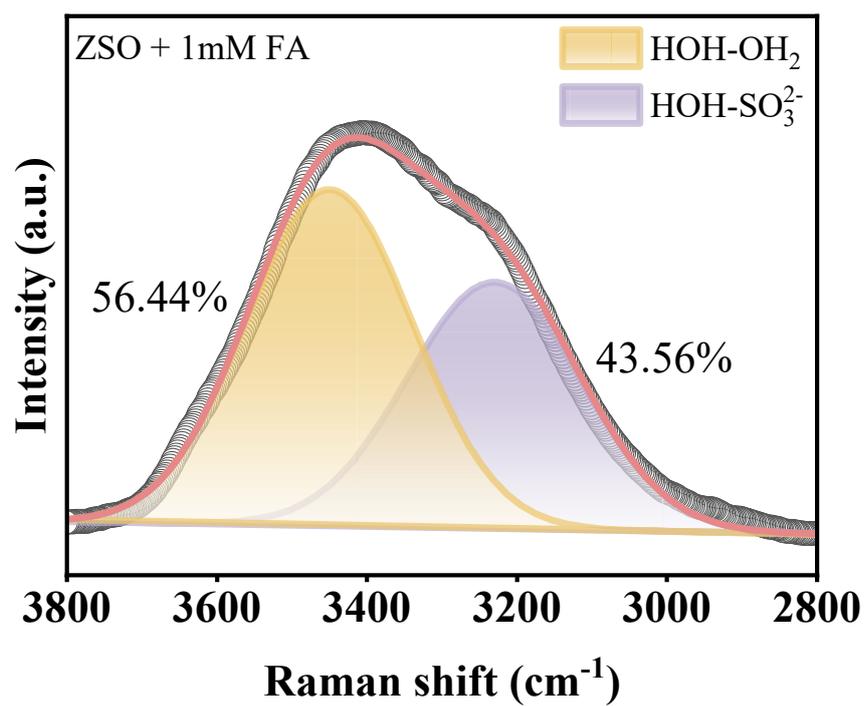


Figure S10: Raman Spectrum of the ZSO + 1 mM FA Electrolyte.

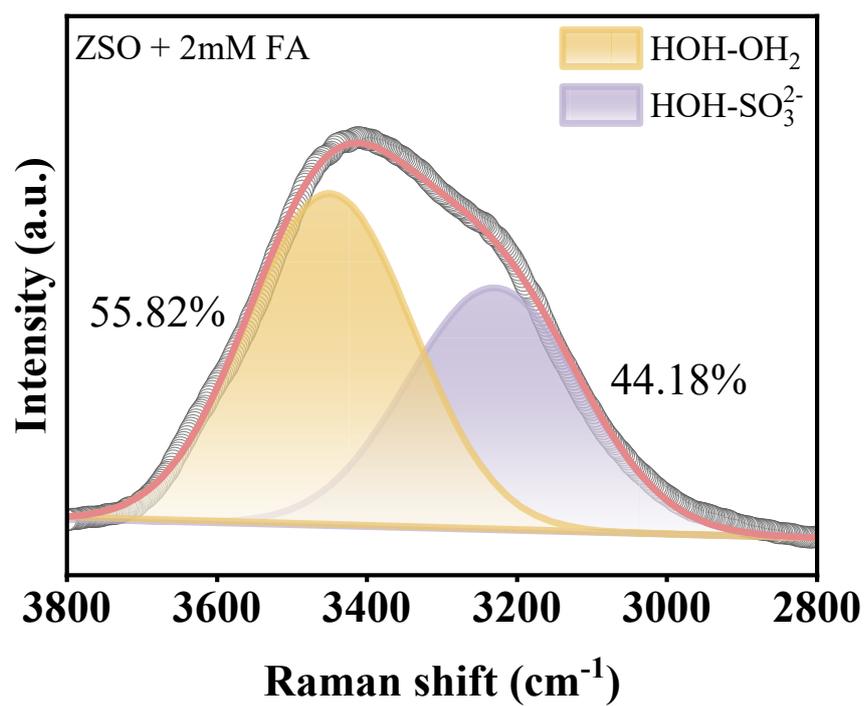


Figure S11: Raman Spectrum of the ZSO + 2 mM FA Electrolyte.

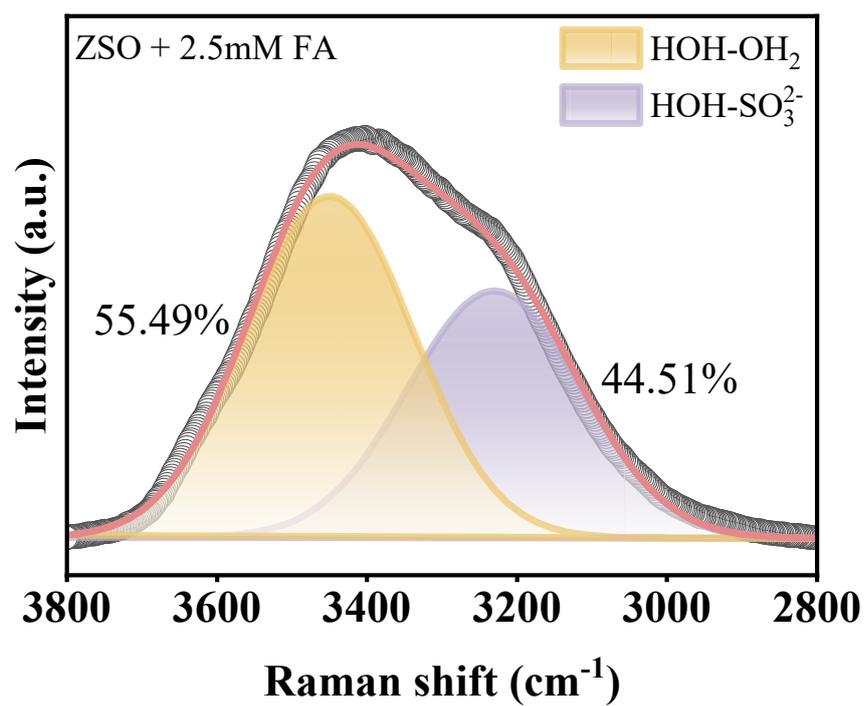


Figure S12: Raman Spectrum of the ZSO + 2.5 mM FA Electrolyte.

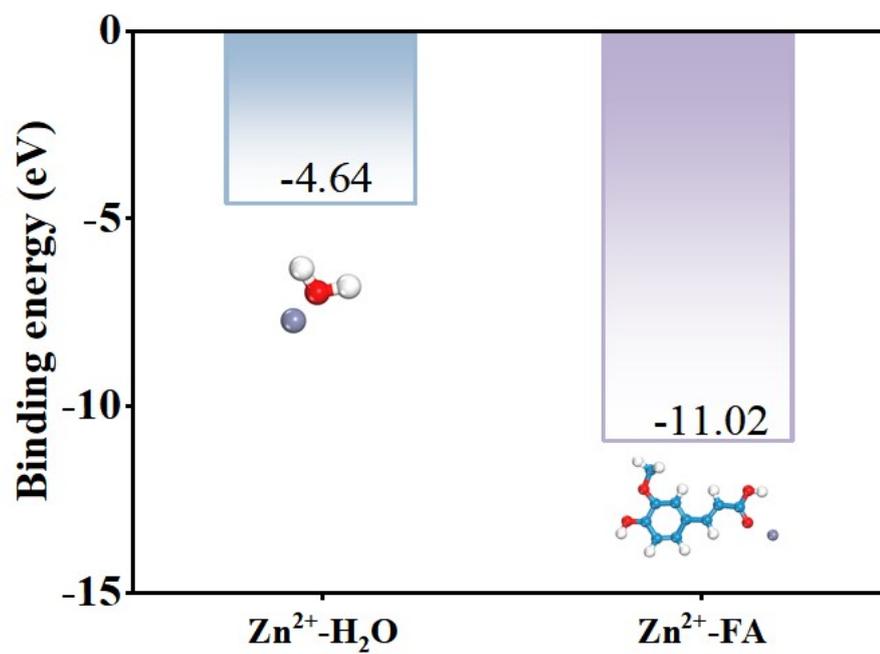


Figure S13: Binding energies of  $H_2O$ , and  $H_2O$ -FA molecules.

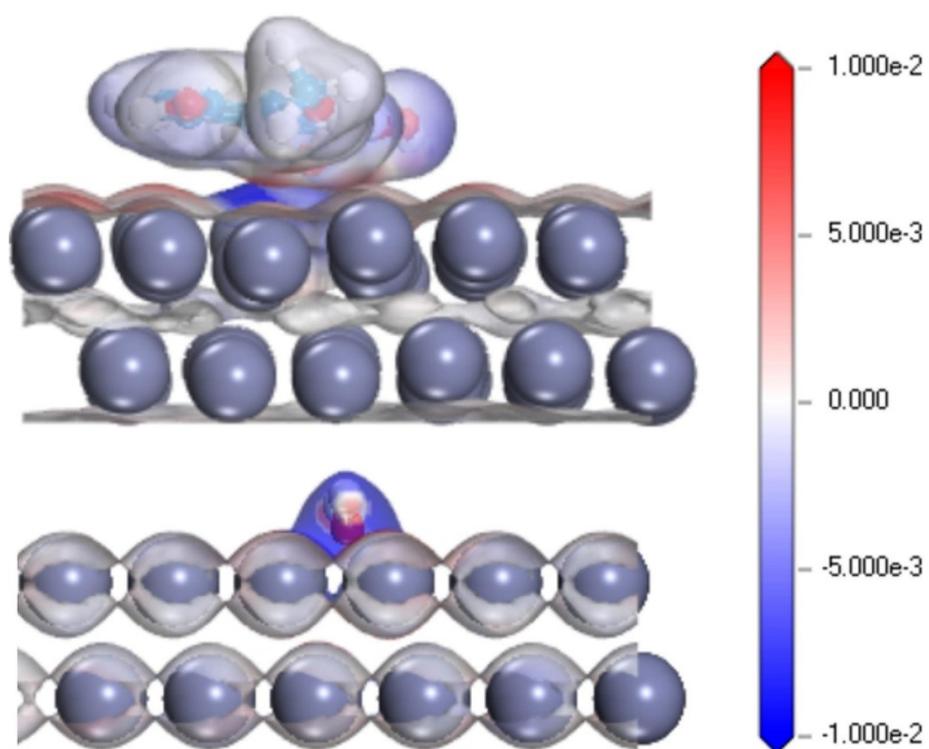


Figure S14: Charge density difference of Zn foil containing FA and H<sub>2</sub>O molecules.

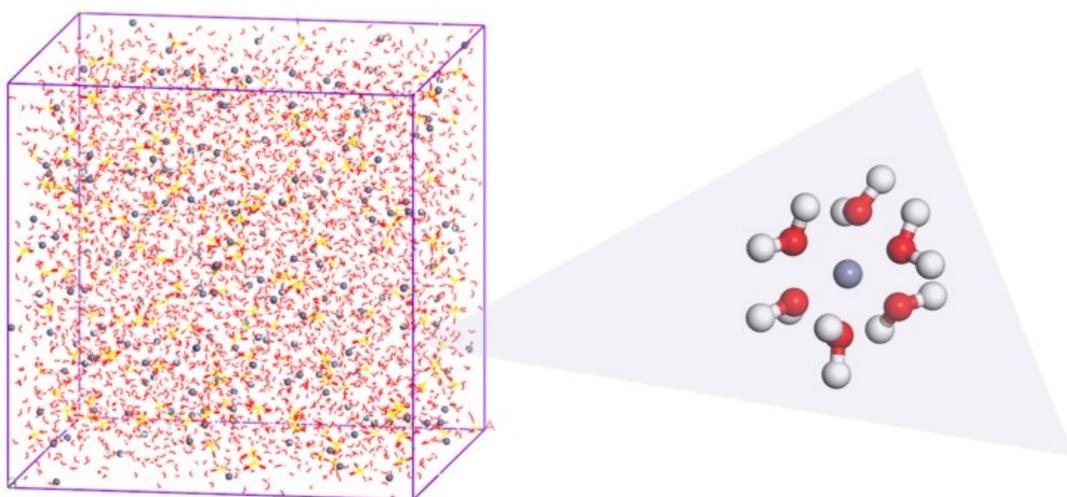


Figure S15: MD snapshots and schematic diagrams of the solvation structure of ZSO.

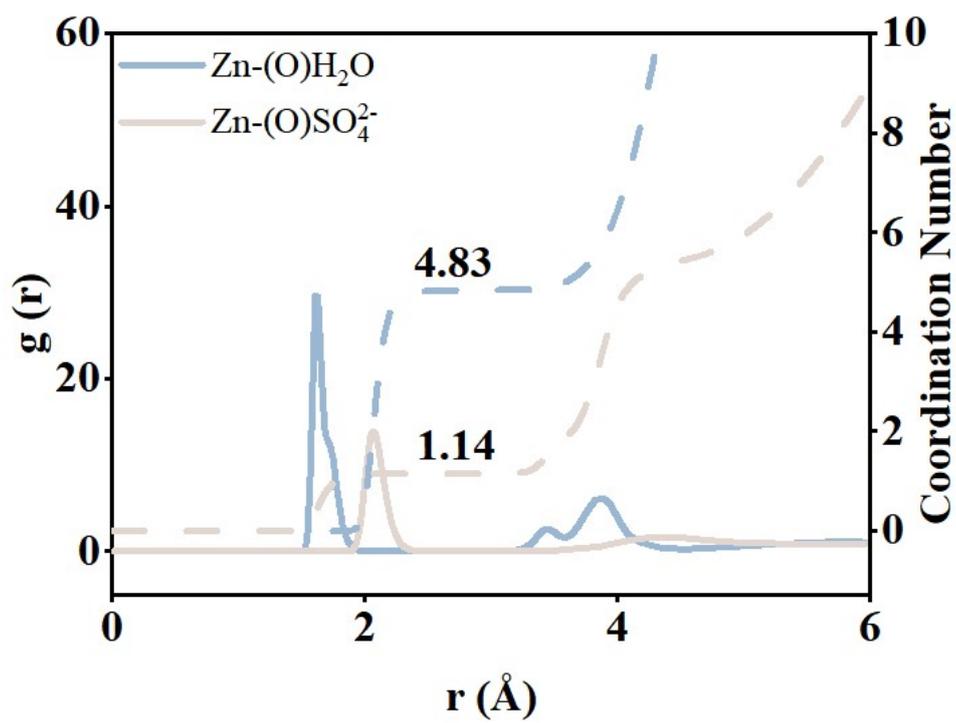
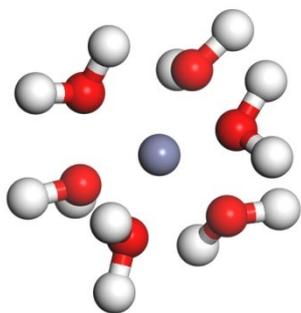
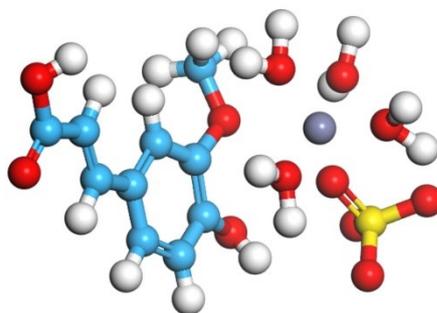


Fig. S16: Radial distribution function of ZSO.



$$\Delta E_{solv} = -0.204829 \text{ Ha}$$



$$\Delta E_{solv} = -2.424139 \text{ Ha}$$

Figure S17: Comparison of solvation energies

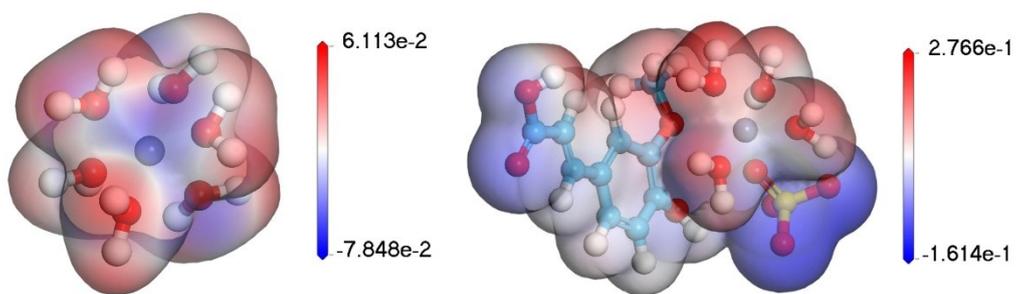


Figure S18: Comparison of electrostatic potential energies

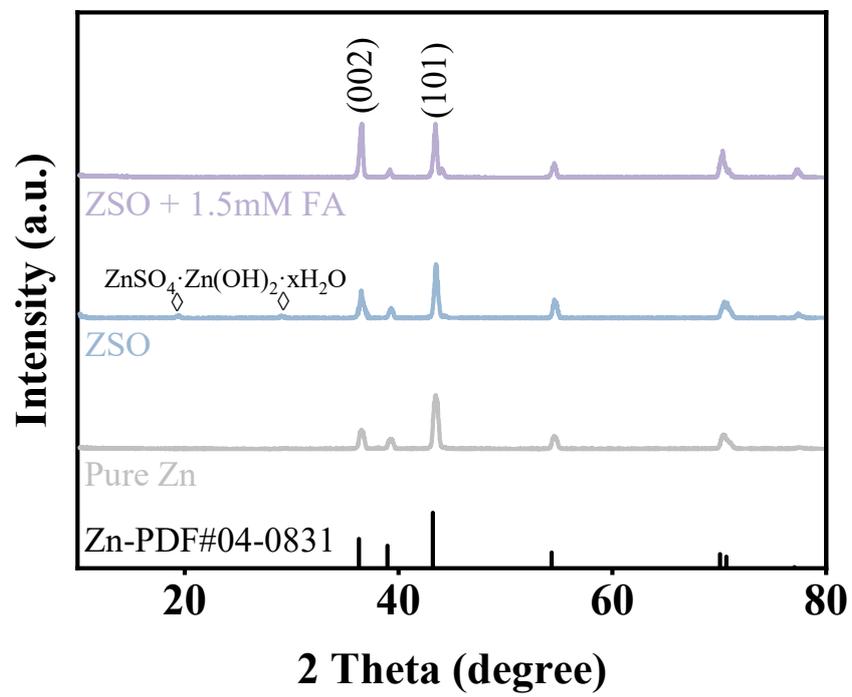


Figure S19: XRD patterns of zinc foils after 7-day immersion.



Figure S20: Thickness comparison of Zn symmetric cells before and after cycling.

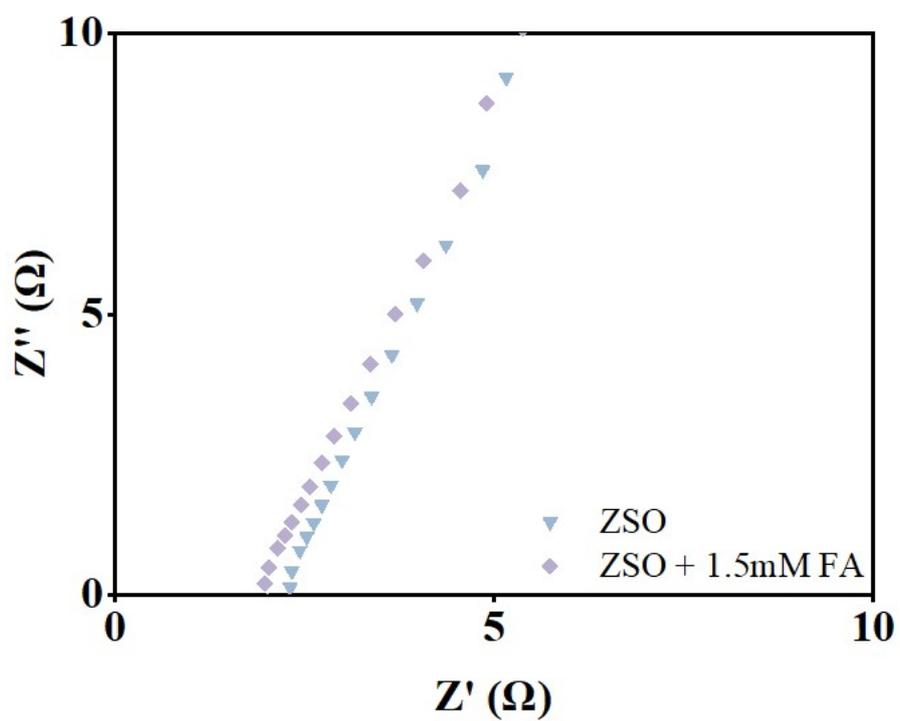


Figure S21: EIS plots of ZSO and ZSO + 1.5 mM FA electrolytes.

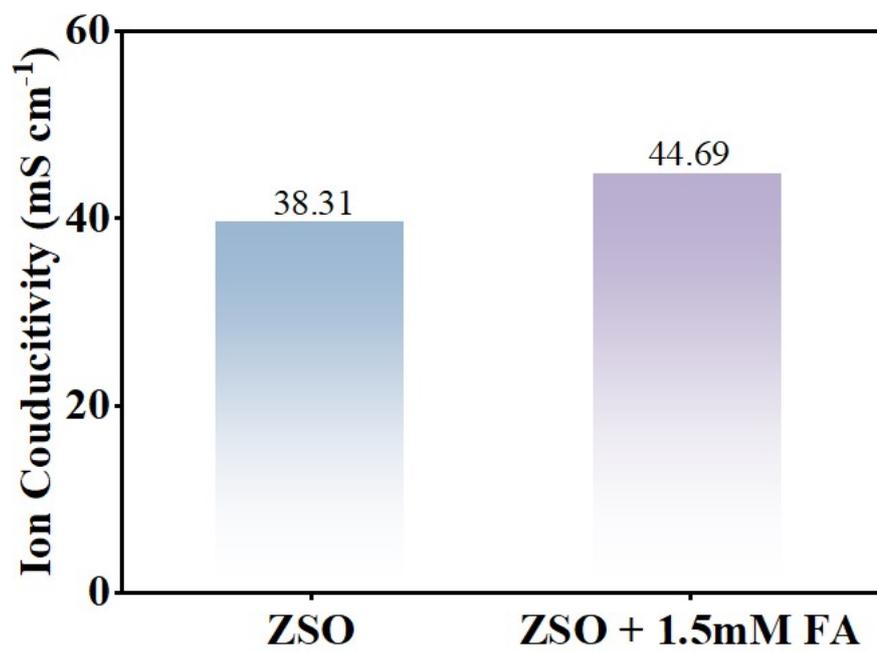


Figure S22: Calculated ionic conductivity.

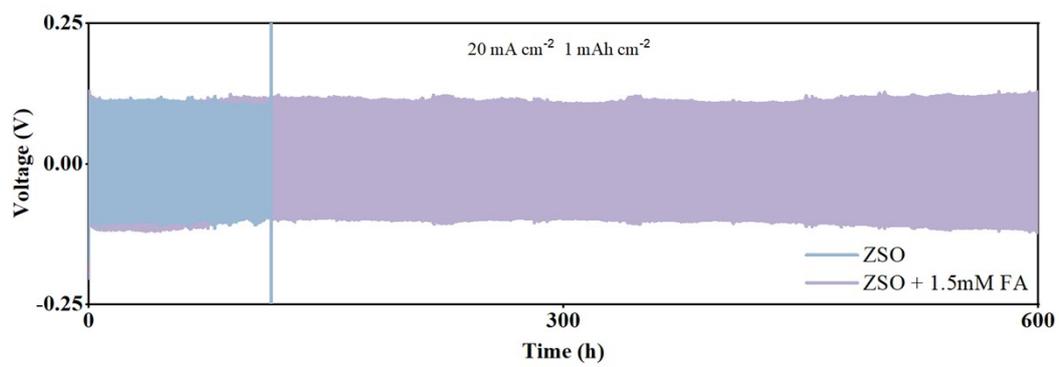


Figure S23: Long-term cycling performance at 20 mA cm<sup>-2</sup> and 1 mAh cm<sup>-2</sup>.

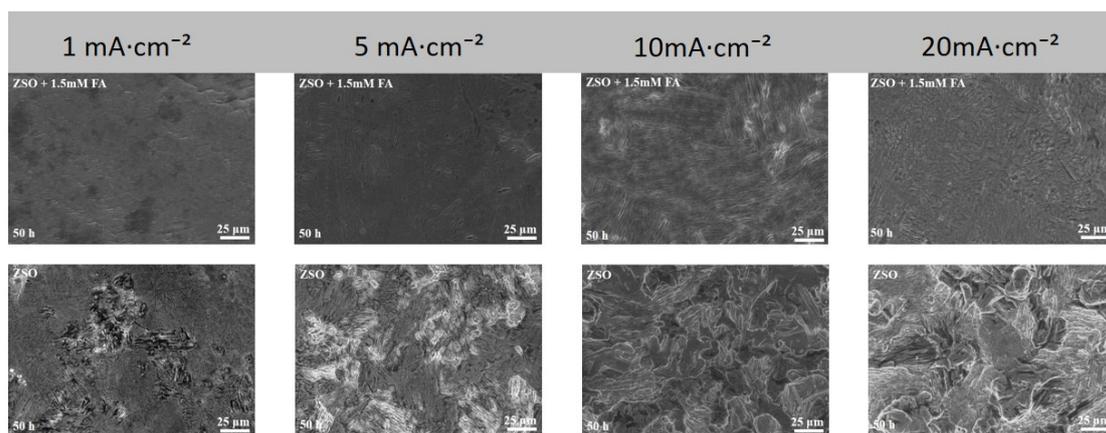


Figure S24: SEM images of zinc anode surfaces at different current densities ( $1 \text{ mA cm}^{-2}$ ,  $5 \text{ mA cm}^{-2}$ ,  $10 \text{ mA cm}^{-2}$  and  $20 \text{ mA cm}^{-2}$ )

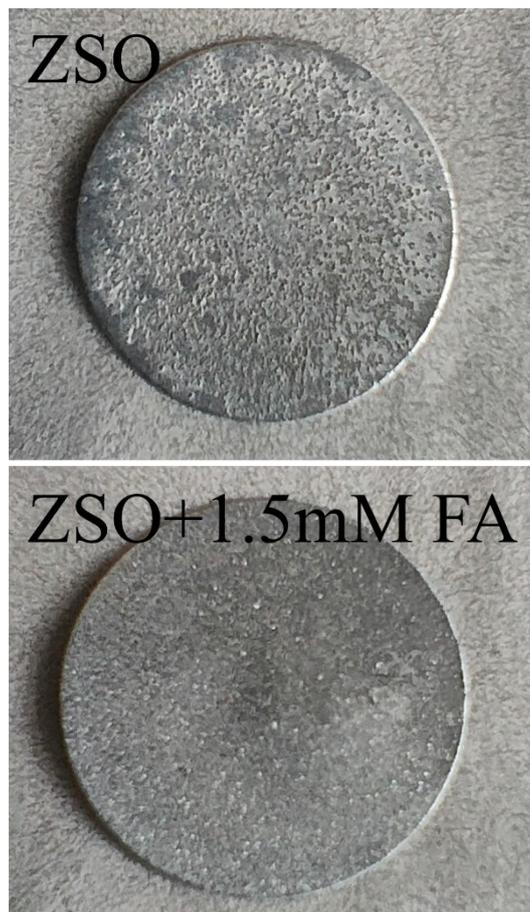


Figure S25: Images of cycled zinc anodes.

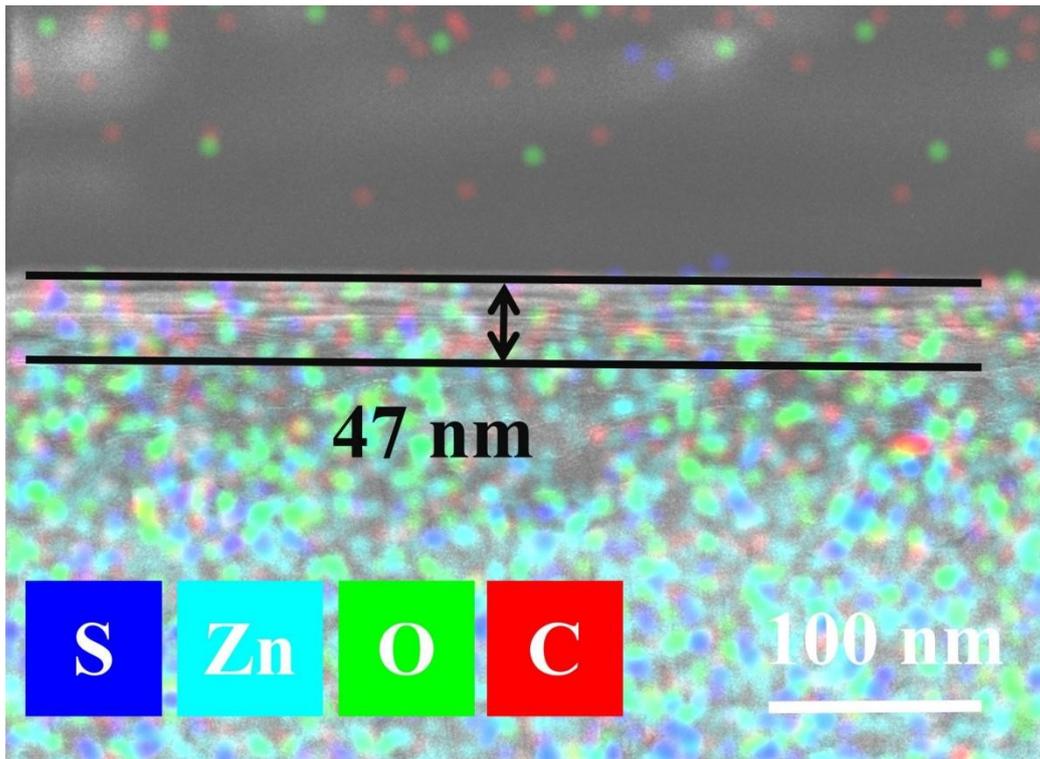


Figure S26: EDS spectrum of the SEI film.

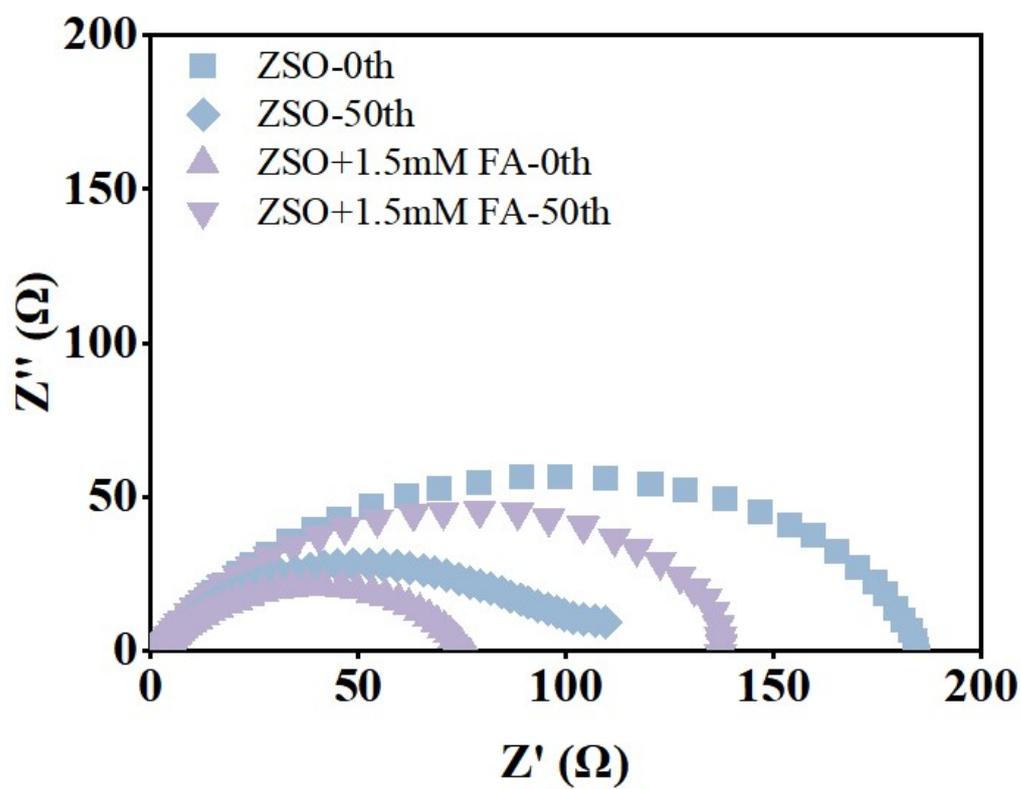


Figure S27: Impedance comparison before and after cycling.

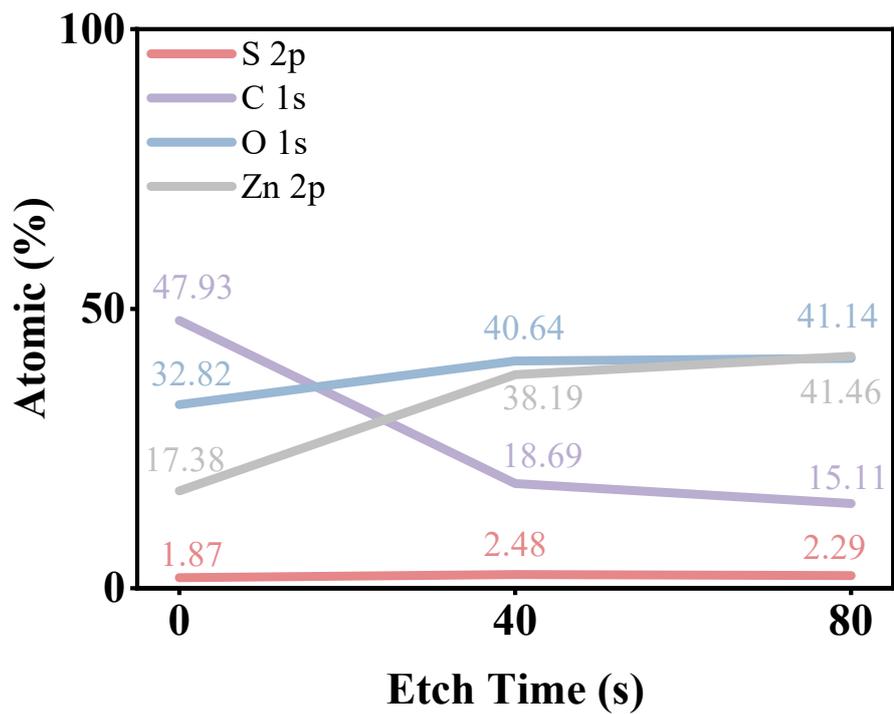


Figure S28: Elemental ratios at different sputtering depths.

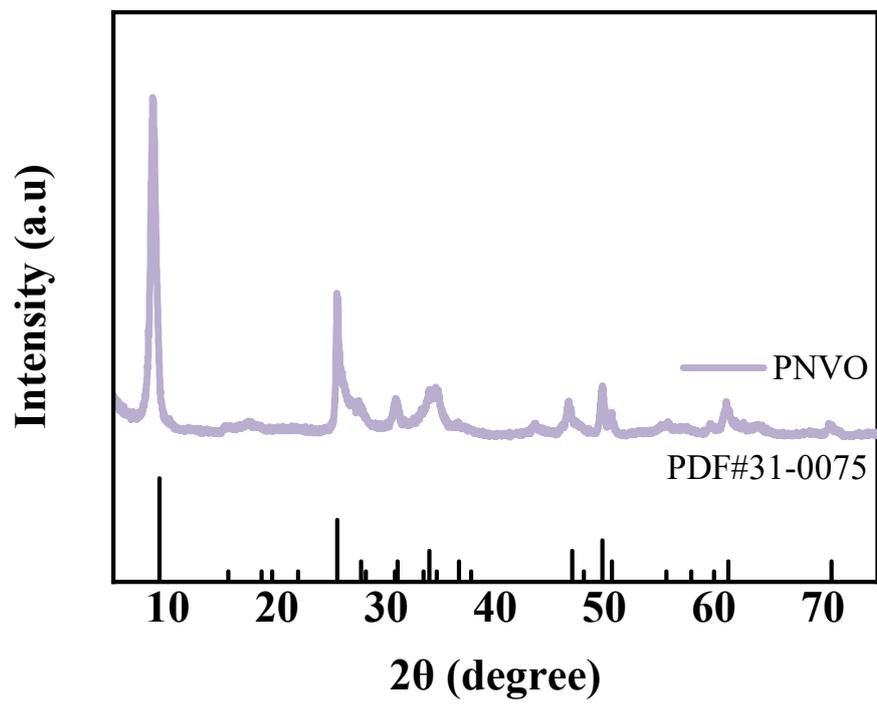


Figure S29: XRD pattern of P-NH<sub>4</sub>V<sub>4</sub>O<sub>10</sub>.

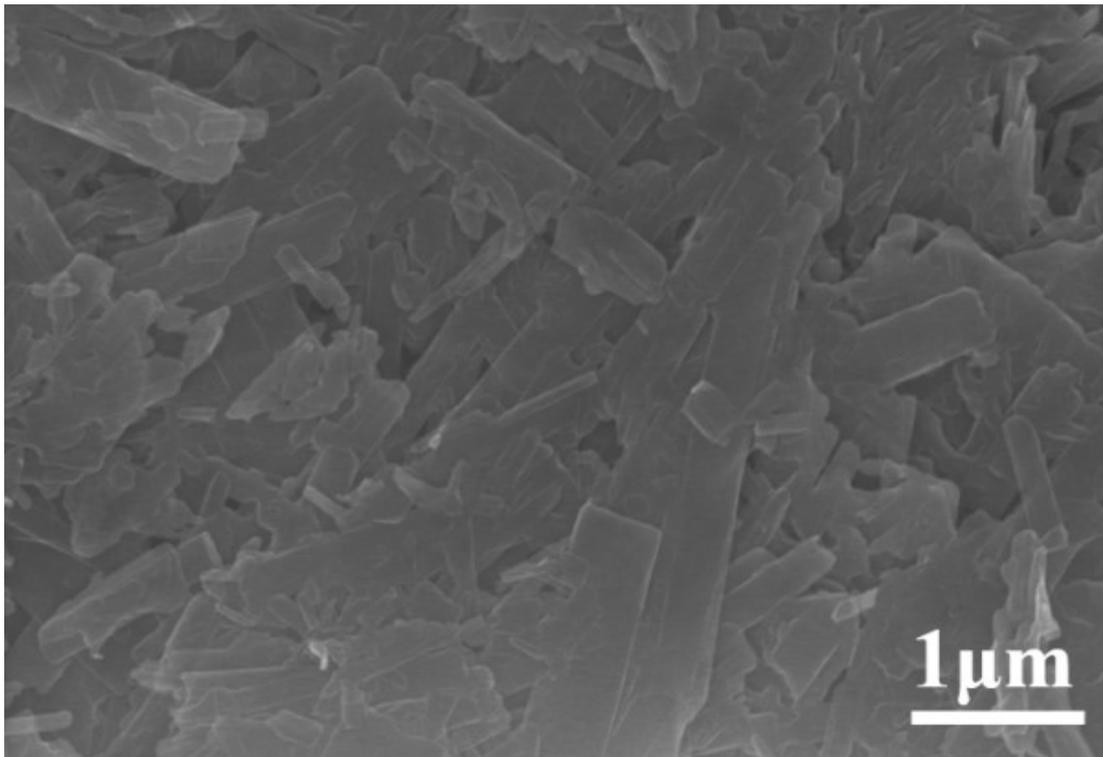


Figure S30: SEM image of P-NH<sub>4</sub>V<sub>4</sub>O<sub>10</sub>.

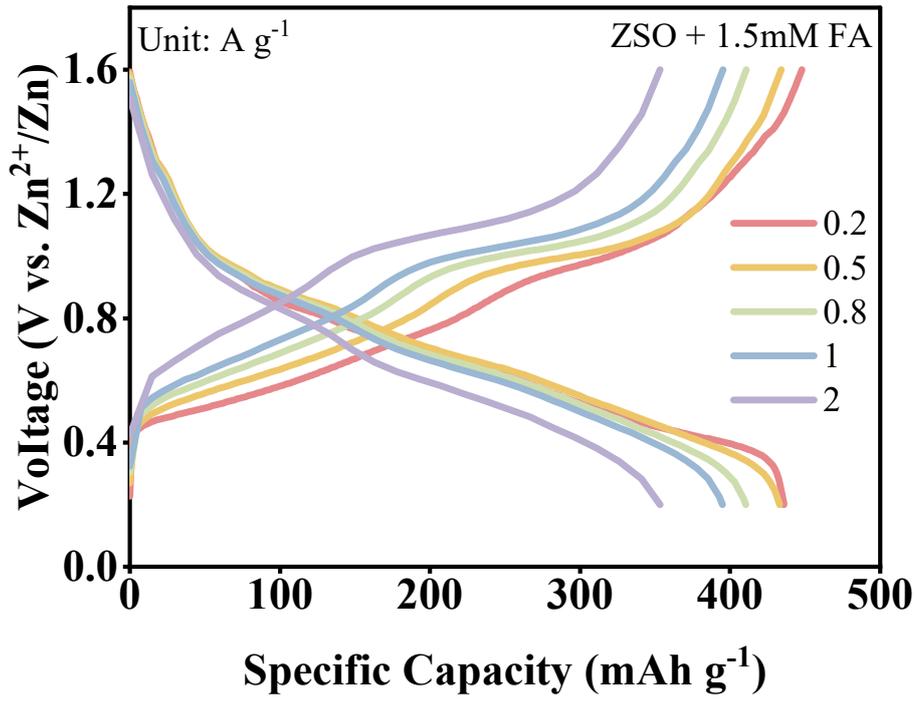


Figure S31: GCD curves of ZSO + 1.5 mM FA electrolyte.

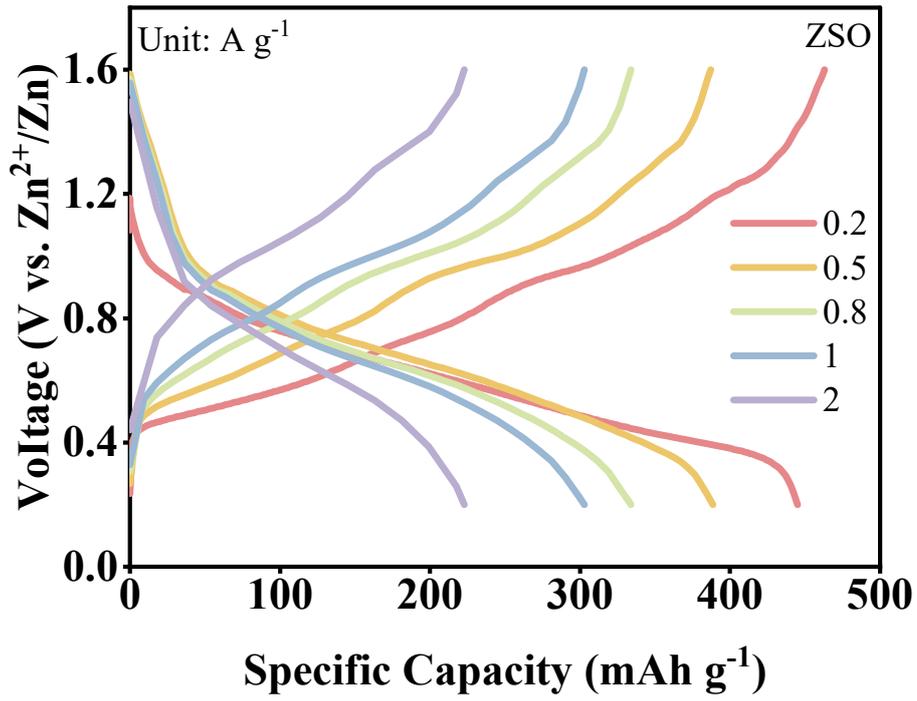


Figure S32: GCD curves of ZSO electrolyte.