

Supporting Information

Negative Thermal Quenching One-dimensional Copper(I)-Iodide Coordination Polymer Scintillator: Enabling High-Resolution X-Ray Imaging

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Supplementary figures

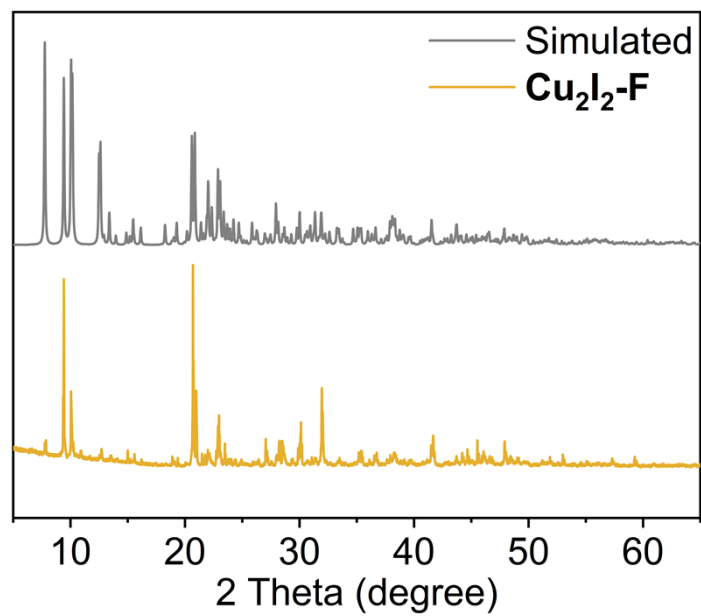


Figure S1. PXRD patterns of $\text{Cu}_2\text{I}_2\text{-F}$.

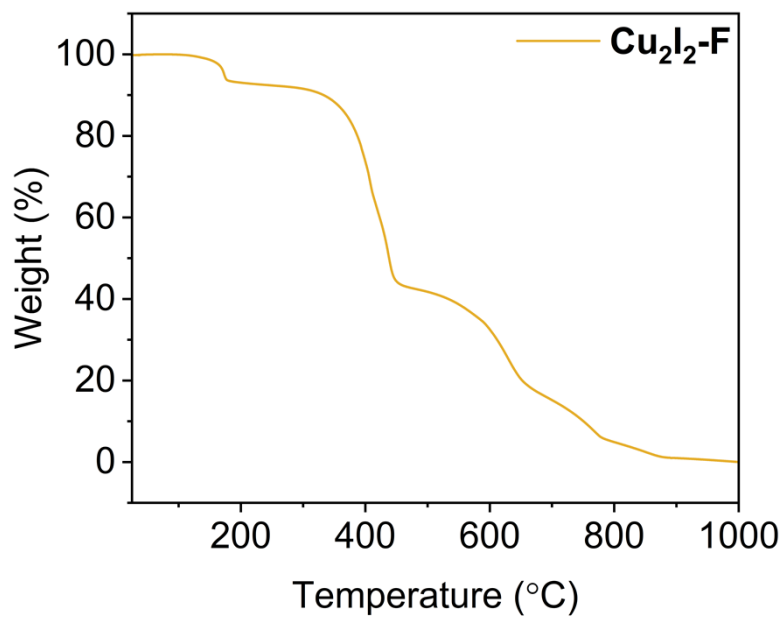


Figure S2. TGA curve for $\text{Cu}_2\text{I}_2\text{-F}$.

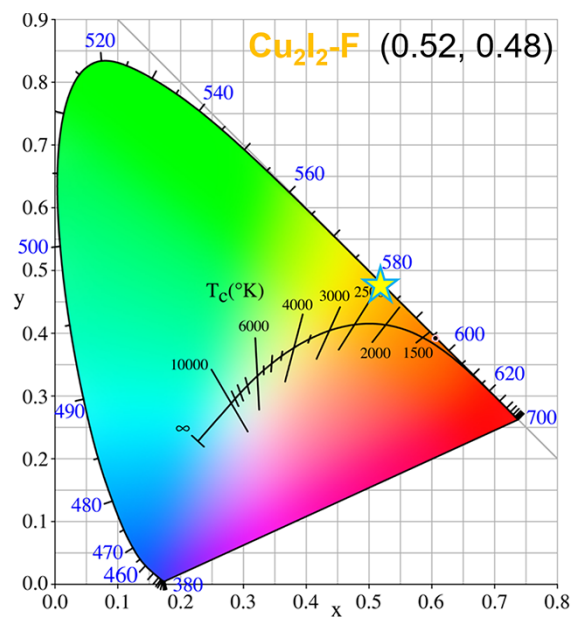


Figure S3. Commission Internationale de l'Eclairage (CIE) chromaticity coordinate diagram of the photoluminescence of the solid $\text{Cu}_2\text{I}_2\text{-F}$.

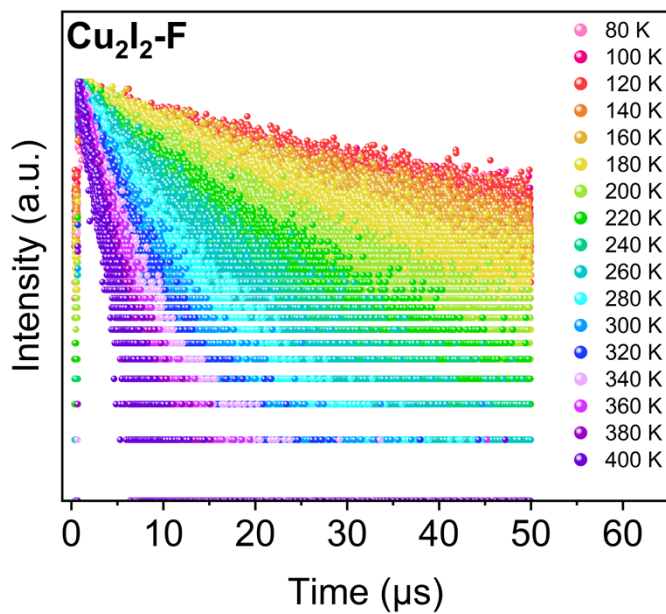


Figure S4. Temperature-dependent time-resolved PL decay of $\text{Cu}_2\text{I}_2\text{-F}$.

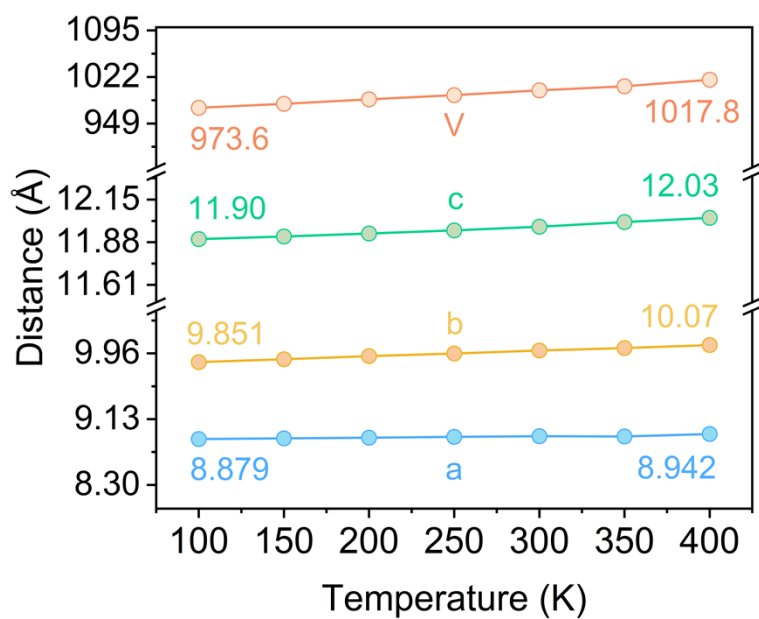


Figure S5. Temperature-dependent cell parameters of $\text{Cu}_2\text{I}_2\text{-F}$.

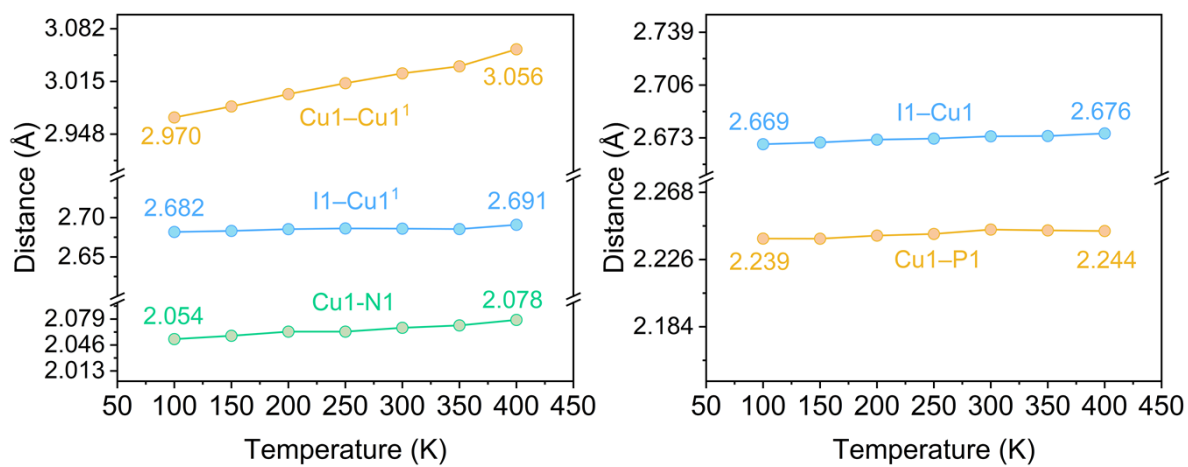


Figure S6. Temperature-dependent Cu...Cu distances and selected bond lengths in $\text{Cu}_2\text{I}_2\text{-F}$.

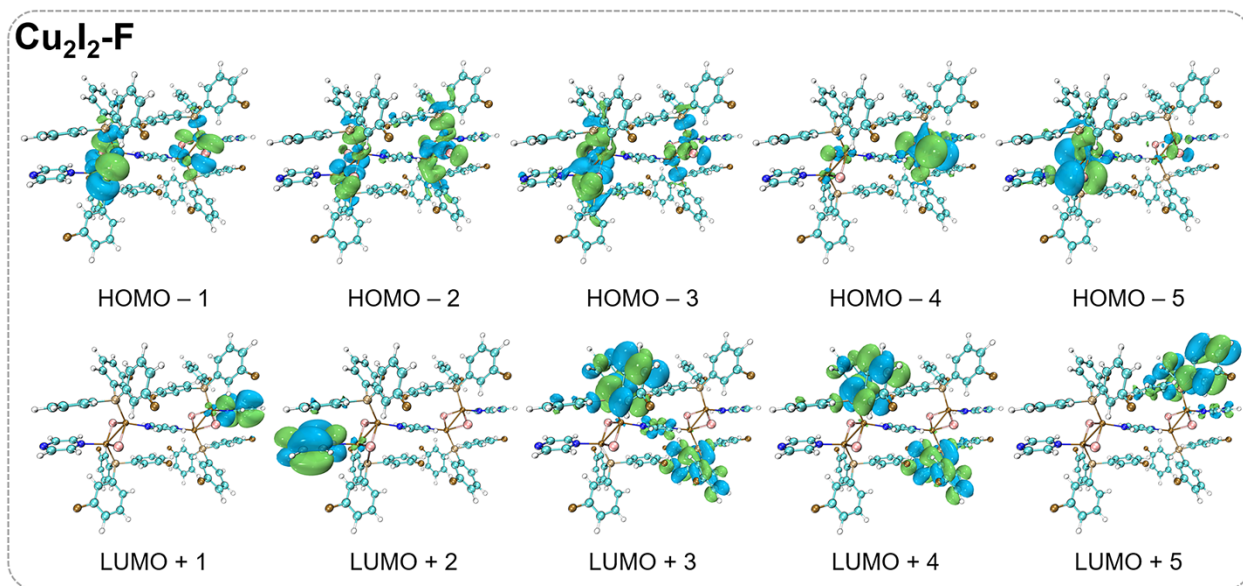


Figure S7. The calculated frontier orbitals of **Cu₂I₂-F**.

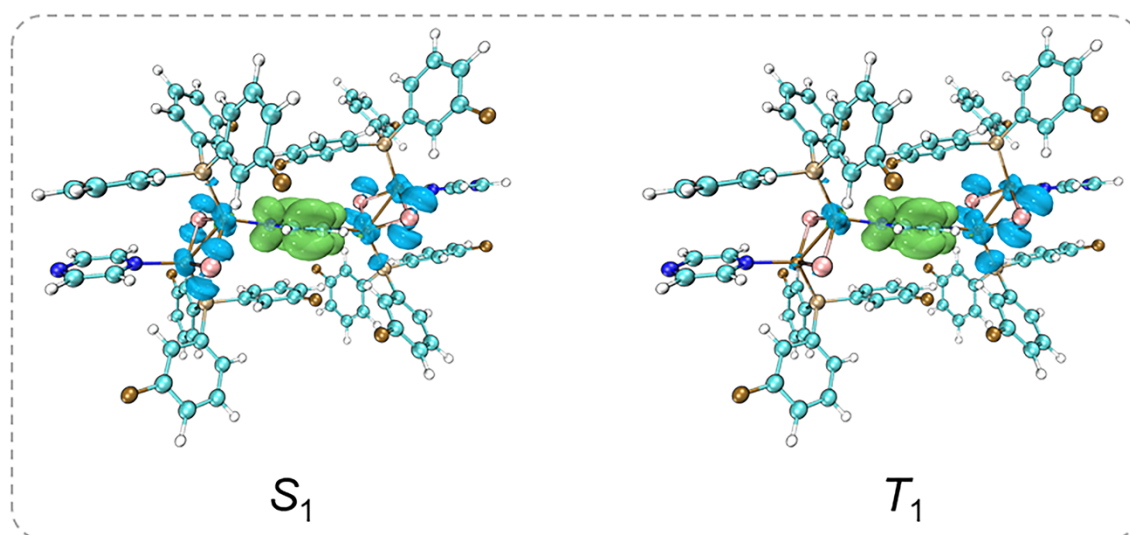


Figure S8. CDD maps of S_1 and T_1 states for **Cu₂I₂-F**.

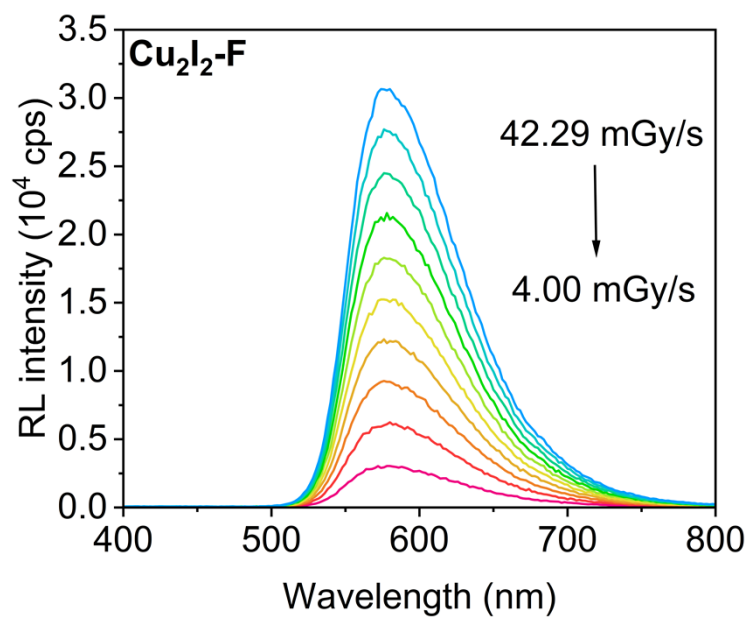


Figure S9. The X-ray dose rate-dependent RL spectra of $\text{Cu}_2\text{I}_2\text{-F}$.

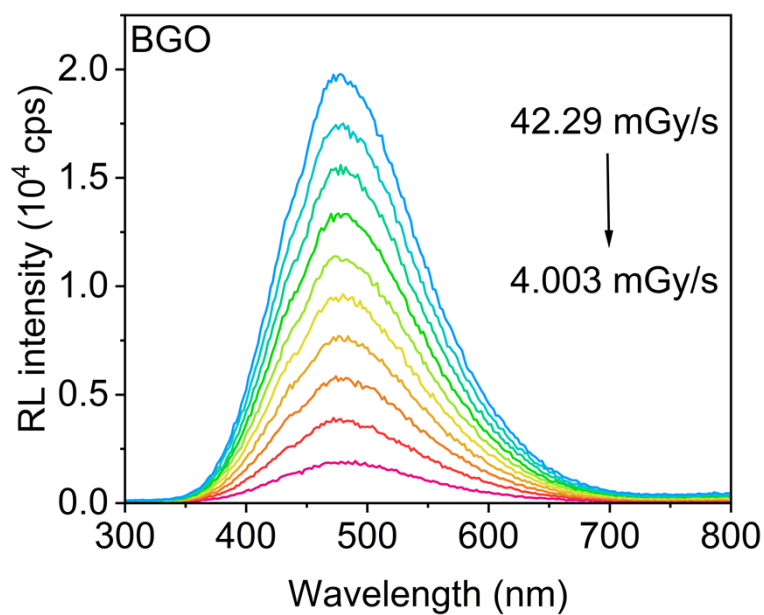


Figure S10. The X-ray dose rate-dependent RL spectra of BGO.

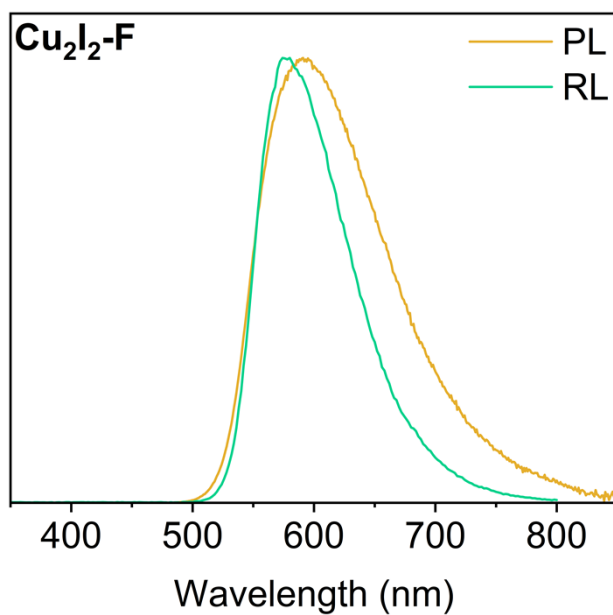


Figure S11. Comparison of PL and RL emission spectra of $\text{Cu}_2\text{I}_2\text{-F}$.

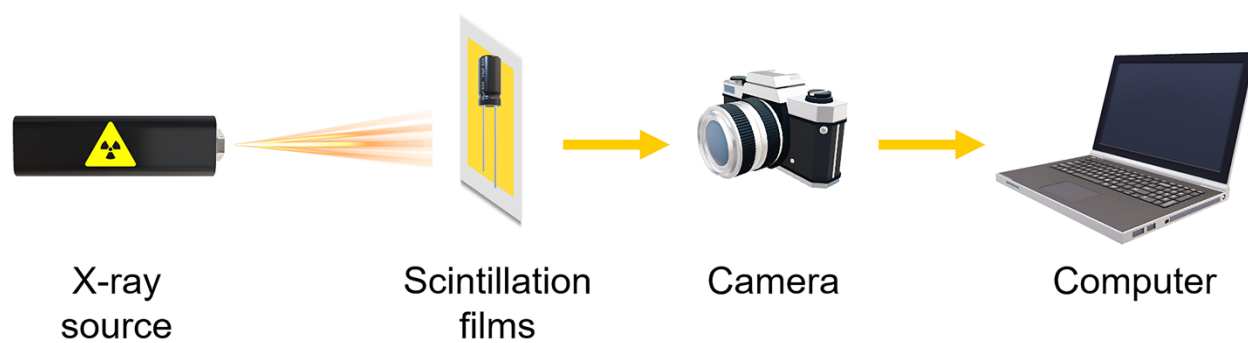


Figure S12. Schematic diagram of the experimental setup utilized for X-ray imaging.

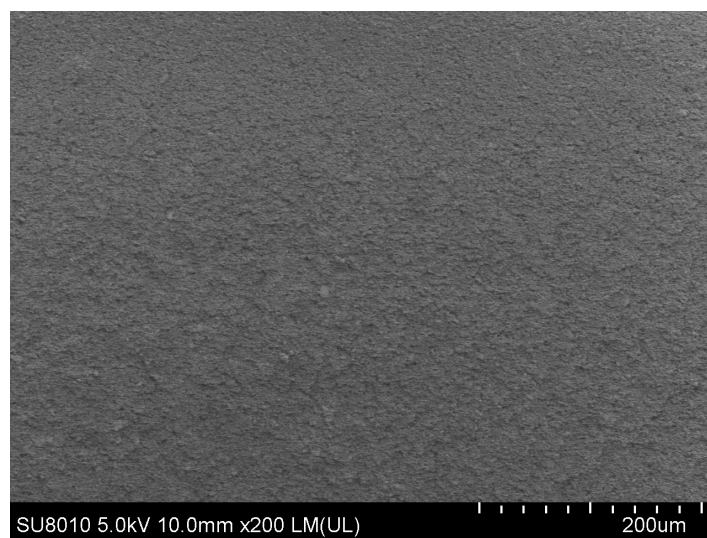


Figure S13. SEM imaging of the $\text{Cu}_2\text{I}_2\text{-F}$ -film.

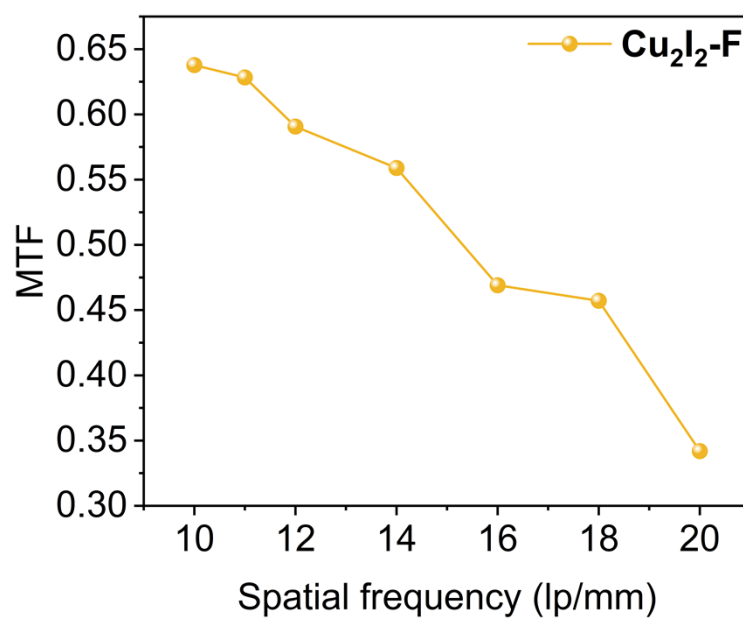


Figure S14. The MTF value of the $\text{Cu}_2\text{I}_2\text{-F}$ -film by the line-pair pattern method.

Supplementary tables

Table S1. Crystal data and structure refinement of **Cu₂I₂-F**.

Compound	Cu₂I₂-F
CCDC	2515892
Formula	C ₄₀ H ₂₈ Cu ₂ F ₆ I ₂ N ₂ P ₂
M_r	1093.46
T/K	300
Crystal system	triclinic
Space group	<i>P</i> -1
$a/\text{\AA}$	8.91430(10)
$b/\text{\AA}$	9.99620(10)
$c/\text{\AA}$	11.9783(2)
$\alpha/^\circ$	71.9450(10)
$\beta/^\circ$	89.0310(10)
$\gamma/^\circ$	80.7840(10)
$V[\text{\AA}^3]$	1001.11(2)
Z	1
<i>Calcd.</i> density (g cm ⁻³)	1.814
μ/mm^{-1}	2.744
$F(000)$	530.0
2θ [deg]	3.578 to 53.668
Reflections collected	19500
Goodness-of-fit on F^2	1.045
Final R indices [$I > 2\sigma(I)$]	$R_1 = 0.0282$ $wR_2 = 0.0678$

$$^a R_1 = \sum(F_o - F_c)/\sum F_o. \quad ^b wR_2 = [\sum w(F_o^2 - F_c^2)^2/\sum w(F_o^2)^2]^{1/2}.$$

Table S2. Bond lengths (Å) and bond angles (°) of **Cu₂I₂-F** at 300 K.

Selected bond lengths (Å)			
I1–Cu1	2.6740(4)	Cu1–P1	2.2447(8)
I1–Cu1 ¹	2.6859(4)	Cu1–N1	2.068(2)
Cu1–Cu1 ¹	3.0253(7)		
Selected bond angles (°)			
Cu1–I1–Cu1 ¹	68.725(14)	N1–Cu1–I1	106.34(6)
I1–Cu1–I1 ¹	111.275(14)	N1–Cu1–Cu1 ¹	120.11(7)
I1 ¹ –Cu1–Cu1 ¹	55.452(13)	N1–Cu1–P1	122.37(7)
I1–Cu1–Cu1 ¹	55.823(12)	C3–P1–Cu1	117.12(11)
P1–Cu1–I1	105.59(2)	C9–P1–Cu1	117.11(11)
P1–Cu1–I1 ¹	104.65(2)	C15–P1–Cu1	110.86(10)
P1–Cu1–Cu1 ¹	117.52(3)	C2–N1–Cu1	122.7(2)
N1–Cu1–I1 ¹	106.56(7)	C1–N1–Cu1	121.7(2)

Symmetric code: ¹ 1 – x, 2 – y, –z

Table S3. Crystal data and structure refinement of **Cu₂I₂-F** at different temperatures (100–400 K).

Temperature	100 K	150 K	200 K
Formula	C ₄₀ H ₂₈ Cu ₂ F ₆ I ₂ N ₂ P ₂	C ₄₀ H ₂₈ Cu ₂ F ₆ I ₂ N ₂ P ₂	C ₄₀ H ₂₈ Cu ₂ F ₆ I ₂ N ₂ P ₂
M_r	1093.46	1093.46	1093.46
Crystal system	triclinic	triclinic	triclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
$a/\text{\AA}$	8.87940(10)	8.88710(10)	8.8964(2)
$b/\text{\AA}$	9.85130(10)	9.8865(2)	9.9262(2)
$c/\text{\AA}$	11.8999(2)	11.9162(2)	11.9355(3)
$\alpha/^\circ$	108.6070(10)	71.5110(10)	71.644(2)
$\beta/^\circ$	90.1600(10)	89.9350(10)	89.677(2)
$\gamma/^\circ$	98.7280(10)	81.1380(10)	80.997(2)
$V[\text{\AA}^3]$	973.64(2)	979.82(3)	986.97(4)
Z	1	1	1
Calcd. density (g cm ⁻³)	1.865	1.853	1.840
μ/mm^{-1}	2.821	2.803	2.783
$F(000)$	530.0	530.0	530.0
2θ [deg]	3.616 to 60.706	3.608 to 60.986	3.6 to 60.692
Reflections collected	22262	22436	22178
Goodness-of-fit on F^2	1.054	1.051	1.049
Final R indices	$R_1 = 0.0258$	$R_1 = 0.0281$	$R_1 = 0.0312$
$[I > 2\sigma(I)]$	$wR_2 = 0.0579$	$wR_2 = 0.0605$	$wR_2 = 0.0694$

$$^a R_1 = \sum(F_o - F_c)/\sum F_o. \quad ^b wR_2 = [\sum w(F_o^2 - F_c^2)^2/\sum w(F_o^2)^2]^{1/2}.$$

Temperature	250 K	300 K	350 K	400 K
Formula	C ₄₀ H ₂₈ Cu ₂ F ₆ I ₂ N ₂ P ₂	C ₄₀ H ₂₈ Cu ₂ F ₆ I ₂ N ₂ P ₂	C ₄₀ H ₂₈ Cu ₂ F ₆ I ₂ N ₂ P ₂	C ₄₀ H ₂₈ Cu ₂ F ₆ I ₂ N ₂ P ₂
M_r	1093.46	1093.46	1093.46	1093.46
Crystal system	triclinic	triclinic	triclinic	triclinic
Space group	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1	<i>P</i> -1
$a/\text{\AA}$	8.90700(10)	8.91430(10)	8.9123(4)	8.9421(4)
$b/\text{\AA}$	9.9592(2)	9.99620(10)	10.0282(6)	10.0653(6)
$c/\text{\AA}$	11.9549(2)	11.9783(2)	12.0080(8)	12.0343(8)
$\alpha/^\circ$	71.765(2)	71.9450(10)	72.114(5)	72.297(5)
$\beta/^\circ$	89.3610(10)	89.0310(10)	88.671(4)	88.128(4)
$\gamma/^\circ$	80.8630(10)	80.7840(10)	80.686(4)	80.587(4)
$V[\text{\AA}^3]$	993.60(3)	1001.11(2)	1007.47(11)	1017.79(11)
Z	1	1	1	1
Calcd. density (g cm ⁻³)	1.827	1.814	1.802	1.784
μ/mm^{-1}	2.764	2.744	2.726	2.699
$F(000)$	530.0	530.0	530.0	530.0
2θ [deg]	3.59 to 60.926	3.578 to 53.668	3.566 to 55.122	3.554 to 55.194
Reflections collected	22804	19500	19349	19575
Goodness-of-fit on F^2	1.040	1.045	1.040	1.064
Final R indices	$R_1 = 0.0334$	$R_1 = 0.0282$	$R_1 = 0.0299$	$R_1 = 0.0317$
$[I > 2\sigma(I)]$	$wR_2 = 0.0723$	$wR_2 = 0.0678$	$wR_2 = 0.0787$	$wR_2 = 0.0862$

$$^a R_1 = \sum(F_o - F_c)/\sum F_o. \quad ^b wR_2 = [\sum w(F_o^2 - F_c^2)^2/\sum w(F_o^2)^2]^{1/2}.$$

Table S4. Cu...Cu distances and selected bond lengths (Å) of **Cu₂I₂-F** at different temperatures (100–400 K).

Temperature	100 K	150 K	200 K	250 K
I1–Cu1	2.6691(3)	2.6702(3)	2.6719(4)	2.6726(4)
I1–Cu1 ¹	2.6817(3)	2.6830(3)	2.6853(4)	2.6862(4)
Cu1–Cu1 ¹	2.9693(6)	2.9833(6)	2.9989(6)	3.0128(7)
Cu1–P1	2.2391(7)	2.2390(7)	2.2410(7)	2.2420(8)
Cu1–N1	2.0535(19)	2.0578(19)	2.063(2)	2.063(2)
Temperature	300 K	350 K	400 K	
I1–Cu1	2.6740(4)	2.6742(4)	2.6758(5)	
I1–Cu1 ¹	2.6859(4)	2.6854(5)	2.6909(5)	
Cu1–Cu1 ¹	3.0253(7)	3.0343(7)	3.0559(8)	
Cu1–P1	2.2447(8)	2.2442(8)	2.2438(9)	
Cu1–N1	2.068(2)	2.071(2)	2.078(2)	

Symmetric codes: ¹ 1 – x, 2 – y, –z