

Supporting Information

Yolk-Shell Structured $\text{MnCo}_2\text{O}_{4.5}$ Nanospheres for High Performance Lithium-Ion Battery Anodes

Yanhui Qiu, Jingchao Zhu, Xintao Zuo, Dapeng Liu, Yu Zhang

School of Chemistry, Beihang University, Beijing 100191, P. R. China

*Corresponding author. E-mail: liudp@buaa.edu.cn (D. P. Liu); jade@buaa.edu.cn (Y. Zhang)

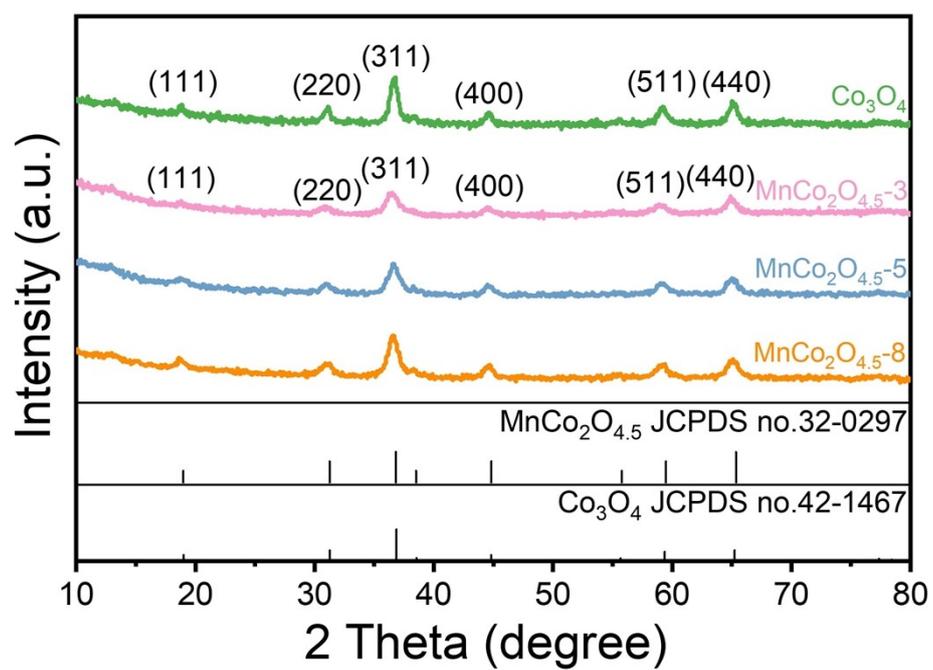


Fig. S1. XRD patterns of Co_3O_4 , $\text{MnCo}_2\text{O}_{4.5-3}$, $\text{MnCo}_2\text{O}_{4.5-5}$, and $\text{MnCo}_2\text{O}_{4.5-8}$.

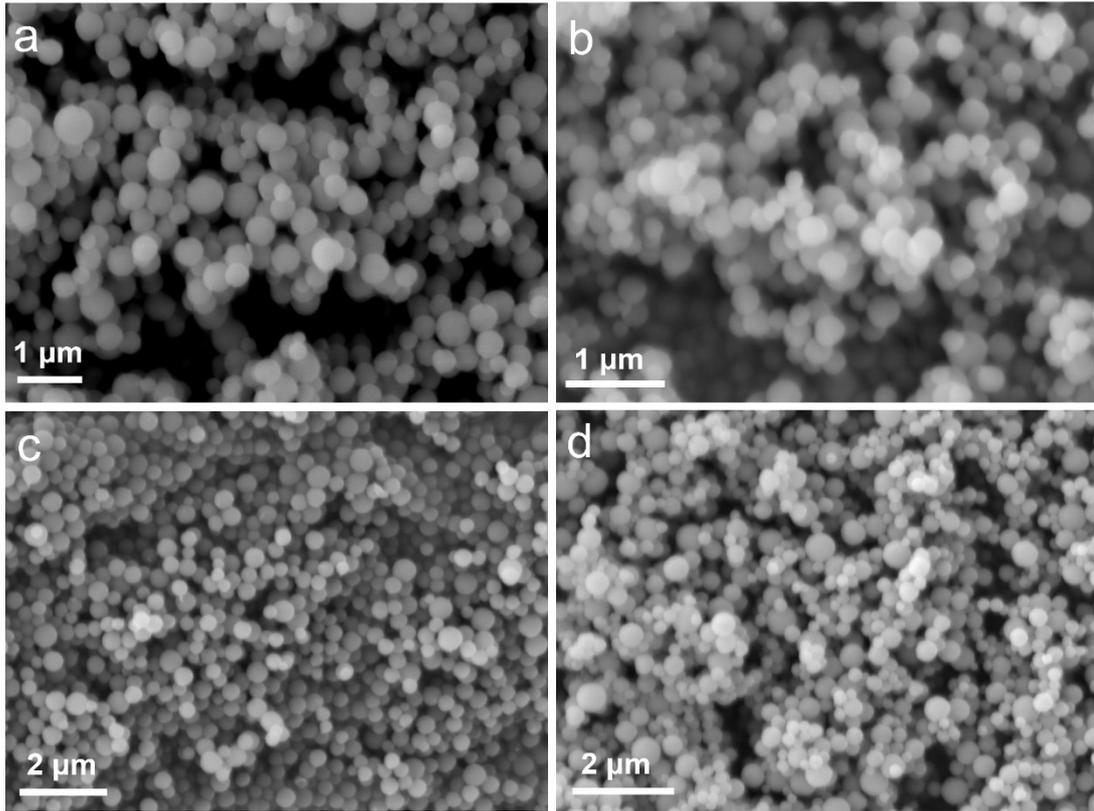


Fig. S2. SEM images of the precursors: (a) Co-gly, (b) CoMn-3-gly, (c) CoMn-5-gly, (d) CoMn-8-gly.

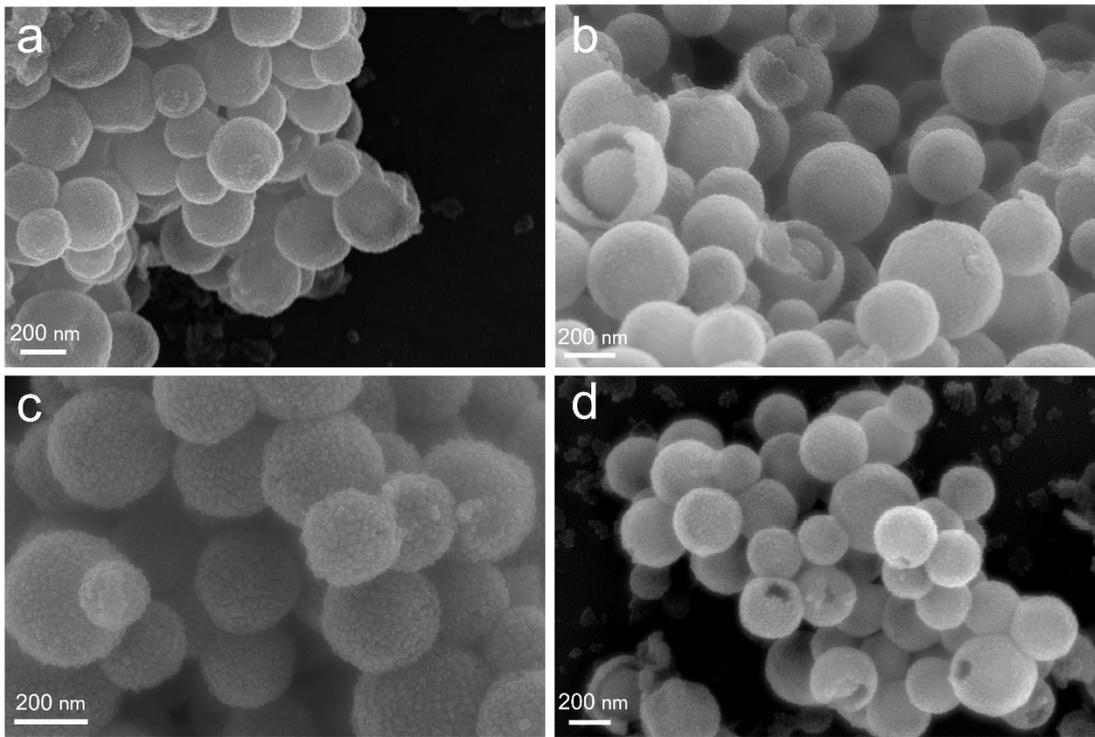


Fig. S3. SEM images of (a) Co_3O_4 , (b) $\text{MnCo}_2\text{O}_{4.5-3}$, (c) $\text{MnCo}_2\text{O}_{4.5-5}$, (d) $\text{MnCo}_2\text{O}_{4.5-8}$.

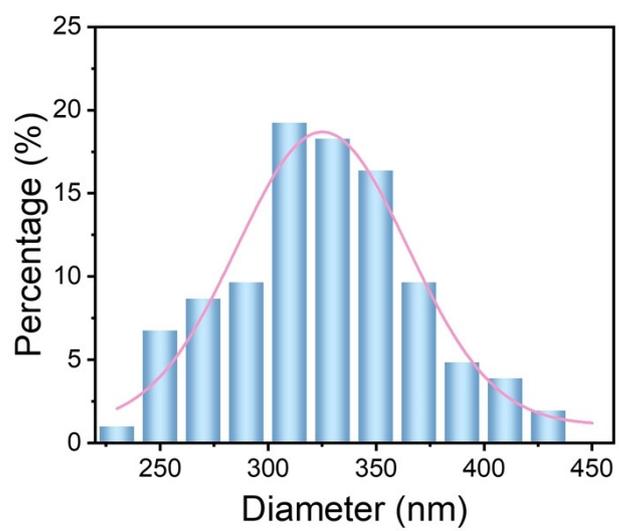


Fig. S4. MnCo₂O_{4.5-5} particle size distribution diagram.

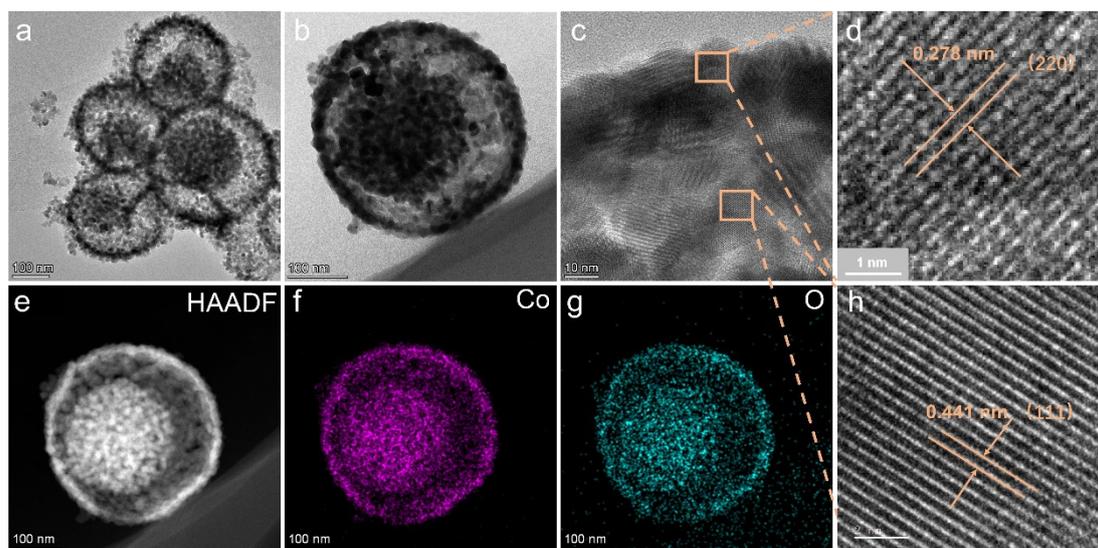


Fig. S5. (a, b) TEM images. (c) HRTEM image. (d) magnified view of the HRTEM image. (e) HAADF-STEM image, and (f, g, h) elemental distribution maps of Co_3O_4 .

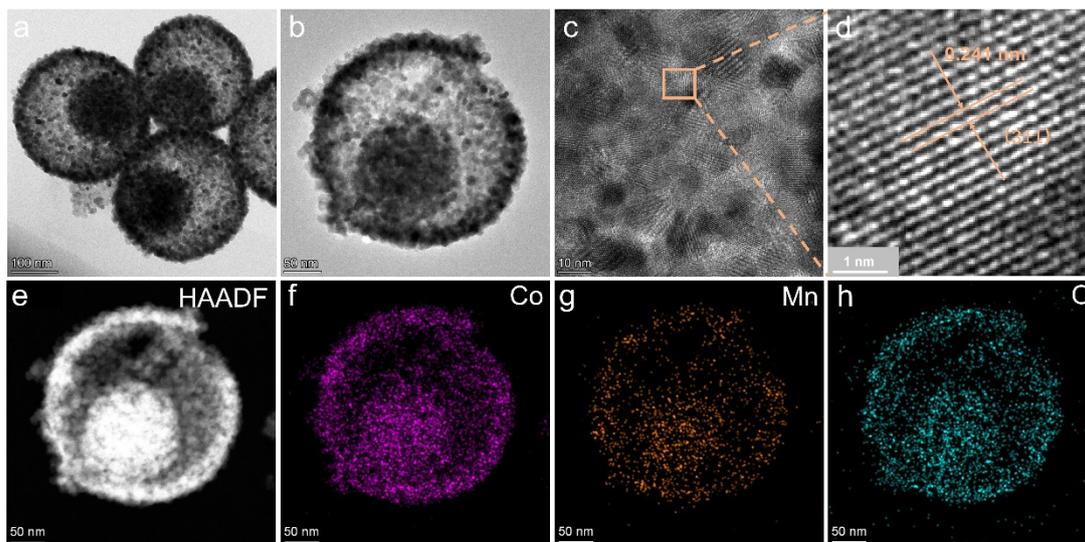


Fig. S6. (a, b) TEM images. (c) HRTEM image. (d) magnified view of the HRTEM image. (e) HAADF-STEM image, and (f, g, h) elemental distribution maps of MnCo₂O_{4.5-3}.

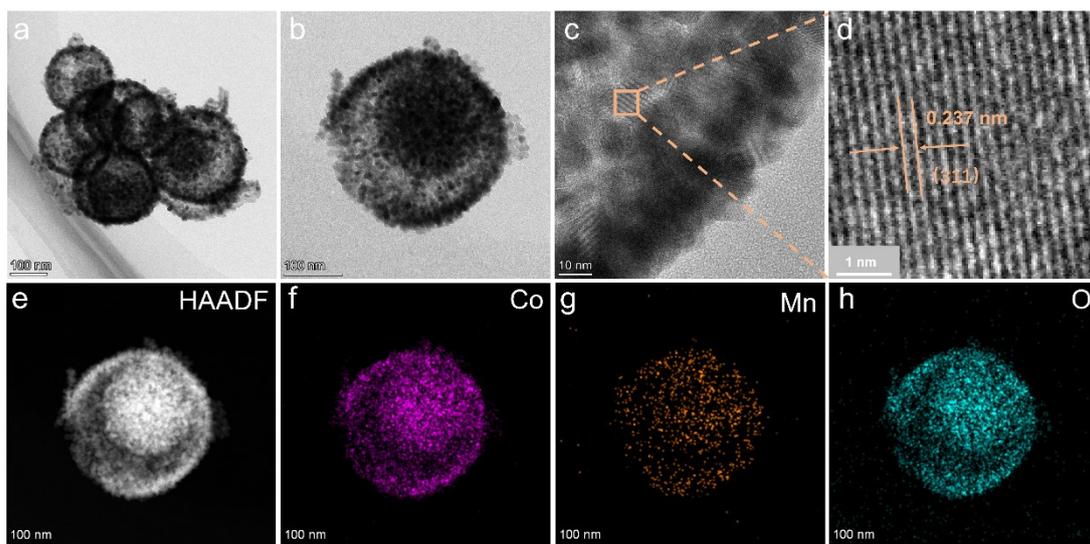


Fig. S7. (a, b) TEM images. (c) HRTEM image. (d) magnified view of the HRTEM image. (e) HAADF-STEM image, and (f, g, h) elemental distribution maps of $\text{MnCo}_2\text{O}_{4.5-8}$.

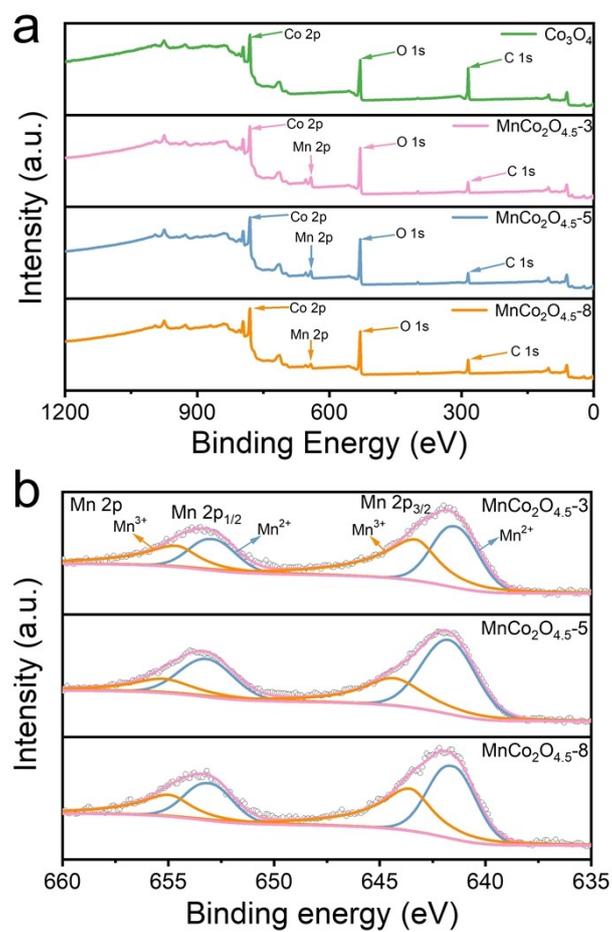


Fig. S8. (a) XPS survey spectrum. (b) XPS spectrum of Mn 2p.

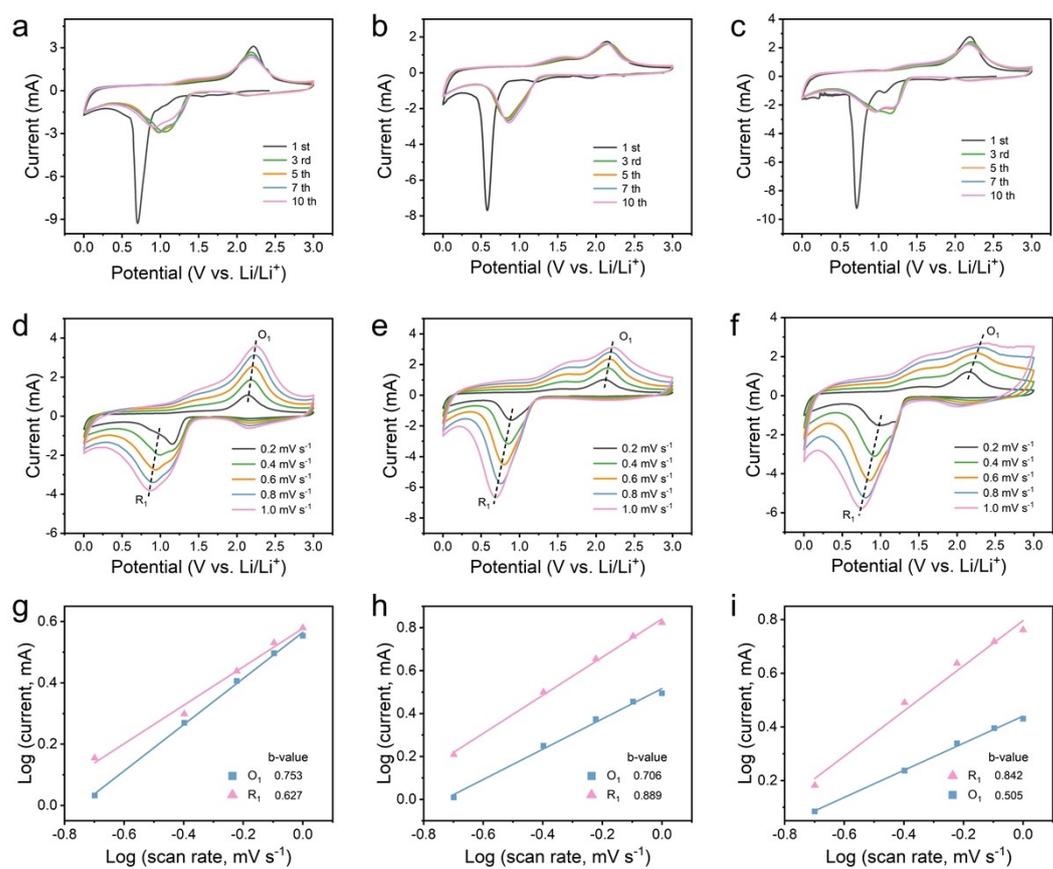


Fig. S9. CV curves of (a) Co_3O_4 . (b) $\text{MnCo}_2\text{O}_{4.5-3}$. (c) $\text{MnCo}_2\text{O}_{4.5-8}$ at a scan rate of 0.5 mV s^{-1} . CV curves at various scanning rates of (d) Co_3O_4 . (e) $\text{MnCo}_2\text{O}_{4.5-3}$. (f) $\text{MnCo}_2\text{O}_{4.5-8}$. (g) Log (i) vs. log (v) plots of (g) Co_3O_4 . (h) $\text{MnCo}_2\text{O}_{4.5-3}$. (i) $\text{MnCo}_2\text{O}_{4.5-8}$.

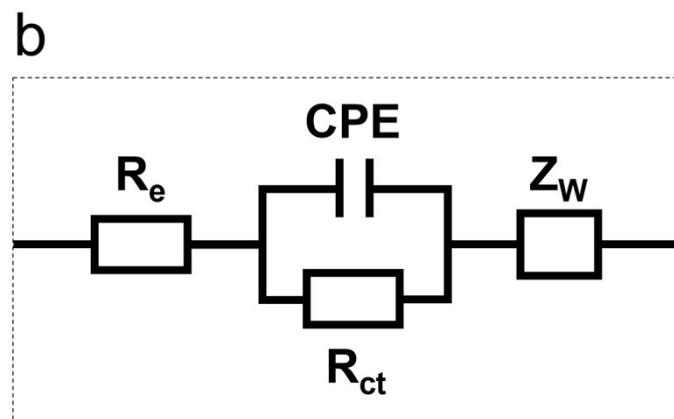
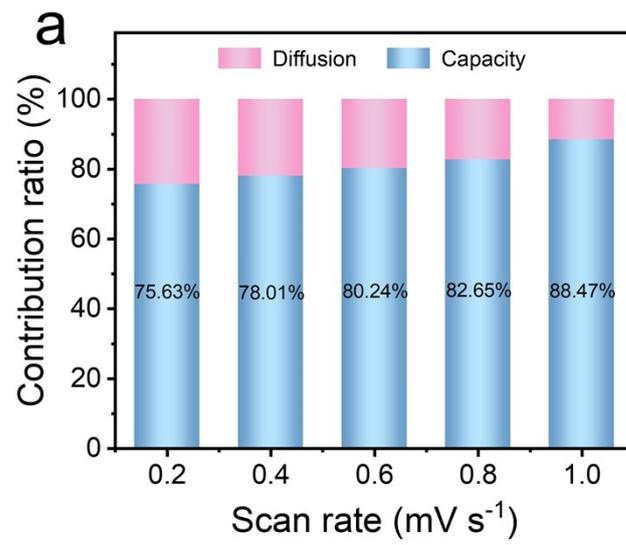


Fig. S10. (a) Percentage of pseudo-capacitive contribution of Co_3O_4 at all scan rates. (b) Equivalent circuit diagram. (R_e : The internal resistance of electrolyte; R_{ct} : The charge-transfer resistance.)

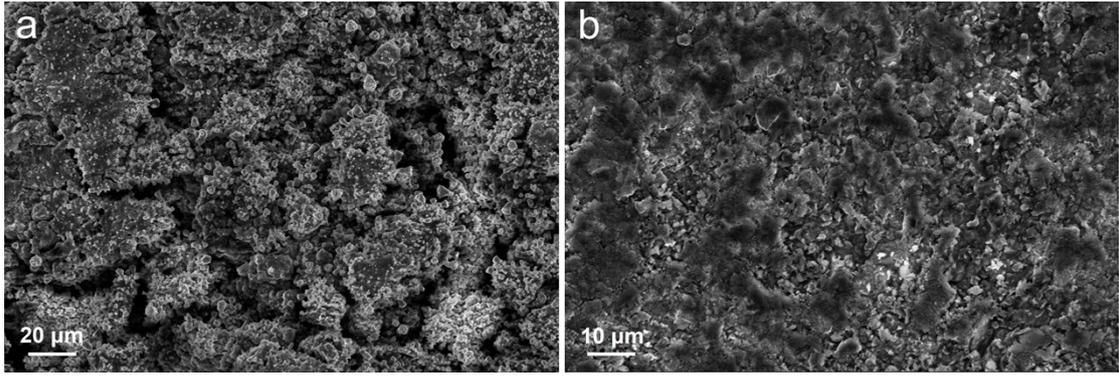


Fig. S11. SEM images of (a) $\text{MnCo}_2\text{O}_{4.5-5}$ and (b) Co_3O_4 after 400 cycles at 2 A g^{-1} .

Table S1. Comparison of electrochemical performance with various anodes for LIBs.

Materials	Current density (A g ⁻¹)	Condition	Reversible capacity (mAh g ⁻¹)	Ref.
CoO-ZnO@NC-450	0.2	After 100 cycles	796.2	[1]
Co ₃ O ₄ /CoP NFAs/CC	2.0	After 300 cycles	451.3	[2]
Co ₃ O ₄ /N-C	1.0	After 200 cycles	573	[3]
Co ₃ O ₄ /AG	0.5	After 100 cycles	510	[4]
NCF/Co ₃ O ₄	1.0	After 300 cycles	576	[5]
Ge-Fe-O _x -700 NWs	0.1	After 50 cycles	750	[6]
NiFeP/C@rGO-21	1.0	After 300 cycles	530.4	[7]
Co _{0.5} Fe _{2.5} O ₄ /PCN	2.0	After 420 cycles	490	[8]
NiCo ₂ S ₄ @GS-1	5.0	Rate capability	460	[9]
Co ₂ AlO ₄	3.2	Rate capability	548	[10]
Ni _x Co _{3-x} O ₄	5.0	Rate capability	546	[11]
L-ZnCo ₂ O ₄ @rGO-30	3.2	Rate capability	686	[12]
MnCo ₂ O _{4.5}	0.2	After 50 cycles	1383.7	This work
	2.0	After 300 cycles	578.5	
	5.0	Rate capability	680.9	

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