

Supporting Information

Unlocking XAT/HAT Selectivity: A Machine Learning-Guided Discovery of Governing Physicochemical Principles

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1. Details of the dataset used in model training

1.1 Detailed substrates

The distribution of organic halides within our dataset is illustrated in Figure S1, with corresponding optimization output files hosted on our GitHub repository (<https://github.com/liusqchem/machine-learning.git>). Sourced from the *iBonD* 2.0 database, the dataset includes 113 distinct C–H bonds and 61 C–X bonds (Figure S1). This selection was designed with a dual purpose. First, it ensures a broad representation of chemical and structural diversity, encompassing alkyl, cycloalkyl, (hetero)benzylic, aldehydic, and (hetero)aryl C–H and C–X motifs. The calculated bond dissociation energies (**BDEs**) range from 80–115 kcal/mol for C–H and 40–130 kcal/mol for C–X bonds (Figure S3), confirming that the dataset covers a wide and representative chemical space. Second, these **BDE** values serve as a benchmark to validate the reliability of our quantum chemical calculations (Figure S2). High-quality, chemically relevant training data is essential, as it directly governs the accuracy of the resulting machine learning (ML) models. To model radical abstraction, we considered a variety of abstracting agents, including C-, B-, N-, Si-, and S-centered radicals. Specifically, we selected 38 carbon-centered radicals—ranging from alkyl and α -heteroatom-substituted species to captodative and sp^2 -hybridized radicals—leading to a comprehensive set of 4,294 HAT and 2,318 XAT reactions. Furthermore, to assess the generalizability of the H/X selectivity patterns, we incorporated an additional array of experimental boryl, nitrogen-centered, silyl, and thiol radicals.

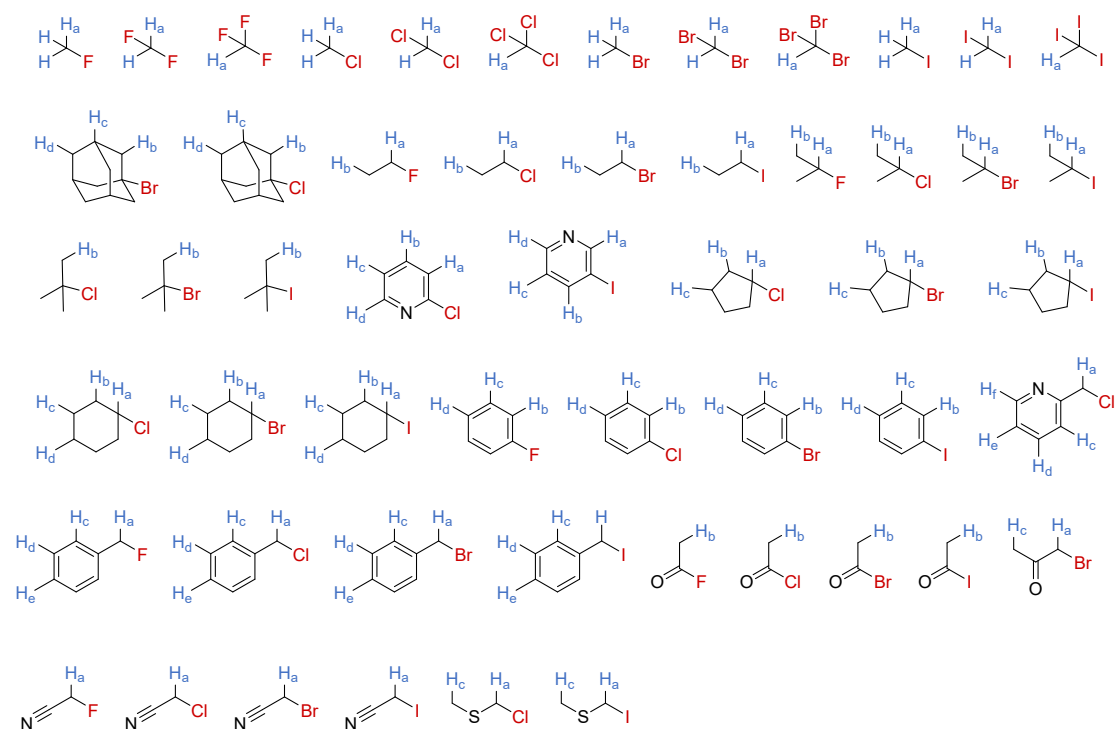


Figure S1. Compound details of organic halides

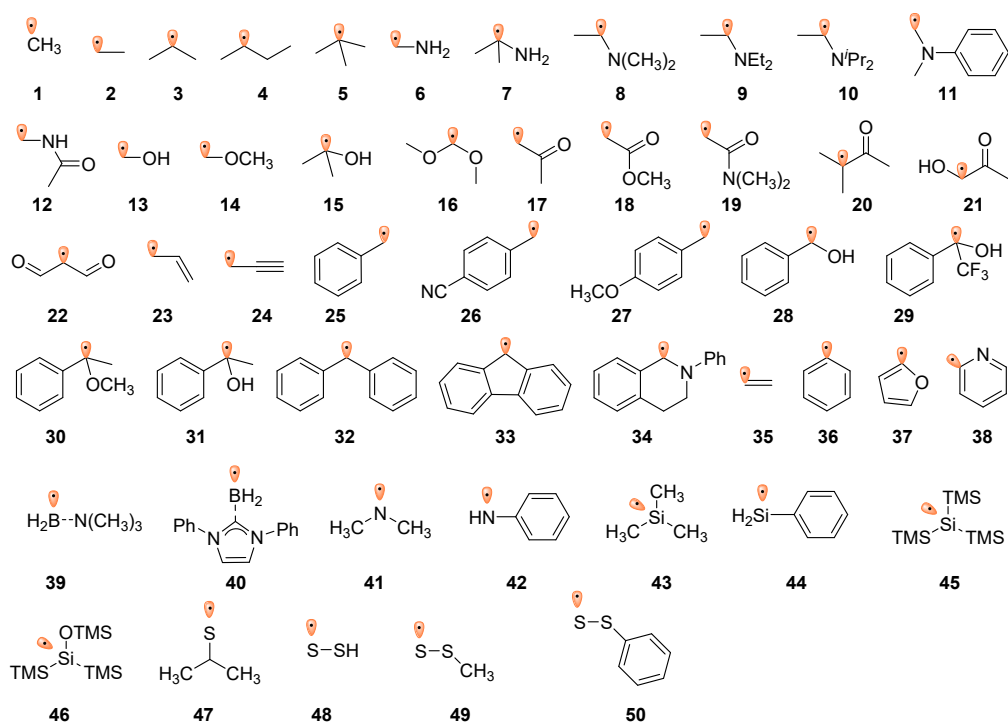


Figure S2. Compound details of radicals

1.2 Benchmark of the theoretical levels for geometry optimization

All DFT calculations were conducted with Gaussian 16². Structural optimizations were performed at the (U)B3LYP-D3(BJ)³⁻⁶ / Lanl2dz+d with ECP for I^{7, 8}, and (U)B3LYP-D3(BJ) / 6-31+G(d,p)⁹ for other atoms in the gas phase. Single point calculations were conducted at the (U)B3LYP-D3(BJ)/6-311++G(3df,3pd)^{9, 10}-SDD^{11, 12}. The calculated BDE_{C-H} and BDE_{C-X} and experimental BDE_{C-H} and BDE_{C-X} of organic halides are shown in Table S1.

Table S1. The calculated BDE_{C-H} and BDE_{C-X} and experimental BDE_{C-H} and BDE_{C-X} of organic halides (kcal/mol).

	Bond type	$BDE_{exp.}$	$BDE_{calc.}$
BDE _{C-H}	CHF ₃	106.4	103.9
	CH ₂ F ₂	103.2	99.1
	CH ₃ F	101.3	99.1
	CHCl ₃	93.8	90.9
	CH ₂ Cl ₂	95.7	93.8
	CH ₃ Cl	100.1	97.5
	CHBr ₃	95.4	91.0
	CH ₂ Br ₂	98.6	95.2
	CH ₃ Br	102.1	98.9
	CHI ₃	101.1	93.0
	CH ₂ I ₂	103.0	97.6
	CH ₃ I	103.5	100.9
	C ₂ H ₅ Br (H _b)	99.2	96.3
	C ₂ H ₅ Cl (H _a)	97.2	95.0

	C ₂ H ₅ Cl (H _b)	101.1	98.8
	C ₂ H ₅ F (H _a)	98.2	96.6
	C ₂ H ₅ F (H _b)	103.6	101.3
	C ₇ H ₇ Cl (H _a)	84.2	84.2
	CNCH ₂ F	88.6	88.6
BDE _{C-X}	CHF ₃	127.6	123.0
	CH ₂ F ₂	118.6	116.7
	CH ₃ F	110.0	109.2
	CHCl ₃	74.4	70.7
	CH ₂ Cl ₂	80.8	76.1
	CH ₃ Cl	83.7	81.5
	CHBr ₃	65.7	59.2
	CH ₂ Br ₂	66.0	65.2
	CH ₃ Br	70.3	70.6
	CHI ₃	52.7	48.1
	CH ₂ I ₂	51.8	53.5
	CH ₃ I	56.9	58.3
	C ₁₀ H ₁₅ Br	70.8	74.8
	C ₁₀ H ₁₅ Cl	90.4	86.9
	C ₂ H ₅ F	111.7	111.0
	C ₂ H ₅ Cl	84.2	81.7
	C ₂ H ₅ Br	69.6	70.3
	C ₂ H ₅ I	56.3	57.8
	C ₃ H ₇ F	110.1	112.2
	C ₃ H ₇ Cl	84.6	81.3
	C ₃ H ₇ Br	70.5	69.4
	C ₃ H ₇ I	55.7	56.8
	C ₅ H ₄ NCl	90.5	90.7
	C ₅ H ₄ NI	66.6	68.7
	C ₅ H ₉ Cl	82.4	79.8
	C ₅ H ₉ Br	69.8	68.1
	C ₅ H ₉ I	55.4	55.8
	C ₆ H ₁₁ F	117.5	112.8
	C ₆ H ₁₁ Cl	86.1	82.7
	C ₆ H ₁₁ Br	69.9	70.8
	C ₆ H ₁₁ I	55.5	58.4
	C ₆ H ₅ F	127.2	125.8
	C ₆ H ₅ Cl	97.1	95.0
	C ₆ H ₅ Br	80.4	82.1
	C ₆ H ₅ I	67.4	68.9
	C ₇ H ₇ F	98.7	96.9
	C ₇ H ₇ Cl	71.7	68.9
	C ₇ H ₇ Br	57.2	58.1

C ₇ H ₇ I	44.9	46.9
CH ₃ COF	122.3	119.3
CH ₃ COCl	84.6	82.3
CH ₃ COBr	69.8	70.2
CH ₃ COI	53.3	56.7
CNCH ₂ Cl	66.4	66.8
CNCH ₂ Br	56.8	56.3
CNCH ₂ I	44.7	45.2
SC ₂ H ₅ Cl	72	73.4
SC ₂ H ₅ I	51.8	50.8
CH ₃ COCH ₂ Br	62.5	60.0
C ₆ H ₆ NCl	69.9	69.1
C ₄ H ₉ Cl	84.1	80.5
C ₄ H ₉ Br	70	68.3
C ₄ H ₉ I	53.4	55.6
Mean absolute error	/	2.3

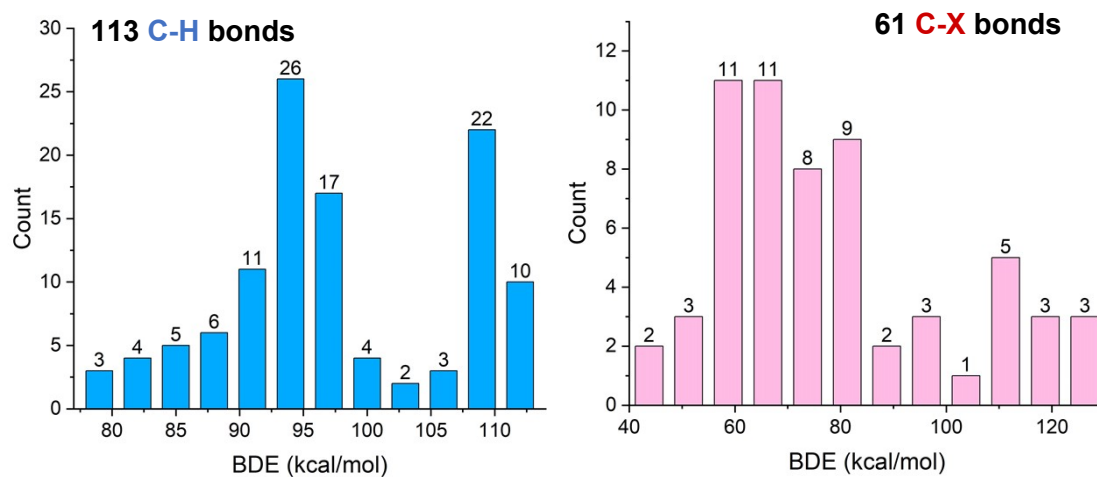


Figure S3. Distribution of calculated *BDEs*(kcal/mol).

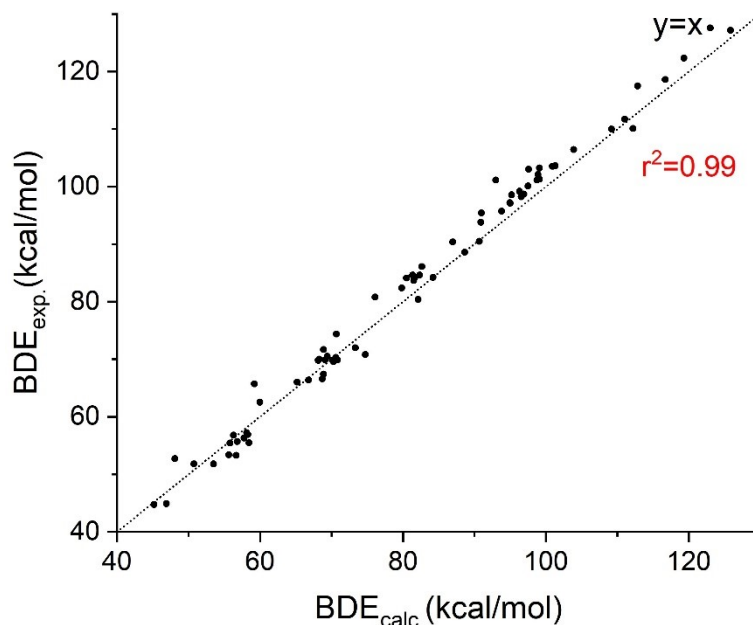


Figure S4. Correlations between the $BDE_{\text{calc.}}$ with $BDE_{\text{exp.}}$ (kcal/mol).

1.3 Calculated BDE and BO

Table S2. The calculated $BDE_{\text{C-H}}$ and $BO_{\text{C-H}}$.

Bond type	BDE	BO	Bond type	BDE	BO	Bond type	BDE	BO
C ₁₀ H ₁₅ Br (H _b)	99.178	0.923	C ₅ H ₉ Cl (H _c)	94.0807	0.9342	C ₇ H ₇ Cl (H _c)	111.8924	0.9261
C ₁₀ H ₁₅ Br (H _c)	97.3249	0.9142	C ₅ H ₉ I (H _a)	94.085	0.9218	C ₇ H ₇ Cl (H _d)	111.5944	0.9277
C ₁₀ H ₁₅ Br (H _d)	97.9949	0.9305	C ₅ H ₉ I (H _b)	79.258	0.9098	C ₇ H ₇ Cl (H _e)	111.7443	0.9277
C ₁₀ H ₁₅ Cl (H _b)	99.1365	0.9242	C ₅ H ₉ I (H _c)	94.251	0.9332	C ₇ H ₇ F (H _a)	84.3371	0.9404
C ₁₀ H ₁₅ Cl (H _c)	97.4937	0.9146	C ₆ H ₁₁ Br (H _a)	94.4278	0.9125	C ₇ H ₇ F (H _c)	111.7713	0.9262
C ₁₀ H ₁₅ Cl (H _d)	98.0437	0.9305	C ₆ H ₁₁ Br (H _b)	97.2602	0.9222	C ₇ H ₇ F (H _d)	111.5955	0.9277
C ₂ H ₅ Br (H _a)	96.1901	0.9313	C ₆ H ₁₁ Br (H _c)	96.1702	0.9291	C ₇ H ₇ F (H _e)	111.6252	0.9278
C ₂ H ₅ Br (H _b)	96.2736	0.9255	C ₆ H ₁₁ Br (H _d)	96.3728	0.9286	C ₇ H ₇ I (H _a)	88.549	0.9296
C ₂ H ₅ Cl (H _a)	94.975	0.9293	C ₆ H ₁₁ Cl (H _a)	93.5252	0.9108	C ₇ H ₇ I (H _c)	111.950	0.9265
C ₂ H ₅ Cl (H _b)	98.7753	0.9298	C ₆ H ₁₁ Cl (H _b)	97.5474	0.9229	C ₇ H ₇ I (H _d)	111.613	0.9276
C ₂ H ₅ F (H _a)	96.5698	0.9424	C ₆ H ₁₁ Cl (H _c)	96.1785	0.929	C ₇ H ₇ I (H _e)	111.822	0.9278
C ₂ H ₅ F (H _b)	101.3099	0.9371	C ₆ H ₁₁ Cl (H _d)	96.3937	0.9287	CH ₂ Br ₂	95.2	0.9252
C ₂ H ₅ I (H _a)	97.78	0.9337	C ₆ H ₁₁ I (H _a)	95.650	0.9158	CH ₂ Cl ₂	93.8	0.9201
C ₂ H ₅ I (H _b)	87.4308	0.9229	C ₆ H ₁₁ I (H _b)	80.73	0.9222	CH ₂ F ₂	99.1	0.9452
C ₃ H ₇ Br (H _a)	94.1239	0.9158	C ₆ H ₁₁ I (H _c)	96.118	0.9286	CH ₂ I ₂	97.6	0.9313
C ₃ H ₇ Br (H _b)	96.1178	0.9254	C ₆ H ₁₁ I (H _d)	96.366	0.9286	CH ₃ Br	98.9	0.9437
C ₃ H ₇ Cl (H _a)	93.1873	0.9139	C ₆ H ₅ Br (H _b)	112.5727	0.9193	CH ₃ Cl	97.5	0.9414
C ₃ H ₇ Cl (H _b)	98.7035	0.9298	C ₆ H ₅ Br (H _c)	111.3946	0.9264	CH ₃ COBr	101.9148	0.9113
C ₃ H ₇ F (H _a)	94.6661	0.9259	C ₆ H ₅ Br (H _d)	112.2021	0.9268	CH ₃ COCH ₂ Br (H _a)	87.2073	0.9092
C ₃ H ₇ F (H _b)	101.2961	0.9378	C ₆ H ₅ Cl (H _b)	113.1296	0.92	CH ₃ COCH ₂ Br (H _c)	94.4225	0.9323
C ₃ H ₇ I (H _a)	95.419	0.9189	C ₆ H ₅ Cl (H _c)	111.5567	0.9266	CH ₃ COCl	95.3537	0.9125
C ₃ H ₇ I (H _b)	83.83	0.9221	C ₆ H ₅ Cl (H _d)	112.2784	0.927	CH ₃ COF	96.0657	0.9161
C ₄ H ₉ Br (H _b)	90.3217	0.9232	C ₆ H ₅ F (H _b)	114.0035	0.92	CH ₃ COI	94.956	0.9104
C ₄ H ₉ Cl (H _b)	98.6455	0.9281	C ₆ H ₅ F (H _c)	111.687	0.9276	CH ₃ F	99.1	0.9564

C ₄ H ₉ I (H _b)	81.26	0.9198	C ₆ H ₅ F (H _d)	112.5339	0.9269	CH ₃ I	100.9	0.9462
C ₅ H ₄ NCl (H _a)	111.1449	0.9252	C ₆ H ₅ I (H _b)	111.79	0.9194	CHBr ₃	91	0.9063
C ₅ H ₄ NCl (H _b)	113.1372	0.9238	C ₆ H ₅ I (H _c)	111.273	0.9258	CHCl ₃	90.9	0.8966
C ₅ H ₄ NCl (H _c)	106.6584	0.9238	C ₆ H ₅ I (H _d)	112.110	0.9268	CHF ₃	103.9	0.9306
C ₅ H ₄ NCl (H _d)	113.097	1.0471	C ₆ H ₆ NCl (H _a)	84.296	0.9226	CHI ₃	93	0.9216
C ₅ H ₄ NI (H _a)	106.541	0.9182	C ₆ H ₆ NCl (H _c)	111.9215	0.9212	CNCH ₂ Br	87.8256	0.8966
C ₅ H ₄ NI (H _b)	111.082	0.923	C ₆ H ₆ NCl (H _d)	110.8864	0.9266	CNCH ₂ Cl	86.6625	0.8946
C ₅ H ₄ NI (H _c)	111.5875	0.9229	C ₆ H ₆ NCl (H _e)	112.4097	0.9247	CNCH ₂ F	88.6433	0.9113
C ₅ H ₄ NI (H _d)	106.4375	1.0058	C ₆ H ₆ NCl (H _f)	105.9481	0.9242	CNCH ₂ I	89.407	0.8989
C ₅ H ₉ Br (H _a)	92.985	0.9183	C ₇ H ₇ Br (H _a)	86.1632	0.9163	SC ₂ H ₅ Cl (H _a)	92.1362	0.9178
C ₅ H ₉ Br (H _b)	88.0989	0.9131	C ₇ H ₇ Br (H _c)	111.9483	0.9143	SC ₂ H ₅ Cl (H _c)	93.3693	0.93
C ₅ H ₉ Br (H _c)	94.1498	0.9337	C ₇ H ₇ Br (H _d)	111.6049	0.928	SC ₂ H ₅ I (H _a)	94.619	0.9224
C ₅ H ₉ Cl (H _a)	92.0736	0.9168	C ₇ H ₇ Br (H _e)	111.7808	0.9278	SC ₂ H ₅ I (H _c)	93.145	0.9264
C ₅ H ₉ Cl (H _b)	91.9867	0.9175	C ₇ H ₇ Cl (H _a)	84.2431	0.9258			

Table S3. The calculated BDE_{C-X} and BO_{C-X} .

Bond type	BDE	BO	Bond type	BDE	BO	Bond type	BDE	BO
C ₁₀ H ₁₅ F	118.106073	0.7914	C ₆ H ₅ Cl	94.99364313	1.056	SC ₂ H ₅ Br	62.585915	0.946
C ₂ H ₅ F	111.0421986	0.826	C ₆ H ₆ NCl	69.13774169	0.9827	CHBr ₃	59.2	1.0145
C ₃ H ₇ F	112.2049737	0.8025	C ₇ H ₇ Cl	68.88171782	0.9701	CH ₂ Br ₂	65.2	1.0259
C ₅ H ₄ NF	124.9703995	0.8731	CH ₃ COCl	82.33426648	0.954	CH ₃ Br	70.6	1.0235
C ₅ H ₉ F	110.2854221	0.7992	CNCH ₂ Cl	66.79650375	1.0045	C ₁₀ H ₁₅ I	62.939	0.9305
C ₆ H ₁₁ F	112.8306007	0.8078	SC ₂ H ₅ Cl	73.35711557	0.9714	C ₂ H ₅ I	57.77856472	0.9828
C ₆ H ₅ F	125.8432652	0.8746	CHCl ₃	70.7	1.035	C ₃ H ₇ I	56.8322804	0.9482
C ₇ H ₇ F	96.94519767	0.8125	CH ₂ Cl ₂	76.1	1.0373	C ₄ H ₉ I	55.6368748	0.9117
CH ₃ COF	119.3416392	0.8287	CH ₃ Cl	81.5	1.0268	C ₅ H ₄ NI	68.69974006	1.0058
CNCH ₂ F	95.09027959	0.8549	C ₁₀ H ₁₅ Br	74.7658744	0.9466	C ₅ H ₉ I	55.81634252	0.9437
SC ₂ H ₅ F	102.8230792	0.8169	C ₂ H ₅ Br	70.28733909	0.9939	C ₆ H ₁₁ I	58.44999989	0.9549
CHF ₃	123	0.8919	C ₃ H ₇ Br	69.40380572	0.96	C ₆ H ₅ I	68.88171781	1.0091
CH ₂ F ₂	116.7	0.8711	C ₄ H ₉ Br	68.25169828	0.927	C ₇ H ₇ I	46.91574777	0.9371
CH ₃ F	109.2	0.8488	C ₅ H ₄ NBr	77.52628869	1.0325	CH ₃ COI	56.67226547	0.892
C ₁₀ H ₁₅ Cl	86.94834383	0.9588	C ₅ H ₉ Br	68.13937408	0.9559	CNCH ₂ I	45.16374124	0.9489
C ₂ H ₅ Cl	81.68542165	0.9998	C ₆ H ₁₁ Br	70.83766493	0.9683	SC ₂ H ₅ I	50.75171334	0.9195
C ₃ H ₇ Cl	81.34154645	0.9695	C ₆ H ₅ Br	82.10961808	1.0441	CHI ₃	48.1	0.9694
C ₄ H ₉ Cl	80.50821383	0.9411	C ₇ H ₇ Br	58.11992989	1.0111	CH ₂ I ₂	53.5	0.997
C ₅ H ₄ NCl	90.68139784	1.0471	CH ₃ COBr	70.16685727	0.9272	CH ₃ I	58.3	1.0115
C ₅ H ₉ Cl	79.84932886	0.9649	CH ₃ COCH ₂ Br	59.9836331	1.0223			
C ₆ H ₁₁ Cl	82.66182643	0.9759	CNCH ₂ Br	56.30893747	0.9844			

2. Details of machine learning

2.1 Generation of transition states geometries and corresponding barriers

The workflow for DFT data generation is detailed in Figure S5. Initially, the organic halides

and radicals were optimized to obtain the equilibrium structures and ground-state energies of the reactants. The optimized geometries served as the basis for constructing initial transition state (TS) guesses. Specifically, Python scripts were employed to automatically assemble fragments of the organic halides and radicals into starting TS geometries for each target reaction. These guesses were then subjected to TS optimization. To ensure the localization of the correct HAT and XAT transition states, all output files were automatically validated; instances of optimization failure were rectified through manual adjustment of the guess geometries. Subsequent single-point energy calculations were performed on all verified TS structures. The automation scripts for geometry generation and output verification are hosted on our GitHub repository (<https://github.com/liusqchem/machine-learning.git>). The H/X selectivity of the carbon radicals is quantified by the relative activation barrier $\Delta\Delta G^\ddagger$, defined as the difference between the lowest-lying HAT and XAT transition states ($\Delta G^\ddagger_H - \Delta G^\ddagger_X$) for each organic halide substrate.

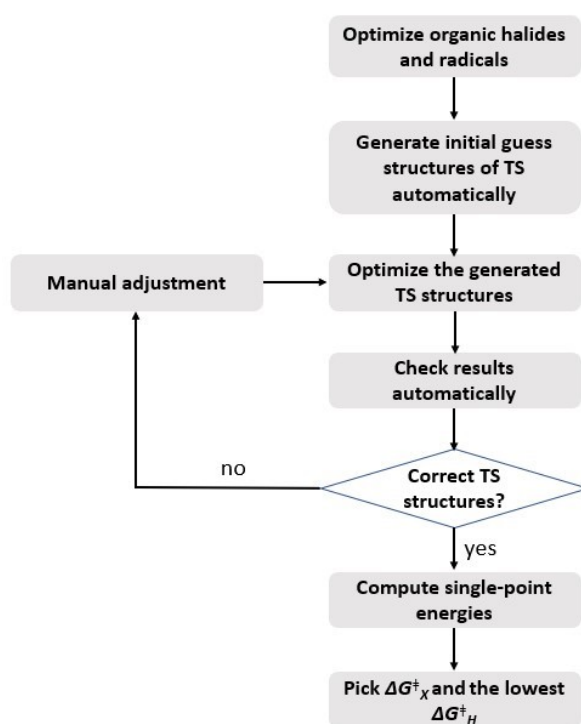


Figure S5. Flow chart of the DFT-computed data generation

2.2 Details and calculations of descriptors

We initially select 18 physical organic descriptors categorized into eight groups: frontier molecular orbital (*FMO*) energies, atom charges, Wiberg bond orders (*BO*), bond dissociation energies (*BDE*), reaction free energies (ΔG), spin densities, ionization potentials (*IP*) and electrostatic potentials (*ESP*). These descriptors were strategically chosen to capture both the thermodynamic and polar effects governing atom transfer reactions. Thermodynamic effects are represented by three categories comprising eight descriptors: the *BDE*s of the breaking and forming C-H/C-X bonds (4 descriptors), the Wiberg bond orders of the scissile C-H and C-X bonds (2 descriptors), and reaction energies for the HAT (ΔG_H) and XAT (ΔG_X) pathways (2 descriptors). Polar effects are encoded by the remaining ten descriptors across five categories: the spin density of carbon radical center (C_1), Natural Population Analysis (NPA) charges (encompassing the radical (C_1), the abstracted H/X atoms, and their adjacent carbon atom (C_2 , C_3); 5 descriptors), *FMO*

energies of the carbon radicals (E(LUMO) and E(SOMO) (i.e. E(HOMO)); 2 descriptors), and the *ESP* and *IP* of the carbon radicals (1 descriptor each).

2.2.1 Bond dissociation energy (*BDE*) and bond order (*BO*)

Tables S2 and S3 provide comprehensive computational details for the bond dissociation energies (*BDE*) of the scissile and nascent C–H/C–X bonds in both reactants and products. BDE_{C-H} and BDE_{C-X} were calculated as:



Additionally, bond order descriptors were derived from Wiberg bond indexes¹³. Our analysis explicitly incorporates the BO_{C-H} and BO_{C-X} values for each potential reacting carbon center.

2.2.2 Frontier orbital energy and ionization potentials

Table S4 includes the calculation details of the **FMO** energy-based descriptors. The SOMO(HOMO) and LUMO energies for radicals were obtained from single-point energy calculation. The calculated gas-phase ionization potentials (eV) were computed as indicated in Eq. (1), -0.03766 eV is the energy value of a free electron at 298K.

$$IP \text{ (eV)} = (H_{\text{radical cation}} - H_{\text{radical}}) - 0.03766 \quad \text{Eq. (1)}$$

2.2.3 Spin, atomic charge and electrostatic potentials

Natural bond orbital (*NBO*) analysis was conducted to obtain the atomic charge information. Charge (C_1) in Table S4 refers to the NPA charge¹⁴ of radical-center carbon atom. Charge (H), charge (C_2), charge(X) and charge(C_3) in Table S5 refer to the NPA charge of abstracted H atom, the adjacent carbon atom(C_2), abstracted halogen atom and adjacent carbon atom(C_3). Spin (C_1) in Table S4 refers to Mulliken atomic spin population¹⁵ of radical-center atom. Electrostatic potentials(*ESP*) of radicals in Table S4 are calculated using Multiwfn¹⁶.

2.2.4 Database for descriptors

Table S4. Phys-org descriptors for radicals.

Radical	Spin (C_1)	Charge (C_1)	IP	E(SOMO)	E(LUMO)	ESP
1	1.083	-0.468	9.8	-0.23781	-0.08075	3.19233
2	1.057	-0.272	8.1	-0.20781	-0.06085	2.55279
3	1.013	-0.089	7.3	-0.18834	-0.04861	2.28046
4	1.04	-0.077	7.1	-0.18846	-0.05038	2.1719
5	0.967	0.071	6.7	-0.17686	-0.0426	2.77247
6	0.883	-0.253	6.3	-0.15677	-0.0285	4.3306
7	0.905	-0.032	5.1	-0.13958	-0.03085	2.61164
8	0.859	-0.032	4.9	-0.13397	-0.02748	3.04769
9	0.894	-0.045	4.7	-0.12858	-0.01797	1.50218
10	0.845	-0.247	5.6	-0.1504	-0.05433	3.4703
11	0.88	-0.21	6.8	-0.18295	-0.07161	3.66203
12	0.857	0.148	5.2	-0.14351	-0.02473	3.5872
13	0.931	-0.061	7.6	-0.18726	-0.04559	4.45321
14	0.924	-0.063	7.0	-0.18146	-0.04818	3.90981
15	0.916	0.317	6.1	-0.16612	-0.03462	3.85535

16	0.875	0.388	5.6	-0.17078	-0.0413	3.18867
17	0.882	-0.239	8.3	-0.27007	-0.13799	3.75155
18	0.947	-0.253	8.4	-0.26757	-0.13399	3.96379
19	0.964	-0.267	8.4	-0.24525	-0.11988	3.69976
20	0.773	0.093	7.6	-0.2268	-0.11256	2.74079
21	0.664	0.168	7.9	-0.22052	-0.10516	4.26836
22	0.747	-0.129	9.6	-0.2924	-0.19318	4.62418
23	0.688	-0.31	8.0	-0.20653	-0.09018	3.86071
24	0.798	-0.351	8.6	-0.2233	-0.10521	3.78623
25	0.797	-0.284	7.1	-0.19203	-0.09497	3.79475
26	0.759	-0.259	7.6	-0.2173	-0.12716	3.79665
27	0.797	-0.312	6.3	-0.17423	-0.08229	3.8505
28	0.647	0.187	6.3	-0.1658	-0.07376	4.50266
29	0.633	0.256	6.8	-0.19375	-0.10072	3.74628
30	0.625	0.371	5.7	-0.15627	-0.06814	3.42916
31	0.597	0.361	6.0	-0.15868	-0.06929	3.61018
32	0.652	-0.11	6.3	-0.17965	-0.10143	4.1306
33	0.629	-0.077	6.8	-0.19498	-0.11999	4.70461
34	0.543	0.022	5.3	-0.14707	-0.07331	3.08519
35	1.055	-0.054	8.5	-0.23292	-0.07954	14.11783
36	1.061	0.17	8.1	-0.25035	-0.09384	15.87553
37	1.124	0.463	9.0	-0.25046	-0.13057	17.81607
38	0.995	0.315	7.7	-0.22646	-0.08468	16.51412
39	1.057	-0.068	5.1	-0.12162	-0.02123	3.35522
40	0.409	0.036	5.2	-0.14153	-0.07161	3.06249
41	0.937	-0.292	8.7	-0.25475	-0.09153	13.91240
42	0.633	-0.521	7.9	-0.22309	-0.12405	14.48714
43	0.998	1.263	6.1	-0.18505	-0.07384	9.48910
44	0.917	0.82	6.6	-0.19973	-0.10169	10.83672
45	0.918	-0.095	6.1	-0.18941	-0.09742	6.48321
46	0.894	0.713	5.5	-0.17703	-0.08298	8.47136
47	0.981	0.108	6.5	-0.24607	-0.15358	31.09472
48	0.705	-0.135	9.3	-0.25454	-0.15118	40.49280
49	0.646	-0.184	8.6	-0.23816	-0.13939	38.27280
50	0.566	-0.203	7.8	-0.22791	-0.14265	34.90718

Table S5. Charge for organic halides.

Atom position	C ₂ (C ₃)	H(X)	Atom position	C ₂ (C ₃)	H(X)	Atom position	C ₂ (C ₃)	H(X)
C ₁₀ H ₁₅ Br (H _b)	-0.404	0.209	C ₅ H ₉ I (H _a)	-0.296	0.209	C ₇ H ₇ I (H _a)	-0.482	0.216
C ₁₀ H ₁₅ Br (H _c)	-0.218	0.203	C ₅ H ₉ I (H _b)	-0.41	0.216	C ₇ H ₇ I (H _c)	-0.182	0.204
C ₁₀ H ₁₅ Br (H _d)	-0.373	0.198	C ₅ H ₉ I (H _c)	-0.39	0.204	C ₇ H ₇ I (H _d)	-0.194	0.205
C ₁₀ H ₁₅ Br (Br)	-0.04	-0.031	C ₅ H ₉ I (I)	-0.296	0.042	C ₇ H ₇ I (H _e)	-0.194	0.205
C ₁₀ H ₁₅ Cl (H _b)	-0.399	0.209	C ₆ H ₁₁ Br (H _a)	-0.21	0.195	C ₇ H ₇ I (I)	-0.482	0.058

C ₁₀ H ₁₅ Cl (H _c)	-0.219	0.202	C ₆ H ₁₁ Br (H _b)	-0.403	0.21	CH ₂ Br ₂ (H)	-0.535	0.212
C ₁₀ H ₁₅ Cl (H _d)	-0.372	0.198	C ₆ H ₁₁ Br (H _c)	-0.375	0.186	CH ₂ Br ₂ (Br)	-0.535	0.056
C ₁₀ H ₁₅ Cl (Cl)	0.008	-0.087	C ₆ H ₁₁ Br (H _d)	-0.379	0.2	CH ₂ Cl ₂ (H)	-0.38	0.205
C ₂ H ₅ Br (H _a)	-0.396	0.199	C ₆ H ₁₁ Br (Br)	-0.21	-0.033	CH ₂ Cl ₂ (Cl)	-0.38	-0.016
C ₂ H ₅ Br (H _b)	-0.592	0.207	C ₆ H ₁₁ Cl (H _a)	-0.151	0.19	CH ₂ F ₂ (H)	0.499	0.125
C ₂ H ₅ Br (Br)	-0.396	-0.033	C ₆ H ₁₁ Cl (H _b)	-0.399	0.198	CH ₂ F ₂ (F)	0.499	-0.375
C ₂ H ₅ Cl (H _a)	-0.331	0.195	C ₆ H ₁₁ Cl (H _c)	-0.376	0.202	CH ₂ I ₂ (H)	-0.771	0.214
C ₂ H ₅ Cl (H _b)	-0.59	0.206	C ₆ H ₁₁ Cl (H _d)	-0.378	0.187	CH ₂ I ₂ (I)	-0.771	0.171
C ₂ H ₅ Cl (Cl)	-0.331	-0.089	C ₆ H ₁₁ Cl (Cl)	-0.151	-0.09	CH ₃ Br(H)	-0.601	0.206
C ₂ H ₅ F (H _a)	0.1	0.151	C ₆ H ₁₁ I (H _a)	-0.292	0.197	CH ₃ Br (Br)	-0.601	-0.018
C ₂ H ₅ F (H _b)	-0.611	0.205	C ₆ H ₁₁ I (H _b)	-0.408	0.198	CH ₃ Cl(H)	-0.527	0.201
C ₂ H ₅ F (F)	0.1	-0.408	C ₆ H ₁₁ I (H _c)	-0.374	0.203	CH ₃ Cl (Cl)	-0.527	-0.076
C ₂ H ₅ I (H _a)	-0.496	0.203	C ₆ H ₁₁ I (H _d)	-0.38	0.187	CH ₃ COBr(H)	-0.675	0.229
C ₂ H ₅ I (H _b)	-0.595	0.207	C ₆ H ₁₁ I (I)	-0.292	0.059	CH ₃ COBr (Br)	0.527	-0.071
C ₂ H ₅ I (I)	-0.496	0.065	C ₆ H ₅ Br (H _b)	-0.226	0.219	CH ₃ COCH ₂ Br (H _a)	-0.525	0.228
C ₃ H ₇ Br (H _a)	-0.219	0.195	C ₆ H ₅ Br (H _c)	-0.185	0.207	CH ₃ COCH ₂ Br (H _c)	-0.668	0.222
C ₃ H ₇ Br (H _b)	-0.589	0.204	C ₆ H ₅ Br (H _d)	-0.205	0.207	CH ₃ COCH ₂ Br (Br)	-0.525	0.033
C ₃ H ₇ Br (Br)	-0.219	-0.046	C ₆ H ₅ Br (Br)	-0.106	0.073	CH ₃ COCl(H)	-0.674	0.229
C ₃ H ₇ Cl (H _a)	-0.164	0.19	C ₆ H ₅ Cl (H _b)	-0.226	0.219	CH ₃ COCl (Cl)	0.569	-0.104
C ₃ H ₇ Cl (H _b)	-0.586	0.213	C ₆ H ₅ Cl (H _c)	-0.185	0.207	CH ₃ COF(H)	-0.697	0.237
C ₃ H ₇ Cl (Cl)	-0.164	-0.098	C ₆ H ₅ Cl (H _d)	-0.206	0.207	CH ₃ COF (F)	0.908	-0.382
C ₃ H ₇ F (H _a)	0.239	0.149	C ₆ H ₅ Cl (Cl)	-0.035	0.003	CH ₃ COI(H)	-0.678	0.243
C ₃ H ₇ F (H _b)	-0.602	0.204	C ₆ H ₅ F (H _b)	-0.267	0.22	CH ₃ COI (I)	0.471	-0.014
C ₃ H ₇ F (F)	0.239	-0.418	C ₆ H ₅ F (H _c)	-0.183	0.206	CH ₃ F(H)	-0.071	0.155
C ₃ H ₇ I (H _a)	-0.304	0.196	C ₆ H ₅ F (H _d)	-0.219	0.207	CH ₃ F (F)	-0.071	-0.394
C ₃ H ₇ I (H _b)	-0.594	0.213	C ₆ H ₅ F (F)	0.417	-0.358	CH ₃ I(H)	-0.717	0.21
C ₃ H ₇ I (I)	-0.304	0.048	C ₆ H ₅ I (H _b)	-0.23	0.218	CH ₃ I (I)	-0.717	0.089
C ₄ H ₉ Br (H _b)	-0.591	0.213	C ₆ H ₅ I (H _c)	-0.186	0.208	CHBr ₃ (H)	-0.546	0.208
C ₄ H ₉ Br (Br)	-0.059	-0.054	C ₆ H ₅ I (H _d)	-0.203	0.207	CHBr ₃ (Br)	-0.546	0.113
C ₄ H ₉ Cl (H _b)	-0.587	0.211	C ₆ H ₅ I (I)	-0.204	0.181	CHCl ₃ (H)	-0.302	0.032
C ₄ H ₉ Cl (Cl)	-0.013	-0.102	C ₆ H ₆ NCl (H _a)	-0.346	0.207	CHCl ₃ (Cl)	-0.302	0.204
C ₄ H ₉ I (H _b)	-0.597	0.203	C ₆ H ₆ NCl (H _c)	-0.22	0.215	CHF ₃ (H)	0.958	0.11
C ₄ H ₉ I (I)	-0.13	0.04	C ₆ H ₆ NCl (H _d)	-0.159	0.208	CHF ₃ (F)	0.958	-0.356
C ₅ H ₄ NCl (H _a)	-0.254	0.226	C ₆ H ₆ NCl (H _e)	-0.236	0.212	CHI ₃ (H)	-0.897	0.201
C ₅ H ₄ NCl (H _b)	-0.151	0.211	C ₆ H ₆ NCl (H _f)	0.064	0.186	CHI ₃ (I)	-0.897	0.232
C ₅ H ₄ NCl (H _c)	-0.247	0.214	C ₆ H ₆ NCl (Cl)	-0.346	-0.076	CNCH ₂ Br(H)	-0.523	0.249
C ₅ H ₄ NCl (H _d)	0.071	0.19	C ₇ H ₇ Br (H _a)	-0.39	0.213	CNCH ₂ Br (Br)	-0.523	0.056
C ₅ H ₄ NCl (Cl)	-0.491	0.82	C ₇ H ₇ Br (H _c)	-0.181	0.205	CNCH ₂ Cl (H)	-0.45	0.245
C ₅ H ₄ NI (H _a)	0.036	0.197	C ₇ H ₇ Br (H _d)	-0.195	0.205	CNCH ₂ Cl (Cl)	-0.45	-0.013
C ₅ H ₄ NI (H _b)	-0.193	0.221	C ₇ H ₇ Br (H _e)	-0.194	0.205	CNCH ₂ F (H)	0.006	0.202
C ₅ H ₄ NI (H _c)	-0.226	0.215	C ₇ H ₇ Br (Br)	-0.39	-0.034	CNCH ₂ F (F)	0.006	-0.364
C ₅ H ₄ NI (H _d)	0.055	0.188	C ₇ H ₇ Cl (H _a)	-0.328	0.207	CNCH ₂ I (H)	-0.628	0.251
C ₅ H ₄ NI (I)	-0.251	0.197	C ₇ H ₇ Cl (H _c)	-0.181	0.205	CNCH ₂ I (I)	-0.628	0.165
C ₅ H ₉ Br (H _a)	-0.213	0.207	C ₇ H ₇ Cl (H _d)	-0.195	0.205	SC ₂ H ₅ Cl (H _a)	-0.525	0.218

C ₅ H ₉ Br (H _b)	-0.405	0.217	C ₇ H ₇ Cl (H _c)	-0.193	0.205	SC ₂ H ₅ Cl (H _c)	-0.707	0.22
C ₅ H ₉ Br (H _c)	-0.389	0.204	C ₇ H ₇ Cl (Cl)	-0.328	-0.088	SC ₂ H ₅ Cl (Cl)	-0.525	-0.079
C ₅ H ₉ Br (Br)	-0.213	-0.049	C ₇ H ₇ F (H _a)	0.089	0.163	SC ₂ H ₅ I (H _a)	-0.697	0.225
C ₅ H ₉ Cl (H _a)	-0.156	0.203	C ₇ H ₇ F (H _c)	-0.178	0.204	SC ₂ H ₅ I (H _c)	-0.71	0.221
C ₅ H ₉ Cl (H _b)	-0.402	0.216	C ₇ H ₇ F (H _d)	-0.199	0.205	SC ₂ H ₅ I (I)	-0.697	0.052
C ₅ H ₉ Cl (H _c)	-0.389	0.203	C ₇ H ₇ F (H _e)	-0.19	0.205			
C ₅ H ₉ Cl (Cl)	-0.156	-0.103	C ₇ H ₇ F (F)	0.089	-0.402			

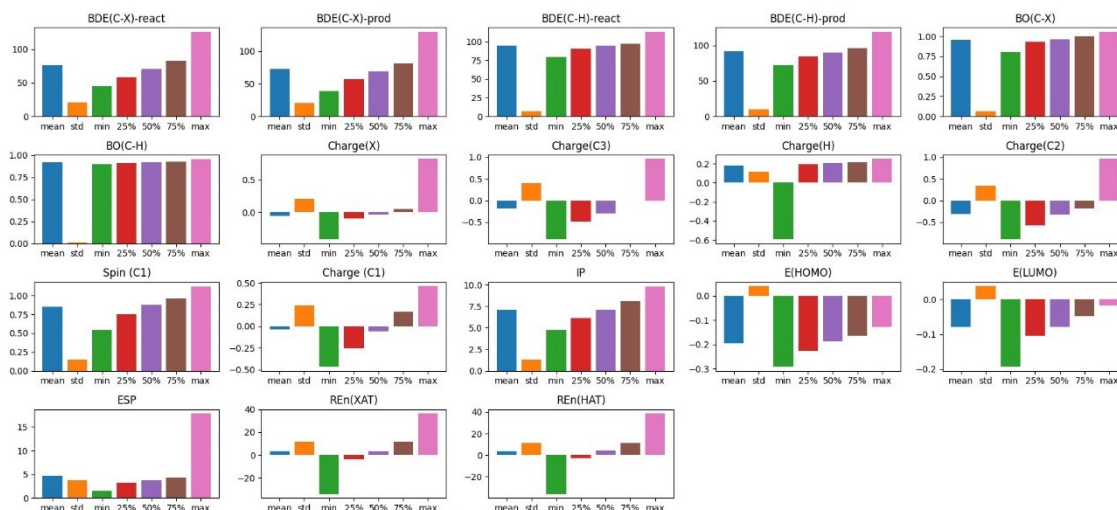


Figure S6. Distribution of phys-organic descriptors

2.3 Feature reduction based on Pearson relation coefficient

Pearson correlation analysis was performed as part of the feature selection process. The correlation coefficients for the 18 initial physical organic descriptors are presented in Figure S7, illustrating the degree of linear relationship between the features.

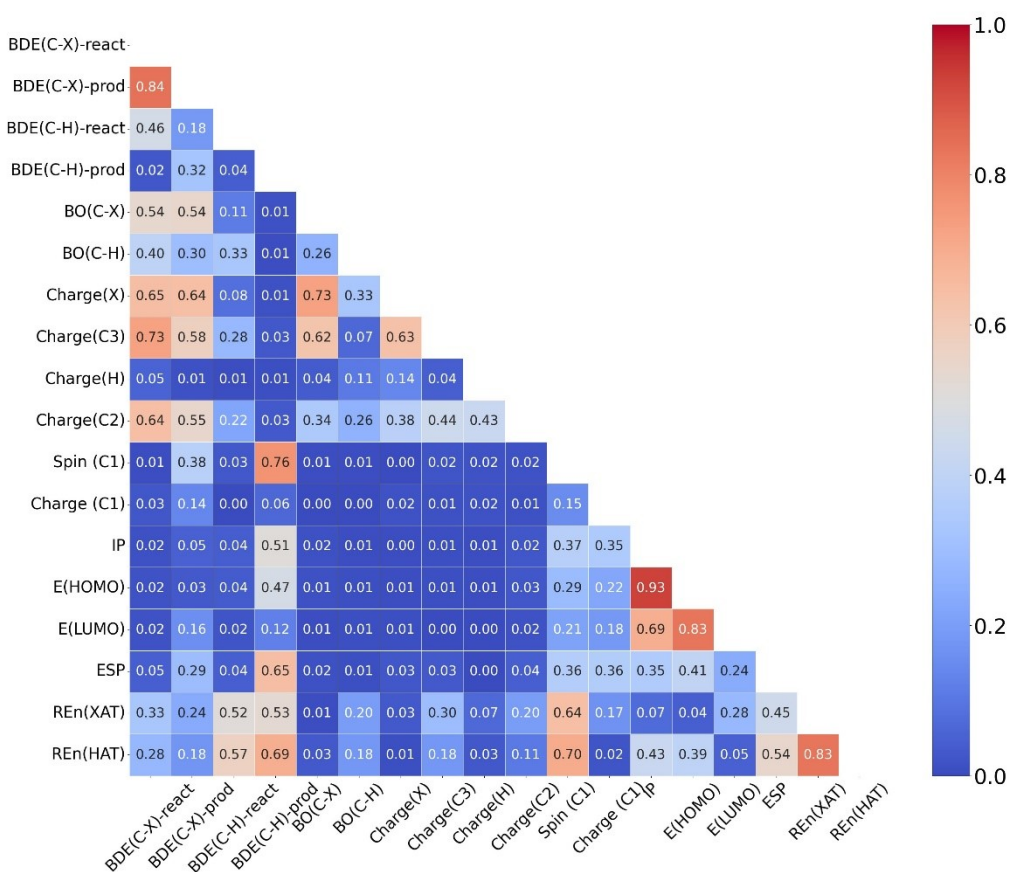


Figure S7. Pearson correlation coefficients of descriptors.

2.4 Machine learning algorithms

A series of widely used machine learning algorithms were evaluated for model development, including Bagging¹⁷, Gradient Boosting (GB)^{18, 19}, Extra Tree, *k*-nearest neighbors regressor (KNN)²⁰, Adaboost²¹, Multivariate Linear Regression (MLR)²², Ridge Regressor²³, Lasso²⁴, Support Vector Machine (SVM)²⁵, Decision Tree (DT)²⁶, Random Forest (RF)²⁷. All models were implemented using the scikit-learn Python library²⁸. Detailed hyperparameter configurations and ensemble settings for each algorithm are provided in Table S6; for any parameters not explicitly listed, default scikit-learn settings were maintained. The complete set of model-training scripts is hosted on our GitHub repository (<https://github.com/liusqchem/machine-learning.git>).

Table S6. 10-fold cross validation performance and hyperparameters for the evaluated machine learning algorithms.

Model	R ²	MAE	RMSE	Main hyperparameters
Bagging	0.98852	0.86511	1.15737	estimator=None, n_estimators=10, max_samples=1.0, max_features=1.0,
Gradient Boosting	0.98856	0.8838	1.15506	loss='squared_error', n_estimators=100
Extra Tree	0.97349	1.20865	1.75862	n_estimators=100, criterion='squared_error', min_samples_split=2
<i>k</i>-nearest neighbors regressor	0.95307	1.85303	2.33995	n_neighbors=5, weights='uniform', leaf_size=30

Adaboost	0.94562	2.08772	2.5189	n_estimators=50, learning_rate=1.0, loss='linear', random_state=None
Multivariate Linear Regressor	0.91773	2.39126	3.09815	-
Ridge Regressor	0.91716	2.39927	3.10893	alpha=1.0
Lasso	0.69257	4.43712	5.98901	alpha=1.0'
Support Vector Machine	0.93982	1.93337	2.64977	kernel='rbf', degree=3
Decision Tree	0.98173	1.06747	1.45981	criterion='squared_error', min_samples_split=2
Random Forest	0.99051	0.78561	1.05247	n_estimators=100, criterion='squared_error', min_samples_split=2,

2.5 Feature importance

2.5.1 Feature importance scores

The relative importance was evaluated using the feature importance metrics from RF, GB, Extra Trees, DT, and AdaBoost. Table S7 summarizes these scores, providing a basis for interpreting the chemical drivers of HAT/XAT selectivity.

Table S7. Feature importance scores of RF, GB, Extra tree, DT, AdaBoost and Bagging regressors.

Feature	RF	AdaBoost	GB	DT	Extra Tree	Bagging
BDE(C-X)	0.67873	0.78153	0.58028	0.82925	0.60333	0.71955
BO(C-X)	0.16003	0.03626	0.26268	0.00613	0.19964	0.11755
IP	0.08507	0.09943	0.07969	0.08121	0.07626	0.08496
E(LUMO)	0.03489	0.04317	0.03973	0.03898	0.02762	0.03516
Charge C2	0.009	0.0132	0.01421	0.01213	0.04274	0.00937
BDE(C-H)	0.00844	0.00606	0.00845	0.00833	0.01201	0.00852
ESP	0.00567	0.00285	0.00294	0.00452	0.00954	0.00611
Spin C1	0.00524	0.01026	0.00519	0.00415	0.00581	0.00548
Charge H	0.00515	0.00197	0.00335	0.00766	0.00993	0.00515
BO(C-H)	0.00436	0.0033	0.00211	0.00313	0.00671	0.00454
Charge C1	0.00342	0.00198	0.00136	0.00451	0.00641	0.00362

2.5.2 Influence of feature set size

We investigated the impact of the number of input features on model performance by adding descriptors sequentially according to their RF-derived importance in Table S8. We found that utilizing the top seven features provides an ideal balance between simplicity and accuracy, delivering an R^2 of 0.99 and an MAE of 0.82 kcal/mol for predicting $\Delta\Delta G^\ddagger$. This suggests that a subset of the most influential descriptors can effectively represent the reaction energetics.

Table S8. Influence of feature set size on the predictive accuracy (MAE and R^2) of the RF regressor.

Feature number	R^2	MAE	Detailed features
2	0.85592	3.29731	BDE(C-X), BO(C-X)
3	0.98045	1.13479	BDE(C-X), BO(C-X), IP
4	0.98603	0.95918	BDE(C-X), BO(C-X), IP, E(LUMO)

5	0.98685	0.93142	BDE(C-X), BO(C-X), IP, E(LUMO), Charge (C ₂)
6	0.98722	0.91648	BDE(C-X), BO(C-X), IP, E(LUMO), Charge (C ₂), BDE(C-H)
7	0.98913	0.82468	BDE(C-X), BO(C-X), IP, E(LUMO), Charge (C ₂), BDE(C-H), ESP
8	0.98961	0.81308	BDE(C-X), BO(C-X), IP, E(LUMO), Charge (C ₂), BDE(C-H), ESP, Spin (C ₁)
9	0.98977	0.8059	BDE(C-X), BO(C-X), IP, E(LUMO), Charge (C ₂), BDE(C-H), ESP, Spin (C ₁), Charge (H)
10	0.98981	0.80482	BDE(C-X), BO(C-X), IP, E(LUMO), Charge (C ₂), BDE(C-H), ESP, Spin (C ₁), Charge (H), BO(C-H)
11	0.99007	0.79587	BDE(C-X), BO(C-X), IP, E(LUMO), Charge (C ₂), BDE(C-H), ESP, Spin (C ₁), Charge H, BO(C-H), Charge (C ₁)

3. Results for silyl radicals

As shown in Figure S8A, the multivariate linear regression (MLR) model using BDE_{C-X} and IP as input features yields a relatively low correlation coefficient ($R^2 = 0.59$) for silyl radicals, suggesting that additional or alternative descriptors may better capture the selectivity trends in silyl radical-induced atom transfer reactions. To identify more suitable features, we repeated the machine-learning-guided workflow by training a model on 116 DFT-computed $\Delta\Delta G^\ddagger$ values and evaluating feature importance using the Random Forest (RF) algorithm.

The RF feature importance analysis revealed that BDE_{C-X} and BDE_{C-H} of the organic halides are the most influential descriptors for silyl radical reactivity. Based on this insight, we constructed a new MLR model using these two features. As illustrated in Figure S8B, this model shows a significantly improved correlation with DFT-computed values ($R^2 = 0.86$), indicating that both C-X and C-H bond strengths play dominant roles in determining atom transfer selectivity for silyl radicals. Figure S8B also reveals that silyl radicals preferentially abstract Cl, Br, and I atoms over H atoms, and a small number of successful F atom transfer (FAT) events are also observed. These results highlight the effectiveness of the machine-learning-guided approach in uncovering reactivity patterns across diverse radical species and atom transfer scenarios.

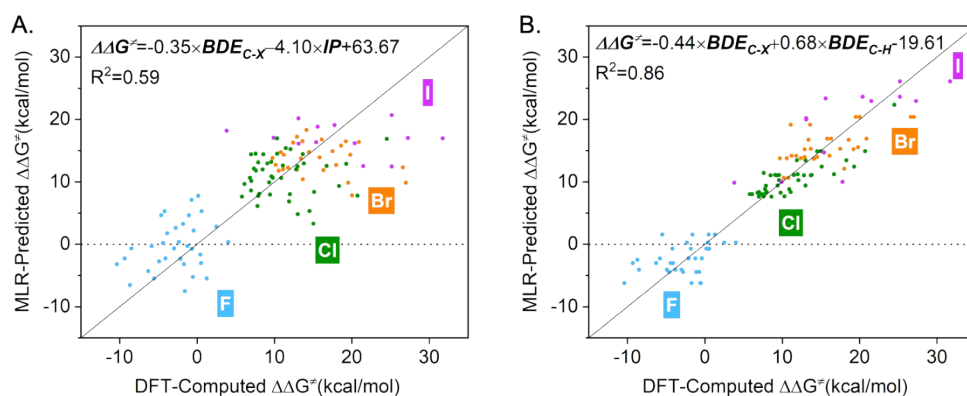


Figure S8. Correlation between DFT-computed $\Delta\Delta G^\ddagger$ values and multivariate linear regression predictions for silyl radicals using (A) BDE_{C-X} and IP , and (B) BDE_{C-X} and BDE_{C-H} as input features.

4. References

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