

Supporting Information

Bio-Inspired Branched Au–Cu Nanoalloys as Synergistic Dual-Pathway Peroxidase like activity

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Keywords-Au-Cu Nanoalloy, Coreduction, Green synthesis, Peroxidase mimicking, Multimodal sensing

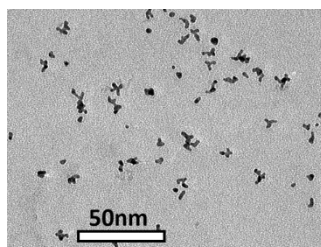


Fig S1a-TEM image after 1 year

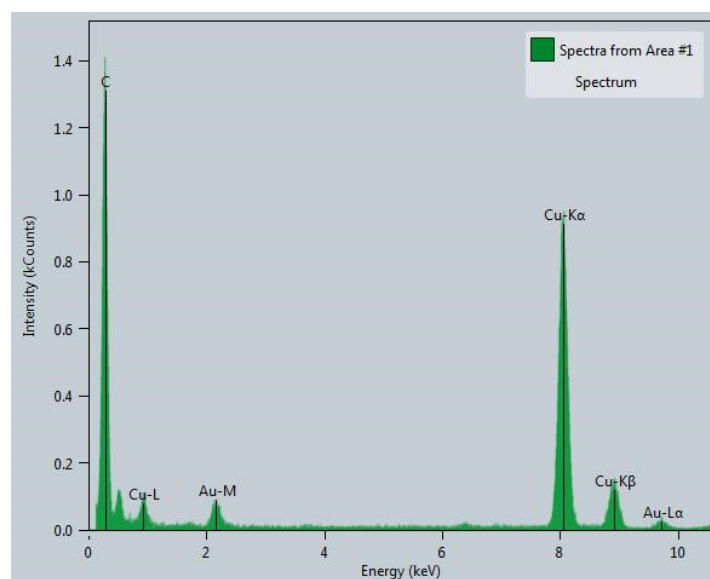


Fig S2 -HAADF EDS spectrum

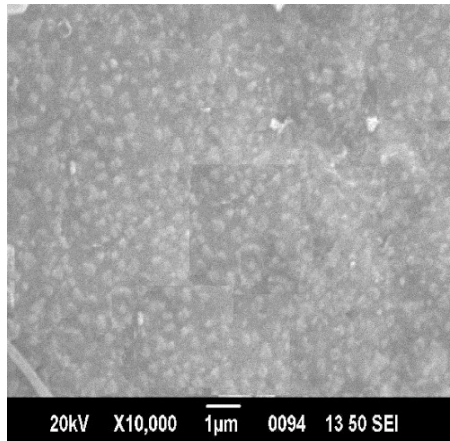


Figure S3-SEM image of Au-Cu NA

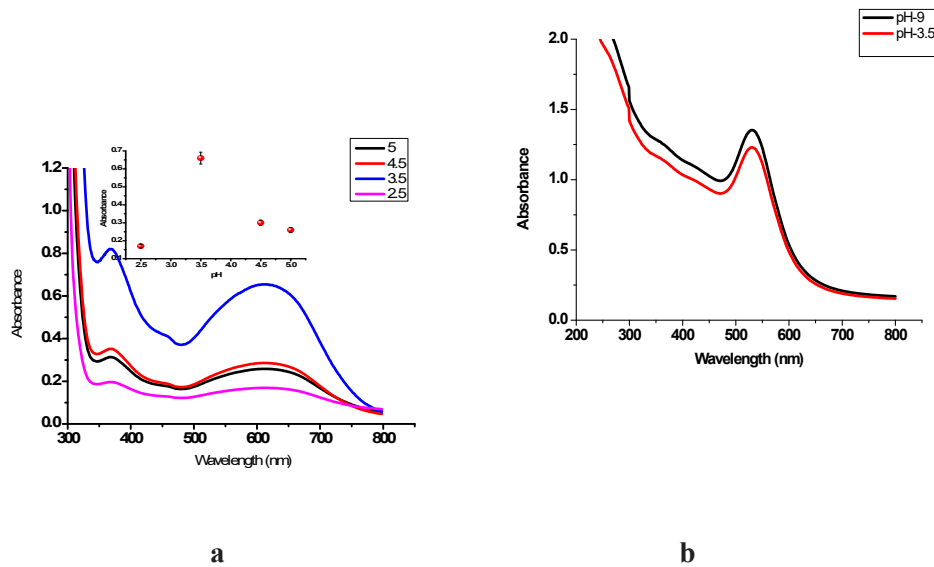


Fig S4 a- Plot of Absorbance for various pH conditions. Inset show the plot of abs vs pH b- UV-Vis spectrum of Au-Cu NA at pH 9 and 3.

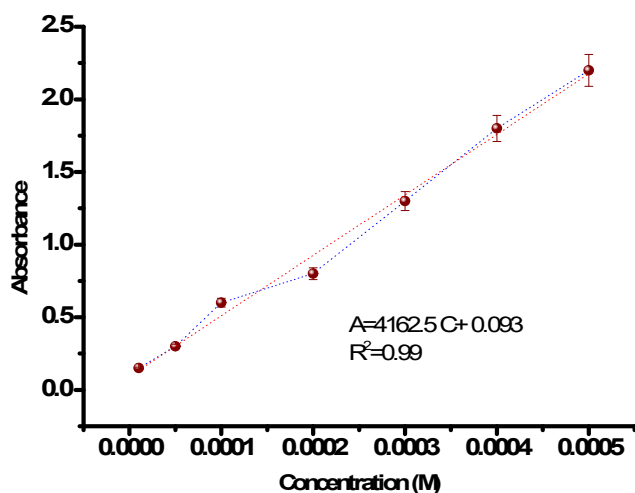


Fig S5- Plot of Abs vs Conc for various concentration of H₂O₂ from 10 μ M to 500 μ M

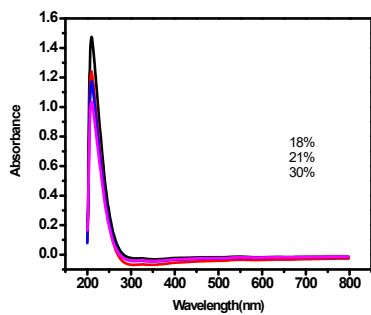
Radical scavenging assay

The hydrogen peroxide (H₂O₂) scavenging activity of the samples was evaluated according to the method described by Ruch et al. A 40 mM solution of H₂O₂ was prepared in phosphate buffer and mixed with the test samples. The remaining concentration of H₂O₂ was measured spectrophotometrically by recording the absorbance at 230 nm. The percentage of H₂O₂ scavenging activity was then calculated accordingly. A₀ represents the absorbance of the control sample, while A₁ denotes the absorbance after treatment with ascorbic acid, Cur-GNP and Au-Cu NA.

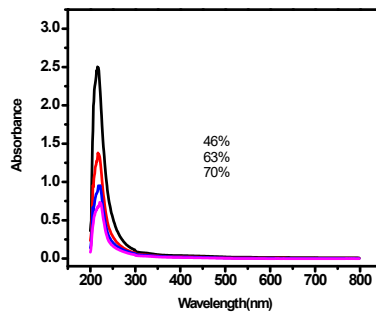
The figures below illustrate the radical scavenging activity of ascorbic acid, and Au-Cu NA. A progressive decrease in absorbance at 230 nm was observed with increasing concentrations of ascorbic acid and Au-CuNA, indicating enhanced H₂O₂ scavenging activity. Scavenging efficiency is found to be higher for Au-Cu NA. This shows that Au-Cu NA has good catalytic efficiency to rupture O-O bond of H₂O₂.

$$\frac{A_0 - A_1}{A_0} \times 100$$

(Reference-Keser, Serhat & Celik, Sait & Türkoğlu, Semra & Yilmaz, Okkes & Turkoglu, Ismail. (2012). Hydrogen Peroxide Radical Scavenging and Total Antioxidant Activity of Hawthorn. Chem J. 2.)

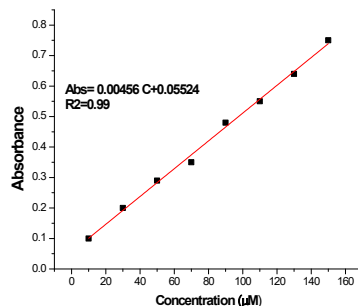
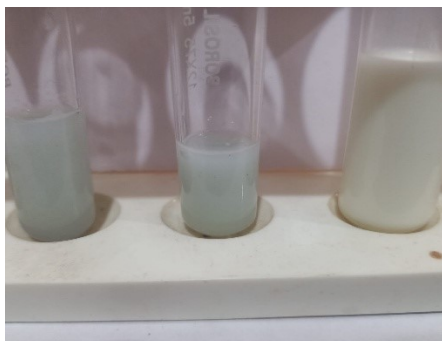


a



b

Fig S6-Radical scavenging efficiency of a- ascorbic acid b-Au-Cu NA



	Concentration Spiked	Recovered concentration	Recovery %
1	90µM	87 µM	96%
2	130 µM	125 µM	95%

Figure-S7 Real sample analysis and colour change observed for various concentrations of H_2O_2

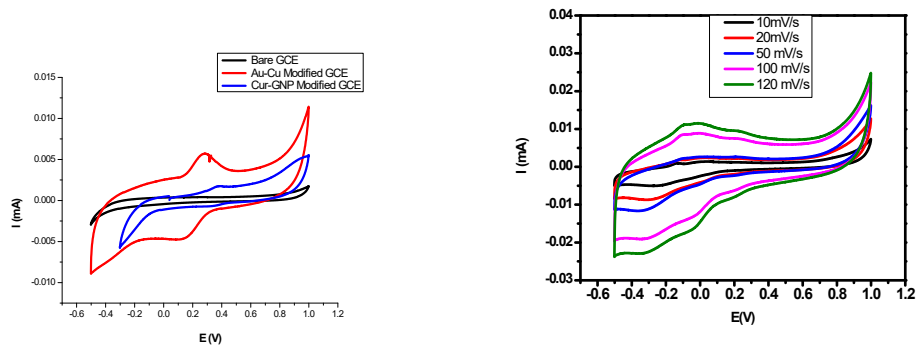


Fig S8 a- Cyclic voltametric response of GCE, Au-Cu nano particle and Gold nanoparticle modified electrode vs SCE. b-Scan rate variation.

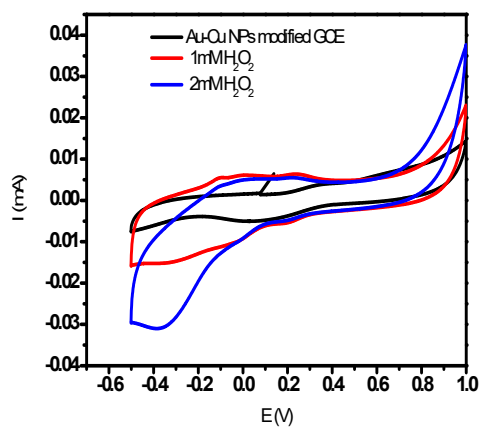
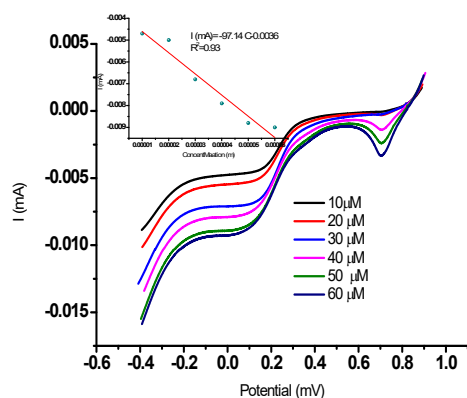


Fig S9- Cyclic voltametric response of Au-Cu modified electrode in various concentration of H_2O_2 .



FigS10- Calibration plot for the concentration range from 10-60 μM .

Table S1-Showing comparison of detection limits of various gold based nanozymes

SI No.	Type	Detection limit	Reference
1	Gold nano rattles	0.5 μM	1
2	Graphene-oxide gold	1.9 nM	2
3	Au@PtNP/GO	1.6 μM	3
4	Fe ₃ O ₄ @PB@Au	13.4 μM	4
5	Au@CMs	7 μM	5
6	Cu doped carbonized bacterial cellulose (BC) nanofibers	2.2 μM	6
7	gold nanoparticles deposited CeO ₂	3 μM	7

References

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