

Development of sphere-shaped bimetallic $\text{Ti}_{0.5}\text{Sr}_{0.5}\text{O}_3$ as an efficient bifunctional electrocatalyst for overall water splitting performance in alkaline media

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1. Synthesis of $\text{Ti}_{0.5}\text{Sr}_{0.5}\text{O}_3$

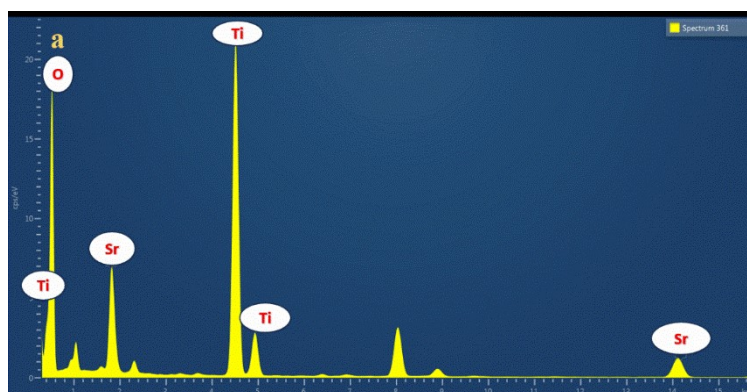


Figure S1. Schematic representation of the synthesis of $\text{Ti}_{0.5}\text{Sr}_{0.5}\text{O}_3$

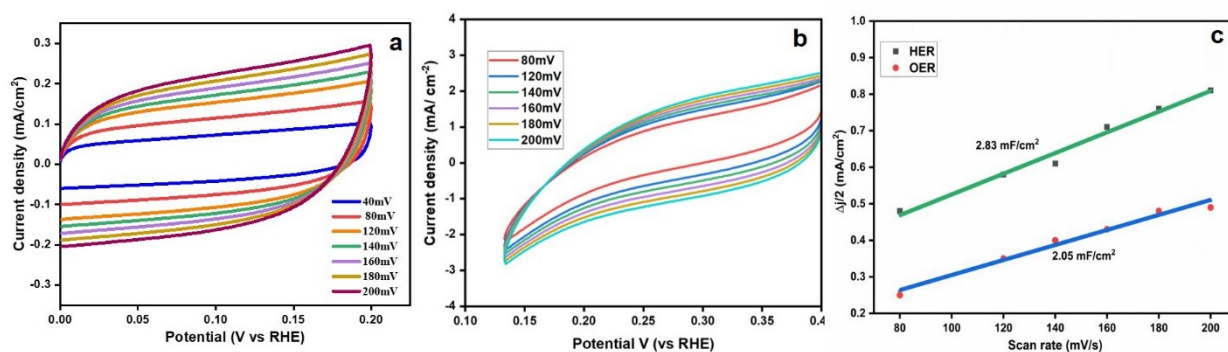
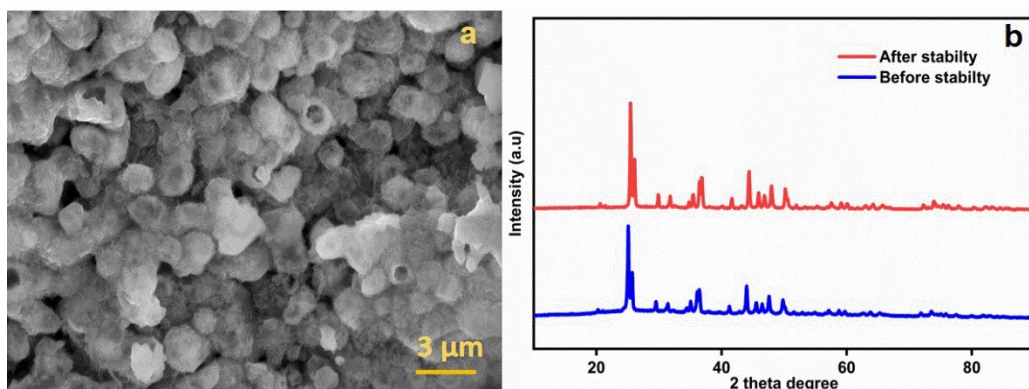


Figure S2 EDAX profile for $\text{Ti}_{0.5}\text{Sr}_{0.5}\text{O}_3$

Figure S3 (a) ECSA profile at various scan rates for OER in non-faradic region; (b) ECSA profile at various scan rates for HER in non-faradic region; (c) Cdl plot for

$\text{Ti}_{0.5}$
 $\text{Sr}_{0.5}$
 O_3/N
 F



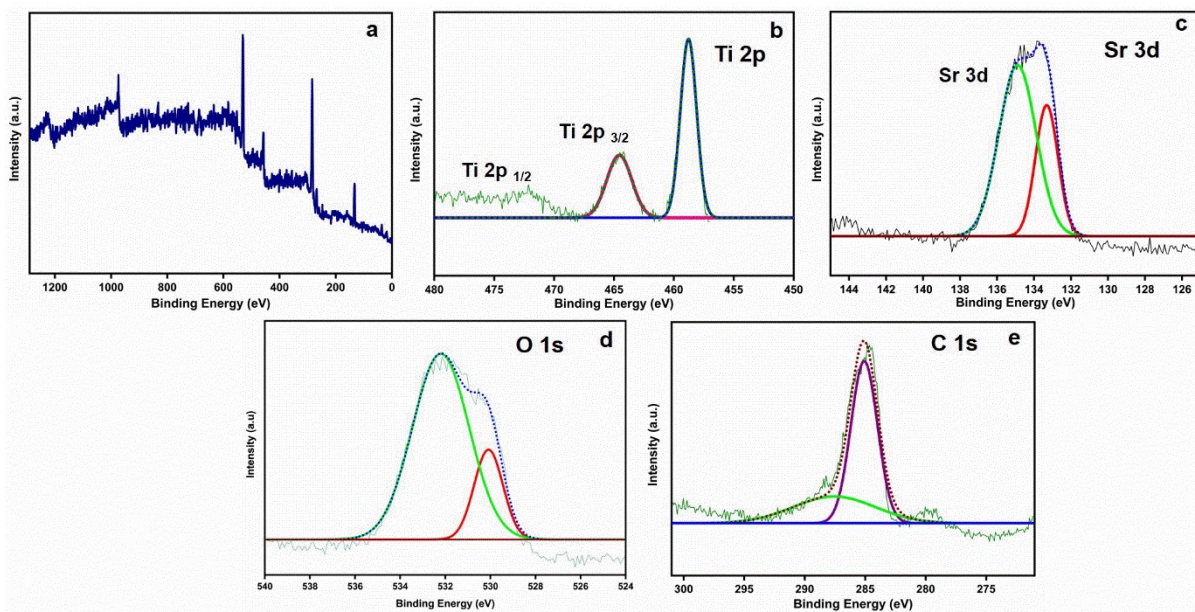
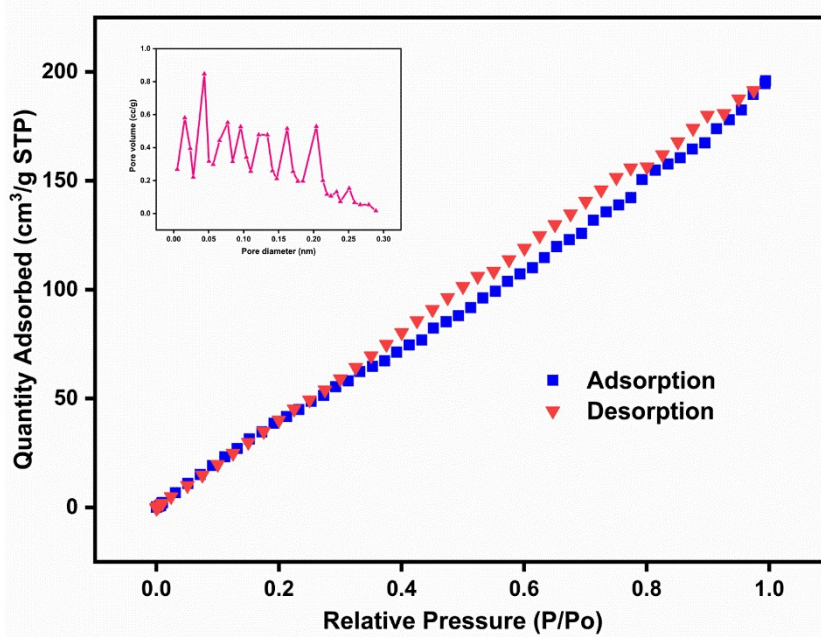


Figure S4 (a) SEM image; (b) XRD spectra of $\text{Ti}_{0.5}\text{Sr}_{0.5}\text{O}_3/\text{NF}$ measured after long-term stability.

Figure
spectra;
survey
(b) Ti 2p,
O 1s, and



S5 XPS
(a) XPS full
spectra of
 $\text{Ti}_{0.5}\text{Sr}_{0.5}\text{O}_3$;
(c) Sr 3d, (d)
(e) C 1s

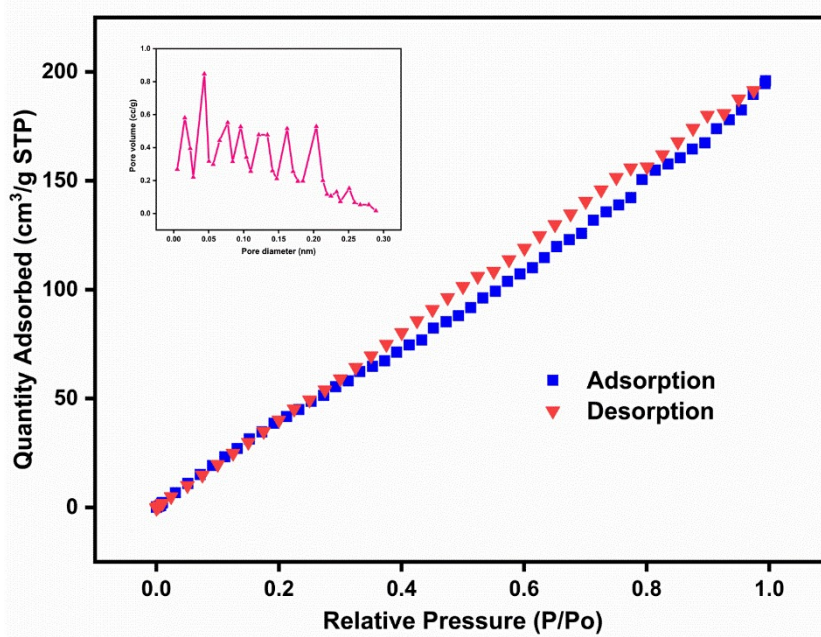


Figure S6 N₂ adsorption–desorption isotherms of Ti_{0.5}Sr_{0.5}O₃ (insets: corresponding pore size distribution).

Table S1: Comparison Table of the performance of Ti_{0.5}Sr_{0.5}O₃ with recently reported catalysts.

S. No	Electrocatalyst	Reaction	Overpotential	Tafel slope	Electrolyte	References
1	CoO _x -TiO ₂	OER	350	65	1.0 M KOH	[1]
2	CoS ₂ -TiO ₂	HER/OER	198/231	61/92	1.0 M KOH	[2]
3	FeTiO ₃	OER	220	155V	1.0 M KOH	[3]
4	Ni/TiO ₂	HER	46	41.8	1.0 M KOH	[4]
5	NiCo ₂ O ₄ /TiO ₂	HER/OER	185/309	120/55	1.0 M KOH	[5]
6	Ti _{0.5} Sr _{0.5} O ₃	HER/OER	101.6/299	179/152.5	1.0 M KOH	[THIS WORK]

1. Yan, Y.; Liu, C.; Jian, H.; Cheng, X.; Hu, T.; Wang, D.; Shang, L.; Chen, G.; Schaaf, P.; Wang, X.; Kan, E.; Zhang, T. Substitutionally Dispersed High-Oxidation CoO_x Clusters in the Lattice of Rutile TiO_2 Triggering Efficient Co□Ti Cooperative Catalytic Centers for Oxygen Evolution Reactions. *Adv Funct Mater* **2021**, *31* (9). <https://doi.org/10.1002/adfm.202009610>.
2. Ganesan, P.; Sivanantham, A.; Shanmugam, S. CoS_2 – TiO_2 Hybrid Nanostructures: Efficient and Durable Bifunctional Electrocatalysts for Alkaline Electrolyte Membrane Water Electrolyzers. *J Mater Chem A Mater* **2018**, *6* (3), 1075–1085. <https://doi.org/10.1039/C7TA09096J>.
3. Kaleeswaran, P.; Praveen Kumar, M.; Mangalaraja, R. V.; Hartley, U. W.; Sasikumar, M.; Venugopalan, R.; Rajesh Kumar, M.; Rajabathar, J. R.; Peera, S. G.; Murugadoss, G. FeTiO_3 Perovskite Nanoparticles for Efficient Electrochemical Water Splitting. *Catalysts* **2021**, *11* (9), 1028. <https://doi.org/10.3390/catal11091028>.
4. Liu, Y.; Wang, X.; Yang, M.; Li, Y.; Xiao, Y.; Zhao, J. Preparation of Ru-Doped TiO_2 Nanotube Arrays through Anodizing TiRu Alloys for Bifunctional HER/OER Electrocatalysts. *Nanoscale* **2023**, *15* (44), 17936–17945. <https://doi.org/10.1039/D3NR03831A>.
5. Vadakkekara, R.; Illathvalappil, R.; Kurungot, S. Layered TiO_2 Nanosheet-Supported NiCo_2O_4 Nanoparticles as Bifunctional Electrocatalyst for Overall Water Splitting. *ChemElectroChem* **2018**, *5* (24), 4000–4007. <https://doi.org/10.1002/celec.201801107>.