

## First-Principles Study of Lead-Free $A_3InX_6$ ( $A = Rb, Cs$ ; $X = Cl, Br, I$ ) Defect -Perovskites: Structural, Electronic, Optical, and Mechanical Properties

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$$n\lambda = 2d\sin\theta \quad (\text{S1})$$

$$d_{hkl} = \frac{a}{\sqrt{h^2 + k^2 + l^2}} \quad (\text{S2})$$

$$\alpha(\omega) = \sqrt{2\omega \left[ \sqrt{\varepsilon_1^2(\omega) + \varepsilon_2^2(\omega)} - \varepsilon_1(\omega) \right]^{\frac{1}{2}}} \quad (\text{S3})$$

$$\sigma_1(\omega) = \frac{2\pi e^2 \hbar}{m^2 \omega \Omega} \sum_{k,v,c} |\langle \psi_{ck} | p | \psi_{vk} \rangle|^2 \delta(E_{ck} - E_{vk} - \hbar\omega) \quad (\text{S4})$$

$$\varepsilon(\omega) = \varepsilon_1(\omega) + jE_2(\omega) \quad (\text{S5})$$

$$\varepsilon_1(\omega) = 1 + \frac{2}{\pi} P \int_0^{\infty} \frac{\omega' \varepsilon_2(\omega')}{\omega'^2 - \omega^2} d\omega' \quad (\text{S6})$$

$$\varepsilon_2(\omega) = \frac{2\pi e^2}{\Omega \varepsilon_0} \sum_{k,v,c} |\langle \Psi_k^c | \hat{u} \cdot \vec{r} | \Psi_k^v \rangle|^2 \delta(E_k^c - E_k^v - E) \quad (\text{S7})$$

$$R(\omega) = \frac{(\eta - 1)^2 + k^2}{(\eta + 1)^2 + k^2} \quad (\text{S8})$$

$$\eta'(\omega) = \eta(\omega + i k(\omega)) \quad (\text{S9})$$

$$n(\omega) = \frac{1}{\sqrt{2}} \left[ (\varepsilon_1(\omega)^2 + \varepsilon_2(\omega)^2)^{\frac{1}{2}} + \varepsilon_1(\omega) \right]^{\frac{1}{2}} \quad (\text{S10})$$

$$k(\omega) = \frac{1}{\sqrt{2}} \left[ (\varepsilon_1(\omega)^2 + \varepsilon_2(\omega)^2)^{\frac{1}{2}} - \varepsilon_1(\omega) \right]^{\frac{1}{2}} \quad (\text{S11})$$

**Table S1.** Optimized structural parameters and atomic fractional coordinates of  $\text{Rb}_3\text{InX}_6$  obtained from first-principles calculations.

Atom	Wyckoff	x	y	z
In	4a	0.0000	0.0000	0.0000
Rb	4b	0.5000	0.0000	0.0000
Rb	8c	0.2500	0.2500	0.7500

X (X= Cl, Br, I)	24e	0.219174	0.0000	0.0000
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**Table S2.** Optimized structural parameters and atomic fractional coordinates of  $\text{Cs}_3\text{InX}_6$  obtained from first-principles calculations.

Atom	Wyckoff	x	y	z
In	4a	0.0000	0.0000	0.0000
Cs	4b	0.5000	0.0000	0.0000
Cs	8c	0.2500	0.2500	0.7500
X (X= Cl, Br, I)	24e	0.219174	0.0000	0.0000

**Table S3.** The energy band gap, lattice constants, unit cell volume & formation enthalpy of  $\text{A}_3\text{InX}_6$  (A=Rb, Cs, and X=Cl, Br, I) Perovskites

Ref.	Compounds	Band Gaps ( $E_g$ )	Lattice Constant, a( $\text{\AA}$ )	Functionals	
This work	$\text{Rb}_3\text{InCl}_6$	3.695	-	GGA-PBESOL	
	$\text{Rb}_3\text{InBr}_6$	2.595	-		
	$\text{Rb}_3\text{InI}_6$	1.429	-		
	$\text{Cs}_3\text{InCl}_6$	3.801	-		
	$\text{Cs}_3\text{InBr}_6$	2.755	-		
	$\text{Cs}_3\text{InI}_6$	1.590	-		
	This work	$\text{Rb}_3\text{InCl}_6$	5.004	-	Hybrid (HSE06)
		$\text{Rb}_3\text{InBr}_6$	3.892	-	
		$\text{Rb}_3\text{InI}_6$	2.550	-	
		$\text{Cs}_3\text{InCl}_6$	5.116	-	
		$\text{Cs}_3\text{InBr}_6$	4.038	-	
		$\text{Cs}_3\text{InI}_6$	2.556	-	
[1]	$\text{Na}_3\text{GaBr}_6$	1.991	7.747	GGA-PBE	
	$\text{K}_3\text{GaBr}_6$	1.445	8.085		
	$\text{Na}_3\text{GaBr}_6$	2.602	-	mGGA-RSCAN	
	$\text{K}_3\text{GaBr}_6$	1.985	-		
	$\text{Na}_3\text{GaBr}_6$	3.096	-	Hybrid (HSE06)	
	$\text{K}_3\text{GaBr}_6$	2.567	-		
[2]	$\text{Cs}_3\text{InCl}_6$	1.45	-		

	$\text{Cs}_3\text{InBr}_6$	1.55	-	
	$\text{Cs}_3\text{InI}_6$	1.60	-	
	$\text{Cs}_3\text{TlCl}_6$	0.93	-	
	$\text{Cs}_3\text{TlBr}_6$	1.01	-	
	$\text{Cs}_3\text{InI}_6$	0.99	-	
[3]	$\text{Rb}_3\text{InCl}_6$	5.22	-	Experimental

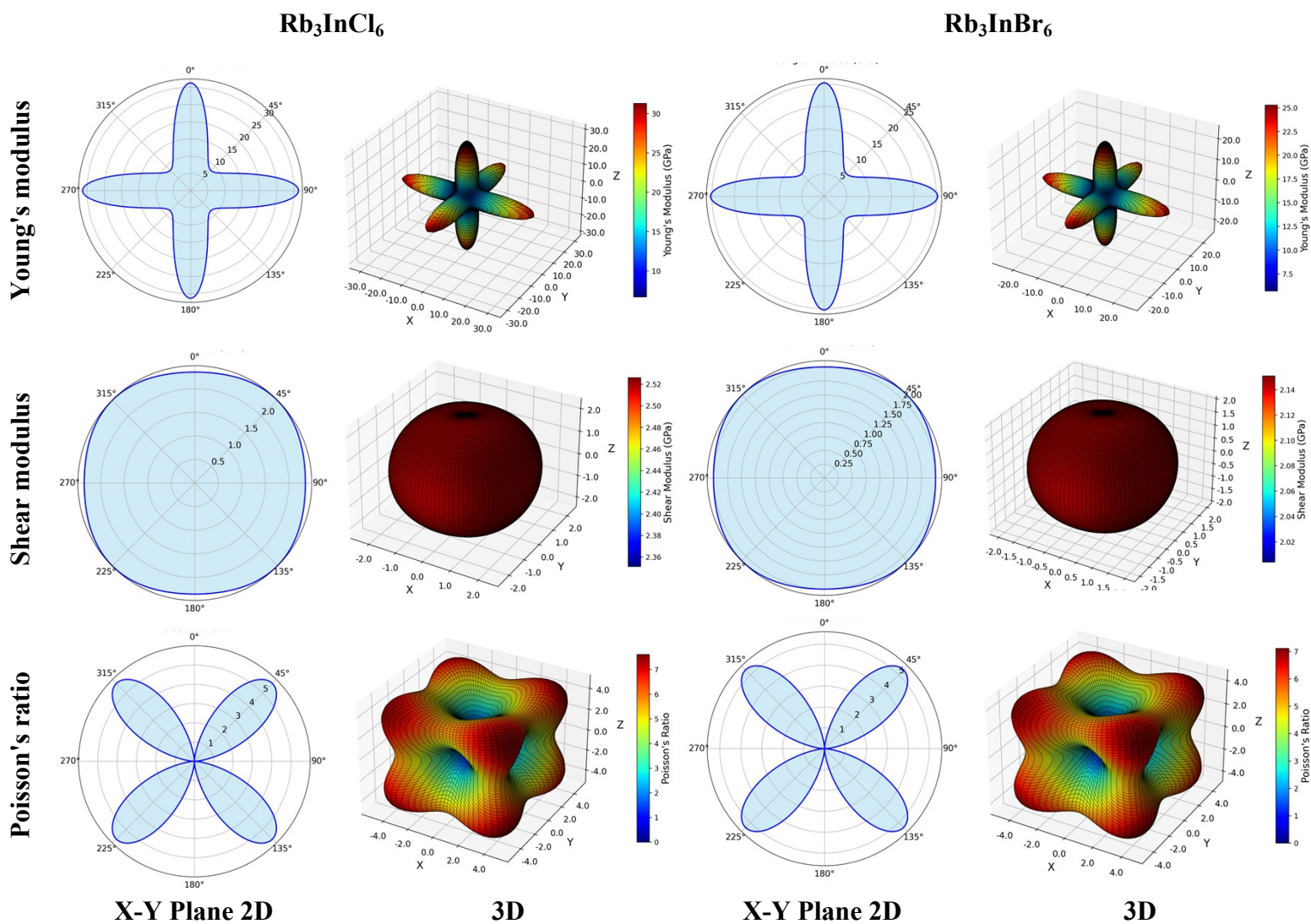
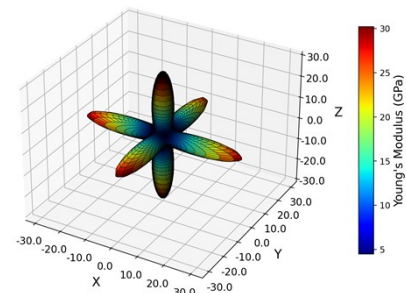
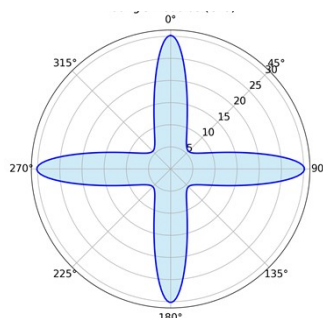
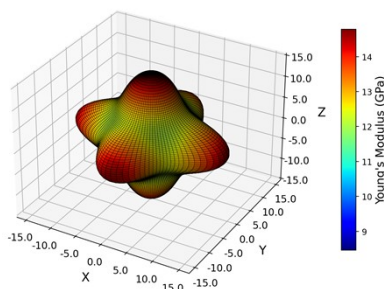
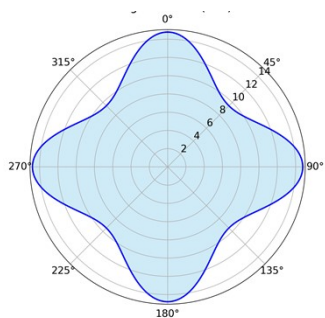


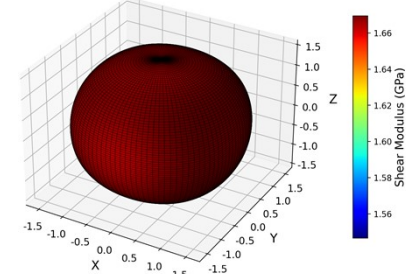
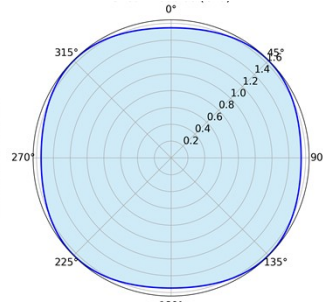
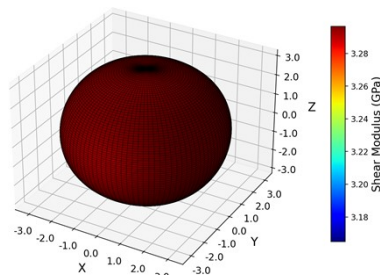
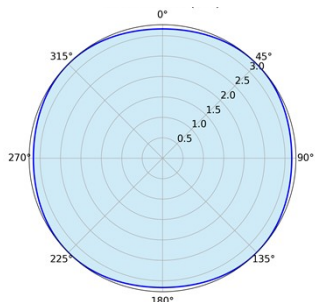
Figure S1. Graphical assessment of  $\text{Rb}_3\text{InCl}_6$  and  $\text{Rb}_3\text{InBr}_6$  Perovskites anisotropy showing 2D and 3D figures.



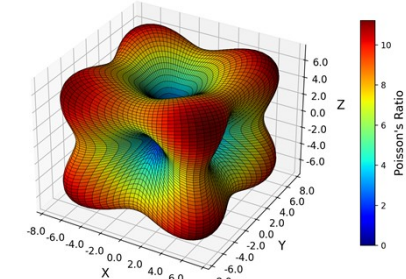
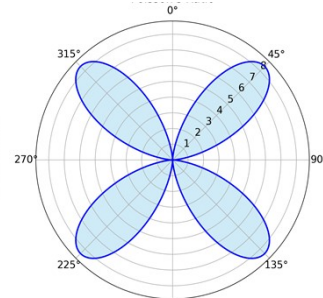
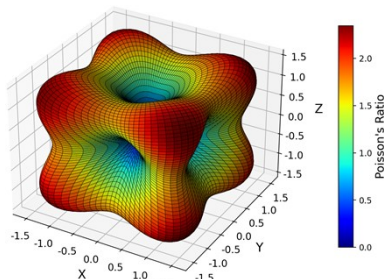
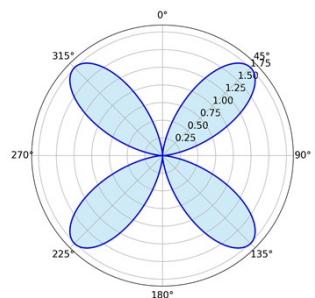
Young's modulus



Shear modulus



Poisson's ratio



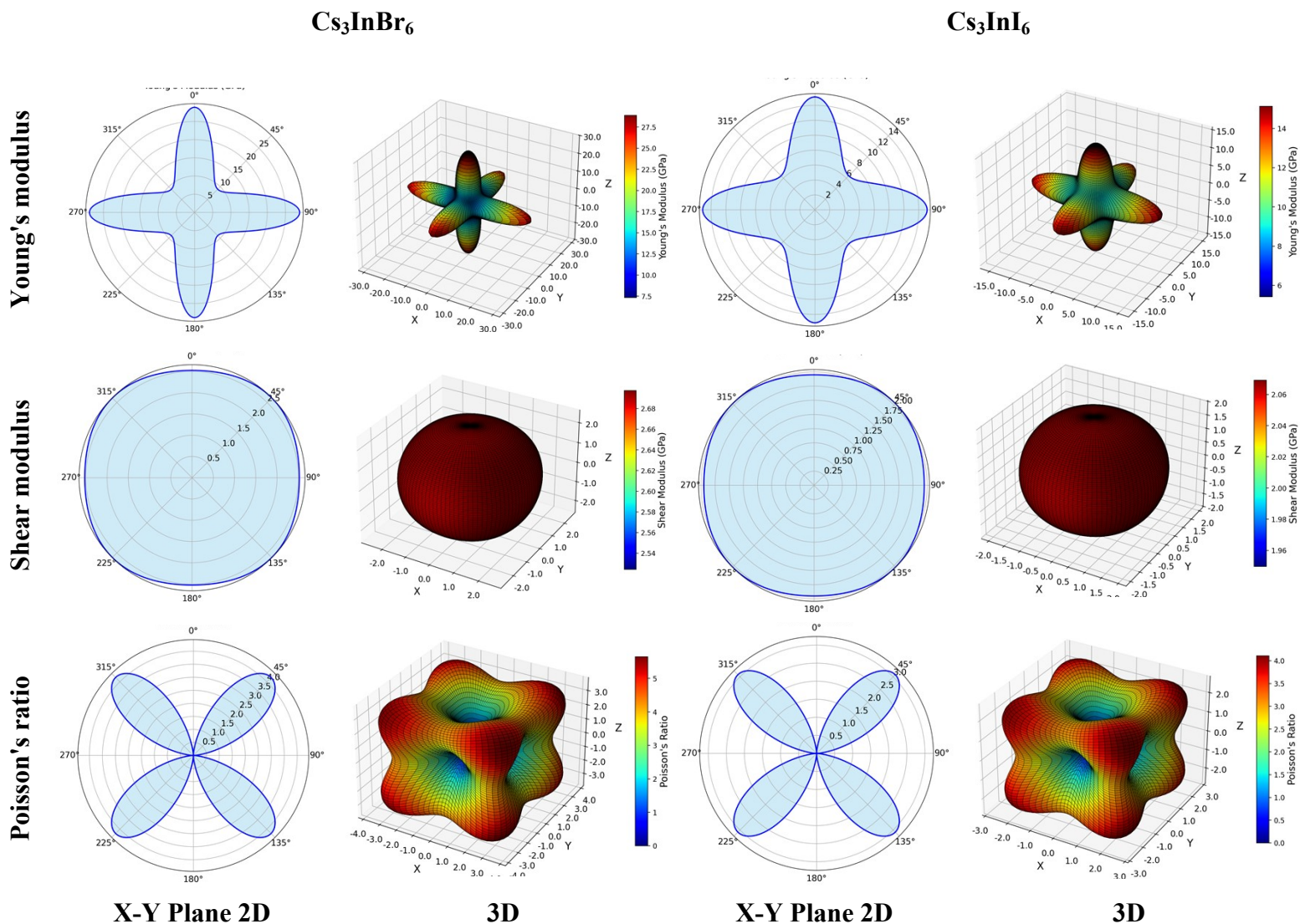
X-Y Plane 2D

3D

X-Y Plane 2D

3D

Figure S2. Graphical assessment of  $\text{Rb}_3\text{InI}_6$  and  $\text{Cs}_3\text{InCl}_6$  Perovskites anisotropy showing 2D and 3D figures.



**Figure S3.** Graphical assessment of  $\text{Cs}_3\text{InBr}_6$  and  $\text{Cs}_3\text{InI}_6$  Perovskites anisotropy showing 2D and 3D figures.

**Table S4.** Mulliken and Hirshfeld charge analysis of different atoms of  $\text{A}_3\text{InX}_6$  (A=Rb, Cs, and X=Cl, Br, I) Perovskites

Compound	Charge	Species	Mulliken atomic populations	Mulliken	Hirshfeld
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	<b>spilling</b>		<b>s</b>	<b>P</b>	<b>d</b>	<b>f</b>	<b>Total</b>	<b>change</b>	<b>change</b>
Rb <sub>3</sub> InCl <sub>6</sub>	0.11%	Rb (1)	2.11	6.18	0.45	0.00	8.73	0.27	0.17
		Rb (2)	1.99	5.98	0.11	0.00	8.08	0.92	0.33
		Rb (3)	1.99	5.98	0.11	0.00	8.08	0.92	0.33
		In	1.11	1.16	10.00	0.00	12.26	0.74	0.41
		Cl	1.94	5.53	0.00	0.00	7.47	-0.47	-0.23
Rb <sub>3</sub> InBr <sub>6</sub>	0.09%	Rb (1)	2.13	6.39	0.53	0.00	9.05	-0.05	0.16
		Rb (2)	2.03	6.15	0.20	0.00	8.38	0.62	0.32
		Rb (3)	2.03	6.15	0.20	0.00	8.38	0.62	0.32
		In	1.41	1.39	10.00	0.00	12.81	0.19	0.35
		Br	1.76	5.47	0.00	0.00	7.23	-0.23	-0.21
Rb <sub>3</sub> InI <sub>6</sub>	0.11%	Rb (1)	2.21	6.36	0.56	0.00	9.13	-0.13	0.14
		Rb (2)	2.08	6.14	0.18	0.00	8.40	0.60	0.32
		Rb (3)	2.08	6.14	0.18	0.00	8.40	0.60	0.32
		In	1.12	1.53	10.00	0.00	12.65	0.35	0.27
		I	1.83	5.41	0.00	0.00	7.24	-0.24	-0.20
Cs <sub>3</sub> InCl <sub>6</sub>	0.12%	Cs (1)	2.11	6.18	0.40	0.00	8.69	0.31	0.24
		Cs (2)	2.00	5.93	0.14	0.00	8.07	0.93	0.38
		Cs (3)	2.00	5.93	0.14	0.00	8.07	0.93	0.38
		In	1.13	1.14	10.00	0.00	12.26	0.74	0.42
		Cl	1.95	5.54	0.00	0.00	7.49	-0.49	-0.24
Cs <sub>3</sub> InBr <sub>6</sub>	0.10%	Cs (1)	2.14	6.39	0.47	0.00	9.00	-0.00	0.21
		Cs (2)	2.05	6.13	0.19	0.00	8.37	0.63	0.35
		Cs (3)	2.05	6.13	0.19	0.00	8.37	0.63	0.35
		In	1.36	1.38	10.00	0.00	12.74	0.26	0.35
		Br	1.78	5.48	0.00	0.00	7.25	-0.25	-0.21
Cs <sub>3</sub> InI <sub>6</sub>	0.10%	Cs (1)	2.19	6.33	0.51	0.00	9.03	-0.03	0.18
		Cs (2)	2.09	6.15	0.18	0.00	8.42	0.58	0.35
		Cs (3)	2.09	6.15	0.18	0.00	8.42	0.58	0.35
		In	1.14	1.51	10.00	0.00	12.65	0.35	0.28
		I	1.83	5.41	0.00	0.00	7.25	-0.25	-0.19

## References

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