

## **Self-Powered Piezoelectric Microfluidic Flow Sensor for Low-Flow Monitoring of Metal-Ion Solutions**

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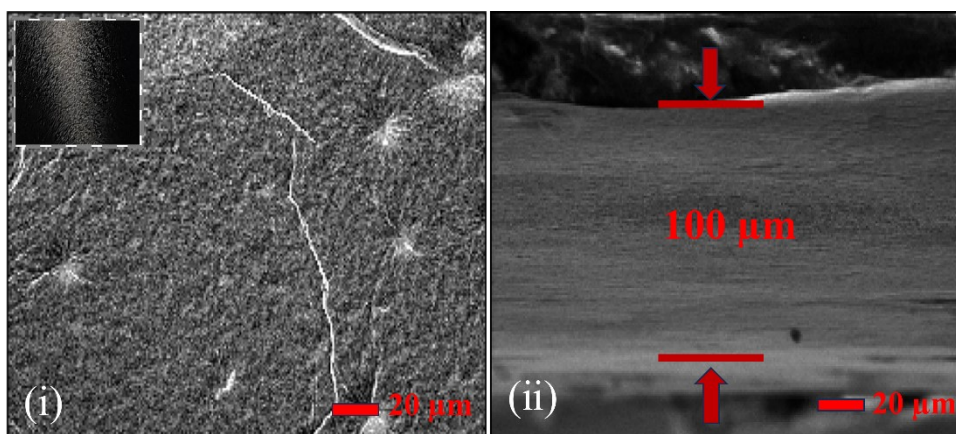
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**Figure S1.** SEM images of the commercial piezoelectric film: (i) plan-view SEM image showing a relatively smooth and uniform surface; (ii) cross-sectional SEM image indicating a film thickness of approximately 100  $\mu\text{m}$ .

**Table S1.** Gravimetric calibration results of the syringe pump at set flow rates of 3, 10, 50, 100, and 200  $\mu\text{L}\cdot\text{min}^{-1}$ .

Set flow rate ( $\mu\text{L}\cdot\text{min}^{-1}$ )	Collection time (min)	Measured flow rate ( $\mu\text{L}\cdot\text{min}^{-1}$ , mean $\pm$ SD, n = 5)	Relative error (%)	RSD (%)
3	60	$3.009 \pm 0.021$	0.30	0.69
10	30	$10.032 \pm 0.031$	0.32	0.31
50	10	$50.050 \pm 0.274$	0.10	0.55
100	10	$100.088 \pm 0.534$	0.09	0.53
200	5	$200.124 \pm 0.192$	0.06	0.19

Note: The actual flow rate was calculated from the collected liquid mass and collection time. Relative error was calculated with respect to the programmed flow rate.

**Table S2.** Comparison of Representative Microfluidic Flow Sensors

Author	Sensing Method	Sensitivity	Limit of detection	Range	External power	Applicable fluid types
Hongdan Wan et al. (2025) <sup>1</sup>	Optical sensing	~0.27 pm/( $\mu\text{L}\cdot\text{min}^{-1}$ )	~1.43 $\mu\text{L}\cdot\text{min}^{-1}$	-	Yes	-
Adam Hawke et al. (2023) <sup>2</sup>	Pressure sensing	-	-	10–500 $\mu\text{L}\cdot\text{min}^{-1}$	Yes	DI water; water-glycerol solutions; blood
Weinan Liu et al. (2025) <sup>3</sup>	Optical sensing	36.5 nm/( $\mu\text{L}\cdot\text{s}^{-1}$ )	~212nL $\cdot\text{s}^{-1}$	0–1.6 $\mu\text{L}\cdot\text{s}^{-1}$	Yes	Water
R. Vilares et al. (2010) <sup>4</sup>	Thermal flow sensing	485 $\mu\text{V}/(\mu\text{L}\cdot\text{min}^{-1})$	40nL $\cdot\text{min}^{-1}$	0–25 $\mu\text{L}\cdot\text{min}^{-1}$	Yes	DI water
Zijun Zhang et al. (2024) <sup>5</sup>	Pressure sensing	4.33–6.67 Pa/(mL $\cdot\text{h}^{-1}$ )	2.9 $\mu\text{L}\cdot\text{h}^{-1}$	11.8–118.0mL $\cdot\text{min}^{-1}$	Yes	DI water
John Collins and Abraham P. Lee (2004) <sup>6</sup>	Electrical admittance measurement	$5.2 \times 10^{-4}$ mA/(mL $\cdot\text{min}^{-1}$ )	0.05mL $\cdot\text{min}^{-1}$	>0.05mL $\cdot\text{min}^{-1}$	Yes	NaOH; CaCl <sub>2</sub> ; KCl; KOH; D-PBS; D-MEM
Martin Seidl (2024) <sup>7</sup>	Mass flow sensing	-	10 $\mu\text{L}\cdot\text{min}^{-1}$	>20mL $\cdot\text{min}^{-1}$	Yes	air
Nadine Noeth et al. (2011) <sup>8</sup>	Optical sensing	-	3nL $\cdot\text{min}^{-1}$	-	Yes	water
Tiange Wu et al. (2020) <sup>9</sup>	Optical sensing	$4.65 \times 10^5$ mV/(m $\cdot\text{s}^{-1}$ )	0.7 mm $\cdot\text{s}^{-1}$	0.6–14 kPa	Yes	Blood; DI water
Wenxue Li et al. (2025) <sup>10</sup>	Optical sensing	22.51 nm/(mm $\cdot\text{s}^{-1}$ )	-	0.17–1.17mm $\cdot\text{s}^{-1}$	Yes	DI water
Alex Baldwin et al. (2016) <sup>11</sup>	Electrochemical sensing	-	-	–400 to 400 $\mu\text{L}\cdot\text{min}^{-1}$	Yes	physiological ionic fluids
Harsh Deswal et al. (2024) <sup>12</sup>	Pressure sensing	0.016 $\Omega/(\mu\text{L}\cdot\text{min}^{-1})$	5 $\mu\text{L}\cdot\text{min}^{-1}$	0–200 $\mu\text{L}\cdot\text{min}^{-1}$	Yes	NaCl (0.1–0.6 M)
Trevor Q. Hudson et al. (2021) <sup>13</sup>	Electrochemical sensing	-	-	43–200 $\mu\text{L}\cdot\text{min}^{-1}$	Yes	physiological ionic fluids
This work	Pressure sensing	0.79 mV/( $\mu\text{L}\cdot\text{min}^{-1}$ )	3 $\mu\text{L}\cdot\text{min}^{-1}$	3–203 $\mu\text{L}\cdot\text{min}^{-1}$	NO (self-powered)	DI water; electrolyte solutions

**Section S3. Quantification of stability metrics**

To quantitatively evaluate the operational stability of the sensor, the steady-state output voltage for each condition was defined as the arithmetic mean of the selected plateau window<sup>14</sup>:

$$V_{ss} = \frac{1}{n} \sum_{i=1}^n V_i \quad (S1)$$

where  $V_i$  is the instantaneous voltage and  $n$  is the number of data points in the steady-

state window.

For electrolyte-dependent measurements, the normalized deviation of the steady-state output under the  $J$ -th condition relative to the reference condition<sup>14</sup> was calculated as

$$\delta(\%) = \frac{\bar{V}_{ss,j} - \bar{V}_{ss,ref}}{\bar{V}_{ss,ref}} \times 100\%, \quad (S2)$$

where  $\bar{V}_{ss,j}$  is the mean steady-state output voltage under the  $J$ -th electrolyte condition, and  $\bar{V}_{ss,ref}$  is the mean steady-state output voltage under the reference condition (1 mM NaCl at the same flow rate). The maximum electrolyte-dependent deviation reported in the main text corresponds to the largest value of  $\delta(\%)$  among all tested electrolyte conditions.

For the long-term continuous-flow test, the relative drift magnitude between the first and last 5 min steady-state windows<sup>14,16</sup> was calculated as

$$D(\%) = \left| \frac{\bar{V}_{end} - \bar{V}_{start}}{\bar{V}_{start}} \right| \times 100\% \quad (S3)$$

where  $\bar{V}_{start}$  and  $\bar{V}_{end}$  are the mean output voltages in the first and last 5 min steady-state windows, respectively, after excluding the startup and shutdown transients. the normalized temperature variation relative to the 20 °C condition<sup>15,16</sup> was calculated as

$$\eta_T(\%) = \left| \frac{\bar{V}_{ss,T} - \bar{V}_{ss,20^\circ C}}{\bar{V}_{ss,20^\circ C}} \right| \times 100\% \quad (S4)$$

where  $\bar{V}_{ss,T}$  is the mean steady-state output voltage at temperature  $T$ , and  $\bar{V}_{ss,20^\circ C}$  is the corresponding mean steady-state output voltage at the reference temperature of 20 °C. In this work, because the stability datasets correspond to continuous time traces under fixed conditions rather than multiple independent replicate runs, temporal SD/RSD and condition-to-condition deviations were used as quantitative descriptors of signal stability.

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