

Supporting information

A Robust Co/SiO₂ Catalyst for Sustainable Sorbitol Production: From Batch to Continuous Flow

Xincheng Wang,* Bangxian Long, Yuxin Liang, Yuze Peng, Haofei Yang, Lei Wang,
Miaomiao Chen, Ruomeng Li, Songqi Tai

(Beijing Key Laboratory of Enze Biomass Fine Chemicals, College of New Materials
and Chemical Engineering, Beijing Institute of Petrochemical Technology, Beijing
102617, PR China)

Contents

Fig. S1 N₂ adsorption-desorption isotherm and pore size distribution of SiO₂ catalyst.

Fig. S2 Comparison of product yield over 10% Ni supported on different supports.

Fig. S3 Comparison of Catalytic Performance of 20% Ni Supported on two different supports.

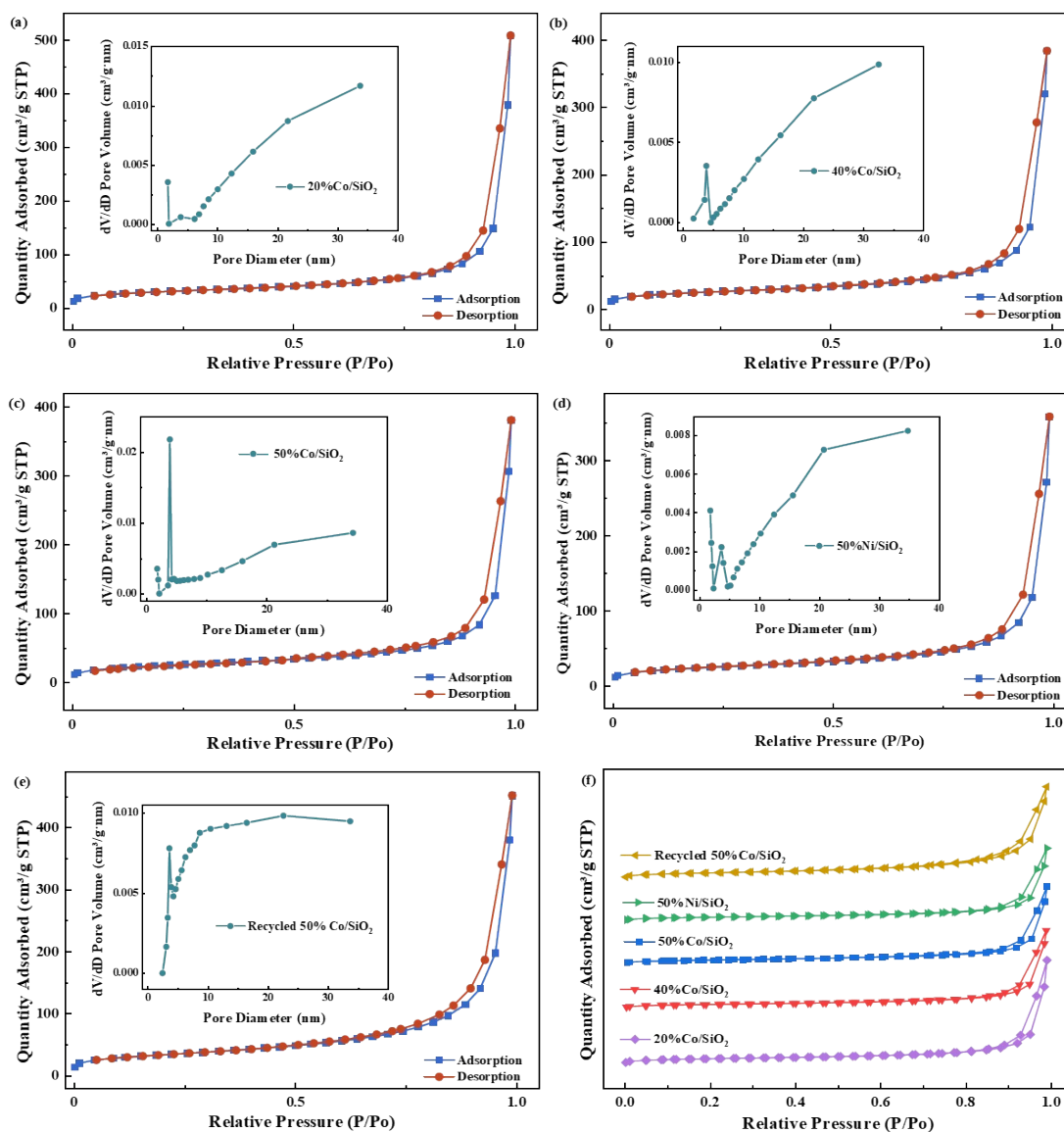


Fig. S1 N₂ adsorption-desorption isotherm and pore size distribution of SiO₂ catalyst: (1) 20% Co/SiO₂, (2) 40% Co/SiO₂, (3) 50% Co/SiO₂, (4) 50% Ni/SiO₂, (5) Recycled 50% Co/SiO₂, (6) SiO₂ catalysts with different loadings.

Effect of the Support on Reaction Activity

Given the widespread application of nickel in conventional hydrogenation processes, metallic Ni was employed to screen suitable supports. Catalysts with a uniform Ni loading of 10% were prepared via the incipient wetness impregnation method. The catalytic performance was evaluated in a 10 wt% glucose solution after 1 hour of reaction, as shown in Fig. S2.

When Al₂O₃, SiO₂, or β zeolite were used as supports, the catalysts exhibited superior performance compared to others, with no significant by-products detected. This indicates that these three supports effectively disperse the Ni active species,

forming selective active sites that promote the directed hydrogenation of glucose to sorbitol while suppressing side reactions.

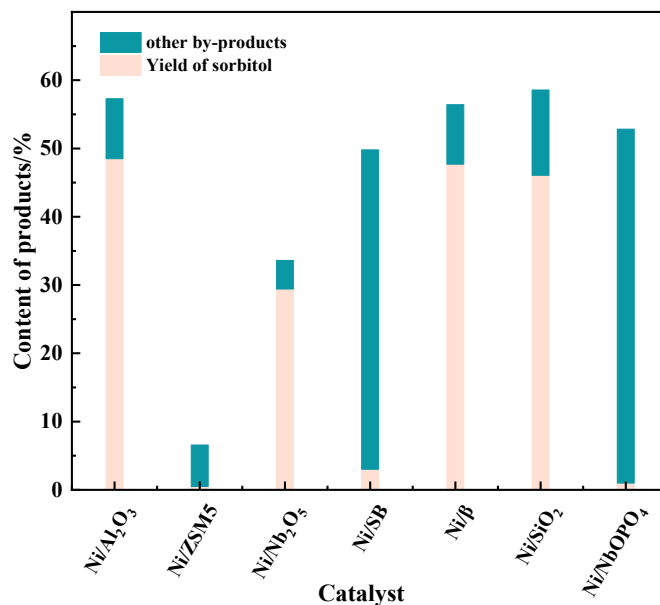


Fig. S2 Comparison of product yield over 10% Ni supported on different supports. Reaction conditions: 2 g catalyst, 5 MPa H₂, 145 °C, 10 wt% glucose solution.

Since Al₂O₃, SiO₂, and β zeolite exhibited comparable catalytic performance, the Ni loading was increased to 20% to investigate the effect of higher loading on these three catalysts, thereby screening for a more efficient catalyst support.

Fig. S3 Comparison of Catalytic Performance of 20% Ni Supported on two different supports: (a) glucose(10 wt%) conversion, (b) sorbitol yield as a function of time. Reaction conditions: 2 g catalyst, 5 MPa H₂, 145 °C.

After increasing the Ni content, the catalytic performance of all catalysts improved (Fig. S3). Among the three supports, the catalyst with SiO₂ demonstrated significantly superior performance compared to those with Al₂O₃ and β zeolite, achieving a glucose conversion of 80.2% and a sorbitol yield of 77.3%. This enhanced activity may be attributed to the high specific surface area and suitable pore structure of the SiO₂ support, which can provide more active sites and facilitate reactant diffusion. The advantage of SiO₂ became more pronounced at the elevated loading of 20%, as it enabled better dispersion of metallic Ni, thereby

increasing the number of accessible active sites.