

Supplementary Information

Hierarchical Polar–Nonpolar Phase Architecture Enabling Excellent Lead-free Capacitive Energy Storage

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Materials and Methods

Sample preparation

$0.5\text{BaTiO}_3-(0.5-x)\text{Bi}(\text{Mg}_{0.5}\text{Ti}_{0.5})\text{O}_3-x\text{CdZrO}_3$ [0.5BT-(0.5-x)BMT-xCZ] ceramics, with $x = 0, 0.1, 0.15, 0.2,$ and 0.25 , were synthesized using the conventional solid-state reaction method. The starting materials included Bi_2O_3 (99.9%), TiO_2 (98%), CdO (99.9%), ZrO_2 (99.99%), MgO (99.8%), and BaCO_3 (99%) powders. The dried powders were weighed according to the stoichiometric ratios and ball-milled in anhydrous ethanol for 10 hours. The resulting mixture was dried and calcined in air at $780\text{ }^\circ\text{C}$ for 2 hours. Subsequently, the calcined powder was mixed with polyvinyl alcohol (PVA) binder, thoroughly ground, and then pressed into pellets 10 mm in diameter and 1 mm thick. To minimize elemental volatilization during sintering, the pellets were covered with powder and sintered at $1050\text{--}1150\text{ }^\circ\text{C}$ for 2 hours. To identify the optimal sintering temperature, we performed sintering experiments on the selected compositions at various temperatures. The corresponding microstructure and elemental analyses are presented in Figures S12 and S13. Within the temperature range of $1000\text{--}1100\text{ }^\circ\text{C}$, samples sintered at $1025\text{ }^\circ\text{C}$ and $1050\text{ }^\circ\text{C}$ exhibit uniform contrast and well-defined phase separation, whereas the sample sintered at $1000\text{ }^\circ\text{C}$ shows inferior ceramic quality. Samples sintered between $1075\text{ }^\circ\text{C}$ and $1100\text{ }^\circ\text{C}$ display rough surfaces and noticeable precipitates, indicating that excessively high sintering temperatures adversely affect the grain-level microstructure. Meanwhile, grain size increases gradually with sintering temperature; however, no significant differences are observed between $1025\text{ }^\circ\text{C}$ and $1050\text{ }^\circ\text{C}$. Energy-dispersive X-ray spectroscopy (EDS) analysis confirms that the phase separation pattern remains consistent across all temperatures, suggesting that the two-phase structure is governed by the reaction process rather than the sintering conditions. Considering grain size, sintering quality, and elemental distribution, the temperature range of $1025\text{--}1050\text{ }^\circ\text{C}$ provides the optimal microstructural characteristics, with the best overall performance at $1050\text{ }^\circ\text{C}$. Therefore, $1050\text{ }^\circ\text{C}$ was selected as the optimal sintering temperature for these ceramics. The relative densities of the ceramics, measured by the Archimedes method, exceeded 95% in all cases.

Characterization

The phase structure was characterized by X-ray diffraction (XRD) using a D8ADVANCE-A25 diffractometer (Bruker, Germany). Microstructure and elemental distribution analyses were conducted with a field-emission scanning electron microscope (FE-SEM; Zeiss Gemini SEM 300, Germany) equipped with energy-dispersive X-ray spectroscopy (EDS; Oxford Instruments X-MaxN SN 78861). Prior to SEM observation, all samples were polished and thermally etched at 1050–1150 °C for 0.5 hours. Selected area electron diffraction (SAED) patterns and high-resolution transmission electron microscopy (HRTEM) images were obtained using an aberration-corrected scanning transmission electron microscope (STEM; FEI Titan Cubed Themis G2 300, Thermo Fisher Scientific, USA) operating at 300 kV. Atomic-scale high-angle annular dark-field (HAADF) STEM images were acquired with a probe- and image-aberration-corrected microscope operated at 200 kV. All STEM images were Fourier-filtered using a lattice mask to reduce noise. The precise positions of the A- and B-site atomic columns were determined by fitting two-dimensional Gaussian functions, and the angles and magnitudes of polar displacements were calculated using a MATLAB script.

Polarization-electric field (P - E) loops under varying electric fields, frequencies, temperatures, and cycle numbers were measured using a ferroelectric test system (FETS-2000, Wuhan Yanhe Technology Co., Ltd., China). Breakdown strength was evaluated with a dielectric breakdown test system at a ramp rate of 500 V/s (PolyK Technologies, USA). Dielectric properties, including dielectric constant and loss, were characterized as functions of temperature (21–500 °C) and frequency (1 kHz–1 MHz) using the FETS-2000 system. Charge–discharge measurements were conducted on a charge–discharge test system (CCDM-1000, Wuhan Yanhe Technology Co., Ltd., China). Alternating current (AC) impedance spectroscopy was performed with an electrochemical workstation (CHI700e, CH Instruments, USA) over the temperature range of 410–570 °C. Ultraviolet–visible (UV-Vis) absorption spectra were recorded using a UV-Vis spectrometer (UV-9000, METASH, China) over the wavelength range of 200–800 nm, with BaSO₄ employed as the reference material.

Phase-field simulations

Phase-field simulations were carried out to investigate the polarization evolution in 0.5BT-(0.5-x)BMT-xCZ bulk ceramic. The spontaneous polarization P_i ($i = 1-3$) are taken as the order parameters, whose temporal evolution is described by the time-dependent Ginzburg-Landau equation:

$$\frac{\partial P_i}{\partial t} = -L \frac{\delta F}{\delta P} \quad (1)$$

Here, t represents the evolution time, and L is the kinetic coefficient associated with domain-wall mobility. The total free energy F of the bulk ceramic is expressed as the volume integral of several energy contributions, including bulk, gradient, elastic, and electrostatic terms¹:

$$F = \int (f_{bulk}(P_i) + f_{grad}(P_{i,j}) + f_{elas}(P_i, \varepsilon_{ij}) + f_{elec}(P_i, E_i)) dV \quad (2)$$

The bulk energy density is expressed as:

$$f_{bulk} = \alpha_{ij} P_i P_j + \alpha_{ijkl} P_i P_j P_k P_l + \alpha_{ijklmn} P_i P_j P_k P_l P_m P_n + \alpha_{ijklmnop} P_i P_j P_k P_l P_m P_n P_o P_p \quad (3)$$

where α_{ij} , α_{ijkl} , α_{ijklmn} , and $\alpha_{ijklmnop}$ denote the Landau coefficients. These coefficients are determined by linearly interpolating the corresponding parameters of constituent compounds BTO, BMT, and CZ.^{2, 3} To incorporate concentration inhomogeneities, local fluctuations of the disordered concentration with magnitude of 0.3 are introduced.

The gradient energy density is given by:

$$f_{grad} = \frac{1}{2} g_{ijkl} P_{i,j} P_{k,l} \quad (4)$$

where g_{ijkl} denotes the gradient energy coefficient.

The elastic energy density is written as:

$$f_{elas} = \frac{1}{2} c_{ijkl} (\varepsilon_{ij} - \varepsilon_{ij}^0) (\varepsilon_{kl} - \varepsilon_{kl}^0) \quad (5)$$

where c_{ijkl} is the elastic stiffness tensor, ε_{ij} is the total strain. The eigenstrain ε_{ij}^0 is related to ferroelectric polarization P_i through $\varepsilon_{ij}^0 = Q_{ijkl} P_k P_l$, where Q_{ijkl} is the electrostrictive coefficient.

The electrostatic energy density is described as:

$$f_{\text{elec}} = -\frac{1}{2}\kappa_0\kappa_b E_i E_j - E_i P_i \quad (6)$$

where κ_0 is the permittivity of vacuum, κ_b is background relative dielectric constant, and E_i is the total electric field. The polarization–electric field (P – E) hysteresis loop was obtained by applying a cyclic electric field along the out-of-plane direction, ranging from 0 to ± 4 MV/cm with a field increment of 0.4 MV/cm.

The discrete quasi-two-dimensional mesh of $100\Delta x \times 2\Delta y \times 100\Delta z$ with a real grid space of $\Delta x = \Delta y = \Delta z = 1.0$ nm was used in the simulation. Periodic boundary conditions were adopted to model the bulk ceramic. The numerical implementation of the phase-field model follows established methods in the literature,⁴ and the normalized timestep was set to 0.004.

Weibull distribution method

This method is consistent with the standard widely adopted in this field.^{5, 6} To determine the statistical E_b value of each composition, 10–12 ceramic samples were tested, with a thickness range of 50–90 μm . The Weibull distribution of the ceramics is calculated using the following formula

$$P_i = i/(n+1) \quad (7)$$

$$X_i = \ln(E_i) \quad (8)$$

$$Y_i = \ln(-\ln(1-P_i)) \quad (9)$$

$$P(E_i) = 1 - \exp((-E_i/E_b)^\beta) \quad (10)$$

where E_i represents the single breakdown value, i indicates the i^{th} measurement number, and n is the total number of tests for each sample (in this study, $n = 10$), E_i is the measured breakdown field, $P(E_i)$ presents the cumulative probability of electric breakdown at E_i , E_b is the statistical breakdown strength, which corresponds to the $P(E_i)$ is equal to 63.2%. The intersection point of the fitting line with $Y_i = 0$ is the theoretical value of E_b . The slope value of the fitting line represents the Voigt modulus β of the ceramic. β represents the dispersion degree of the E_b data, wherein a higher β value corresponds to a more concentrated distribution and a more uniform breakdown field strength.

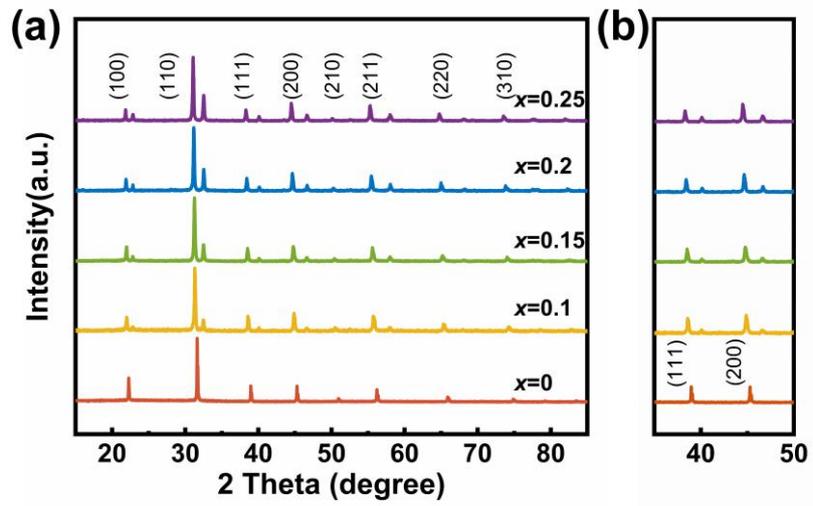


Fig. S1 (a) XRD patterns of $0.5\text{BT}-(0.5-x)\text{BMT}-x\text{CZ}$ ceramics and (b) an enlarged (200) profile of the $x = 0.2$ ceramic at room temperature.

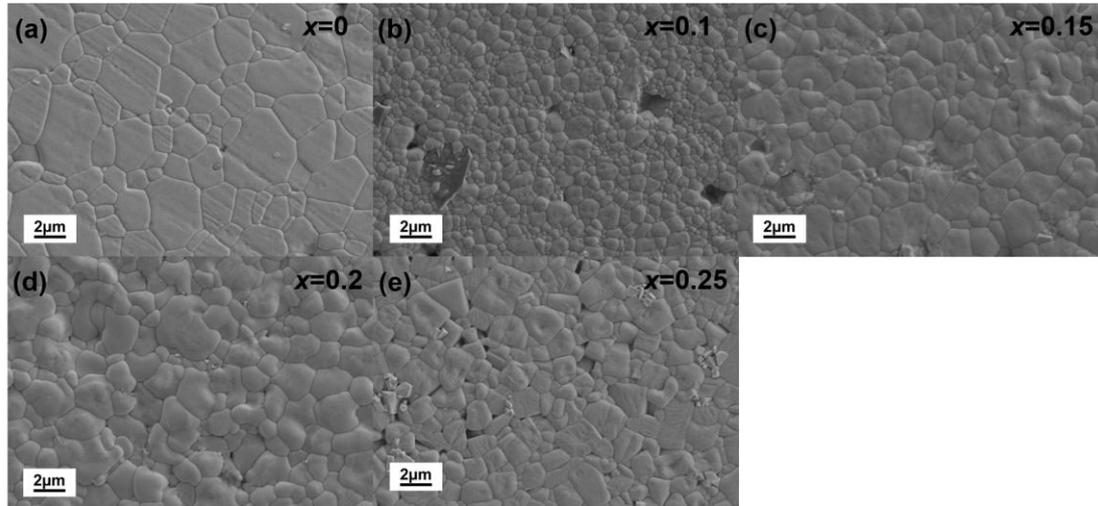


Fig. S2 Representative SEM surface morphologies of $0.5\text{BT}-(0.5-x)\text{BMT}-x\text{CZ}$ ceramics are shown for different compositions: (a) $x = 0$, (b) $x = 0.1$, (c) $x = 0.15$, (d) $x = 0.2$, and (e) $x = 0.25$.

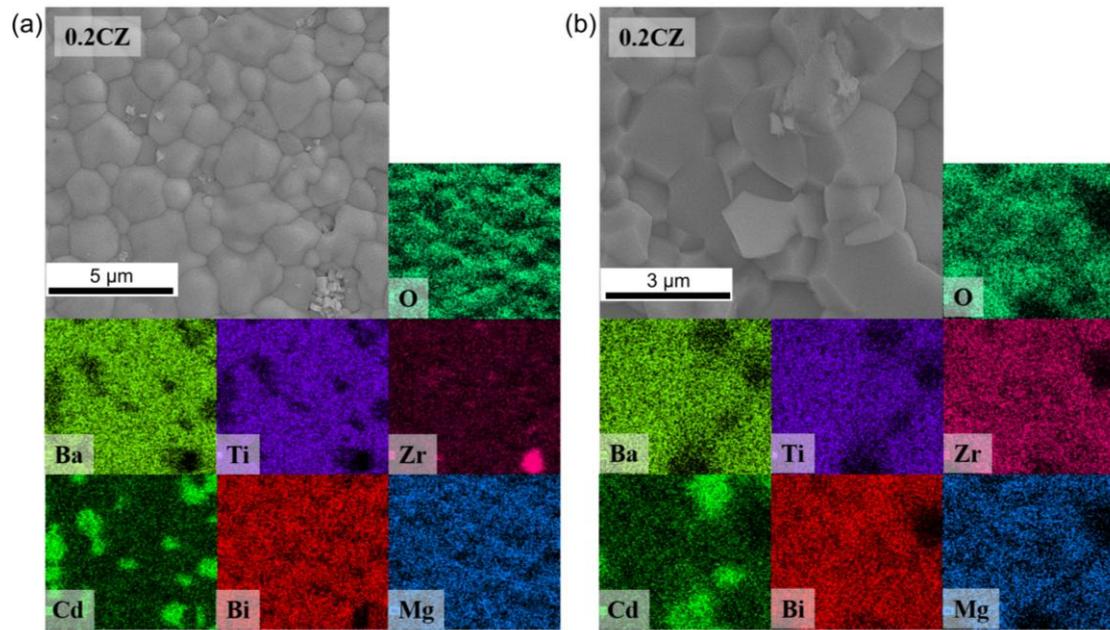


Fig. S3 SEM images of polished and thermally etched surfaces, accompanied by elemental mapping of dual-phase ferroelectric ceramics, are presented as follows: (a) surface morphology of the ceramic with $x = 0.2$, and (b) cross-sectional images of the ceramic with $x = 0.2$.

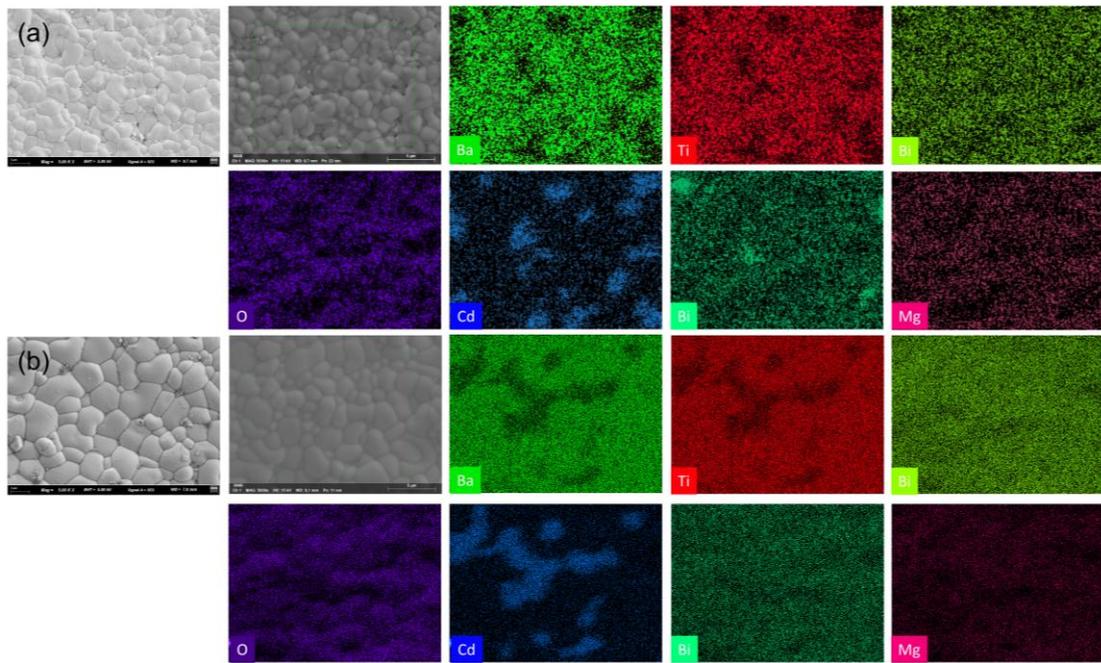


Fig. S4 SEM surface morphology and elemental distribution of the $x = 0.2$ ceramic: (a) before and (b) after high-temperature aging at 300 °C. These images reveal consistent elemental segregation at the grain level, with no evidence of further phase separation.

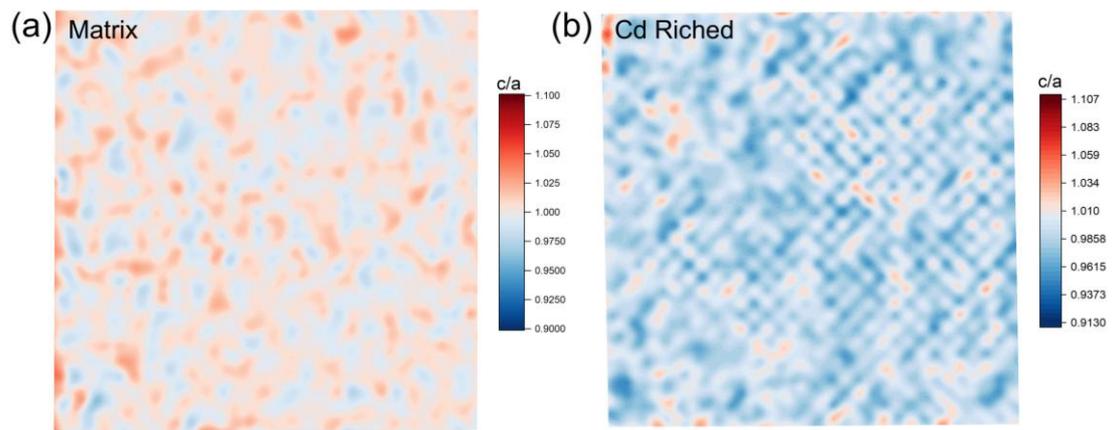


Fig. S5 Lattice distortion (c/a ratio) mapping of the ceramic with $x = 0.2$: (a) main phase; (b) Cd-rich secondary phase.

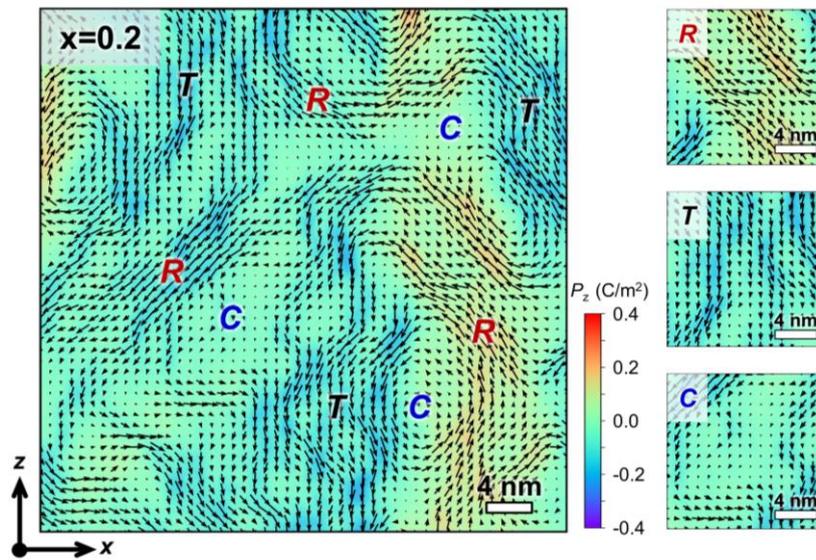


Fig. S6 Equilibrium domain configurations and dielectric phases of the $x = 0.2$ ceramic, as determined by phase-field simulation. The arrows represent the magnitude and orientation of the spontaneous polarization.

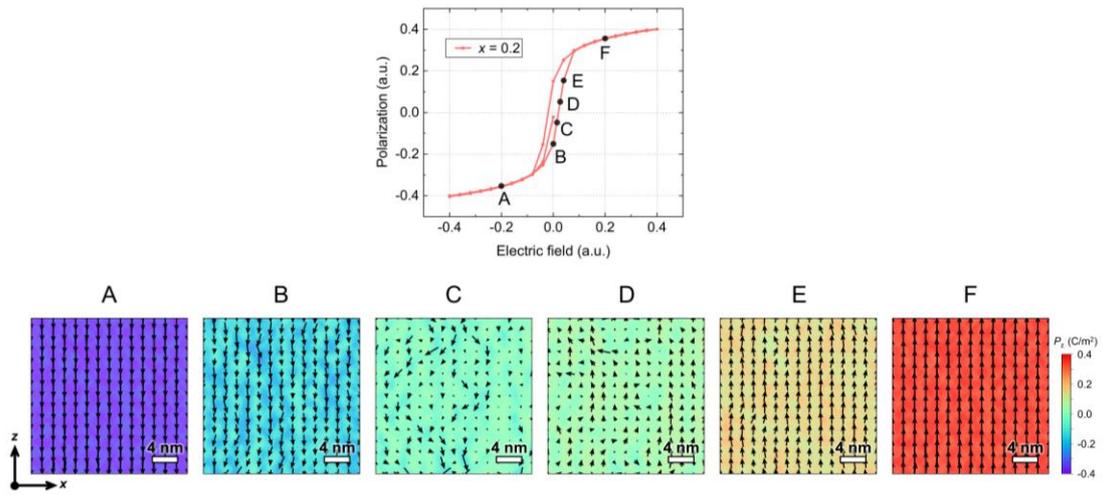


Fig. S7 Calculated P - E loop and dynamic evolution of polar configuration for $x = 0.2$ ceramic at room temperature based on phase-field simulation.

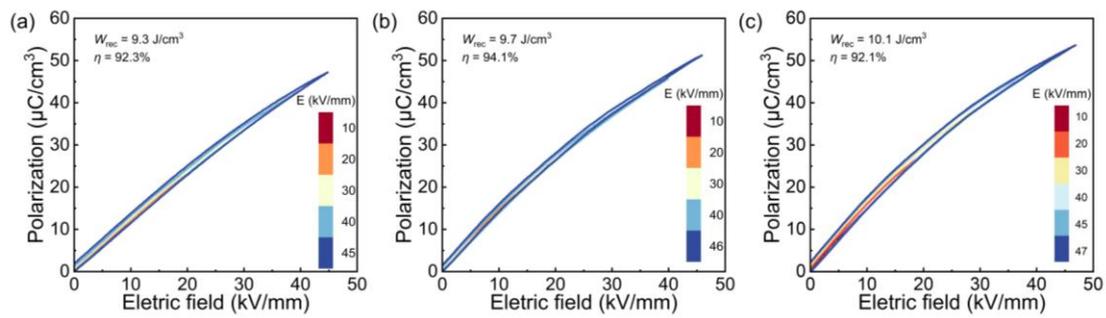


Fig. S8 Unipolar P - E loops of $x = 0.2$ ceramic under multiple measurements on three ceramics.

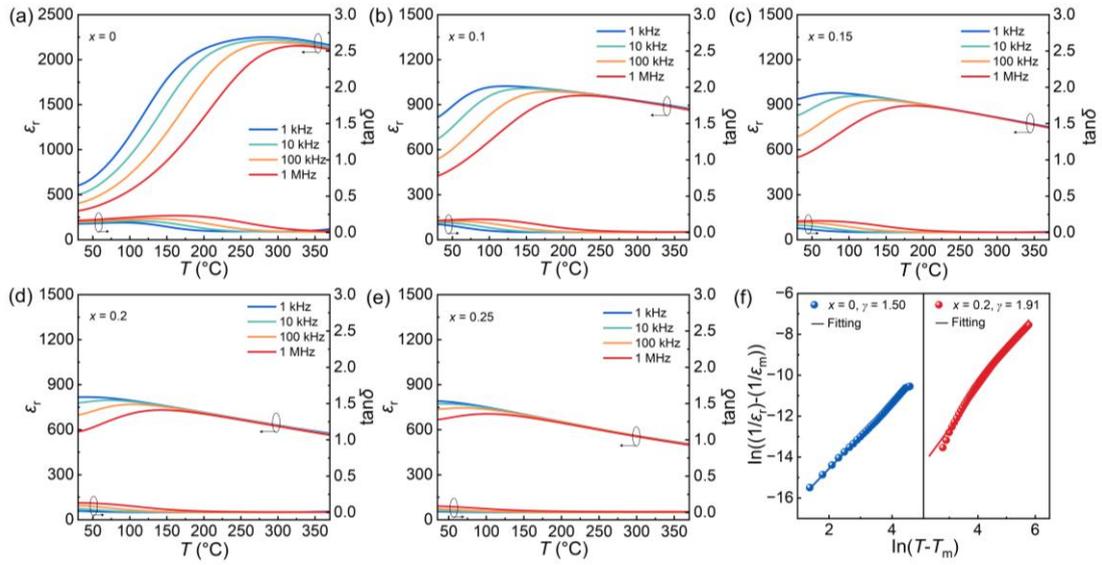


Fig. S9 Dielectric-temperature spectra of 0.5BT-(0.5-x)BMT-xCZ ceramics: (a) $x = 0$, (b) $x = 0.1$, (c) $x = 0.15$, (d) $x = 0.2$, (e) $x = 0.25$. (f) Relaxor degree (γ) of $x = 0$ and $x = 0.2$ ceramics.

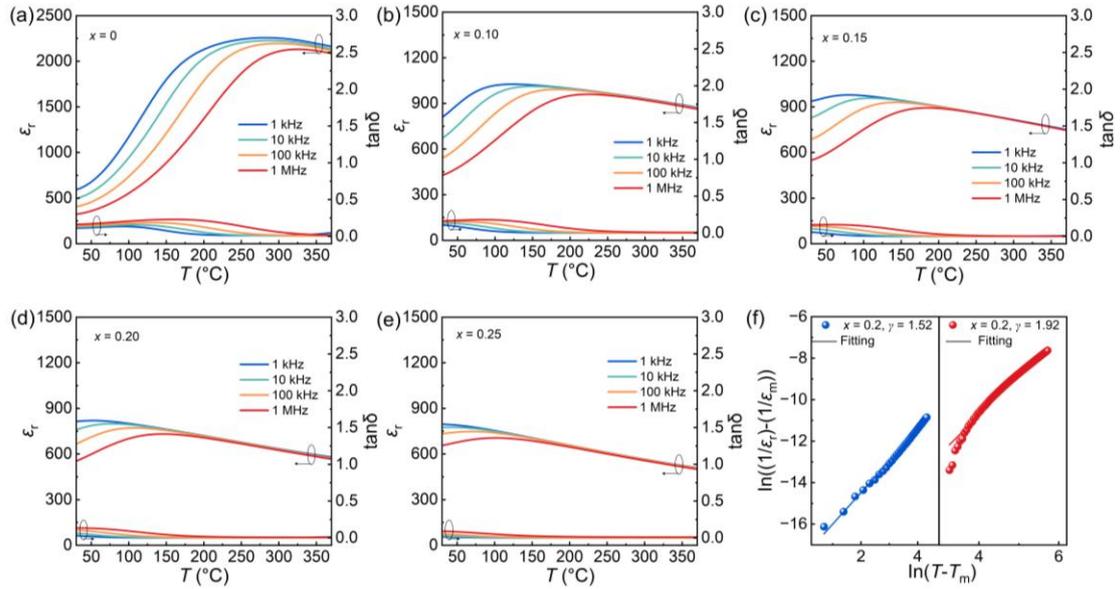


Fig. S10 Temperature-dependent dielectric spectra of 0.5BT-(0.5-x)BMT-xCZ ceramics after 10 months: (a) $x = 0$, (b) $x = 0.1$, (c) $x = 0.15$, (d) $x = 0.2$, (e) $x = 0.25$; (f) Relaxor degree (γ) for $x = 0$ and $x = 0.2$ ceramics. The dielectric-temperature spectra were remeasured on the same samples after 10 months, showing negligible changes and thereby confirming the long-term stability of the dielectric properties.

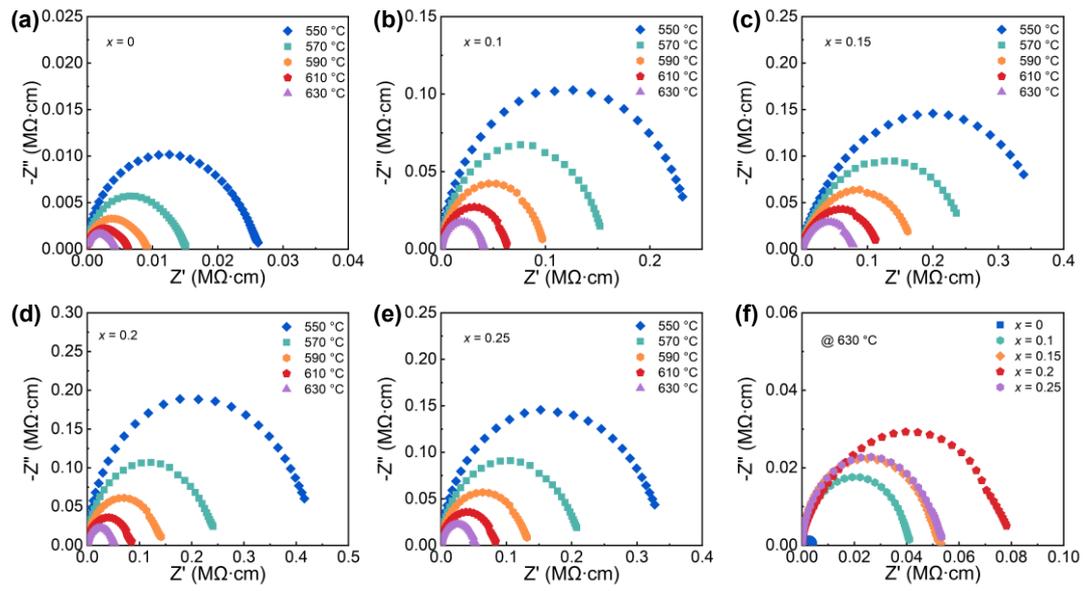


Fig. S11 Complex impedance spectra of 0.5BT-(0.5-x)BMT-xCZ ceramics measured between 550 and 630 °C: (a) $x = 0$, (b) $x = 0.1$, (c) $x = 0.15$, (d) $x = 0.2$, and (e) $x = 0.25$. (f) Impedance spectra of 0.5BT-(0.5-x)BMT-xCZ ceramics at 630 °C.

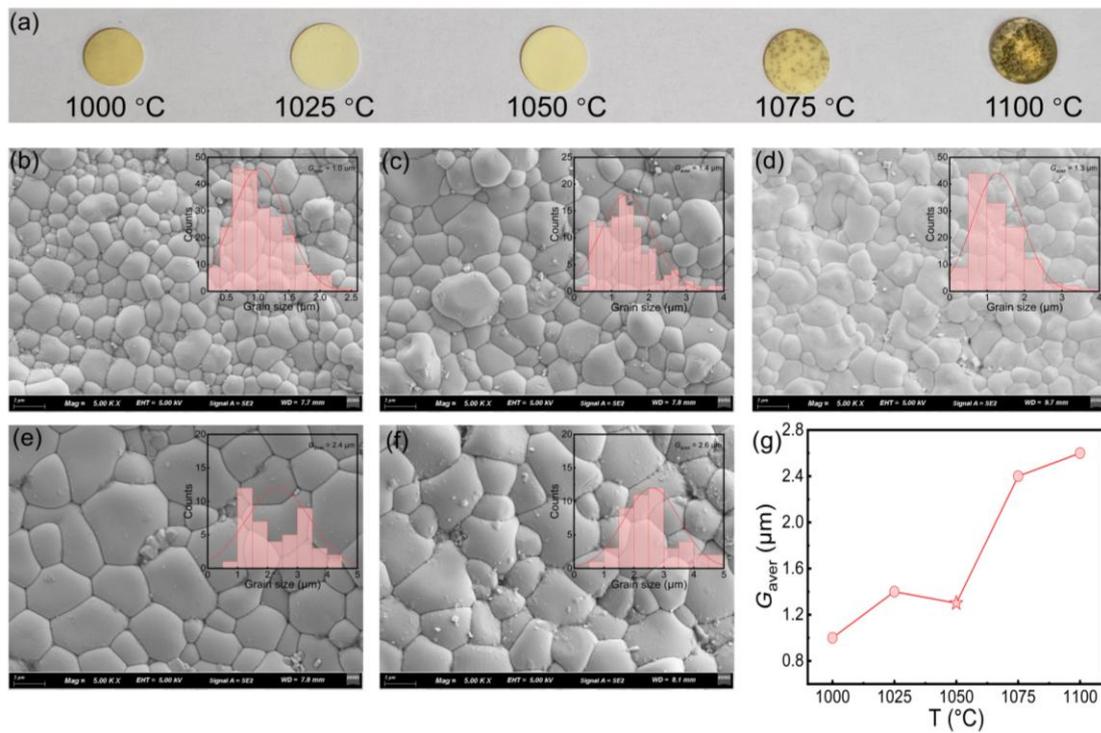


Fig. S12 (a) Digital photograph of the sample; (b–f) SEM morphologies and (g) grain size statistics of $x = 0.2$ ceramics sintered at 1000 °C, 1025 °C, 1050 °C, 1075 °C, and 1100 °C, respectively.

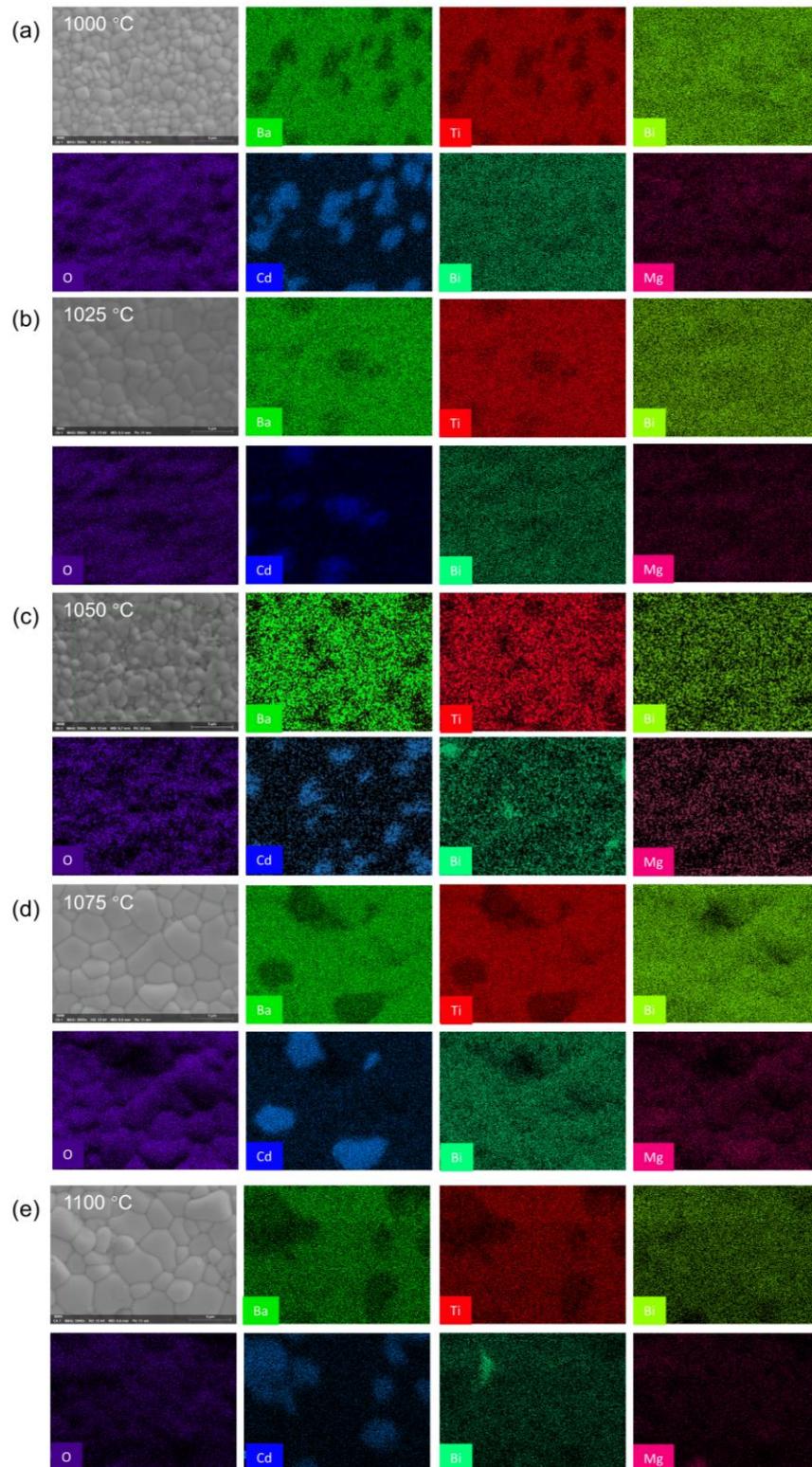


Fig. S13 SEM surface morphologies and corresponding elemental mappings of $x = 0.2$ ceramics sintered at (a) 1000 °C, (b) 1025 °C, (c) 1050 °C, (d) 1075 °C, and (e) 1100 °C.

Table S1. A comparison of W_{rec} and η between 0.5BT–0.3BMT–0.2CZ and other reported two-phase or multi-phase systems ceramics.⁷⁻²⁷

Composition	W_{rec} (J/cm ³)	η (%)	Reference
0.94BT–0.06BMN	4.55	90	Ref. 7
0.45NBT–0.55SBT/6AlN	5.53	90	Ref. 8
BTAN–0.08	6.04	86.8	Ref. 9
0.92SBT–0.08BMH	3.5	92	Ref. 10
BF–BCZT	5.8	91.3	Ref. 11
BT–BMZ@SiO ₂	3.41	85.1	Ref. 12
PLSZST–1 wt% ZnO	4.25	95.5	Ref. 13
0.3PLZS–Al ₂ O ₃	13.2	74	Ref. 14
0.55BNT–0.45ST	4.94	88.45	Ref. 15
0.55(BNT–CS)–0.45SLT	4.55	83	Ref. 16
0.91BST–0.09NBN–VPP	8.5	93.4	Ref. 17
0.4BST–0.4BNT–0.2SZ	10	91	Ref. 18
00.62BF–0.3BT–0.08NZZ	10.5	87	Ref. 19
0.96(BKT–SBT)–0.04BMH	4.25	87	Ref. 20
BNTNN@0.8 wt% SiO ₂	6.17	70	Ref. 21
BNNT–SBT–0.3Mn/BNNT–SBT– 0.2Mn	6.03/5.84	71/85	Ref. 22
B–0.5SB	7.19	93.8	Ref. 23
BSTN–0.1Ta	12.2	89.5	Ref. 24
BNKLST	16.4	90	Ref. 25
0.5NN–0.5BNT core-shell	12.7	87.2	Ref. 26
SMT3	10.3	85.4	Ref. 27
0.5BT–0.3BMT–0.2CZ	9.7	92.8	This work

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