

Supporting Information

Centimeter-Sized Single Crystals of 3D perovskitoid (4-AP)Pb₂X₆ (X = Br, I) for Efficient and Stable X-ray Detection

Hongliang Dai,^a Zeng-Kui Zhu,^{*b} Shihai You,^c Junhua Luo^{*c}

^aSchool of Chemistry and Chemical Engineering, Key Laboratory of Jiangxi Province for Special Optoelectronic Artificial Crystal Materials, Jingtangshan University, Ji'an, Jiangxi, 343009, China

^bCollege of Chemistry and Materials

Jiangxi Province Key Laboratory of Porous Functional Materials; Jiangxi Normal University, Nanchang, Jiangxi, 330022, China. E-mail: zkzhu@jxnu.edu.cn

^cState Key Laboratory of Functional Crystals and Devices

Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, Fujian, 350002, China. E-mail: jhluo@fjirsm.ac.cn

Materials 4-Amidinopyridinium (4-AP) chloride (98%, Aladdin); lead acetate trihydrate (Pb(Ac)₂•3H₂O, 99.5%, Aladdin), hydrobromic acid (HBr, 48%, Aladdin), hydroiodic acid (HI, 55-57%, Aladdin). All the chemicals were bought and used without further purification.

Synthesis Crystalline materials of (4-AP)Pb₂Br/I₆ were obtained from saturated HBr/I solution (45 mL) containing stoichiometric amounts of Pb(Ac)₂•3H₂O (4 mmol, 1.516 g) and corresponding 4-Amidinopyridinium (4-AP) chloride (1 mmol, 0.157 g). The saturated solution was then heated and stirred for 30 min at 100°C to get the clear solution. With the solution temperature decreasing to saturated status, microcrystals were obtained by spontaneous nucleation. Bulk crystals were grown by slowly cooling (1°C/day) the above solution from 80°C to 30°C.

Powder X-ray diffraction Powder X-ray diffraction (PXRD) was recorded on a Rigaku MiniFlex diffractometer at room temperature. The diffraction patterns were collected in the 2θ range of 5°-50° with a step size of 0.02°. The experimental PXRD patterns match fairly well with the simulated data based on the single-crystal structure, which confirm the pure phase of (4-AP)Pb₂Br/I₆ (Fig. S1-2). For the sample after exposed to ambient air for 90 days, it shows environmental stability with negligible phase change. (Fig. 2d-e)

Thermogravimetric analysis (TGA) measurement TGA was performed on STA449C Thermal Analyser ranging from the room temperature to 1000°C (Fig. 1c).

Single-crystal structure determination: Single-crystal X-ray diffraction was performed on a Bruker 8 diffractometer with the Mo Kα radiation. The data were processed by the Crystalclear software package. The structures were solved by direct methods and then refined by the full-matrix least-squares refinements on F2 using SHELXLTL software package. Crystallographic data and structure refinements for (4-AP)Pb₂Br/I₆ are given in Table S1.

X-ray detection The I-V traces and I-t curves under X-ray irradiation were recorded using the 6517B high precision electrometer (Keithley, USA). A commercially available Ag target X-ray tube with X-ray photons energy up to 50 keV and peak intensity at 22 keV was used as the X-ray source (4 W, Mini-X2, Amptek, USA). The dose rate of X-ray tube was modulated by changing its tube current and measured by a commercial X-ray dosimeter (Accu-Gold, Radcal, USA) attached with the ion chamber (10X6-180 model) in an integrating mode.

Calculations of mobility lifetime product ($\mu\tau$) the $\mu\tau$ product has been extracted by fitting the voltage-dependent photoconductivity under X-ray irradiation with the standard Hecht equation,

$$I = \frac{I_0 \mu \tau V}{L^2} \left[1 - \exp\left(-\frac{L^2}{\mu \tau V}\right) \right]$$

where I is the photocurrent, I_0 is the saturated photocurrent, L is the electrode spacing (1 mm for both A and B detectors), V is the bias voltage, μ is carrier mobility, and τ is carrier lifetime.

Calculations of sensitivity (S) and signal-to-noise ratio (SNR) S is defined as the collected charge per unit area under X-ray irradiation and can be determined by

$$S = (I_{x\text{-ray}} - I_d) / (D \times A)$$

where $I_{x\text{-ray}}$ and I_d are the currents recorded under X-ray irradiation and in the dark, respectively, D is the irradiation dose rate, and A is the effective area of detector.

SNR is calculated by

$$SNR = (I_{x\text{-ray}} - I_d) / I_{\text{noise}}$$

where I_{noise} is the noise current and is obtained by calculating the standard deviation of the $I_{x\text{-ray}}$.

Calculation of Dark Current Drift (I_{drift}) The dark current drift is determined by the following equation

$$I_{\text{drift}} = (I_t - I_0) / (E \times S \times t)$$

where I_0 is the current immediately after stabilization, I_t is the current at time t , E is the electric field, S is the device area.

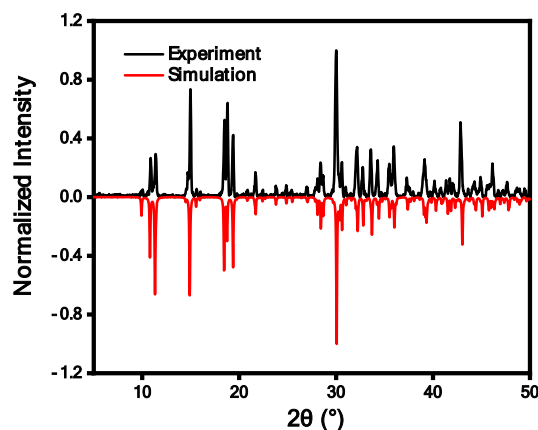


Fig. S1 Experimental and simulated powder X-ray diffraction patterns of (4-AP)Pb₂Br₆.

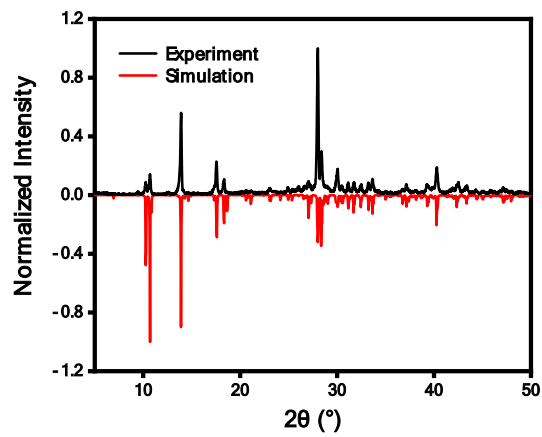


Fig. S2 Experimental and simulated powder X-ray diffraction patterns of (4-AP)Pb₂I₆.

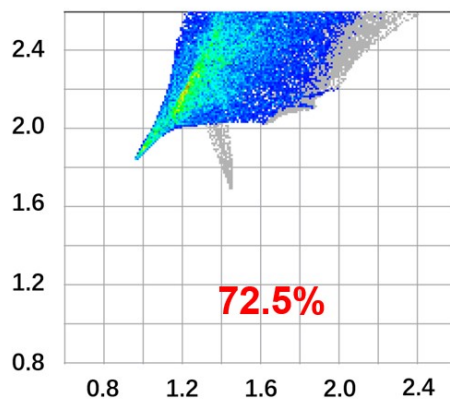


Fig. S3 2D fingerprint plots for 4-AP²⁺ cations in (4-AP)Pb₂I₆.

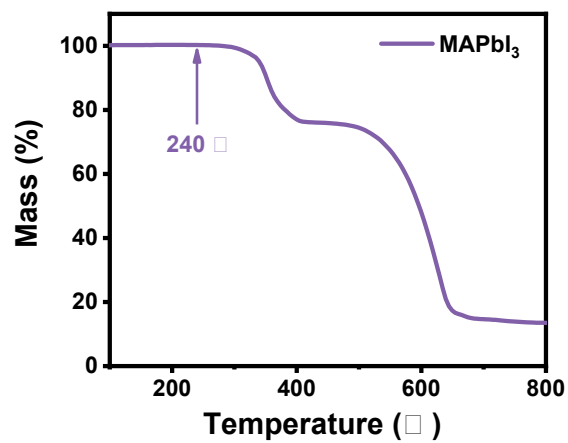


Fig. S4 The thermogravimetric cure of MAPbI₃.

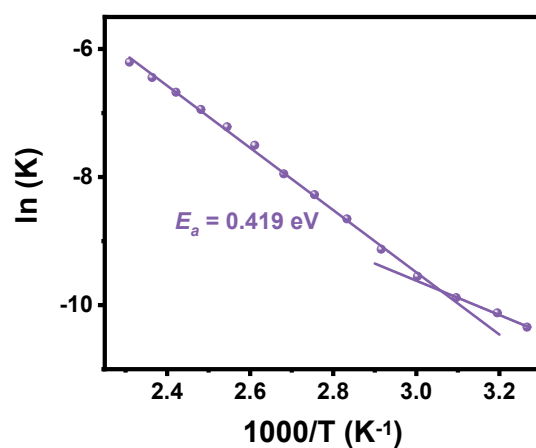


Fig. S5 Temperature-dependent conductivity of MAPbI₃ SC.

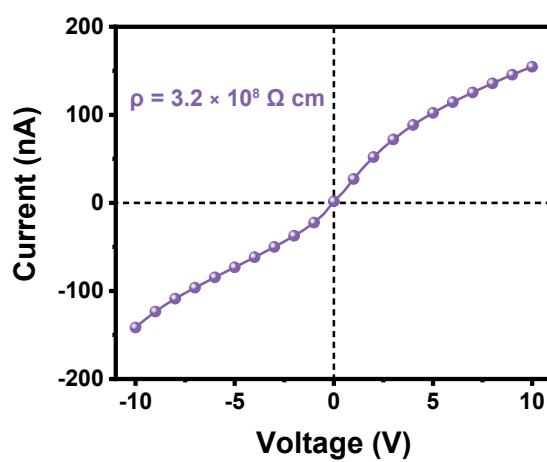


Fig. S6 Resistivity of MAPbI₃ SC.

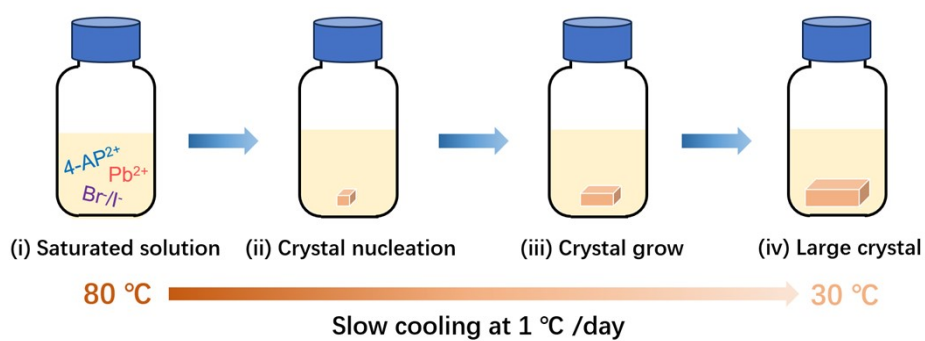


Fig. S7 Schematic diagram of the crystal growth process.

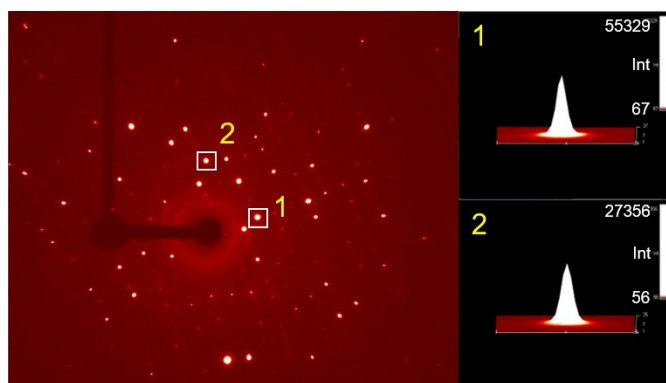


Fig. S8 SCXRD diffraction spots of (4-AP)Pb₂I₆.

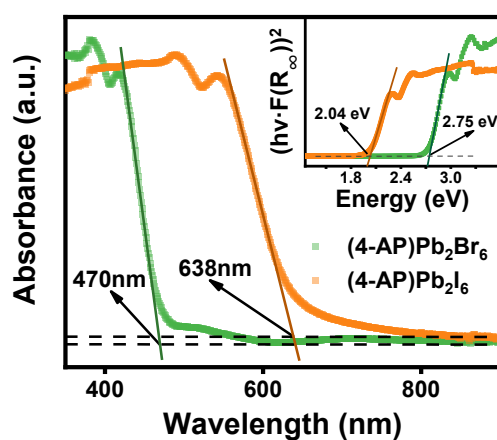


Fig. S9 UV-vis absorption spectra of (4-AP)Pb₂Br₆ and (4-AP)Pb₂I₆. Insert: the energy band estimated by a Tauc's plot.

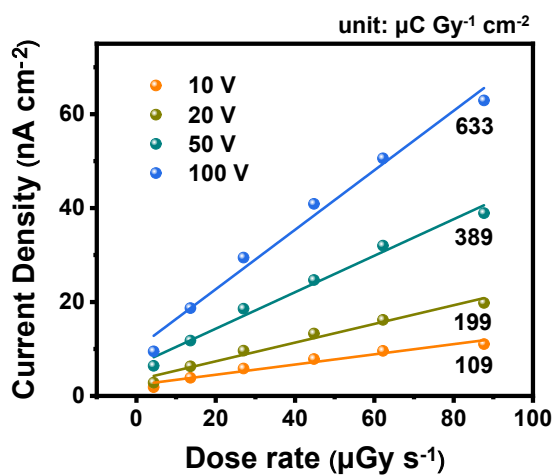


Fig. S10 Photocurrent density of (4-AP)Pb₂Br₆ detectors under various dose rates at different bias voltage.

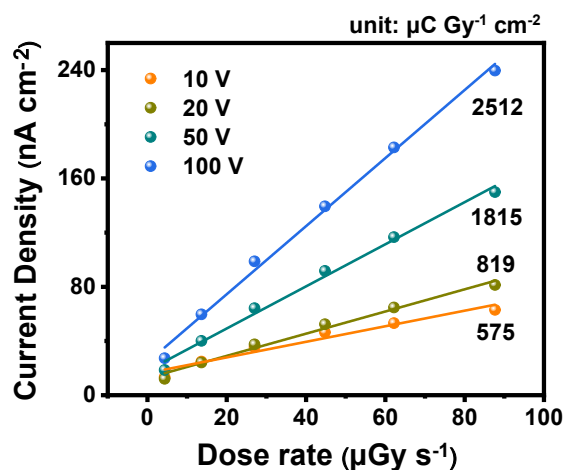


Fig. S11 Photocurrent density of (4-AP)Pb₂I₆ detectors under various dose rates at different bias voltage.

Table S1. Crystallographic Data and Structure Refinements of (4-AP)Pb₂Br/I₆.

	(4-AP)Pb ₂ Br ₆	(4-AP)Pb ₂ I ₆
CCDC Identification code	2312099	2312100
Empirical formula	C ₆ H ₉ Br ₆ N ₃ Pb ₂	C ₆ H ₈ I ₆ N ₃ Pb ₂
Formula weight	1008.068	1297.93
Temperature/K	295.00(10)	295.00(10)
Crystal system	orthorhombic	orthorhombic
Space group	<i>Pnna</i>	<i>Pnna</i>
<i>a</i> /Å	23.9011(15)	25.5050(12)
<i>b</i> /Å	15.7772(10)	16.5271(9)
<i>c</i> /Å	9.6867(5)	10.0933(6)
α /°	90	90
β /°	90	90
γ /°	90	90
Volume/Å ³	3652.8(4)	4254.6(4)
Z	8	8
ρ_{calc} /cm ³	3.666	4.053
μ /mm ⁻¹	31.525	24.503
<i>F</i> (000)	3399.3	4376.0
Radiation	Mo K α (λ = 0.71073)	Mo K α (λ = 0.71073)
2 θ range for data collection/°	6.82 to 57.68	6.57 to 57.634
Index ranges	-31 $\leq h \leq$ 26, -19 $\leq k \leq$ 18, -10 $\leq l \leq$ 12	-29 $\leq h \leq$ 33, -18 $\leq k \leq$ 22, -13 $\leq l \leq$ 10
Reflections collected	17692	17964
Independent reflections	4042 [R_{int} = 0.0459, R_{sigma} = 0.0391]	4758 [R_{int} = 0.0408, R_{sigma} = 0.0366]
Data/restraints/parameters	4042/0/157	4758/0/157
Goodness-of-fit on <i>F</i> ²	1.061	1.054
Final R indexes [$I > 2\sigma(I)$]	R_1 = 0.0352, wR_2 = 0.0731	R_1 = 0.0419, wR_2 = 0.0917
Final R indexes [all data]	R_1 = 0.0615, wR_2 = 0.0867	R_1 = 0.0640, wR_2 = 0.1038

Table S2. Performances of some reported MHP and perovskitoid X-ray single-crystal detectors.

Materials	Dimensionality	Sensitivity ($\mu\text{C Gy}^{-1} \text{cm}^{-2}$)	Detection limit (nGy s^{-1})	Current drift ($\text{nA cm}^{-1} \text{s}^{-1} \text{V}^{-1}$)	Ref.
(4-AP)Pb ₂ Br ₆	3D	633	14900	3.1×10^{-6}	This work
(4-AP)Pb ₂ I ₆		2512	720	1.48×10^{-5}	
Cs ₂ AgBiBr ₆	3D	1974	45.7	-	[1]
CsPbBr ₃	3D	46180	10.81	1.68×10^{-6}	[2]
MAPbI ₃	3D	5.2×10^6	0.1	-	[3]
MAPbBr ₃	3D	96000	2.8	-	[4]
FAPbBr ₃	3D	185.64	133	-	[5]
CsFAGAPb(I _{0.9} Br _{0.1}) ₃ : Sr	3D	26000	7.09	8.77×10^{-5}	[6]
FAPbBr ₃	3D	130	300	-	[7]
(3AMPY)Pb ₂ I ₆	3D	207	-	-	[8]
MhyPbBr ₃	3D	220 (0 V)	203	0	[9]
		514 (100 V)	5428	5.77×10^{-5}	
(R/S-BPEA)EA ₆ Pb ₄ Cl ₁₅	3D	1709.09	3500	-	[10]
[(Cu(O ₂ C-(CH ₂) ₃ -NH ₃) ₂) ₂]PbBr ₄	2D	1.14×10^5	56	-	[11]
(BZA) ₂ (MA)Pb ₂ Br ₇	2D	97.9	266	3.9×10^{-6}	[12]
(CH ₃ OC ₃ H ₉ N) ₂ CsPb ₂ Br ₇	2D	410	-	-	[13]
BA ₂ EA ₂ Pb ₃ I ₁₀	2D	5555.7	17.2	9.25×10^{-6}	[14]
(4ABA)PbI ₄	2D	150	7.5	5.18×10^{-8}	[15]
(BA) ₂ PbI ₄	2D	204	241	-	[16]
(PMA) ₂ PbI ₄	2D	283	2130	-	[17]
(F-PEA) ₂ PbI ₄	2D	3402	23	4.9×10^{-8}	[18]
(o-F-PEA) ₂ PbI ₄	2D	1724.5	460	8.48×10^{-8}	[19]
(S-BPEA) ₂ FAPb ₂ I ₇	2D	87.8	161	-	[20]
		1985.9	1100	4.4×10^{-7}	
(2IPA) ₂ FAPb ₂ I ₇	2D	438	20	7.76×10^{-6}	[21]
(1,3-BMACH)(MA)Pb ₂ I ₇	2D	199	106	-	[22]
BDAPbI ₄	2D	242	430	6.06×10^{-9}	[23]
(1,3-BMACH)BiI ₅	1D	170	49	-	[22]
(BZA) ₂ (R/S-PPA)BiI ₆	1D	53.2	18.5	-	[24]
		2170	-	-	
(BAH)BiI ₄	1D	1181.8	77	-	[25]
MA ₃ Bi ₂ I ₉	0D	1947	83	5.0×10^{-10}	[26]
(R/S-PPA) ₂ BiI ₅	0D	31	270	0	[27]
		150 (10 V)		1.40×10^{-4}	
(HIS)BiI ₅	0D	1230	36.4	2.559×10^{-7}	[28]

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