

# Supporting Information

## Fe-Assisted Nitridation-Induced Reconstruction of Mo-Fe-Ni Molybdates Enables Durable Alkaline Seawater Oxygen Evolution and Zn-Air Batteries

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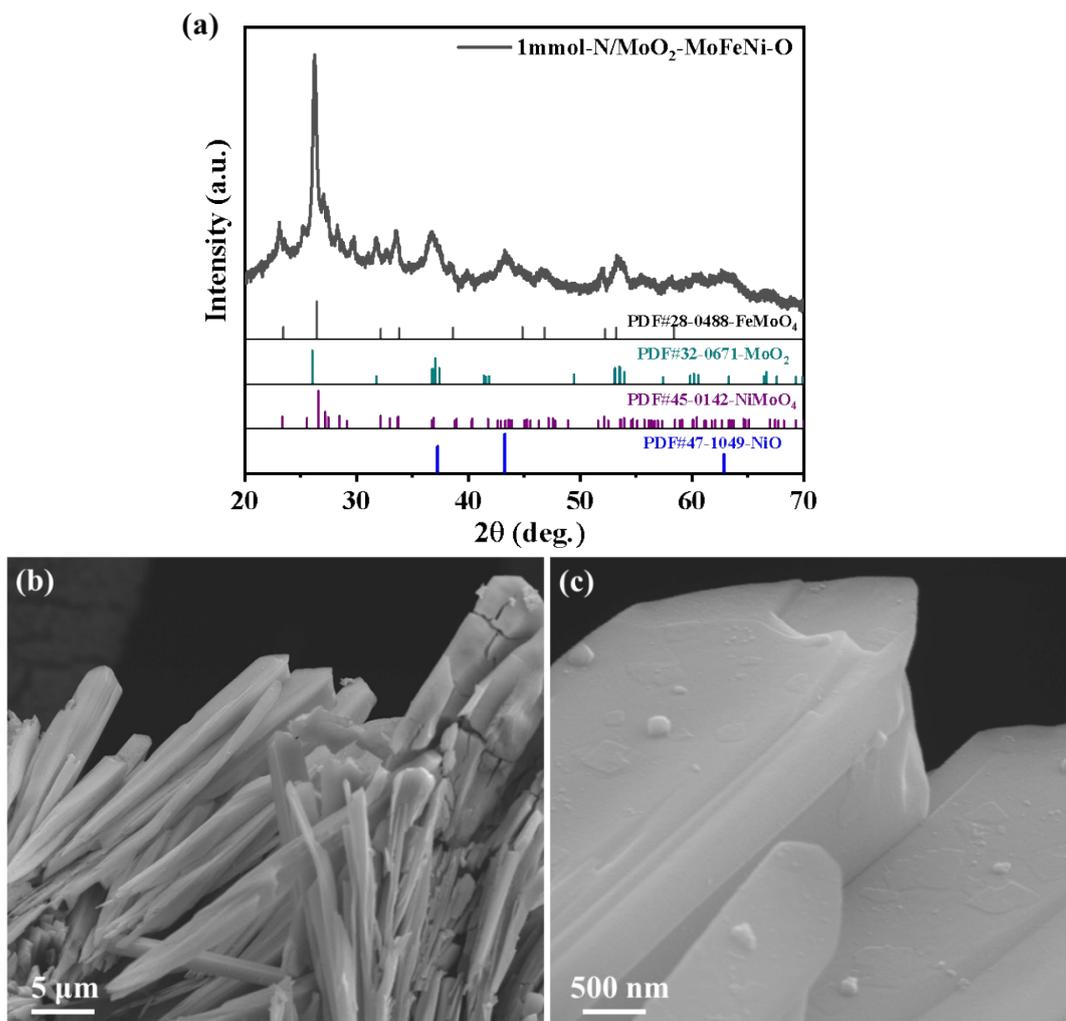
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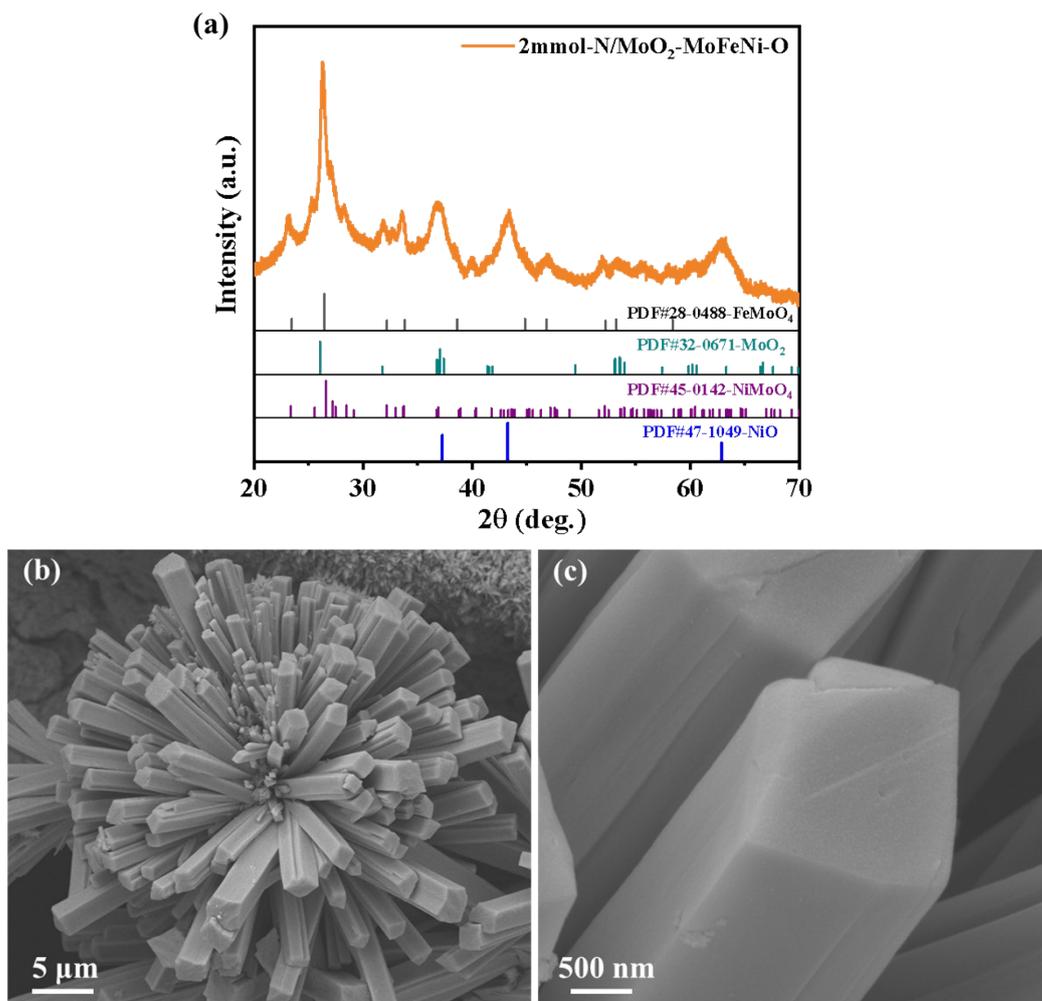
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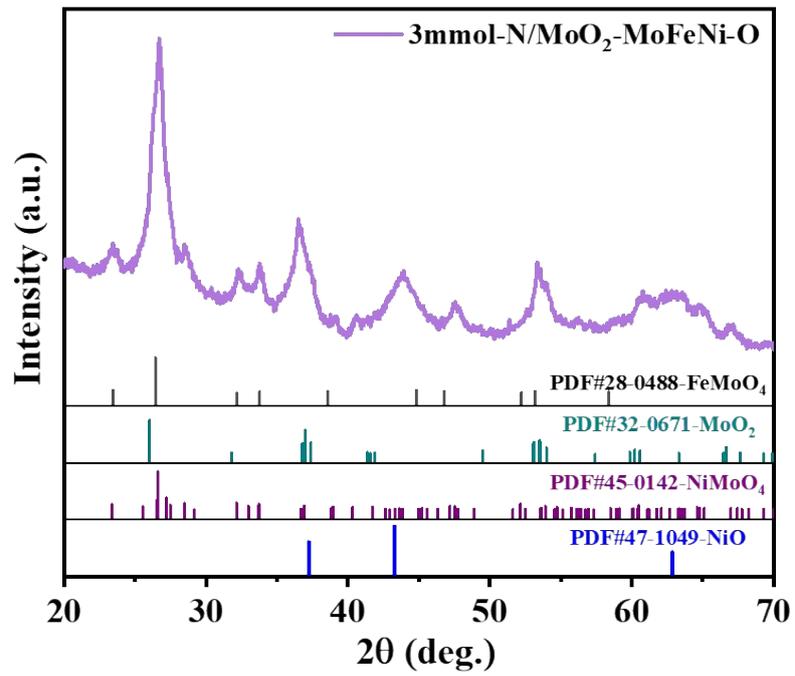
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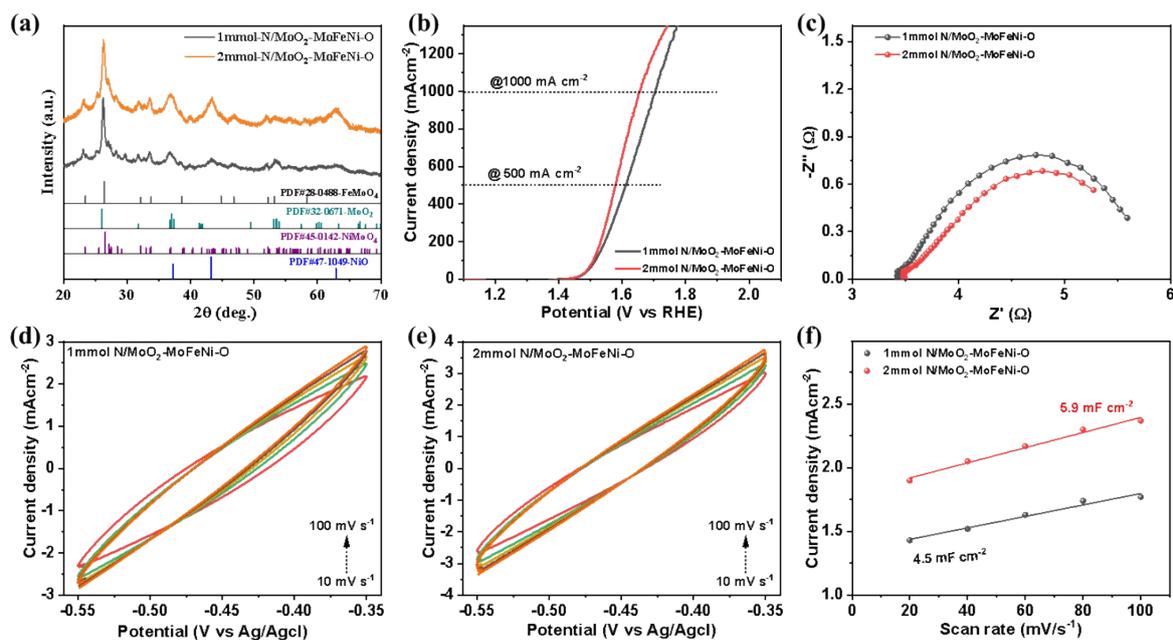
**Figure S1.** (a) XRD pattern, (b-c) SEM images of 1mmol-N/MoO<sub>2</sub>-MoFeNi-O.



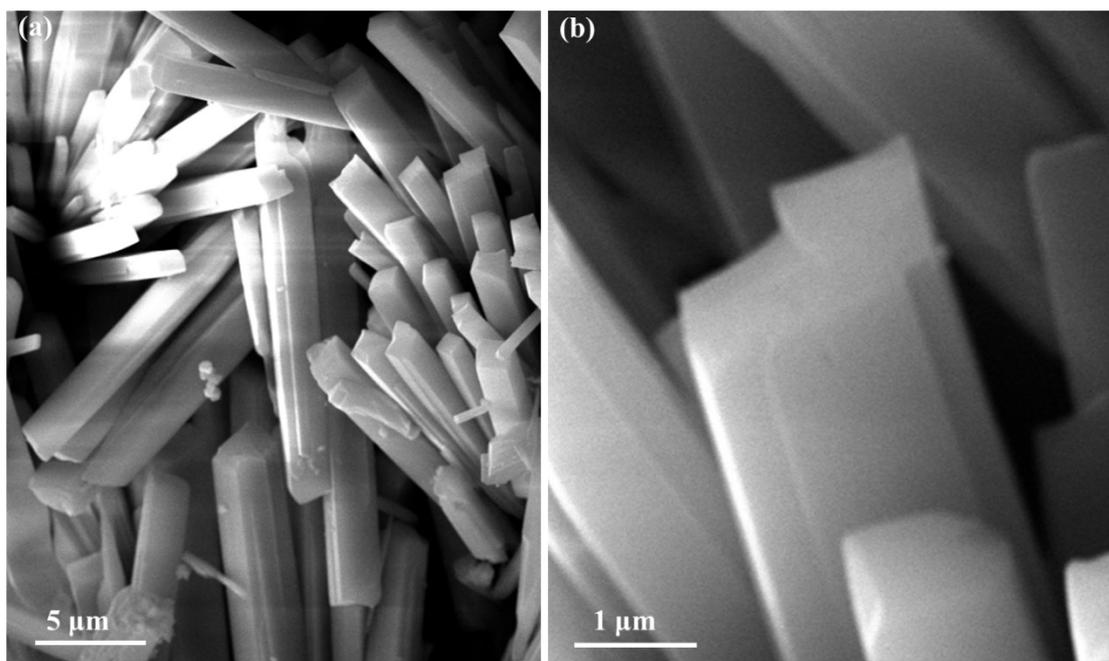
**Figure S2.** (a) XRD pattern, (b-c) SEM images of 2mmol-N/MoO<sub>2</sub>-MoFeNi-O.



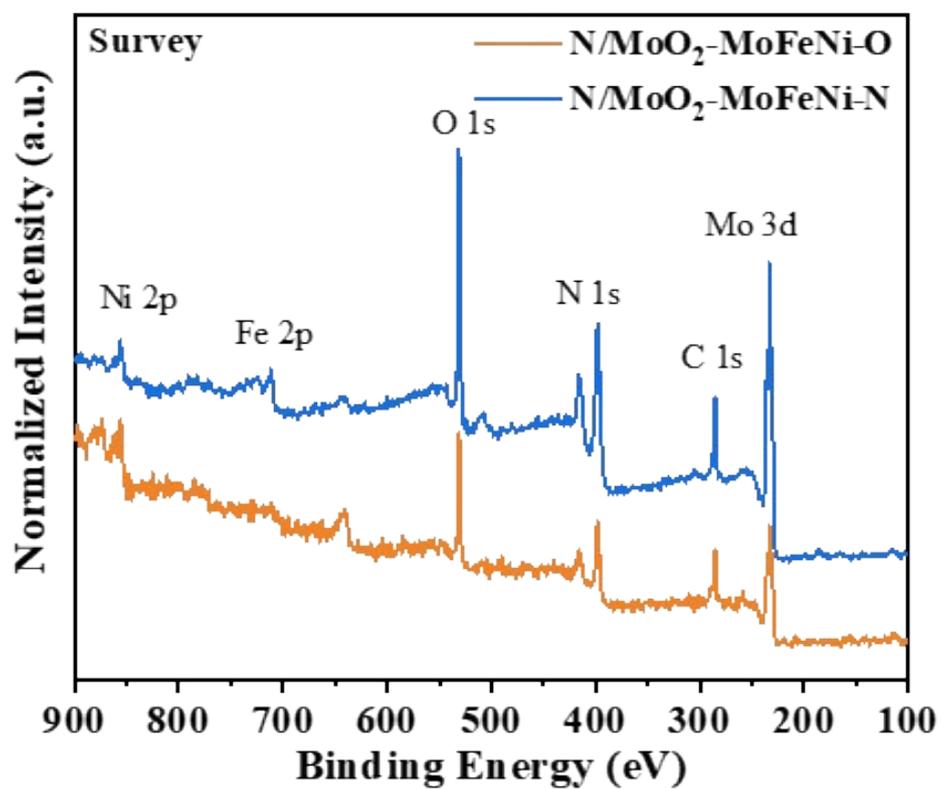
**Figure S3.** XRD pattern of 3mmol-N/MoO<sub>2</sub>-MoFeNi-O.



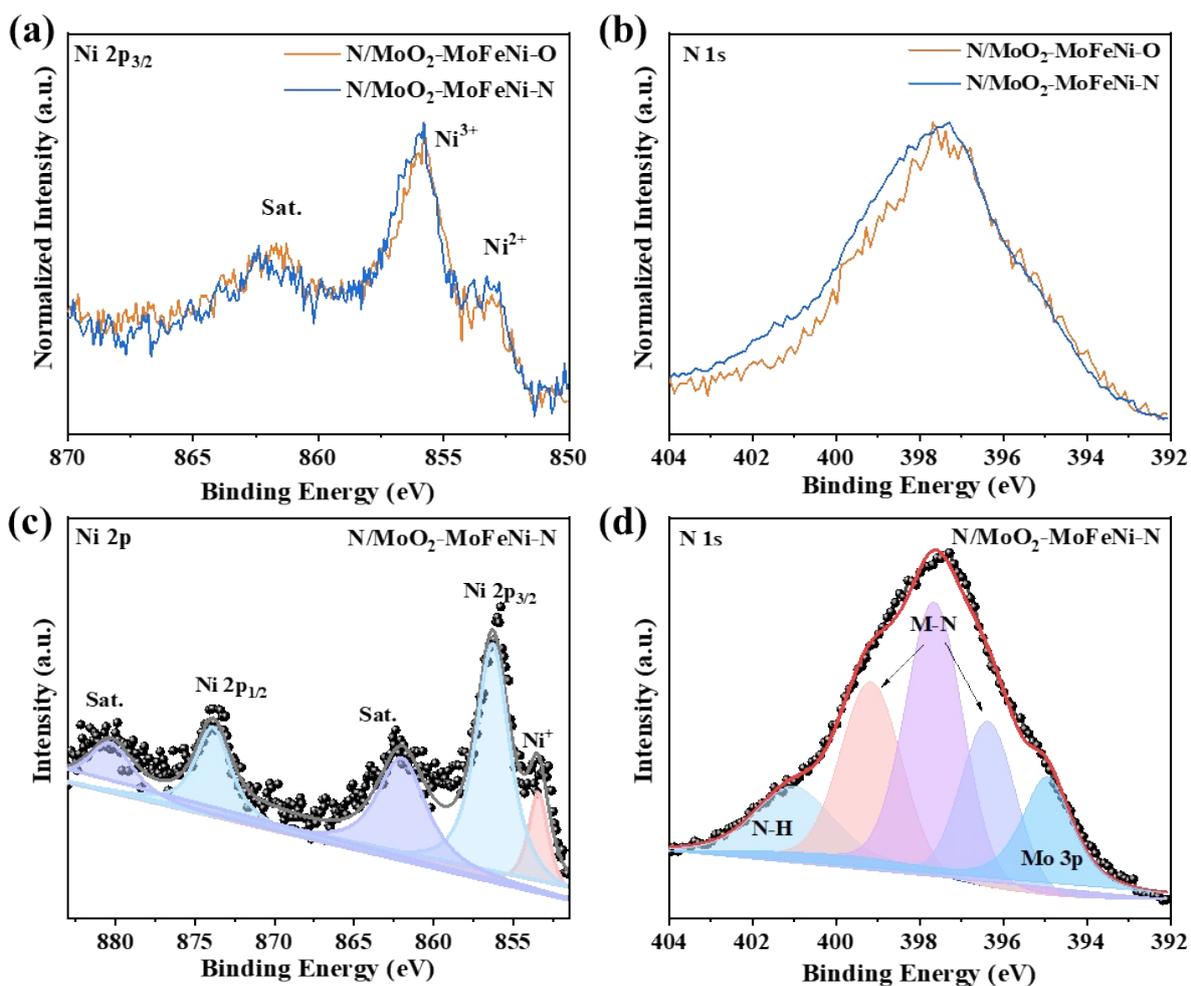
**Figure S4.** Structural and electrochemical analysis of 1mmol- and 2mmol-N/MoO<sub>2</sub>-MoFeNi-O. (a) XRD patterns, (b) OER LSV curves, and (c) EIS spectra of 1mmol- and 2mmol-N/MoO<sub>2</sub>-MoFeNi-O. CV curves of (d) mmol-N/MoO<sub>2</sub>-MoFeNi-O and (e) 2mmol-N/MoO<sub>2</sub>-MoFeNi-O. (f) ECSA comparison of 1mmol- and 2mmol-N/MoO<sub>2</sub>-MoFeNi-O.



**Figure S5.** SEM images of the 4:3 (Fe:Mo) FeMo precursor.



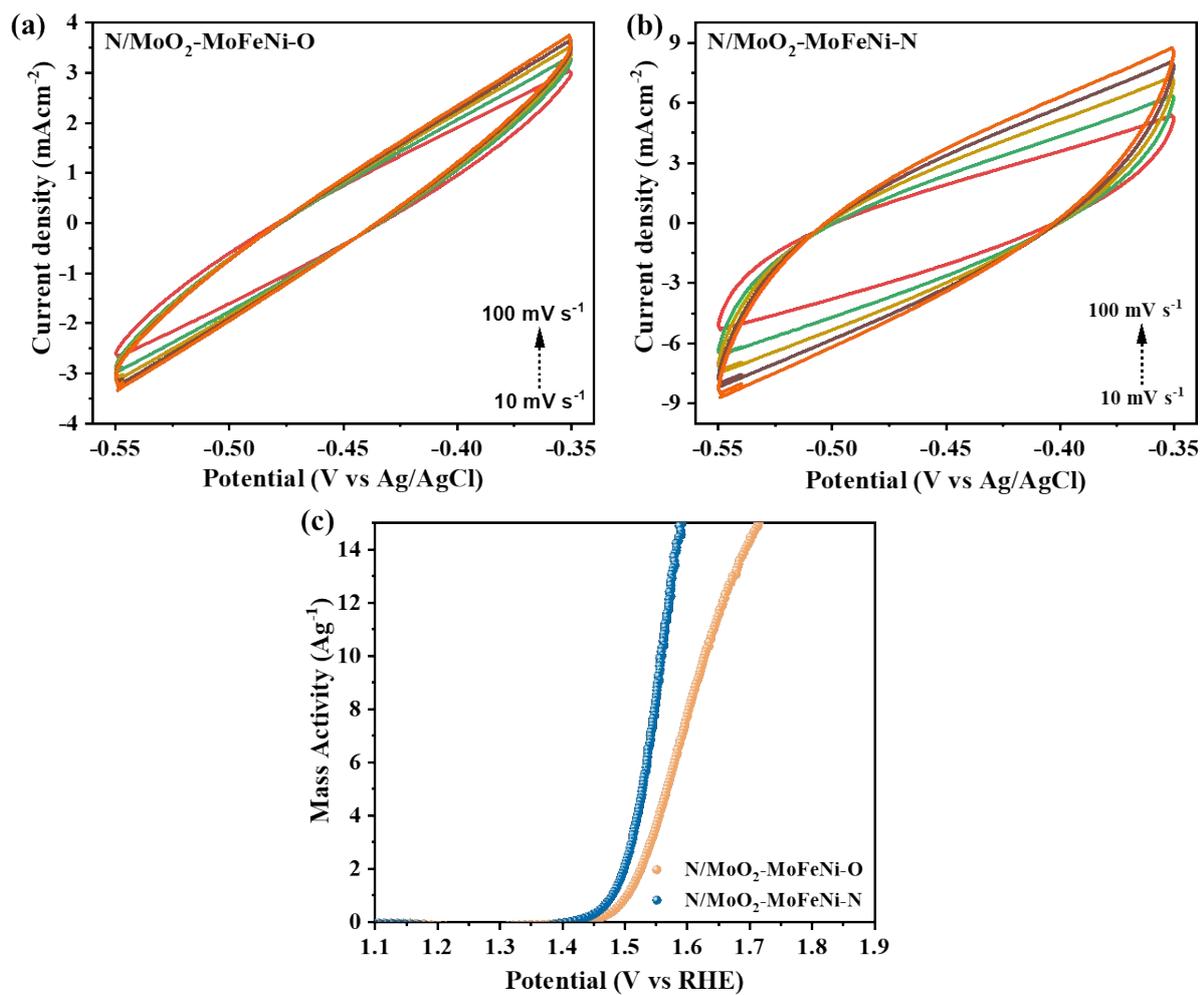
**Figure S6.** XPS survey spectra of N/MoO<sub>2</sub>-MoFeNi-O and N/MoO<sub>2</sub>-MoFeNi-N.



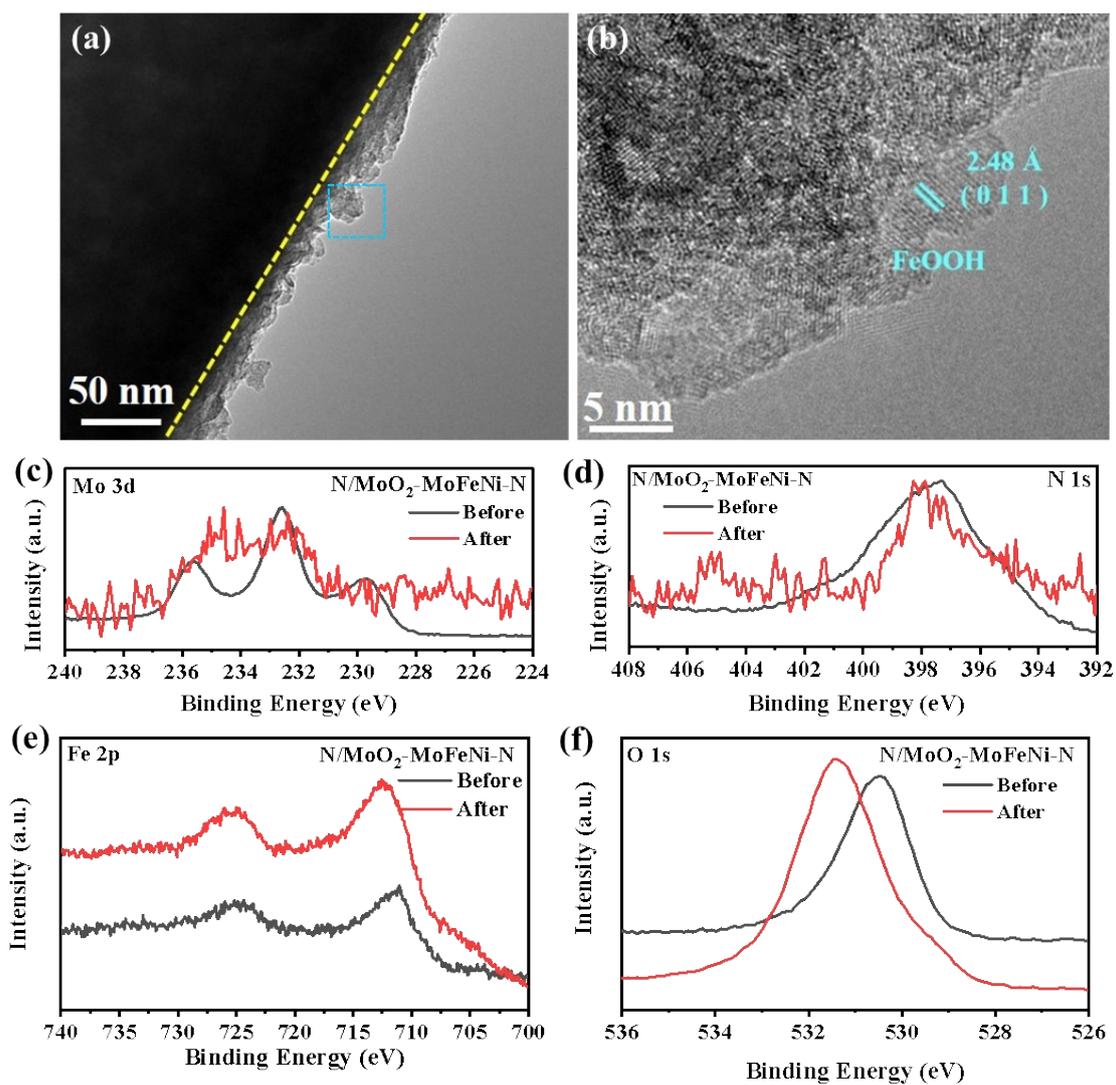
**Figure S7.** Normalized XPS spectra of (a) Ni 2p and (b) N 1s for N/MoO<sub>2</sub>-MoFeNi-O and N/MoO<sub>2</sub>-MoFeNi-N. Deconvoluted (c) Ni 2p and (d) N 1s XPS spectra of spectra of N/MoO<sub>2</sub>-MoFeNi-N.

**Table S1.** Comparison of some recently reported representative transition metal-based electrocatalysts for OER in alkaline electrolytes.

<b>Electrocatalysts</b>	<b><math>\eta_{10}</math> (mV)</b>	<b>Tafel Slope (mV dec<sup>-1</sup>)</b>	<b>Ref.</b>
<b>N/MoO<sub>2</sub>-MoFeNi-N</b>	<b>200</b>	<b>62</b>	<b>This work</b>
N/MoO <sub>2</sub> -MoFeNi-O	220	65	This work
Co <sub>5</sub> Mo <sub>1.0</sub> O NSs@NF	270	54.4	Nano Energy 45 (2018) 448-455.
Co <sub>4</sub> Mo <sub>2</sub> @NC	330	49	J. Mater. Chem. A 5 (2017) 16929-16935.
Fe <sub>1</sub> Co <sub>1</sub> -ONS	308	36.8	Adv. Mater. 29 (2017) 1606793.
PNG-NiCo <sub>2</sub> O <sub>4</sub>	349	156	ACS Nano 7 (2013) 10190- 10196.
CoFe <sub>2</sub> O <sub>4</sub>	410	82	Nanoscale 7 (2015) 8920-8930.
CoCr <sub>2</sub> O <sub>4</sub>	422	63	Small 12 (2016) 2866-2871.
NiMnO NM	275	41	Electrochim. Acta 245 (2017) 32- 40.
MnFe <sub>2</sub> O <sub>4</sub>	520	114	Nanoscale 7 (2015) 8920-8930.
Co <sub>6</sub> Mo <sub>6</sub> C <sub>2</sub> /NCRGO	260	50	ACS Appl. Mater. Interfaces 2017, 9, 20, 16977-16985
Co <sub>3</sub> Mo <sub>3</sub> C/N-C	316	89.9	Int. J. Hydrogen. Energ. 46 (2021) 22268-e22276



**Figure S8.** CV curves of (a) N/MoO<sub>2</sub>-MoFeNi-O and (b) N/MoO<sub>2</sub>-MoFeNi-N.



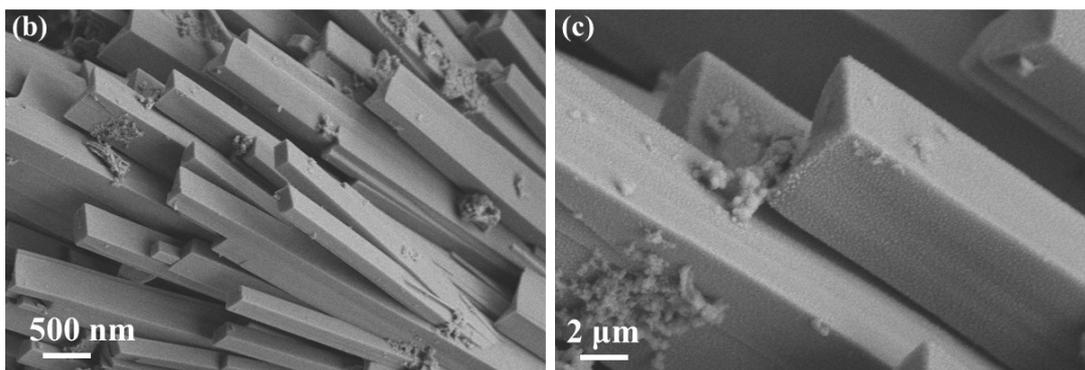
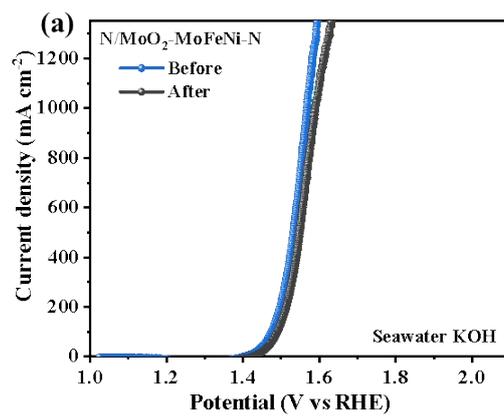
**Figure S9.** (a) TEM and (b) HRTEM images, and corresponding XPS spectra showing (c) Mo 3d, (d) N 1s, (e) Fe 2p, and (f) O 1s of N/MoO<sub>2</sub>-MoFeNi-N after the stability test in freshwater KOH electrolyte.

**Table S2.** Comparison of OER performance of N/MoO<sub>2</sub>-MoFeNi-N in freshwater and seawater electrolytes.

<b>Electrolyte</b>	<b><math>\eta</math> @ 500 mA cm<sup>-2</sup></b>	<b><math>\eta</math> @ 1000 mA cm<sup>-2</sup></b>	<b>Tafel slope</b>
Freshwater KOH	309 mV	392 mV	56.6 mV dec <sup>-1</sup>
Seawater KOH	334 mV	499 mV	71.1 mV dec <sup>-1</sup>

**Table S3.** Comparison of OER metrics in freshwater and seawater KOH electrolytes and with other reported literatures.

Electrocatalysts	Freshwater KOH $\eta$ (mV) @500/1000mAcm <sup>-2</sup>	Seawater KOH $\eta$ (mV) @500/1000mAcm <sup>-2</sup>	References
<b>N/MoO<sub>2</sub>-MoFeNi-N</b>	<b>309/334</b>	<b>392/499</b>	<b>This work</b>
RuMoNi	390/480	397/484	Nat Commun 14, 3607 (2023).
Cr-Co <sub>x</sub> P	NA	392/423	Adv. Funct. Mater. 2023, 33, 2214081
NixCryO	NA	460/-	Angew. Chem., Int. Ed. 2023, 62, e202309854
S-(Ni,Fe)OOH	NA	398/462	Energy Environ. Sci., 2020, 13, 3439-3446
Ni <sub>3</sub> FeN@C/NF	341/-	394/-	J. Mater. Chem. A, 2021, 9, 13562
NiCoS/NF	330/390	440/470	Appl. Catal. B Environ., 2021, 291, 120071
NiFe-PBA-gel-cal /GC	398/-	NA	Adv. Sci. 2022, 9, 2200146
MnCo <sub>2</sub> O <sub>4</sub> @NiFe-LDH/NF	NA	578/-	J. Colloid Interface Sci. 2023, 632, 54
S-Cu <sub>2</sub> O-CuO NDLS	400/450	440/-	Catal. Today 2022, 400-401, 14-25
NiCoHPi@Ni <sub>3</sub> N/NF	405/-	474/-	ACS Appl. Mater. Interfaces 2022, 14, 22061-22070
FeCoNiCrMn HEA/NM	-/330	NA	J. Mater. Chem. A, 2025,13, 17384-17392



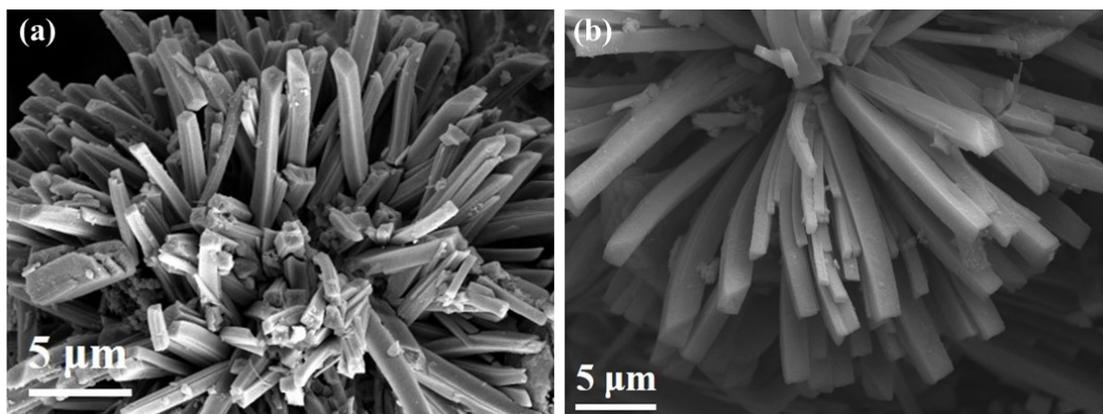
**Figure S10.** (a) LSV curves before and after seawater stability test and (b-c) SEM images of N/MoO<sub>2</sub>-MoFeNi-N after the stability test.



**Figure S11.** Digital image of pH-adjusted 6 M KOH seawater electrolyte after DPD addition.

**Table S4.** ICP analysis of electrolyte before and after electrolysis in alkaline seawater

<b>Element</b>	<b>Before electrolysis (mg L<sup>-1</sup>)</b>	<b>After electrolysis (mg L<sup>-1</sup>)</b>
Mo	0.0	0.0
Fe	0.0	1942.2
Ni	192.0	12446.5



**Figure S12.** SEM images of N/MoO<sub>2</sub>-MoFeNi-N after the OWS stability test in (a) alkaline freshwater and (b) alkaline seawater.

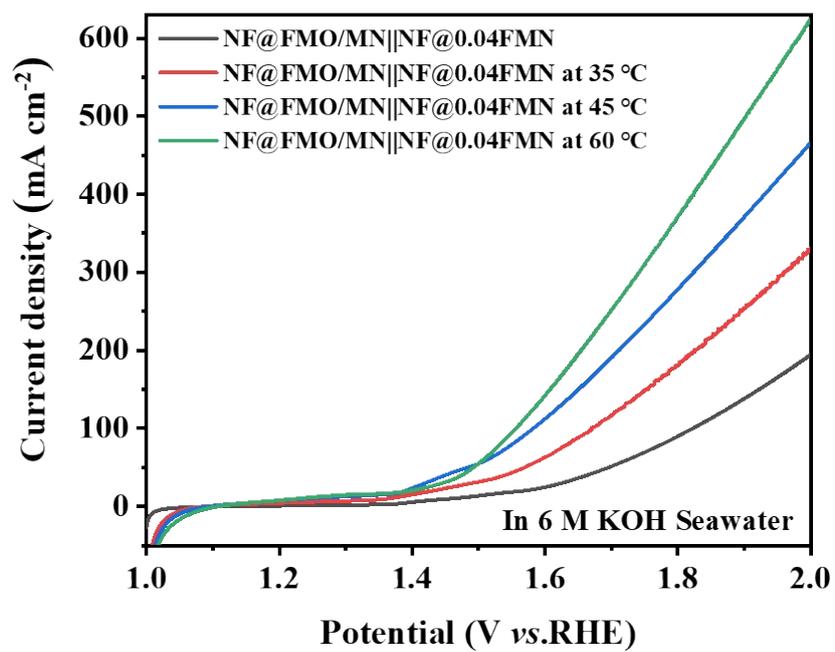
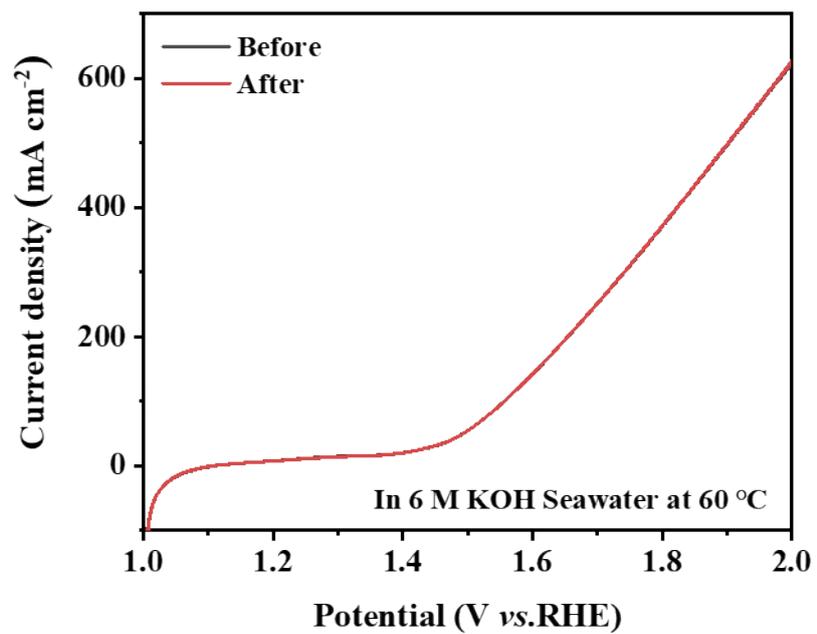
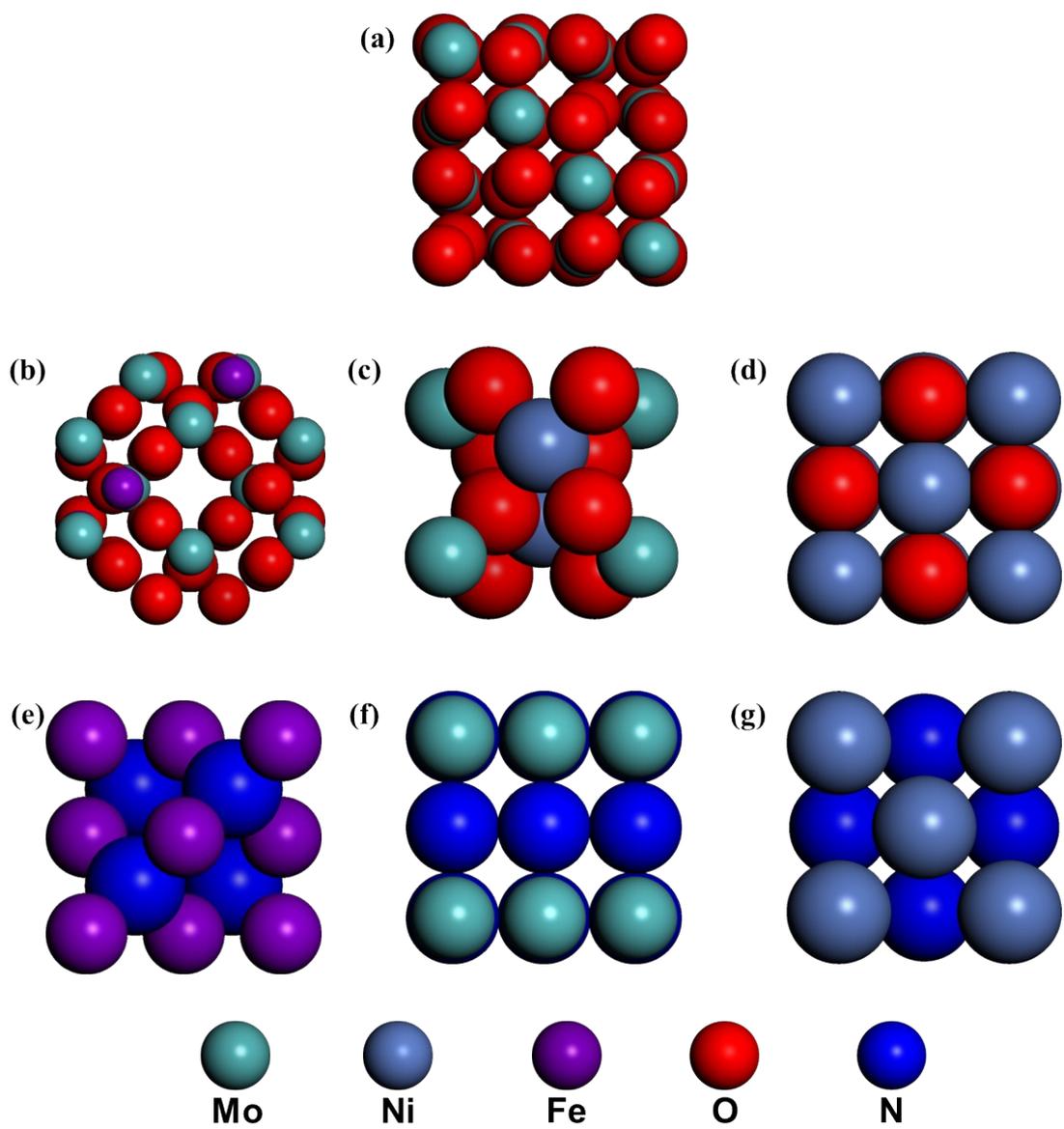


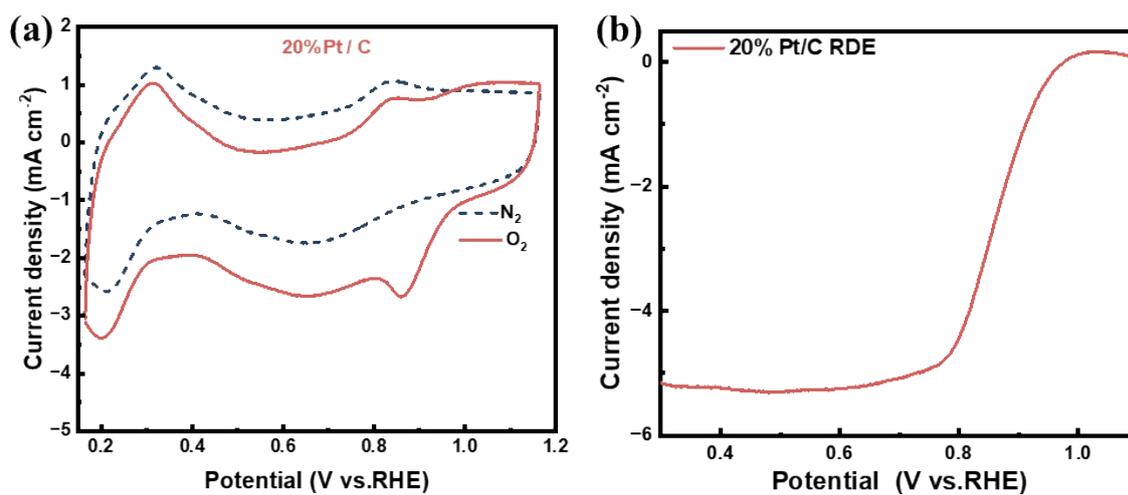
Figure S13. OWS LSV curves at different temperatures in 6 M KOH.



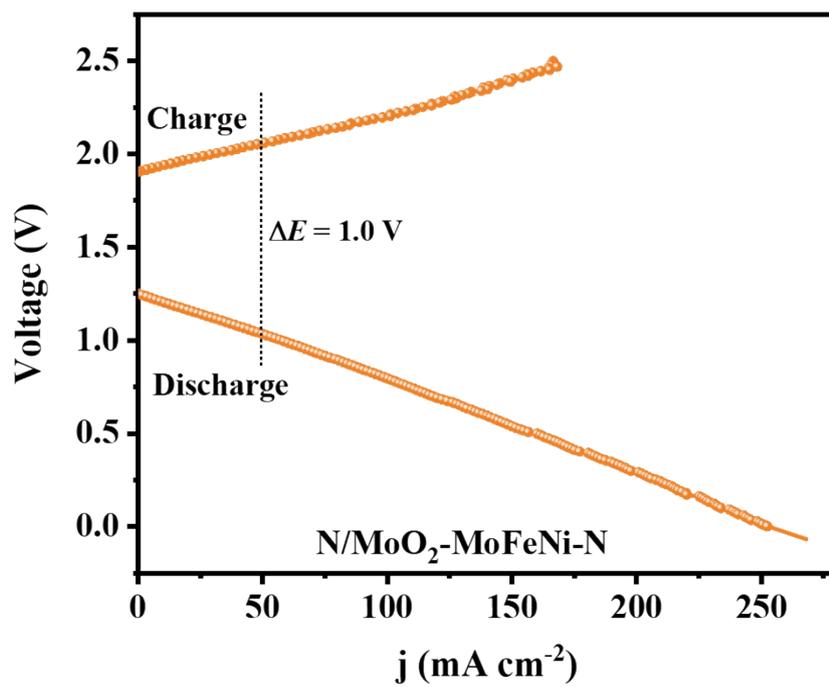
**Figure S14.** (a) LSV curves of N/MoO<sub>2</sub>-MoFeNi-N before and after seawater stability test in 6.0 M seawater KOH electrolyte over 60 °C.



**Figure S15.** Optimized structural models of (a)  $\text{MoO}_2$ , (b)  $\text{FeMoO}_4$ , (c)  $\text{NiMoO}_4$ , (d)  $\text{NiO}$ , (e)  $\text{FeN}$ , (f)  $\text{Mo}_2\text{N}$ , and (g)  $\text{Ni}_4\text{N}$ .



**Figure S16.** (a) CV curves recorded in 0.1 M KOH saturated with O<sub>2</sub> (solid lines) and N<sub>2</sub> (dotted lines) at a scan rate of 50 mV s<sup>-1</sup> and (b) ORR LSV curves at rotation speeds of 1600 rpm of the Pt/C electrocatalyst.



**Figure S17.** ORR charging and discharging I-V polarization curves of N/MoO<sub>2</sub>-MoFeNi-N.

**Table S5.** Comparison of the as-prepared N/MoO<sub>2</sub>-MoFeNi-N with reported ORR catalysts.

<b>Electrocatalysts</b>	<b>Peak power density (mW cm<sup>-2</sup>)</b>	<b>Cycling stability (h)</b>	<b>References</b>
<b>N/MoO<sub>2</sub>-MoFeNi-N</b>	<b>84.2 mW cm<sup>-2</sup></b>	<b>140 @10mA cm<sup>-2</sup></b>	<b>This work</b>
Cu <sub>0.81</sub> Ni <sub>0.19</sub> @CoN/Mo <sub>2</sub> N	81.72	150@10mA cm <sup>-2</sup>	Chem. Eng. J., 2026, 528, 172033
NS@Co <sub>3-x</sub> Ni <sub>x</sub> O <sub>4</sub> /Co <sub>3</sub> O <sub>4</sub>	-	200 @5mA cm <sup>-2</sup>	Energy Stor. Mater., 24 (2020) 272-280
MnCo <sub>2</sub> O <sub>4.5</sub> NPs	-	6 @10mA cm <sup>-2</sup>	J. Appl. Electrochem., 50 (2020) 907-915
Co <sub>3</sub> O <sub>4</sub> nanowire	35.7	333@10mA cm <sup>-2</sup>	Appl. Catal. B., 241 (2019) 104-112
GCCCO	81.6	20@5mA cm <sup>-2</sup>	Appl. Catal. B, 358 (2024) 124354
2%Cu-Co <sub>3</sub> O <sub>4</sub>	-	100@5mA cm <sup>-2</sup>	ACS Appl. Mater. Interfaces, 16 (2024) 17574
L8S2MCO/Vulcan	44.5	20@5mA cm <sup>-2</sup>	J. Power Sources, 632 (2025) 236364
VSM0.5CO	38.4	20@5mA cm <sup>-2</sup>	Appl. Surf. Sci. Adv., 27 (2025) 100725
Co <sub>9</sub> S <sub>8</sub> /Co <sub>1-x</sub> S/WS <sub>2</sub>	52.2	100@10mA cm <sup>-2</sup>	Electrochim. Acta, 512 (2025) 145467
5%Cu-Co <sub>3</sub> O <sub>4</sub>	84.4	140@5mA cm <sup>-2</sup>	ACS Appl. Mater. Interfaces, 16 (2024) 17574
10%Cu-Co <sub>3</sub> O <sub>4</sub>	-	80@5mA cm <sup>-2</sup>	ACS Appl. Mater. Interfaces, 16 (2024) 17574
CoNi <sub>2</sub> S <sub>4</sub> @CoNi-LDH	62.2	320@5mA cm <sup>-2</sup>	J. Mater. Chem. C, 11 (2023) 16384-16389