

# **Synergistic Ru-Mediated Activation of CoP/Co Heterostructures on Carbon Nanofibers toward Efficient Hydrogen Evolution**

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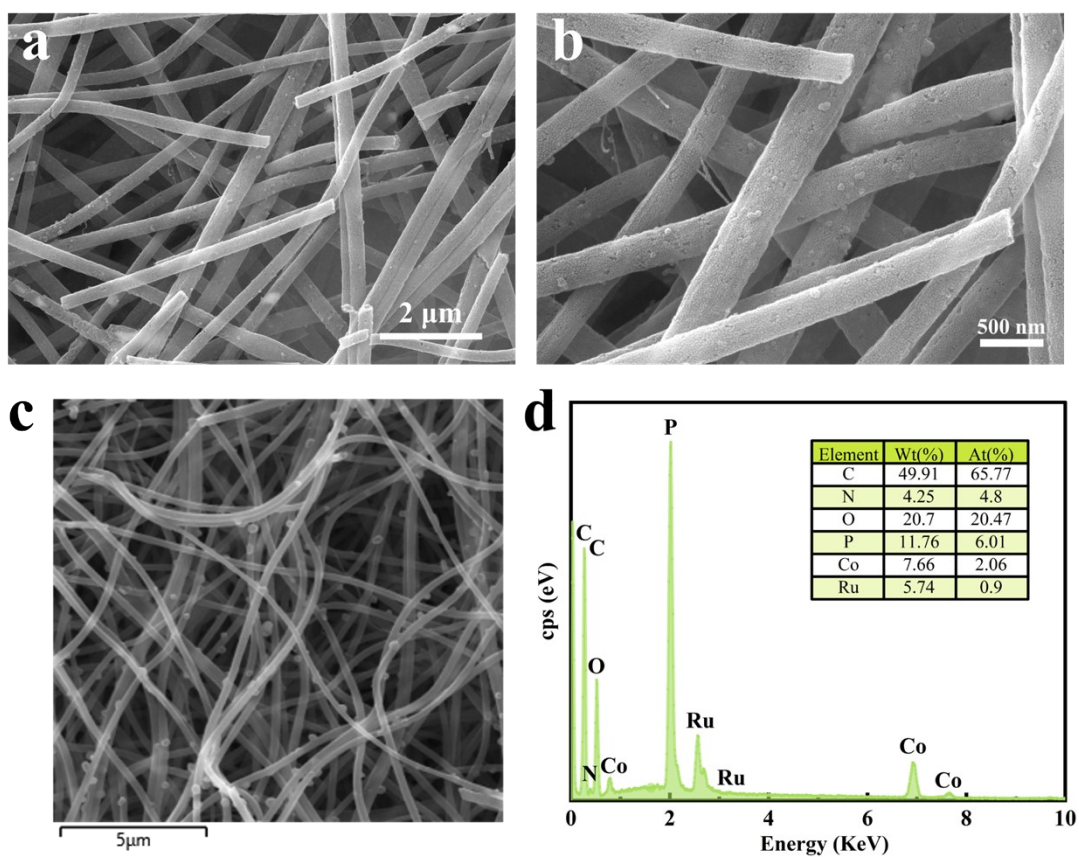


Fig. S1. (a-b) EM images of Co-NCNFs. (c) SEM image and (d) EDS of Ru-CoP/Co-NCNFs.

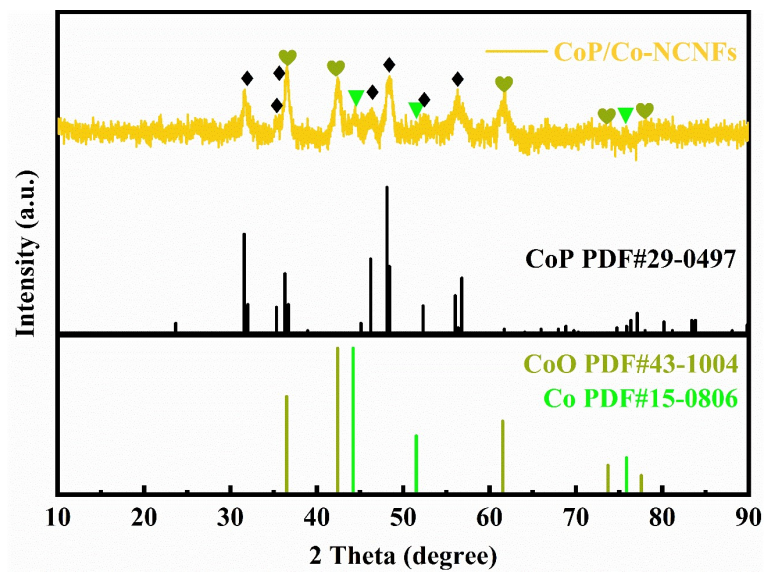


Fig. S2. XRD patterns of CoP/Co-NCNFs.

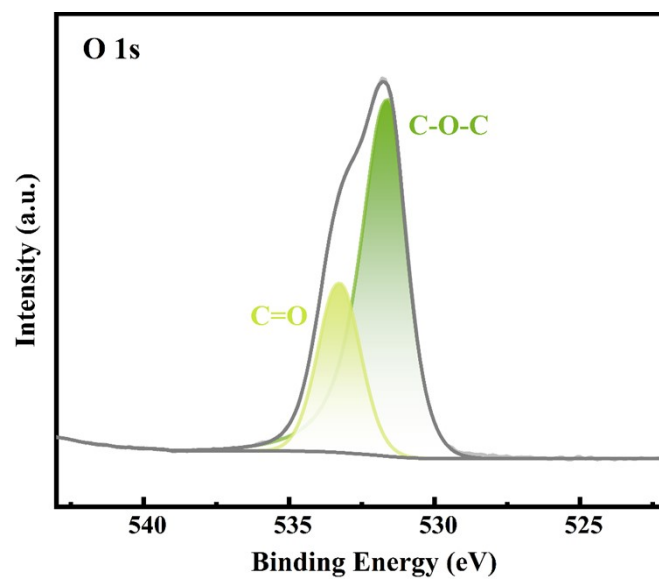


Fig. S3. XPS high-resolution O 1s spectra of the Ru-CoP/Co-NCNFs.

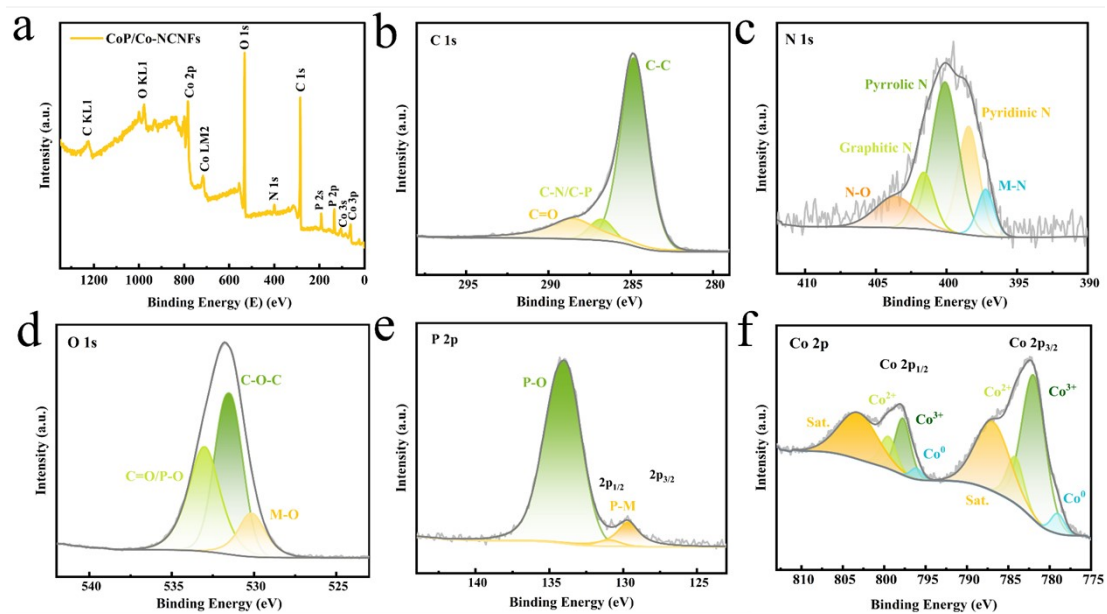


Fig. S4. (a) XPS survey spectrum and high-resolution (b) C 1s, (c) N 1s, (d) O 1s, (e) P 2p, and (f) Co 2p spectra of the CoP/Co-NCNFs.

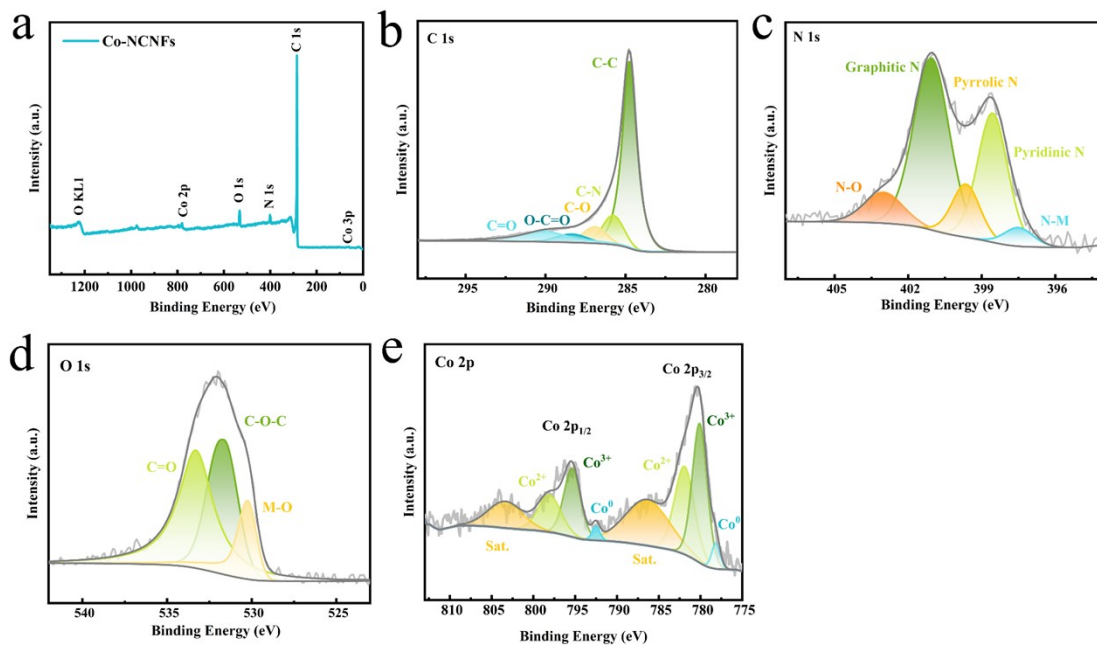


Fig. S5. (a) XPS survey spectrum and high-resolution (b) C 1s, (c) N 1s, (d) O 1s, and (e) Co 2p spectra of the Co-NCNFs.

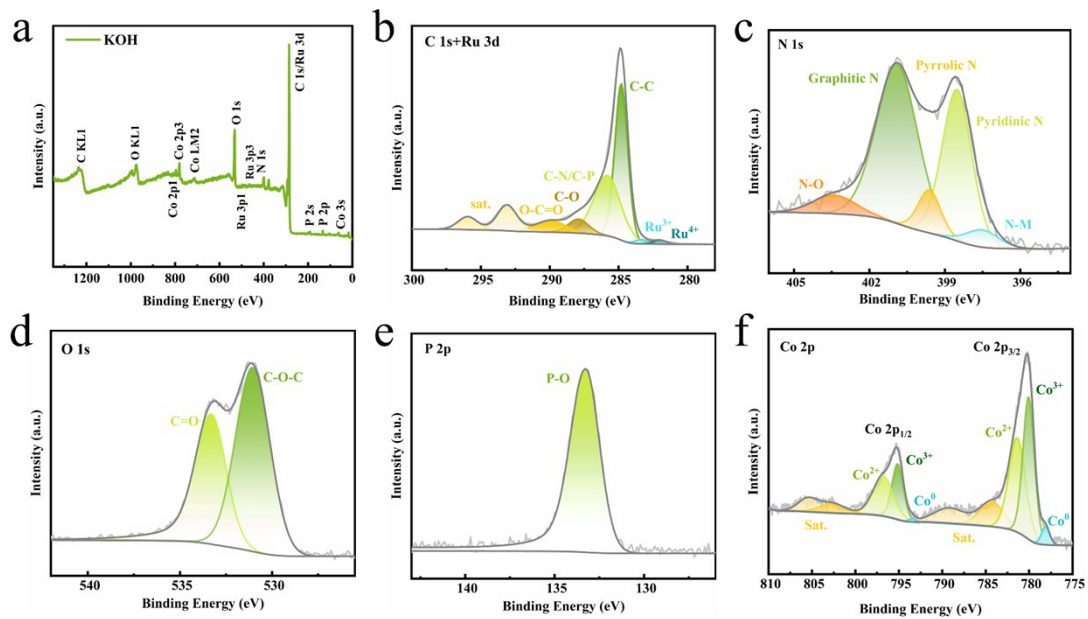


Fig. S6. (a) XPS survey spectrum and high-resolution (b) C 1s+Ru 3d, (c) N 1s, (d) O 1s, (e) P 2p, and (f) Co 2p spectra of the Ru-CoP/Co-NCNFs after chronoamperometric testing in 1.0 M KOH solution.

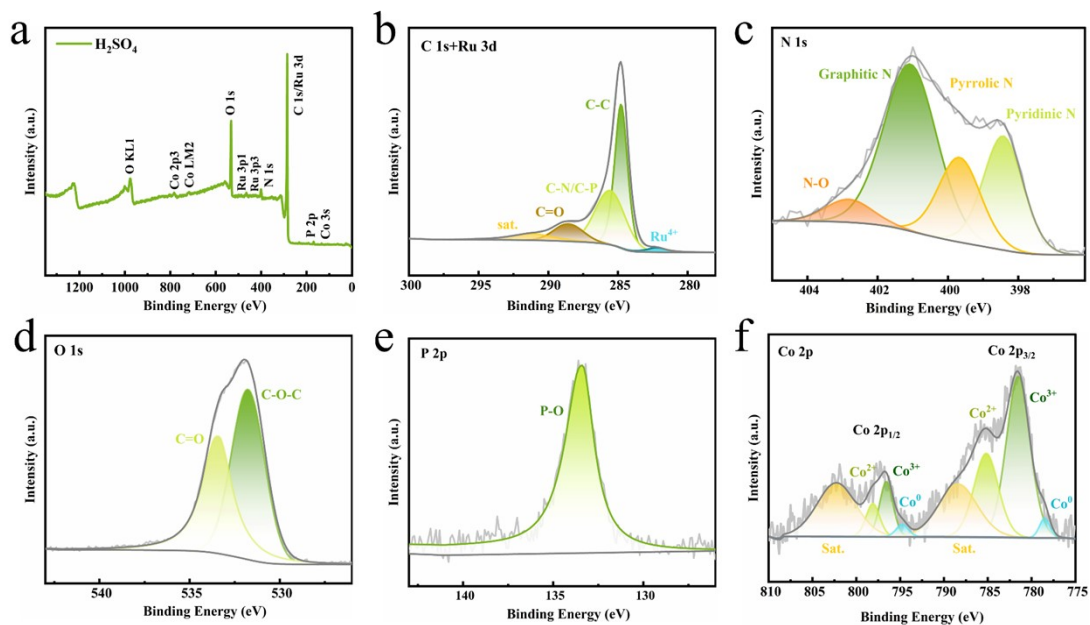


Fig. S7. (a) XPS survey spectrum and high-resolution (b) C 1s+Ru 3d, (c) N 1s, (d) O 1s, (e) P 2p, and (f) Co 2p spectra of the Ru-CoP/Co-NCNFs after chronoamperometric testing in 0.5 M  $\text{H}_2\text{SO}_4$  solution.

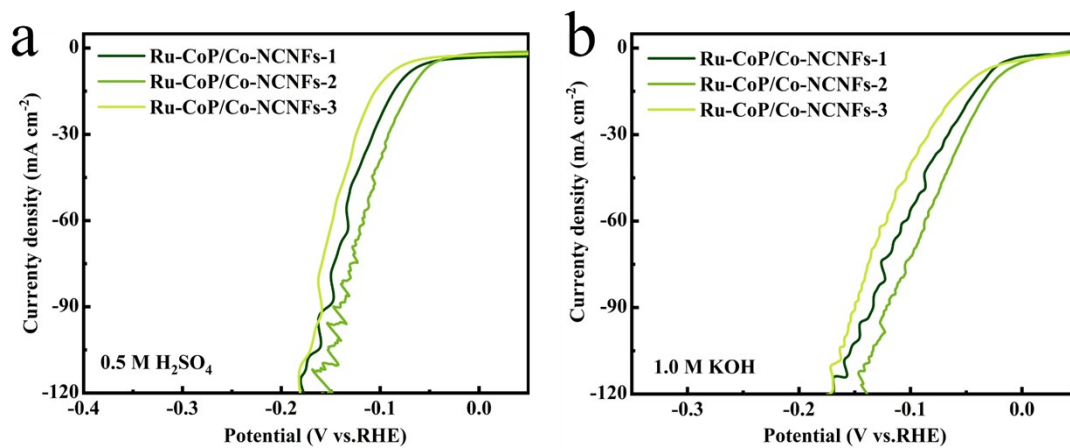


Fig. S8. Polarization curves for the different contents of Ru (a) in 0.5 M H<sub>2</sub>SO<sub>4</sub> and (b) 1.0 M KOH solution.

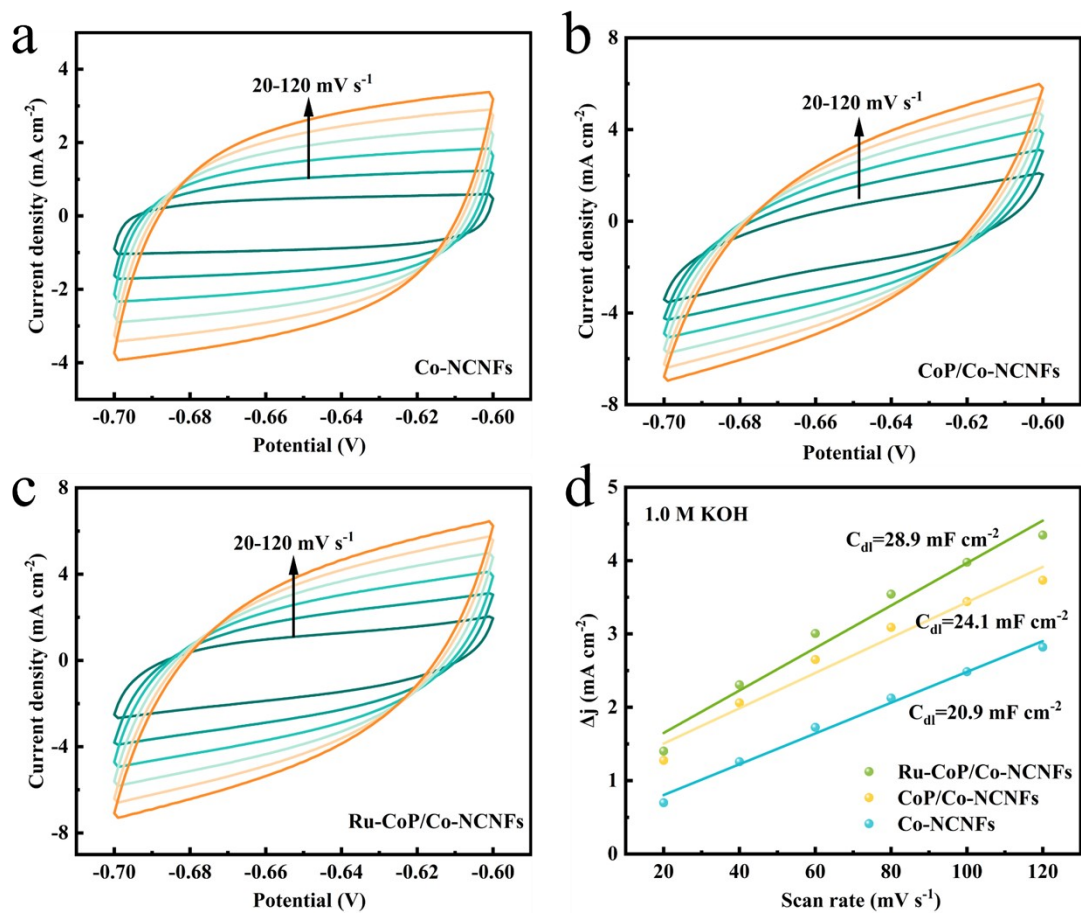


Fig. S9. (a-c) CV curves with 20 to 120  $\text{mV s}^{-1}$  scan rate and (d)  $C_{dl}$  for different samples in 1.0 M KOH solution.

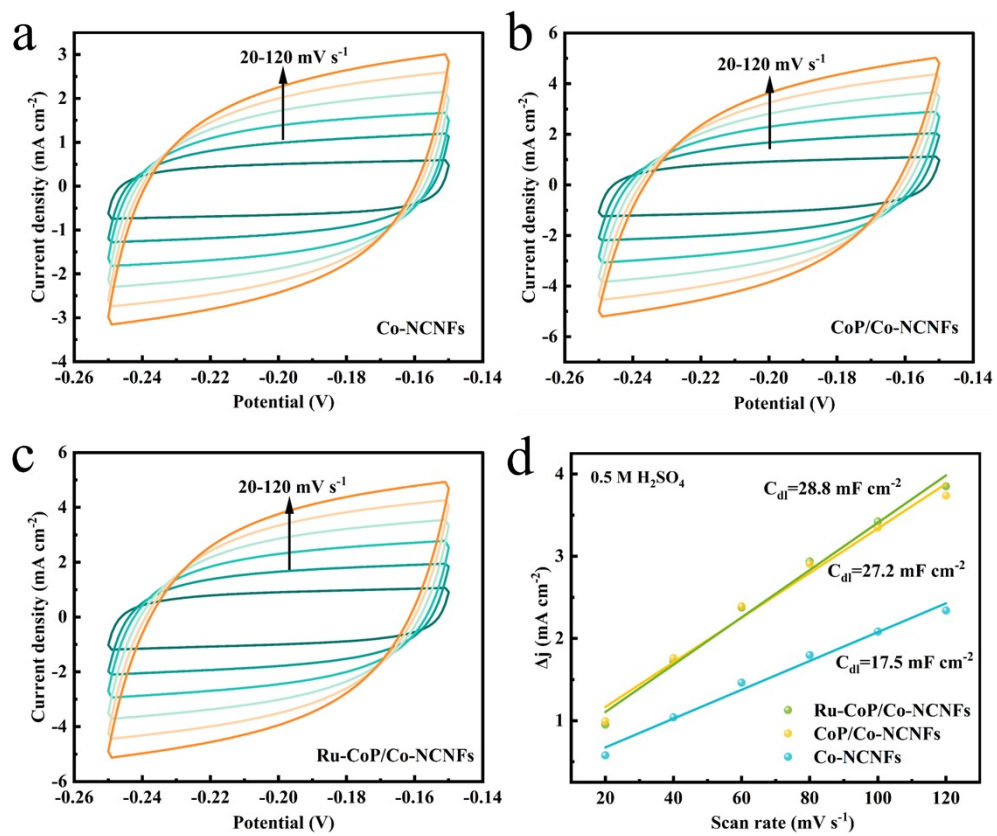


Fig. S10. (a-c) CV curves with 20 to 120 mV s<sup>-1</sup> scan rate and (d) C<sub>dl</sub> for different samples in 0.5 M H<sub>2</sub>SO<sub>4</sub> solution.

The turnover frequency (TOF) for HER was calculated by the following equation:

$$TOF = \frac{I}{Q} = \frac{|j|S}{2nF}$$

Where  $I$  (A) is the current,  $Q$  (C) is voltammetric charges,  $|j|$  ( $A \cdot cm^{-2}$ ) is the current density,  $S$  is the geometric area of the working electrode ( $1.0 \text{ cm}^2$ ),  $F$  is the Faraday constant ( $96485 \text{ C} \cdot \text{mol}^{-1}$ ), and  $n$  is the number of active sites (mol). 2 is the electron transfer numbers of HER. The number of active sites ( $n$ ) was calculated by the method previously reported. CV was used to scan a voltage range of 0-0.4V (vs. RHE) at a rate of  $50 \text{ mV s}^{-1}$  in a 1.0 M PBS solution.

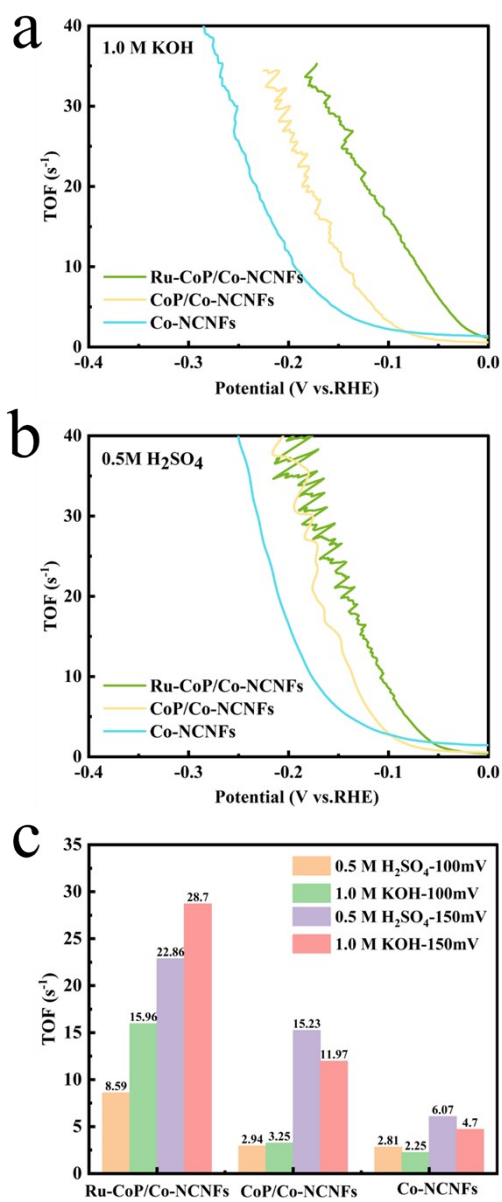


Fig. S11. TOF curves of the different samples in (a) 1.0 M KOH, (b) 0.5 M H<sub>2</sub>SO<sub>4</sub> solutions, and (c) comparison of values at different potentials.

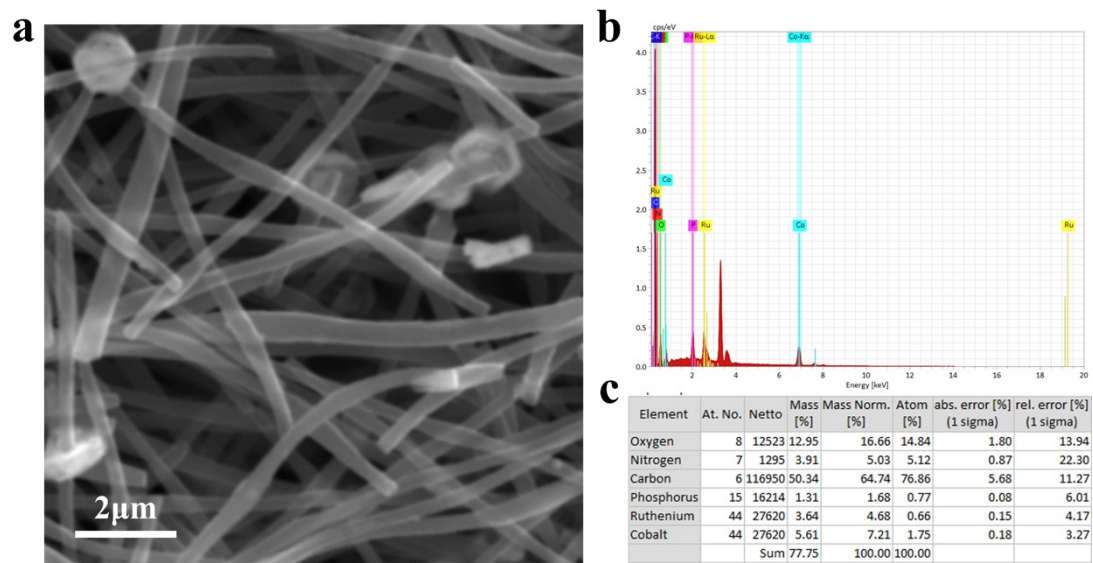


Fig. S12. (a) SEM image, (b) EDS, and (c) element content table of Ru-CoP/Co-NCNFs after stability test in 1.0 M KOH.

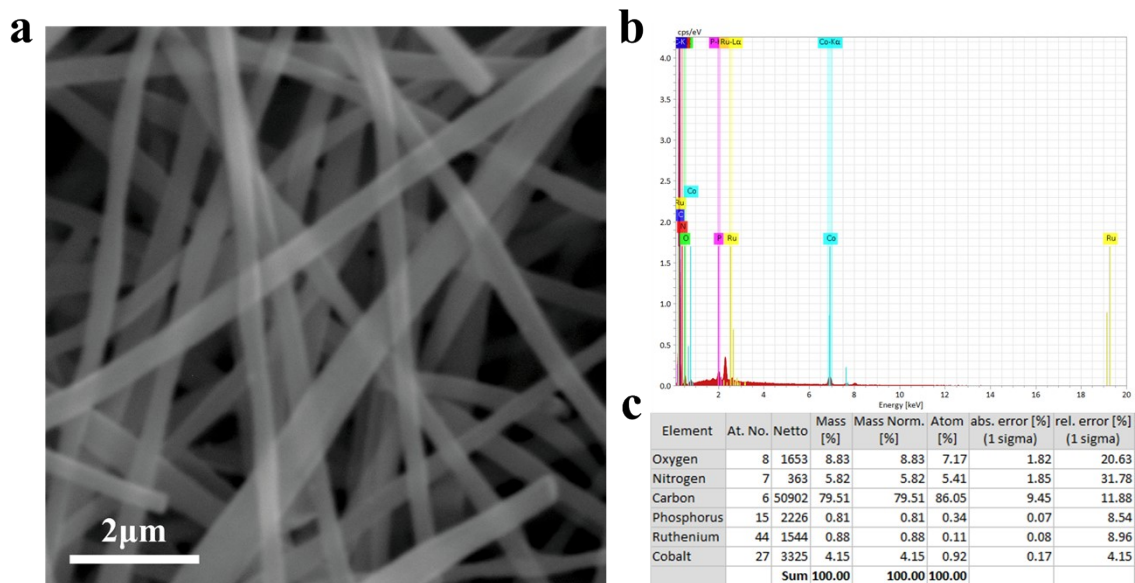


Fig. S13. (a) SEM image, (b) EDS, and (c) element content table of Ru-CoP/Co-NCNFs after stability test in 0.5 M H<sub>2</sub>SO<sub>4</sub>.

Table S1. Elemental content in different samples and after stability testing by XPS.

Catalysts	C (At %)	O (At %)	N (At %)	P (At %)	Co (At %)	Ru (At %)
Ru-CoP/Co-NCNFs	36.99	34.78	3.22	12.52	11.19	2.96
CoP/Co-NCNFs	43.28	25.53	2.11	6.09	25.99	-
Co-NCNFs	88.22	6.85	3.08	-	1.85	-
Ru-CoP/Co-NCNFs- H <sub>2</sub> SO <sub>4</sub>	71.82	17.47	4.08	1.32	3.54	1.79
Ru-CoP/Co-NCNFs- KOH	71.52	17.41	4.06	1.32	3.51	1.78

Table S2. The value of  $j_0$  of the different samples in 0.5 M  $\text{H}_2\text{SO}_4$  and 1.0 M KOH solutions.

Catalysts	$j_0$ (mA cm <sup>-2</sup> )	$j_0$ (mA cm <sup>-2</sup> )
	0.5 M $\text{H}_2\text{SO}_4$	1.0 M KOH
Ru-CoP/Co-NCNFs	6.52	2.48
CoP/Co-NCNFs	4.99	1.26
Co-NCNFs	0.96	0.53
RuCoP/CP	1.16	0.79
RuCoP/CC	1.14	0.70
20% Pt/C	6.80	4.46

Table S3. EIS parameters of the different samples in 0.5 M H<sub>2</sub>SO<sub>4</sub> and 1.0 M KOH solutions.

Catalysts	R <sub>s</sub>	R <sub>ct</sub>	R <sub>s</sub>	R <sub>ct</sub>
	(Ω)	(Ω)	(Ω)	(Ω)
	0.5 M H <sub>2</sub> SO <sub>4</sub>		1.0 M KOH	
Ru-CoP/Co-NCNFs	5.58	6.28	4.09	12.17
CoP/Co-NCNFs	5.76	6.70	5.56	17.51
Co-NCNFs	6.08	9.36	7.88	23.55

Table S4.  $C_{dl}$  and ECSA values of the different samples in 0.5 M  $H_2SO_4$  and 1.0 M KOH solutions.

Catalysts	$C_{dl}$	ECSA	$C_{dl}$	ECSA
	( $mF\ cm^{-2}$ )	( $cm^{-2}\ mg^{-1}$ )	( $mF\ cm^{-2}$ )	( $cm^{-2}\ mg^{-1}$ )
	0.5 M $H_2SO_4$		1.0 M KOH	
Ru-CoP/Co-NCNFs	28.8	720.0	28.9	722.5
CoP/Co-NCNFs	27.2	680.0	24.1	602.5
Co-NCNFs	17.5	437.5	20.9	522.5

Table S5. Comparison of the catalytic performance of reported HER catalysts in 1.0 M KOH solution.

Catalysts	$\eta_{10}$ (mV)	Tafel (mV dec <sup>-1</sup> )	Stability	Ref.
Ru-CoP/Co-NCNFs	21	59.5	retaining 90% after 200 h 10,000 cycles	This work
Ru-CoP <sub>x</sub> /NF	41	68	12h	1
Ru-CoP/CC	44	64	60 h	2
Ru@CoP/CC	48	103	27.8h	3
CoP@NC-Ru	35	72	5000 cycles	4
Ru/B-CoP	52	63.6	20 h	5
Ru-CoP-2.5 NAs	52	70.1	50 h	6
Ru-CoP/CC-2	45	98.11	20 h	7
Ru, Ni-CoP	45	53.9	40 h	8
Ru-CoP/CDs	51	73.4	2000 cycles	9
MnRuPOGO-500	27	57.35	48 h	10
Ru-MoS <sub>2</sub> -Mo <sub>2</sub> C/TiN	25	58	30 h	11

Table S6. Comparison of the catalytic performance of reported HER catalysts in 0.5 M H<sub>2</sub>SO<sub>4</sub> solution.

Catalysts	$\eta_{10}$ (mV)	Tafel (mV dec <sup>-1</sup> )	Stability	Ref.
Ru-CoP/Co-NCNFs	60	50.6	retaining 90% after 200h 6000 cycles	This work
Ru-CoPx/NF	73	59	12 h	1
Ru@CoP/CC	66	84	27.8h	3
Ru-CoP/NC	98	56	-	12
MnRuPOGO-500	109	38.55	60 h	10
CNT-V-Fe-Ru	64	51	20 h	13
RuMo/NCN	82	54	12 h	14
NiZn@C-CoP	89	74	24 h	15
Br-Ru-RuO <sub>2</sub> /MCC	77	49	12 h	16
NF@CoFeP	82	60	50 h	17
CoSe <sub>2</sub> /a-CoP	65	54	50 h	18

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