

## Supplementary Information

# Dual-Function Pt/Ir-Doped CdS Catalysts: Defect Engineering for Visible-Light HMF Photo Oxidation and Hydrogen Evolution Reaction

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**Table S1.** Binding energies obtained from XPS analysis for the four CdS NPs.

Samples	Binding energy (eV)		
	Cd <i>3d</i> CdS/CdS <sub>1-x</sub>	S <i>2p</i> CdS/MS	M
CdS	404.8/0.0	161.1/0.0	
Li <sub>dp</sub> -CdS	404.8/0.0	161.1/0.0	Li 1s: 55.6
Ir <sub>dp</sub> -CdS	405.0/404.1	160.85/160.35	Ir 4f <sub>7/2</sub> : 64.95
Pt <sub>dp</sub> -CdS	404.95/404.05	160.0/160.30	Pt 4f <sub>7/2</sub> : 72.1

**Table S2.** Bandgap,  $E_F$  energy, and work function obtained from DRS and XPS analysis for the four CdS NPs.

Samples	Bandgap (eV)	$E_F$ (eV)	Work function (eV)
CdS	2.30	1.75	4.45
Li <sub>dp</sub> -CdS	2.34	1.72	4.30
Ir <sub>dp</sub> -CdS	2.24	1.65	4.60
Pt <sub>dp</sub> -CdS	2.21	1.60	4.70

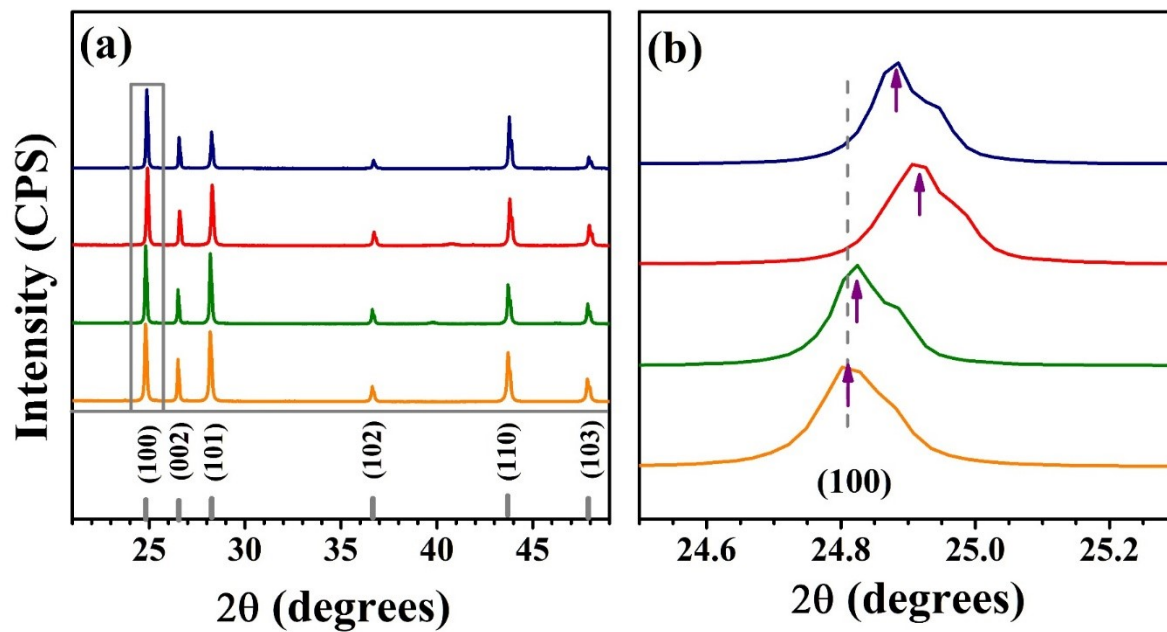
**Table S3.** Comparison of Ir<sub>dp</sub>-CdS with representative recently reported photocatalytic systems for HMF oxidation in aqueous media.

Catalyst	Light Source & Power	Base (additive) / Reaction time / HMF Conc.	Target Product	Performance (HMF Conv. / Sel. or Yield)	Ref.
Au/ZnO <sub>D</sub> <sub>P-H</sub>	300W Xe lamp ( $\lambda > 420$ nm)	Na <sub>2</sub> CO <sub>3</sub> / 5 h / HMF conc.: 31.7 mM	FDCA	HMF Conv.: 55.6% FDCA Yield: 53.9% Sel.: 96.9%	[1]
UCNT/Pt (3 wt.%)	300W Xe lamp ( $\lambda \geq 420$ nm)	Base free / 5 h / HMF conc.: 10 mM	DFF	DFF Sel.: 95%	[2]
$\alpha$ -Fe <sub>2</sub> O <sub>3</sub> /Zn <sub>0.5</sub> Cd <sub>0.5</sub> S	12W LED (413 nm)	(Air or Ar) / 24–32 h HMF conc.: 100 mM	DFF FDCA	HMF Conv.: 75%, DFF Sel. (24 h, under Ar): 99% HMF Conv.: 99%, FDCA Sel. (32 h, air) : 85%	[3]
CdS@M Xene	LED ( $\lambda > 420$ nm)	Base free / 6 h / HMF conc.: 1 mM	DFF	HMF Conv.: 63.89% DFF Sel: 91.63% DFF Yield 58.49%	[4]
CdS/WO <sub>3</sub>	6W LED (445 nm)	Base free / 24 h HMF conc.: 25 mM	FDCA	HMF Conv: >97% FDCA Yield: 61% (24h)	[5]
Ir <sub>dp</sub> -CdS	6W LED (445 nm)	Base free / 24 h HMF conc.: 25 mM	FDCA	HMF Conv: >95% FDCA Yield: 59% (24h)	This Work

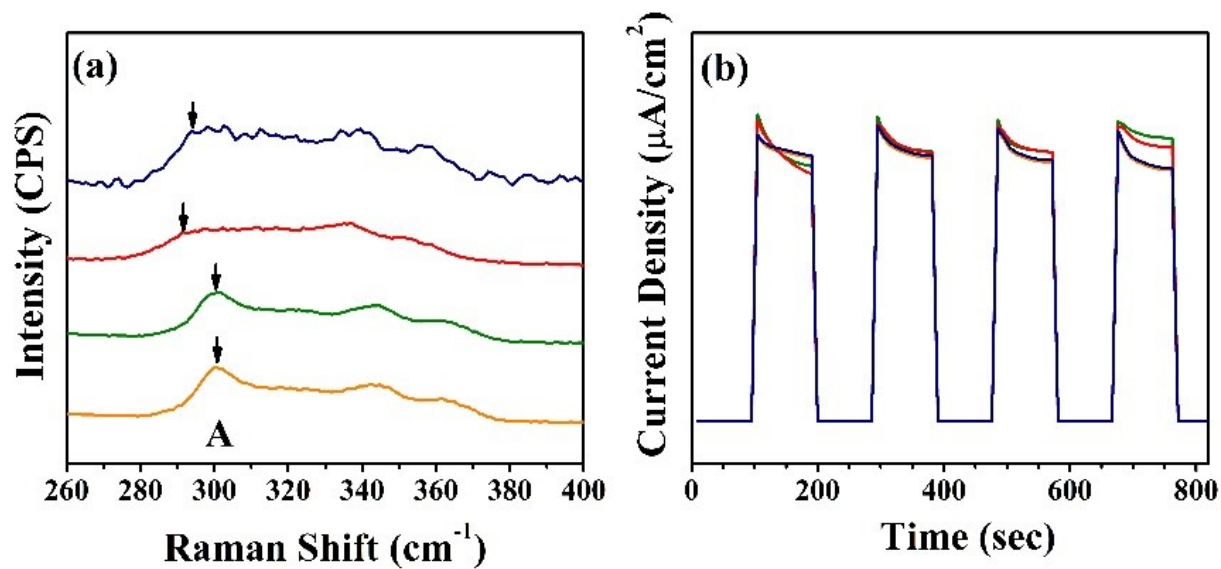
**Table S4.** Comparison of the electrocatalytic activity with other previously reported CdS-based catalysts for HER.

Catalysts	Tafel slope (mV·dec <sup>-1</sup> )	Overpotential (mV)	Solution	Ref.
CdS <sub>1-x</sub>	131.6	401 ( $\eta_{10}$ )	1 M KOH	[6]
20 wt % CdS <sub>1-x</sub> in WO <sub>3-x</sub>	61.9	191 ( $\eta_{10}$ )	1 M KOH	[6]
ZnS/CdS	313.6	299 ( $\eta_{10}$ )	0.5 M H <sub>2</sub> SO <sub>4</sub>	[7]
VS <sub>2</sub> /ZnS/CdS	74.4	86 ( $\eta_{10}$ )	0.5 M H <sub>2</sub> SO <sub>4</sub>	[7]
CdS	151	334 ( $\eta_{10}$ )	1 M KOH	[8]
CdS-V <sub>2</sub> O <sub>5</sub>	104	238 ( $\eta_{10}$ )	1 M KOH	[8]
CdS-V <sub>2</sub> O <sub>5</sub> / g-C <sub>3</sub> N <sub>4</sub>	63	202 ( $\eta_{10}$ )	1 M KOH	[8]
CdS	111	225 ( $\eta_{10}$ )	1 M KOH	[9]
CdS/MIL- 101(Fe) (3:1)	47	108 ( $\eta_{10}$ )	1 M KOH	[9]
Ir <sub>dp</sub> -CdS	57.5	198.3 ( $\eta_1$ )	1 M KOH	This Work
Pt <sub>dp</sub> -CdS	128.8	575.8 ( $\eta_1$ )	1 M KOH	This Work

**Fig. S1** (a) Wide-angle XRD patterns and (b) Corresponding enlarged XRD view of the selected region for CdS (orange),  $\text{Li}_{\text{dp}}\text{-CdS}$  (green),  $\text{Ir}_{\text{dp}}\text{-CdS}$  (red), and  $\text{Pt}_{\text{dp}}\text{-CdS}$  NPs (blue).



**Fig. S2** (a) Zoom-in Raman spectra, (b) Photocurrents generated by CdS (black),  $\text{Li}_{\text{dp}}$ -CdS (green),  $\text{Ir}_{\text{dp}}$ -CdS (red), and  $\text{Pt}_{\text{dp}}$ -CdS (blue) NPs in the presence of  $\text{Na}_2\text{SO}_3$  (0.025 M) as a hole scavenging agent.



**Fig. S3** (a) UV-Vis DRS spectra, (b) Work function measurements, and (c) Valence band spectra of CdS (orange),  $\text{Li}_{\text{dp}}$ -CdS (green),  $\text{Ir}_{\text{dp}}$ -CdS (red), and  $\text{Pt}_{\text{dp}}$ -CdS (blue) NPs.

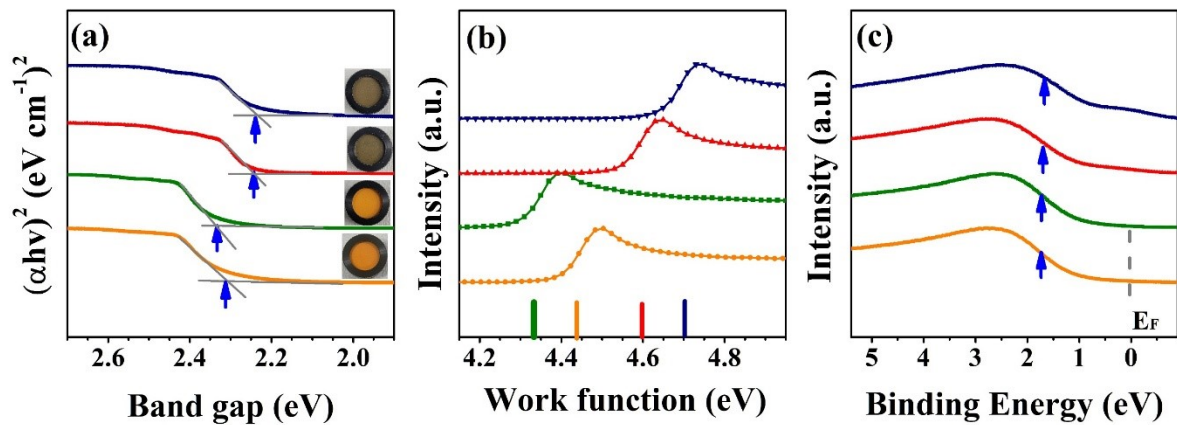
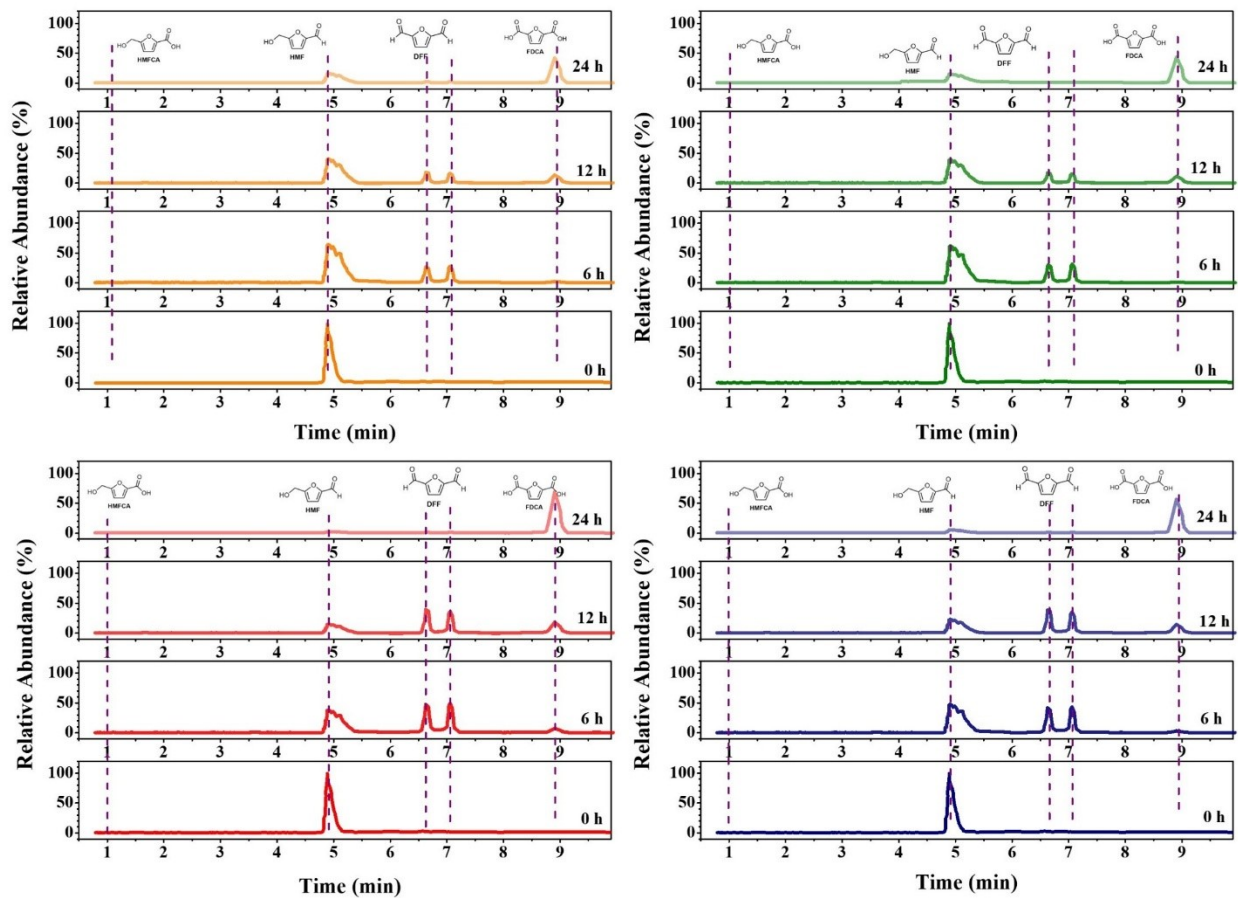
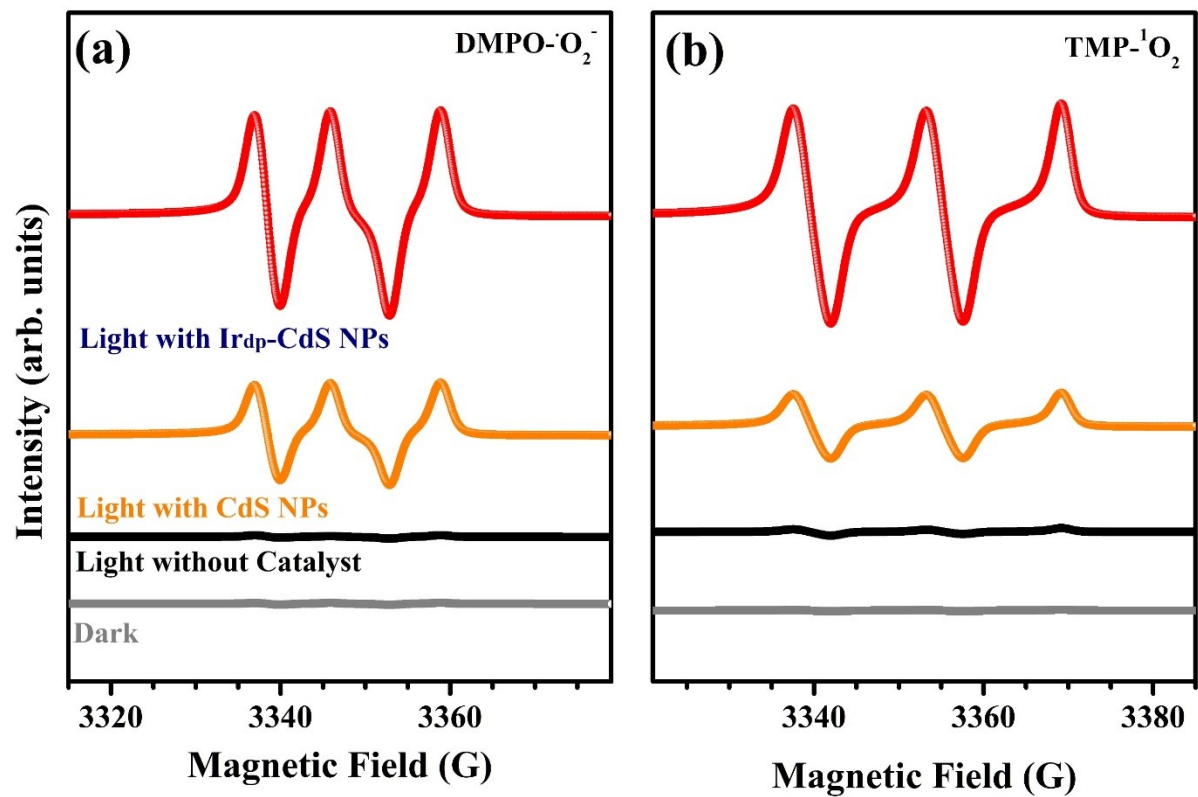


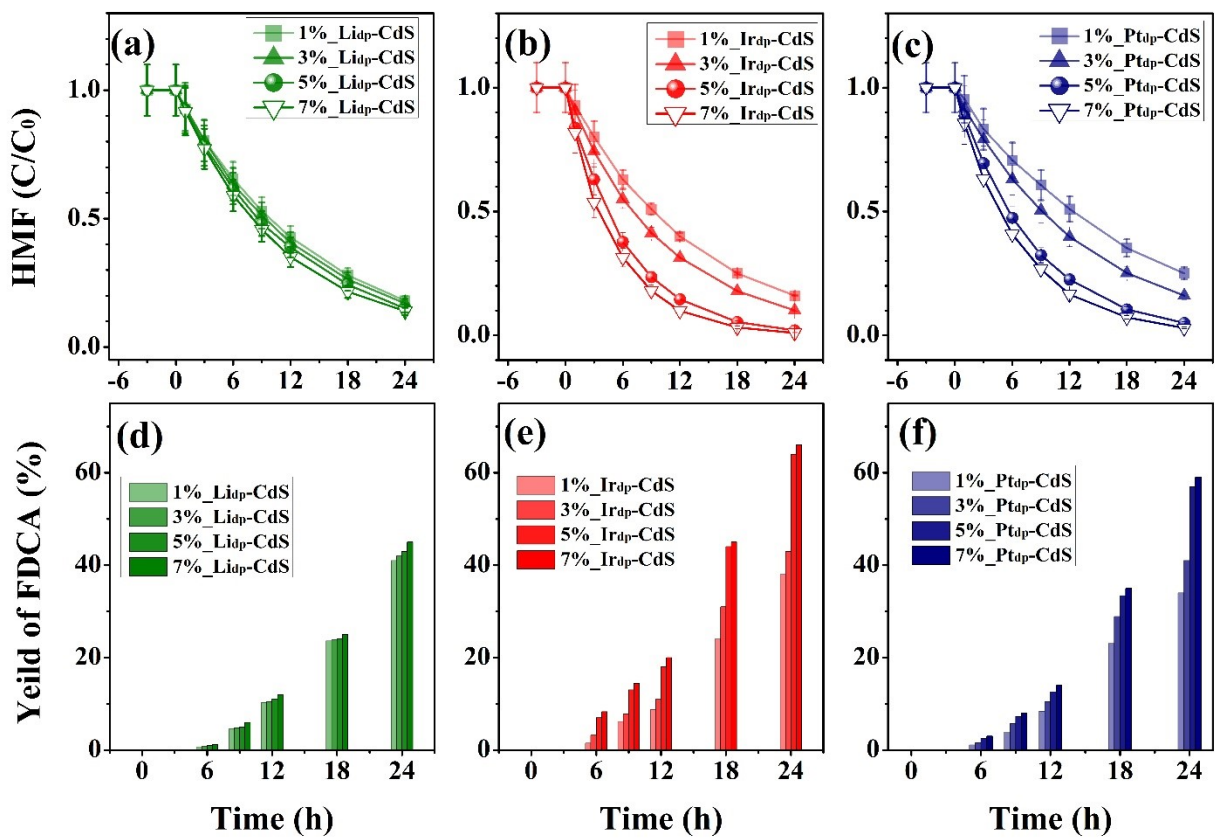
Fig. S4 LC-MS spectra of the four CdS NPs.



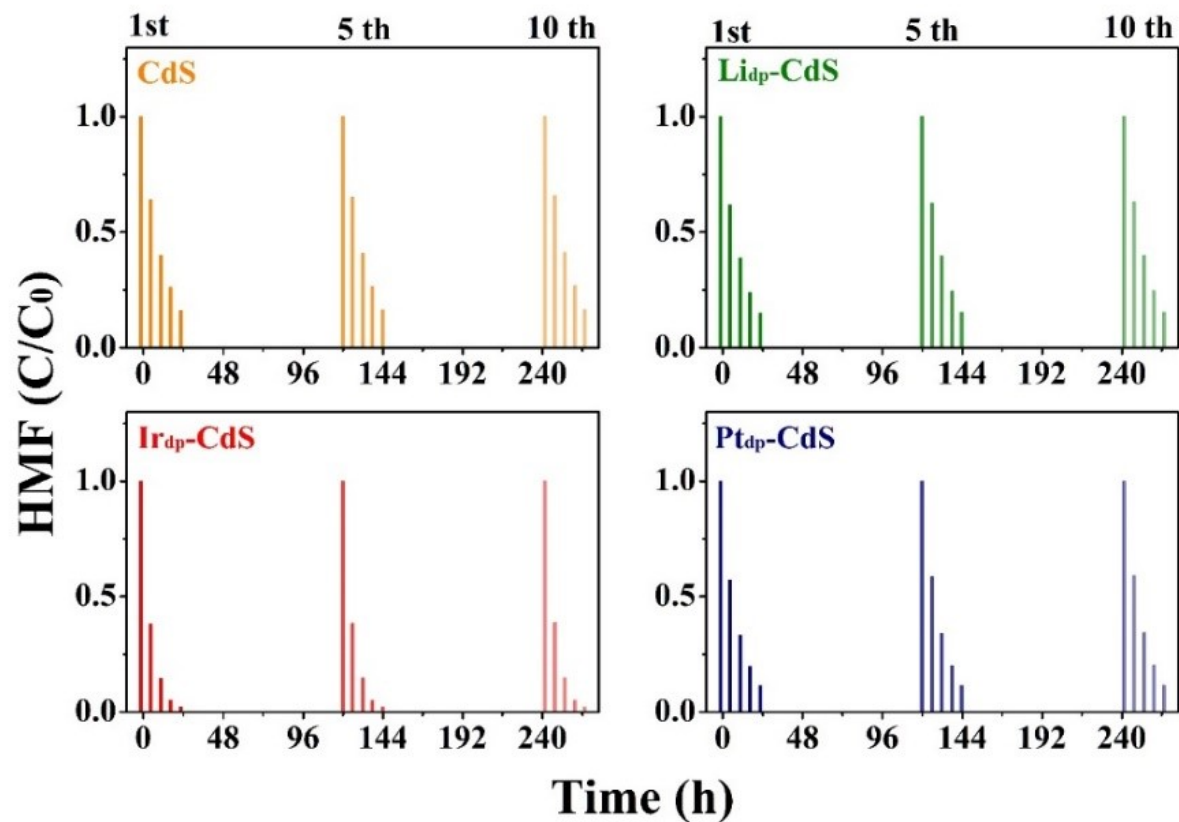
**Fig. S5** EPR spectra of CdS and Ir<sub>dp</sub>-CdS NPs using (a) DMPO and (b) TMP as  $\cdot\text{O}_2^-$ -trapping agent and  $^1\text{O}_2$ -trapping agents.



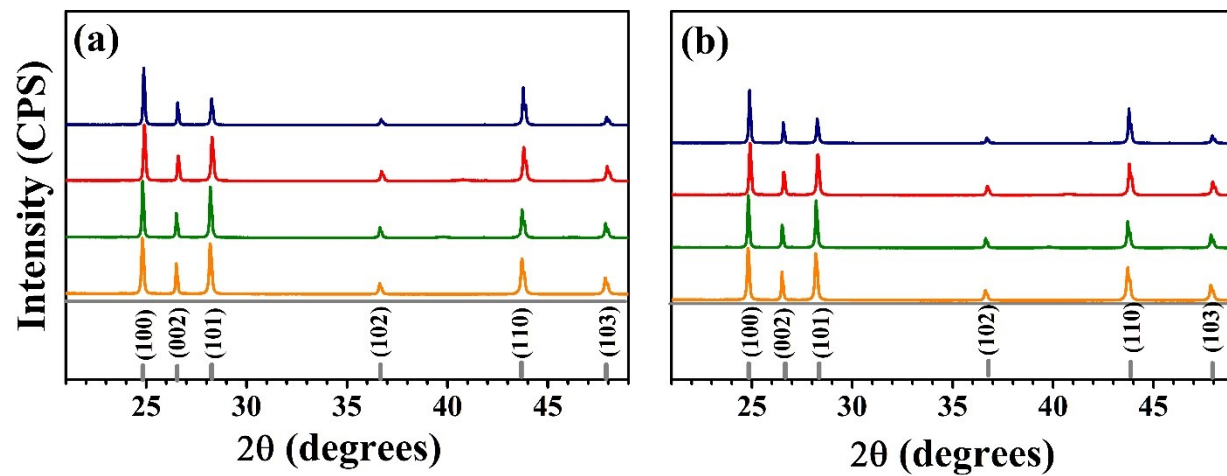
**Fig. S6** Conversion rate of HMF and FDCA yields in the presence of 1, 3, 5, and 7 wt.% doping of Li (a and d), Ir (b and e), and Pt (c and f) ions.



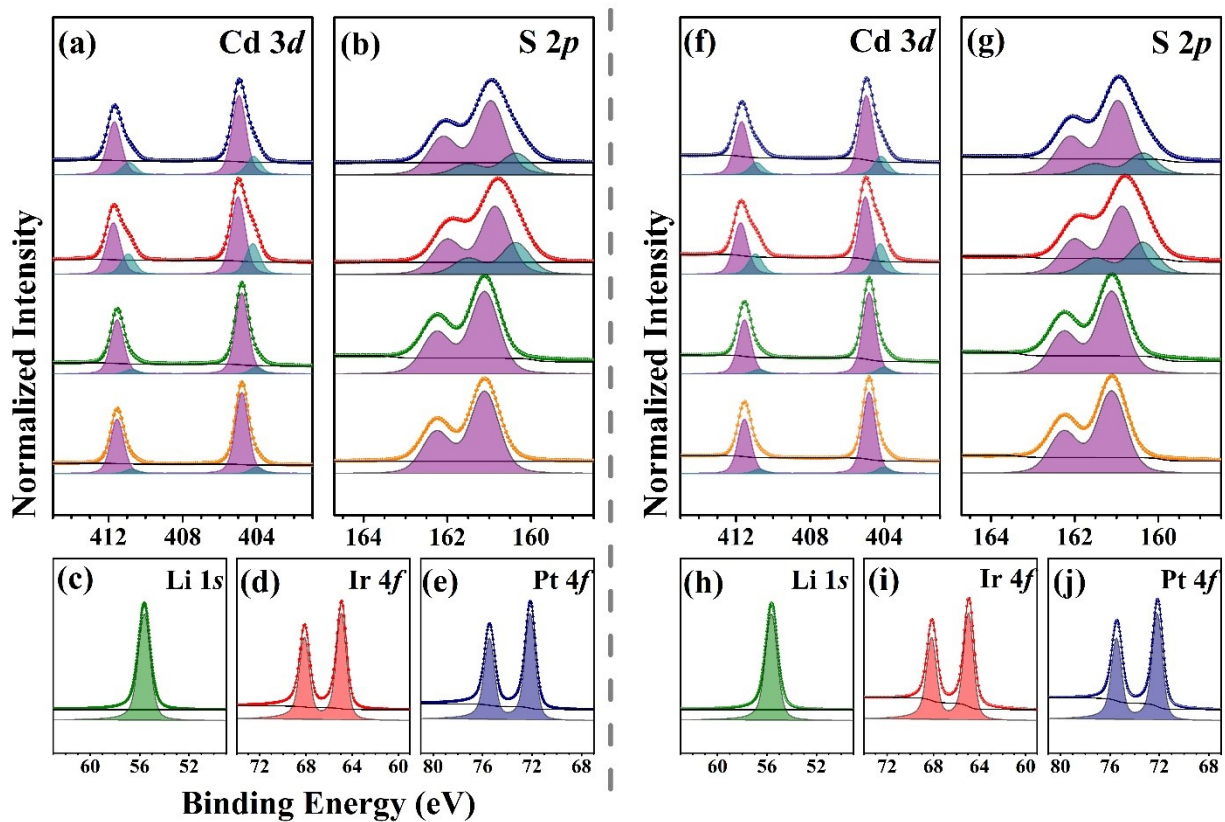
**Fig. S7** Cyclic photocatalytic HMF-decomposition data for CdS (orange),  $\text{Li}_{\text{dp}}\text{-CdS}$  (green),  $\text{Ir}_{\text{dp}}\text{-CdS}$  (red), and  $\text{Pt}_{\text{dp}}\text{-CdS}$  (blue) NPs at 445 nm.



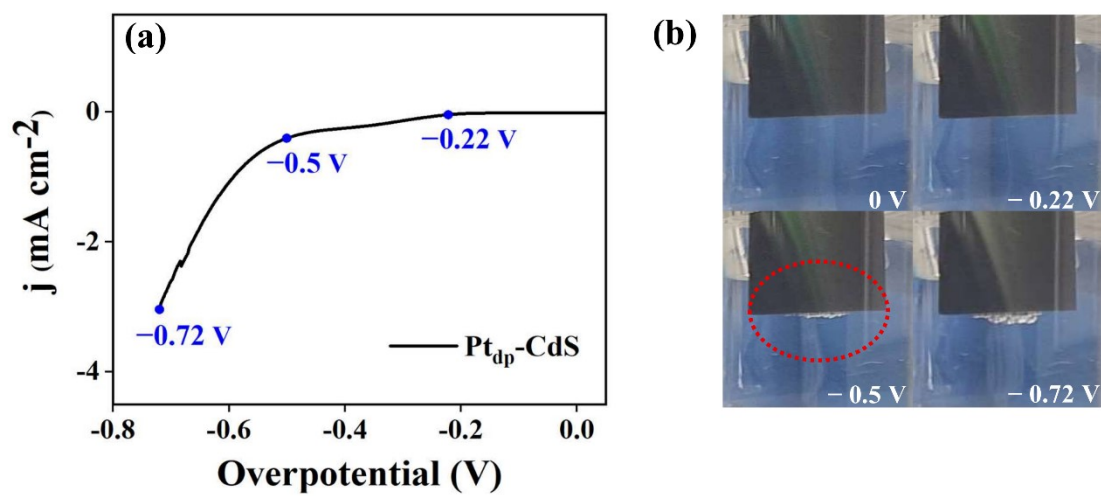
**Fig. S8** XRD patterns (a) before and (b) after PCR experiments for CdS (orange), Li<sub>dp</sub>-CdS (green), Ir<sub>dp</sub>-CdS (red), and Pt<sub>dp</sub>-CdS (blue) NPs.



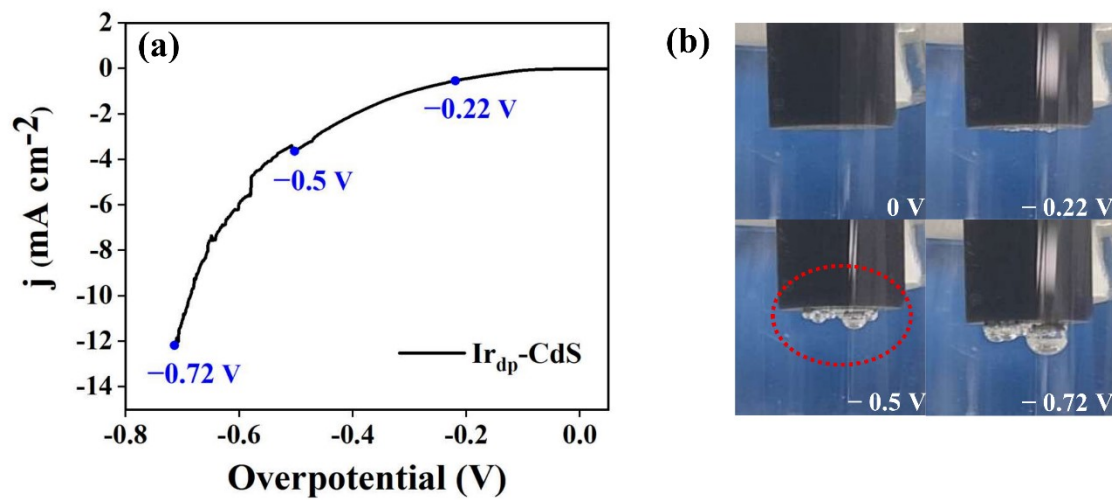
**Fig. S9** XPS spectra of (a–e) before and (f–j) after PCR experiments for CdS (orange), Li<sub>dp</sub>-CdS (green), Ir<sub>dp</sub>-CdS (red), and Pt<sub>dp</sub>-CdS (blue) NPs.



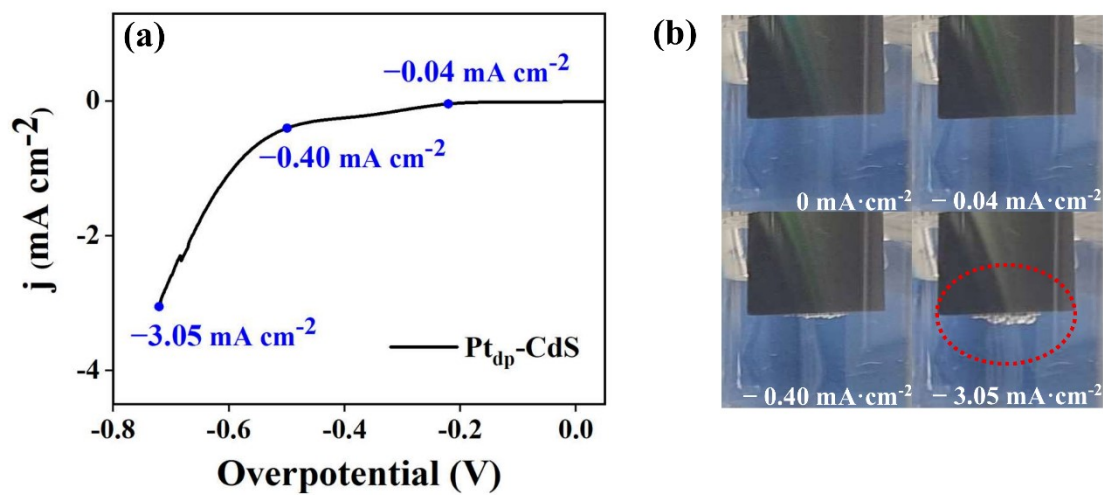
**Fig. S10** (a) LSV curve of  $\text{Pt}_{\text{dp}}\text{-CdS}$  NPs in 1 M KOH. Marked points indicate the overpotential at which bubble behavior was observed. (b) Corresponding optical images showing bubble dynamics at selected overpotentials of 0,  $-0.22$ ,  $-0.5$ , and  $-0.72$  V.



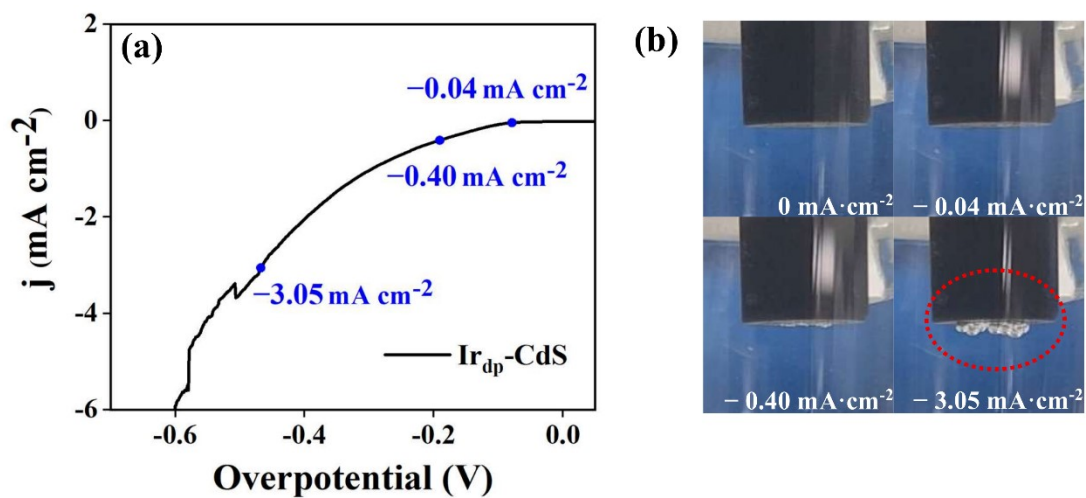
**Fig. S11** (a) LSV curve of Ir<sub>dp</sub>-CdS NPs in 1 M KOH. Marked points indicate the overpotential at which bubble behavior was observed. (b) Corresponding optical images showing bubble dynamics at selected overpotentials of 0, -0.22, -0.5, and -0.72 V.



**Fig. S12** (a) LSV curve of Pt<sub>dp</sub>-CdS NPs in 1 M KOH. Marked points indicate the current density at which bubble behavior was observed. (b) Corresponding optical images showing bubble dynamics at selected current densities of 0, -0.04, -0.40, and -3.05 mA·cm<sup>-2</sup>.



**Fig. S13** (a) LSV curve of Ir<sub>dp</sub>-CdS in 1 M KOH. Marked points indicate the current density at which bubble behavior was observed. (b) Corresponding optical images showing bubble dynamics at selected current densities of 0, -0.04, -0.40, and -3.05 mA·cm<sup>-2</sup>.



## References

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