

Supporting Information

Magnetic-Field-Assisted Cascade Electrocatalysis: Simultaneous H₂O₂ Generation and ·OH Radical Production for Coupled Hydrogen Evolution and Water Disinfection

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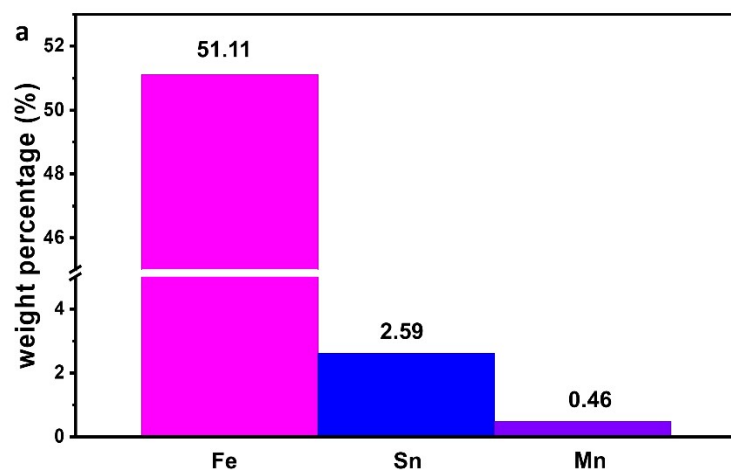


Fig. S1 The atomic weight percentage of (a) MTOFO.

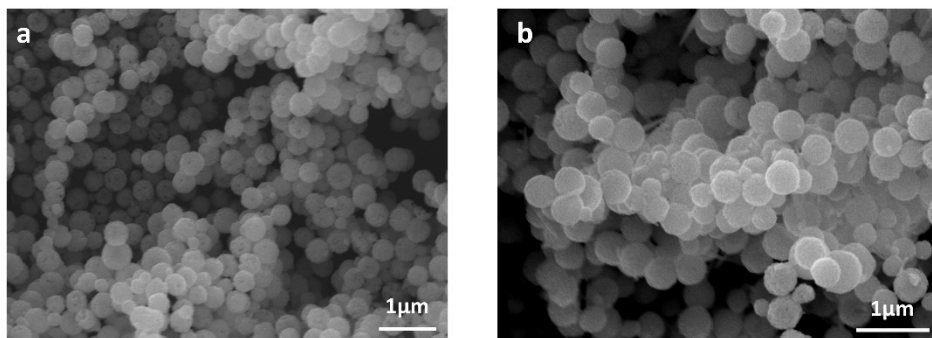


Fig. S2 SEM images of (a) Fe₃O₄ and (b) TOFO.

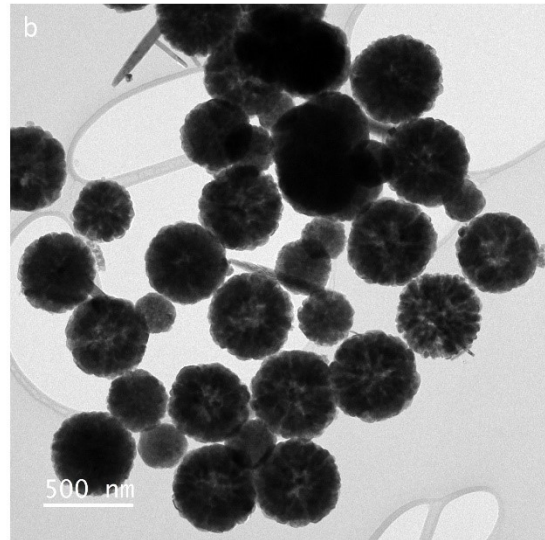
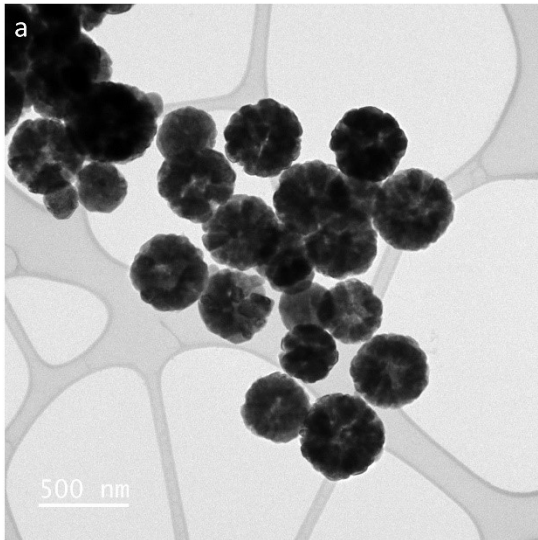


Fig. S3 Low-magnification TEM images of (a) TOFO and (b) MTOFO.

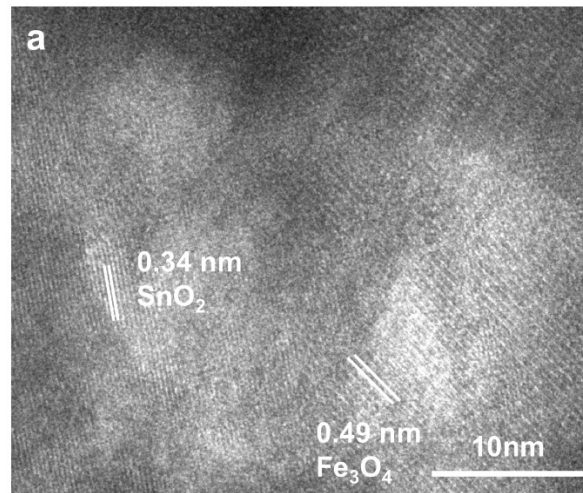


Fig. S4 (a) HR-TEM image of TOFO.

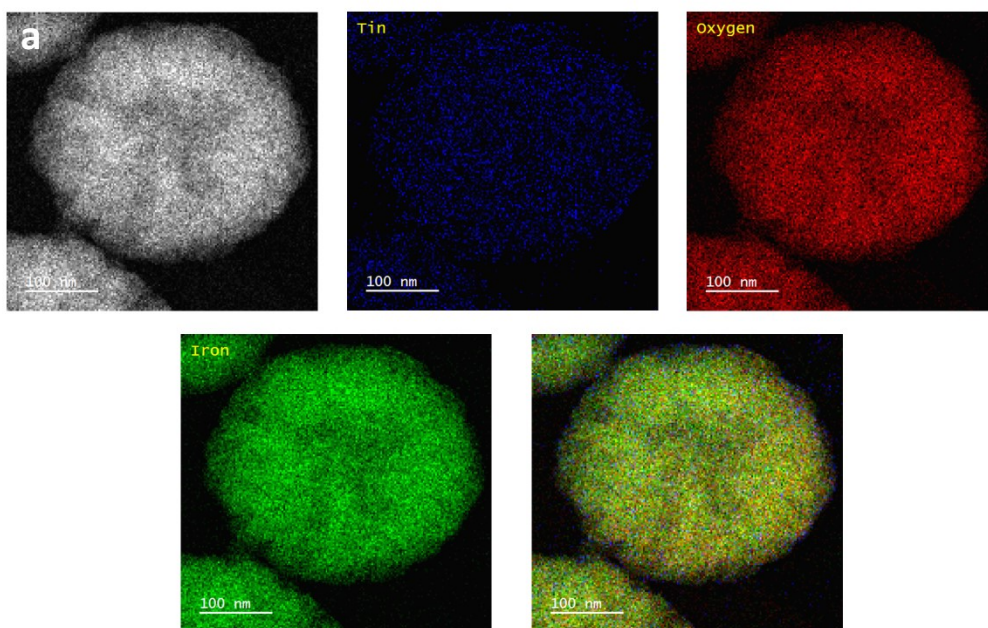


Fig. S5 HAADF-STEM images of TOFO and EDS corresponding elemental and mixed mapping.

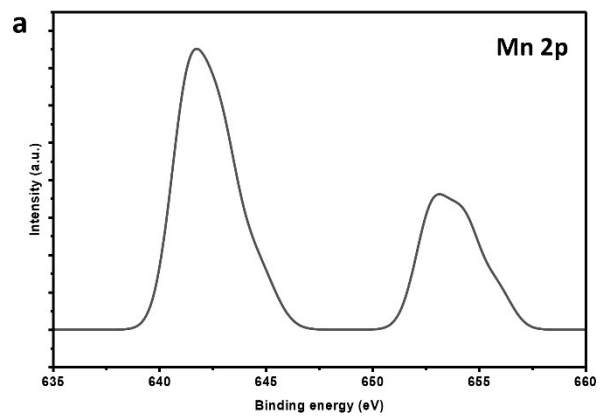


Fig. S6 (a) Mn 2p XPS of MTOFO.

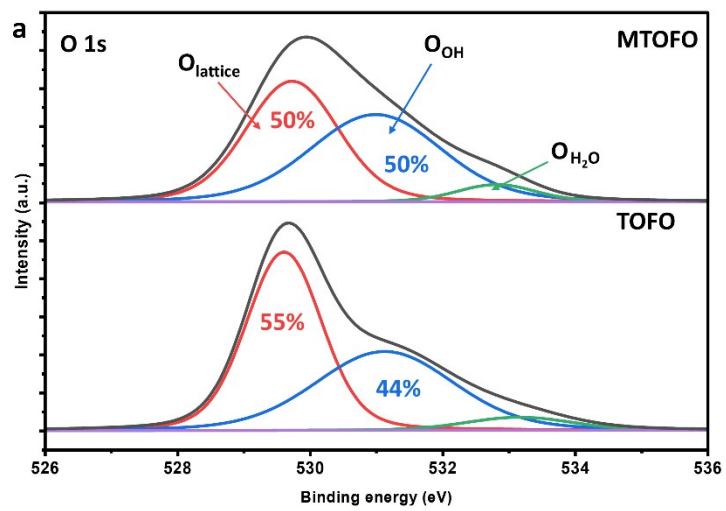


Fig. S7 O 1s XPS of (a) MTOFO and (b) TOFO.



Fig. S8 Electrochemical cell set up with magnet.

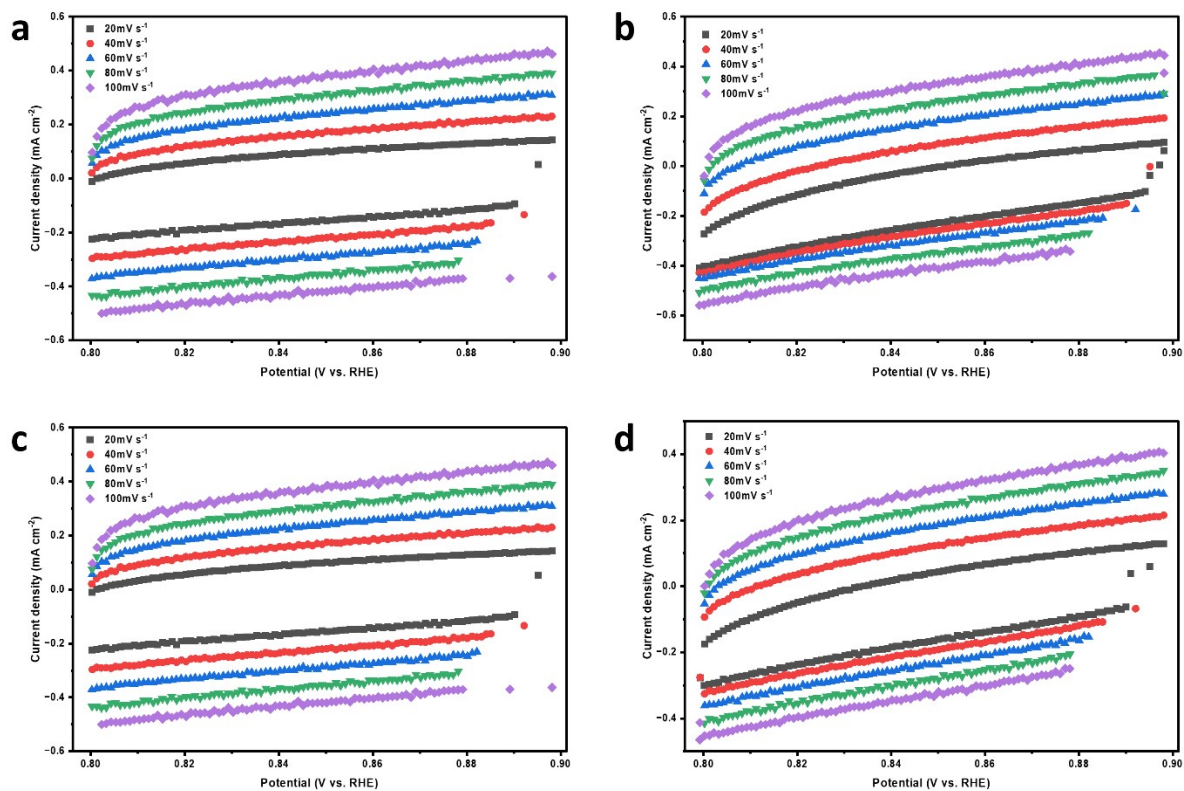


Fig. S9 Various scan rate CV of (a) TOFO without magnet, (b) with magnet, (c) MTOFO without magnet, and (d) with magnet.

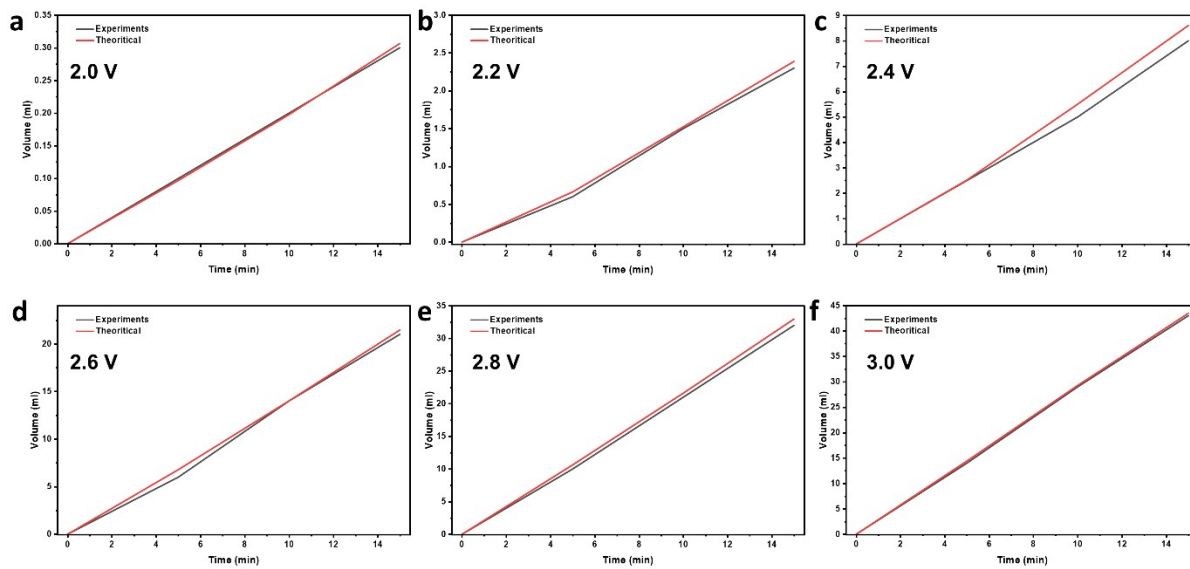


Fig. S10 The amount of hydrogen production at each potential (a) - (f).

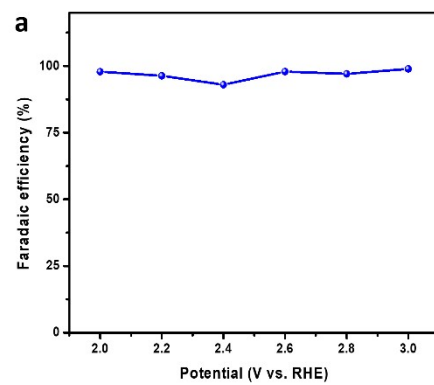


Fig. S11 The faradaic efficiency of hydrogen at each potential.

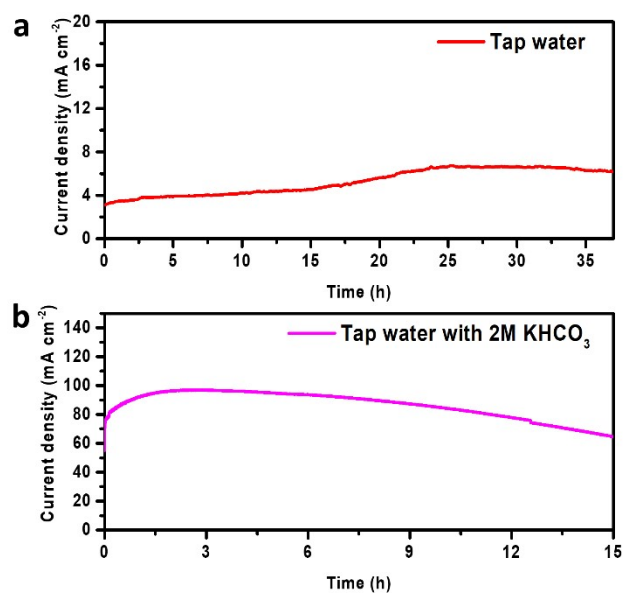


Fig. S12 The Stability of MTOFO at 2.6 V vs. RHE in (a) tap water and (b) tap water with 2M KHCO₃.

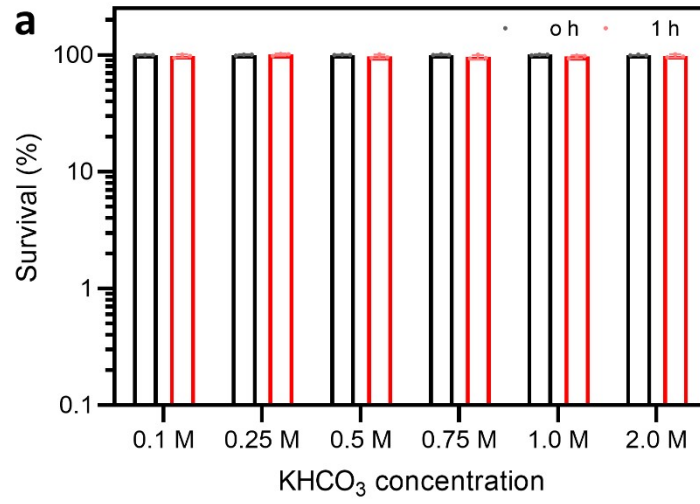


Fig. S13 Viability of *E. coli* depending on KHCO₃ concentration ($n = 3$ independent experiments per bacterial concentration).

Table. S1 Comparison table of 2-electron WOR performance.

Electrodes (catalysts)	Solution	E vs. RHE	Production rate $\mu\text{mol min}^{-1} \text{cm}^{-2}$	Reference
Mn-SnO₂@Fe₃O₄	2M KHCO₃	2.6 V	60.2	This work
BiVO ₄	1M NaHCO ₃	3.1 V	5.9	[1]
CaSnO ₃	2M KHCO ₃	3.2 V	4.7	[2]
Gd-BiVO ₄	2M KHCO ₃	3.2 V	10.6	[3]
PTFE/CPF	1M Na ₂ CO ₃	2.4 V	23.4	[4]
BDD	2M KHCO ₃	3.5 V	19.7	[5]
Bi ₂ WO ₆ : Mo	2M KHCO ₃	3.2 V	300	[6]
F _{0.4} -WO ₃	1M H ₂ SO ₄	3.4 V	79.92	[7]
SnO ₂	1M NaHCO ₃	3.1 V	1.6	[8]
CuWO ₄ /CC	2M KHCO ₃	3.0 V	11.8	[9]
TiO ₂	1M NaHCO ₃	3.3 V	1	[10]
MgSnO ₃ @NF	2M KHCO ₃	2.6 V	46.07	[11]
Mg _{1-x} SnWO _{6-x} 700	2M KHCO ₃	2.6 V	68.07	[12]

Table. S2 Comparison table of catalytic disinfection performance.

Electrodes (catalysts)	Solution	Time	Operate condition	Performance	Referenc e
Mn-SnO₂@Fe₃O₄	2M KHCO₃	60m	2.6 V vs. RHE	99.9% at 2 × 10⁶ CFU mL⁻¹ E. coli	This work
Cathode: Reticulated vitreous carbon Anode: Ti/RuO ₂	250mg L ⁻¹ Na ₂ SO ₄	150 min	32 mA	99.99% at 10 ⁶ CFU mL ⁻¹ E. coli	[13]
Pt-coated titanium electrodes	Tap water	45 min	6 V	99.999% at 2 × 10 ² CFU mL ⁻¹ Legionella	[14]
Nitrogen-doped reduced graphene oxide	Tap water	15 min	5.8 mA cm ⁻²	99.999% at 10 ⁶ CFU mL ⁻¹ E. coli	[15]
Co ₃ O ₄	5mM NaCl	160 min	2.5 V	99.9999% at 5*10 ⁶ CFU mL ⁻¹ E. coli	[16]
CuO	9.0 g L ⁻¹ sodium chloride	960 min	6 V	99.9999% at 10 ⁷ CFU mL ⁻¹ E. coli)	[17]
Cu ₃ P	DI water	720 min	1 V	99.9999% at 10 ⁷ CFU mL ⁻¹ E. coli and S. aureus	[18]
Ag ₁ /CN (Ag single atom/C ₃ N ₄)	DI water	240 min	Xe lamp (<400nm, 300W)	99.9% at 10 ⁵ CFU mL ⁻¹ E. coli, S. aureus, MRSA and KREC	[19]
h ³ -FNCs (Fe single atom/NC)	DI water	240 min	Oxidase-like Activity	>90% and 99% at 10 ⁸ CFU mL ⁻¹ P. aeruginosa and S. aureus	[20]
Cu Sazyme (Cu single-atom nanozyme)		240 min (white ligh for 30 min and NIR for 10min)	Dual-light activation (808nm, 1.0W cm ⁻² NIR + 400-700 nm 5mW cm ⁻² White light)	97% at 10 ⁶ CFU mL ⁻¹ E. coli, MDRPA, S. aureus and MRSA	[21]

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