

Supplementary Information

Unveiling the Mechanism of Fe-Doping-Enhanced Oxygen Evolution Reaction of CoSe

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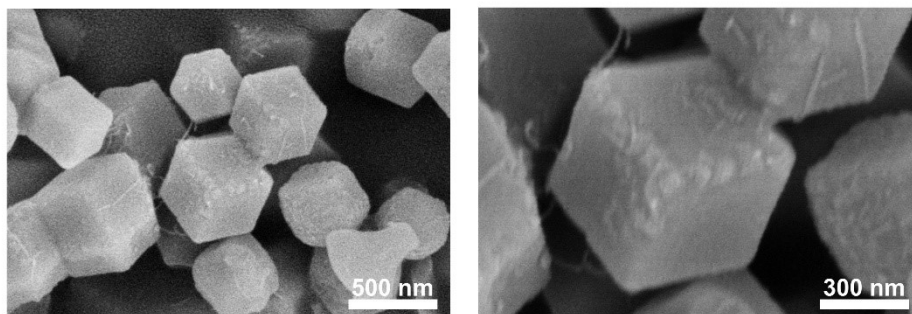


Figure S1. SEM images of Co-ZIF.

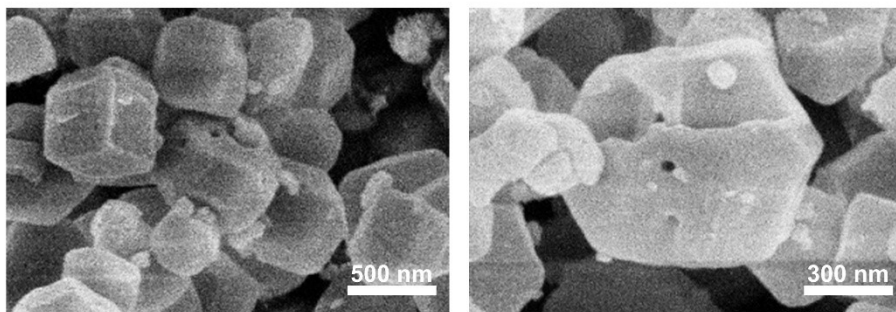


Figure S2. SEM images of CoFe-ZIF.

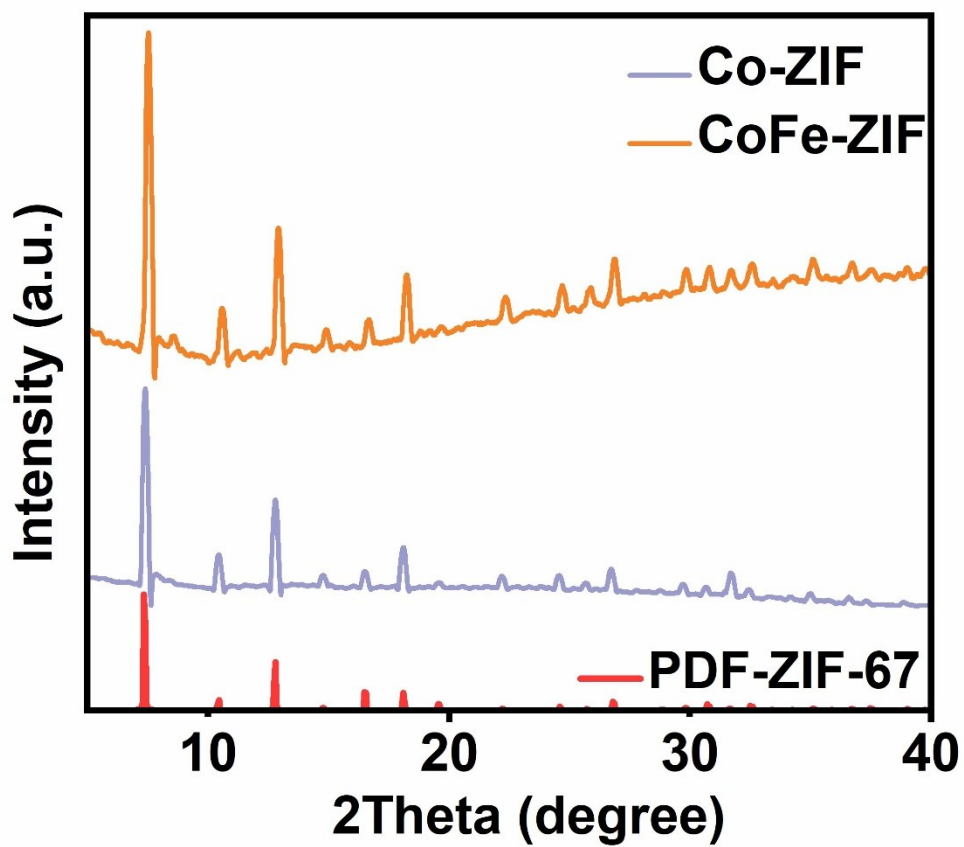


Figure S3. XRD patterns of Co-ZIF and CoFe-ZIF.

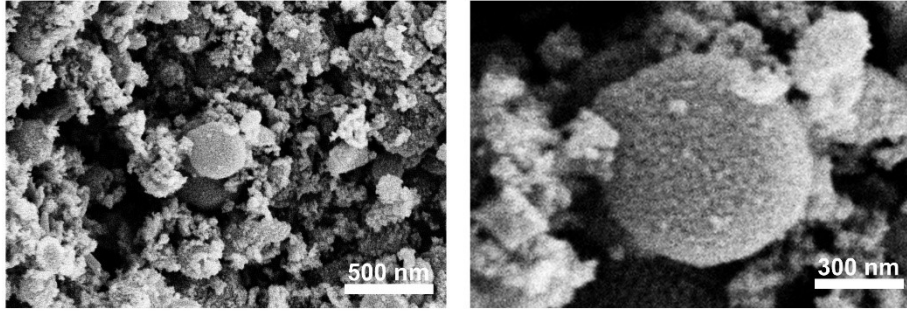


Figure S4. SEM images of CoFeSe.

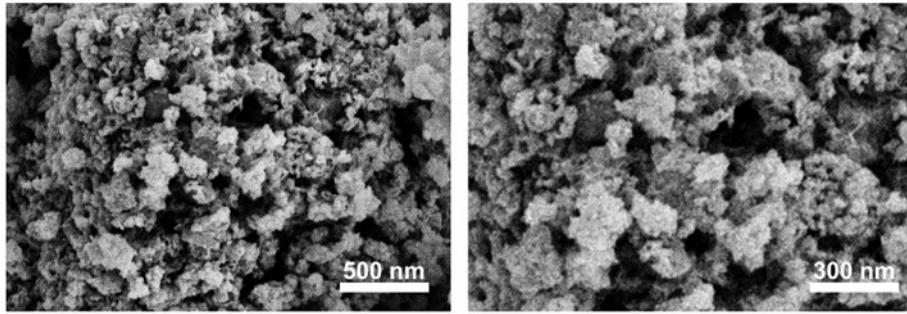


Figure S5. SEM images of CoSe.

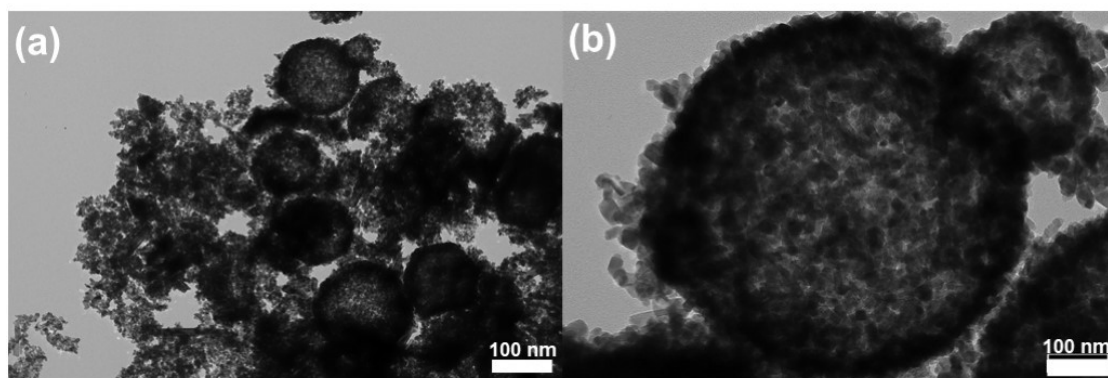


Figure S6. TEM images of CoFeSe.

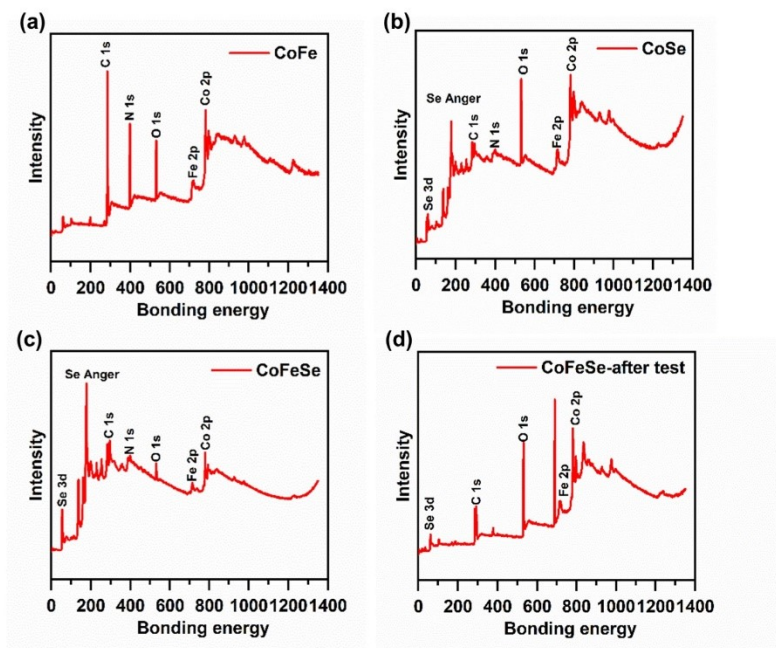


Figure S7. XPS overall spectrum of (a) CoFe-ZIF, (b) CoSe, (c) CoFeSe and (d) CoFeSe-after test in high-resolution.

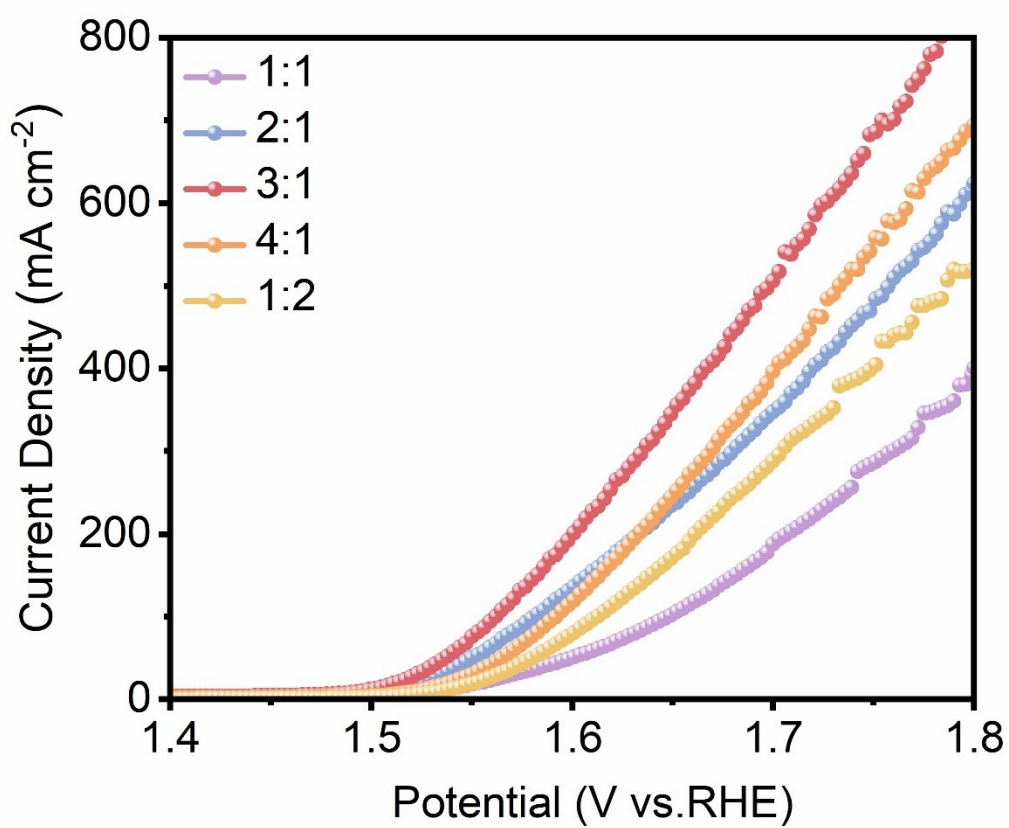


Figure S8. LSV curves of CoFeSe with different Se/Co ratios.

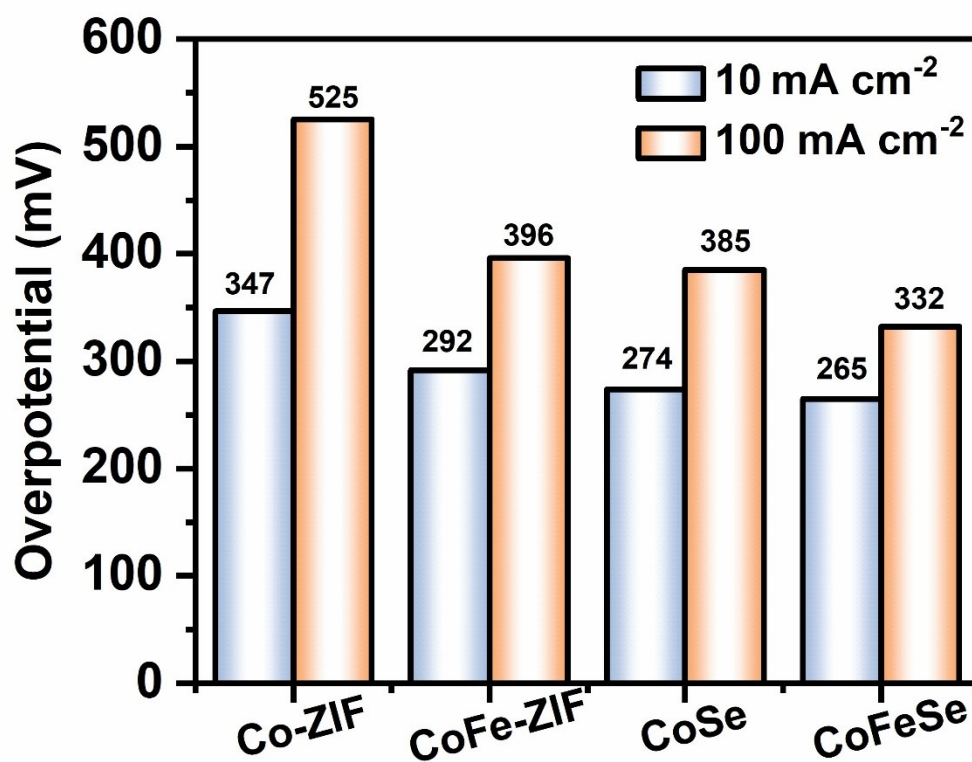


Figure S9. Overpotentials of the fabricated samples at the current densities 10 and 100 mA·cm⁻².

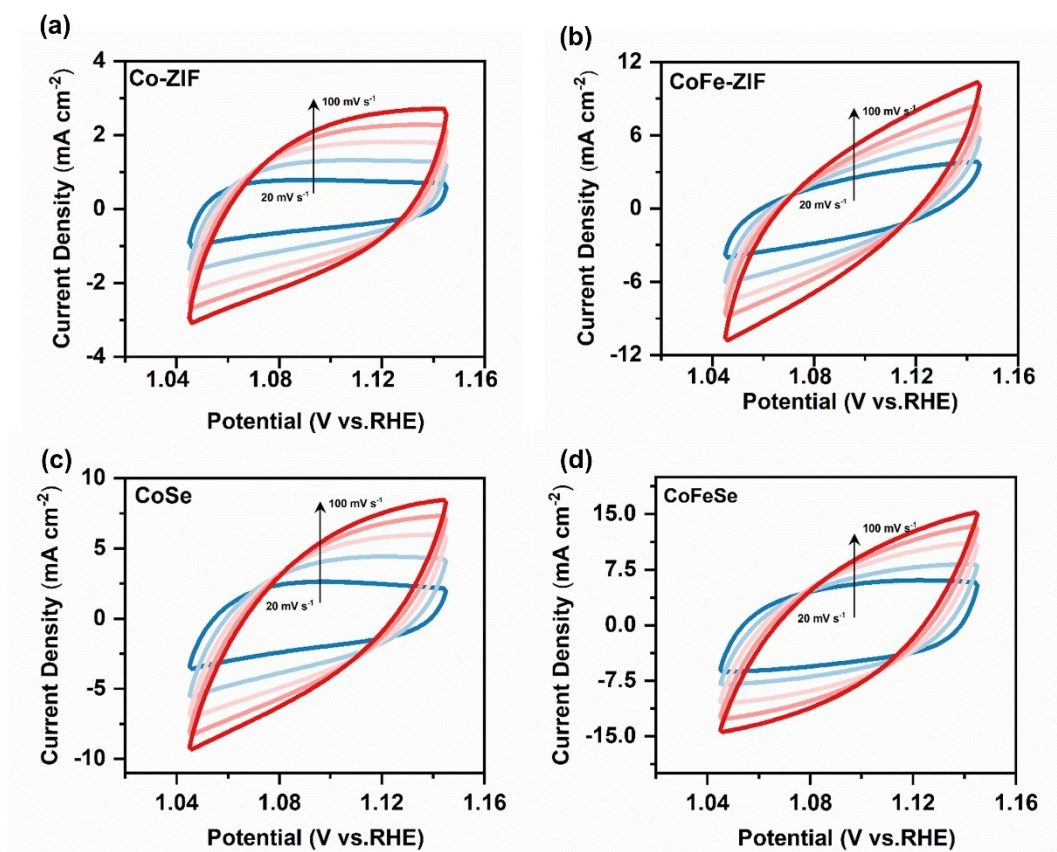


Figure S10. CV curves obtained in non-Faraday intervals for (a) Co-ZIF, (b) CoFe-ZIF, (c) CoSe, and (d) CoFeSe at different scan speeds from 20 to 100 mV s^{-1} .

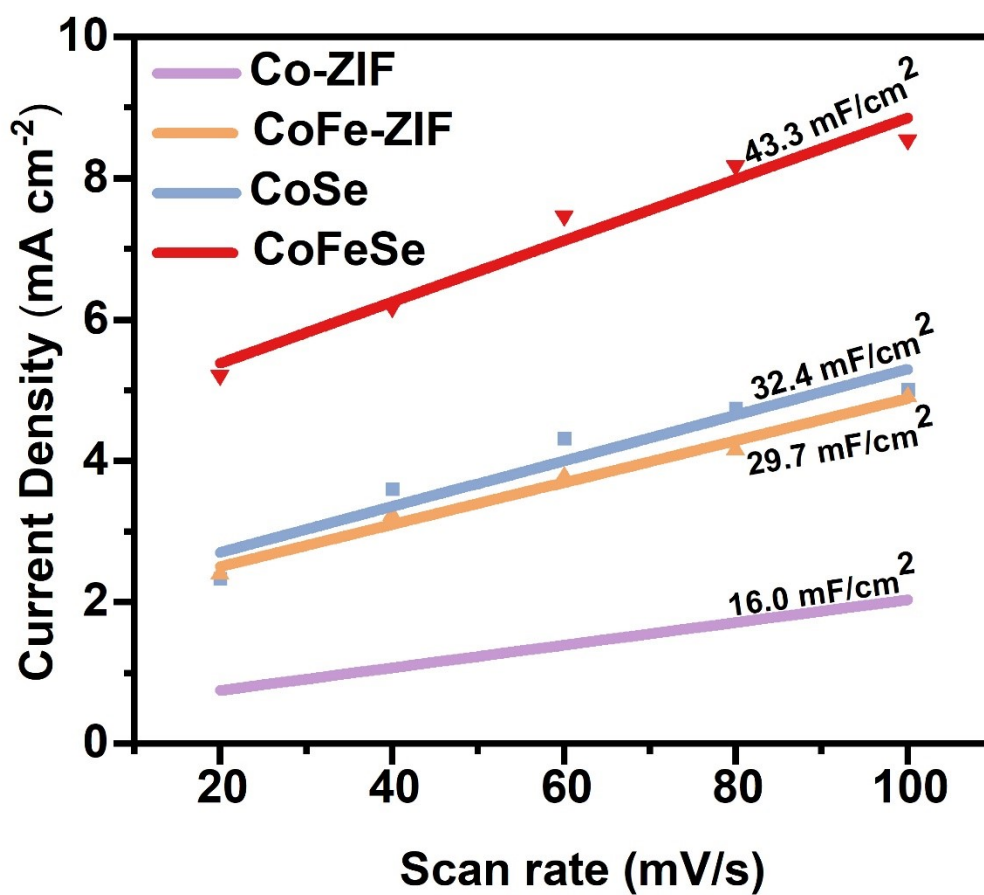


Figure S11. The C_{dl} of Co-ZIF, CoFe-ZIF, CoSe and CoFeSe.

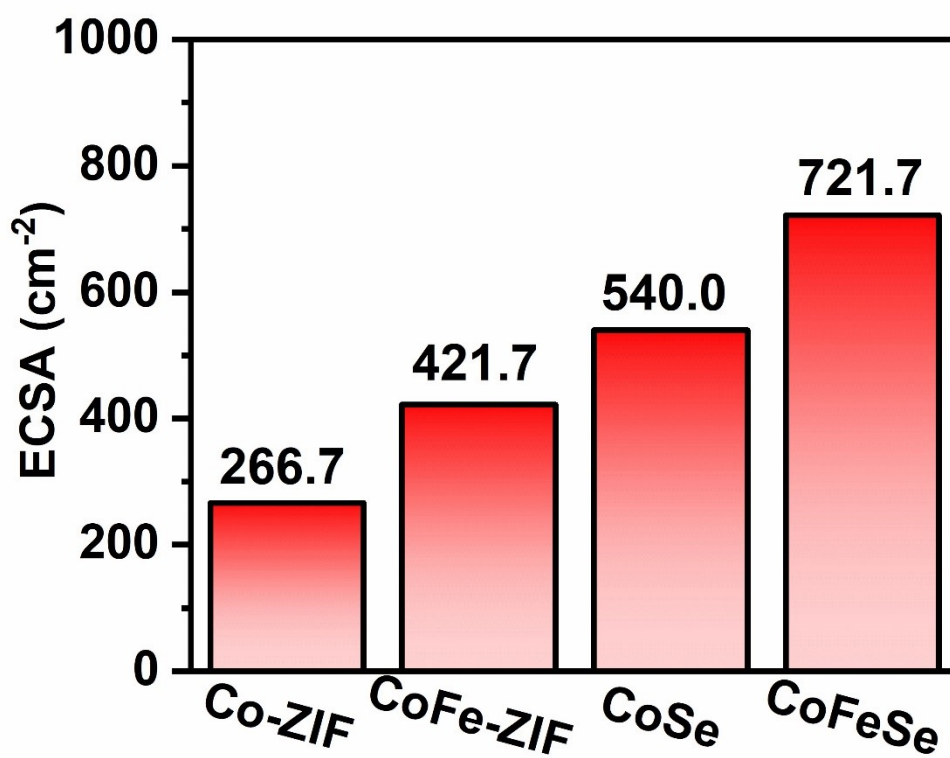


Figure S12. The ECSA of Co-ZIF, CoFe-ZIF, CoSe and CoFeSe.

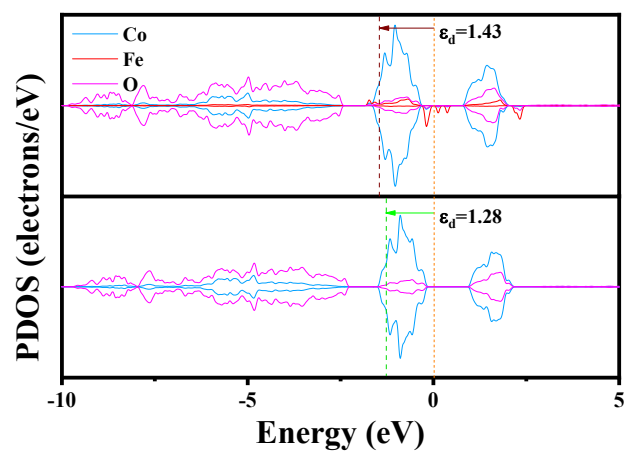


Figure S13 PDOS for CoOOH and CoFeOOH.

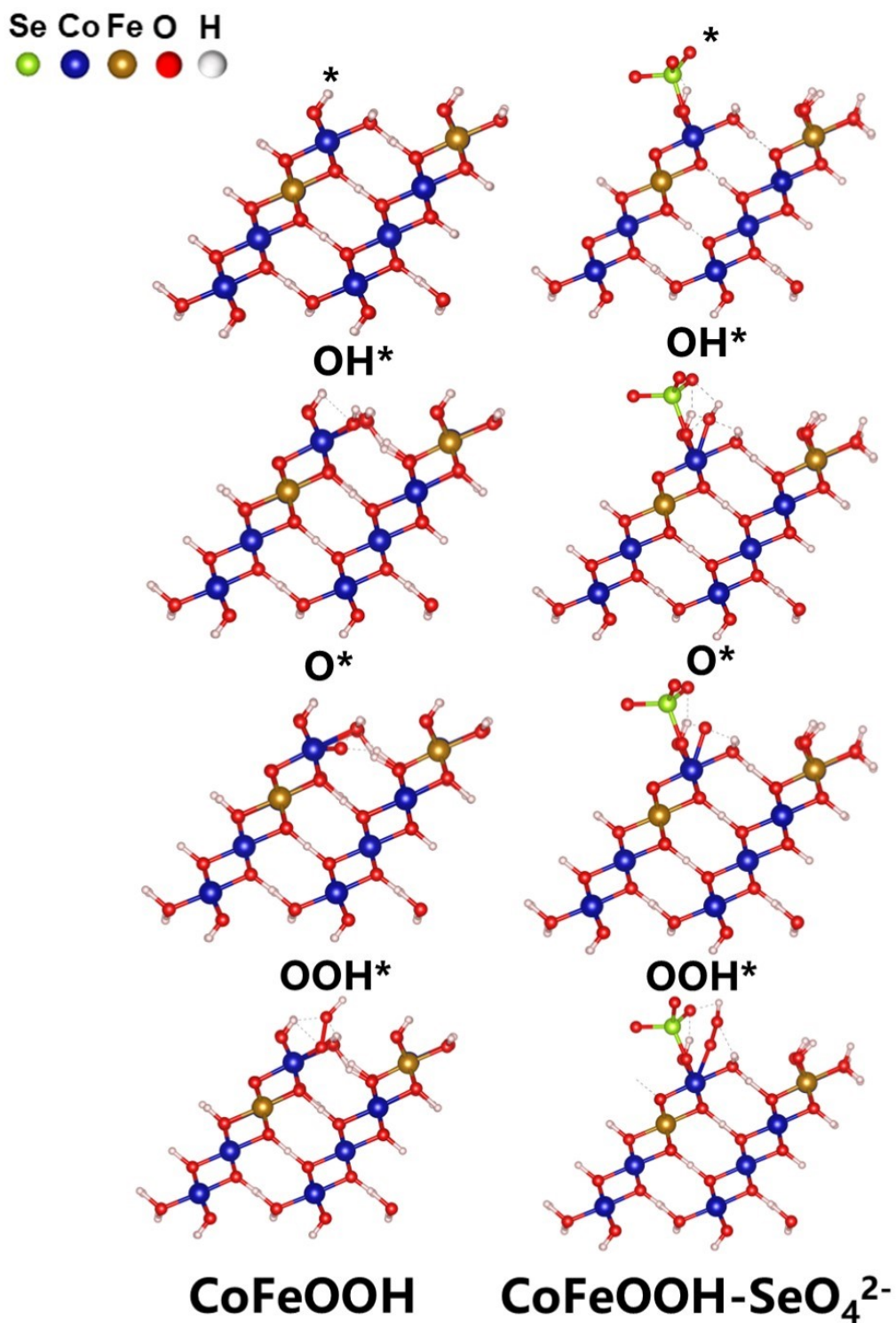


Figure S14 The balanced geometric configuration and intermediate adsorption structure of CoFeOOH and CoFeOOH-SeO₄²⁻.

Table S1. The elements content in the fabricated samples based on the ICP-OES analysis

Catalysts	Co (wt.%)	Fe (wt.%)
CoFeSe	35.6%	3.29%
CoFe	34.8%	7.6%
CoSe	38.1%	

Table S2. Comparison of the overpotential and Tafel slope of CoFeSe with reported ZIF-67 derived catalysts in alkaline electrolyte.

Catalysts	Overpotential (mV)	Tafel (mV dec ⁻¹)	References
CoFeSe	265	57.3	This Work
Co(OH) ₂ -MoO ₃ /NF	273	72	1
NiCoP	273	88.9	2
Co-S@CA	291	61.8	3
ZIF@NiBiCo	300	77	4
CoNC@FePc	314	73.4	5
Z-67/Ni-V	320	71	6
ZIF-67@Ni@FeOOH	329	42.95	7
Cu ₄ -ZIF-67	340	65.64	8
ZIF-8/ZIF-67@Pd	340	85	9
ZIF-67 200	347	105.1	10
75NZ67	350	120	11
Co-LDH-1	354	79.06	12
NiFe-LDH-Sn/ZIF-67	360	79	13
ZIF-67@Fe28	379	130	14

References

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