

Supplementary Information Associated with

Unraveling the therapeutic potentials of a thiosemicarbazone based Zn(II) metal-organic triangle complex

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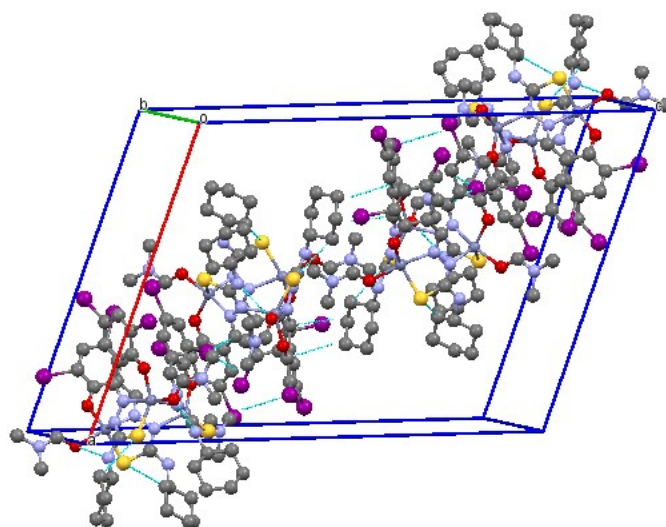


Figure S1. Packing diagram of complex **1** along the b-axis showing hydrogen bonding interactions.

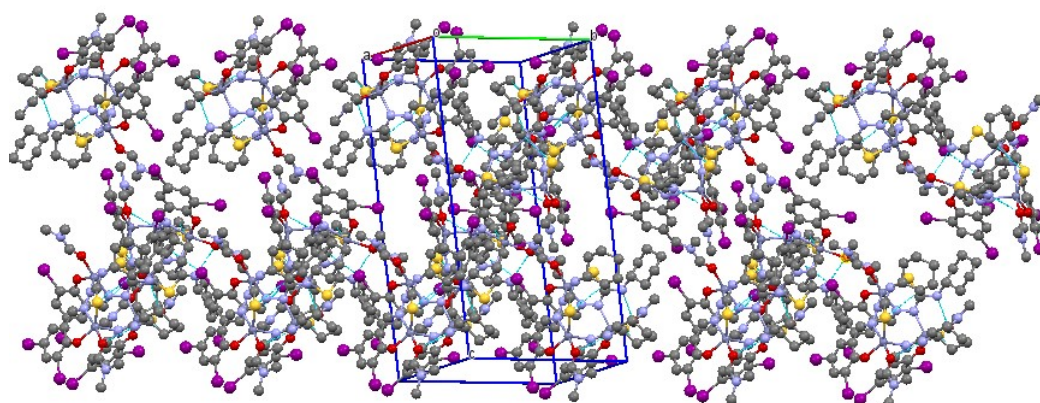


Figure S2. Packing of complex **1** showing molecules are packed along the a-axis and the formation of 3D motif through hydrogen bonding and other short contacts.

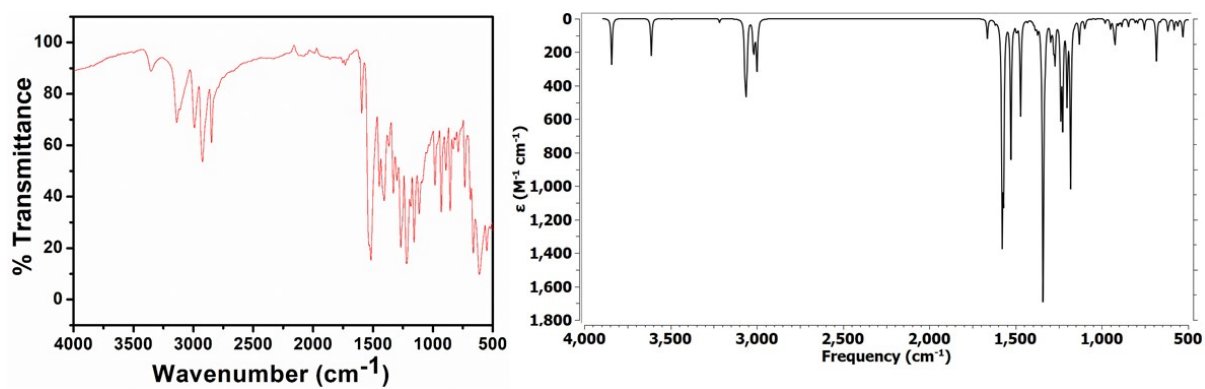


Figure S3. Experimental (left) and theoretical (right) IR spectra of the ligand H₂L.

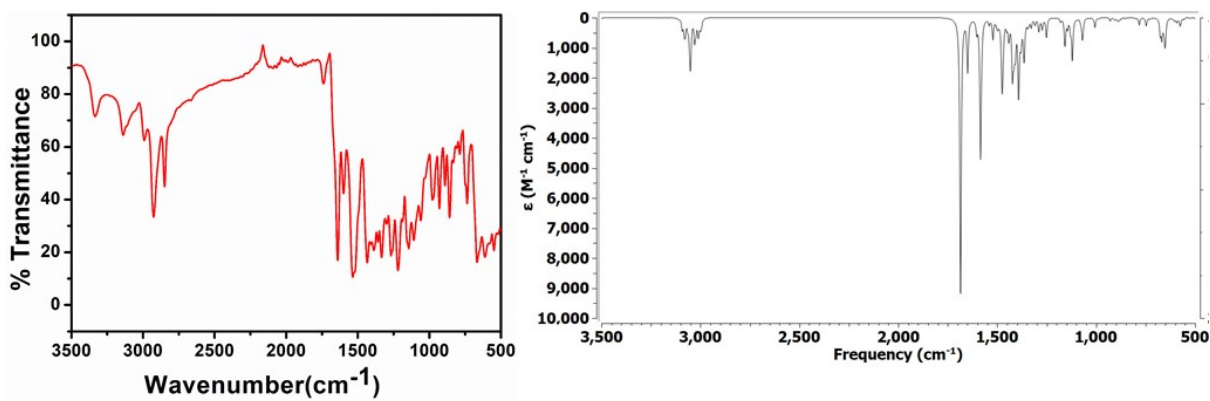


Figure S4. Experimental (left) and theoretical (right) IR spectra of the complex **1**.

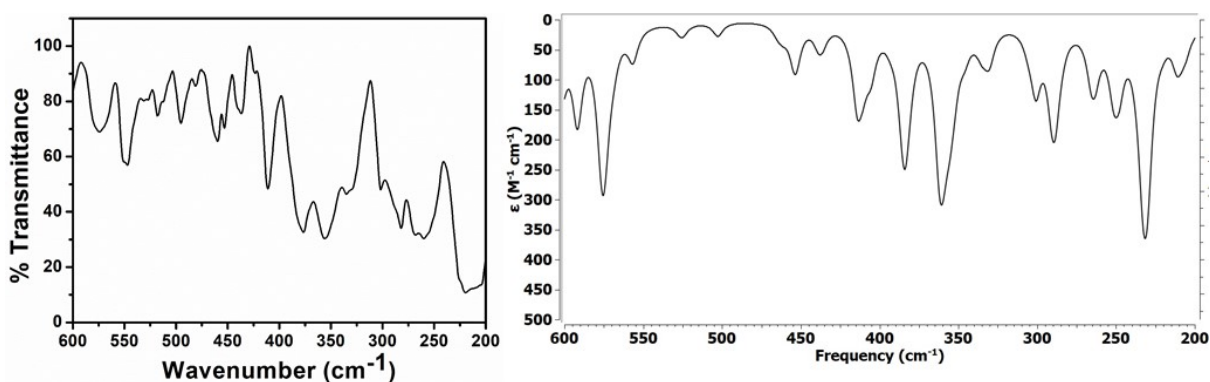


Figure S5. Experimental (left) and theoretical (right) Far IR spectra of the complex **1**.

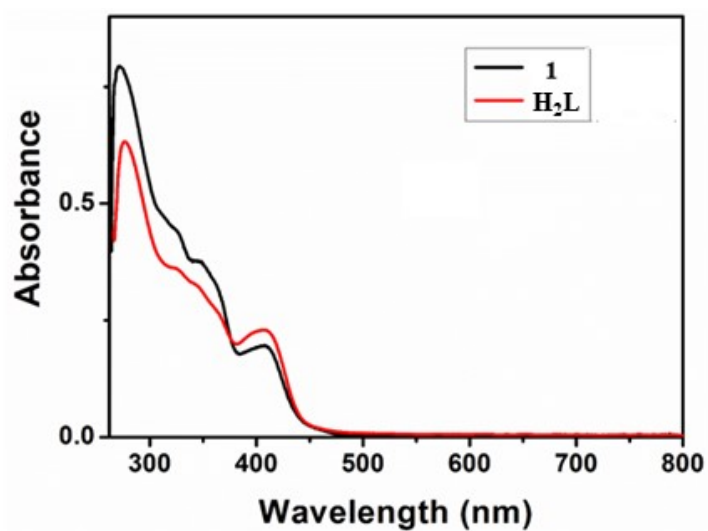


Figure S6. UV-vis spectra of ligand H₂L and the complex **1** in 10⁻⁵ M DMF solution.

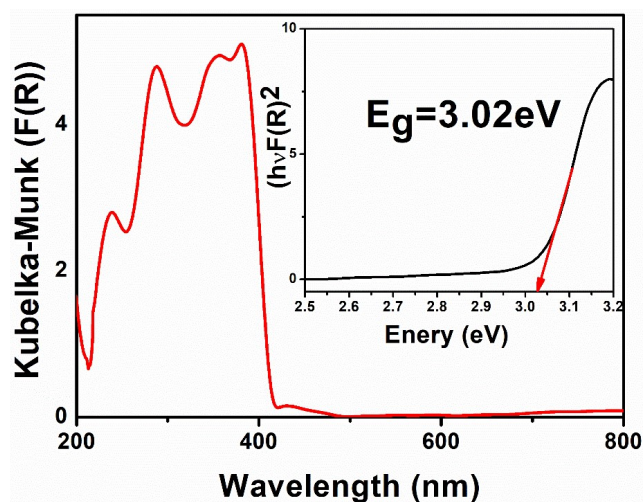


Figure S7. UV-vis diffuse reflectance spectrum of H₂L. Inset shows the band gap calculations.

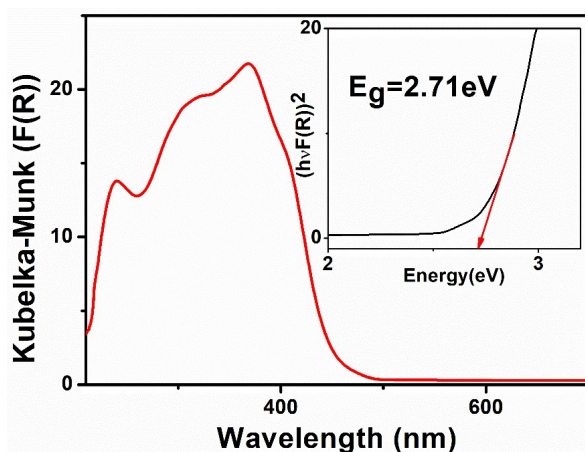


Figure S8. UV-vis diffuse reflectance spectrum of complex 1. Inset shows the band gap calculations.

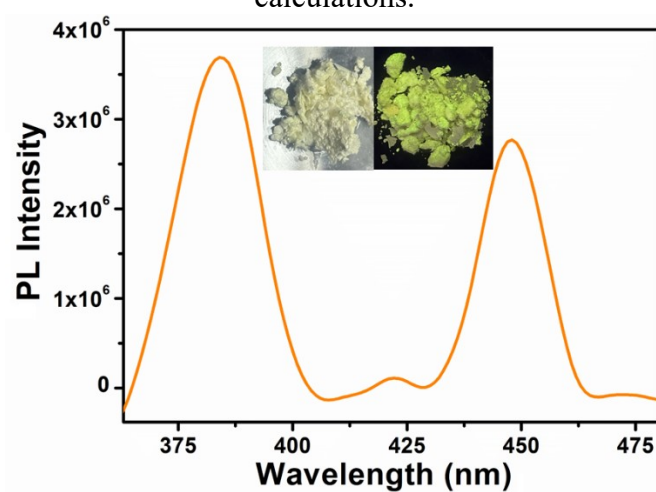


Figure S9. Solid state emission spectrum of H₂L ($\lambda_{\text{exc}} = 307 \text{ nm}$). Inset shows the photographs under daylight and UV light.

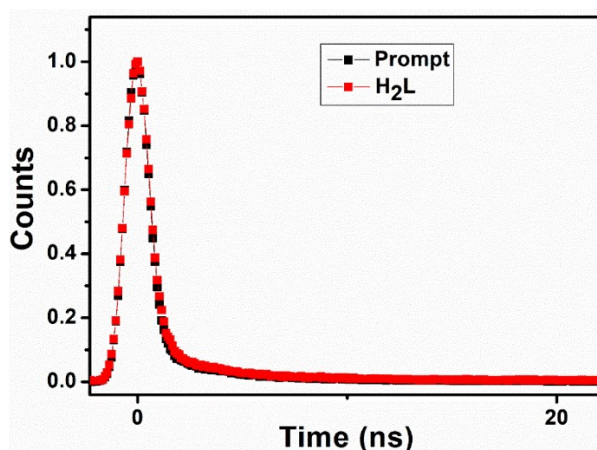


Figure S10. Time resolved fluorescence decay curve of the ligand H₂L.

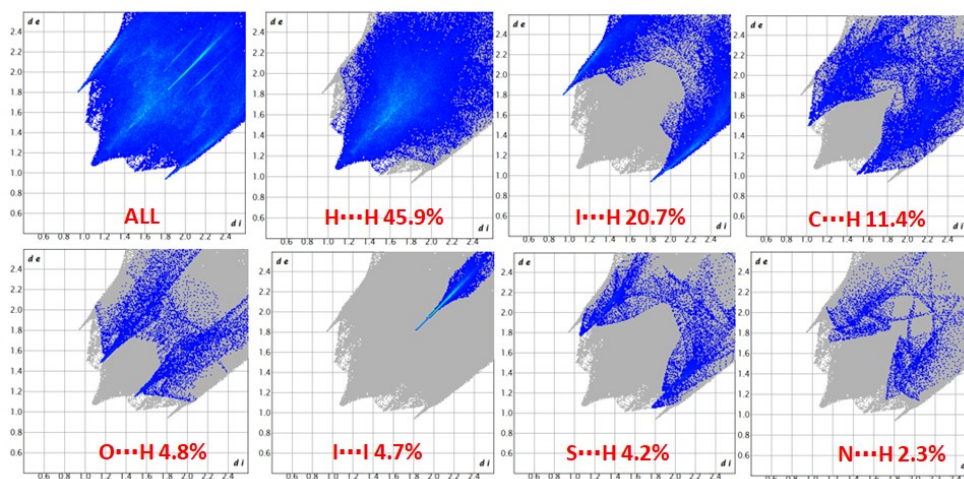


Figure S11. 2D Fingerprint plots showing the percentages of contacts contributed to the total Hirshfeld surface area of the complex **1**.

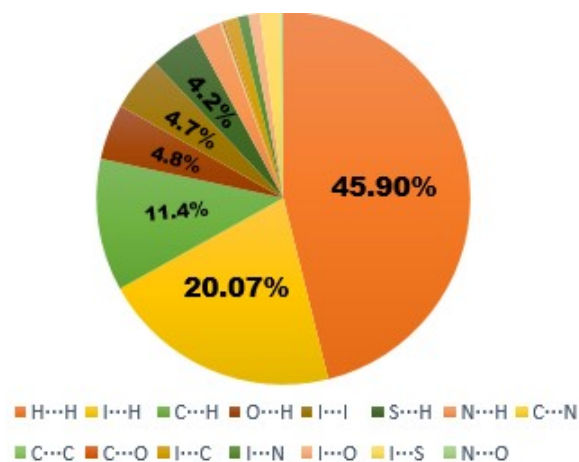


Figure S12. Relative contributions of various intermolecular contacts to the Hirshfeld surface area in complex **1**.

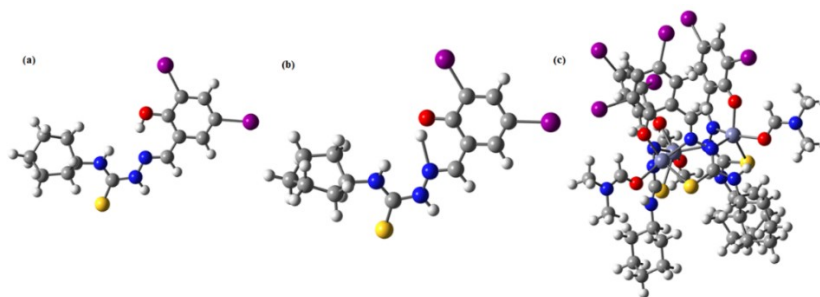


Figure S13. Optimized geometries of (a) H₂L, (b) H₂L-keto, (c) complex 1.

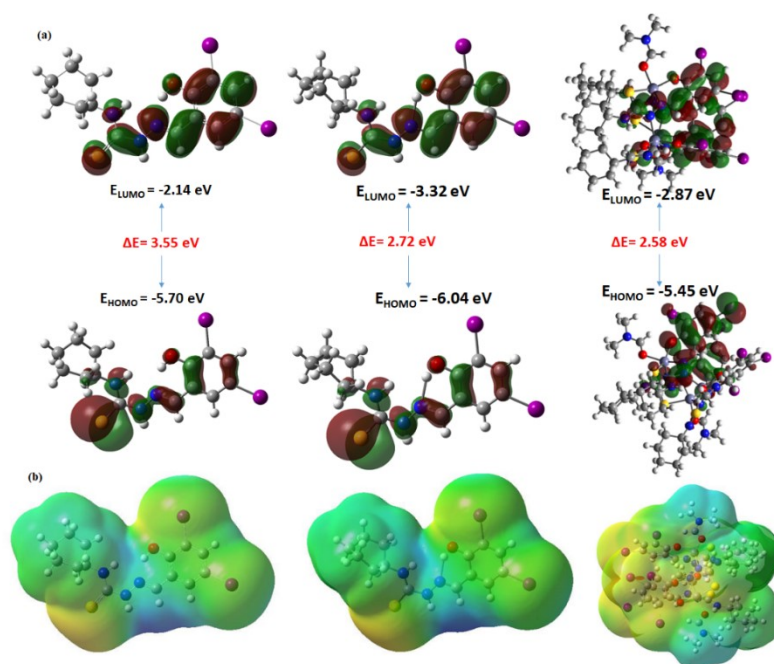


Figure S14. (a) Frontier molecular orbitals, (b) MEP plots of H₂L, H₂L-keto, complex 1.

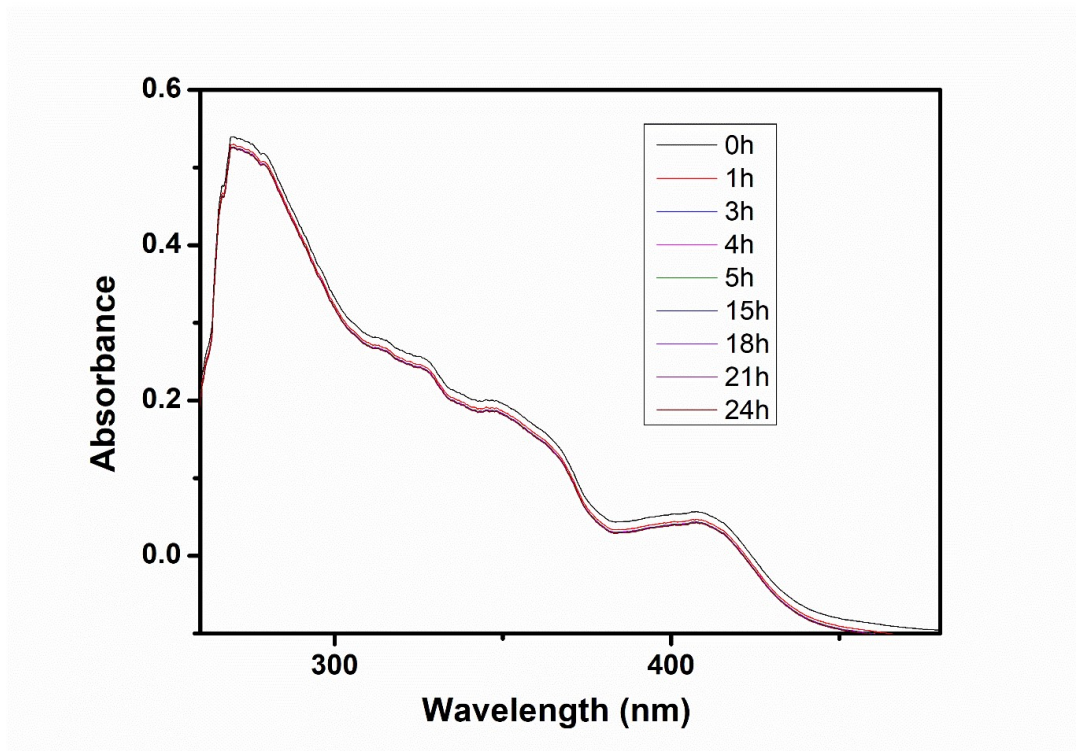


Figure S15. Time dependent absorption spectra of complex 1 in buffer solution at pH =7.4.

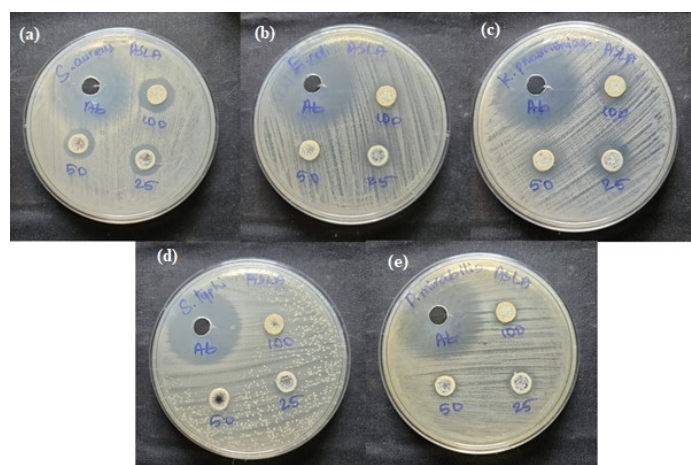


Figure S16. Zone of inhibition of the ligand H₂L against (a) *S. aureus*, (b) *E. coli*, (c) *K. pneumoniae*, (d) *S. typhi*. (e) *P. mirabilis*.

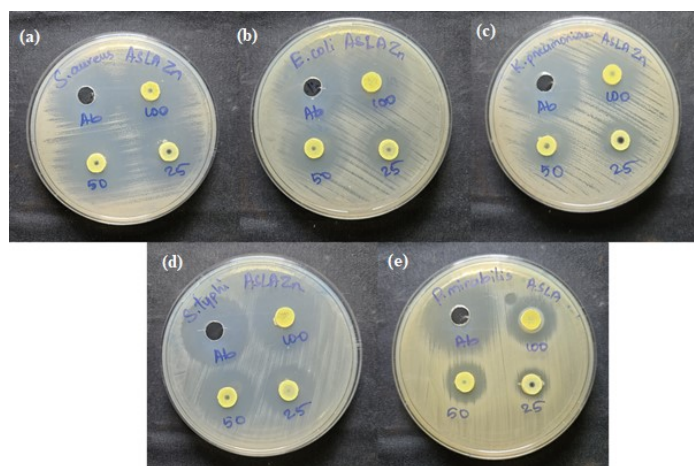


Figure S17. Zone of inhibition of the complex 1 against (a) *S. aureus*, (b) *E. coli*, (c) *K. pneumoniae*, (d) *S. typhi*. (e) *P. mirabilis*.

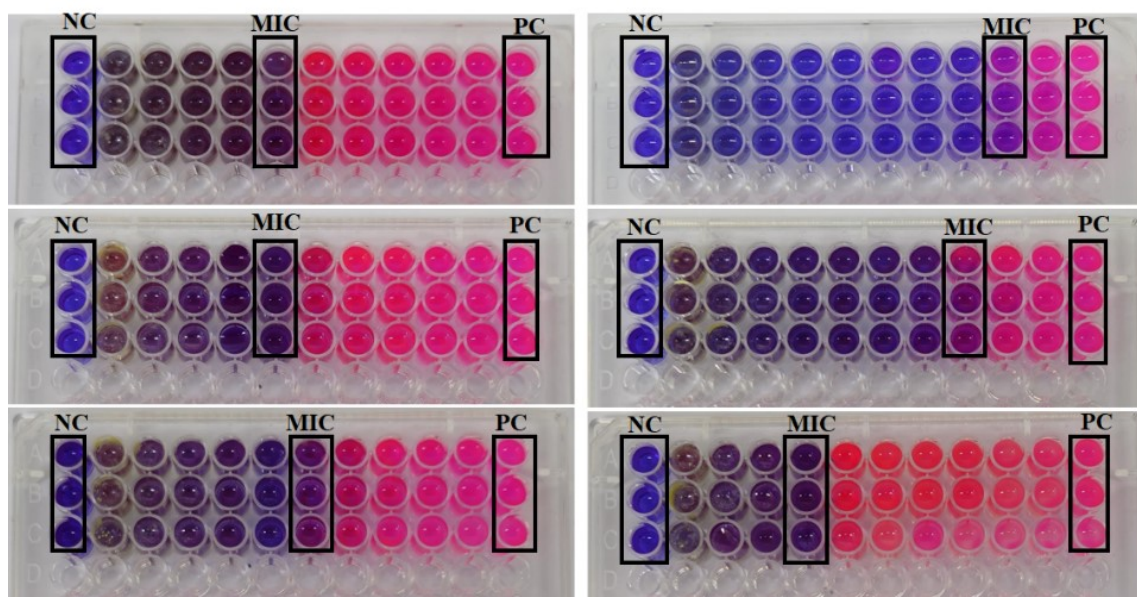


Figure S18. MIC determination against *S. aureus* of ligand H₂L (top left), complex 1 (top right), *E. coli* of complex 1 (middle left), *K. pneumoniae* of complex 1 (middle right), *S. typhi* of complex 1 (bottom left), *P. mirabilis* of complex 1 (bottom right).

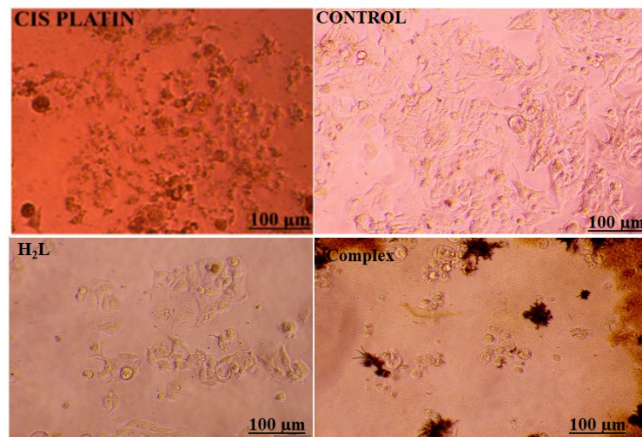


Figure S19. Morphological changes induced by H₂L and complex **1** against MCF7 cell lines compared with cisplatin at 100 µg/mL.

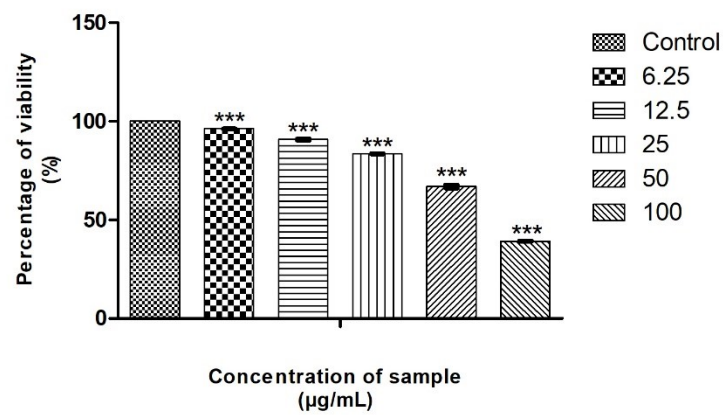


Figure S20. Assessment of HEK 293 cell survival in the presence of complex **1**, conducted by treating the cells with different concentrations. (*) indicate $p < 0.001$ when compared to control group.

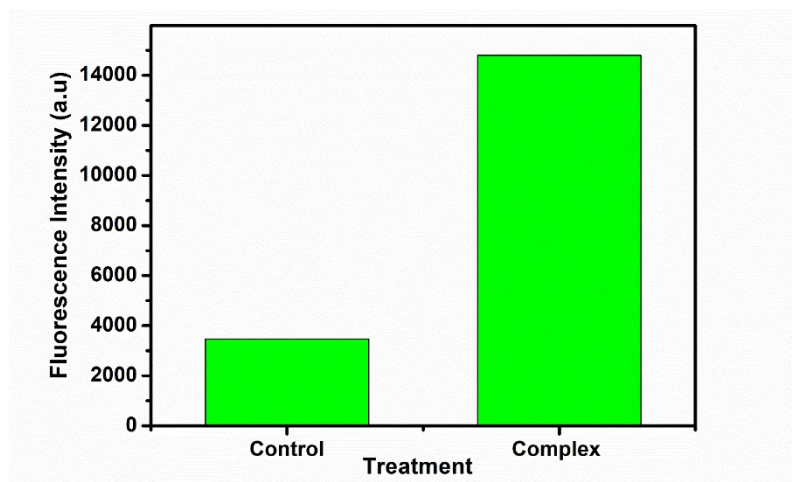


Figure S21. Fluorescent intensity plot of DCFDA/DCF assay performed for the detection of ROS generation in MCF7 cells by complex 1 compared with untreated control.

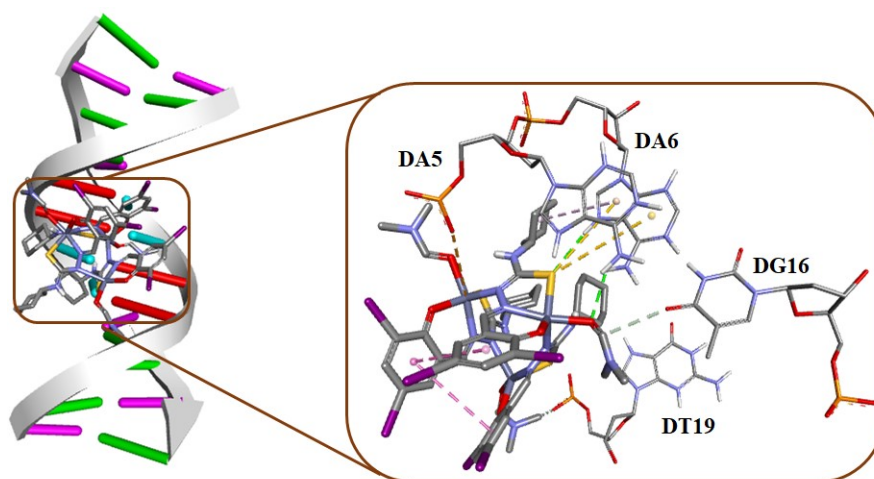


Figure S22. Binding mode of $[Zn_3(\mu-L)_3 \cdot 3DMF]$ (1) and its focused view of interactions with 1BNA.

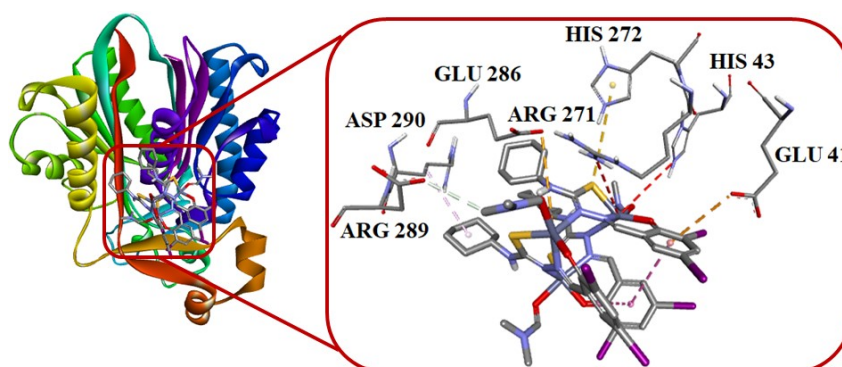


Figure S23. Binding mode of $[Zn_3(\mu-L)_3 \cdot 3DMF]$ (1) and its focused view of interactions with 1HNJ.

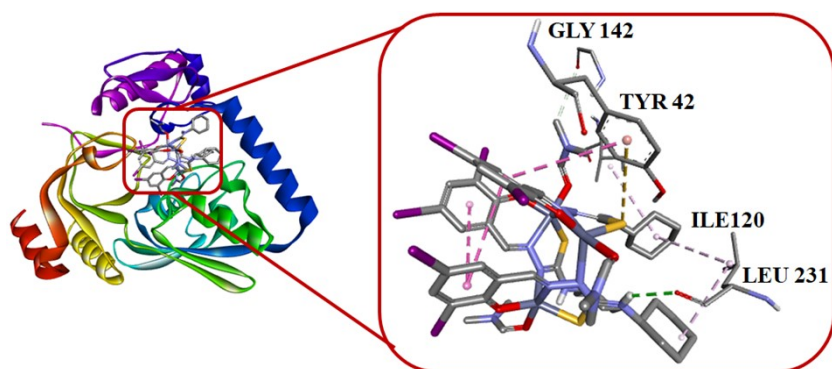


Figure S24. Binding mode of $[Zn_3(\mu-L)_3 \cdot 3DMF]$ (1) and its focused view of interactions with 1HSK.

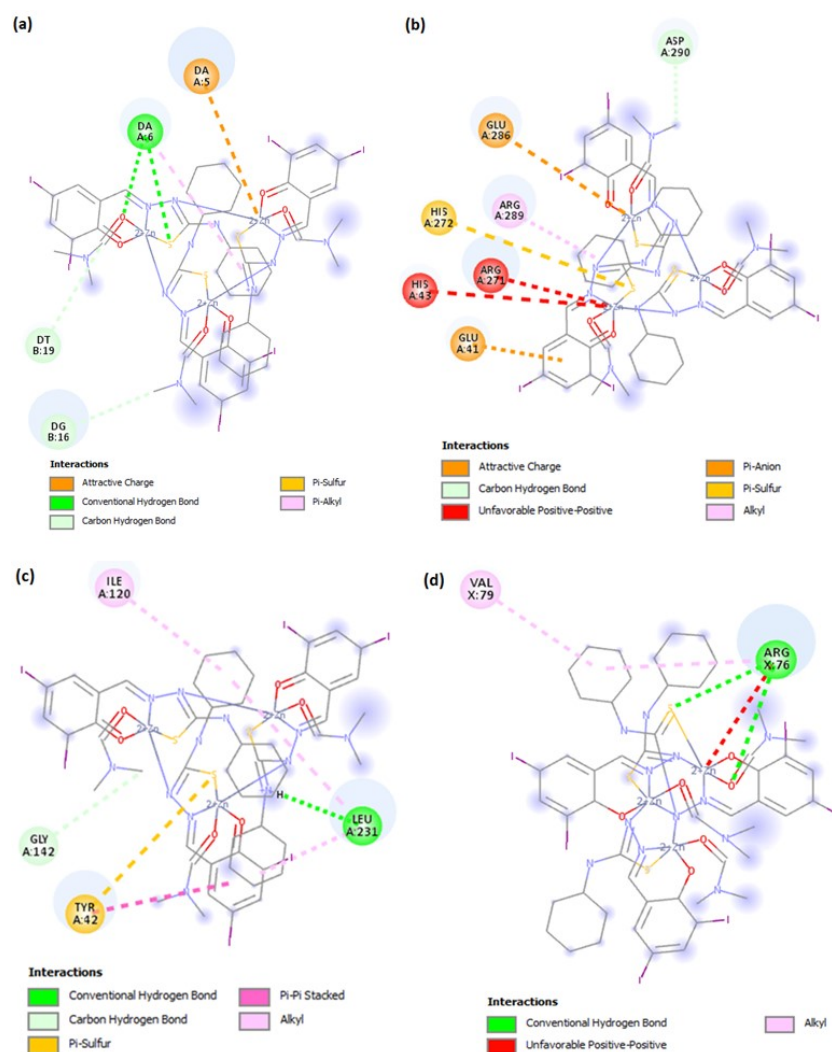


Figure S25. 2D representation of $[Zn_3(\mu-L)_3 \cdot 3DMF]$ (1) with the active site residues of (a) 1BNA, (b) 1HNJ, (c) 1HSK and (d) 3HB5.

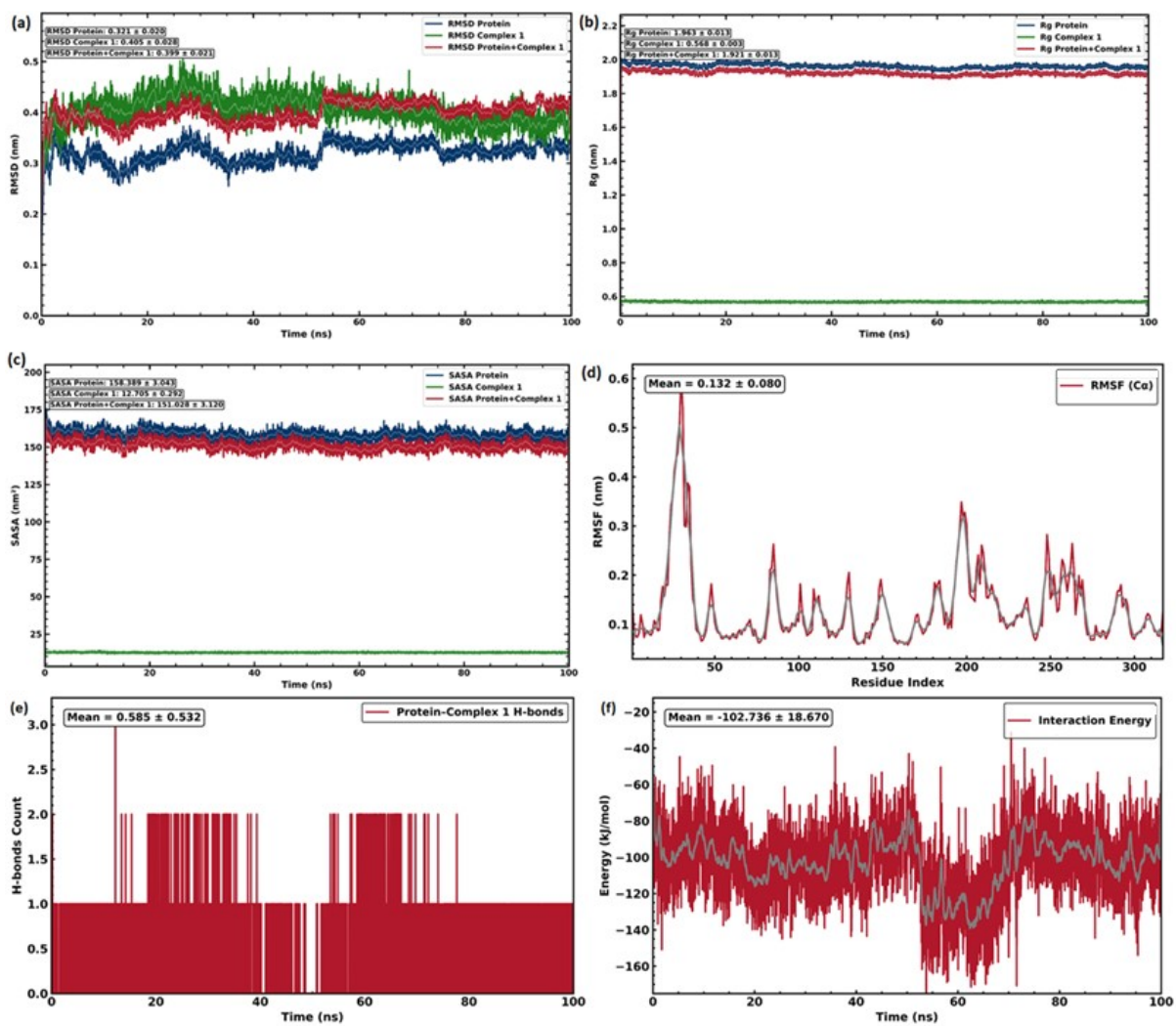


Figure S26. MD simulation profile of the Complex 1-1HNJ system over 100 ns (a) RMSD (b) Rg (c) SASA (d) RMSF (e) hydrogen bonds (f) interaction energy.

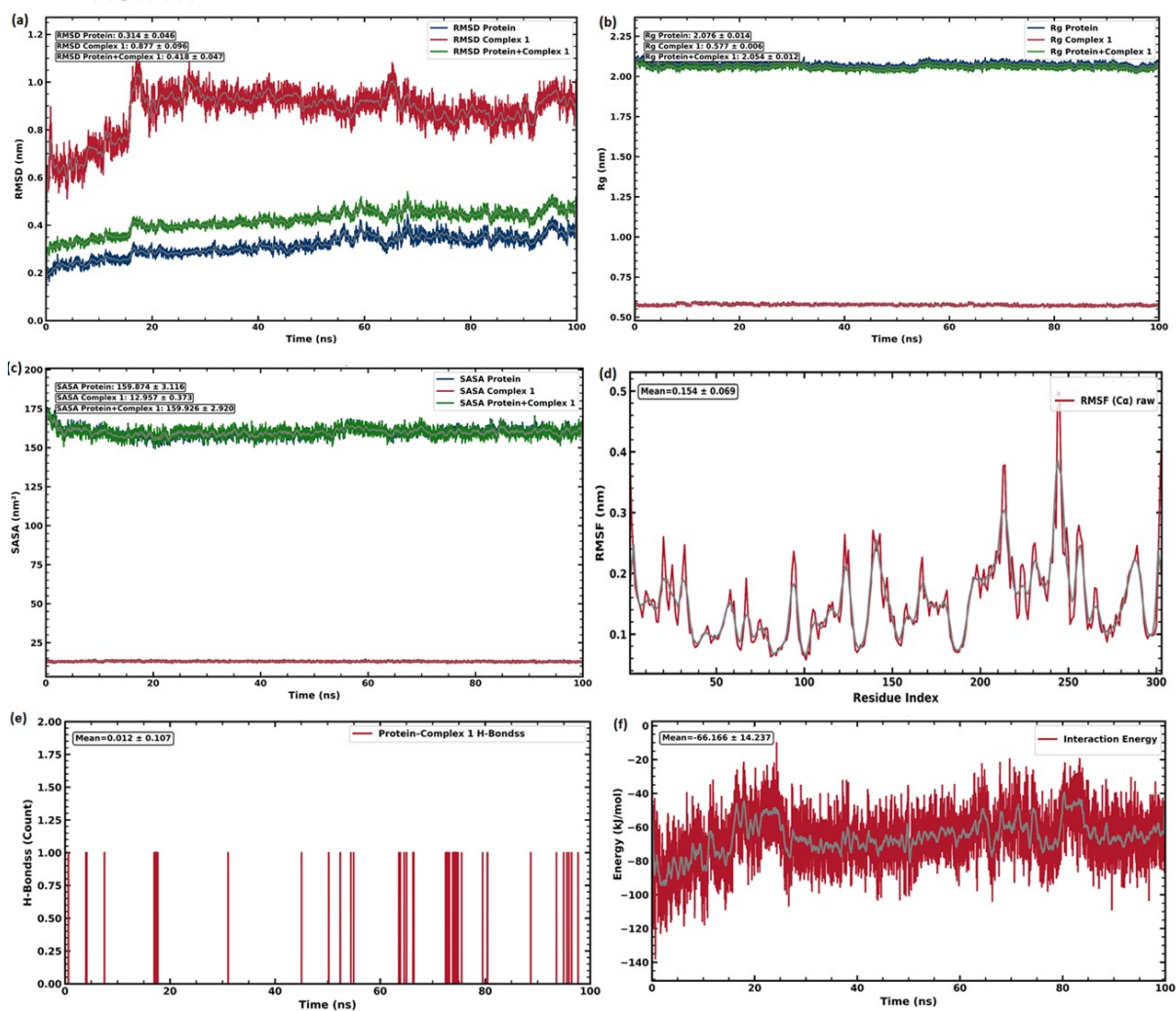


Figure S27. MD simulation profile of the Complex 1-IHSK system over 100 ns (a) RMSD (b) Rg (c) SASA (d) RMSF (e) hydrogen bonds (f) interaction energy.

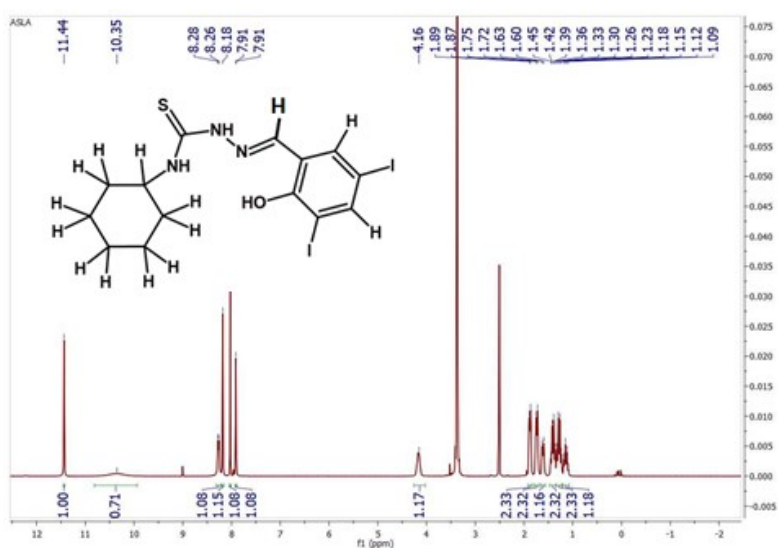


Figure S28. ¹H NMR spectrum of H₂L.

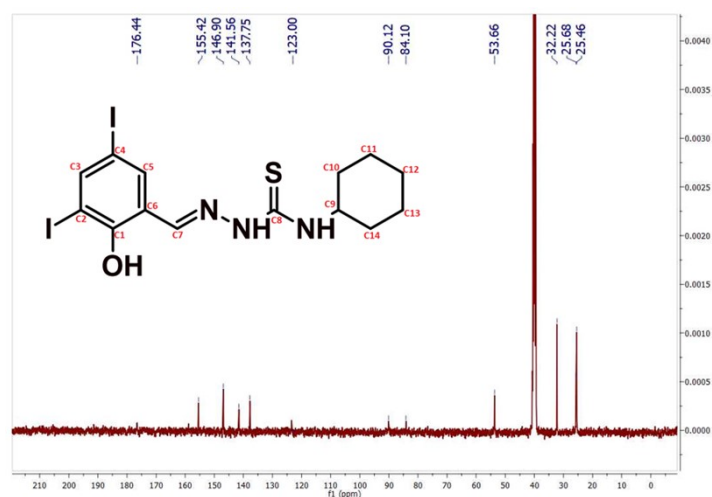


Figure S29. ^{13}C NMR spectrum of H_2L .

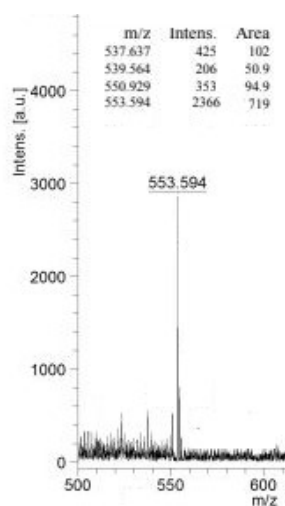


Figure S30. Mass spectrum of H_2L .

Table S1. The bond lengths (\AA) and bond angles ($^\circ$) of the complex $[\text{Zn}_3(\mu\text{-L})_3\cdot 3\text{DMF}]$ (**1**).

Bond lengths (\AA)		Bond angles ($^\circ$)	
Zn(1)–N(2)	2.077(3)	O(2)–Zn(1)–N(2)	113.35(11)
Zn(1)–N(7)	2.106(3)	O(2)–Zn(1)–N(7)	87.59(11)
Zn(1)–S(2)	2.3229(10)	N(2)–Zn(1)–N(7)	100.01(12)
Zn(1)–O(2)	1.975(3)	O(2)–Zn(1)–O(6)	88.78(11)
Zn(1)–O(6)	2.179(3)	N(2)–Zn(1)–O(6)	93.34(11)
Zn(2)–O(3)	1.975(3)	N(7)–Zn(1)–O(6)	166.53(11)
Zn(2)–O(5)	2.120(3)	O(2)–Zn(1)–S(2)	139.09(8)

Zn(2)–S(3)	2.3551(10)	N(2)–Zn(1)–S(2)	107.44(9)
Zn(2)–N(8)	2.091(3)	N(7)–Zn(1)–S(2)	82.40(8)
Zn(2)–N(6)	2.130(3)	O(6)–Zn(1)–S(2)	91.94(8)
Zn(3)–O(4)	2.113(3)	O(3)–Zn(2)–N(8)	115.14(11)
Zn(3)–O(1)	1.983(3)	O(3)–Zn(2)–O(5)	90.43(11)
Zn(3)–S(1)	2.3545(10)	N(8)–Zn(2)–O(5)	92.79(11)
Zn(3)–N(1)	2.094(3)	O(3)–Zn(2)–N(6)	87.53(11)
Zn(3)–N(5)	2.096(3)	N(8)–Zn(2)–N(6)	98.49(12)
N(7)–N(8)	1.396(4)	O(3)–Zn(2)–S(3)	135.12(8)
N(6)–N(5)	1.395(4)	N(8)–Zn(2)–S(3)	109.31(9)
N(1)–N(2)	1.396(4)	O(5)–Zn(2)–S(3)	93.38(8)
S(1)–C(8)	1.740(4)	N(6)–Zn(2)–S(3)	80.07(8)
S(2)–C(36)	1.737(4)	O(5)–Zn(2)–N(6)	168.32(11)
S(3)–C(22)	1.744(4)	O(1)–Zn(3)–N(1)	87.19(11)
C(8)–N(2)	1.336(5)	O(1)–Zn(3)–N(5)	116.55(12)
N(5)–C(22)	1.337(5)	N(1)–Zn(3)–N(5)	97.96(12)
N(8)–C(36)	1.345(5)	O(1)–Zn(3)–O(4)	89.93(10)

Table S2. Interaction parameters of the complex $[\text{Zn}_3(\mu\text{-L})_3\cdot 3\text{DMF}]$ (**1**).

Hydrogen bonding interactions				
D–H \cdots A	D–H (Å)	H \cdots A (Å)	D–A (Å)	D–H \cdots A (°)
N3–H3A \cdots O6	0.84(4)	2.17(4)	2.959(5)	156(4)
N4–H4A \cdots O4	0.84(5)	2.46(6)	3.316(5)	138(5)
N9–H9A \cdots O5	0.85(3)	2.16(3)	2.957(5)	157(4)
C7–H7 \cdots O2	0.95	2.40	3.294(4)	157
C9–H9 \cdots S1	1.00	2.65	3.060(4)	105
C21–H21 \cdots O1	0.95	2.50	3.348(5)	149
C35–H35 \cdots O3	0.95	2.46	3.352(5)	157
C42–H42 ^B \cdots S2	0.99	2.76	3.365(4)	120
C43–H43 \cdots O2	0.95	2.33	2.942(5)	122
C44–H44A \cdots I1 ^a	0.98	2.86	3.837(4)	177
C44–H44B \cdots I6 ^a	0.98	2.86	3.819(4)	166
C46–H46 \cdots O3	0.95	2.48	3.042(5)	118
C49–H49 \cdots O1	0.95	2.33	2.935(5)	121

D = donor, A = acceptor, Equivalent position codes a = x,1-y,2-z.

C–X \cdots π interactions			
C–X \cdots Cg	X \cdots Cg (Å)	C \cdots Cg (Å)	C–X \cdots Cg (°)
C5–H5 \cdots Cg(11) ^a	2.78	3.629(4)	150
C15–H15 \cdots Cg(7) ^a	2.87	3.594(4)	134
C29–H29 \cdots Cg(9) ^a	2.95	3.742(4)	141

Equivalent position codes: a= x,y,z; Cg(7)= C(1), C(2), C(3), C(4), C(5), C(6); Cg(9)= C(15), C(16), C(17), C(18), C(19), C(20); Cg(11)= C(29), C(30), C(31), C(32), C(33), C(34).

Table S3. The Frontier molecular orbital energies and calculated chemical reactivity parameters for the ligand and complex.

Energy parameters (eV)	H₂L	H₂L-keto	1
HOMO	-5.70	-6.04	-5.45
HOMO-1	-5.95	-6.25	-5.95
LUMO	-2.14	-3.32	-2.87
LUMO+1	-1.48	-1.36	-1.48
E _{HOMO} -E _{LUMO} : ΔE	3.55	2.72	2.58
Ionization Energy, I	5.70	6.04	5.45
Electron Affinity, A	2.14	3.32	2.87
Chemical hardness, η	1.78	1.36	1.29
Electronegativity, χ	3.92	4.68	4.16
Chemical potential, μ	-3.92	-4.68	-4.16
Electrophilicity, ω	4.31	8.05	6.70
Total Energy	-32772.12	-32804.77	-123877.59
Global softness, σ (eV ⁻¹)	0.56	0.73	0.77
Nucleophilicity, ε (eV ⁻¹)	0.23	0.12	0.12
Dipole moment (Debye)	8.20	4.99	12.07

Table S4. Results of the inhibitory zones of the ligand and complex **1**.

Compounds	Zone of Inhibition (mm)				
	<i>S. aureus</i>	<i>E. coli</i>	<i>S. typhi</i>	<i>K. pneumoniae</i>	<i>P. mirabilis</i>
H₂L	17±1	11±1	0	11.33±1.52	0
1	33±1.73	23±1	26.67±1.53	23±1	20.67±1.53
Ciprofloxacin	27.33±0.57	29.67±0.57	27.67±1.15	32±1	29.33±1.52

Table S5. The percentage of cell viability of MCF-7 cell with the concentrations of ligand H₂L and complex **1**.

Samples	Concentration (μg/mL)	Average Absorbance @	% Cell Viability
		540 nm	MCF-7
H₂L	Control	0.5766	100
	6.25	0.5561	96.44
	12.5	0.4686	81.26
	25	0.3782	65.59
	50	0.3098	53.73
	100	0.2467	42.79
	6.25	0.4800	83.25
	12.5	0.48087	70.88

Complex 1	25	0.2373	41.15
	50	0.1772	30.73
	100	0.1096	19.01

Table S6. Physicochemical properties and in silico ADMET profile of **1**.

Properties	Complex 1
Physicochemical and medicinal properties	
MW	1996.84
TPSA	581.945
LogP _{o/w}	11.064
Absorption	
Caco-2 Permeability	0.78
HIA	85.25
P-gp substrate	0.869
Distribution	
BBB	-2.923
FU	0.394
Metabolism	
CYP3A4 inhibitor	1(yes)
Excretion	
CL _{plasma}	-34.217
Toxicity	
Human hepatotoxicity (H-HT)	No
<p>Note. MW: molecular weight (≤ 500 g/mol), TPSA: topological polar surface area (\AA^2), LogP: logarithm of partition coefficient (≤ 5) of compound between n-octanol and water, HIA: Human intestinal absorption ($\text{HIA} + < 30\% < \text{HIA}$-), P-gp: P-glycoprotein inhibitor/ substrate (0 to 0.3, excellent; 0.3 to 0.7, medium; 0.7 to 1.0, poor), BBB: Blood-Brain Barrier penetration (~ 1: BBB+; ~ 0: BBB-), FU: Fraction unbound in plasmas (Low; $< 5\%$, Middle; $5 \sim 20\%$, High; $> 20\%$), CYP3A4 inhibitor: Cytochrome P450 monooxygenase enzyme (~ 1; inhibitor, ~ 0; non-inhibitor), CL_{plasma}: ml/min/kg. > 15 ml/min/kg; high clearance, $5 \sim 15$ ml/min/kg; moderate clearance, < 5 ml/min/kg; low clearance, H-HT: Human hepatotoxicity (0 to 0.3, low; 0.3 to 0.7, medium; 0.7 to 1.0, high).</p>	