

Supplementary Material

Table S1. a/ Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cu}(\text{Me}_2\text{SO})_6(\text{ClO}_4)_2 \cdot 2\text{Me}_2\text{SO}$. U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor, b/ anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cu}(\text{OSMe}_2)_6(\text{ClO}_4)_2 \cdot 2\text{Me}_2\text{SO}$. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$ and c/ hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cu}(\text{OSMe}_2)_6(\text{ClO}_4)_2 \cdot 2\text{Me}_2\text{SO}$

Table S2. Interatomic bond distances (\AA) and angles ($^\circ$) in the structure of hexakis(dimethylsulfoxide)copper(II) perchlorate didimethylsulfoxide.

Table S3. Fractional atomic coordinates and equivalent isotropic displacement coefficients (\AA^2) for the structure of hexakis(dimethylsulfoxide) tetrakis(dimethylsulfoxide)-bishydroxocopper(II) perchlorate.

Table S4. Interatomic bond distances (\AA) and angles ($^\circ$) in the structure of tetrakis(dimethylsulfoxide)bishydroxocopper(II) perchlorate.

Figure S1. The Fourier transforms and the fit of solid hexaaquacopper(II) bromate (offset 10.5), solid hexaaquacopper(II) hexafluorosilicate (offset 9.0), an aqueous solution of copper(II) perchlorate (offset 7.5), solid hexakis(pyridine-1-oxide)copper(II) perchlorate (offset 6.0), solid tetrakis(pyridine-1-oxide)copper(II) perchlorate (offset 4.5), solid hexakis(dimethylsulfoxide)copper(II) perchlorate (offset 3.0), dimethylsulfoxide solution of copper(II) perchlorate (offset 1.5) and *N,N'*-dimethylpropyleneurea solution of copper(II) trifluoromethanesulfonate.

Figure S2. The fit and the individual contribution of the different scattering paths of the EXAFS data of

a/ solid hexaaquacopper(II) bromate, solid line - experimental data and dashed line - model. Individual contributions: Cu- O_{eq} single scattering - solid line (offset -4), Cu- O_{ax} single scattering - dashed line (offset -10), Cu- O_{eq} - O_{eq} linear multiple scattering - dotted line (offset -12), Cu- O_{eq} -Cu- O_{eq} linear multiple scattering - dash-dotted line (offset -14) and Cu- O_{eq} -Cu- O_{eq} linear multiple scattering - dash-dotted line (offset -16).

b/ solid hexaaquacopper(II) hexafluorosilicate, solid line - experimental data and dashed line - model. Individual contributions: Cu- O_{eq} single scattering - solid line (offset -4), Cu- O_{ax} single scattering - dashed line (offset -10), Cu- O_{eq} - O_{eq} linear multiple scattering - dotted line (offset -12), Cu- O_{eq} -Cu- O_{eq} linear multiple scattering - dash-dotted line (offset -14) and Cu- O_{eq} -Cu- O_{eq} linear multiple scattering - dash-dotted line (offset -16).

c/ aqueous solution of copper(II) perchlorate, solid line - experimental data and dashed line - model. Individual contributions: Cu- O_{eq} single scattering - solid line (offset -4), Cu- O_{ax} single scattering - dashed line (offset -10), Cu- O_{eq} - O_{eq} linear multiple scattering - dotted

line (offset -12), Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -14) and Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -16).

d/ solid hexakis(pyridine-1-oxide)copper(II) perchlorate, solid line - experimental data and dashed line - model. Individual contributions: Cu-O_{eq} single scattering - solid line (offset -4), Cu···N_{eq} single scattering - dashed line (offset -10), Cu-O_{eq}-N_{eq} three-leg scattering path - dotted line (offset -12), Cu-O_{eq}-O_{eq} linear multiple scattering - dotted line (offset -14), Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -16) and Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -18).

e/ solid tetrakis(pyridine-1-oxide)copper(II) perchlorate, solid line - experimental data and dashed line - model. Individual contributions: Cu-O_{eq} single scattering - solid line (offset -4), Cu···N_{eq} single scattering - dashed line (offset -10), Cu-O_{eq}-N_{eq} three-leg scattering path - dotted line (offset -12), Cu-O_{eq}-O_{eq} linear multiple scattering - dotted line (offset -14), Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -16) and Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -18).

f/ solid hexakis(dimethylsulfoxide)copper(II) perchlorate, solid line - experimental data and dashed line - model. Individual contributions: Cu-O_{eq} single scattering - solid line (offset -4), Cu-O_{ax} single scattering - dashed line (offset -12), Cu···S_{eq} single scattering - dotted line (offset -16), Cu-O_{eq}-S_{eq} three-leg scattering path - dash-dotted line (offset -20), Cu-O_{eq}-O_{eq} linear multiple scattering - dotted line (offset -22), Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -24) and Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -26).

g/ dimethylsulfoxide solution of copper(II) perchlorate, solid line - experimental data and dashed line - model. Individual contributions: Cu-O_{eq} single scattering - solid line (offset -4), Cu···S_{eq} single scattering - dotted line (offset -12), Cu-O_{eq}-S_{eq} three-leg scattering path - dash-dotted line (offset -16), Cu-O_{eq}-O_{eq} linear multiple scattering - dotted line (offset -18), Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -20) and Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -22).

h/ *N,N'*-dimethylpropyleneurea solution of copper(II) trifluoromethanesulfonate, solid line - experimental data and dashed line - model. Individual contributions: Cu-O single scattering - solid line (offset -4), Cu···C single scattering - dotted line (offset -10), Cu-O_{eq}-O_{eq} linear multiple scattering - dotted line (offset -12), Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -14) and Cu-O_{eq}-Cu-O_{eq} linear multiple scattering - dash-dotted line (offset -16).

Figure S3. Normalized absorption edges of the studied samples. A help line is inserted at 8998 eV.

Table S1a. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cu(OSMe}_2)_6(\text{ClO}_4)_2 \cdot 2\text{Me}_2\text{SO}$. U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	<i>x</i>	<i>y</i>	<i>z</i>	U_{eq}
Cu(1)	5000	0	0	49(1)
S(2)	2915(2)	-60(2)	2758(2)	82(1)
S(3)	6521(2)	72(2)	1935(2)	62(1)
S(1)	5201(2)	-3192(2)	1410(2)	68(1)
S(4)	822(3)	-4063(2)	1158(2)	97(1)
Cl(1)	2841(2)	2757(2)	4397(2)	81(1)
O(1)	6036(4)	-1719(4)	556(4)	60(1)
O(2)	3329(5)	-941(4)	1991(4)	76(1)
O(3)	6231(4)	990(4)	615(4)	65(1)
O(4)	-157(6)	-3158(5)	445(5)	112(2)
O(13)	3223(7)	3429(6)	3005(5)	132(2)
O(12)	4044(9)	2242(8)	4772(7)	187(3)
O(11)	2441(10)	3725(9)	4892(8)	194(3)
O(14)	1760(10)	1678(8)	4953(7)	204(4)
C(1A)	6262(10)	-4181(7)	602(8)	118(3)
C(1B)	5665(10)	-3870(7)	2885(7)	108(3)
C(2A)	2269(10)	-1322(9)	4455(7)	127(3)
C(2B)	1269(8)	581(8)	2570(8)	102(3)
C(3A)	6823(9)	1298(7)	2518(7)	95(2)
C(3B)	8317(7)	-359(7)	1461(7)	80(2)
C(4A)	-168(10)	-5695(8)	2269(10)	140(4)
C(4B)	933(9)	-3492(8)	2351(8)	102(3)

Table 1b. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cu(OSMe}_2)_6(\text{ClO}_4)_2 \cdot 2\text{Me}_2\text{SO}$. The anisotropic displacement factor exponent takes the form: $-2\pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$

	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Cu(1)	41(1)	56(1)	46(1)	-20(1)	-15(1)	8(1)
S(2)	56(1)	112(2)	90(1)	-60(1)	-16(1)	3(1)
S(3)	50(1)	76(1)	50(1)	-20(1)	-16(1)	5(1)
S(1)	54(1)	64(1)	77(1)	-23(1)	-23(1)	4(1)
S(4)	96(2)	86(1)	103(2)	-37(1)	-32(1)	13(1)
Cl(1)	79(2)	83(1)	64(1)	-21(1)	-18(1)	7(1)
O(1)	43(3)	65(3)	69(3)	-23(2)	-22(2)	8(2)
O(2)	76(3)	73(3)	80(3)	-39(2)	-20(3)	5(2)
O(3)	63(3)	71(3)	53(3)	-19(2)	-22(2)	9(2)
O(4)	127(5)	80(3)	122(4)	-16(3)	-71(4)	17(3)

O(13)	149(6)	143(5)	65(4)	-18(3)	-28(4)	9(4)
O(12)	148(7)	234(8)	154(6)	-53(5)	-76(6)	97(6)
O(11)	222(9)	200(7)	182(7)	-126(6)	-40(7)	58(7)
O(14)	190(9)	207(7)	144(6)	-32(5)	-13(6)	-118(7)
C(1A)	135(8)	77(5)	133(7)	-58(5)	-14(6)	5(5)
C(1B)	147(9)	79(5)	86(6)	-15(4)	-52(6)	-3(5)
C(2A)	135(9)	167(8)	80(6)	-49(6)	-43(6)	25(7)
C(2B)	63(6)	137(7)	109(6)	-60(5)	-26(5)	25(5)
C(3A)	103(7)	111(6)	85(5)	-53(5)	-34(5)	21(5)
C(3B)	59(5)	82(5)	93(5)	-31(4)	-28(4)	13(4)
C(4A)	135(9)	91(6)	175(9)	-13(6)	-91(8)	-6(6)
C(4B)	101(7)	102(5)	122(7)	-59(5)	-51(6)	25(5)

Table 1c. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cu}(\text{OSMe}_2)_6(\text{ClO}_4)_2 \cdot 2\text{Me}_2\text{SO}$.

	<i>x</i>	<i>y</i>	<i>z</i>	U_{eq}
H(1AA)	6119	-3922	-254	177
H(1AB)	5971	-5174	1171	177
H(1AC)	7268	-3980	454	177
H(1BA)	5166	-3429	3448	162
H(1BB)	6692	-3682	2654	162
H(1BC)	5392	-4874	3369	162
H(2AA)	3061	-1785	4726	191
H(2AB)	1574	-2010	4556	191
H(2AC)	1816	-865	5015	191
H(2BA)	1441	1308	1661	153
H(2BB)	854	964	3203	153
H(2BC)	612	-179	2743	153
H(3AA)	5925	1650	2822	142
H(3AB)	7508	2072	1791	142
H(3AC)	7199	839	3257	142
H(3BA)	8348	-1038	1111	120
H(3BB)	8636	-753	2241	120
H(3BC)	8942	481	775	120
H(4AD)	-327	-6159	1762	210
H(4AE)	-1084	-5564	2806	210
H(4AF)	358	-6265	2855	210
H(4BA)	1467	-2570	1884	152
H(4BB)	1417	-4146	2934	152
H(4BC)	-26	-3447	2886	152

Table S2. Bond lengths [Å] and angles [°] for Cu(OSMe₂)₆(ClO₄)₂·2Me₂SO.

Cu(1)-O(1)	1.990(4)
Cu(1)-O(1)#1	1.990(4)
Cu(1)-O(3)	2.083(4)
Cu(1)-O(3)#1	2.083(4)
Cu(1)-O(2)#1	2.182(4)
Cu(1)-O(2)	2.182(4)
S(2)-O(2)	1.507(4)
S(2)-C(2A)	1.746(8)
S(2)-C(2B)	1.757(7)
S(3)-O(3)	1.519(4)
S(3)-C(3B)	1.769(6)
S(3)-C(3A)	1.770(7)
S(1)-O(1)	1.527(4)
S(1)-C(1B)	1.738(7)
S(1)-C(1A)	1.783(7)
S(4)-O(4)	1.496(5)
S(4)-C(4A)	1.732(7)
S(4)-C(4B)	1.760(7)
Cl(1)-O(11)	1.362(7)
Cl(1)-O(14)	1.368(7)
Cl(1)-O(13)	1.375(5)
Cl(1)-O(12)	1.386(7)
O(1)-Cu(1)-O(1)#1	180.0(2)
O(1)-Cu(1)-O(3)	90.97(15)
O(1)#1-Cu(1)-O(3)	89.03(15)
O(1)-Cu(1)-O(3)#1	89.03(15)
O(1)#1-Cu(1)-O(3)#1	90.97(15)
O(3)-Cu(1)-O(3)#1	180.00(18)
O(1)-Cu(1)-O(2)#1	90.32(15)
O(1)#1-Cu(1)-O(2)#1	89.68(15)
O(3)-Cu(1)-O(2)#1	86.83(16)
O(3)#1-Cu(1)-O(2)#1	93.17(16)
O(1)-Cu(1)-O(2)	89.68(15)
O(1)#1-Cu(1)-O(2)	90.32(15)
O(3)-Cu(1)-O(2)	93.17(16)
O(3)#1-Cu(1)-O(2)	86.83(16)
O(2)#1-Cu(1)-O(2)	180.0(2)
O(2)-S(2)-C(2A)	103.4(3)
O(2)-S(2)-C(2B)	106.8(3)
C(2A)-S(2)-C(2B)	98.9(4)
O(3)-S(3)-C(3B)	106.0(3)
O(3)-S(3)-C(3A)	103.7(3)
C(3B)-S(3)-C(3A)	98.6(4)

O(1)-S(1)-C(1B)	104.5(3)
O(1)-S(1)-C(1A)	101.9(3)
C(1B)-S(1)-C(1A)	99.2(4)
O(4)-S(4)-C(4A)	106.3(4)
O(4)-S(4)-C(4B)	107.0(3)
C(4A)-S(4)-C(4B)	99.0(4)
O(11)-Cl(1)-O(14)	111.2(6)
O(11)-Cl(1)-O(13)	109.1(4)
O(14)-Cl(1)-O(13)	111.4(4)
O(11)-Cl(1)-O(12)	105.0(5)
O(14)-Cl(1)-O(12)	110.1(5)
O(13)-Cl(1)-O(12)	109.7(4)
S(1)-O(1)-Cu(1)	120.9(2)
S(2)-O(2)-Cu(1)	117.8(2)
S(3)-O(3)-Cu(1)	116.4(2)

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y,-z