

Tuning facial-meridional isomerisation in monometallic nine-co-ordinate lanthanide complexes with unsymmetrical tridentate ligands.

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Supporting Information

(5 Pages)

Table S1 Lifetimes (τ) of the ligand-centred $^3\pi\pi^*$, and Eu(5D_0) excited level in $[\text{Eu}(\mathbf{L2})_3]^{3+}$ and Tb(5D_4) in $[\text{Tb}(\mathbf{L2})_3]^{3+}$ in acetonitrile.

Compound	Conc / mol·dm ⁻³	T /K	$\tilde{\nu}_{\text{ex}} / \text{cm}^{-1}$	τ/ms
L2	10 ⁻⁵	77	31940	≥ 100
$[\text{Lu}(\mathbf{L2})_3]^{3+}$	6·10 ⁻⁴	77	39525	130(15)
$[\text{Eu}(\mathbf{L2})_3]^{3+}$	8·10 ⁻⁴	295	37040	2.25(2)
$[\text{Eu}(\mathbf{L2})_3]^{3+}$	8·10 ⁻⁴	77	39060	1.86(7)
$[\text{Tb}(\mathbf{L2})_3]^{3+}$	8·10 ⁻⁴	295	26109	0.029(1)
$[\text{Tb}(\mathbf{L2})_3]^{3+}$	8·10 ⁻⁴	77	39215	1.22(4)

Table S2 Elemental analyses for the $[\text{Ln}(\mathbf{L2})_3](\text{ClO}_4)_3 \cdot x\text{H}_2\text{O}$ (Ln = Eu, $x = 4$: **1**; Ln = Tb, $x = 5.5$: **2**; Ln = Lu, $x = 4$: **3**) and $[\text{Ln}(\mathbf{L4})_3](\text{CF}_3\text{SO}_3)_3 \cdot x\text{C}_7\text{H}_{15}$ (Ln = Eu, $x = 0$: **4**; Ln = Lu, $x = 0.4$: **5**). Calculated values are given in parentheses.

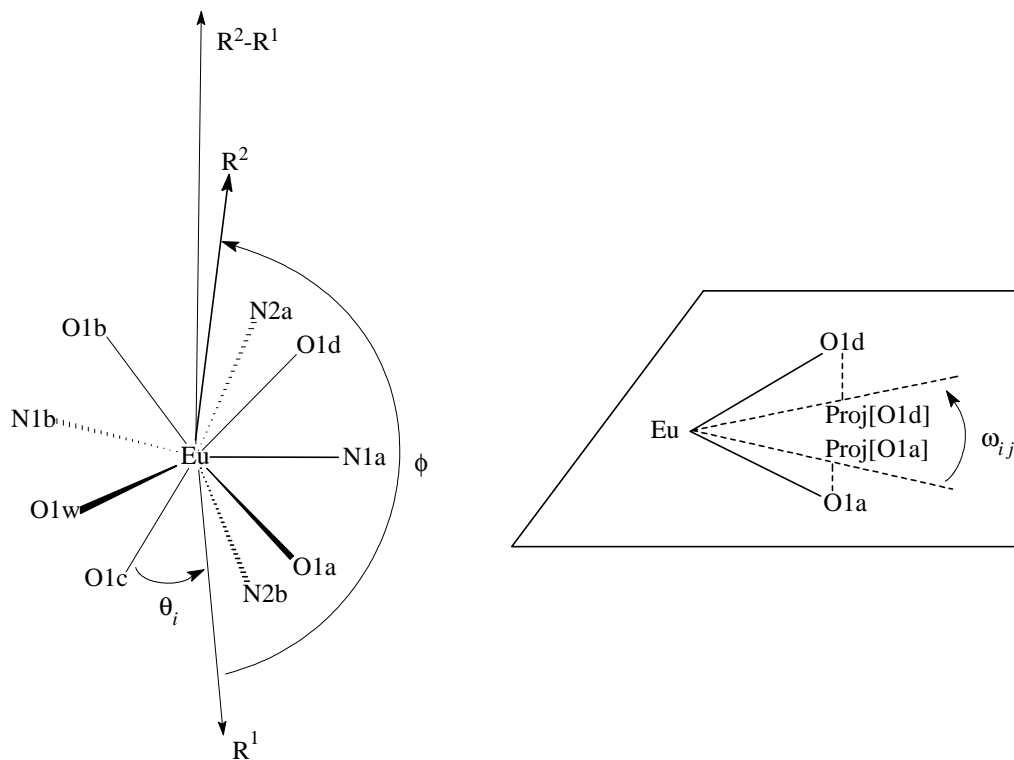
Compound	% C	% H	% N
$[\text{Eu}(\mathbf{L2})_3](\text{ClO}_4)_3 \cdot 4\text{H}_2\text{O}$ (1)	46.05 (45.96)	4.79 (5.01)	11.20 (11.28)
$[\text{Tb}(\mathbf{L2})_3](\text{ClO}_4)_3 \cdot 5.5\text{H}_2\text{O}$ (2)	44.82 (44.94)	4.59 (5.09)	10.69 (11.03)
$[\text{Lu}(\mathbf{L2})_3](\text{ClO}_4)_3 \cdot 4\text{H}_2\text{O}$ (3)	45.30 (45.26)	4.68 (4.93)	11.01 (11.11)
$[\text{Eu}(\mathbf{L4})_3](\text{CF}_3\text{SO}_3)_3$ (4)	46.55 (46.60)	5.05 (4.62)	9.79 (9.79)
$[\text{Lu}(\mathbf{L4})_3](\text{CF}_3\text{SO}_3)_3 \cdot 0.4\text{C}_7\text{H}_{15}$ (5)	46.93 (46.97)	4.83 (5.08)	9.53 (9.93)

Table S3 Selected structural data for the lanthanide coordination spheres in [Eu(L1)₂(CF₃SO₃)₂(H₂O)](CF₃SO₃)(C₃H₅N)₂(H₂O) (**6**) and [Eu(L4)₂(CF₃SO₃)₂(H₂O)](CF₃SO₃)(THF)_{1.5} (**7**) analysed as tricapped trigonal prims TTP).

		Angles $\phi^a / ^\circ$	
Compd	6	7	Perfect TTP ^b
R ¹ -Eu-R ²	170.9	170.7	180
Angles $\theta_t^a / ^\circ$ (distal tetrapodes)			
Compd	6	7	Perfect TTP ^b
R ¹ -Eu-N2b	45.0	44.7	α
R ¹ -Eu-O1c	50.3	52.4	α
R ¹ -Eu-O1a	45.3	43.6	α
R ² -Eu-N2a	48.6	46.5	α
R ² -Eu-O1d	44.0	42.8	α
R ² -Eu-O1b	43.9	44.0	α
Angles $\omega_{ij}^a / ^\circ$			
Compd	6	7	Perfect TTP ^b
Proj[N2b]-Eu-Proj[O1c] ^c	133.0	135.0	120
Proj[O1c]-Eu-Proj[O1a]	115.1	116.8	120
Proj[O1a]-Eu-Proj[N2b]	111.9	108.2	120
Proj[N1b]-Eu-Proj[O1w]	117.3	115.9	120
Proj[O1w]-Eu-Proj[N1a]	130.0	132.0	120
Proj[N1a]-Eu-Proj[N1b]	112.8	112.0	120
Proj[O1d]-Eu-Proj[N2a]	133.3	134.5	120
Proj[N2a]-Eu-Proj[O1b]	117.5	113.5	120
Proj[O1b]-Eu-Proj[O1d]	109.2	111.9	120
Proj[O1a]-Eu-Proj[O1b]	11.5	9.2	0
Proj[N2b]-Eu-Proj[O1d]	8.8	12.8	0
Proj[O1c]-Eu-Proj[N2a]	9.1	12.4	0
Proj[O1a]-Eu-Proj[O1w]	179.0	177.1	180
Proj[N2b]-Eu-Proj[N1a]	162.9	159.1	180
Proj[O1c]-Eu-Proj[N1b]	176.8	178.0	180
Proj[N1a]-Eu-Proj[O1d]	171.7	171.9	180

Proj[N1b]-Eu-Proj[N2a]	167.7	165.6	180
Proj[O1w]-Eu-Proj[O1b]	167.5	168.1	180

^a For the definition of ϕ , θ_i and ω_{ij} , see the scheme below. The error in the angles is typically 0.5° . ^b TTP = tricapped trigonal prism. ^c Proj[O_i] and Proj[N_i] are the projections of O_i and respectively N_i along the R^2 - R^1 direction onto a perpendicular plane passing through the lanthanide atom. $R^1 = \text{Eu-N2b} + \text{Eu-O1c} + \text{Eu-O1a}$ and $R^2 = \text{Eu-N2a} + \text{Eu-O1b} + \text{Eu-O1d}$.



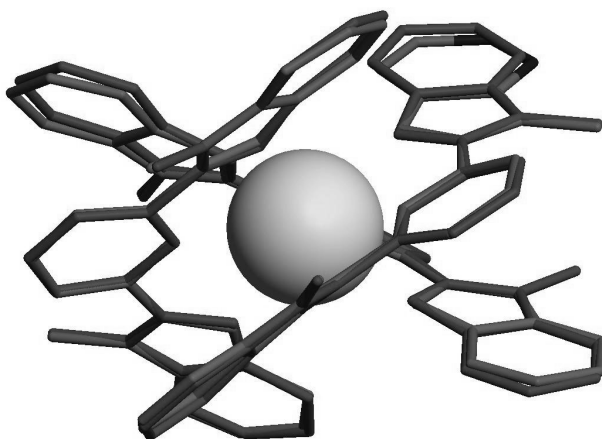


Figure S1 Perspective view of the optimized superimposition of the two molecular structures of $[\text{Eu}(\text{L5})_3]^{3+}$ obtained by X-ray diffraction in the solid state (in blue) and by MM simulation in the gas-phase (in red) with the parameters reported by Van Veggel and Reinhoudt.³¹

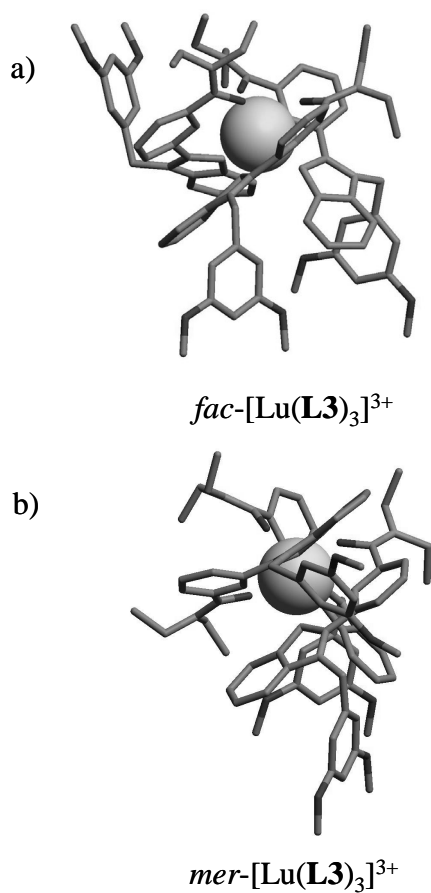


Figure S2 Perspective views of the molecular structures of the two cations a) $\text{fac-}[\text{Lu}(\text{L3})_3]^{3+}$ and b) $\text{mer-}[\text{Lu}(\text{L3})_3]^{3+}$ obtained by molecular mechanic in the gas-phase.

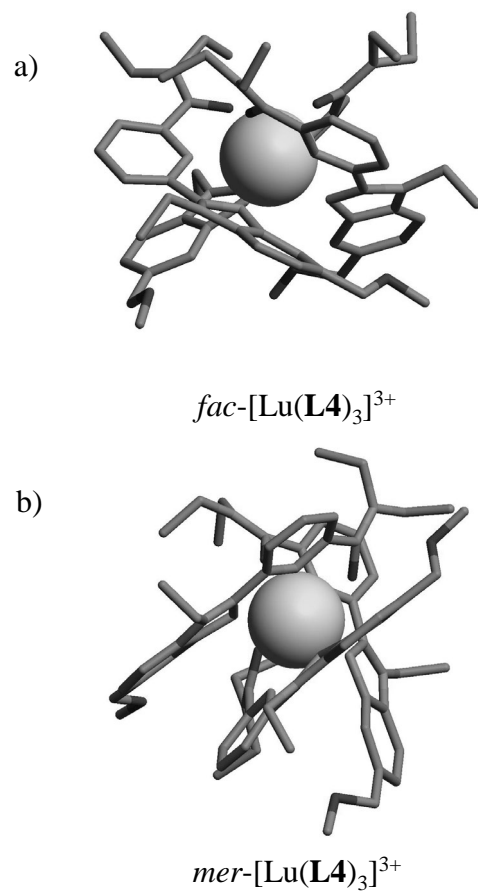


Figure S3 Perspective views of the molecular structures of the two cations a) *fac*-[Lu(**L4**)₃]³⁺ and b) *mer*-[Lu(**L4**)₃]³⁺ obtained by molecular mechanic in the gas-phase.