

## Supplementary Electronic Information

### Synthesis and structural characterization of lanthanide (III) nitrate complexes of a tetraiminodiphenol macrocycle in the solid state and in solution †

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† Electronic supplementary information (ESI) available: COSY spectrum of **1** (**Fig. S1**), <sup>1</sup>H NMR spectrum of **1** at different temperatures (**Fig. S2**), plots of  $\Delta_{ij} / \langle S_z \rangle_j$  vs  $C_j / \langle S_z \rangle_j$  according to equation (4) for the lanthanide complexes (**Fig. S3**), absorption spectrum of [Nd(LH<sub>2</sub>)(NO<sub>3</sub>)<sub>3</sub>] (**4**) (**Fig. S4**) and absorption spectral data for some of the lanthanide complexes (**Table S1**).

See <http://www.rsc.org/suppdata>

**Electronic supplementary information (ESI)**

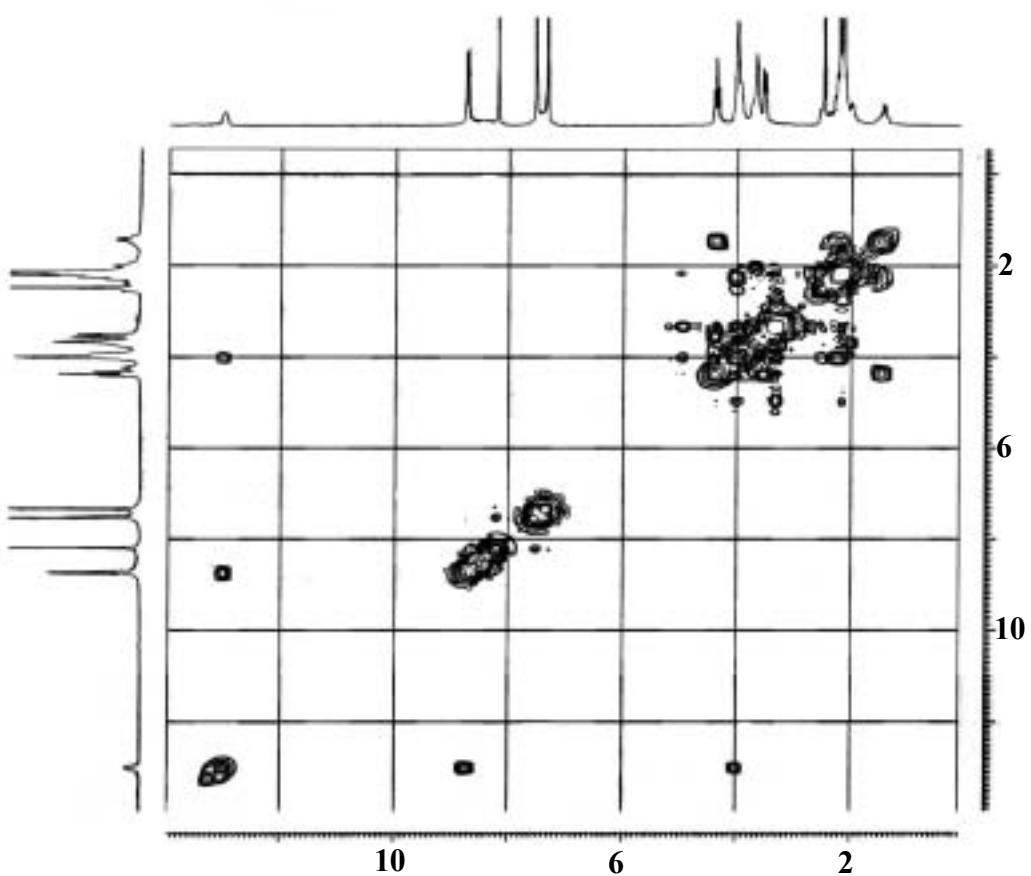
**Fig. S1**  $\{^1\text{H} - ^1\text{H}\}$ COSY spectrum of  $[\text{La}(\text{LH}_2)(\text{NO}_3)_3]$  (**1**) in  $(\text{CD}_3)_2\text{SO}$ .

**Fig. S2**  $^1\text{H}$  NMR spectra of  $[\text{La}(\text{LH}_2)(\text{NO}_3)_3]$  (**1**) in  $(\text{CD}_3)_2\text{SO}$  at different temperatures.

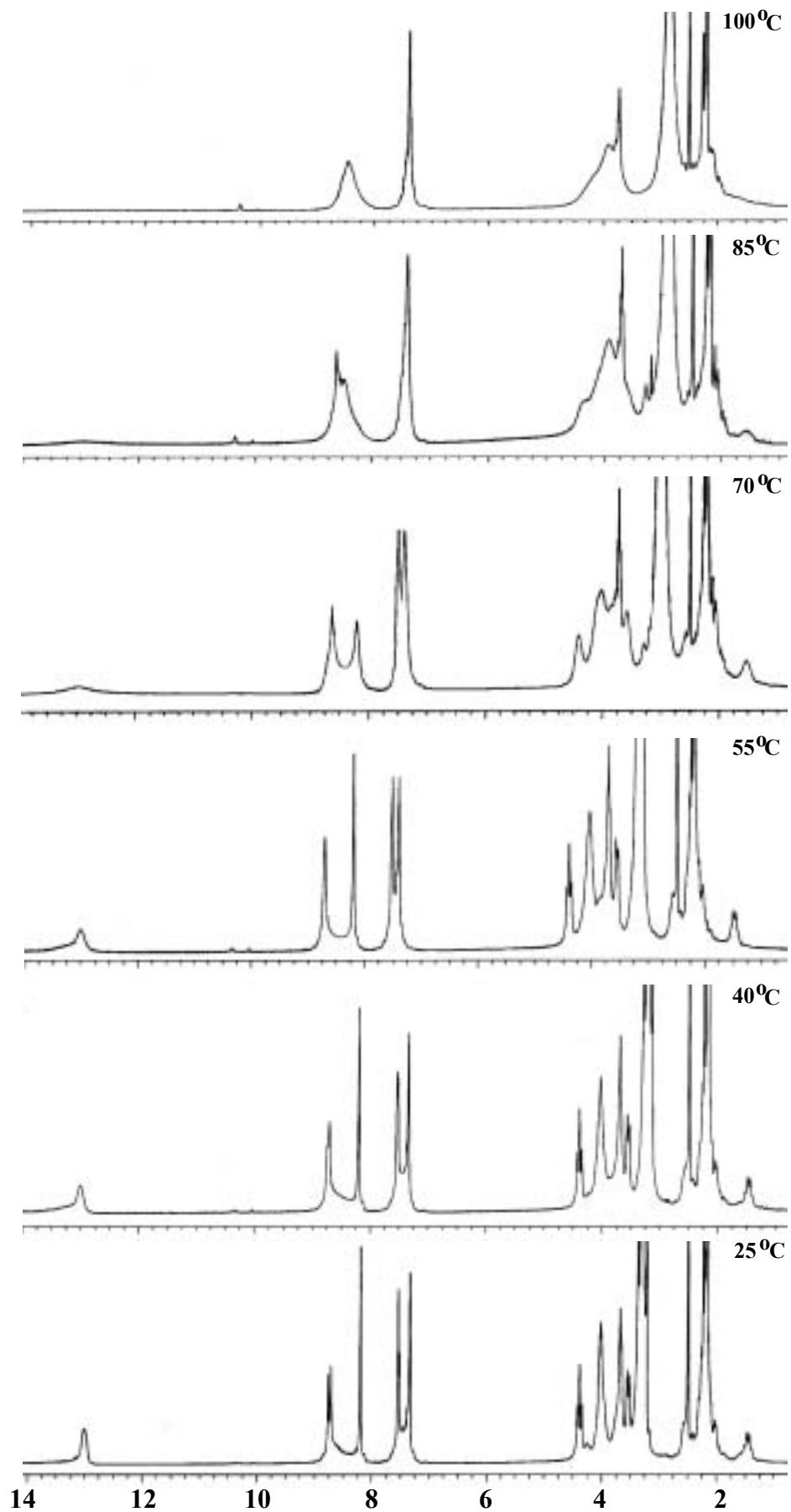
**Fig. S3** Plots of  $\Delta_{ij} / \langle S_z \rangle_j$  vs.  $C_j / \langle S_z \rangle_j$  according to equation (4) for the proton e in  $\text{Ln}(\text{LH}_2)(\text{NO}_3)_3$  compounds.

**Fig. S4** The absorption spectrum of  $[\text{Nd}(\text{LH}_2)(\text{NO}_3)_3]$  (**4**) in *N,N*-dimethylformamide.

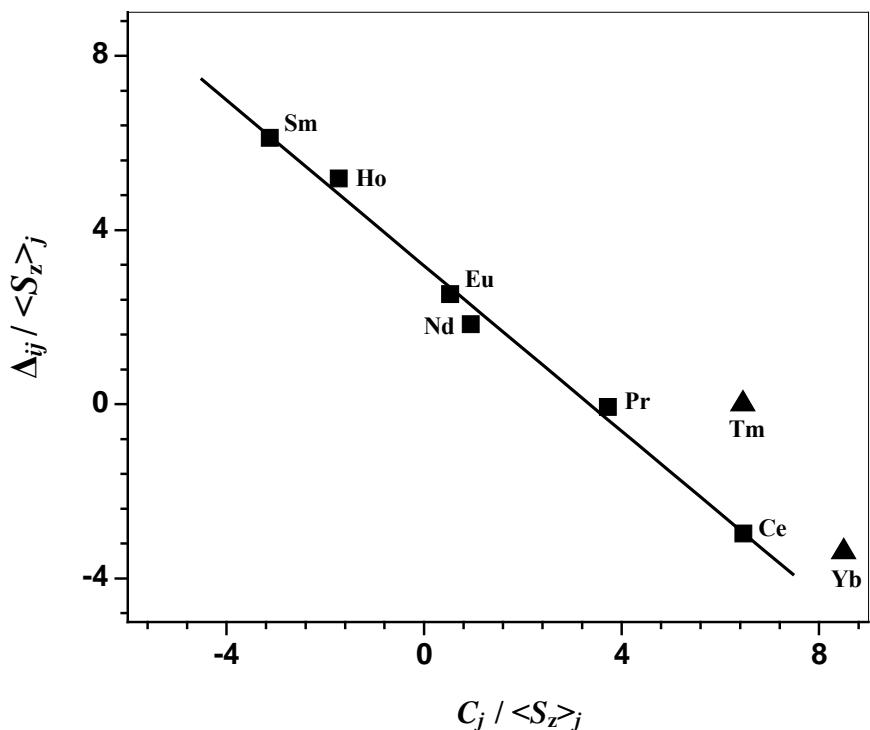
**Table S1** Absorption spectral data for some of the lanthanide (III) complexes.



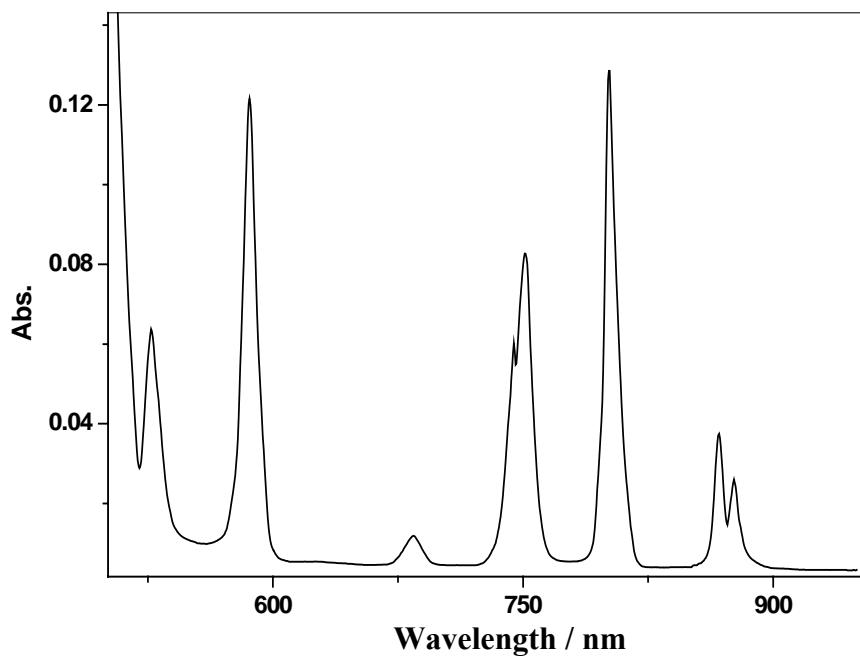
**Fig. S1**  $\{^1\text{H} - ^1\text{H}\}$ COSY spectrum of  $[\text{La}(\text{LH}_2)(\text{NO}_3)_3]$  (**1**) in  $(\text{CD}_3)_2\text{SO}$ .



**Fig. S2**  $^1\text{H}$  NMR spectra of  $[\text{La}(\text{LH}_2)(\text{NO}_3)_3]$  (1) in  $(\text{CD}_3)_2\text{SO}$  at different temperatures.



**Fig. S3** Plots of  $\Delta_{ij} / \langle S_z \rangle_j$  vs.  $C_j / \langle S_z \rangle_j$  according to equation (4) for the proton e in  $\text{Ln}(\text{LH}_2)(\text{NO}_3)_3$  compounds.



**Fig. S4** The absorption spectrum of  $[\text{Nd}(\text{LH}_2)(\text{NO}_3)_3]$  (4) in N,N-dimethylformamide

**Table S1** Absorption spectral data for some of the lanthanide (III) complexes<sup>a</sup>

[Ln(LH <sub>2</sub> )(NO <sub>3</sub> ) <sub>3</sub> ]		
Ln <sup>3+</sup>	λ <sub>max</sub> / nm (ε / dm <sup>3</sup> mol <sup>-1</sup> cm <sup>-1</sup> )	Transition
Pr <b>3</b>	585 (3.5)	<sup>3</sup> H <sub>4</sub> → <sup>1</sup> D <sub>2</sub> <sup>b</sup>
	1000 (1.5)	→ <sup>1</sup> G <sub>4</sub>
Nd <b>4</b>	530 (6.3)	<sup>4</sup> I <sub>9/2</sub> → <sup>4</sup> G <sub>7/2</sub> , <sup>4</sup> G <sub>9/2</sub> <sup>b</sup>
	585 (12.2)	→ <sup>4</sup> G <sub>5/2</sub> , <sup>2</sup> G <sub>7/2</sub> <sup>b</sup>
	685 (1.3)	→ <sup>4</sup> F <sub>9/2</sub>
	738 (7.5)	→ <sup>4</sup> F <sub>7/2</sub>
	802 (12.9)	→ <sup>4</sup> F <sub>5/2</sub>
	867 (3.8)	→ <sup>4</sup> F <sub>3/2</sub>
Ho <b>10</b>	550 (9.8)	<sup>5</sup> I <sub>8</sub> → <sup>5</sup> G <sub>2</sub>
	645 (5.4)	→ <sup>5</sup> G <sub>3</sub>
Er <b>11</b>	520 (13.5)	<sup>4</sup> I <sub>15/2</sub> → <sup>2</sup> H <sub>11/2</sub> , <sup>4</sup> G <sub>11/2</sub> <sup>b</sup>
	630 (3)	→ <sup>4</sup> F <sub>9/2</sub>
	980 (1)	→ <sup>4</sup> I <sub>11/2</sub>
Tm <b>12</b>	688 (7.0)	
	790 (5.4)	<sup>3</sup> H <sub>6</sub> → <sup>3</sup> H <sub>4</sub>

<sup>a</sup> In N,N-dimethylformamide. <sup>b</sup> Hypersensitive transition.