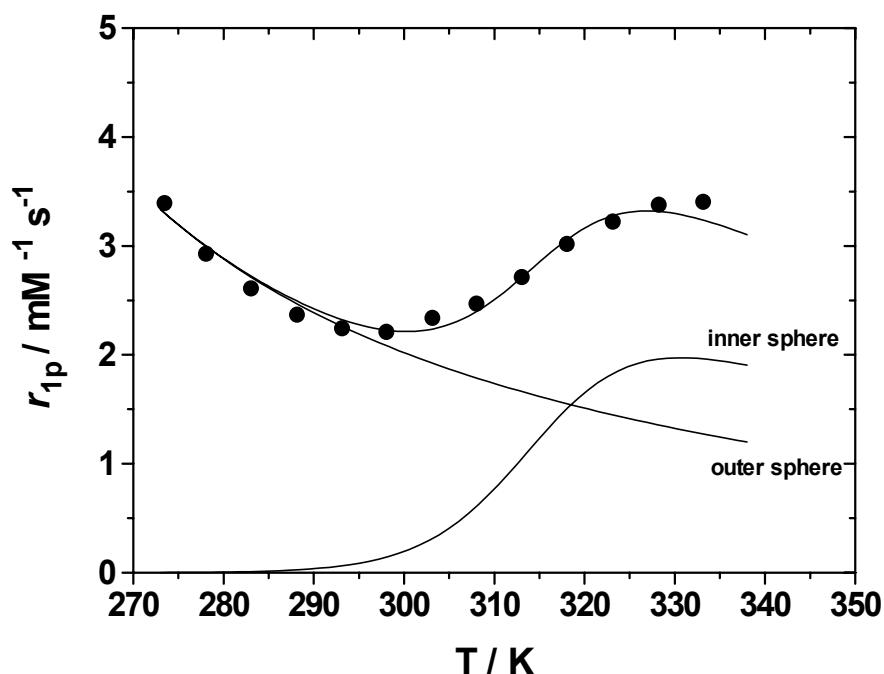


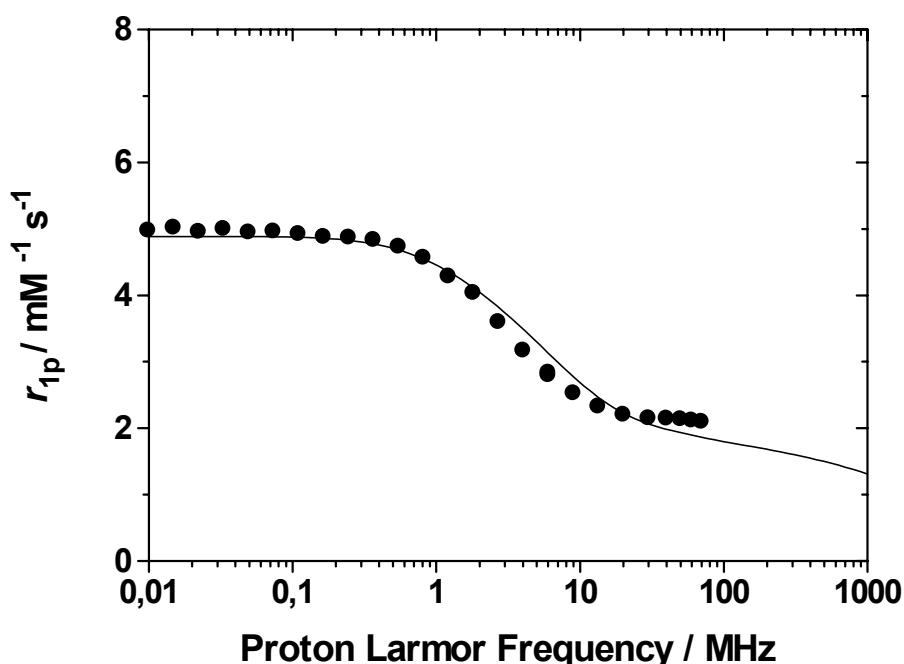
ESI Analyses of the T dependence of the relaxivity and (for selected complexes) the field dependence of the relaxivity for salts of $[\text{Gd.2}]^{3+}$.

Acetate

a) VT ^1H - 20 MHz

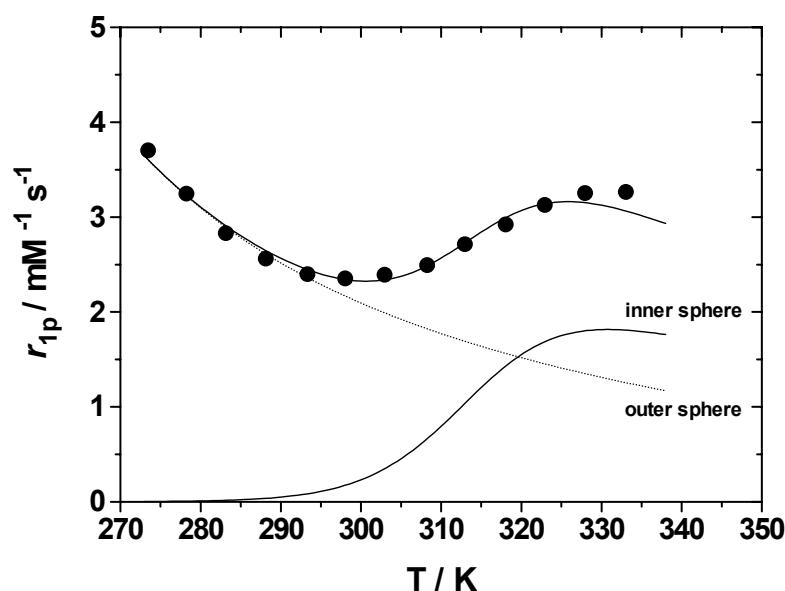


b) NMRD - 25°C, pH 6



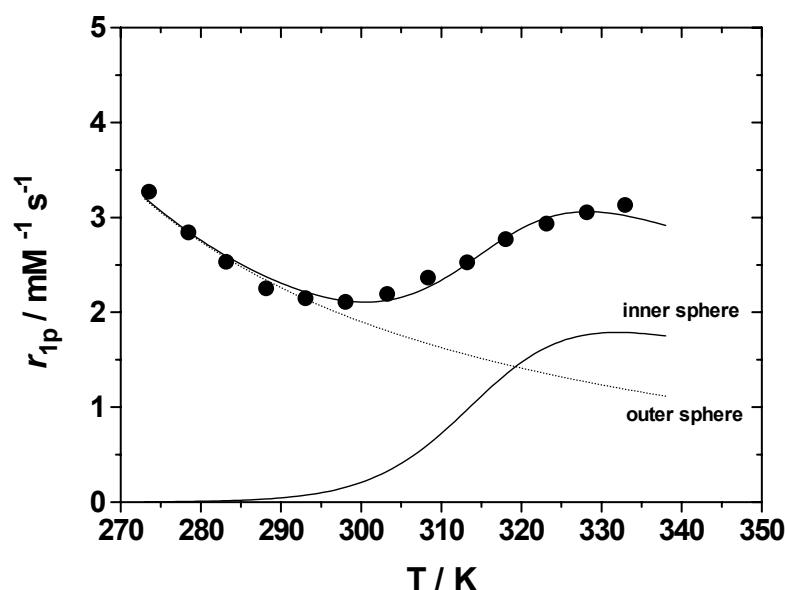
Nitrate

VT ^1H - 20 MHz



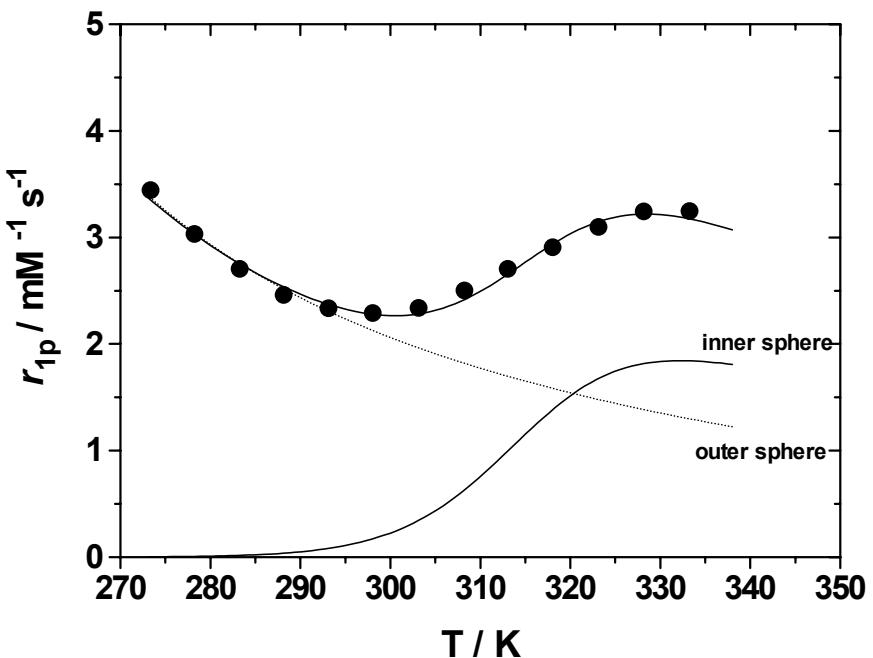
Triflate

VT ^1H - 20 MHz

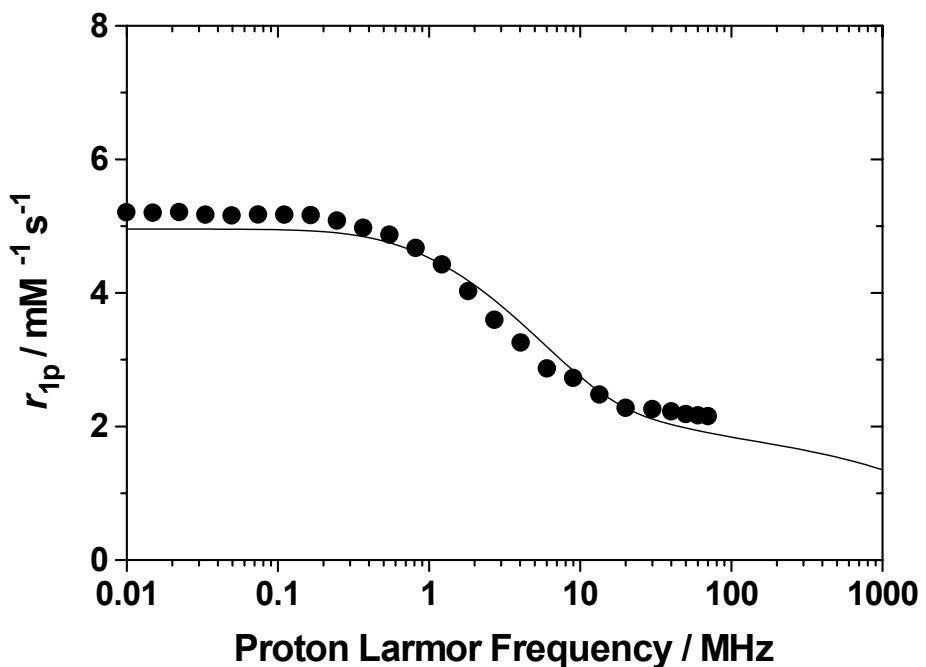


Chloride

a) VT ^1H - 20 MHz

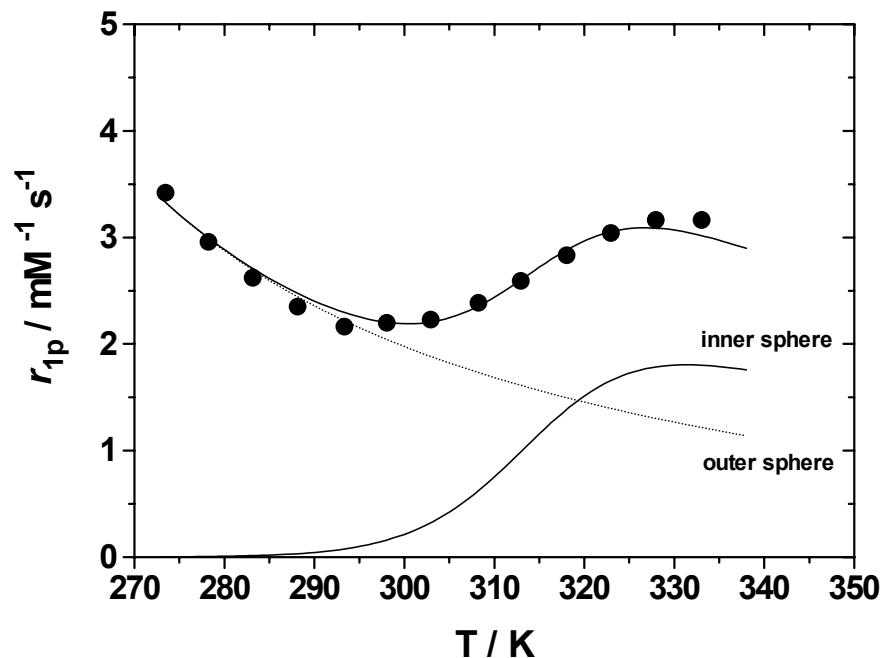


b) NMRD - 25°C, pH 6

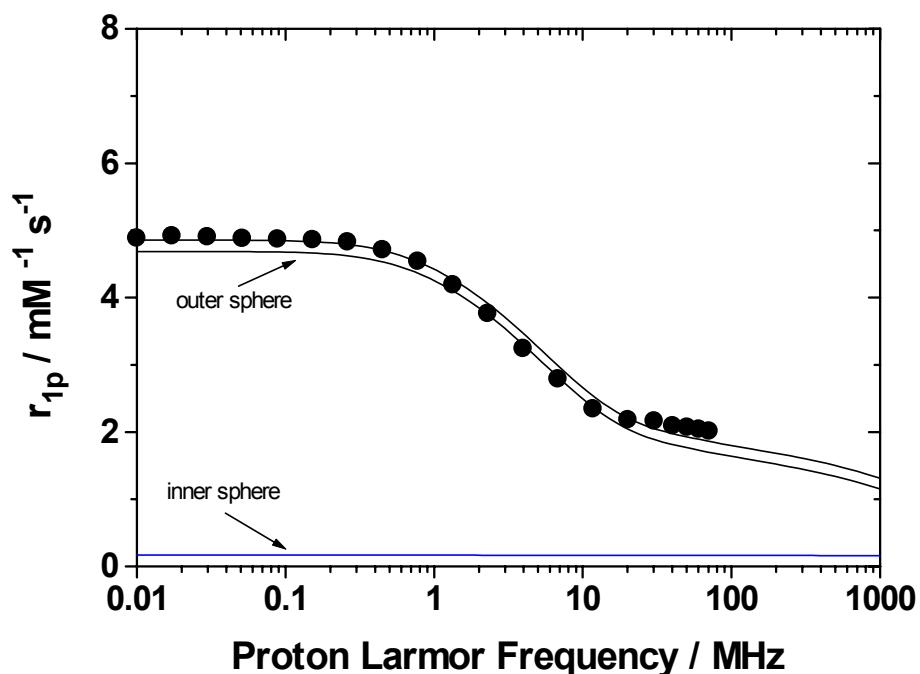


Bromide

a) VT ^1H - 20 MHz

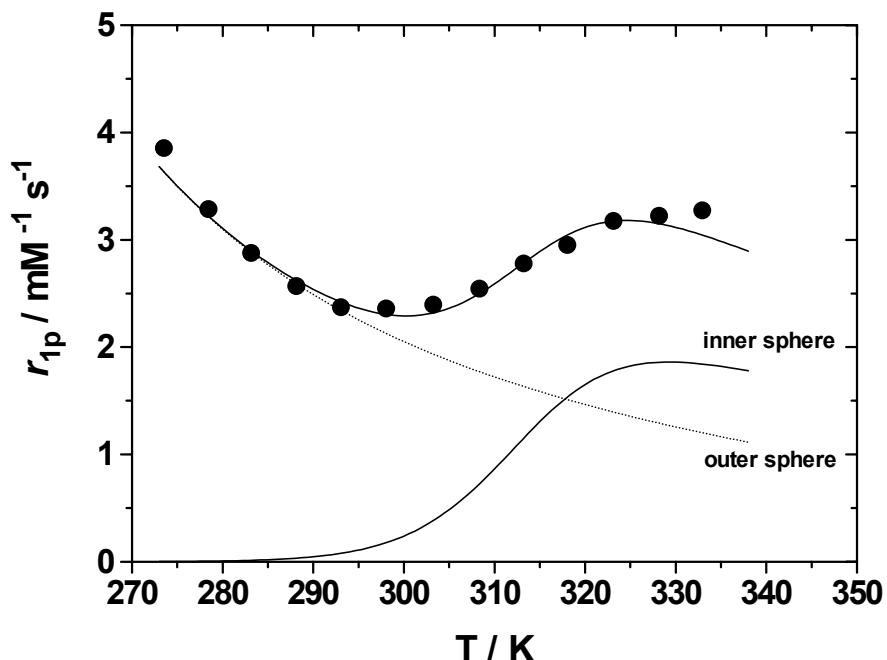


b) NMRD - 25°C, pH 6



Iodide

a) VT ^1H - 20 MHz



b) NMRD - 25°C, pH 6

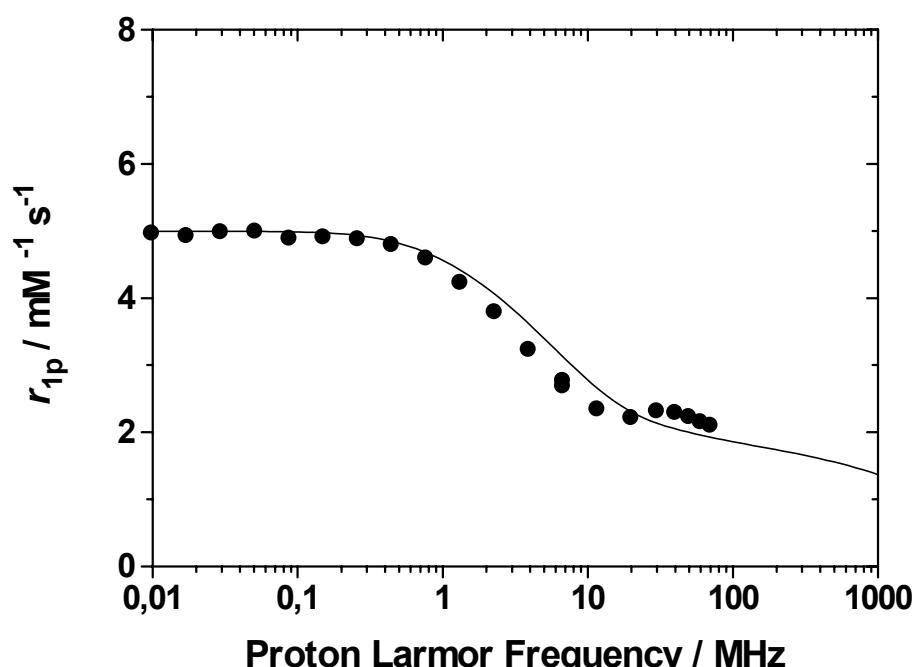


Table: Relaxation parameters for [Gd.2] salts (298 K)

	Acetate	Nitrate	Triflate	Chloride	Bromide	Iodide
r_{1p} (mM ⁻¹ s ⁻¹) ^a	2.20	2.34	2.10	2.28	2.19	2.35
Δ^2 (s ⁻¹ ; $\times 10^{19}$) ^b	1.5	1.5	1.5	1.5	1.5	1.5
τ_V (ps) ^b	8.0	8.0	8.0	8.0	8.0	8.0
ΔH_V (kJ) ^c	5.0	5.0	5.0	5.0	5.0	5.0
τ_R (ps)	100	90	98	100	90	103
ΔH_R (kJ) ^b	14	14	14	14	14	14
τ_M (μ s)	119.6	96.6	95	100	105	101
ΔH_M (kJ)	122	114	110	106.9	114	117
q ^c	1	1	1	1	1	1
r (Å) ^c	3.0	3.0	3.0	3.0	3.0	3.0
a (Å)	4.25	4.15	4.40	4.22	4.30	4.20
D (cm ² s ⁻¹ ; $\times 10^{-5}$) ^c	2.24	2.24	2.24	2.24	2.24	2.24
ΔH_D (kJ)	-21	-23	-22	-21	-22	-25

^a 20 MHz, 298 K; ^b adjusted for bromide and fixed for the other salts; ^c fixed in the fitting procedure

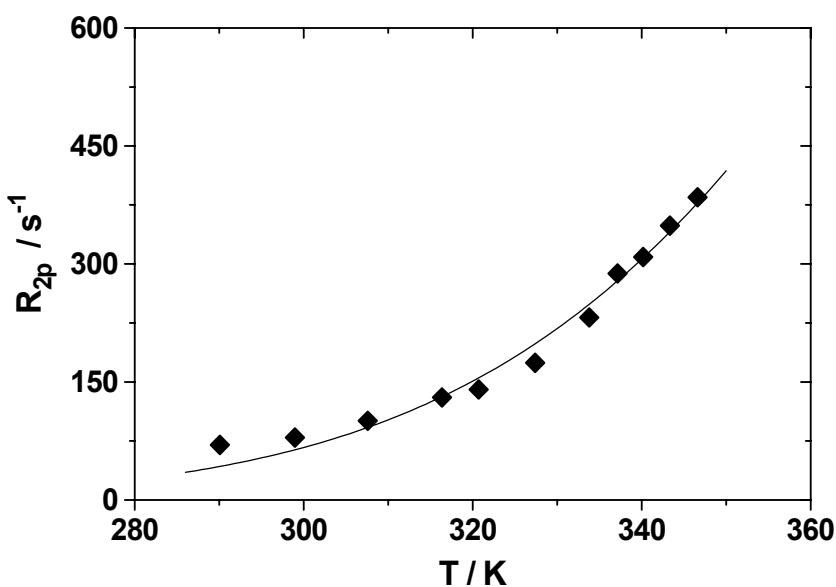


Figure A Variable temperature ^{17}O NMR measurements of the transverse relaxation rate $R_{2\text{p}}$ for $[\text{Gd.1}]\text{Cl}_3$ (2.1 T, 47 mM, pH 5.5) showing the fit to the observed data

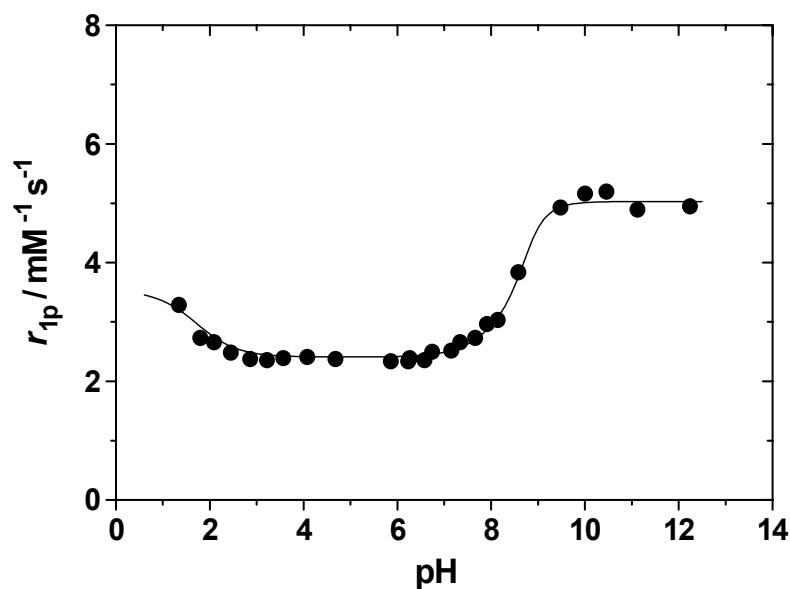


Figure B Variation of the proton relaxivity of $[\text{Gd.1}] \text{Cl}_3$ with pH (298K, 20MHz), showing the fit to the observed data (ref. 21b for details of the fitting process)