

Insertion reaction of carbon dioxide into Sn-O bond. NMR, crystal structures of di- and tetranuclear isopropoxycarbonato tin(IV) complexes, and DFT calculations

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Supporting Information

NMR and IR spectra

The ^1H , $^{119}\text{Sn}\{^1\text{H}\}$, and ^{13}C NMR experiments were run on a Bruker Avance 300 spectrometer at 300.131, 111.910, and 75.475 MHz, respectively; J values are given in Hz. ^1H and ^{13}C chemical shifts (δ , ppm) were determined relative to the solvent (^1H , CHCl_3 , δ 7.24; CDCl_3 , δ 77.00) and converted to the scale downfield from Me_4Si . $^{119}\text{Sn}\{^1\text{H}\}$ chemical shifts (δ , ppm) are reported downfield from Me_4Sn , as the internal standard. IR spectra were recorded on a Bruker Vector 22 equipped with a Specac Golden GateTM ATR device.

Figure 1.

ATR IR spectra of neat $n\text{-Bu}_2\text{Sn}(\text{O}^i\text{Pr})_2$ (**1**) before and after reaction with CO_2

Figure 2.

ATR IR spectra of neat $[n\text{-Bu}_2(^i\text{PrO})\text{Sn}]_2\text{O}$ (**3**) before and after reaction with CO_2

Figure 3.

^1H , $^{119}\text{Sn}\{^1\text{H}\}$, $^{13}\text{C}\{^1\text{H}\}$, and ^{13}C DEPT-135 NMR of $n\text{-Bu}_2\text{Sn}(\text{O}^i\text{Pr})_2$ (**1**) at 295 K

Figure 4.

^1H , $^{119}\text{Sn}\{^1\text{H}\}$, $^{13}\text{C}\{^1\text{H}\}$, and ^{13}C DEPT-135 NMR of $n\text{-Bu}_2\text{Sn}(\text{O}^i\text{Pr})(\text{OCO}_2^i\text{Pr})$ (**2**) at 253 K

Figure 5.

^1H , $^{119}\text{Sn}\{^1\text{H}\}$, $^{13}\text{C}\{^1\text{H}\}$, and ^{13}C DEPT-135 NMR of $[n\text{-Bu}_2(\text{O}^i\text{Pr})\text{Sn}]_2\text{O}$ (**3**) at 295 K

Figure 6.

^1H , $^{119}\text{Sn}\{^1\text{H}\}$, $^{13}\text{C}\{^1\text{H}\}$, and ^{13}C DEPT-135 NMR of $n\text{-Bu}_2(^i\text{PrO})\text{SnOSn}(\text{OCO}_2^i\text{Pr})n\text{-Bu}_2$ (**4**) at 295 K

Figure 7.

^{13}C DEPT-135 and ^1H NMR of $n\text{-Bu}_2\text{Sn}(\text{O}^i\text{Pr})(\text{O}^{13}\text{CO}_2^i\text{Pr})$ (**2**) after enrichment with labelled $^{13}\text{CO}_2$

Figure 8.

$^{119}\text{Sn}\{^1\text{H}\}$ and $^{13}\text{C}\{^1\text{H}\}$ NMR of $n\text{-Bu}_2\text{Sn}(\text{O}^i\text{Pr})(\text{O}^{13}\text{CO}_2^i\text{Pr})$ (**4**) after enrichment with labelled $^{13}\text{CO}_2$

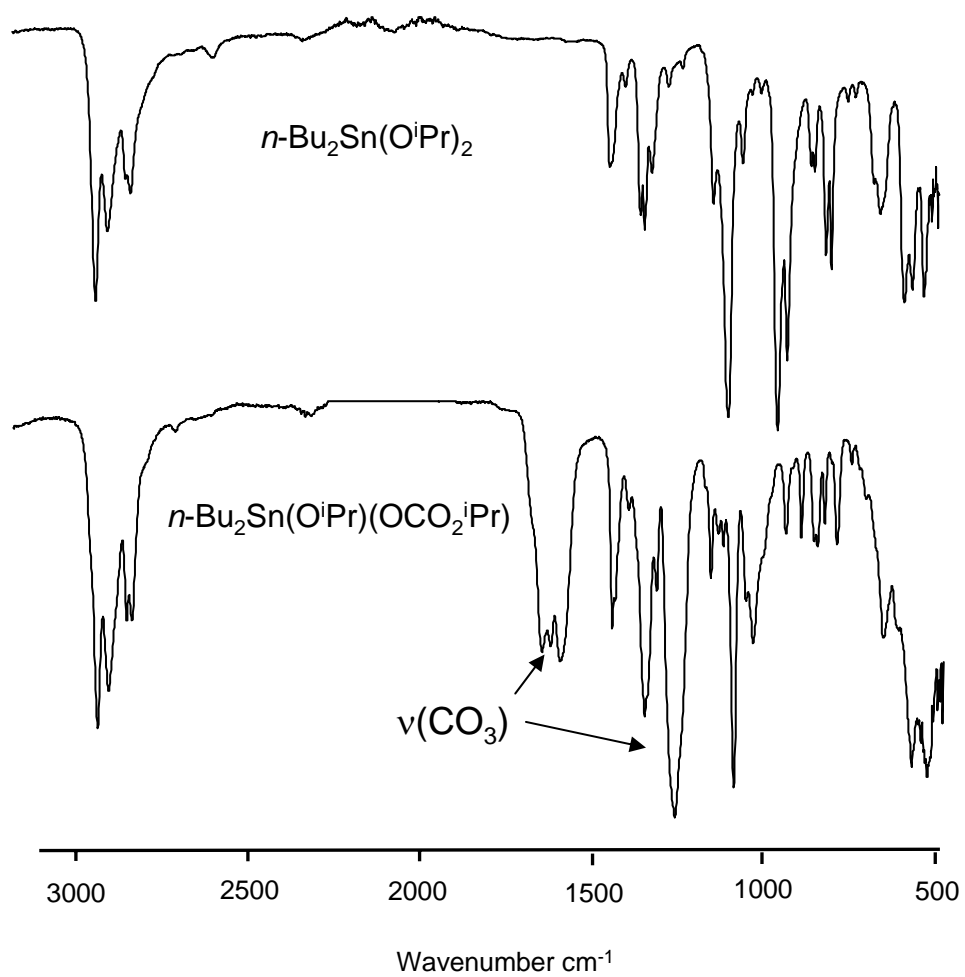


Figure 1.
ATR IR spectra of neat $n\text{-Bu}_2\text{Sn}(\text{O}^i\text{Pr})_2$ (**1**) before and after reaction with CO_2

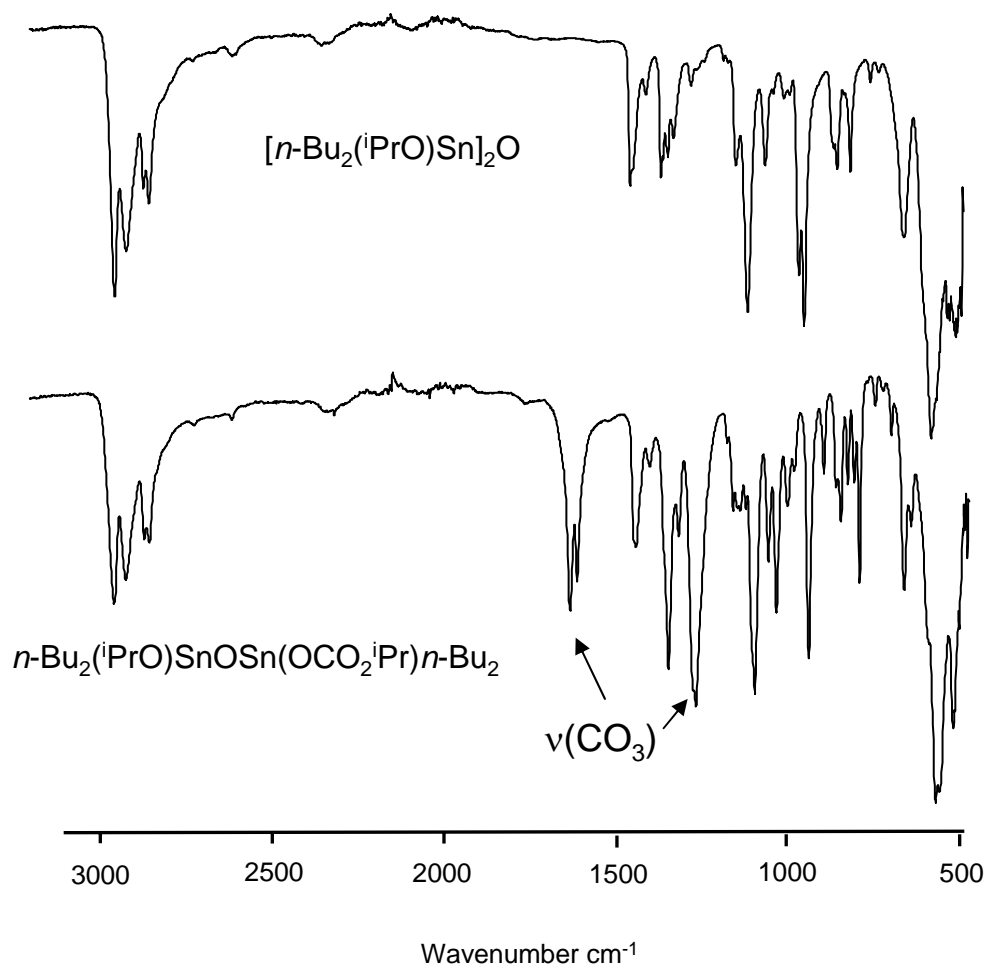


Figure 2.
ATR IR spectra of neat $[n\text{-Bu}_2(\text{iPrO})\text{Sn}]_2\text{O}$ (**3**) before and after reaction with CO_2

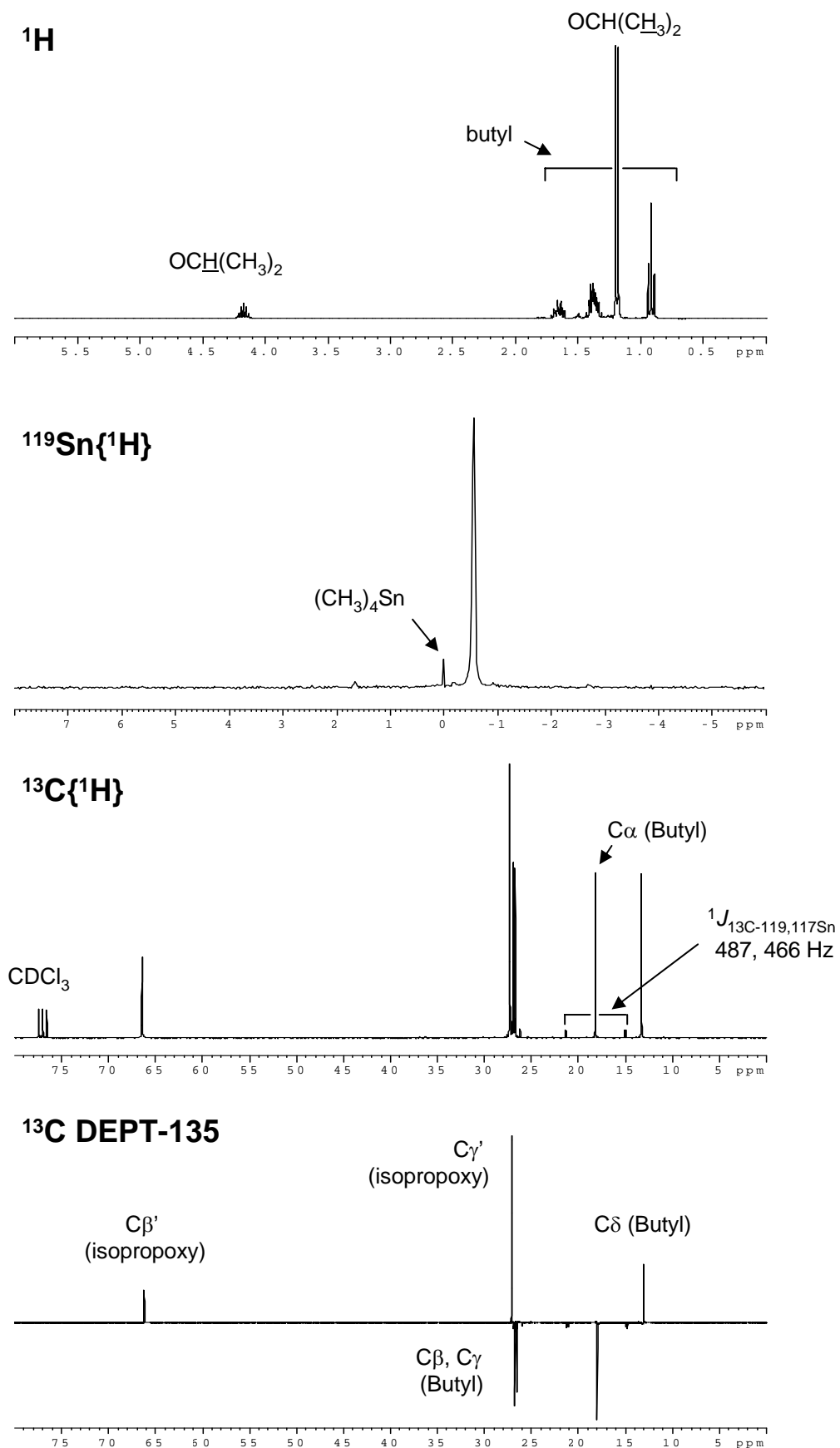


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^1H , $^{119}\text{Sn}\{^1\text{H}\}$, $^{13}\text{C}\{^1\text{H}\}$, and ^{13}C DEPT-135 NMR of $n\text{-Bu}_2\text{Sn}(\text{O}^i\text{Pr})_2$ (**1**) at 295 K

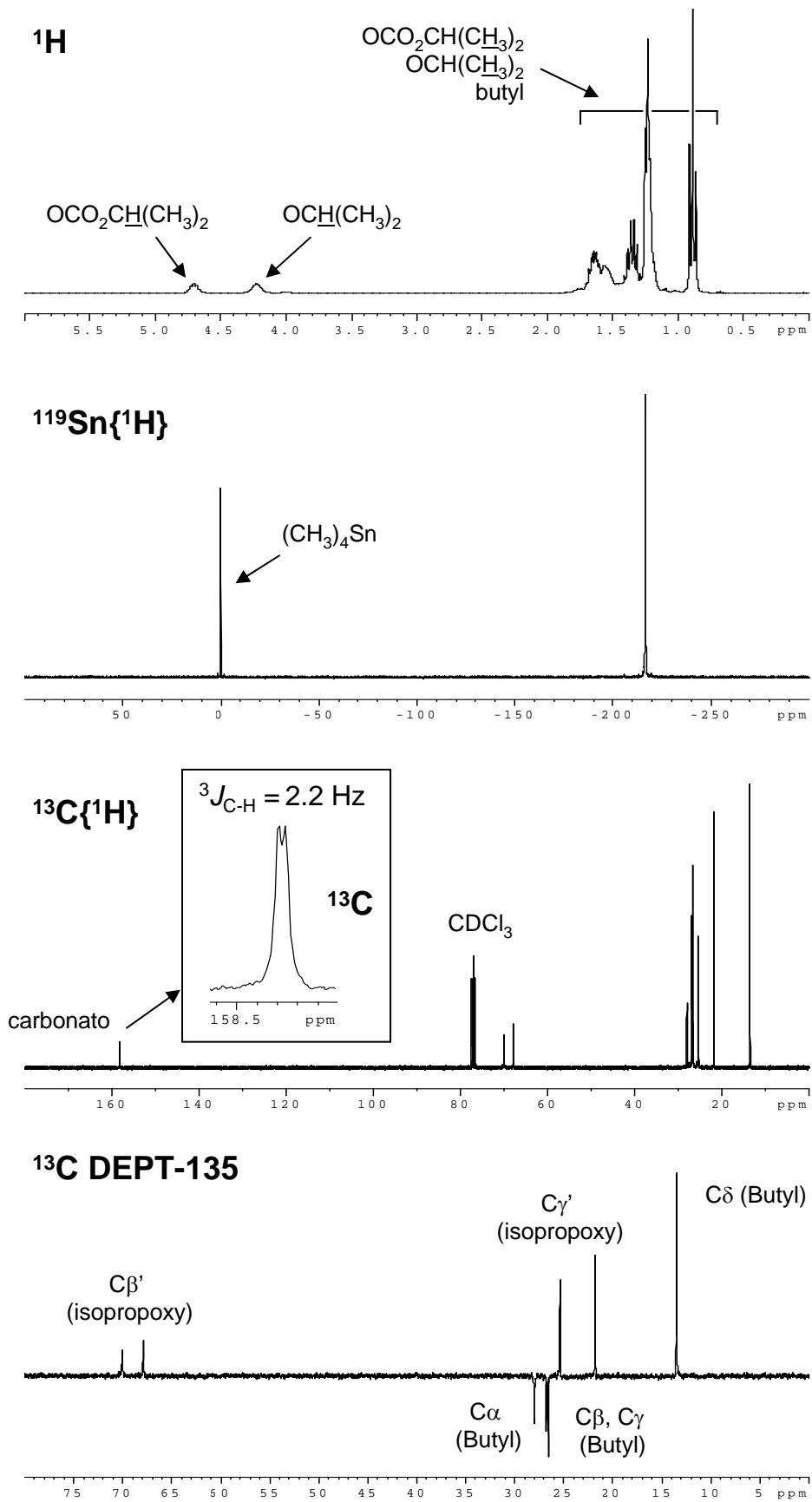


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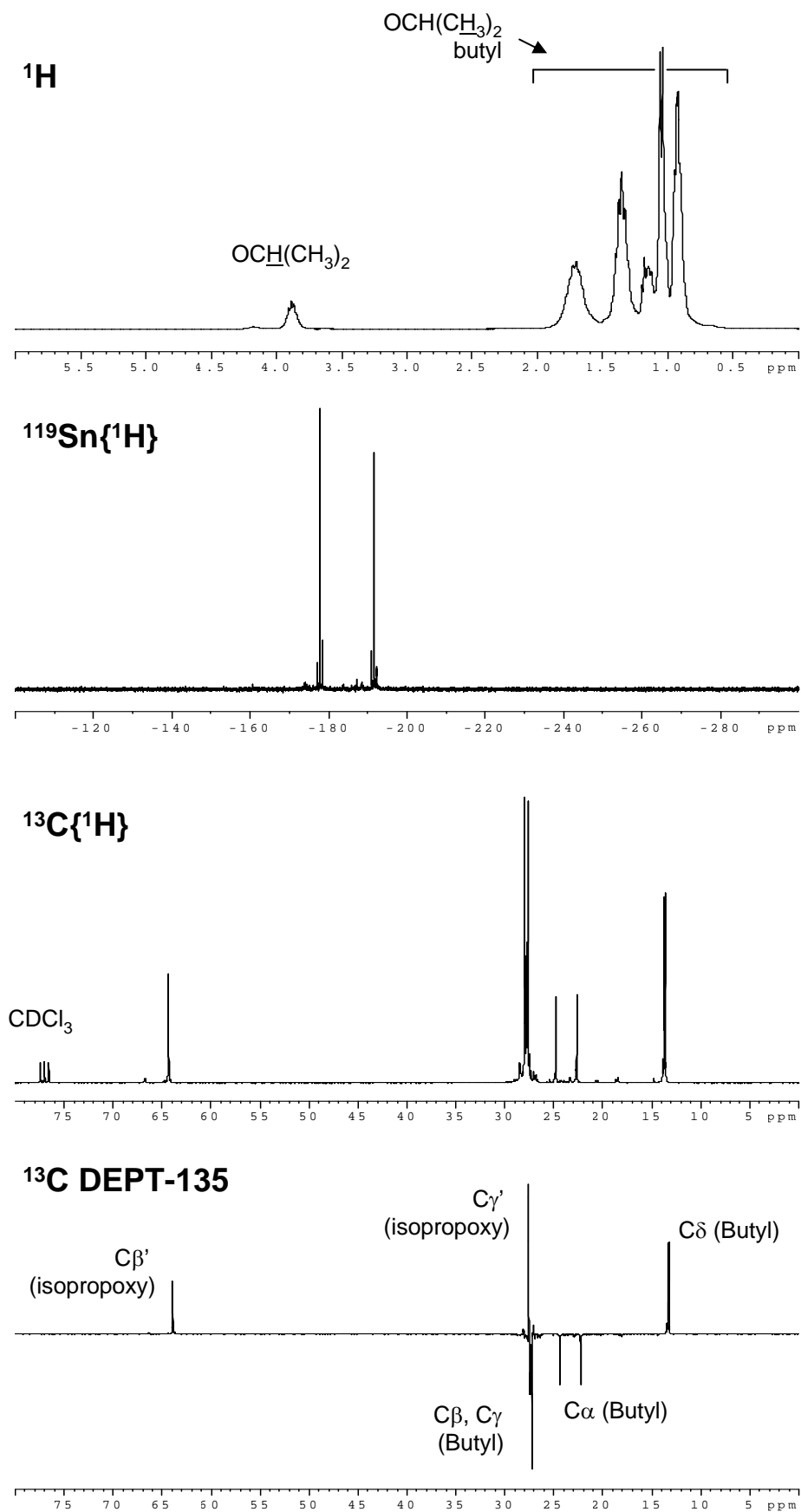


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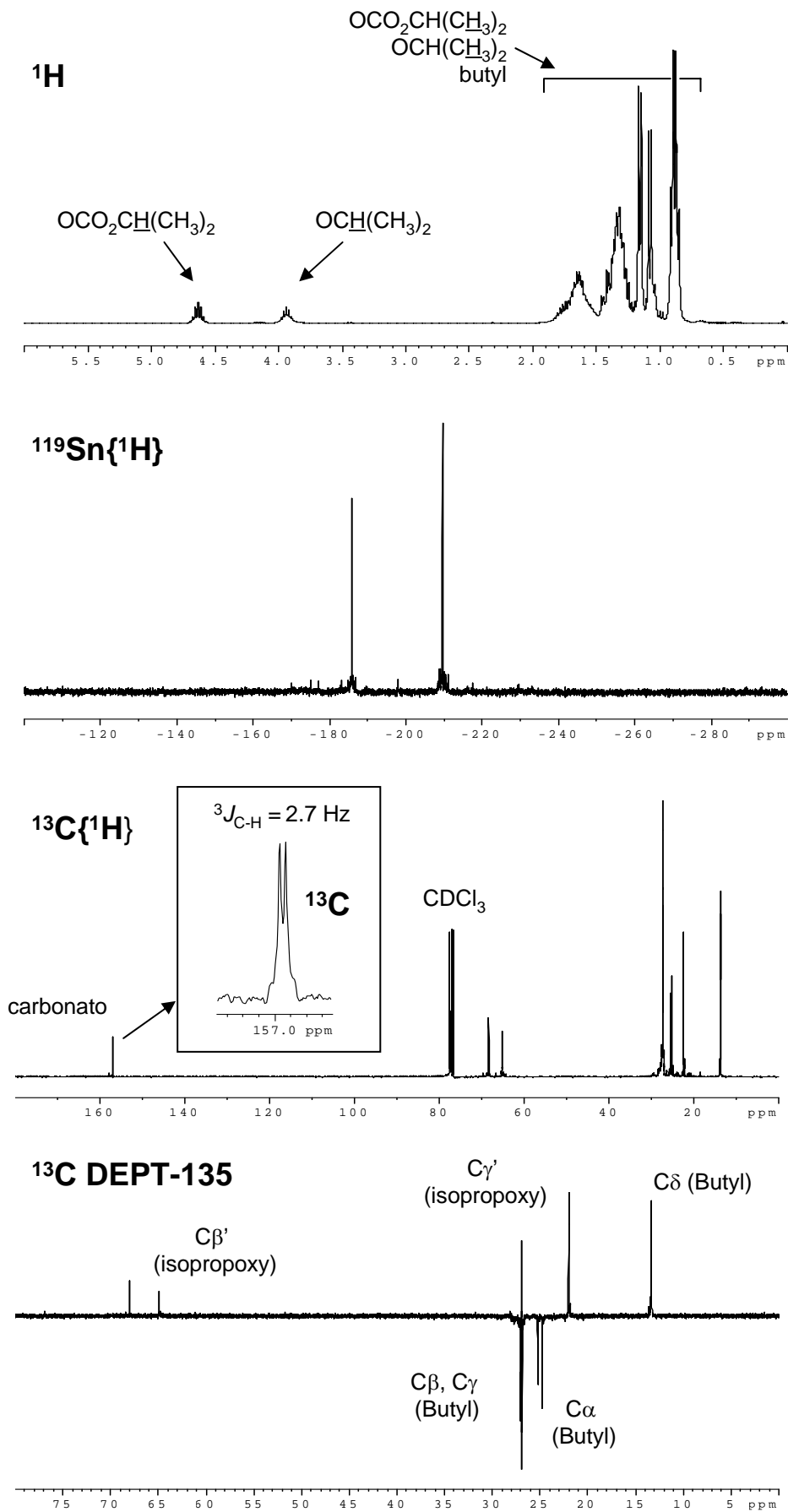


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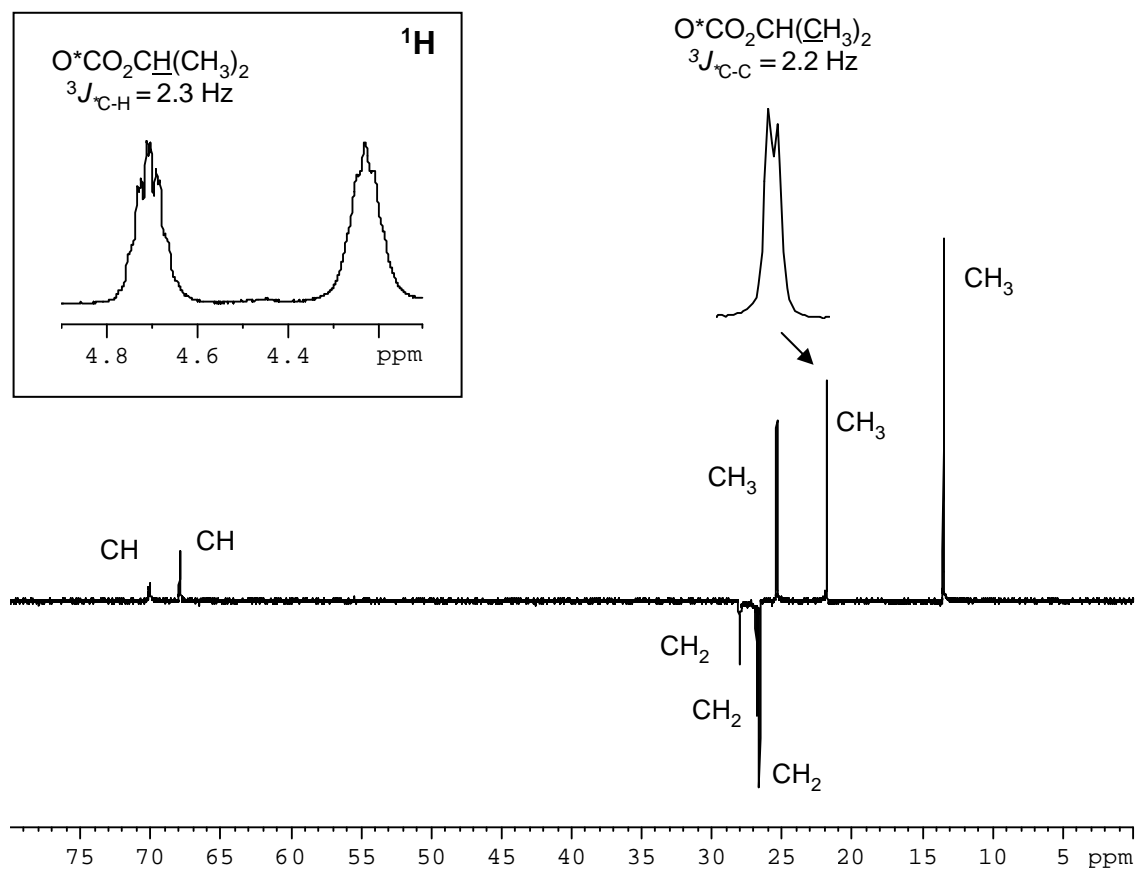


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${}^{13}\text{C}$ DEPT-135 and ${}^1\text{H}$ NMR of $n\text{-Bu}_2\text{Sn}(\text{O}^i\text{Pr})(\text{O}^{13}\text{CO}_2^i\text{Pr})$ (**2**) after enrichment with labelled ${}^{13}\text{CO}_2$

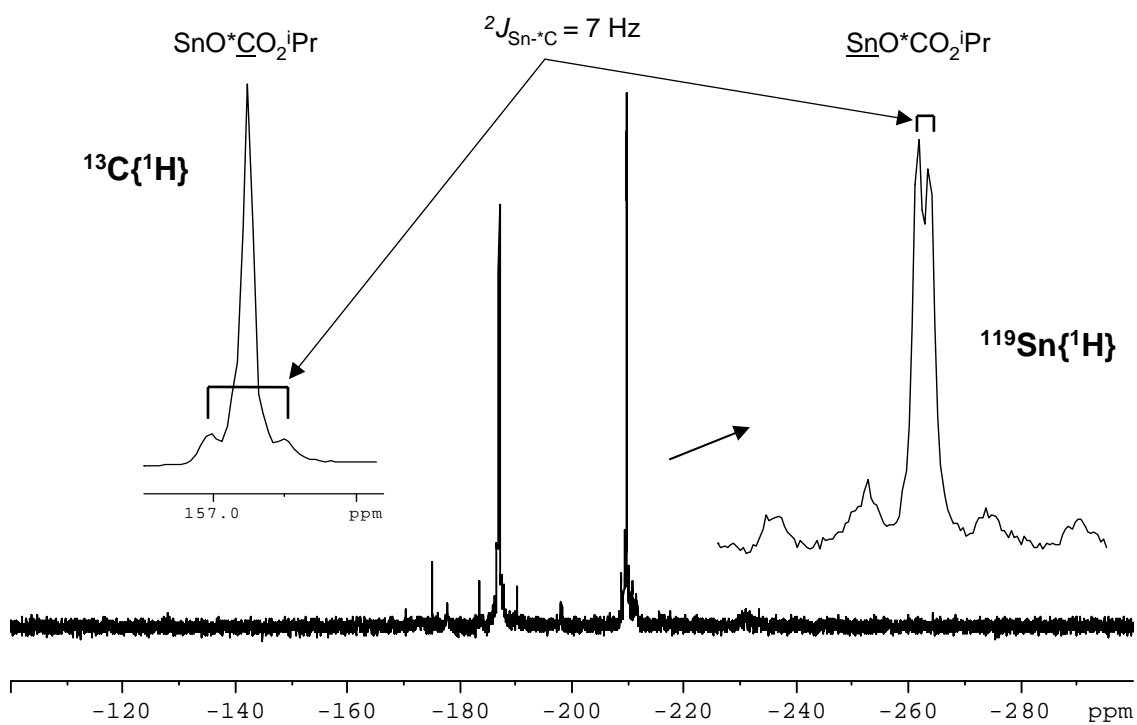


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