# The Role of Halogenated Carborane Monoanions in Olefin Hydrogenation Catalysed by Cationic Iridium Phosphine Complexes.

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#### **Electronic Supporting Information.**

# X-ray structure of $[{Ir(PPh_3)_2H}_2(\mu-H)_3][BAr^F_4]$ , 4d



Figure S1. Solid state structure of the cationic portion of **4d**.

Table S2. Crystal data and structure refinement for 4d.

	4u	
Empirical formula	$C_{108}H_{85.5}BCl_3F_{24}P_4Ir_2$	
M/g mol <sup>-1</sup>	2464.70	
Crystal colour	Pale yellow	
Crystal size /mm	0.38  imes 0.2  imes 0.13	
Crystal system	triclinic	
Space group	P -1 (no.2)	
a /Å	14.0950(1)	
b /Å	18.4279(2)	
c /Å	22.3514(2)	
α /°	67.904(1)	
ß /°	81.927(1)	
$\gamma/^{\circ}$	80.160(1)	
$V/Å^3$	5281.51(8)	
Ζ	2	
$D_c /g cm^{-3}$	1.549	
$\mu(Mo-K_{\alpha})/mm^{-1}$	2.742	
F (000)	2435	
$2\Theta$ range/°	9-60	
Collected data	92810	
Unique Data (I>2oI)	30494 (R(int) = 0.0416)	
Refined parameter	1384	
min./max density /e Å-3	-1.986 / 2.132	
$R_1^{a}$ (I>2 $\sigma$ I)	0.0442	
wR2 <sup><i>b</i></sup> (I>2 $\sigma$ I)	0.1158	
GOF <sup>cd</sup>	1.053	

4d

Table S2. Bond lengths [A] for 4d.

Ir(1) - P(2)	2.2639(11)
Tr(1) - D(1)	2 2827(11)
	2.2027(11)
Ir(1) - Ir(2)	2.5132(2)
$Tr(1) - H(1\Delta)$	1 71 (4)
	1.01(1)
lr(l)-H(lB)	1.91(5)
Ir(1) - H(1C)	1.81(6)
$T_{r}(1)$ II (100)	$1 \in (E)$
	1.05(5)
Ir(2) - P(4)	2.2665(10)
Tr(2) - P(3)	2.2873(10)
$T_{m}(2) = (0)$	1 04(4)
II(Z) - H(IA)	1.84(4)
Ir(2)-H(1B)	1.71(5)
Tr(2) - H(1C)	1 88(6)
II(2) $II(10)$	1.50(0)
Ir(2) - H(200)	1.59(6)
P(2)-C(25)	1.823(5)
P(2) = C(19)	1 833 (1)
F(2) = C(1)	1.033(4)
P(2)-C(31)	1.844(5)
P(1) - C(7)	1.822(4)
P(1) = Q(12)	1,000(4)
P(1) = C(13)	1.020(4)
P(1)-C(1)	1.839(5)
P(4) - C(55)	1,814(4)
P(4) = Q(c1)	1,020(4)
P(4) = C(61)	1.828(4)
P(4)-C(67)	1.833(4)
P(3) - C(43)	1,824(4)
$P(2) = Q(2\pi)$	1 000 (4)
P(3) = C(37)	1.828(4)
P(3)-C(49)	1.831(4)
C(1) - C(2)	1.387(7)
C(1) - C(6)	1 205 (9)
C(1) = C(0)	1.393(8)
C(2) - C(3)	1.389(7)
C(3) - C(4)	1.380(10)
$C(\Lambda) = C(5)$	1 386(10)
C(4) = C(5)	1.388(10)
C(5) - C(6)	1.391(8)
C(7) - C(12)	1.378(7)
C(7) = C(8)	$1 \ 1 \ 2 \ (7)$
C(7) = C(8)	1.412(7)
C(8) - C(9)	1.397(7)
C(10) - C(9)	1.373(9)
C(10) - C(11)	1 398 (9)
C(10) C(11)	1.396(5)
C(11) - C(12)	1.396(7)
C(13)-C(18)	1.390(7)
C(13) - C(14)	1 397(6)
C(14) $C(11)$	1,200(5)
C(14) - C(15)	1.388(7)
C(15)-C(16)	1.391(8)
C(16) - C(17)	1 370(8)
C(10) C(17)	1,200(5)
C(17) - C(18)	1.392(7)
C(19)-C(24)	1.395(6)
C(19) - C(20)	1 400(6)
C(19) $C(20)$	1,204(6)
C(20) - C(21)	1.394(6)
C(21)-C(22)	1.393(7)
C(22) - C(23)	1,378(8)
C(22) = C(24)	1 201(7)
$\cup (23) = \cup (24)$	T. J J T ( / )
C(25)-C(30)	1.392(7)
C(25)-C(26)	1.394(7)
C(26) = C(27)	1 398 (7)
C(20) = C(27)	1.370(/)
C(27)-C(28)	1.370(10)

C(28)-C(29)	1.410(10)
C(29)-C(30)	1.377(8)
C(31) - C(36)	1.383(7)
C(31) - C(32)	1.390(7)
C(32) - C(33)	1.388(7)
C(33) = C(34) C(34) = C(35)	1.392(9) 1.272(9)
C(35) - C(36)	1.575(9) 1 405(8)
C(37) - C(38)	1.385(6)
C(37)-C(42)	1.408(6)
C(38)-C(39)	1.398(7)
C(39)-C(40)	1.386(7)
C(40)-C(41)	1.387(7)
C(41) - C(42)	1.382(6)
C(43) - C(48)	1.387(7)
C(43) - C(44) C(44) - C(45)	1.401(6) 1.400(7)
C(45) - C(45)	1.400(7) 1.388(9)
C(46) - C(47)	1.370(9)
C(47) - C(48)	1.383(6)
C(49)-C(54)	1.391(6)
C(49)-C(50)	1.399(6)
C(50)-C(51)	1.393(7)
C(51)-C(52)	1.366(8)
C(52) - C(53)	1.387(8)
C(53) - C(54)	1.397(6)
C(55) - C(56)	1.393(6) 1.204(6)
C(55) - C(57)	1.394(0) 1.398(6)
C(57) - C(58)	1.372(8)
C(58)-C(59)	1.373(8)
C(59)-C(60)	1.405(6)
C(61)-C(66)	1.389(6)
C(61) - C(62)	1.398(6)
C(62) - C(63)	1.395(6)
C(63) - C(64)	1.395(7) 1.292(6)
C(64) - C(65)	1 391(6)
C(67) - C(72)	1.384(6)
C(67)-C(68)	1.407(6)
C(68)-C(69)	1.403(7)
C(69)-C(70)	1.378(8)
C(70)-C(71)	1.376(8)
C(71)-C(72)	1.399(7)
B-C(101)	1.611(10)
B = C(P1)	1.620(10)
B = C(91)	1 667(10)
C(81) - C(82)	1.403(8)
C(81)-C(88)	1.427(8)
C(82)-C(83)	1.386(7)
C(83)-C(85)	1.388(8)
C(83)-C(84)	1.503(8)
C(84) - F(1)	1.331(7)
C(84) - F(3)	1.349(6)
C(85) - C(85)	1.351(7) 1.201(0)
C(00) = C(00)	エ・フラエ (の)

C(86) - C(88) $C(86) - C(87)$ $C(87) - F(5A)$ $C(87) - F(4)$ $C(87) - F(4)$ $C(87) - F(4)$ $C(87) - F(5)$ $C(87) - F(6A)$ $C(91) - C(92)$ $C(91) - C(93)$ $C(92) - C(93)$ $C(92) - C(93)$ $C(93) - C(94)$ $C(94) - F(9)$ $C(94) - F(7)$ $C(95) - C(96)$ $C(96) - C(98)$ $C(96) - C(97)$ $C(97) - F(12)$ $C(97) - F(12)$ $C(97) - F(12)$ $C(101) - C(102)$ $C(101) - C(102)$ $C(102) - C(103)$ $C(102) - C(103)$ $C(102) - C(104)$ $C(104) - F(13)$ $C(104) - F(13)$ $C(104) - F(14)$ $C(104) - F(14)$ $C(106) - C(107)$ $C(107) - F(17)$ $C(107) - F(18)$ $C(107) - F(18)$ $C(111) - C(112)$ $C(112) - C(113)$ $C(113) - C(115)$ $C(112) - C(113)$ $C(113) - C(115)$ $C(115) - C($	$\begin{array}{c} 1.384(7)\\ 1.490(8)\\ 1.230(15)\\ 1.255(10)\\ 1.315(13)\\ 1.318(11)\\ 1.345(10)\\ 1.349(13)\\ 1.347(8)\\ 1.390(8)\\ 1.427(9)\\ 1.372(9)\\ 1.438(11)\\ 1.290(12)\\ 1.315(12)\\ 1.315(12)\\ 1.340(11)\\ 1.388(9)\\ 1.403(8)\\ 1.486(8)\\ 1.326(8)\\ 1.326(8)\\ 1.359(8)\\ 1.365(8)\\ 1.378(10)\\ 1.459(11)\\ 1.362(9)\\ 1.372(12)\\ 1.372(12)\\ 1.512(13)\\ 1.197(10)\\ 1.265(14)\\ 1.270(12)\\ 1.415(14)\\ 1.299(17)\\ 1.39(2)\\ 1.372(8)\\ 1.39(11)\\ 1.380(11)\\ 1.380(11)\\ 1.380(11)\\ 1.462(11)\\ 1.380(11)\\ 1.462(11)\\ 1.380(11)\\ 1.462(11)\\ 1.380(11)\\ 1.462(11)\\ 1.380(11)\\ 1.462(11)\\ 1.380(11)\\ 1.462(11)\\ 1.380(11)\\ 1.462(11)\\ 1.462(11)\\ 1.380(11)\\ 1.462(11)\\ $
C(107) - F(18) $C(107) - F(16)$ $C(111) - C(118)$ $C(111) - C(112)$ $C(112) - C(113)$ $C(113) - C(115)$ $C(113) - C(114)$ $C(114) - F(19)$ $C(114) - F(20)$ $C(114) - F(21)$ $C(114) - F(21)$ $C(115) - C(116)$ $C(116) - C(118)$ $C(116) - C(117)$ $C(116) - C(117)$ $C(117) - F(23)$ $C(117) - F(24)$ $C(117) - F(24)$ $C(117) - F(24)$ $C(117) - F(22)$ $C(80) - C1(2)$ $C(80) - C1(2)$ $C(80) - C1(3)$ $C(100) - C1(5)$ $C(110) - C1(7) #1$	1.299(17) 1.39(2) 1.372(8) 1.447(9) 1.399(11) 1.380(11) 1.463(11) 1.224(14) 1.288(14) 1.333(16) 1.389(9) 1.407(8) 1.485(9) 1.327(8) 1.327(8) 1.334(7) 1.352(9) 1.73(2) 1.82(2) 1.710(16) 1.742(17) 1.769(10) 1.784(10) 1.57(3)

C(110)-Cl(7)	1.750(10)
Cl(7)-C(110)#1	1.57(3)
C(120)-C(121)	1.29(13)
C(121)-C(122)	1.41(4)
C(122)-C(122)#2	1.21(5)

Symmetry transformations used to generate equivalent atoms: #1 - x + 2, -y + 1, -z #2 - x, -y, -z + 2

Table S3. Bond angles [deg] for 4d.

P(2) - Ir(1) - P(1)	98.77(4)
P(2) - Ir(1) - Ir(2)	127.95(3)
P(1) - Ir(1) - Ir(2)	122.65(3)
P(2) - Ir(1) - H(1A)	89.9(15)
$P(1) - Tr(1) - H(1\Delta)$	$169 \ 8(15)$
$Tr(2) - Tr(1) - H(1\lambda)$	47 2 (15)
$D(2) T_{\infty}(1) U(1D)$	$\frac{1}{100}$ 0(16)
P(2) = II(1) = H(1D) P(1) = Irr(1) = H(1D)	109.0(10)
P(1) - 1r(1) - H(1B)	96.5(16)
Ir(2) - Ir(1) - H(IB)	42.8(16)
H(1A) - Ir(1) - H(1B)	76(2)
P(2) - Ir(1) - H(1C)	171.1(18)
P(1) - Ir(1) - H(1C)	89.3(18)
Ir(2) - Ir(1) - H(1C)	48.1(18)
H(1A) - Ir(1) - H(1C)	82(2)
H(1B)-Ir(1)-H(1C)	74(2)
P(2) - Ir(1) - H(100)	84.0(18)
P(1) - Ir(1) - H(100)	90.3(18)
Ir(2) - Ir(1) - H(100)	121.9(18)
H(1A) - Tr(1) - H(100)	96(2)
H(1R) = Tr(1) = H(100)	164(2)
H(1C) - Tr(1) - H(100)	92(3)
D(4) = Tr(2) = D(2)	52(3)
P(4) - II(2) - P(3) P(4) - Im(2) - Im(1)	99.50(4)
P(4) - II(2) - II(1)	129.25(3)
P(3) - Ir(2) - Ir(1)	120.77(3)
P(4) - Ir(2) - H(1A)	105.8(14)
P(3) - Ir(2) - H(1A)	99.1(14)
Ir(1) - Ir(2) - H(1A)	43.0(14)
P(4) - Ir(2) - H(1B)	92.5(18)
P(3) - Ir(2) - H(1B)	168.0(18)
Ir(1) - Ir(2) - H(1B)	49.3(18)
H(1A)-Ir(2)-H(1B)	77(2)
P(4) - Ir(2) - H(1C)	167.8(17)
P(3) - Ir(2) - H(1C)	91.6(17)
Tr(1) - Tr(2) - H(1C)	45.8(17)
H(1A) - Tr(2) - H(1C)	77(2)
H(1R) - Ir(2) - H(1C)	76(2)
P(4) = Tr(2) = H(200)	83 (2)
D(2) = Tr(2) = H(200)	90(2)
F(3) = II(2) = H(200)	90 (Z) 104 (D)
Ir(1) - Ir(2) - H(200) H(10) - Ir(2) - H(200)	124(2)
H(1A) - Ir(2) - H(200)	167(2)
H(1B) - Ir(2) - H(200)	92(3)
H(1C) - Ir(2) - H(200)	92 (3)
C(25) - P(2) - C(19)	104.7(2)
C(25) - P(2) - C(31)	102.9(2)
C(19) - P(2) - C(31)	103.8(2)
C(25) - P(2) - Ir(1)	118.81(16)
C(19) - P(2) - Ir(1)	114.76(14)
C(31) - P(2) - Ir(1)	110.34(15)
C(7) - P(1) - C(13)	106.0(2)
C(7) - P(1) - C(1)	101.5(2)
C(13) - P(1) - C(1)	103.9(2)
C(7) - P(1) - Tr(1)	120 33(16)
C(13) = D(1) = Tr(1)	111 02(11)
C(TO) = E(T) = TT(T)	エエキ・ラン(エキ)

C(1) - P(1) - Ir(1)	108.13(15)
C(55) - P(4) - C(61)	105.8(2)
C(55) - P(4) - C(67)	101.72(19)
C(61) - P(4) - C(67)	103.11(19)
C(55) - P(4) - Ir(2)	116.07(14)
C(61) - P(4) - Ir(2)	116.31(13)
C(67) - P(4) - Ir(2) C(42) - P(2) - C(27)	112.05(14)
C(43) - P(3) - C(37) C(43) - D(3) - C(49)	101.85(19)
C(43) - P(3) - C(49)	103.0(2)
C(37) = P(3) = C(49) C(43) = P(3) = Tr(2)	103.93(19) 120.12(16)
C(37) - P(3) - Tr(2)	120.12(10) 106.62(13)
C(49) - P(3) - Tr(2)	116.54(13)
C(2) - C(1) - C(6)	118.9(5)
C(2) - C(1) - P(1)	117.6(4)
C(6) - C(1) - P(1)	123.5(4)
C(1) - C(2) - C(3)	120.8(5)
C(4)-C(3)-C(2)	119.9(6)
C(3)-C(4)-C(5)	120.0(5)
C(4)-C(5)-C(6)	120.1(6)
C(5) - C(6) - C(1)	120.2(6)
C(12) - C(7) - C(8)	119.3(4)
C(12) - C(7) - P(1)	122.4(3)
C(8) - C(7) - P(1)	118.2(4)
C(9) - C(8) - C(7)	119.1(5)
C(9) - C(10) - C(11)	119.6(5)
C(12) - C(11) - C(10)	119.7(5)
C(18) = C(12) = C(11)	121.0(5)
C(18) - C(13) - P(1)	1177(3)
C(14) - C(13) - P(1)	123.6(4)
C(15) - C(14) - C(13)	120.2(5)
C(14) - C(15) - C(16)	120.3(5)
C(17) - C(16) - C(15)	119.7(5)
C(16)-C(17)-C(18)	120.5(5)
C(13)-C(18)-C(17)	120.5(5)
C(24)-C(19)-C(20)	118.8(4)
C(24)-C(19)-P(2)	122.7(4)
C(20) - C(19) - P(2)	118.5(3)
C(21)-C(20)-C(19)	120.7(4)
C(22) - C(21) - C(20)	119.7(5)
C(23) - C(22) - C(21)	119.8(5)
C(22) - C(23) - C(24)	120.8(5)
C(23) - C(24) - C(19) C(20) - C(25) - C(26)	120.2(5)
C(30) - C(25) - C(20)	120.5(3)
C(26) - C(25) - P(2)	120.3(4) 120 8(4)
C(25) - C(26) - C(27)	120.8(5)
C(28) - C(27) - C(26)	119.6(6)
C(27) - C(28) - C(29)	120.4(5)
C(30)-C(29)-C(28)	119.2(6)
C(29)-C(30)-C(25)	121.3(6)
C(36)-C(31)-C(32)	119.4(5)
C(36)-C(31)-P(2)	119.7(4)
C(32)-C(31)-P(2)	120.8(4)
C(33)-C(32)-C(31)	120.8(5)
C(32)-C(33)-C(34)	119.4(5)

$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(35)-C(34)-C(33)	120.3(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(34)-C(35)-C(36)	120.1(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(31)-C(36)-C(35)	119.9(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(38)-C(37)-C(42)	119.4(4)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(38)-C(37)-P(3)	122.7(3)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(42)-C(37)-P(3)	117.4(3)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(37)-C(38)-C(39)	120.2(4)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(40) - C(39) - C(38)	120.1(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(39) - C(40) - C(41)	119.9(5)
$\begin{array}{ccccc} C(41) - C(42) - C(37) & 120.0(4) \\ C(48) - C(43) - C(44) & 119.2(4) \\ C(48) - C(43) - P(3) & 121.5(3) \\ C(44) - C(43) - P(3) & 120.1(5) \\ C(46) - C(44) - C(43) & 120.1(5) \\ C(46) - C(47) - C(48) & 121.1(6) \\ C(46) - C(47) - C(48) & 121.1(6) \\ C(46) - C(47) - C(48) & 121.1(6) \\ C(46) - C(47) - H(47) & 119.5 \\ C(47) - C(48) - C(43) & 120.1(5) \\ C(47) - C(48) - C(43) & 120.1(5) \\ C(47) - C(48) - H(48) & 120.0 \\ C(43) - C(48) - H(48) & 120.0 \\ C(43) - C(48) - H(48) & 120.0 \\ C(54) - C(49) - P(3) & 117.6(3) \\ C(50) - C(49) - P(3) & 122.6(3) \\ C(51) - C(50) - C(49) & 119.1(5) \\ C(51) - C(50) - H(50) & 120.4 \\ C(52) - C(51) - H(50) & 120.4 \\ C(52) - C(51) - H(51) & 119.5 \\ C(51) - C(52) - H(51) & 119.5 \\ C(51) - C(52) - H(51) & 119.5 \\ C(51) - C(52) - H(52) & 129.8 \\ C(52) - C(53) - H(53) & 120.3 \\ C(49) - C(54) - H(54) & 119.9 \\ C(52) - C(53) - H(53) & 120.3 \\ C(49) - C(54) - H(54) & 119.9 \\ C(53) - C(55) - P(4) & 120.7(3) \\ C(49) - C(54) - H(54) & 119.9 \\ C(56) - C(55) - P(4) & 120.7(3) \\ C(56) - C(55) - P(4) & 120.4(5) \\ C(57) - C(58) - C(59) & 120.4(5) \\ C(55) - C(60) - C(59) & 119.9(5) \\ C(55) - C(60) - C(59) & 119.9(5) \\ C(55) - C(60) - C(59) & 119.9(5) \\ C(56) - C(61) - P(4) & 118.3(3) \\ C(62) - C(61) - P(4) & 122.8(3) \\ C(66) - C(61) - P(4) & 122.8(3) \\ C(66) - C(61) - P(4) & 120.1(4) \\ C(64) - C(65) - C(66) & 120.1(4) \\ C(64) - C(65) - C(65) & 120.1(4) \\ C(64) - C(65) - C(65) & 120.1(4) \\ C(64) - C(65) - C(66) & 120.1(4) \\ C(64) - C(65) - C(66) & 120.1(4) \\ C(64) - C(65) - C(65) & 120.1(4) \\ C(61) - C(66) - C(65) & 120.1(4) \\ $	C(42) - C(41) - C(40)	120.4(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(41) - C(42) - C(37)	120.0(4)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(48) - C(43) - C(44)	119.2(4)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(48)-C(43)-P(3)	121.5(3)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(44)-C(43)-P(3)	119.3(4)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(45)-C(44)-C(43)	120.1(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(46)-C(45)-C(44)	119.4(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(47)-C(46)-C(45)	120.1(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(46)-C(47)-C(48)	121.1(6)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	С(46)-С(47)-Н(47)	119.5
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(48)-C(47)-H(47)	119.5
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(47)-C(48)-C(43)	120.1(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(47)-C(48)-H(48)	120.0
$\begin{array}{llllllllllllllllllllllllllllllllllll$	С(43)-С(48)-Н(48)	120.0
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(54)-C(49)-C(50)	119.7(4)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(54)-C(49)-P(3)	117.6(3)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(50)-C(49)-P(3)	122.6(3)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(51)-C(50)-C(49)	119.1(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(51)-C(50)-H(50)	120.4
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(49)-C(50)-H(50)	120.4
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(52)-C(51)-C(50)	121.1(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(52)-C(51)-H(51)	119.5
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(50)-C(51)-H(51)	119.5
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(51) - C(52) - C(53)	120.4(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(51)-C(52)-H(52)	119.8
$\begin{array}{llllllllllllllllllllllllllllllllllll$	С(53)-С(52)-Н(52)	119.8
$\begin{array}{llllllllllllllllllllllllllllllllllll$	C(52) - C(53) - C(54)	119.5(5)
$\begin{array}{llllllllllllllllllllllllllllllllllll$	С(52)-С(53)-Н(53)	120.3
$\begin{array}{llllllllllllllllllllllllllllllllllll$	С(54)-С(53)-Н(53)	120.3
C(49) - C(54) - H(54) $119.9$ $C(53) - C(54) - H(54)$ $119.9$ $C(53) - C(55) - C(60)$ $119.4(4)$ $C(56) - C(55) - P(4)$ $120.7(3)$ $C(60) - C(55) - P(4)$ $119.6(3)$ $C(55) - C(56) - C(57)$ $119.8(5)$ $C(58) - C(57) - C(56)$ $120.4(5)$ $C(57) - C(58) - C(59)$ $120.6(5)$ $C(55) - C(60) - C(59)$ $119.9(5)$ $C(55) - C(60) - C(59)$ $119.9(5)$ $C(66) - C(61) - P(4)$ $118.3(3)$ $C(62) - C(61) - P(4)$ $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$	C(49) - C(54) - C(53)	120.2(4)
C(53) - C(54) - H(54) $119.9$ $C(56) - C(55) - C(60)$ $119.4(4)$ $C(56) - C(55) - P(4)$ $120.7(3)$ $C(60) - C(55) - P(4)$ $119.6(3)$ $C(55) - C(56) - C(57)$ $119.8(5)$ $C(58) - C(57) - C(56)$ $120.4(5)$ $C(57) - C(58) - C(59)$ $120.6(5)$ $C(55) - C(60) - C(59)$ $119.9(5)$ $C(55) - C(60) - C(59)$ $119.9(5)$ $C(66) - C(61) - C(62)$ $118.9(4)$ $C(66) - C(61) - P(4)$ $118.3(3)$ $C(62) - C(61) - P(4)$ $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(49) - C(54) - H(54)	119.9
C(56) - C(55) - C(60) $119.4(4)$ $C(56) - C(55) - P(4)$ $120.7(3)$ $C(60) - C(55) - P(4)$ $119.6(3)$ $C(55) - C(56) - C(57)$ $119.8(5)$ $C(58) - C(57) - C(56)$ $120.4(5)$ $C(57) - C(58) - C(59)$ $120.6(5)$ $C(58) - C(59) - C(60)$ $119.9(5)$ $C(55) - C(60) - C(59)$ $119.9(5)$ $C(66) - C(61) - C(62)$ $118.9(4)$ $C(66) - C(61) - P(4)$ $118.3(3)$ $C(62) - C(61) - P(4)$ $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(53) - C(54) - H(54)	119.9
C(56) - C(55) - P(4) $120.7(3)$ $C(60) - C(55) - P(4)$ $119.6(3)$ $C(55) - C(56) - C(57)$ $119.8(5)$ $C(58) - C(57) - C(56)$ $120.4(5)$ $C(57) - C(58) - C(59)$ $120.6(5)$ $C(58) - C(59) - C(60)$ $119.9(5)$ $C(55) - C(60) - C(59)$ $119.9(5)$ $C(66) - C(61) - P(4)$ $118.3(3)$ $C(62) - C(61) - P(4)$ $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(56) - C(55) - C(60)	119.4(4)
C(60) - C(55) - P(4) $I19.6(3)$ $C(55) - C(56) - C(57)$ $I19.8(5)$ $C(55) - C(56) - C(57)$ $I20.4(5)$ $C(57) - C(58) - C(59)$ $I20.6(5)$ $C(57) - C(58) - C(59)$ $I20.6(5)$ $C(55) - C(60) - C(59)$ $I19.9(5)$ $C(66) - C(61) - C(62)$ $I18.9(4)$ $C(66) - C(61) - P(4)$ $I18.3(3)$ $C(62) - C(61) - P(4)$ $I22.8(3)$ $C(63) - C(62) - C(61)$ $I20.5(4)$ $C(64) - C(63) - C(62)$ $I19.6(4)$ $C(64) - C(65) - C(66)$ $I20.1(4)$ $C(61) - C(66) - C(65)$ $I20.7(4)$	C(56) - C(55) - P(4)	120.7(3)
C(55) - C(56) - C(57) $119.8(5)$ $C(58) - C(57) - C(56)$ $120.4(5)$ $C(57) - C(58) - C(59)$ $120.6(5)$ $C(58) - C(59) - C(60)$ $119.9(5)$ $C(55) - C(60) - C(59)$ $119.9(5)$ $C(66) - C(61) - C(62)$ $118.9(4)$ $C(66) - C(61) - P(4)$ $118.3(3)$ $C(62) - C(61) - P(4)$ $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(60) - C(55) - P(4)	119.6(3)
C(58) - C(57) - C(56) $120.4(5)$ $C(57) - C(58) - C(59)$ $120.6(5)$ $C(57) - C(59) - C(60)$ $119.9(5)$ $C(55) - C(60) - C(59)$ $119.9(5)$ $C(66) - C(61) - C(62)$ $118.9(4)$ $C(66) - C(61) - P(4)$ $118.3(3)$ $C(63) - C(62) - C(61)$ $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(55) - C(56) - C(57)	119.8(5)
C(57) - C(58) - C(59) $120.8(5)$ $C(58) - C(59) - C(60)$ $119.9(5)$ $C(55) - C(60) - C(59)$ $119.9(5)$ $C(66) - C(61) - C(62)$ $118.9(4)$ $C(66) - C(61) - P(4)$ $118.3(3)$ $C(62) - C(61) - P(4)$ $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(58) - C(57) - C(56)	120.4(5)
C(53) - C(63) - C(60) $119.9(5)$ $C(55) - C(60) - C(59)$ $119.9(5)$ $C(66) - C(61) - C(62)$ $118.9(4)$ $C(66) - C(61) - P(4)$ $118.3(3)$ $C(62) - C(61) - P(4)$ $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(57) - C(58) - C(59)	120.6(5)
C(63) - C(60) - C(63) $119.9(3)$ $C(66) - C(61) - C(62)$ $118.9(4)$ $C(66) - C(61) - P(4)$ $118.3(3)$ $C(62) - C(61) - P(4)$ $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(58) = C(59) = C(80)	119.9(5)
C(60) - C(61) - P(4) $118.3(3)$ $C(62) - C(61) - P(4)$ $118.3(3)$ $C(62) - C(61) - P(4)$ $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(65) - C(64) - C(63)$ $120.1(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(55) = C(50) = C(53)	119.9(3)
C(62) - C(61) - P(4) $122.8(3)$ $C(63) - C(62) - C(61)$ $120.5(4)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(65) - C(64) - C(63)$ $120.1(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(66) - C(61) - P(4)	118 2(2)
C(63) - C(62) - C(61) $122.0(3)$ $C(64) - C(63) - C(62)$ $119.6(4)$ $C(65) - C(64) - C(63)$ $120.1(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(62) - C(61) - P(4)	122 8(3)
C(64) - C(63) - C(62) $119.6(4)$ $C(65) - C(64) - C(63)$ $120.1(4)$ $C(64) - C(65) - C(66)$ $120.1(4)$ $C(61) - C(66) - C(65)$ $120.7(4)$	C(63) - C(62) - C(61)	120.5(3)
C(65) - C(64) - C(63)       120.1(4)         C(64) - C(65) - C(66)       120.1(4)         C(61) - C(66) - C(65)       120.7(4)	C(64) - C(63) - C(62)	119.6(4)
C(64) -C(65) -C(66) 120.1(4) C(61) -C(66) -C(65) 120.7(4)	C(65) - C(64) - C(63)	120.1(4)
C(61)-C(66)-C(65) 120.7(4)	C(64) - C(65) - C(66)	120.1(4)
	C(61)-C(66)-C(65)	120.7(4)

C(72)-C(67)-C(68)	119.4(4)
C(72) - C(67) - P(4)	120.9(3)
C(68) - C(67) - P(4)	119.5(3)
C(70) = C(68) = C(68)	120.9(5)
C(71) - C(70) - C(69)	120.0(5) 120.4(5)
C(70) - C(71) - C(72)	119.6(5)
C(67) - C(72) - C(71)	120.8(5)
C(101)-B-C(111)	107.6(5)
C(101)-B-C(81)	110.1(5)
C(111)-B-C(81)	114.5(5)
C(101)-B-C(91)	110.9(6)
C(111) - B - C(91)	108.3(5)
C(81) - B - C(91)	105.4(5)
C(82) - C(81) - C(80)	122.7(5)
C(88) - C(81) - B	121.9(5)
C(83) - C(82) - C(81)	122.1(5)
C(82)-C(83)-C(85)	121.8(5)
C(82)-C(83)-C(84)	120.5(5)
C(85)-C(83)-C(84)	117.7(5)
F(1) - C(84) - F(3)	107.1(5)
F(1) - C(84) - F(2)	106.7(5)
F(3) - C(84) - F(2) F(1) - C(84) - C(82)	105.1(5)
F(1) = C(84) = C(83) F(3) = C(84) = C(83)	112.7(5) 112.4(5)
F(2) - C(84) - C(83)	112.3(5)
C(83)-C(85)-C(86)	117.8(5)
C(88)-C(86)-C(85)	120.8(5)
C(88)-C(86)-C(87)	120.2(5)
C(85) - C(86) - C(87)	119.0(5)
F(5A) - C(87) - F(6)	125.3(11)
F(5A) - C(87) - F(4A) F(6) - C(87) - F(4A)	100.8(17) 44 2(9)
F(5A) - C(87) - F(4)	37.8(17)
F(6)-C(87)-F(4)	106.4(10)
F(4A)-C(87)-F(4)	67.7(12)
F(5A)-C(87)-F(5)	67.6(17)
F(6) - C(87) - F(5)	106.4(10)
F(4A) - C(87) - F(5) F(4) - C(87) - F(5)	135.8(9)
F(5A) - C(87) - F(5A)	113.7(16)
F(6) - C(87) - F(6A)	53.1(12)
F(4A)-C(87)-F(6A)	95.5(15)
F(4)-C(87)-F(6A)	132.7(10)
F(5)-C(87)-F(6A)	56.8(12)
F(5A) - C(87) - C(86)	117.3(9)
F(6) - C(87) - C(86)	115.1(7)
F(4A) - C(87) - C(86) F(4) - C(87) - C(86)	112.9(7) 112.9(8)
F(5) - C(87) - C(86)	111.1(6)
F(6A)-C(87)-C(86)	114.4(7)
C(86)-C(88)-C(81)	122.4(5)
C(10) - C(9) - C(8)	121.3(5)
C(92) - C(91) - C(98)	116.6(6)
C(92) - C(91) - B C(98) - C(91) - B	121.5(5)
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C(91) - C(92) - C(93) $C(95) - C(93) - C(94)$ $C(92) - C(93) - C(94)$ $F(9) - C(94) - F(8)$ $F(9) - C(94) - F(7)$ $F(8) - C(94) - F(7)$ $F(9) - C(94) - C(93)$ $F(7) - C(94) - C(93)$ $F(7) - C(94) - C(93)$ $C(93) - C(95) - C(96)$ $C(95) - C(96) - C(97)$ $F(12) - C(96) - C(97)$ $F(12) - C(97) - F(10)$ $F(12) - C(97) - F(11)$ $F(12) - C(97) - F(11)$ $F(12) - C(97) - F(11)$ $F(12) - C(97) - C(96)$ $F(10) - C(97) - C(96)$ $F(11) - C(97) - C(96)$ $F(11) - C(97) - C(96)$ $C(102) - C(101) - C(108)$ $C(102) - C(101) - B$ $C(103) - C(102) - C(104)$ $C(102) - C(103) - C(104)$ $F(13) - C(104) - F(14)$ $F(13) - C(104) - F(14)$ $F(13) - C(104) - F(14)$ $F(13) - C(104) - C(103)$ $F(15) - C(104) - C(103)$	121.9(6) $119.7(6)$ $122.4(7)$ $117.8(7)$ $105.1(10)$ $96.2(8)$ $105.9(9)$ $118.1(8)$ $113.4(9)$ $116.1(8)$ $119.2(6)$ $120.2(5)$ $118.1(6)$ $121.6(6)$ $104.8(5)$ $104.0(5)$ $113.6(6)$ $111.9(5)$ $113.2(6)$ $122.1(6)$ $122.1(6)$ $122.1(6)$ $122.2(6)$ $122.1(6)$ $122.5(7)$ $128.6(6)$ $118.7(7)$ $126.1(7)$ $120.4(8)$ $119.4(7)$ $120.1(7)$ $95.8(11)$ $118.4(12)$ $100.9(11)$ $119.7(7)$ $106.4(12)$
C(103)-C(105)-C(106) C(108)-C(106)-C(107) C(108)-C(106)-C(105) C(107)-C(106)-C(105) F(17)-C(107)-F(18)	117.8(8) 123.0(11) 120.5(8) 116.3(10) 109.1(13)
F(17) - C(107) - F(16) $F(18) - C(107) - F(16)$ $F(17) - C(107) - C(106)$ $F(18) - C(107) - C(106)$ $F(16) - C(107) - C(106)$ $C(106) - C(108) - C(101)$ $C(118) - C(111) - C(112)$	94.1(13) 99.4(17) 120.9(15) 119.2(11) 108.5(14) 122.5(9) 113.5(6)
C(118) - C(111) - B $C(112) - C(111) - B$ $C(113) - C(112) - C(111)$ $C(115) - C(113) - C(112)$ $C(115) - C(113) - C(114)$ $C(112) - C(113) - C(114)$ $F(19) - C(114) - F(20)$ $F(19) - C(114) - F(21)$ $F(20) - C(114) - F(21)$ $F(19) - C(114) - F(21)$ $F(20) - C(114) - C(113)$ $F(20) - C(114) - C(113)$	121.4(6) 124.6(6) 122.6(6) 121.0(6) 119.9(8) 119.1(8) 101.3(15) 109.6(12) 102.5(12) 118.1(10) 114.0(9)

F(21)-C(114)-C(113)	110.0(12)
C(113)-C(115)-C(116)	117.9(7)
C(115)-C(116)-C(118)	120.5(6)
C(115)-C(116)-C(117)	119.8(6)
C(118)-C(116)-C(117)	119.7(5)
F(23)-C(117)-F(24)	106.6(6)
F(23)-C(117)-F(22)	106.8(6)
F(24)-C(117)-F(22)	105.6(6)
F(23)-C(117)-C(116)	113.4(6)
F(24)-C(117)-C(116)	114.5(5)
F(22)-C(117)-C(116)	109.4(6)
C(111)-C(118)-C(116)	124.3(5)
Cl(2) - C(80) - Cl(1)	114.9(9)
Cl(4) - C(90) - Cl(3)	118.8(14)
Cl(6) - C(100) - Cl(5)	98.0(9)
Cl(7) #1-C(110) - Cl(7)	130.4(13)
C(110) #1-Cl(7)-C(110)	49.6(13)
C(120)-C(121)-C(122)	133(4)
C(122)#2-C(122)-C(121)	133(4)

Symmetry transformations used to generate equivalent atoms: #1 - x + 2, -y + 1, -z #2 - x, -y, -z + 2