

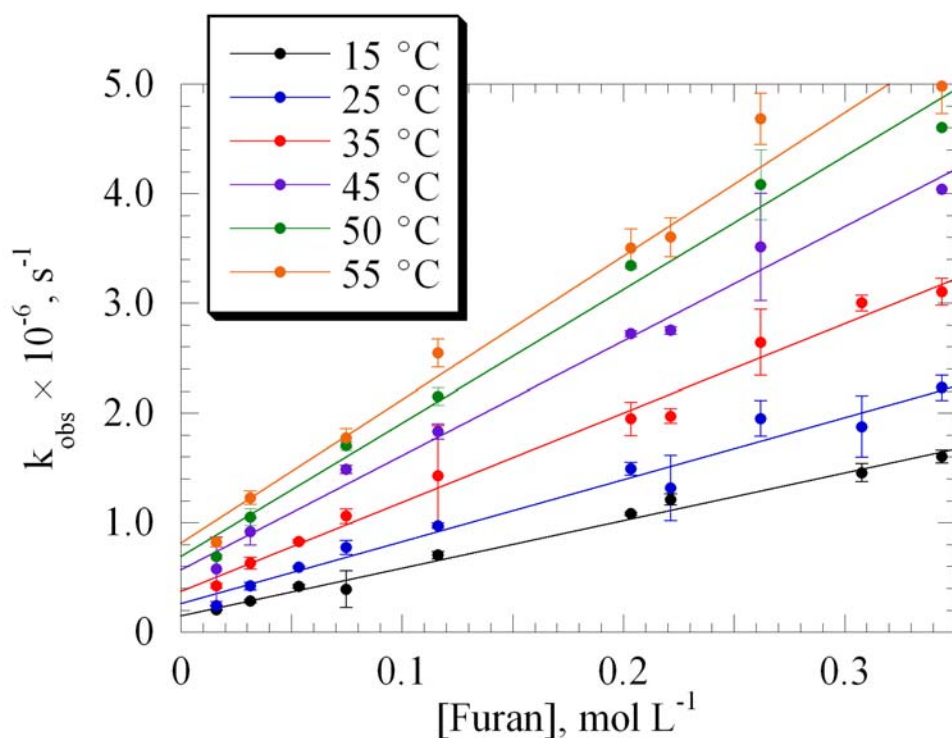
**Electronic Supporting Information**

**Reactions of the Transient Species  $\text{Cr}(\text{CO})_5(\text{cyclohexane})$  with  
 $\text{C}_4\text{H}_n\text{E}$  ( $n = 4, 8$ ;  $\text{E} = \text{O}, \text{NH}, \text{S}$ ) Studied by Time-Resolved  
Infrared Absorption Spectroscopy**

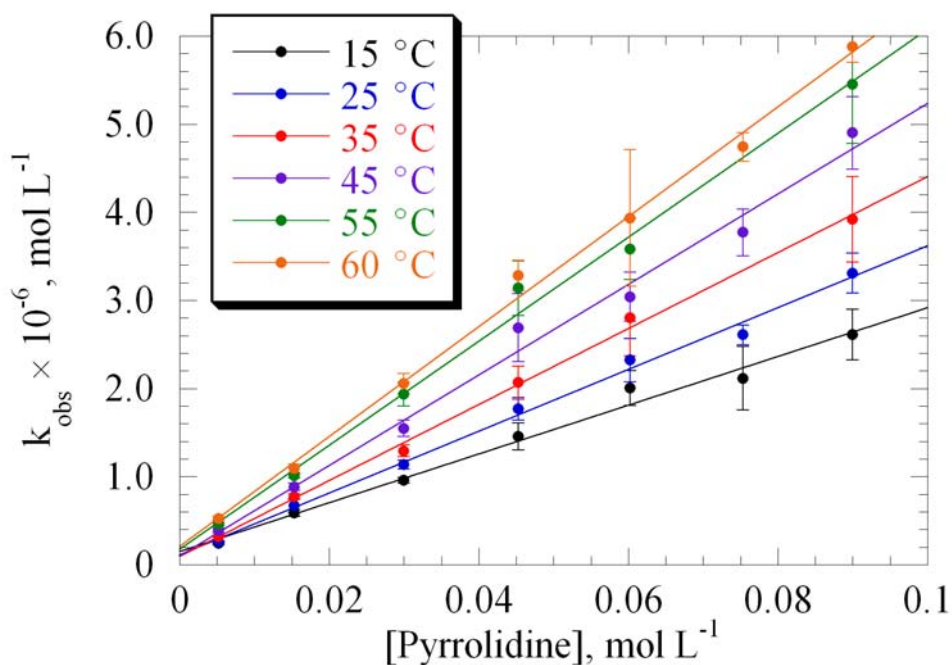
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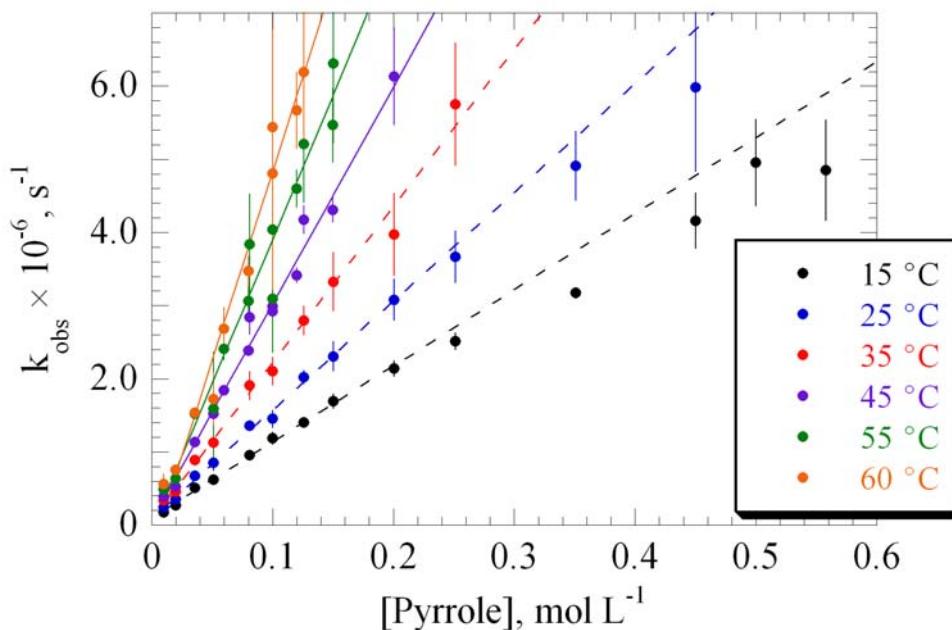
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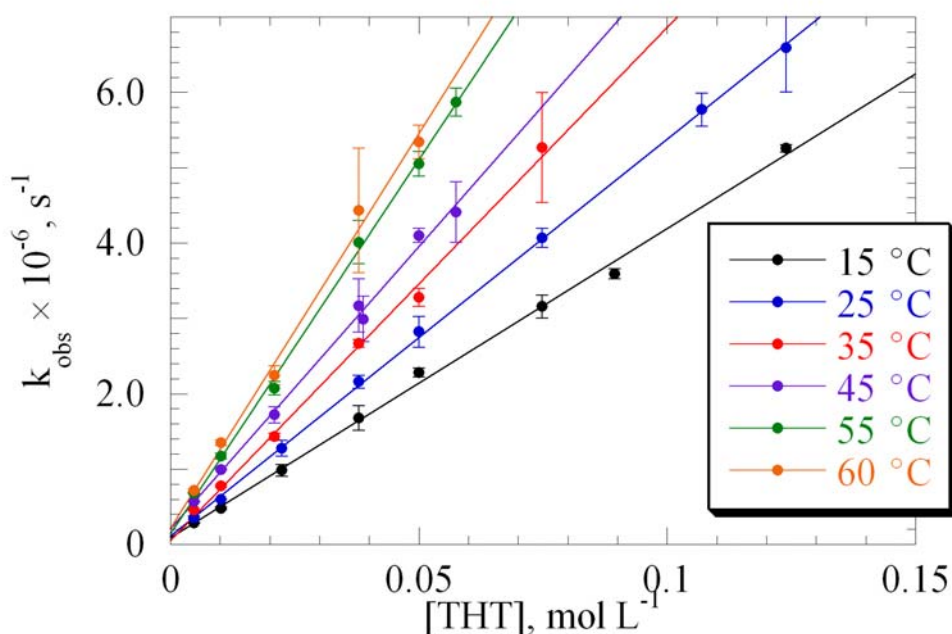
**Fig. S-1**  $k_{\text{obs}}$  as a function of  $[\text{L}]$  for reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with furan over the temperature range 15 EC - 55EC. The solid lines represent least-squares linear fits to the data.  $1\sigma$  error bars are shown.



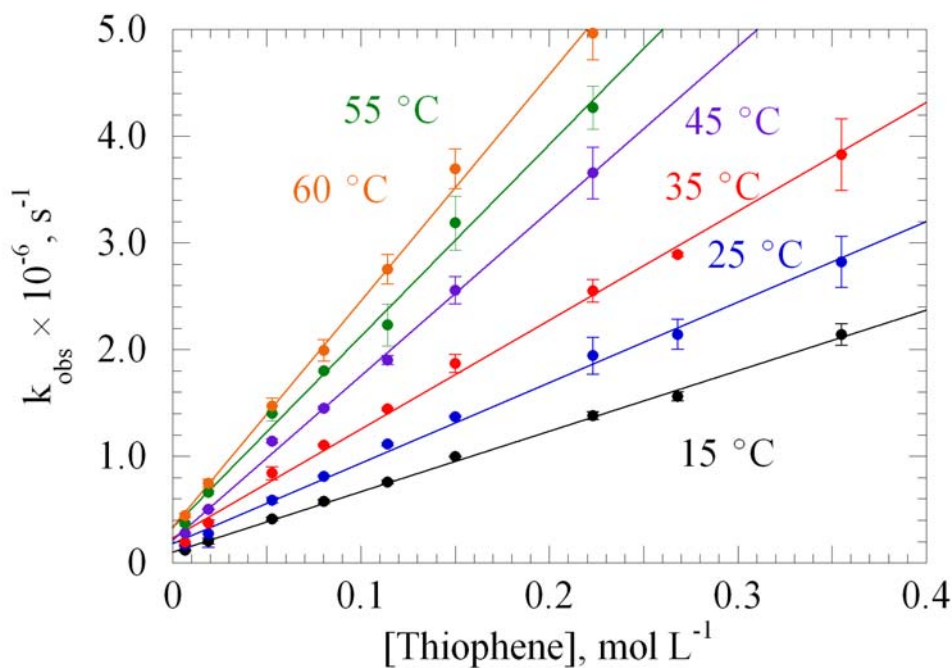
**Fig. S-2**  $k_{\text{obs}}$  as a function of  $[\text{L}]$  for reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with pyrrolidine over the temperature range 15 EC - 60 EC. The solid lines are least-squares linear fits to the data.  $1\sigma$  error bars are shown.



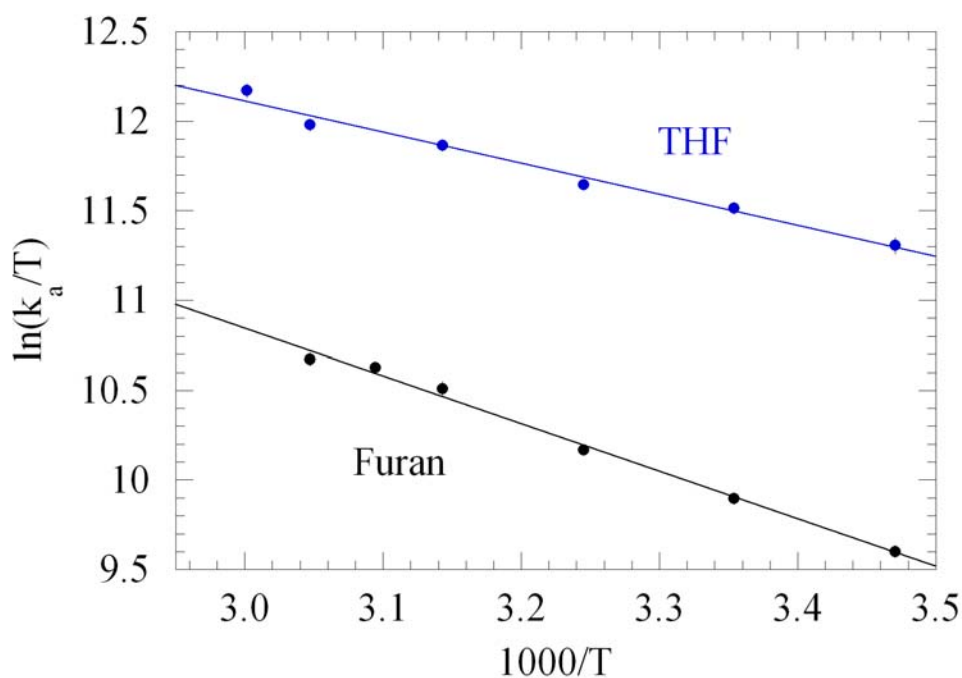
**Fig. S-3**  $k_{\text{obs}}$  as a function of the nominal concentration of [L] for reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with pyrrole, measured over the temperature range 15 EC - 60 EC. The dashed lines indicate least-squares linear fits to the data at 15 EC - 35 EC for  $[\text{Pyr}]_{\text{nom}} < 0.25 \text{ mol L}^{-1}$  (see the text). The solid lines represent least-squares linear fits to the entire data sets at higher temperatures.  $1\sigma$  error bars are shown.



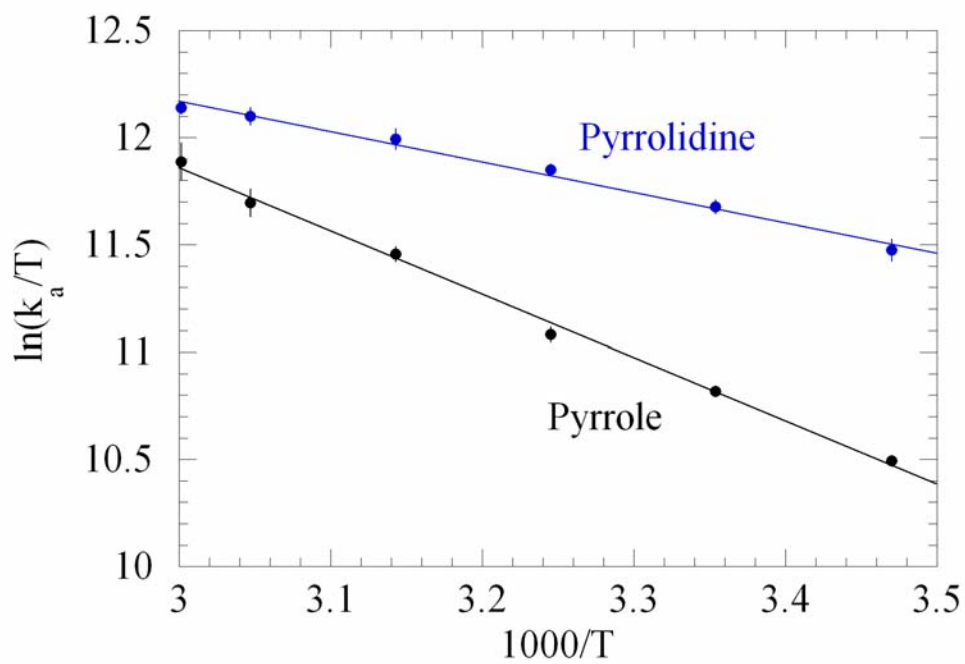
**Fig. S-4**  $k_{\text{obs}}$  as a function of [L] for the reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with tetrahydrothiophene over the temperature range 15 EC - 60 EC. The solid lines represent least-squares linear fits to data.  $1\sigma$  error bars are shown.



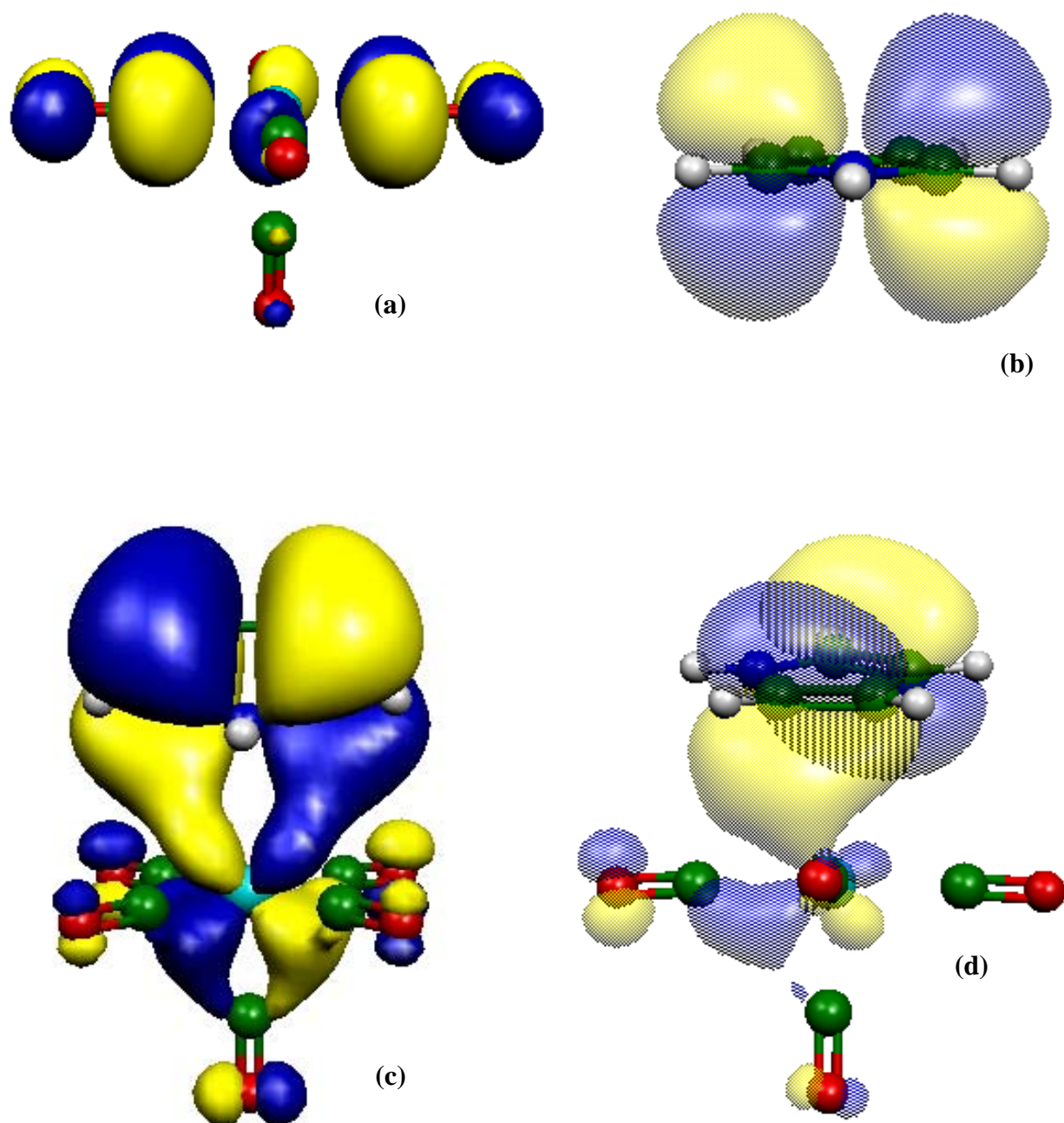
**Fig. S-5**  $k_{\text{obs}}$  as a function of  $[L]$  for the reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with thiophene over the temperature range 15 EC - 60 EC. The solid lines represent least-squares linear fits to data.  $1\sigma$  error bars are shown.



**Fig. S-6** Eyring plot for the reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with THF (blue) and furan (black). The solid lines represent least-squares fits to the data, yielding the activation parameters given in Table 3 of the text.



**Fig. S-7** Eyring plot for the reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with pyrrolidine (blue) and pyrrole (black). The solid lines represent least-squares fits to the data, yielding the activation parameters given in Table 3 of the text.



**Fig. S-8** MO structures as calculated by DFT. All MOs shown in this figure were calculated by using the B3LYP functional, the Stuttgart 1997 ECP basis set for Cr, and the 6-311++G(2d,2p) basis set for the other atoms. **(a)** Unoccupied MO (LUMO+2) of  $\text{Cr}(\text{CO})_5$ ; **(b)** HOMO of pyrrole; **(c)** MO (HOMO-3) of  $\text{Cr}(\text{CO})_5(\eta^1\text{-pyrrole})$  showing the metal-ligand interaction; **(d)** MO (HOMO-3) of  $\text{Cr}(\text{CO})_5(\eta^2\text{-pyrrole})$  showing the metal-ligand interaction.

**Table S-1** Pseudo-first order rate constants  $k_{\text{obs}}$  ( $\times 10^{-6}$ ,  $\text{s}^{-1}$ ) for reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with THF ( $1\sigma$  uncertainties in parentheses)

[THF], $\text{mol L}^{-1}$	15 EC	25 EC	35 EC	45 EC	55 EC	60 EC
0.0050	0.221 (0.021)	0.284 (0.025)	0.363 (0.006)	0.446 (0.008)	0.563 (0.013)	0.593 (0.051)
<b>0.0110</b>	<b>0.404 (0.011)</b>	<b>0.516 (0.010)</b>	<b>0.718 (0.07)</b>	<b>0.831 (0.030)</b>	<b>1.06 (0.03)</b>	<b>1.11 (0.02)</b>
0.0201	0.746 (0.032)	0.811 (0.015)	1.13 (0.04)	1.52 (0.029)	1.74 (0.11)	1.90 (0.07)
<b>0.0212</b>	<b>0.738 (0.016)</b>	<b>0.920 (0.094)</b>	<b>1.14 (0.03)</b>	<b>1.37 (0.041)</b>	<b>1.61 (0.04)</b>	<b>1.73 (0.10)</b>
0.0407	1.00 (0.01)	1.20 (0.03)	1.89 (0.11)	2.34 (0.075)	2.61 (0.22)	2.80 (0.21)
<b>0.0518</b>	<b>1.36 (0.04)</b>	<b>1.73 (0.09)</b>	<b>2.15 (0.17)</b>	<b>2.66 (0.22)</b>	<b>3.31 (0.18)</b>	<b>3.56 (0.22)</b>
0.0691	1.99 (0.23)	2.32 (0.25)	2.70 (0.26)	3.53 (0.32)	3.98 (0.54)	5.08 (0.32)
<b>0.0789</b>	<b>1.87 (0.09)</b>	<b>2.54 (0.12)</b>	<b>3.20 (0.23)</b>	<b>3.83 (0.39)</b>	<b>4.81 (0.54)</b>	<b>5.35 (0.37)</b>
0.0916	2.28 (0.11)	2.78 (0.05)	3.55 (0.16)	4.35 (0.61)	5.13 (0.58)	
<b>0.102</b>	<b>2.59 (0.22)</b>	<b>3.30 (0.18)</b>	<b>3.86 (0.14)</b>	<b>5.17 (0.22)</b>		

**Table S-2** Pseudo-first order rate constants  $k_{\text{obs}}$  ( $\times 10^{-6}$ ,  $\text{s}^{-1}$ ) for reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with furan ( $1\sigma$  uncertainties in parentheses)

[Furan], $\text{mol L}^{-1}$	15 EC	25 EC	35 EC	45 EC	50 EC	55 EC
0.0162	0.200 (0.004)	0.242 (0.006)	0.421 (0.019)	0.573 (0.03)	0.686 (0.021)	0.821 (0.022)
<b>0.0316</b>	<b>0.281 (0.014)</b>	<b>0.420 (0.034)</b>	<b>0.628 (0.054)</b>	<b>0.915 (0.12)</b>	<b>1.05 (0.08)</b>	<b>1.23 (0.11)</b>
0.0534	0.414 (0.016)	0.590 (0.011)	0.828 (0.011)			
<b>0.0749</b>	<b>0.390 (0.017)</b>	<b>0.773 (0.062)</b>	<b>1.06 (0.06)</b>	<b>1.48 (0.04)</b>	<b>1.70 (0.04)</b>	<b>1.77 (0.10)</b>
0.116	0.700 (0.036)	0.971 (0.025)	1.43 (0.46)	1.83 (0.07)	2.15 (0.08)	2.54 (0.15)
<b>0.203</b>	<b>1.08 (0.01)</b>	<b>1.49 (0.06)</b>	<b>1.95 (0.15)</b>	<b>2.72 (0.03)</b>	<b>3.34 (0.03)</b>	<b>3.50 (0.30)</b>
0.221	1.21 (0.05)	1.32 (0.30)	1.97 (0.07)	2.75 (0.03)		3.60 (0.70)
<b>0.262</b>		<b>1.95 (0.16)</b>	<b>2.64 (0.30)</b>	<b>3.51 (0.49)</b>	<b>4.08 (0.32)</b>	<b>4.68 (0.30)</b>
0.308	1.45 (0.08)	1.87 (0.28)	3.00 (0.07)			
<b>0.344</b>	<b>1.60 (0.06)</b>	<b>2.22 (0.11)</b>	<b>3.10 (0.12)</b>	<b>4.69 (0.32)</b>	<b>5.12 (0.60)</b>	<b>5.49 (0.59)</b>

**Table S-3** Pseudo-first order rate constants  $k_{\text{obs}}$  ( $\times 10^{-6}$ ,  $\text{s}^{-1}$ ) for reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with pyrrolidine ( $1\sigma$  uncertainties in parentheses)

[L], $\text{mol L}^{-1}$	15 EC	25 EC	35 EC	45 EC	55 EC	60 EC
0.0052	0.240 (0.012)	0.264 (0.006)	0.323 (0.008)	0.391 (0.027)	0.461 (0.018)	0.528 (0.020)
<b>0.0153</b>	<b>0.587 (0.039)</b>	<b>0.670 (0.076)</b>	<b>0.773 (0.031)</b>	<b>0.885 (0.043)</b>	<b>1.01 (0.03)</b>	<b>1.09 (0.05)</b>
0.0300	0.958 (0.030)	1.14 (0.05)	1.29 (0.07)	1.55 (0.091)	1.94 (0.13)	2.06 (0.12)
<b>0.0453</b>	<b>1.46 (0.15)</b>	<b>1.77 (0.13)</b>	<b>2.07 (0.19)</b>	<b>2.69 (0.39)</b>	<b>3.14 (0.31)</b>	<b>3.28 (0.17)</b>
0.0603	2.01 (0.20)	2.32 (0.25)	2.80 (0.43)	3.04 (0.28)	3.58 (0.35)	3.94 (0.78)
<b>0.0753</b>	<b>2.12 (0.36)</b>	<b>2.61 (0.11)</b>		<b>3.77 (0.26)</b>		<b>4.74 (0.16)</b>
0.0900	2.61 (0.29)	3.31 (0.23)	3.93 (0.48)	4.90 (0.41)	5.46 (0.67)	5.88 (0.18)



**Table S-4** Pseudo-first order rate constants  $k_{\text{obs}}$  ( $\times 10^{-6}$ ,  $\text{s}^{-1}$ ) for reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with pyrrole ( $1\sigma$  uncertainties in parentheses)

[Pyrrole], $\text{mol L}^{-1}$	15 EC	25 EC	35 EC	45 EC	55 EC	60 EC
0.0100	0.176 (0.008)	0.242 (0.034)	0.336 (0.048)	0.386 (0.037)	0.487 (0.17)	0.559 (0.13)
<b>0.0200</b>	<b>0.273 (0.012)</b>	<b>0.350 (0.012)</b>	<b>0.455 (0.015)</b>	<b>0.522 (0.046)</b>	<b>0.634 (0.038)</b>	<b>0.751 (0.73)</b>
0.0356	0.511 (0.009)	0.670 (0.017)	0.885 (0.017)	1.14 (0.022)	1.52 (0.07)	1.54 (0.021)
<b>0.0514</b>	<b>0.621 (0.018)</b>	<b>0.847 (0.10)</b>	<b>1.13 (0.15)</b>	<b>1.52 (0.078)</b>	<b>2.41 (0.15)</b>	<b>1.72 (0.35)</b>
0.0601				1.84 (0.036)	3.06 (0.24)	2.68 (0.29)
<b>0.0801</b>				<b>2.39 (0.058)</b>	<b>3.84 (0.68)</b>	<b>3.47 (0.21)</b>
0.0812	0.956 (0.026)	1.355 (0.053)	1.91 (0.19)	2.84 (0.23)	4.04 (0.24)	
<b>0.100</b>				<b>2.92 (0.10)</b>		<b>4.80 (0.24)</b>
0.100	1.19 (0.08)	1.45 (0.11)	2.10 (0.19)	2.99 (0.53)	4.60 (0.25)	5.44 (2.20)
<b>0.120</b>				<b>3.41 (0.10)</b>	<b>5.21 (0.80)</b>	<b>5.67 (0.52)</b>
0.126	1.40 (0.04)	2.02 (0.09)	2.79 (0.19)	4.17 (0.19)	5.47 (0.51)	6.19 (0.99)
<b>0.150</b>				<b>4.31 (0.17)</b>	<b>6.31 (1.09)</b>	
0.151	1.69 (0.10)	2.31 (0.20)	3.32 (0.40)			
<b>0.201</b>	<b>2.14 (0.11)</b>	<b>3.08 (0.28)</b>	<b>3.97 (0.56)</b>	<b>6.13 (0.66)</b>		
0.252	2.51 (0.11)	3.67 (0.35)	5.75 (0.83)			
<b>0.351</b>	<b>3.17 (0.07)</b>	<b>4.91 (0.47)</b>				
0.450	4.16 (0.38)	5.98 (1.15)				
<b>0.500</b>	<b>4.95 (0.59)</b>					
0.558	4.85 (0.68)					
<b>0.750</b>	<b>6.04 (0.04)</b>					

**Table S-5** Pseudo-first order rate constants  $k_{\text{obs}}$  ( $\times 10^{-6}$ ,  $\text{s}^{-1}$ ) for reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with tetrahydrothiophene ( $1\sigma$  uncertainties in parentheses)

$[\text{THT}]_0$ , $\text{mol L}^{-1}$	15 EC	25 EC	35 EC	45 EC	55 EC	60 EC
0.00480	0.287 (0.007)	0.350 (0.020)	0.452 (0.006)	0.567 (0.006)	0.678 (0.007)	0.716 (0.014)
<b>0.0102</b>	<b>0.481 (0.015)</b>	<b>0.597 (0.003)</b>	<b>0.777 (0.015)</b>	<b>0.994 (0.013)</b>	<b>1.17 (0.04)</b>	<b>1.35 (0.04)</b>
0.0210			1.43 (0.045)	1.72 (0.11)	2.07 (0.09)	2.24 (0.13)
<b>0.0224</b>	<b>0.981 (0.078)</b>	<b>1.28 (0.11)</b>				
0.0380	1.67 (0.17)	2.16 (0.09)	2.67 (0.05)	3.17 (0.35)	4.01 (0.29)	4.43 (0.83)
<b>0.0388</b>				<b>2.99 (0.30)</b>		
0.0500	2.28 (0.06)	2.82 (0.20)	3.28 (0.12)	4.10 (0.09)	5.05 (0.16)	5.34 (0.22)
<b>0.0574</b>				<b>4.41 (0.40)</b>	<b>5.87 (0.19)</b>	
0.0749	3.16 (0.15)	4.07 (0.13)	5.27 (0.73)			
<b>0.0895</b>	<b>3.59 (0.07)</b>					
0.107		5.77 (0.22)				
<b>0.124</b>	<b>5.26 (0.05)</b>	<b>6.59 (0.59)</b>				

**Table S-6** Pseudo-first order rate constants  $k_{\text{obs}}$  ( $\times 10^{-6}$ ,  $\text{s}^{-1}$ ) for reaction of  $\text{Cr}(\text{CO})_5(\text{CyH})$  with thiophene ( $1\sigma$  uncertainties in parentheses)

[Thiophene], mol L <sup>-1</sup>	15 EC	25 EC	35 EC	45 EC	55 EC	60 EC
0.0068	0.117 (0.008)	0.165 (0.008)	0.192 (0.001)	0.275 (0.005)	0.368 (0.020)	0.443 (0.018)
<b>0.0190</b>	<b>0.205 (0.026)</b>	<b>0.273 (0.013)</b>	<b>0.372 (0.001)</b>	<b>0.503 (0.005)</b>	<b>0.661 (0.015)</b>	<b>0.746 (0.020)</b>
0.0529	0.410 (0.015)	0.585 (0.028)	0.839 (0.063)	1.14 (0.02)	1.40 (0.07)	1.47 (0.06)
<b>0.0804</b>	<b>0.573 (0.019)</b>	<b>0.811 (0.007)</b>	<b>1.10 (0.01)</b>	<b>1.45 (0.01)</b>	<b>1.80 (0.01)</b>	<b>1.99 (0.01)</b>
0.114	0.754 (0.016)	1.11 (0.01)	1.44 (0.01)	1.90 (0.04)	2.23 (0.20)	2.75 (0.10)
<b>0.150</b>	<b>0.996 (0.006)</b>	<b>1.37 (0.10)</b>	<b>1.87 (0.08)</b>	<b>2.55 (0.13)</b>	<b>3.19 (0.25)</b>	<b>3.69 (0.24)</b>
0.223	1.38 (0.036)	1.94 (0.17)	2.55 (0.11)	3.65 (0.24)	4.27 (0.20)	4.96 (0.50)
<b>0.268</b>	<b>1.56 (0.042)</b>	<b>2.14 (0.14)</b>	<b>2.89 (0.02)</b>			
0.355	2.14 (0.10)	2.82 (0.24)	3.83 (0.33)			