

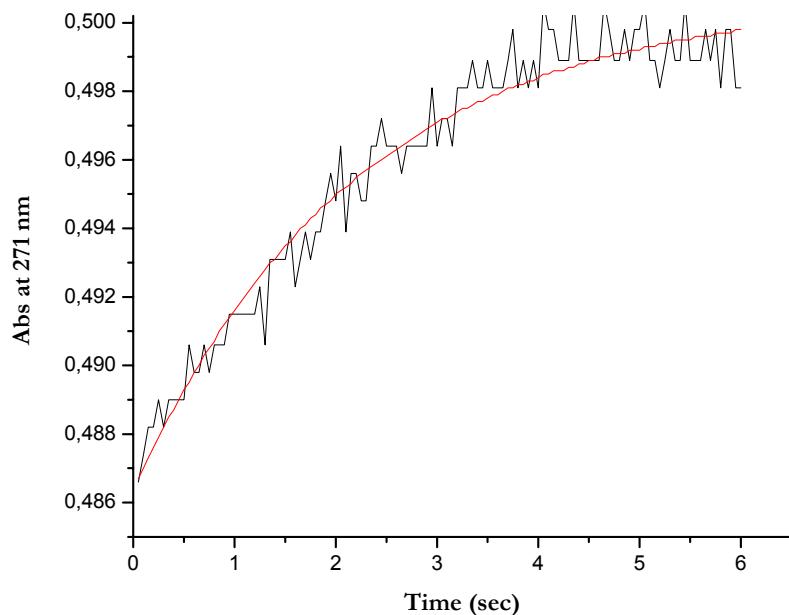
Supporting Information

Reaction of $[\text{Ru}^{\text{III}}(\text{edta})(\text{H}_2\text{O})]^-$ with H_2O_2 in aqueous solution. Kinetic and mechanistic investigation

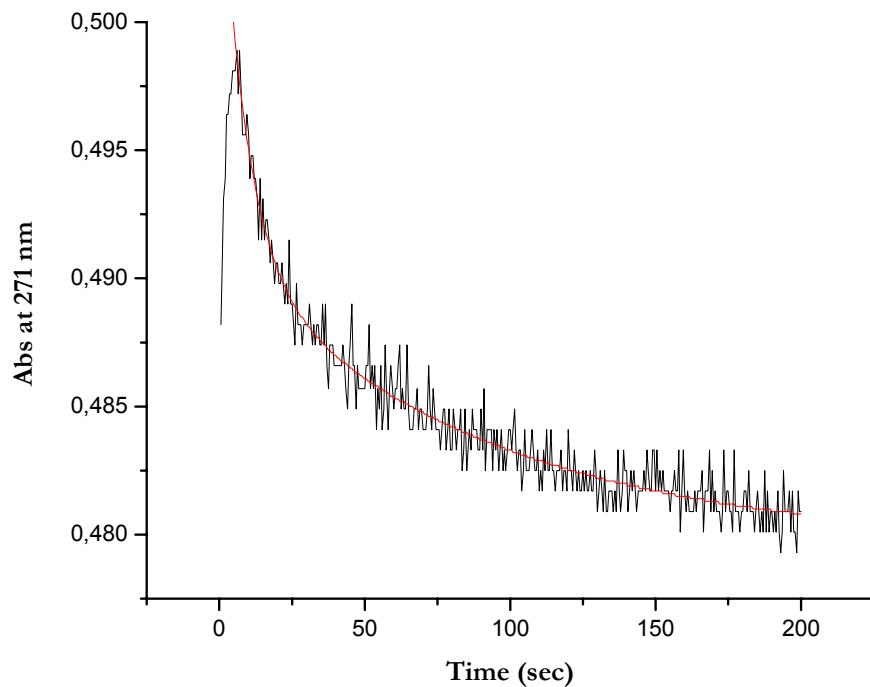
Debabrata Chatterjee,* Anannya Mitra and Rudi van Eldik*

Kinetic traces at 271 nm

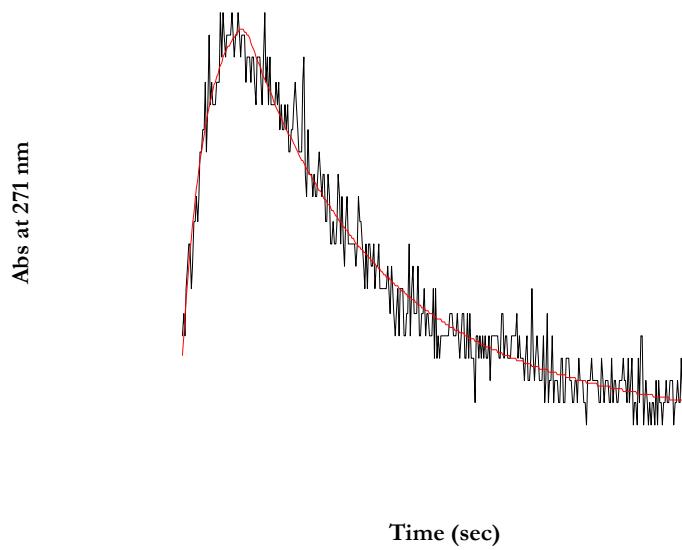
Experimental conditions: $[\text{Ru}^{\text{III}}(\text{edta})] = 5 \times 10^{-5}$ M, $[\text{H}_2\text{O}_2] = 0.015$ M, Temp. = 4.5 °C, pH = 5.1



$$k_{\text{obs}} = 0.34 \text{ s}^{-1}$$



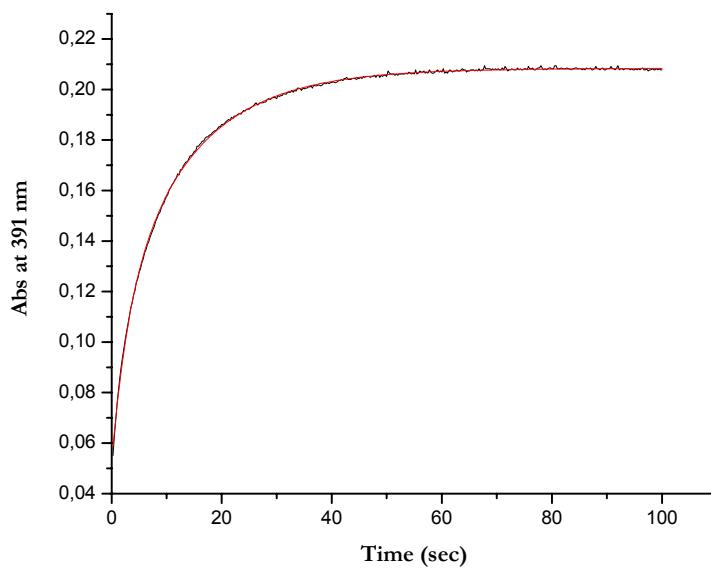
Two-exponential fit: $k_{\text{fast}} = 0.086 \text{ s}^{-1}$ and $k_{\text{slow}} = 0.007 \text{ s}^{-1}$



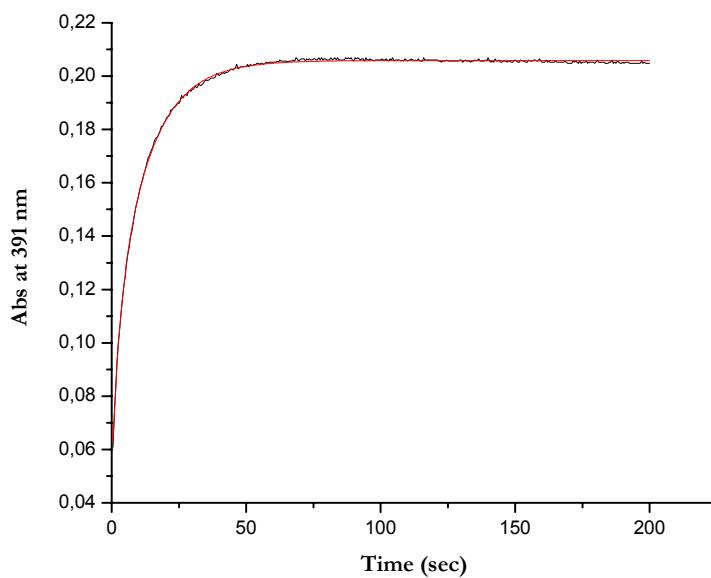
Two-exponential fit: $k_{\text{fast}} = 0.36 \text{ s}^{-1}$ and $k_{\text{slow}} = 0.091 \text{ s}^{-1}$

Kinetic traces at 391 nm

Experimental conditions: $[Ru^{III}(\text{edta})] = 5 \times 10^{-5} \text{ M}$, $[\text{H}_2\text{O}_2] = 0.015 \text{ M}$, Temp. = $4.5 \text{ }^{\circ}\text{C}$, pH = 5.1



Two-exponential fit: $k_{\text{fast}} = 0.33 \text{ s}^{-1}$ and $k_{\text{slow}} = 0.075 \text{ s}^{-1}$



Two-exponential fit: $k_{\text{fast}} = 0.34 \text{ s}^{-1}$ and $k_{\text{slow}} = 0.080 \text{ s}^{-1}$