## [Si(SiMe<sub>3</sub>)<sub>3</sub>]<sub>6</sub>Ge<sub>18</sub>M (M = Cu, Ag, Au): Metalloid cluster compounds as unusual building blocks for a supramolecular chemistry.

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## **Supporting Information**

1.) Gas phase investigations of  $[AgGe_{18}R_6]^-$  (R = Si(SiMe<sub>3</sub>)<sub>3</sub>) **3** 

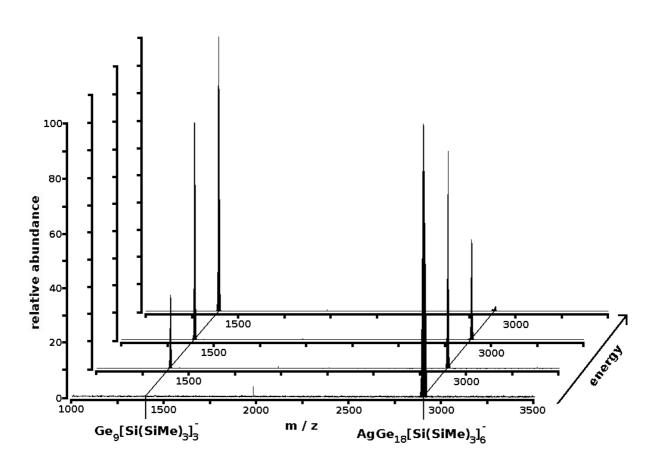


Figure S1: The FT/ICR-mass spectra obtained from a solution of  $[AgGe_{18}R_6]^-$  (R = Si(SiMe<sub>3</sub>)<sub>3</sub>) **3** in thf using ESI as ionisation method. In the dissociation experiments (SORI-CAD) the parent ion **3** dissociates to give the fragment ion Ge<sub>9</sub>R<sub>3</sub><sup>-</sup> (R = Si(SiMe<sub>3</sub>)<sub>3</sub>) **1**. The relative abundance of the fragment ion **1** depends on the translation energy of the parent ion **3** prior to fragementation. Low energy gives less fragement ion than higher.



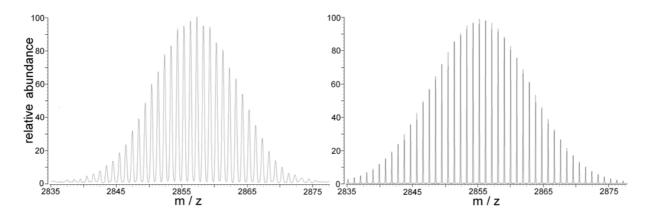


Figure S2: Measured (left) and calculated (right) isotopic pattern of the molecular peak of  ${CuGe_{18}[Si(SiMe_3)_3]_6}^- 3$ .

3.) Schematic presentation of subsequent reactions of  $[AuGe_{18}R_6]^- 2$ ,  $[AgGe_{18}R_6]^- 3$ ,  $[CuGe_{18}R_6]^- 4$  (R = Si(SiMe\_3)\_3.

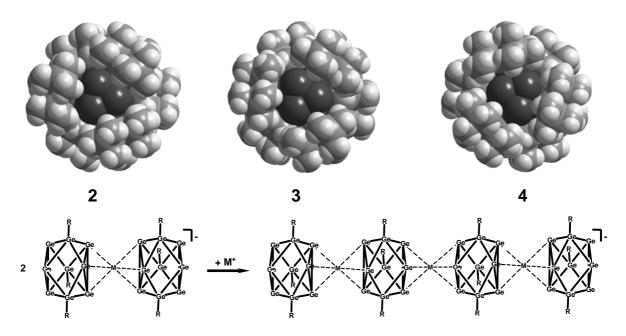


Figure S3: Top: Space filling models of  $[AuGe_{18}R_6]^- 2$ ,  $[AgGe_{18}R_6]^- 3$ ,  $[CuGe_{18}R_6]^- 4$  (R = Si(SiMe<sub>3</sub>)<sub>3</sub>; view along the threefold axis). Bottom: Schematic presentation of the formation of an M<sub>3</sub>Ge<sub>36</sub>R<sub>12</sub><sup>-</sup> unit.