

**Supplementary Material For Dalton Trans. Manuscript (Version: 13 March 2008)**

**“Gas Phase Synthesis, Structure and Unimolecular Reactivity of Silver Iodide Cluster Cations,  $\text{Ag}_n\text{I}_m^+$  ( $n=2-5$ ,  $0 < m < n$ ).” by George N. Khairallah and Richard A. J. O’Hair\*, School of Chemistry and Bio21 Institute of Molecular Science and Biotechnology, The University of Melbourne, Victoria 3010, AUSTRALIA**

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**List of Supplementary Material:**

- (A) Cartesian Coordinates for structures shown in Figures 3 and 4 (pages S1-S33)(The point groups of the output geometries (using GaussView with tolerance 0.01) are listed in brackets);
- (B) Supplementary Figures S1 and S2 (pages S34-S36);
- (C) Supplementary Tables S1, S2, and S3 (pages S37-S44).

Coordinates for  $\text{Ag}_x\text{I}_y^{0/+}$  (SDD basis set, all energies are in Hartrees; where there are several isomers, relative energies are in kcal mol<sup>-1</sup>):

**AgI neutral: (Used in Figure 3a)**

E(B3LYP) = -158.4732823; Zero-point correction = 0.000436

Ag	0.000000	0.000000	-1.393829
I	0.000000	0.000000	1.236037

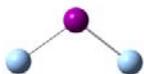
**Ag<sub>2</sub> neutral:**

E(B3LYP) = -294.0430056; Zero-point correction = 0.000406

Ag	0.000000	0.000000	1.298211
Ag	0.000000	0.000000	-1.298211

**Ag<sub>2</sub>I<sup>+</sup> (C<sub>2v</sub>):**

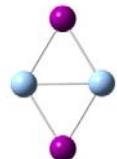
**Unique isomer (Used in Figure 4a)**



E(B3LYP) = -305.25137; Zero-point correction = 0.000816

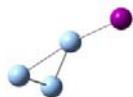
Ag	2.158825	-0.595263	0.000000
Ag	-2.158825	-0.595265	0.000000
I	0.000000	1.055751	0.000000

**Ag<sub>2</sub>I<sub>2</sub> neutral (D<sub>2h</sub>): (Used in Figure 3b)**



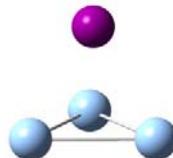
E(B3LYP) = -317.0025911; Zero-point correction = 0.001390

Ag	0.000000	1.430794	0.000000
Ag	0.000000	-1.430794	0.000000
I	0.000000	0.000000	2.438866
I	0.000000	0.000000	-2.438866

**Ag<sub>3</sub>I<sup>+</sup>:****a) Unique isomer (Used in Figure 4b)**

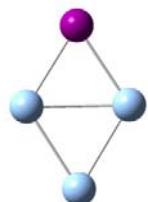
E(B3LYP) = -452.262528; Zero-point correction = 0.001215

Ag	1.948597	-1.351896	-0.109717
Ag	1.948559	1.351920	-0.109707
Ag	-0.408997	-0.000026	0.412142
I	-3.093273	0.000002	-0.170901

**b) In and Out (C<sub>3v</sub>):**

E(B3LYP) = -452.2770734; Zero-point correction = 0.001158

Ag	0.000000	1.648187	-0.670612
Ag	1.427372	-0.824093	-0.670612
Ag	-1.427372	-0.824093	-0.670612
I	0.000000	0.000000	1.784082

**c) In and Out (C<sub>2v</sub>):**

E(B3LYP) = -452.2805541; Zero-point correction = 0.001250

Ag	0.00000000	-0.00000000	2.46753230
Ag	0.00000000	-1.48552515	0.05113636
Ag	-0.00000000	1.48552515	0.05113636
I	0.00000000	-0.00000000	-2.27888370

**Ag<sub>3</sub>I<sub>2</sub><sup>+</sup> (C<sub>2h</sub>):****a) Isomer 1 (C<sub>2h</sub>) (Used in Figure 4c)**

The following two different input guesses



Converged to the same structure:

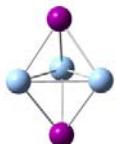
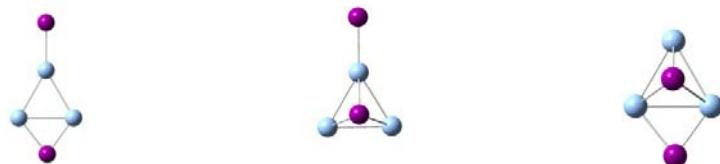


E(B3LYP) = -463.7818205; Zero-point correction = 0.001758

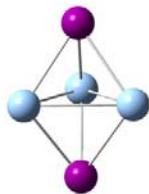
**Relative energy = 0 kcal mol<sup>-1</sup>**

Ag	4.118484	0.832083	0.088397
Ag	-0.000002	-0.000033	-0.049426
Ag	-4.118511	-0.832042	0.088414
I	-2.421270	1.255075	-0.056505
I	2.421296	-1.255083	-0.056459

b) Isomer 2 ( $D_{3h}$ ): The following four different input guesses



Converged to the same structure:



$E(B3LYP) = -463.7798878$ ; Zero-point correction = 0.001768

**Relative energy = 1.2 kcal mol<sup>-1</sup>**

Ag	0.000000	1.797790	0.000000
Ag	-1.556932	-0.898895	0.000000
Ag	1.556932	-0.898895	0.000000
I	0.000000	0.000000	2.279405
I	0.000000	0.000000	-2.279405

c) **Isomer 3:** has 1 small imaginary frequency  $-7.3473 \text{ cm}^{-1}$



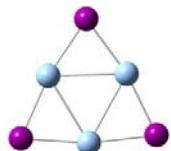
$E(B3LYP) = -463.7793828$ ; Zero-point correction = 0.001744

**Relative energy = 1.5 kcal mol<sup>-1</sup>**

I	2.718192	0.996524	0.000000
Ag	0.000000	0.717177	0.000000
I	-2.718203	0.996487	0.000000
Ag	-3.759736	-1.482330	0.000000
Ag	3.759748	-1.482284	0.000000

**Ag<sub>3</sub>I<sub>3</sub> neutral:**

a) **Isomer 1 (D<sub>3h</sub>) (Used in Figure 3c)**



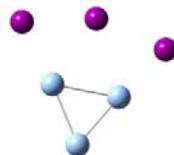
$E(B3LYP) = -475.552877$ ; Zero-point correction = 0.002637

**Relative energy = 0 kcal mol<sup>-1</sup>**

Ag	0.000000	1.817581	0.000000
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Ag	1.574072	-0.908791	0.000000
Ag	-1.574072	-0.908791	0.000000
I	0.000000	-3.146658	0.000000
I	2.725086	1.573329	0.000000
I	-2.725086	1.573329	0.000000

b) **Isomer 2 ( $C_{2v}$ )**

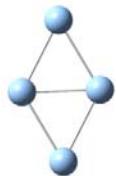


$E(B3LYP) = -475.429427$ ; Zero-point correction = 0.002286

**Relative energy = 77.2 kcal mol<sup>-1</sup>**

Ag	0.000000	0.000000	-3.293988
Ag	0.000000	1.418989	-1.001714
Ag	0.000000	-1.418989	-1.001714
I	0.000000	0.000000	2.043908
I	0.000000	3.070964	1.326900
I	0.000000	-3.070964	1.326900

**$\text{Ag}_4^+$  ( $D_{2h}$ ):**



$E(B3LYP) = -587.8724978$ ; Zero-point correction = 0.001283

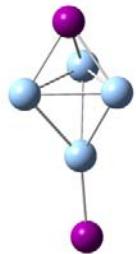
Ag	0.000000	1.350967	0.000000
Ag	0.000000	-1.350967	0.000000
Ag	0.000000	0.000000	2.462614
Ag	0.000000	0.000000	-2.462614

**$\text{Ag}_4\text{I}^+$ :** four isomers have been previously reported in: Khairallah, G.N.; O'Hair, R. A. J., "Activation of the C-I and C-OH Bonds of 2-Iodoethanol by Gas Phase Silver Cluster Cations Yields Subvalent Silver-Iodide and -Hydroxide Cluster Cations.", *Dalton Trans.*, 2007, 3149 - 3157. **The structure shown in Figure 4e is taken from this paper.**

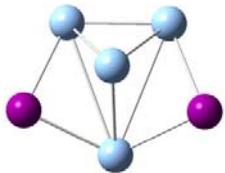
**$\text{Ag}_4\text{I}_2^+$ :**

a) **Isomer 1 ( $C_s$ ) (Used in Figure 4e):**

**The following input guess**



Converged to:

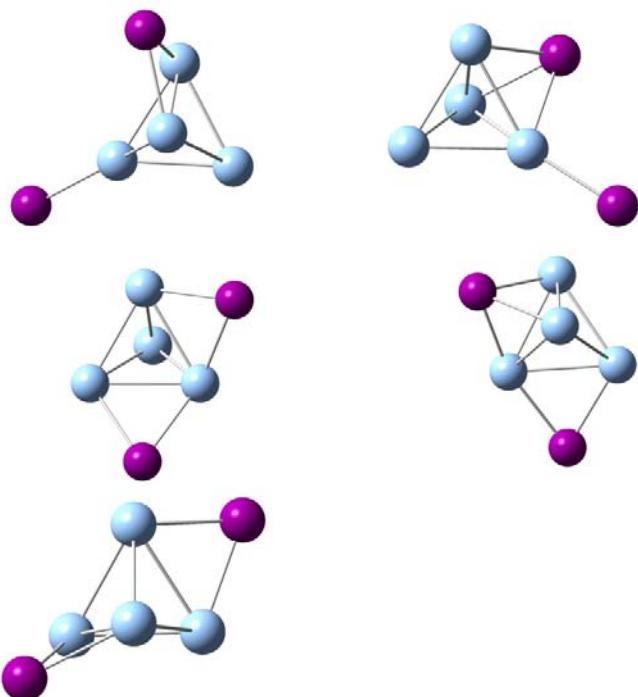


E(B3LYP) = -610.8151529; Zero-point correction = 0.002226

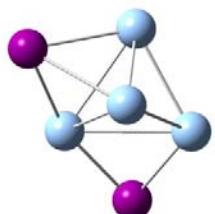
**Relative energy = 0 kcal mol<sup>-1</sup>**

I	-2.494500	-0.834081	0.101026
Ag	-1.406880	1.820034	-0.400834
Ag	1.408128	1.819757	-0.400329
Ag	0.000026	0.059964	1.433184
Ag	-0.000620	-1.817211	-0.859580
I	2.493920	-0.835345	0.100771

b) **Isomer 2 (C<sub>s</sub>) : The following five different input guesses**



**Converged to the same structure:**

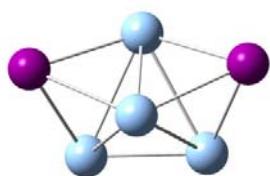


E(B3LYP) = -610.813766; Zero-point correction = 0.002325

**Relative energy = 0.9 kcal mol<sup>-1</sup>**

I	2.934328	-0.808843	-0.000841
Ag	0.273600	-1.406917	-0.001328
Ag	-1.070287	1.028614	1.407446
Ag	1.399990	1.510970	-0.000251
Ag	-1.073757	1.033479	-1.403997
I	-2.517133	-1.112080	-0.000817

c) **Isomer 3:** has 1 small imaginary frequency -21.2 cm<sup>-1</sup>



E(B3LYP) = -610.8135422; Zero-point correction = 0.002130

**Relative energy = 1.0 kcal mol<sup>-1</sup>**

I	2.515394	-0.922171	0.003282
Ag	1.385517	1.729986	-0.007721
Ag	-1.391184	1.727205	-0.006344
Ag	0.001050	-0.698839	-1.448860
Ag	0.001286	-0.674340	1.456125
I	-2.512440	-0.925915	0.002749

d) **Isomer 4:** has 1 small imaginary frequency -4.9034 cm<sup>-1</sup>



E(B3LYP) = -610.8000989; Zero-point correction = 0.002088

**Relative energy = 9.4 kcal mol<sup>-1</sup>**

Ag	1.541014	-1.849522	0.000000
Ag	-1.291008	-2.036170	0.000000
Ag	0.000000	0.509479	0.000000
I	-1.192879	3.033989	0.000000
Ag	0.791190	4.853811	0.000000
I	0.269554	-4.344311	0.000000

e) **Isomer 5:** has 1 small imaginary frequency -1.3667 cm<sup>-1</sup>

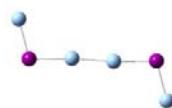


E(B3LYP) = -610.7983978; Zero-point correction = 0.002197

**Relative energy = 10.5 kcal mol<sup>-1</sup>**

Ag	-1.084803	-0.625709	-1.269039
Ag	1.088624	-1.261557	0.640860
Ag	1.094728	1.274003	-0.621291
Ag	-1.098221	0.636323	1.266317
I	-3.502468	-0.010063	-0.007030
I	3.502178	-0.010386	-0.007910

f) **Isomer 6 (C<sub>2h</sub>):** has 1 small imaginary frequency -5.9439 cm<sup>-1</sup>

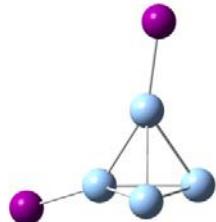


$E(B3LYP) = -610.7959515$ ; Zero-point correction = 0.001937

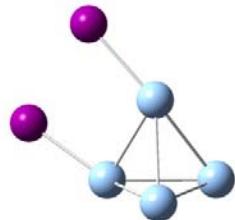
**Relative energy = 11.9 kcal mol<sup>-1</sup>**

Ag	-2.317912	5.010329	0.000000
Ag	0.000000	1.372156	0.000000
Ag	0.000008	-1.372176	0.000000
Ag	2.317913	-5.010308	0.000000
I	0.228699	4.163068	0.000000
I	-0.228707	-4.163069	0.000000

g) **Isomer 7 ( $C_{2v}$ ): The following input guess**



**Converged to:**



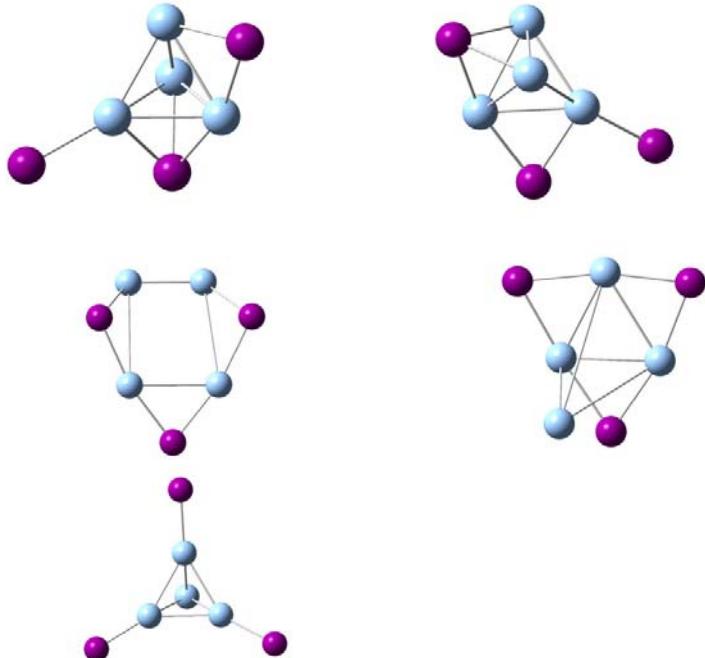
$E(B3LYP) = -610.7862828$ ; Zero-point correction = 0.002424

**Relative energy = 18.2 kcal mol<sup>-1</sup>**

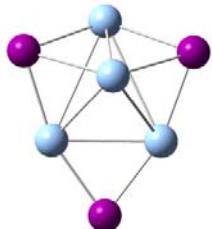
I	-2.383612	-1.762679	0.000035
Ag	0.302651	-1.413319	-0.000193
Ag	2.385233	-0.000168	1.357355
Ag	2.385223	0.000123	-1.357343
Ag	0.302661	1.413367	0.000165
I	-2.383577	1.762676	-0.000022

**$\text{Ag}_4\text{I}_3^+$ :**

a) **Isomer 1 ( $C_s$ ): (Used in Figure 4f)**  
**The following four different input guesses**



**Converged to the same structure:**

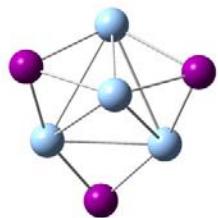


$E(B3LYP) = -622.3239667$ ; Zero-point correction = 0.002983

**Relative energy = 0 kcal mol<sup>-1</sup>**

I	1.424303	2.485671	0.068635
Ag	0.851292	0.000009	1.475822
Ag	2.191163	0.000777	-1.153800
Ag	-1.210871	-1.602753	-0.130863
Ag	-1.212576	1.602371	-0.130134
I	1.426190	-2.484849	0.068868
I	-3.399425	-0.001180	-0.191620

b) **Isomer 2:** has 1 small imaginary frequency  $-15.7800 \text{ cm}^{-1}$



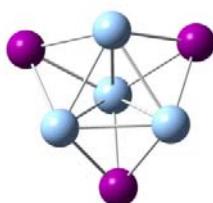
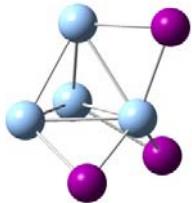
E(B3LYP) = -622.3232744; Zero-point correction = 0.002837

**Relative energy = 0.3 kcal mol<sup>-1</sup>**

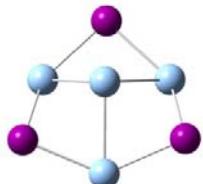
Ag	-1.960030	0.292998	0.000000
Ag	0.765457	1.827772	0.000000
Ag	0.765457	-1.359093	1.478959
Ag	0.765457	-1.359093	-1.478959
I	-1.466305	-2.445954	0.000000
I	2.851720	-0.013552	0.000000
I	-1.683678	2.989290	0.000000

c) **Isomer 3:** has 1 small imaginary frequency -16.3 cm<sup>-1</sup>

**The following two different input guesses**



**Converged to the same structure:**

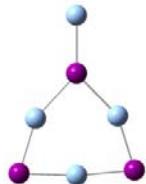


E(B3LYP) = -622.320497; Zero-point correction = 0.002781

**Relative energy = 2.1 kcal mol<sup>-1</sup>**

I	0.853784	2.830987	0.130433
Ag	-1.612335	1.686108	-0.524555
Ag	0.029306	0.027091	1.199633
Ag	-0.676335	-2.214092	-0.491741
Ag	2.290731	0.529968	-0.565444
I	2.040750	-2.167990	0.114894
I	-2.922351	-0.688780	0.093523

d) **Isomer 4:** has 1 small imaginary frequency  $-26.7 \text{ cm}^{-1}$

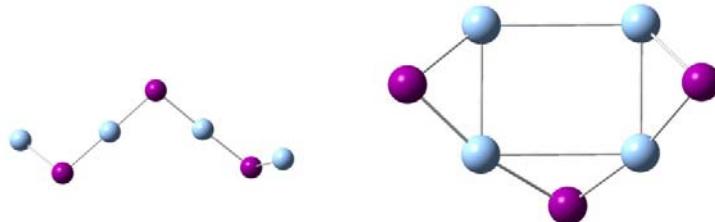


$E(\text{B3LYP}) = -622.306748$ ; Zero-point correction = 0.002701

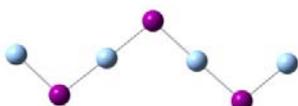
**Relative energy = 10.6 kcal mol<sup>-1</sup>**

Ag	0.000000	1.940896	0.184783
Ag	0.000000	-1.940896	0.184783
Ag	0.000000	0.000000	-2.551873
I	0.000000	0.000000	2.270360
Ag	0.000000	0.000000	5.007237
I	0.000000	2.728701	-2.387744
I	0.000000	-2.728701	-2.387744

e) **Isomer 5 ( $C_{2v}$ ): The following four different input guesses**



**Converged to the same structure:**



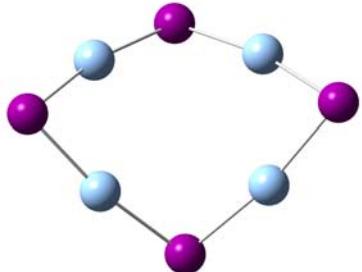
$E(\text{B3LYP}) = -622.3064865$ ; Zero-point correction = 0.002688

**Relative energy = 10.8 kcal mol<sup>-1</sup>**

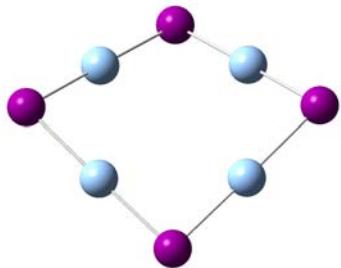
Ag	6.175719	0.283422	0.002481
Ag	2.054454	0.258854	-0.001190
Ag	-2.054500	0.258803	0.000872
Ag	-6.175602	0.283496	-0.001617
I	4.158337	-1.491250	-0.002365
I	-4.158356	-1.491335	0.002128
I	-0.000044	2.020792	-0.000247

**Ag<sub>4</sub>I<sub>4</sub> neutral:**

a) Isomer ( $D_{2d}$ ): (used in figure 3d):  
**The following input guess:**



**Converged to:**

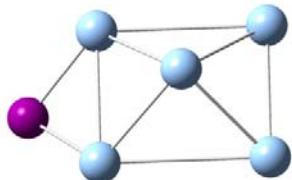


E(B3LYP) = -634.0762313; Zero-point correction = 0.003608

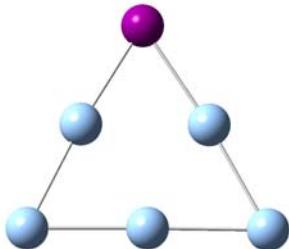
Ag	1.885402	1.885383	-0.000126
Ag	-1.885402	1.885383	-0.000126
Ag	-1.885402	-1.885383	-0.000126
Ag	1.885402	-1.885383	-0.000126
I	0.000000	3.742718	0.599787
I	0.000000	-3.742718	0.599787
I	-3.742900	0.000000	-0.599563
I	3.742900	0.000000	-0.599563

**Ag<sub>5</sub>I<sup>+</sup>:**

a) **Isomer 1 (C<sub>2v</sub>): (Used in Figure 4g)**  
**The following input guess:**



**Converged to:**



E(B3LYP) = -746.39529; Zero-point correction = 0.002320

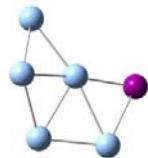
**Relative energy = 0 kcal mol<sup>-1</sup>**

Ag	0.774539	-1.481713	-0.000595
Ag	0.776809	1.480450	0.000732
Ag	-1.687099	2.783356	-0.000360
Ag	-1.691491	-2.780837	0.000347
Ag	-1.670850	0.001351	-0.000053
I	3.102082	-0.002312	-0.000062

b) **Isomer 2 (C<sub>1</sub>): The following two different input guesses**



**Converged to the same structure:**

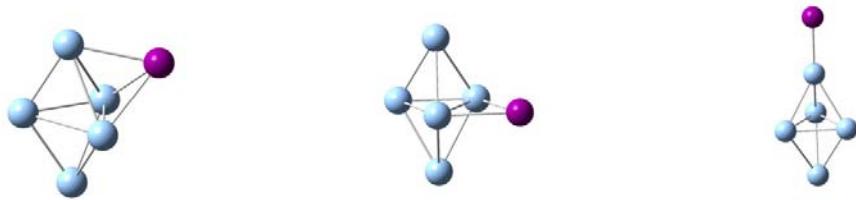


E(B3LYP) = -746.3933769; Zero-point correction = 0.002315

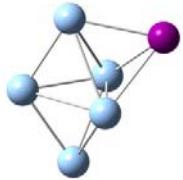
**Relative energy = 1.2 kcal mol<sup>-1</sup>**

Ag	2.603761	-1.699923	-0.341211
Ag	0.123510	2.236835	-0.472330
Ag	0.067878	-0.563589	0.090673
Ag	-2.405241	1.067504	0.114369
Ag	2.379428	0.868555	0.560481
I	-2.455827	-1.693227	0.042582

c) **Isomer 3 (C<sub>s</sub>): The following three different input guesses**



**Converged to the same structure:**

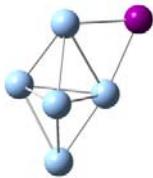


E(B3LYP) = -746.392747; Zero-point correction = 0.002308

**Relative energy = 1.6 kcal mol<sup>-1</sup>**

Ag	1.419429	1.705540	-0.000234
Ag	-1.374498	1.616720	-0.000621
Ag	-0.116729	-0.591510	1.385330
Ag	-0.116991	-0.593041	-1.385008
Ag	-2.569089	-0.975318	0.000429
I	2.445666	-1.030800	0.000092

d) **Isomer 4 (C<sub>s</sub>):**

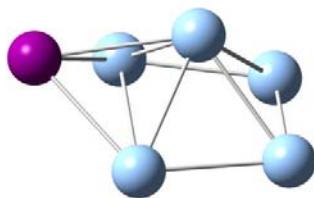


E(B3LYP) = -746.3920387; Zero-point correction = 0.002357

**Relative energy = 2.1 kcal mol<sup>-1</sup>**

Ag	-2.597154	-1.149806	-0.002055
Ag	0.163136	-1.301660	-0.003819
Ag	-1.100347	0.946007	1.356976
Ag	-1.102357	0.957112	-1.351228
Ag	1.390147	1.515608	0.000987
I	2.879037	-0.857760	-0.000763

e) **Isomer 5 (C<sub>s</sub>):**

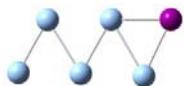


E(B3LYP) = -746.3919542; Zero-point correction = 0.002309

**Relative energy = 2.1 kcal mol<sup>-1</sup>**

Ag	-2.076107	1.448649	-0.111450
Ag	0.652143	1.957052	-0.620993
Ag	0.651718	-1.957074	-0.621015
Ag	-2.076334	-1.448419	-0.111421
Ag	-0.133136	0.000043	1.269724
I	2.644164	-0.000222	0.173061

f) **Isomer 6 (C<sub>s</sub>)**

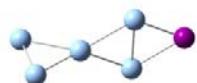


E(B3LYP) = -746.3833394; Zero-point correction = 0.002201

**Relative energy = 7.4 kcal mol<sup>-1</sup>**

Ag	2.518168	-0.172410	0.000000
Ag	2.574902	-2.885130	0.000000
Ag	0.074831	-1.430981	0.000000
Ag	0.000000	1.360293	0.000000
Ag	-2.487728	-0.012918	0.000000
I	-2.376757	2.785544	0.000000

g) **Isomer 7 (C<sub>2v</sub>)**



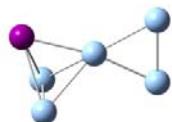
E(B3LYP) = -746.3830176; Zero-point correction = 0.002171

**Relative energy = 7.6 kcal mol<sup>-1</sup>**

Ag	0.000387	3.367379	1.337450
Ag	0.000387	3.367379	-1.337450

Ag	-0.000645	0.909850	0.000000
Ag	1.407316	-1.572597	0.000000
Ag	-1.407883	-1.573353	0.000000
I	0.000387	-3.989376	0.000000

**h) Isomer 8 ( $C_s$ ):**

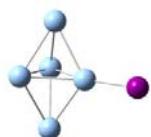


$E(B3LYP) = -746.3814152$ ; Zero-point correction = 0.002095

**Relative energy = 8.6 kcal mol<sup>-1</sup>**

Ag	3.070293	1.212201	-0.000038
Ag	-1.851517	-0.952750	1.394412
Ag	0.431860	0.334212	-0.005582
Ag	-1.800122	-1.043232	-1.363690
Ag	2.527017	-1.415255	0.016519
I	-2.108376	1.653712	-0.036909

**i) Isomer 9 ( $C2v$ ):** has 1 small imaginary frequency -15.3 cm<sup>-1</sup>



$E(B3LYP) = -746.3732857$ ; Zero-point correction = 0.002339

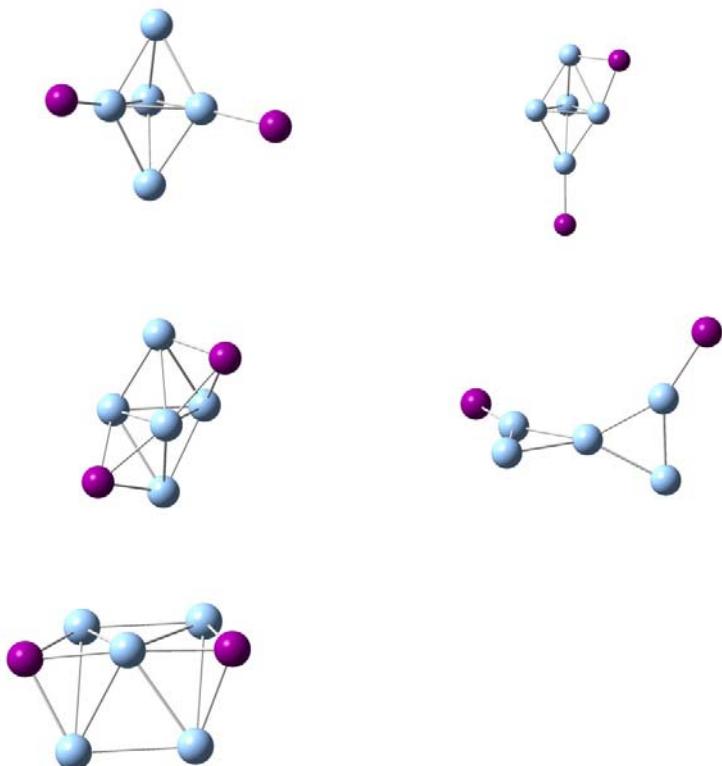
**Relative energy = 13.8 kcal mol<sup>-1</sup>**

Ag	0.006082	0.844008	0.000000
Ag	-1.376506	-1.601919	0.000000
Ag	1.369357	-1.613087	0.000000
Ag	0.000341	-0.757793	2.380600
Ag	0.000341	-0.757793	-2.380600
I	0.000341	3.446594	0.000000

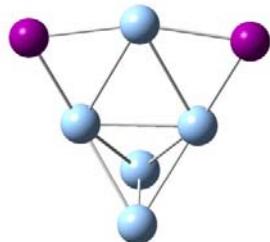
**$\text{Ag}_5\text{I}_2^+$ :**

**a) Isomer 1 ( $C2v$ ): (Used in Figure 4h)**

**The following five different input guesses**



**Converged to the same structure:**

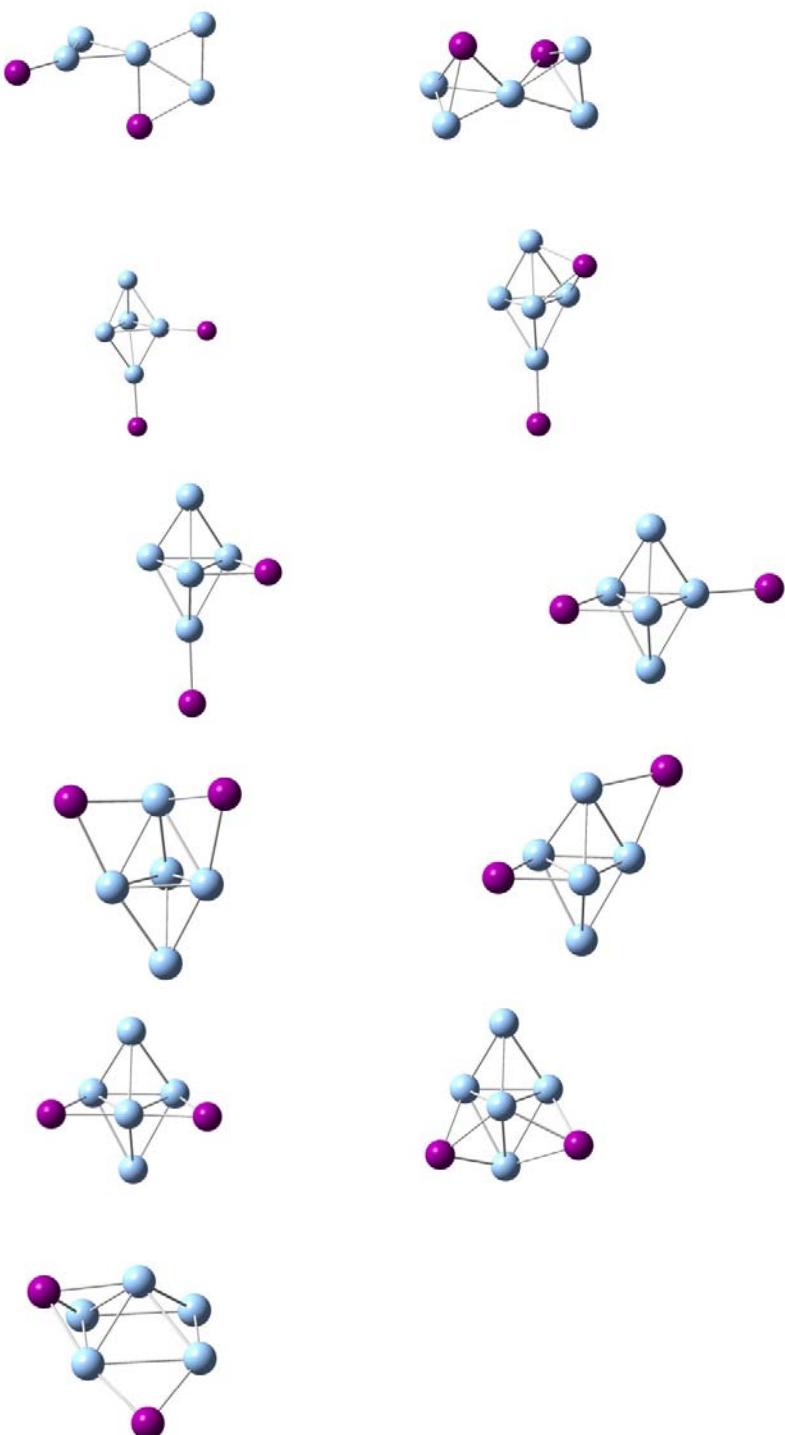


E(B3LYP) = -757.9026117; Zero-point correction = 0.003149

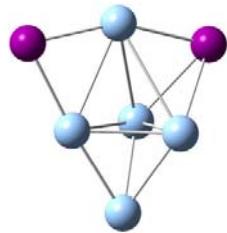
**Relative energy = 0 kcal mol<sup>-1</sup>**

Ag	0.000000	1.433869	-0.569957
Ag	-1.360364	0.000000	-2.611312
Ag	0.000000	-1.433869	-0.569957
Ag	0.000000	0.000000	2.247853
Ag	1.360364	0.000000	-2.611312
I	0.000000	2.710702	1.824435
I	0.000000	-2.710702	1.824435

b) **Isomer 2 (C<sub>s</sub>): The following 11 different input guesses**



**Converged to the same structure:**

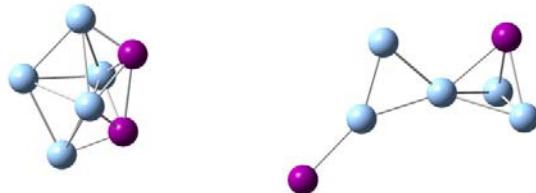


$E(B3LYP) = -757.9014567$ ; Zero-point correction = 0.003061

**Relative energy = 0.7 kcal mol<sup>-1</sup>**

Ag	1.303386	0.109268	-1.382361
Ag	-1.509946	1.398550	-0.000029
Ag	1.304234	0.109229	1.382410
Ag	-0.609189	-1.593252	0.000055
Ag	2.068576	-2.246300	-0.000229
I	0.945011	2.696660	-0.000058
I	-3.212593	-0.725759	0.000196

c) **Isomer 3 ( $C_s$ ): The following two different input guesses**



**Converged to the same structure:**

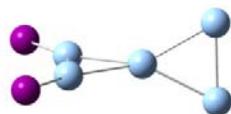


$E(B3LYP) = -757.8980473$ ; Zero-point correction = 0.002933

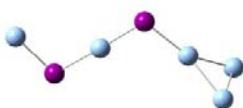
**Relative energy = 2.7 kcal mol<sup>-1</sup>**

Ag	-1.262274	-1.398514	-0.189996
Ag	2.195098	0.000116	-1.085611
Ag	0.782616	-0.000026	1.492984
Ag	-3.584530	-0.001518	-0.157318
Ag	-1.264834	1.399694	-0.186221
I	1.389401	2.485242	0.055021
I	1.389739	-2.485022	0.056858

d) **Isomer 4 ( $C_1$ ): The input guess:**



Converged to:

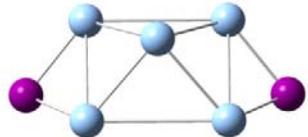
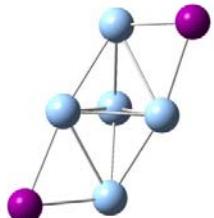
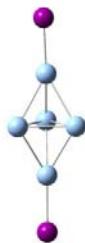


E(B3LYP) = -757.8864396; Zero-point correction = 0.002720

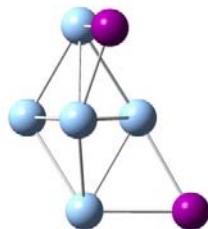
**Relative energy = 9.9 kcal mol<sup>-1</sup>**

Ag	-4.632255	-0.547244	1.360182
Ag	-4.486515	-0.967404	-1.269665
Ag	-2.497927	0.492683	-0.040637
I	-0.246845	2.064355	-0.131081
Ag	1.744097	0.241151	-0.024350
I	3.812451	-1.549186	0.089910
Ag	5.851811	0.199879	0.020895

e) **Isomer 5 (C<sub>2</sub>): The following three different input guesses**



Converged to the same structure:

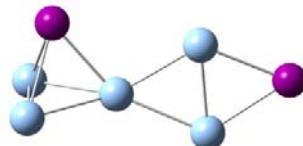
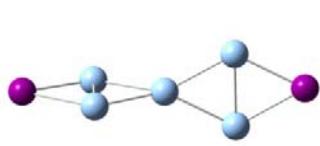


$E(B3LYP) = -757.8726038$ ; Zero-point correction = 0.002949

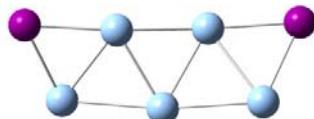
**Relative energy = 18.7 kcal mol<sup>-1</sup>**

Ag	-0.000134	2.308170	-0.000240
Ag	0.751106	-0.155069	1.234315
Ag	-0.750625	-0.155402	-1.233697
Ag	2.100213	0.630867	-1.148207
Ag	-2.100729	0.630785	1.148069
I	-3.080942	-1.445260	-0.427543
I	3.081092	-1.445108	0.427330

f) Isomer 6 ( $C_2$ ): The following two different input guesses



Converged to the same structure:



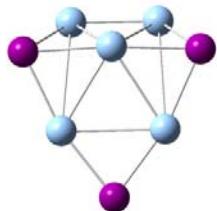
$E(B3LYP) = -757.865901$ ; Zero-point correction = 0.002794

**Relative energy = 22.8 kcal mol<sup>-1</sup>**

Ag	-1.372893	-0.866587	-0.314169
Ag	1.372818	-0.866737	0.313874
Ag	0.000303	1.498717	-0.000717
Ag	3.015432	1.283038	-0.642325
Ag	-3.015682	1.282887	0.643206
I	4.140984	-1.033743	0.322824
I	-4.140965	-1.033652	-0.322708

$\text{Ag}_5\text{I}_3^+$ :

a) **Isomer 1 ( $C_s$ ) (Used in Figure 4i)**



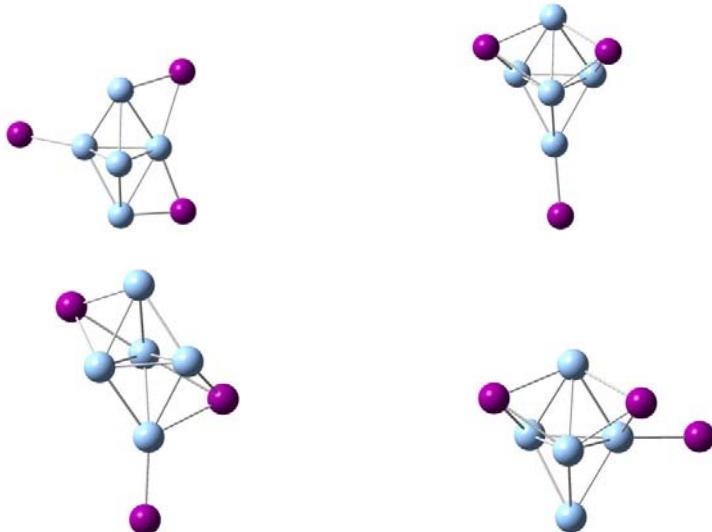
$E(B3LYP) = -769.3593624$ ; Zero-point correction = 0.003411

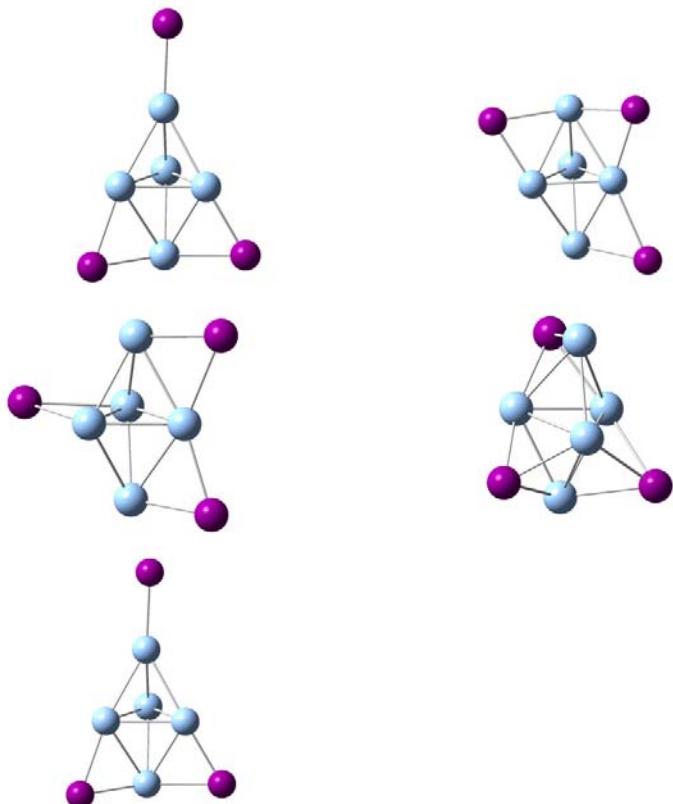
**Relative energy = 0 kcal mol<sup>-1</sup>**

Ag	1.617557	1.653932	-0.020155
Ag	-0.840609	0.000019	-1.027211
Ag	1.617405	-1.653992	-0.020129
I	-0.760649	-2.822589	-0.874008
I	-0.760469	2.822627	-0.873989
I	3.626307	-0.000118	0.729898
Ag	-2.384185	-1.388942	1.107786
Ag	-2.384104	1.389073	1.107779

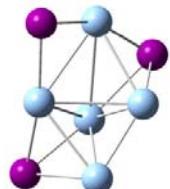
b) **Isomer 2 ( $C_1$ )** : has 1 small imaginary frequency  $-4.8 \text{ cm}^{-1}$

The following nine different input guesses:





**Converged to the same structure:**

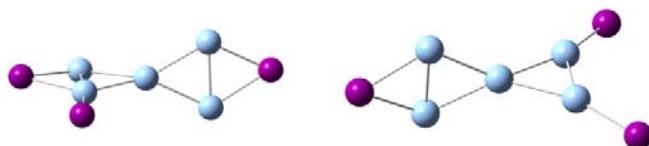


E(B3LYP) = -769.3569614; Zero-point correction = 0.003313

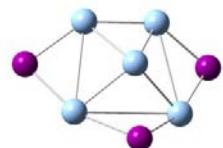
**Relative energy = 1.4 kcal mol<sup>-1</sup>**

Ag	-0.502687	1.584238	-0.287645
Ag	0.367512	-0.829574	1.705836
Ag	-0.786849	-1.472991	-0.976984
Ag	2.332111	0.080239	-0.396141
Ag	-2.229314	0.119929	1.685771
I	1.672975	-2.625507	-0.301788
I	-2.907525	0.336337	-1.172644
I	1.961035	2.748670	-0.060461

c) **Isomer 3 (C<sub>s</sub>) : The following two different input guesses**



**Converged to the same structure:**



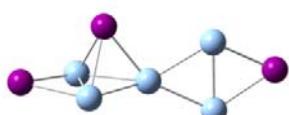
E(B3LYP) = -769.3535597; Zero-point correction = 0.003495

**Relative energy = 3.7 kcal mol<sup>-1</sup>**

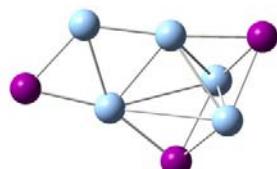
Ag	1.431974	-2.107064	0.440369
Ag	2.084979	1.334595	-0.459761
Ag	0.000003	0.188091	1.255309
Ag	-1.432279	-2.107003	0.440775
Ag	-2.084839	1.334728	-0.459889
I	0.000101	2.986521	0.403091
I	3.614408	-0.891819	-0.741010
I	-3.614364	-0.891633	-0.741133

d) **Isomer 4 (C<sub>s</sub>)** : has 1 small imaginary frequency -5.8 cm<sup>-1</sup>

**The following input guess**



**Converged to the structure:**



E(B3LYP) = -769.3402865; Zero-point correction = 0.003128

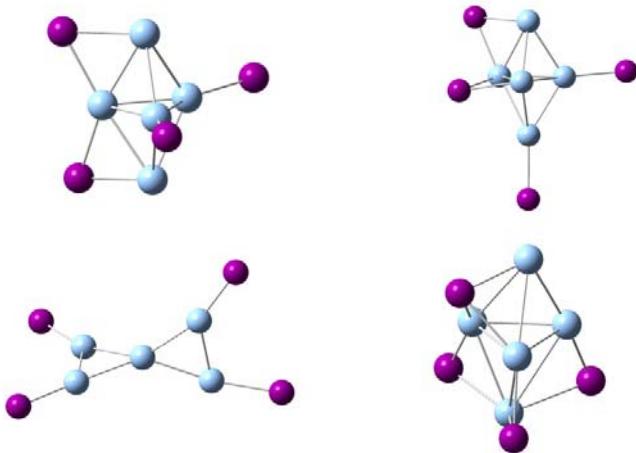
**Relative energy = 11.8 kcal mol<sup>-1</sup>**

Ag	2.040945	0.545514	-1.480894
Ag	-1.470620	0.910009	0.000451

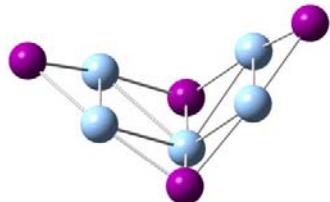
Ag	0.231673	-1.609973	-0.003531
Ag	-2.509387	-1.867823	0.003429
Ag	2.026465	0.532350	1.484811
I	0.771552	2.645320	0.005026
I	-4.174201	0.341825	-0.006104
I	3.119694	-1.665893	-0.002704

**Ag<sub>5</sub>I<sub>4</sub><sup>+</sup>:**

a) **Isomer 1 (C<sub>2v</sub>) (Used in Figure 4j):**  
**The following four different input guesses**



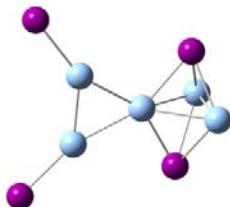
**Converged to the same structure:**



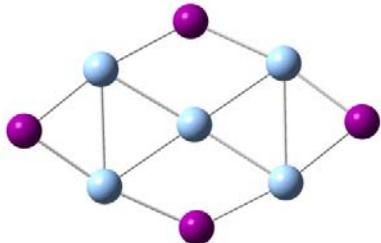
E(B3LYP) = -780.8694168; Zero-point correction = 0.004152  
**Relative energy = 0 kcal mol<sup>-1</sup>**

I	3.698147	-0.000003	1.568482
Ag	2.008541	1.645983	0.241181
Ag	2.008543	-1.645982	0.241170
Ag	0.000649	0.000004	-1.511226
I	0.000253	2.800018	-1.324810
I	0.000255	-2.800011	-1.324825
Ag	-2.008862	1.646217	0.240399
I	-3.698667	-0.000005	1.567187
Ag	-2.008857	-1.646220	0.240396

b) **Isomer 2:** has 1 small imaginary frequency  $-18.0 \text{ cm}^{-1}$   
**The following input guess**



**Converged to the structure:**

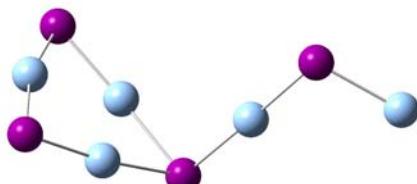


$E(\text{B3LYP}) = -780.8593291$ ; Zero-point correction = 0.004037

**Relative energy = 6.3 kcal mol<sup>-1</sup>**

Ag	2.503838	-1.620965	0.031879
Ag	-2.489666	-1.647160	-0.126438
Ag	-0.020528	0.014287	0.050426
Ag	-2.526190	1.637435	0.068854
Ag	2.504802	1.624981	-0.058083
I	4.657480	-0.006570	-0.247028
I	0.010412	-2.851027	0.283413
I	-4.646991	-0.015321	-0.232621
I	0.003703	2.865311	0.225821

c) **Isomer 3 ( $C_s$ ):**

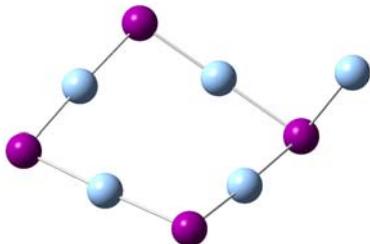


$E(\text{B3LYP}) = -780.8469822$ ; Zero-point correction = 0.003870

**Relative energy = 13.9 kcal mol<sup>-1</sup>**

Ag	-1.505191	-1.569466	0.796859
Ag	-1.453134	1.572792	0.791851

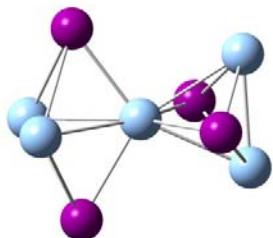
Ag	-3.648340	0.037181	-0.913038
I	-3.341917	2.750588	-0.736705
I	0.369841	-0.026245	2.220781
I	-3.433685	-2.684336	-0.729222
Ag	2.377140	-0.068468	0.341394
I	4.321772	-0.124602	-1.560830
Ag	6.579556	0.123355	-0.108199

d) Isomer 4 ( $C_1$ ):

E(B3LYP) = -780.837153; Zero-point correction = 0.003906

**Relative energy = 20.1 kcal mol<sup>-1</sup>**

Ag	2.496500	-1.920990	0.126636
Ag	-1.345153	-1.971224	-0.442124
Ag	-1.345152	1.971224	-0.442124
Ag	2.496501	1.920990	0.126636
I	0.499229	-3.788657	0.210219
I	-3.260115	0.000000	-1.037749
I	0.499230	3.788657	0.210219
I	4.394615	0.000000	0.036472
Ag	-4.707947	0.000000	1.285965

e) Isomer 5 ( $D_{2d}$ ):

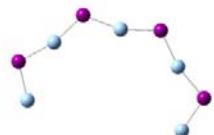
E(B3LYP) = -780.81918; Zero-point correction = 0.003379

**Relative energy = 31.0 kcal mol<sup>-1</sup>**

Ag	2.649139	0.604181	-1.418768
Ag	-2.642461	1.421641	0.605400
Ag	-0.000105	-0.004088	-0.003724
Ag	-2.652629	-1.417981	-0.602571
Ag	2.645213	-0.603376	1.420424

I	2.042680	-2.142698	-0.911650
I	-2.042498	0.911450	-2.142111
I	-2.039283	-0.911812	2.142276
I	2.039848	2.142727	0.910810

f) **Isomer 6:** has 2 small imaginary frequencies -12.5676 and -4.5454cm<sup>-1</sup>

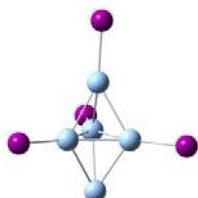


E(B3LYP) = -780.8177435; Zero-point correction = 0.003538

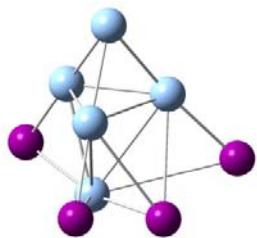
**Relative energy = 32.0 kcal mol<sup>-1</sup>**

Ag	0.000000	0.000000	2.325176
Ag	0.000000	4.283974	0.595595
Ag	0.000000	-4.283974	0.595595
Ag	0.000000	5.338836	-3.592776
Ag	0.000000	-5.338836	-3.592776
I	0.000000	2.685225	2.769012
I	0.000000	-2.685225	2.769012
I	0.000000	6.414264	-1.142109
I	0.000000	-6.414264	-1.142109

g) **Isomer 7 (C<sub>s</sub>): The following input guess**



**Converged to the structure:**

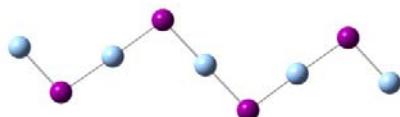


E(B3LYP) = -780.7555356; Zero-point correction = 0.003977

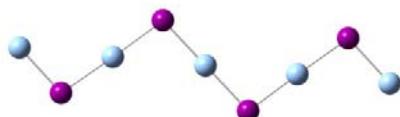
**Relative energy = 71.4 kcal mol<sup>-1</sup>**

Ag	0.029808	-1.478786	1.289360
Ag	2.218561	-0.001233	0.181229
Ag	0.030735	1.473635	1.294759
Ag	1.743024	-0.006180	2.912979
Ag	-0.018078	0.004253	-2.025153
I	2.647882	0.004353	-2.525108
I	-1.861781	-3.067887	-0.041940
I	-1.859757	3.069040	-0.030578
I	-2.477106	0.001865	-0.641981

g) **Isomer 8 ( $C_2$ ): The following input guess**



**Converged to the structure:**



$E(B3LYP) = -780.8288073$ ; Zero-point correction = 0.003623

**Relative energy = 18.9 kcal mol<sup>-1</sup>**

Ag	-4.043527	0.260958	0.014768
Ag	-0.000006	0.000074	0.170078
Ag	4.043537	-0.260920	0.014860
Ag	8.110437	-0.548146	-0.129258
I	-1.920792	1.921141	0.174893
I	6.237033	1.373621	-0.148916
I	1.920732	-1.921033	0.175047
I	-6.236952	-1.373679	-0.149037
Ag	-8.110464	0.547979	-0.129070

**Neutrals used in calculations for reactions of cluster cations with allyl iodide:**

**Allyl iodide ( $C_3H_5I$ ):**

C	3.135366	-0.468335	-0.252017
C	2.149599	0.068822	0.473956
C	1.084754	0.920323	-0.100124
I	-0.935226	-0.086954	-0.008734
H	3.921634	-1.059910	0.207677

H 3.191902 -0.332539 -1.329837  
H 2.113063 -0.094337 1.549618  
H 1.221064 1.136324 -1.158379  
H 0.901009 1.834171 0.462946

E(B3LYP) = -128.7198675; Zero-point correction = 0.070932

**Allyl Radical (C<sub>3</sub>H<sub>5</sub>):**

C 1.230252 0.196157 0.000011  
C 0.000044 -0.443164 -0.000070  
C -1.230271 0.196066 0.000017  
H 2.161085 -0.361223 0.000234  
H 1.299657 1.280844 -0.000110  
H 0.000135 -1.533677 -0.000021  
H -2.161253 -0.361072 0.000135  
H -1.299774 1.280778 0.000017

E(B3LYP) = -117.2579397; Zero-point correction = 0.066578

**Propene (C<sub>3</sub>H<sub>6</sub>):**

C -1.283184 0.220545 -0.000067  
C -0.133803 -0.455933 -0.000186  
C 1.235076 0.163011 0.000037  
H -2.245583 -0.284386 0.000620  
H -1.305021 1.308841 0.000029  
H -0.163952 -1.546669 0.000243  
H 1.811694 -0.148618 0.881676  
H 1.181011 1.256961 -0.001949  
H 1.813318 -0.151868 -0.879324

E(B3LYP) = -117.90521; Zero-point correction = 0.080097

**1,5-hexadiene, (C<sub>3</sub>H<sub>6</sub>)<sub>2</sub>:**

C -0.572990 -0.297257 -0.520358  
C -1.934984 -0.338121 0.117232  
C -2.873995 0.600181 -0.012816  
H 0.448335 -1.172903 1.176072  
H -0.483599 0.589179 -1.160256  
H -2.140627 -1.210375 0.741202  
H -2.713406 1.487796 -0.622006  
H -3.839819 0.517423 0.478733  
C 2.874002 0.600175 0.012798  
C 1.934981 -0.338121 -0.117221  
C 0.572985 -0.297223 0.520364  
H 3.839828 0.517386 -0.478742  
H 2.713421 1.487813 0.621956  
H 2.140617 -1.210399 -0.741159  
H 0.483590 0.589252 1.160209  
H -0.448336 -1.172979 -1.176010

E(B3LYP) = -234.6071736; Zero-point correction = 0.14252

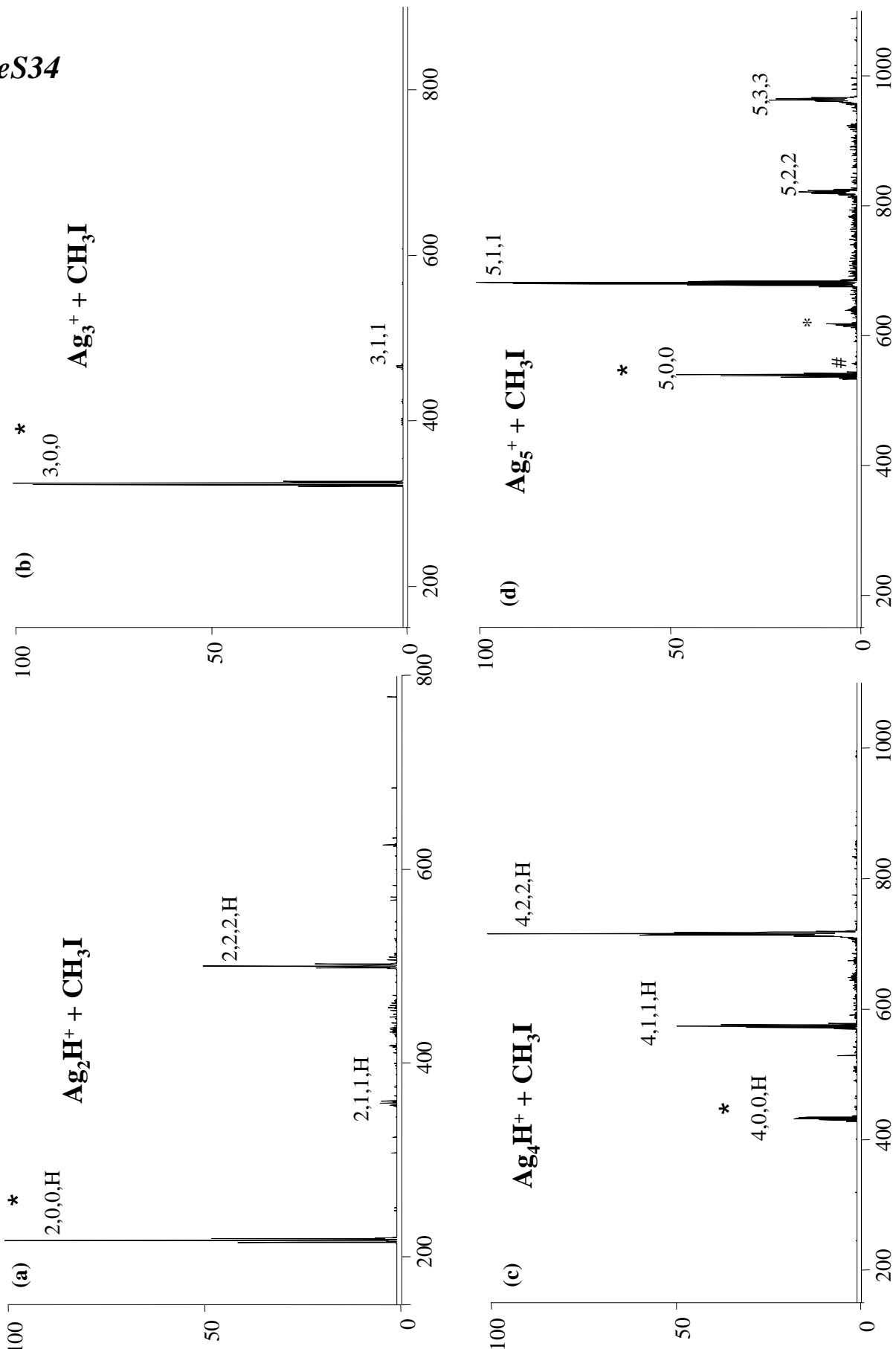
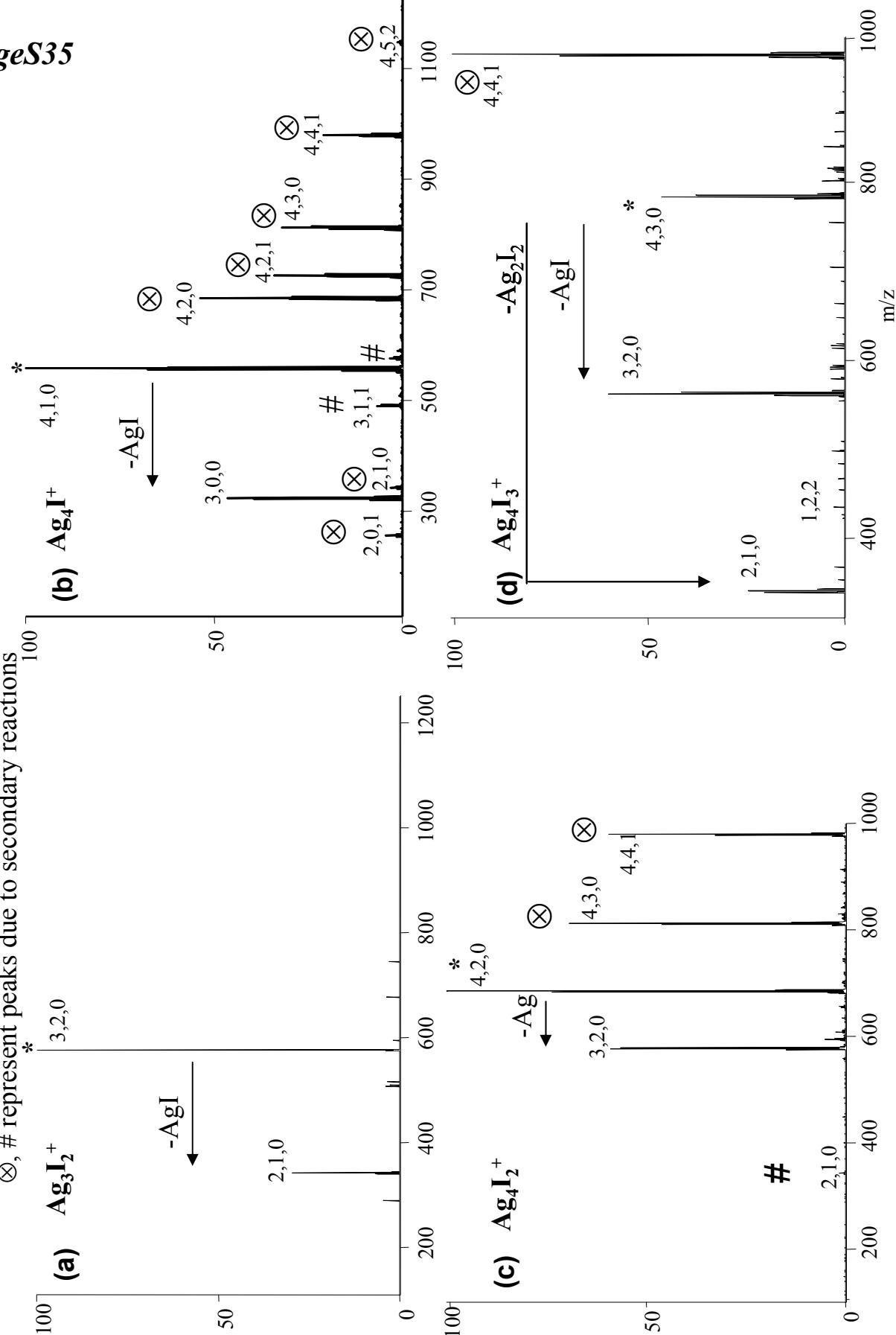
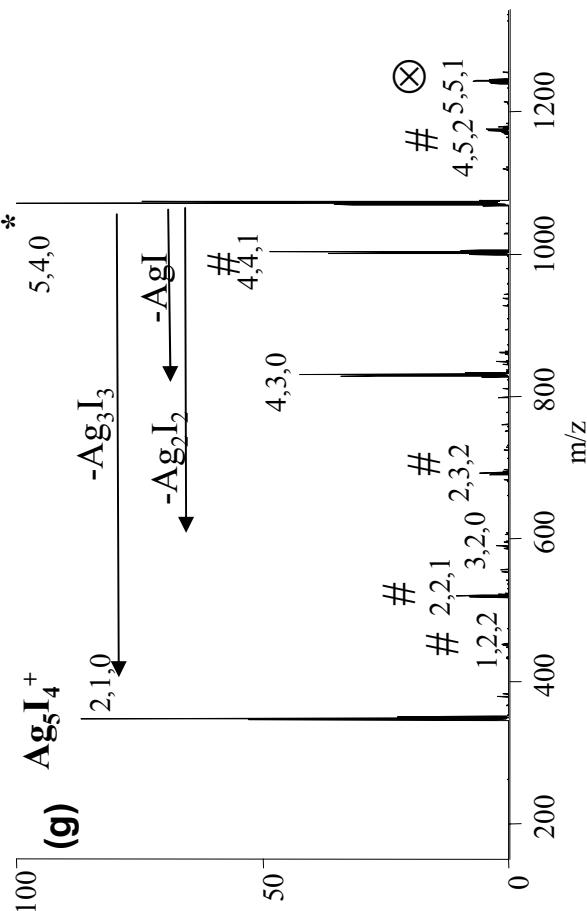
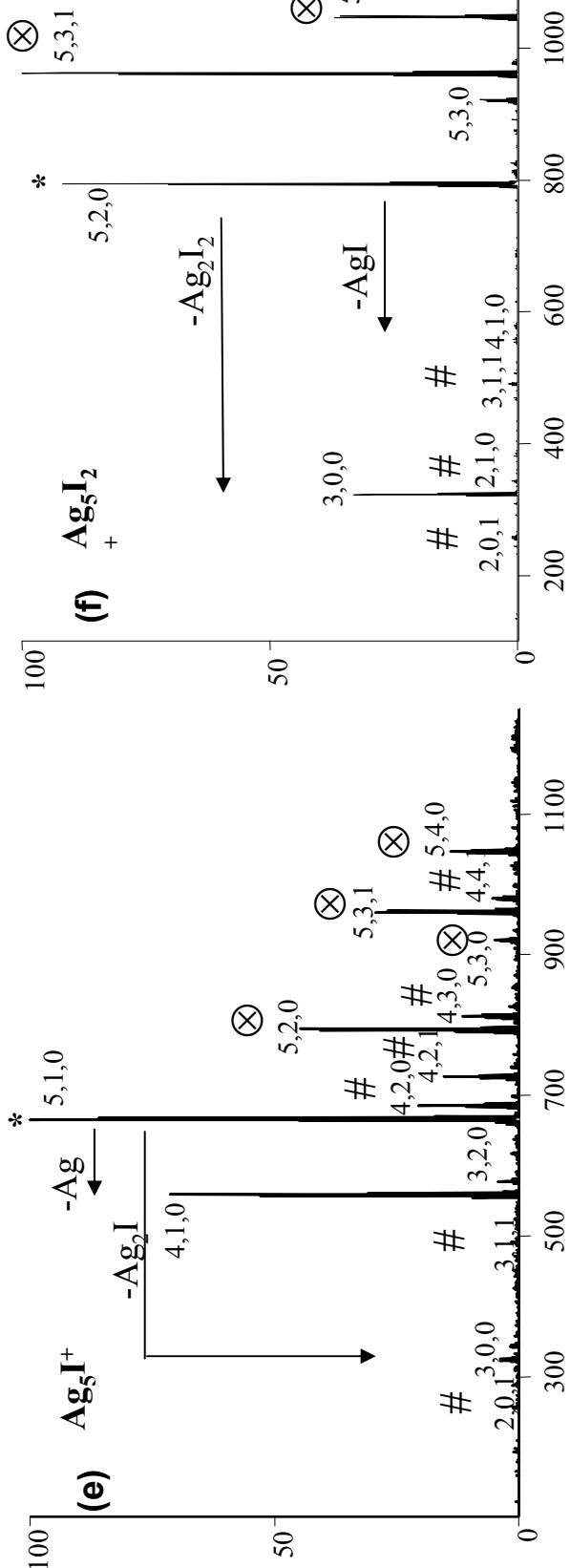
Fig S1: Ion-molecule reactions of  $\text{CH}_3\text{I}$  with: (a)  $\text{Ag}_2\text{H}^+$ , (b)  $\text{Ag}_3^+$ , (c)  $\text{Ag}_4\text{H}^+$  and (d)  $\text{Ag}_5^+$ 

Fig S2: CID spectra of the observed  $\text{Ag}_n\text{I}_m^+$ . The \* represent the selected peak for CID



**Table S1:** DFT calculated energies for ground state structures relevant to the silver clusters used in the thermochemistry calculations.  
The scaling factor used is 0.9806 and the basis sets were LanL2DZ for Ag and I, and 6-31G\* for C, N and H.

Species	Isomer <sup>a</sup>	Comments <sup>b</sup>	E <sup>c</sup>	ZPE <sup>d</sup>	ZPE corr <sup>e</sup>	E+ZPEcorr <sup>f</sup>	Rel E (kcal mol <sup>-1</sup> ) <sup>g</sup>
Ag <sup>+</sup>			-146.70064			-146.70064	
Ag <sub>2</sub> H <sup>+</sup>	from DT		-294.36431	0.005742	0.00563061	-294.35868	
Ag <sub>2</sub> I <sup>+</sup>		<b>Figure 4a</b>	-305.25137	0.000816	0.00080017	-305.25057	
Ag <sub>3</sub> <sup>+</sup>	from DT		-440.84253	0.000896	0.00087862	-440.84165	
Ag <sub>3</sub> I <sup>+</sup>	<b>Isomer 1</b>		-452.26253	0.001215	0.00119143	-452.26134	11.28977024
	<b>Isomer 2</b>		-452.27707	0.001158	0.00113553	-452.27594	2.127563117
	<b>Isomer 3</b>	<b>Figure 4b</b>	-452.28055	0.00125	0.00122575	-452.27933	0
Ag <sub>3</sub> I <sub>2</sub> <sup>+</sup>	<b>Isomer 1</b>	<b>Figure 4c</b>	-463.78182	0.001758	0.00172389	-463.7801	0
	<b>Isomer 2</b>		-463.77989	0.001768	0.0017337	-463.77815	1.21894194
	<b>Isomer 3</b>		-463.77938	0.001744	0.00171017	-463.77767	1.521066419
Ag <sub>4</sub> <sup>+(h)</sup>			-587.872498	0.001283	0.00125811	-587.87124	
Ag <sub>4</sub> H <sup>+</sup>	from DT		-588.475	0.007106	0.00696814	-588.46803	
Ag <sub>4</sub> I <sup>+</sup>	from DT		-599.36216	0.00179	0.00175527	-599.3604	
Ag <sub>4</sub> I <sub>2</sub> <sup>+</sup>	<b>Isomer 1</b>	<b>Figure 4e</b>	-610.81515	0.002226	0.00218282	-610.81297	0
	<b>Isomer 2</b>		-610.81377	0.002325	0.0022799	-610.81149	0.931211913

<b>Species</b>	<b>Isomer<sup>a</sup></b>	<b>Comments<sup>b</sup></b>	<b>E<sup>c</sup></b>	<b>ZPE<sup>d</sup></b>	<b>ZPE corr<sup>e</sup></b>	<b>E+ZPEcorr<sup>f</sup></b>	<b>Rel E (kcal mol<sup>-1</sup>)<sup>g</sup></b>
	<b>Isomer 3</b>	ifreq -21.2 cm <sup>-1</sup>	-610.81354	0.00213	0.00208868	-610.81145	0.951658072
	<b>Isomer 4</b>	ifreq -4.9 cm <sup>-2</sup>	-610.8001	0.002088	0.00204749	-610.79805	9.36161913
	<b>Isomer 5</b>	ifreq -1.4 cm <sup>-3</sup>	-610.7984	0.002197	0.00215438	-610.79624	10.49614805
	<b>Isomer 6</b>	ifreq -5.9 cm <sup>-3</sup>	-610.79595	0.001937	0.00189942	-610.79405	11.87123832
	<b>Isomer 7</b>		-610.78628	0.002424	0.00237697	-610.78391	18.23811304
<b>Ag<sub>4</sub>I<sub>3</sub><sup>+</sup></b>	<b>Isomer 1</b>	<b>Figure 4f</b>	-622.32397	0.002983	0.00292513	-622.32104	0
	<b>Isomer 2</b>	ifreq -15.8 cm <sup>-1</sup>	-622.32327	0.002837	0.00278196	-622.32049	0.344586072
	<b>Isomer 3</b>	ifreq -16.3 cm <sup>-2</sup>	-622.3205	0.002781	0.00272705	-622.31777	2.052973513
	<b>Isomer 4</b>	ifreq -26.7 cm <sup>-3</sup>	-622.30675	0.002701	0.0026486	-622.3041	10.6313816
	<b>Isomer 5</b>		-622.30649	0.002688	0.00263585	-622.30385	10.78747609
<b>Ag<sub>5</sub>I<sup>+</sup></b>	<b>Isomer 1</b>	<b>Figure 4g</b>	-746.39529	0.00232	0.00227499	-746.39302	0
	<b>Isomer 2</b>		-746.39338	0.002315	0.00227009	-746.39111	1.197412699
	<b>Isomer 3</b>		-746.39275	0.002308	0.00226322	-746.39048	1.588373894
	<b>Isomer 4</b>		-746.39204	0.002357	0.00231127	-746.38973	2.062990706
	<b>Isomer 5</b>		-746.39195	0.002309	0.00226421	-746.38969	2.086479159
	<b>Isomer 6</b>		-746.38334	0.002201	0.0021583	-746.38118	7.425895986

<b>Species</b>	<b>Isomer<sup>a</sup></b>	<b>Comments<sup>b</sup></b>	<b>E<sup>c</sup></b>	<b>ZPE<sup>d</sup></b>	<b>ZPE corr<sup>e</sup></b>	<b>E+ZPEcorr<sup>f</sup></b>	<b>Rel E (kcal mol<sup>-1</sup>)<sup>g</sup></b>
	<b>Isomer 7</b>		-746.38302	0.002171	0.00212888	-746.38089	7.609368614
	<b>Isomer 8</b>		-746.3814	0.002061	0.00202102	-746.37938	8.555926034
	<b>Isomer 9</b>	ifreq -15.3 cm <sup>-3</sup>	-746.37329	0.002339	0.00229362	-746.37099	13.81960968
$\text{Ag}_5\text{I}_2^+$	<b>Isomer 1</b>	<b>Figure 4h</b>	-757.90261	0.003149	0.00308791	-757.89952	0
	<b>Isomer 2</b>		-757.90146	0.003061	0.00300162	-757.89846	0.670624455
	<b>Isomer 3</b>		-757.89805	0.002933	0.0028761	-757.89517	2.731294002
	<b>Isomer 4</b>		-757.88644	0.00272	0.00266723	-757.88377	9.884175196
	<b>Isomer 5</b>		-757.8726	0.002949	0.00289179	-757.86971	18.70719007
	<b>Isomer 6</b>		-757.8659	0.002794	0.0027398	-757.86316	22.81788697
$\text{Ag}_5\text{I}_3^+$	<b>Isomer 1</b>	<b>Figure 4i</b>	-769.35936	0.003411	0.00334483	-769.35602	0
	<b>Isomer 2</b>	ifreq -4.8 cm <sup>-3</sup>	-769.35696	0.003313	0.00324873	-769.35371	1.446348552
	<b>Isomer 3</b>		-769.35356	0.003495	0.0034272	-769.35013	3.692940527
	<b>Isomer 4</b>	ifreq -5.8 cm <sup>-3</sup>	-769.34029	0.003128	0.00306732	-769.33722	11.79617783
$\text{Ag}_5\text{I}_4^+$	<b>Isomer 1</b>	<b>Figure 4j</b>	-780.86942	0.004152	0.00407145	-780.86535	0
	<b>Isomer 2</b>	ifreq -18.0 cm <sup>-3</sup>	-780.85933	0.004037	0.00395868	-780.85537	6.259368952
	<b>Isomer 3</b>		-780.84698	0.00387	0.00379492	-780.84319	13.90441101

Species	Isomer <sup>a</sup>	Comments <sup>b</sup>	E <sup>c</sup>	ZPE <sup>d</sup>	ZPE corr <sup>e</sup>	E+ZPEcorr <sup>f</sup>	Rel E (kcal mol <sup>-1</sup> ) <sup>g</sup>
	<b>Isomer 4</b>		-780.83715	0.003906	0.00383022	-780.83332	20.09448441
	<b>Isomer 5</b>		-780.81918	0.003379	0.00331345	-780.81586	31.05013368
	<b>Isomer 6</b> 2 x iffeq -12.6, -4.5 cm <sup>-3</sup>		-780.81774	0.003538	0.00346936	-780.81427	32.04769599
	<b>Isomer 7</b>		-780.75554	0.003977	0.00389985	-780.75164	71.35390796
	<b>Isomer8</b>		-780.82881	0.003623	0.00355271	-780.82525	18.89798549
	<b>I</b>		-11.394691			-11.394691	
	<b>CH<sub>2</sub>=CHCH<sub>2</sub></b>		-117.25794	0.066578	0.06528639	-117.19265	
	<b>CH<sub>2</sub>=CHCH<sub>2</sub>I</b>		-128.71987	0.070932	0.06955592	-128.65031	
	<b>CH<sub>2</sub>=CHCH<sub>3</sub></b>		-117.90521	0.080097	0.07854312	-117.82667	
	<b>(CH<sub>2</sub>=CHCH<sub>2</sub>)<sub>2</sub></b>		-234.60717	0.142525	0.13976002	-234.46741	
	<b>Ag<sup>·</sup></b>		-146.9936			-146.9936	
	<b>AgI</b>	<b>Figure 3a</b>	-158.47328	0.000436	0.00042754	-158.47285	
	<b>AgI<sub>2</sub></b>	<b>Figure 3b</b>	-317.00259	0.00139	0.00136303	-317.00123	
	<b>Ag<sub>3</sub>I<sub>3</sub></b>	<b>Isomer 1</b> <b>Isomer 2</b>	-475.55288	0.002637	0.00258584	-475.55029	0
			-475.42943	0.002286	0.00224165	-475.42719	77.25012646
	<b>Ag<sub>2</sub></b>		-294.04301	0.000406	0.00039812	-294.04261	

- (a) The isomer number as it appears in the supplementary material part A.
- (b) The figure where the relevant structure is shown.
- (c) The converged geometries B3LYP/SDD energies.
- (d) Zero point energy.
- (e) ZPE x 0.9806.
- (f) The sum of the B3LYP/SDD energy and ZPE scaled.
- (g) The relative energy to the most stable isomer in each case.
- (h) This structure has been previously calculated; e.g.; P. Weis, T. Bierweiler, S. Gilb, M. M. Kappes, *Chem. Phys. Lett.*, 2002, **355**, 355

**Table S2:** Calculated energetics for the ion-molecule reactions.

Reaction	Experimental value	Reaction $\Delta H$ in kcal.mol <sup>-1</sup> <sup>b</sup>
<b>Benchmark BDE of allyliodide</b>		
$\text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \cdot + \text{CH}_2=\text{CHCH}_2\text{I}$	$+45.6 \pm 0.9$ <sup>c</sup>	+39.5
<b>Ion-molecule reactions with allyliodide</b>	Relevant eq. in text (reactants, products) <sup>a</sup>	
$\text{Ag}_2\text{H}^+ + \text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_2\text{I}^+ + \text{CH}_2=\text{CHCH}_3$	(6)	-42.8
$\text{Ag}_3^+ + \text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_3\text{I}^+ + \text{CH}_2=\text{CHCH}_2\text{I}$	does not occur	+12.5
$\text{Ag}_3^+ + 2\text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_3\text{I}_2^+ + (\text{CH}_2=\text{CHCH}_2)_2$	(7) and (8)	-66.0
$\text{Ag}_4\text{H}^+ + \text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_4\text{I}^+ + \text{CH}_2=\text{CHCH}_3$	(9)	-43.1
$\text{Ag}_4\text{I}^+ + \text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_4\text{I}_2^+ + \text{CH}_2=\text{CHCH}_2\text{I}$	(10a)	+3.2
$\text{Ag}_4\text{I}_2^+ + \text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_4\text{I}_3^+ + \text{CH}_2=\text{CHCH}_2\text{I}$	(11)	-31.6
$\text{Ag}_4\text{I}^+ + 2\text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_4\text{I}_3^+ + (\text{CH}_2=\text{CHCH}_2)_2$	(10b) and (12)	-79.9
$\text{Ag}_5^+ + \text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_5\text{I}^+ + \text{CH}_2=\text{CHCH}_2\text{I}$	(13b)	+2.3
$\text{Ag}_5^+ + 2\text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_5\text{I}^+ + \text{AgI} + (\text{CH}_2=\text{CHCH}_2)_2$		-38.2
$\text{Ag}_5\text{I}^+ + \text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_5\text{I}_2^+ + \text{CH}_2=\text{CHCH}_2\text{I}$	(14)	-30.7
$\text{Ag}_5\text{I}_2^+ + 2\text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_5\text{I}_4^+ + (\text{CH}_2=\text{CHCH}_2)_2$	(16b) and (17)	-83.2

$\text{Ag}_5\text{I}_2^+ + \text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_5\text{I}_3^+ + \text{CH}_2=\text{CHCH}_2\cdot$	(16a)	+0.7
$\text{Ag}_5\text{I}_3^+ + \text{CH}_2=\text{CHCH}_2\text{I} \rightarrow \text{Ag}_5\text{I}_4^+ + \text{CH}_2=\text{CHCH}_2\cdot$	(18)	-32.4

**Table S3:** Calculated reaction energetics for CID reactions

Reactions (eq. number in text)	SDD Reaction $\Delta H$ in kcal.mol $^{-1}$ <sup>b</sup>
$\text{Ag}_2\text{I}^+ \rightarrow \text{Ag}^+ + \text{AgI}$ (eq. 24c)	+48.4
$\text{Ag}_3\text{I}_2^+ \rightarrow \text{Ag}_2\text{I}^+ + \text{AgI}$ (eq. 24c)	+35.6
$\text{Ag}_3\text{I}_2^+ \rightarrow \text{Ag}^+ + \text{Ag}_2\text{I}_2$ (eq. 24d)	+49.1
$\text{Ag}_4\text{I}^+ \rightarrow \text{Ag}_3^+ + \text{AgI}$ (eq. 24c)	+28.8
$\text{Ag}_4\text{I}^+ \rightarrow \text{Ag}_2\text{I}^+ + \text{Ag}_2$ (eq. 24b)	+42.2
$\text{Ag}_4\text{I}_2^+ \rightarrow \text{Ag}_3\text{I}_2^+ + \text{Ag}$ (eq. 24a)	+24.6
$\text{Ag}_4\text{I}_2^+ \rightarrow \text{Ag}_3\text{I}^+ + \text{AgI}$ (eq. 24c)	+38.1
$\text{Ag}_4\text{I}_3^+ \rightarrow \text{Ag}_3\text{I}_2^+ + \text{AgI}$ (eq. 24c)	+42.7
$\text{Ag}_4\text{I}_3^+ \rightarrow \text{Ag}_2\text{I}^+ + \text{Ag}_2\text{I}_2$ (eq. 24d)	+43.5
$\text{Ag}_4\text{I}_3^+ \rightarrow \text{Ag}^+ + \text{Ag}_3\text{I}_3$ (eq. 24e)	+43.9
$\text{Ag}_5\text{I}^+ \rightarrow \text{Ag}_2\text{I}^+ + \text{Ag}$ (eq. 24a)	+24.5
$\text{Ag}_5\text{I}^+ \rightarrow \text{Ag}_3\text{I}^+ + \text{Ag}_2$ (eq. 24b)	+44.6

$\text{Ag}_5\text{I}^+ \rightarrow \text{Ag}_4^+ + \text{AgI}$ (eq. 24c)	+30.7
$\text{Ag}_5\text{I}_2^+ \rightarrow \text{Ag}_4\text{I}^+ + \text{AgI}$ (eq. 24c)	+41.6
$\text{Ag}_5\text{I}_2^+ \rightarrow \text{Ag}_3\text{I}_2^+ + \text{Ag}_2$ (eq. 24b)	+48.2
$\text{Ag}_5\text{I}_2^+ \rightarrow \text{Ag}_3^+ + \text{Ag}_2\text{I}_2$ (eq. 24d)	+35.6
$\text{Ag}_5\text{I}_4^+ \rightarrow \text{Ag}_4\text{I}_3^+ + \text{AgI}$ (eq. 24c)	+44.8
$\text{Ag}_5\text{I}_4^+ \rightarrow \text{Ag}_3\text{I}_2^+ + \text{Ag}_2\text{I}_2$ (eq. 24d)	+52.7
$\text{Ag}_5\text{I}_4^+ \rightarrow \text{Ag}_2\text{I}^+ + \text{Ag}_3\text{I}_3$ (eq. 24e)	+40.5

- (a) The numbers refer to those equation numbers used in the text and figures.  
 (b) The reaction enthalpy is calculated using the (B3LYP/SDD + ZPE) values  
 (c) Experimental bond enthalpy value from Blanksby S. J.; Ellison G. B., *Acc. Chem. Res.* **2003**, *36*, 255