

## Supporting Information

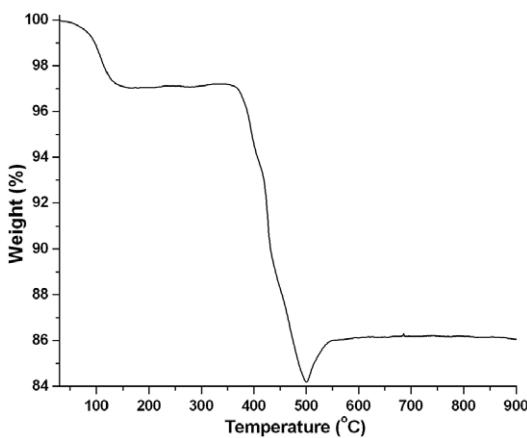
### The First Solid Composed of {As<sub>4</sub>V<sub>16</sub>O<sub>42</sub>(H<sub>2</sub>O)} Clusters

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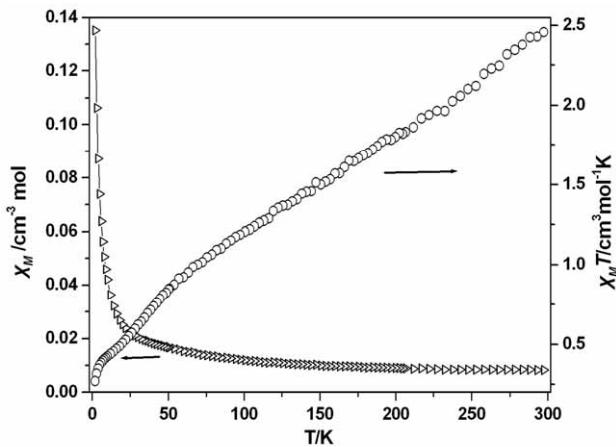
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**Figure S1.** TG curve of **1**.



TGA of **1** shows two major steps of weight losses (Figure S1). The first weight loss of **1** corresponds to the removal of lattice water molecules at relatively lower temperature (below  $\sim 150$  °C). The remainder stages of **1** (in the range 400-500°C) are attributed to the loss of the organic ligands and the sublimation of part As<sub>2</sub>O<sub>3</sub> molecules. A weight gain was shown between 500 and 550 °C, which is consistent with the oxidation of V atoms form V<sup>4+</sup> to V<sup>5+</sup> and some of the As atoms from As<sup>3+</sup> to As<sup>5+</sup>.

**Figure S2.** Temperature dependence of  $\chi_m$  ( $\Delta$ ) and  $\chi_m T$  (O) values for **1**.



The variable temperature magnetic susceptibility of **1** was measured between 2 and 300 K. As shown in Figure S2, the  $\chi_m T$  value at 300 K is 2.45 cm<sup>3</sup> K mol<sup>-1</sup> (4.43  $\mu_B$ ), much smaller than that expected value 6.00 cm<sup>3</sup> K mol<sup>-1</sup> ( $\mu_{eff} = 6.93 \mu_B$ ) for the sixteen uncoupled  $S = 1/2$  spins of V<sup>4+</sup> atoms with  $g = 2$ . The  $\chi_m T$  value decreases nearly linearly with decreasing temperature from 2.45 cm<sup>3</sup> Kmol<sup>-1</sup> at 300 K to 0.41 cm<sup>3</sup> Kmol<sup>-1</sup> at 8 K, and then decreases rapidly from 8 K and reaches

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a minimum value of  $0.27 \text{ cm}^3 \text{ Kmol}^{-1}$  at 2 K. The temperature-dependences of  $\chi_M T$  demonstrate the presence of strong antiferromagnetic coupling interactions, which is a common feature for most polyoxovanadates.(A. Müller, F. Peters, M. T. Pope, D. Gatteschi, *Chem. Rev.* **1998**, *98*, 239). Unfortunately, it is too difficult to fit the experimental magnetic data of these extended heteropolymetallic spin system using a suitable theoretical model.(O. Kahn, *Molecular Magnetism*, VCH, Weinheim, Germany, **1993**)