

Figure 8

Concentration distribution of the complexes formed in the copper(II)-HVVH system.
 $[\text{Cu}^{2+}] = 2 \cdot 10^{-3} \text{ mol/dm}^3$, $c_{\text{HVVH}} = 2 \cdot 10^{-3} \text{ mol/dm}^3$

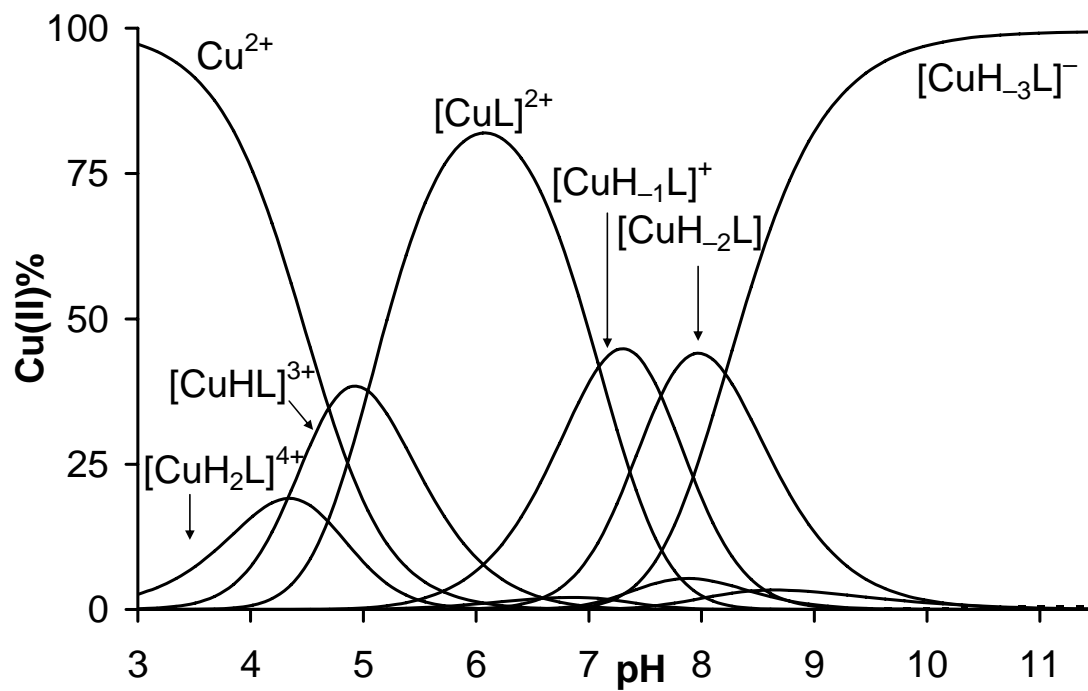


Figure 9
Concentration distribution of the complexes formed in the copper(II)-HAAHVH system.
 $[\text{Cu}^{2+}] = 2 \cdot 10^{-3} \text{ mol/dm}^3$, $c_{\text{HAAHVH}} = 2 \cdot 10^{-3} \text{ mol/dm}^3$

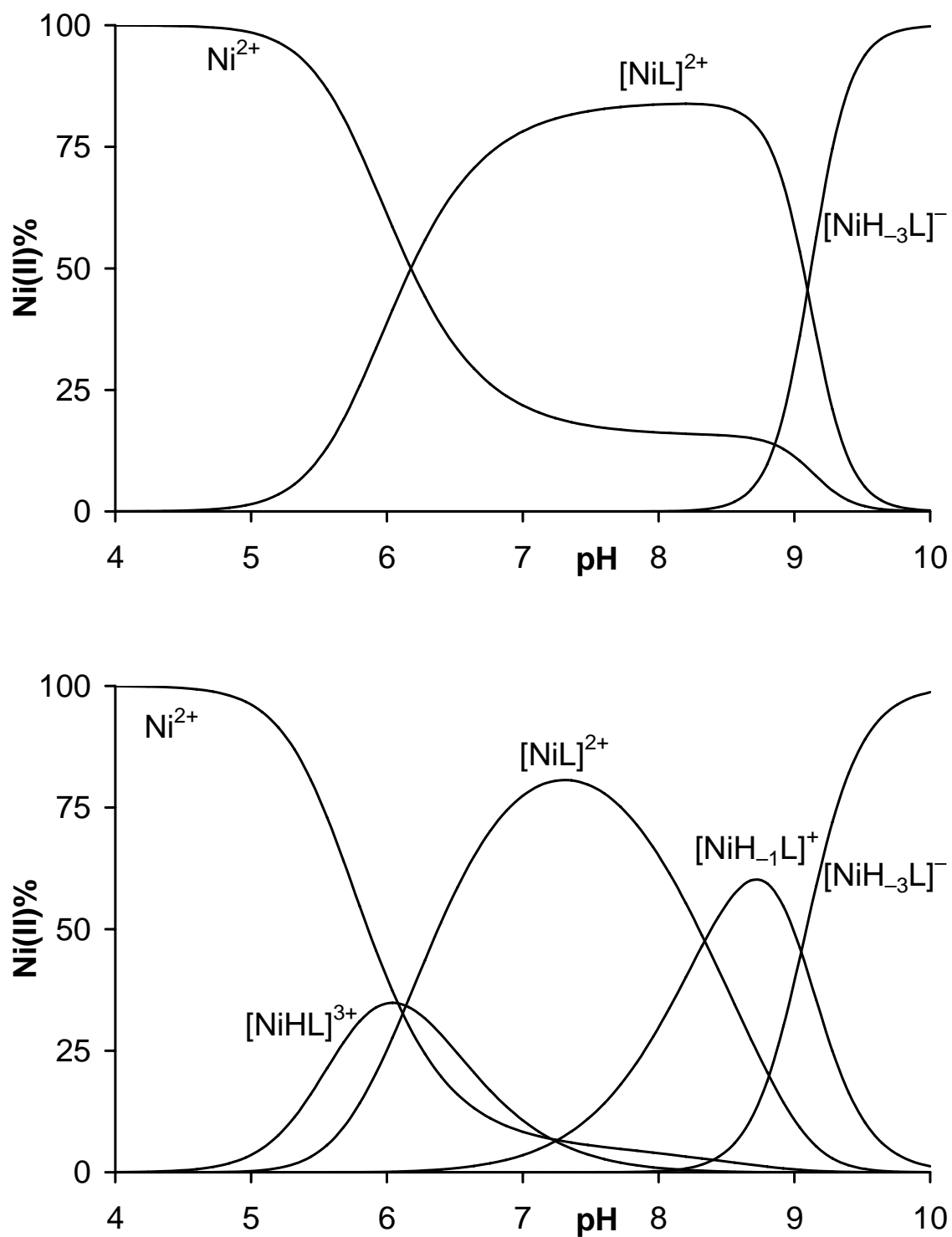


Figure 10.

Concentration distribution of the complexes formed in the nickel(II)-HVVH (a) and HAAHVVH (b) systems.

$[\text{Ni}^{2+}] = 1 \cdot 10^{-3} \text{ mol/dm}^3$, $c_L = 2 \cdot 10^{-3} \text{ mol/dm}^3$