

d¹⁰ Metal coordination polymers based on analogue di(pyridyl)imidazole derivatives and 4,4'-oxydibenzoic acid: influence of flexible and angular characters of neutral ligands on structural diversity

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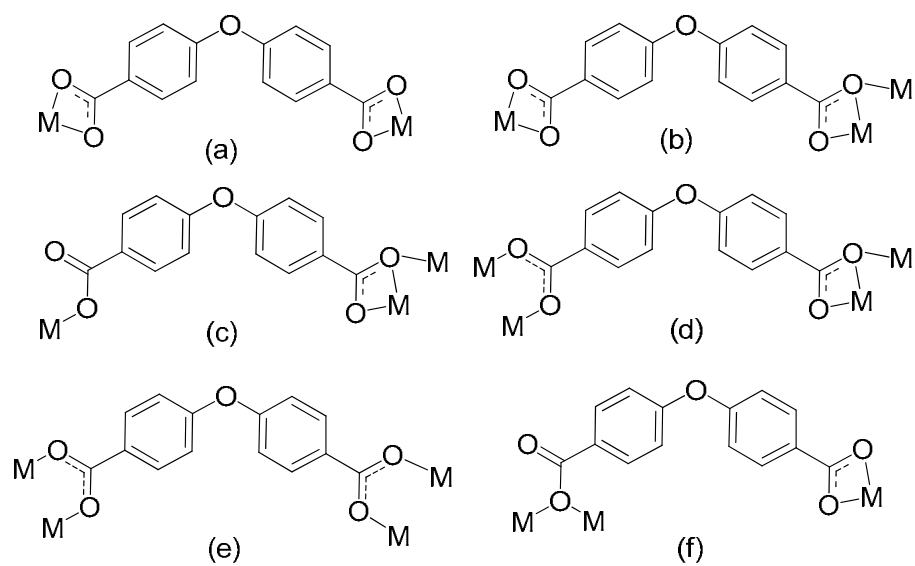


Chart S1. The coordination modes of oba^{2-} anions in this case.

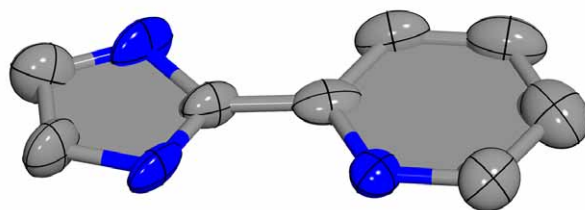


Fig. S1. View of the L^1 ligand in **1**.

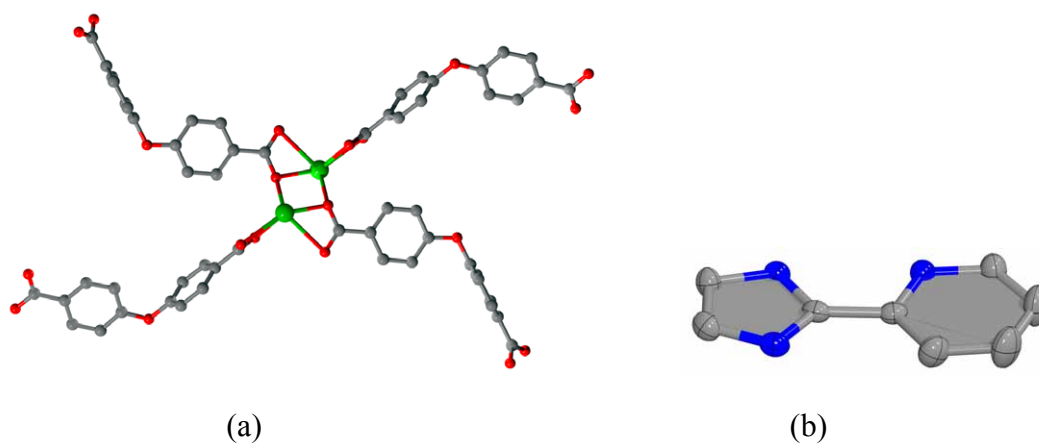


Fig. S2. (a) Ball-and stick representation of the $[\text{Cd}_2(\text{CO}_2)_4]$ unit. (b) View of the L^1 ligand in **2**.

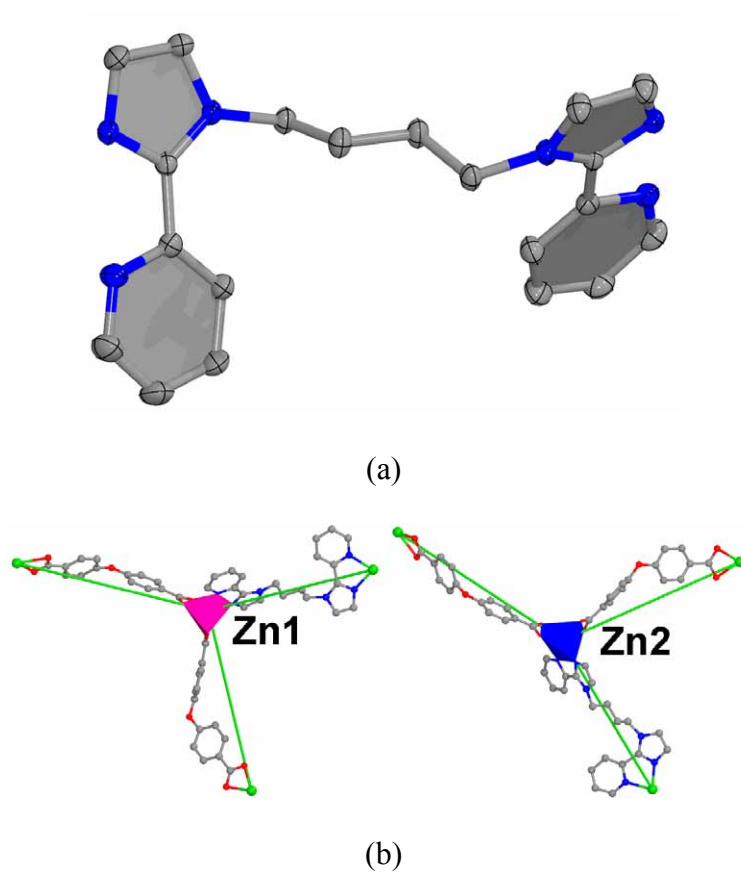
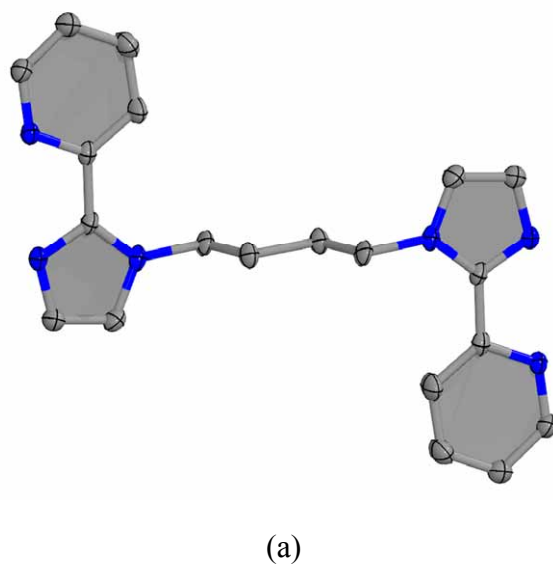


Fig. S3. (a) View of the L^2 ligand in **3**. (b) Ball-and-stick and polyhedral representation of the 3-connected node.



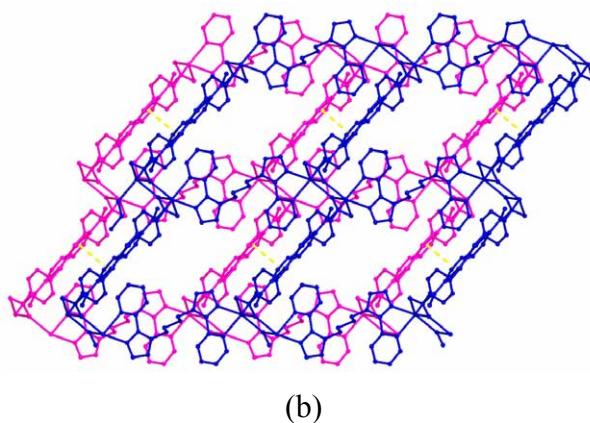


Fig. S4. (a) View of the L^2 ligand in **4**. (b) Ball-and-stick representation of the 3D supramolecular structure of **4**.

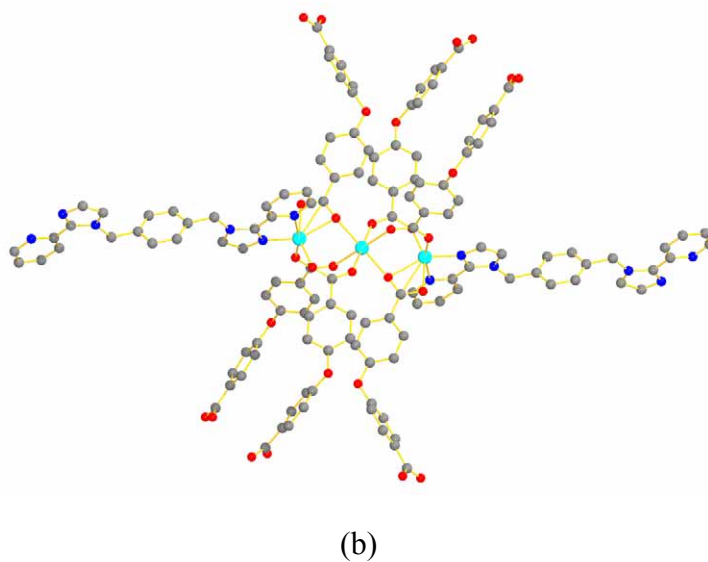
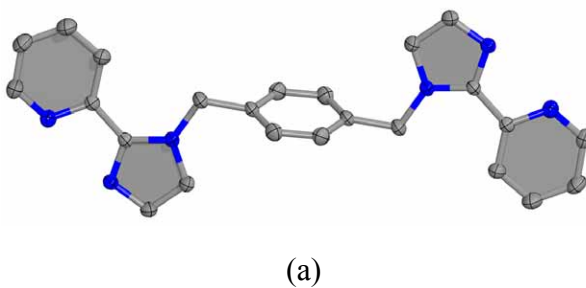


Fig. S5. (a) View of the L^3 ligand in **5**. (b) Ball-and-stick representation of the 8-connected node in **5**.

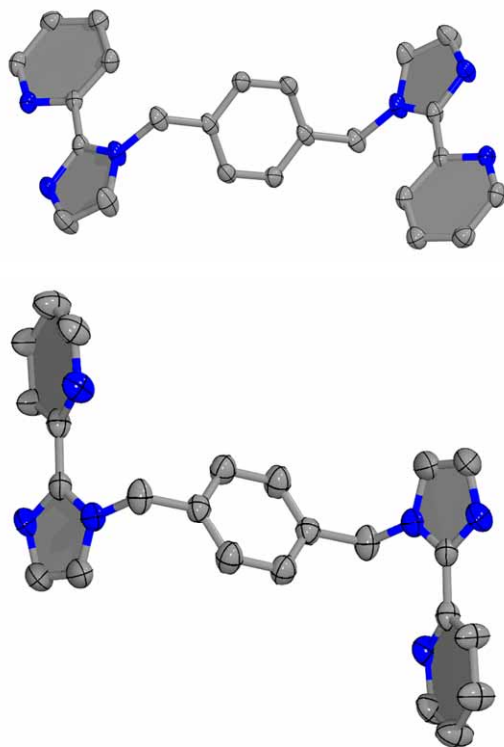
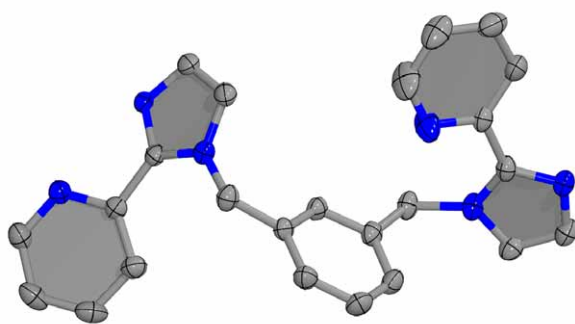
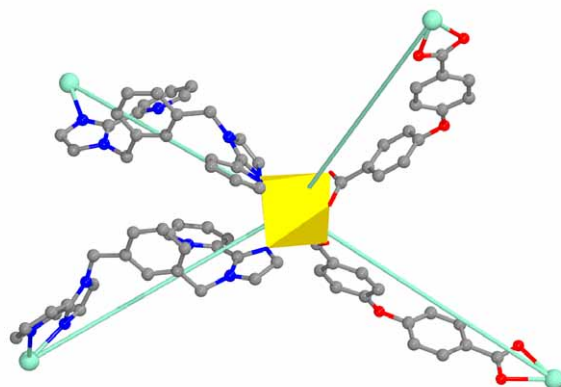


Fig. S6. View of two kinds of L^3 ligands in **6**.



(a)



(b)

Fig. S7. (a) View of the L^4 ligand in **7**. (b) Ball-and-stick and polyhedral representation of the 4-connected node in **7**.

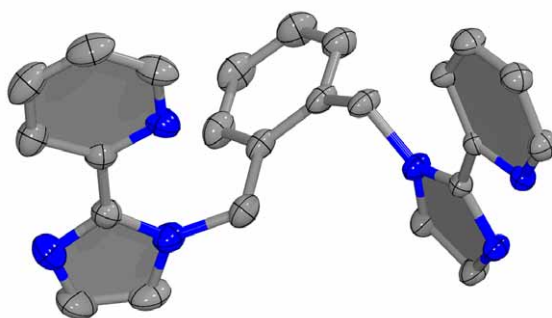


Fig. S8. (a) View of the L^5 ligand in **8**.

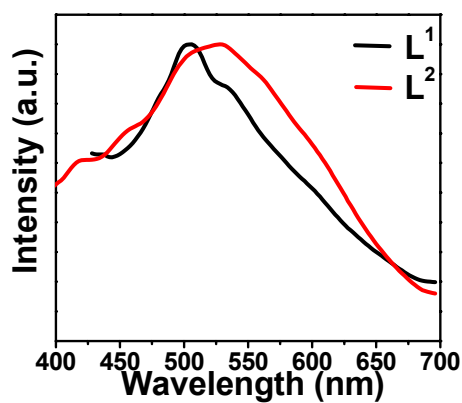
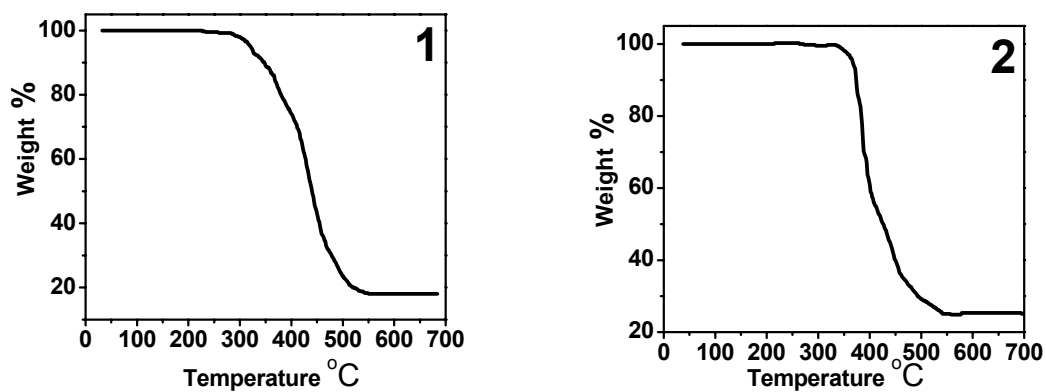


Fig. S9. Solid-state photoluminescent spectra of free ligands L^1 - L^2 at room temperature.



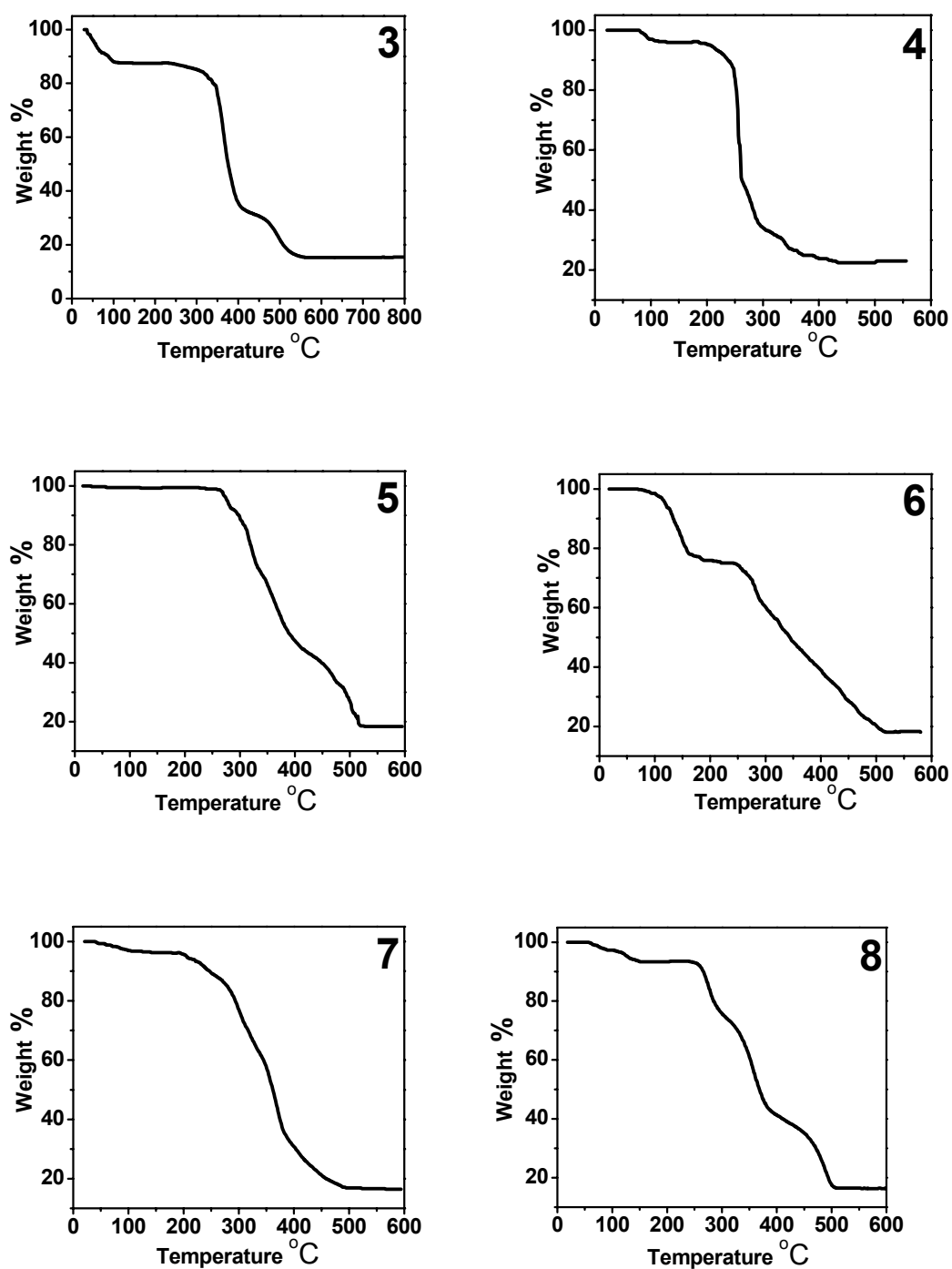


Fig. S10. TGA of compounds **1-8**.

Table S1. Selected Hydrogen-Bonding Geometry for **1, 2, 3, 4, 7** and **8** (in Å and deg)

D-H...A	d(D-H)	d(H...A)	d(D...A)	∠(D-H...A)
Compound 1				
N(1)-H(1A)...O(5)#3	0.86	2.36	3.116(7)	146.4

Compound 2

N(2)-H(1N)...O(4)#5	0.87(3)	1.92(3)	2.767(2)	164(3)
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Compound 3

O(2W)-H(2A)...O(1W)#5	0.80(8)	2.00(9)	2.787(5)	169(12)
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O(1W)-H(1A)...O(8W)	0.81(7)	2.13(8)	2.770(8)	136(10)
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O(1W)-H(1B)...O(7)	0.79(10)	2.12(7)	2.854(5)	152(13)
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O(2W)-H(2B)...O(6W)	0.81(8)	2.09(7)	2.878(6)	165(11)
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O(3W)-H(3A)...O(5W)#6	0.81(9)	2.22(8)	2.854(6)	136(11)
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O(3W)-H(3B)...O(6)	0.79(8)	2.12(7)	2.840(5)	150(12)
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O(4W)-H(4A)...O(3W)#7	0.81(10)	2.11(10)	2.895(7)	163(12)
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O(4W)-H(4B)...O(2)	0.80(10)	2.38(6)	3.124(6)	155(13)
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O(5W)-H(5A)...O(2W)#8	0.80(9)	2.09(10)	2.882(6)	172(12)
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O(5W)-H(5B)...O(4W)#4	0.80(10)	2.11(9)	2.910(6)	175(13)
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O(6W)-H(6A)...O(7W)#9	0.81(6)	1.93(3)	2.722(6)	168(15)
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O(6W)-H(6B)...O(4)#10	0.81(10)	2.07(10)	2.857(6)	164(10)
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O(7W)-H(7A)...O(3W)#7	0.80(9)	2.00(10)	2.769(6)	162(13)
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O(7W)-H(7B)...O(2)	0.80(10)	2.07(9)	2.861(6)	168(12)
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O(8W)-H(8B)...O(10)#2	0.82(9)	2.22(10)	2.889(7)	130(13)
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Compound 4

O(1W)-H(1B)...O(3)	0.85(6)	1.99(5)	2.825(5)	167(6)
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Compound 7

O(1W)-H(1A)...O(1)	0.77(4)	2.09(4)	2.859(3)	173(5)
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O(1W)-H(1B)...O(5)#5	0.77(4)	2.15(4)	2.920(3)	176(5)
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Compound 8

O(1W)-H(1A)...N(5)#4	0.81(4)	2.22(3)	2.950(5)	151(5)
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O(1W)-H(1B)...O(4)#1	0.81(4)	2.08(4)	2.870(3)	163(5)
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O(2W)-H(2A)...O(1W)	0.82(4)	2.00(5)	2.804(5)	167(5)
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O(2W)-H(2B)...O(3W)	0.81(4)	2.26(3)	2.958(9)	144(4)
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Symmetry transformations used to generate equivalent atoms for **1**: #3 -x+3/2, y-1/2, -z+3/2; for **2**: #5 -x+1, -y+1, -z+1; for **3**: #2 x+1/2, -y-1/2, z-1/2; #4 x+1/2, -y+3/2,

$z+1/2$; #5 $-x+1/2, y+1/2, -z+1/2$; #6 $x, y-1, z-1$; #7 $-x+1/2, y+1/2, -z+3/2$; #8 $-x+1, -y+1, -z+2$; #9 $x+1/2, -y+1/2, z-1/2$; #10 $-x+3/2, y-1/2, -z+3/2$; for 7: #5 $-x+3/2, y-1/2, -z+3/2$; for 8: #1 $x, y, z-1$; #4 $-x+1, -y, -z$.