

P,N-Containing Cyclophanes with Large Helical Hydrophobic Cavities: Prospective Precursors for the Design of a Molecular Reactor

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Electronic Supplementary Information

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1) 1D ^1H ROESY spectra of 1

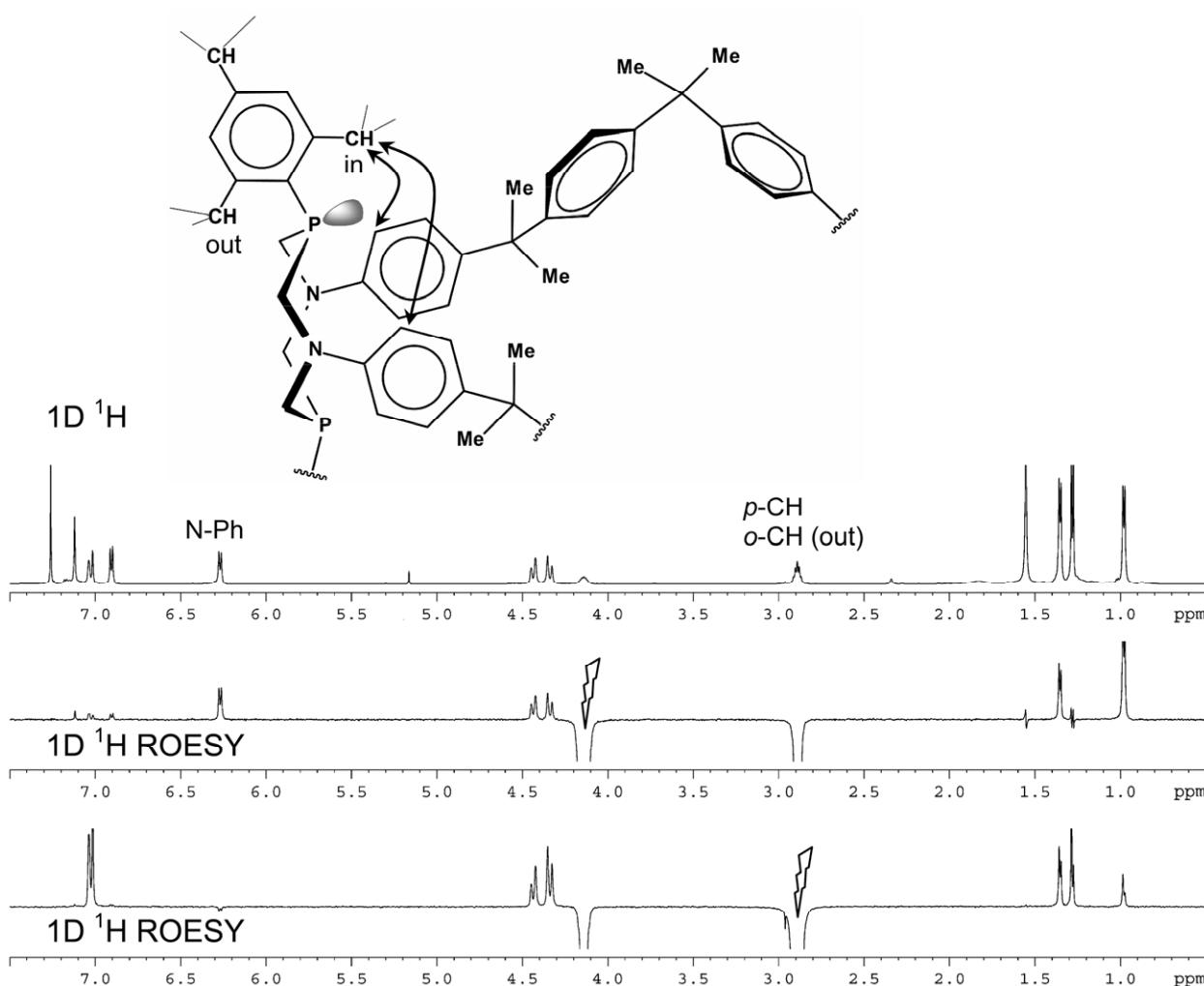


Fig. S1 1D ^1H ROESY spectra of **1** in CDCl_3 at $T=288\text{K}$ and fragment of the molecule with observed nOEs.

2) The binding constants measurement (30°C , CDCl_3).

The binding constants values K_a were calculated from single host and guest concentrations according to eqs 1 and 2, because the large excess of the guest was used.

$$K_a = \frac{P}{(1 - P) \cdot C_{\text{guest}}} \quad (1)$$

$$P = \frac{\Delta\delta_{\text{observed}}}{\Delta\delta_{\text{max}}} \quad (2)$$

Table 1 “Single-point” binding constants K_a of hosts **1** and **2** and aromatic guests in CDCl_3

Host	Guest	$C_{\text{host}}, 10^{-3} \text{M}$	$C_{\text{guest}}, 10^{-2} \text{M}$	$\Delta\delta, \text{ppm}^c$	K_a, M^{-1}
1^a	C_6D_6	6.73	6.73	-0.020	0.65
1^a	C_6D_6	6.57	13	-0.039	0.68
1^a	C_6D_6	2	4	-0.012	0.64
2^b	C_6D_6	2	4	-0.012	0.56
1^c	$\text{C}_6\text{D}_5\text{CD}_3$	2	4	-0.055	2.75
2^d	$\text{C}_6\text{D}_5\text{CD}_3$	2	4	-0.016	0.66

²⁰ ^a $\Delta\delta_{\text{max}} = -0.48 \text{ ppm}$; ^b $\Delta\delta_{\text{max}} = -0.55 \text{ ppm}$; ^c $\Delta\delta_{\text{max}} = -0.55 \text{ ppm}$; ^d $\Delta\delta_{\text{max}} = -0.61 \text{ ppm}$; ^e for the signal of the central phenylene group