

## Supporting Information

### Regioselective Cyclometalation Reactions of Cobalt in Arylketones: C-H versus C-F Activation

Sebnem Camadanli<sup>\*,a,b</sup>, Robert Beck,<sup>a</sup> Ulrich Flörke,<sup>c</sup> and Hans-Friedrich Klein<sup>a</sup>

[<sup>a</sup>] Dr. Sebnem Camadanli, Dr. Robert Beck, Prof. Dr. Hans-Friedrich Klein  
Eduard-Zintl-Institut für Anorganische und Physikalische Chemie  
Technische Universität Darmstadt,  
Petersenstraße 18, Darmstadt  
D-64287, Germany

[<sup>b</sup>] Present Address:  
Leibniz Institute for Surface Modification  
Permoserstrasse 15, Leipzig  
D-04318, Germany

[<sup>c</sup>] Anorganische und Analytische Chemie  
Universität Paderborn  
Warburger Strasse 100, Paderborn  
D-33098, Germany

**General Remarks.** All manipulations were done under an atmosphere of purified argon (BTS - Catalyst) by using standard vacuum techniques.<sup>1-2</sup> Solvents were dried according to known procedures and freshly distilled prior to use. Deuterated tetrahydrofuran were freeze-pump-thaw degassed, dried from sodium/potassium alloy and distilled. Purchased reagents were used without any further purification. Literature methods were applied for the preparation of  $[\text{CoCH}_3(\text{PMe}_3)_4]^3$  and  $\text{PMe}_3$ .<sup>4</sup>

Air-sensitive samples were provided in capillaries sealed under vacuum and were analyzed by H. Kolbe Microanalytical Laboratory, Mülheim/Ruhr. C, H, N analyses identification of contents of air stable substances were performed in the microanalysis laboratory of Clemens-Schöpf-Institut for Organic Chemistry and Biochemistry in TU-Darmstadt using Perkin-Elmer CHN 240 A and CHN 240 B devices. The melting and decomposition points were measured on a Büchi 510 Melting point apparatus and are uncorrected. Air-sensitives substances were sealed in capillaries under 1 bar of Argon.

IR spectra were obtained from as Nujol mulls between KBr plates using a Bruker FRA 106 spectrophotometer and recorded in the range 4000 - 400  $\text{cm}^{-1}$ .  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra (500 MHz and 125 MHz, respectively) were recorded on a Bruker ARX-300 or Bruker DRX 500 spectrometers at the Organic Chemistry Institute of TU Darmstadt,  $^{31}\text{P}$  NMR spectra (202 MHz) were recorded on a Bruker AM-200 instrument.  $^1\text{H}$  and  $^{13}\text{C}$  NMR chemical shifts were referenced to external TMS and  $^{31}\text{P}$  chemical shifts were referenced to external  $\text{H}_3\text{PO}_4$  85%.  $^{13}\text{C}$  and  $^{31}\text{P}$  resonances were obtained with broad-band proton decoupling. Magnetic susceptibility data were obtained by the Faraday method using a Cahn D 200 torsion balance (Bruker) at 1.5 Tesla. Crystallographic data were collected on STOE-IPDS (Eduard-Zintl Institute of TU-Darmstadt) and Bruker AXS SMART APEX CCD (University of Paderborn) diffractometer. Structures were solved by Direct and Patterson methods using SHELXTL program library.<sup>5</sup>

## Synthesis of [2-(Benzoyl- $\kappa\text{O}$ )phenyl- $\kappa\text{C}$ ]tris(trimethylphosphine)cobalt(I) – (1)

920 mg (2.43 mmol) of  $\text{CoCH}_3(\text{PMe}_3)_4$  in THF were combined with 443 mg (2.43 mmol) of benzophenone at  $-70\text{ }^\circ\text{C}$ . The mixture was warmed to  $20\text{ }^\circ\text{C}$  and kept stirring for 16 h. The volatiles were then removed in vacuo. The residue was extracted with pentane and combined solutions were cooled to afford brown crystals of **1** at  $4\text{ }^\circ\text{C}$ . Yield 924 mg (81%); m. p.  $111\text{--}113\text{ }^\circ\text{C}$ . IR (Nujol):  $\tilde{\nu} = 3057\text{ cm}^{-1}$  (v H–C=C);  $1582\text{ cm}^{-1}$  m (v C=C);  $1484\text{ cm}^{-1}$  m (v C=O),  $1422\text{ cm}^{-1}$  w ( $\delta_{\text{as}}\text{PCH}_3$ ),  $1273\text{ cm}^{-1}$  m ( $\delta_{\text{s}}\text{PCH}_3$ ),  $935\text{ cm}^{-1}$  vs ( $\rho_1\text{PCH}_3$ ),  $842\text{ cm}^{-1}$  m ( $\rho_2\text{PCH}_3$ ),  $753\text{ vw}$  ( $\gamma\text{C-H}_{\text{arom}}$ ),  $701\text{ cm}^{-1}$  m ( $\nu_{\text{as}}\text{PC}_3$ ),  $661\text{ cm}^{-1}$  vs ( $\nu_{\text{s}}\text{PC}_3$ ).  $^1\text{H NMR}$  (500 MHz, THF- $d_8$ , 300 K, ppm):  $\delta = 1.30$  (d,  $^2J_{\text{P,H}} = 6.5$  Hz, 27H,  $\text{PCH}_3$ ); 6.61 (dt,  $^4J_{\text{H,H}} = 0.8$  Hz,  $^3J_{\text{H,H}} = 7.8$  Hz, 1H, Ar–H); 7.12 (t,  $^3J_{\text{H,H}} = 7.8$  Hz, 2H, Ar–H); 7.33 (tt,  $^4J_{\text{H,H}} = 1.2$  Hz,  $^3J_{\text{H,H}} = 8.0$  Hz, 1H, Ar–H); 7.46 (dt,  $^4J_{\text{H,H}} = 0.9$  Hz,  $^3J_{\text{H,H}} = 7.4$  Hz, 1H, Ar–H); 8.11 (d,  $^3J_{\text{H,H}} = 8.1$  Hz, 2H, Ar–H); 8.19 (d<sub>br</sub>,  $^3J_{\text{H,H}} = 8.1$  Hz, 1H, Ar–H); 8.22 (d,  $^3J_{\text{H,H}} = 8.2$  Hz, 1H, Ar–H).  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz, THF- $d_8$ , 300 K, ppm):  $\delta = 21.3$  (m,  $\text{PCH}_3$ ); 118.3 (q,  $^4J_{\text{P,C}} = 4.6$  Hz, CH); 121.9 (s, CH); 122.6 (s, CH); 123.7 (d,  $^5J_{\text{P,C}} = 2.4$  Hz, CH); 124.2 (dd,  $^4J_{\text{P,C}} = 3.4$  Hz,  $^4J_{\text{P,C}} = 2.2$  Hz, CH); 130.4 (s, CH); 145.2 (q,  $^4J_{\text{P,C}} = 4.6$  Hz, C); 146.2 (s, C); 147.8 (q,  $^5J_{\text{P,C}} = 2.2$  Hz, CH); 170.8 (q,  $^3J_{\text{P,C}} = 11.8$  Hz; C=O), 173.9 (m, Co–C).  $^{31}\text{P}\{^1\text{H}\}$  NMR (202 MHz, THF- $d_8$ , 300 K, ppm):  $\delta = 5.1$  (s<sub>br</sub>,  $\text{PCH}_3$ ). Anal. Calcd. for  $\text{C}_{22}\text{H}_{36}\text{CoOP}_3$ : C 56.42; H 7.75; P 19.84. Found: C 56.95; H 7.53; P 20.68 %.

### Synthesis of [2-(Benzoyl- $\kappa\text{O}$ )benzyl- $\kappa\text{C}$ ]tris(trimethylphosphine)cobalt(I) - (2)

1.15 g (3.04 mmol) of  $\text{CoCH}_3(\text{PMe}_3)_4$  in THF were combined with 0.60 g (3.04 mmol) of benzylphenylketone at  $-70\text{ }^\circ\text{C}$ . The mixture was warmed to  $20\text{ }^\circ\text{C}$  and kept stirring for 16 h. The volatiles were then removed in vacuo. The residue was extracted with pentane and combined solutions were cooled to afford brown crystals of **2** at  $-27\text{ }^\circ\text{C}$ . Yield 970 mg (66 %); m. p.  $107\text{--}109\text{ }^\circ\text{C}$ . IR (Nujol):  $\tilde{\nu} = 3023\text{ cm}^{-1}$  m (v H–C=C),  $1547\text{ cm}^{-1}$  s (v C=C),  $1483\text{ cm}^{-1}$  m (v C=O),  $1421\text{ s}$  ( $\delta_{\text{as}}\text{PCH}_3$ ),  $1278\text{ cm}^{-1}$  m ( $\delta_{\text{s}}\text{PCH}_3$ ),  $937\text{ cm}^{-1}$  vs ( $\rho_1\text{PCH}_3$ );  $874\text{ cm}^{-1}$  m ( $\rho_2\text{PCH}_3$ ),  $748\text{ cm}^{-1}$  ( $\gamma\text{C-H}_{\text{arom}}$ ),  $698\text{ cm}^{-1}$  vs ( $\nu_{\text{as}}\text{PC}_3$ ),  $657\text{ cm}^{-1}$  vs ( $\nu_{\text{s}}\text{PC}_3$ ).  $^1\text{H NMR}$  (500 MHz, THF- $d_8$ , 300 K, ppm):  $\delta = 1.05$  (s<sub>br</sub>, 27H,  $\text{PCH}_3$ ); 4.20 (s, 2H,  $\text{CH}_2$ ); 7.25–7.55 (m, 6H, Ar–H); 8.03 (m, 1H, Ar–H); 8.95 (m,

1H, Ar-H); 9.19 (m, 1H, Ar-H).  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz, THF- $d_8$ , 300 K, ppm):  $\delta$  = 20.9 (m, PCH<sub>3</sub>); 46.1 (s, CH<sub>2</sub>); 127.5 (s, CH); 129.4 (s, CH); 129.6 (d,  $^3J_{\text{P,C}}$  = 8.2 Hz, CH); 130.6 (s, CH); 133.8 (s, CH); 136.5 (s, CH); 138.2 (s, C); 151.2 (m, C); 158.4 (s, CH); 166.1 (s, Co-C=O); 183.7 (m, Co-C).  $^{31}\text{P}\{^1\text{H}\}$  NMR (202 MHz, THF- $d_8$ , 300 K, ppm):  $\delta$  = 1.5 (s<sub>br</sub>, 3P, PCH<sub>3</sub>). Anal. Calcd. for C<sub>23</sub>H<sub>38</sub>CoOP<sub>3</sub>: C 57.27; H 7.94; P 19.26. Found: C 56.49; H 7.39; P 19.53 %.

### Synthesis of [2-(3,4,5,6-Tetrafluoro-benzoyl- $\kappa\text{O}$ )phenyl- $\kappa\text{C}$ ]tris-(trimethylphosphine)cobalt(I) - (3)

1.20 g (3.17 mmol) of CoCH<sub>3</sub>(PMe<sub>3</sub>)<sub>4</sub> in THF were combined with 0.86 g (3.17 mmol) of 2,3,4,5,6-pentafluorobenzophenone at -70 °C. The mixture was warmed to 20 °C and kept stirring for 16 h. The volatiles were then removed in vacuo. The residue was extracted with pentane and combined solutions were cooled to afford brown crystals of **3** at -27 °C. Yield: 1.08 g (63 %); m. p. 152 -154 °C. IR (Nujol):  $\tilde{\nu}$  = 3077 cm<sup>-1</sup> vw (v H-C=C); 1617 cm<sup>-1</sup> m (v C=C); 1590 cm<sup>-1</sup> m (v C=C); 1490 cm<sup>-1</sup> m (v C=O), 1420 cm<sup>-1</sup> w ( $\delta_{\text{as}}$  PCH<sub>3</sub>); 1339 cm<sup>-1</sup> s (v F-C=C), 1277 cm<sup>-1</sup> s ( $\delta_{\text{s}}$  PCH<sub>3</sub>), 939 cm<sup>-1</sup> vs ( $\rho_1$  PCH<sub>3</sub>); 830 cm<sup>-1</sup> m ( $\rho_2$  PCH<sub>3</sub>), 760 m ( $\gamma$  C-H<sub>arom</sub>), 701 cm<sup>-1</sup> m ( $\nu_{\text{as}}$  PC<sub>3</sub>), 665 cm<sup>-1</sup> m ( $\nu_{\text{s}}$  PC<sub>3</sub>).  $^1\text{H}$  NMR (500 MHz, THF- $d_8$ , 300 K, ppm):  $\delta$  = 1.32 (d,  $^2J_{\text{P,H}}$  = 6.1 Hz, 27H, PCH<sub>3</sub>); 7.09 (t,  $^3J_{\text{H,H}}$  = 7.2 Hz, 2H, Ar-H); 7.48 (t,  $^3J_{\text{H,H}}$  = 7.0 Hz, 1H, Ar-H); 7.81 (t,  $^3J_{\text{H,H}}$  = 6.5 Hz, 2H, Ar-H).  $^{13}\text{C}\{^1\text{H}\}$  NMR (125 MHz, THF- $d_8$ , 300 K, ppm):  $\delta$  = 20.5 (d,  $^1J_{\text{P,C}}$  = 20.5 Hz, PCH<sub>3</sub>); 124.0 (s, CH); 124.3 (s, CH); 128.5 (m, CF); 128.8 (m, CF); 129.1 (s, CH); 129.4 (m, CF); 134.5 (m, CF); 141.5 (m, C); 147.4 (s, C); 167.9 (s, C=O); 168.7 (m, Co-C).  $^{31}\text{P}\{^1\text{H}\}$  NMR (202 MHz, THF- $d_8$ , 300 K, ppm):  $\delta$  = 3.6 (s<sub>br</sub>, PCH<sub>3</sub>). Anal. Calcd. for C<sub>22</sub>H<sub>32</sub>CoF<sub>4</sub>OP<sub>3</sub>: C 48.90; H 5.97; P 17.20. Found: C 48.58; H 5.70; P 16.88 %.

### Synthesis of (2-Fluoro-benzophenone- $\eta^2$ -C,O)tris(trimethylphosphine)cobalt(0) - (4)

1.05 g (2.79 mmol) of CoCH<sub>3</sub>(PMe<sub>3</sub>)<sub>4</sub> in THF were combined with 0.56 g (2.79 mmol) of 2-fluorobenzophenone at -70 °C. The mixture was warmed to 20 °C and kept stirring for 16 h. The

volatiles were then removed in vacuo. The residue was extracted with pentane and combined solutions were cooled to afford brown crystals of **4** at -27 °C. Yield: 1.05 g (77 %); m. p. 98 -102 °C. IR (Nujol):  $\tilde{\nu}$  = 3000  $\text{cm}^{-1}$  vw ( $\nu$  H-C=C); 1878  $\text{cm}^{-1}$  s ( $\nu$  C=O); 1604  $\text{cm}^{-1}$  m ( $\nu$  C=C); 1587  $\text{cm}^{-1}$  m ( $\nu$  C=C), 1423  $\text{cm}^{-1}$  w ( $\delta_{\text{as}}$  PCH<sub>3</sub>), 1274  $\text{cm}^{-1}$  s ( $\delta_{\text{s}}$  PCH<sub>3</sub>), 933  $\text{cm}^{-1}$  vs ( $\rho_1$  PCH<sub>3</sub>), 841  $\text{cm}^{-1}$  m ( $\rho_2$  PCH<sub>3</sub>), 765 m ( $\gamma$  C-H<sub>arom</sub>), 695  $\text{cm}^{-1}$  m ( $\nu_{\text{as}}$  PC<sub>3</sub>), 656  $\text{cm}^{-1}$  m ( $\nu_{\text{s}}$  PC<sub>3</sub>). Anal. Calcd. for C<sub>22</sub>H<sub>36</sub>CoFOP<sub>3</sub>: C 54.22; H 7.45; P 19.07. Found: C 54.19; H 7.43; P 18.55 %.

### References:

- (1) Schulze, M. *Angew. Chem.* **1958**, *70*, 697.
- (2) Strohmeier, W. *Chem. Ber.* **1955**, *88*, 1218.
- (3) Klein, H.-F.; Karsch, H. H. *Chem. Ber.* **1975**, *108*, 944.
- (4) Wolfsberger, W.; Schmidbaur, H. *Syn. React. Inorg. Metal-Org. Chem.* **1974**, *4*, 149.
- (5) Sheldrick, G. M.; Krüger, C.; Goddard, R. *SHELXS-97* and *SHELXL-97*, Universität Göttingen, **1997**.

### Crystallographic Data

**Table 1.** Crystal data and structure refinement for **1**

Identification code	k1096
Empirical formula	C22 H36 Co O P3

Formula weight	468.35	
Temperature	298(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	Pna2(1)	
Unit cell dimensions	a = 16.4913(9) Å	$\alpha = 90^\circ$ .
	b = 9.5041(5) Å	$\beta = 90^\circ$ .
	c = 15.8471(9) Å	$\gamma = 90^\circ$ .
Volume	2483.8(2) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.252 Mg/m <sup>3</sup>	
Absorption coefficient	0.893 mm <sup>-1</sup>	
F(000)	992	
Crystal size	0.50 x 0.45 x 0.40 mm <sup>3</sup>	
Theta range for data collection	2.47 to 28.30°.	
Index ranges	-21 ≤ h ≤ 21, -12 ≤ k ≤ 12, -19 ≤ l ≤ 21	
Reflections collected	23133	
Independent reflections	5960 [R(int) = 0.0234]	
Completeness to theta = 28.30°	99.8 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.983 and 0.699	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	5960 / 1 / 254	
Goodness-of-fit on F <sup>2</sup>	1.051	
Final R indices [I > 2σ(I)]	R1 = 0.0295, wR2 = 0.0705	
R indices (all data)	R1 = 0.0339, wR2 = 0.0727	
Absolute structure parameter	0.005(9)	
Extinction coefficient	0.0011(3)	
Largest diff. peak and hole	0.256 and -0.266 e.Å <sup>-3</sup>	

**Table 2.** Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **1**.  $U(\text{eq})$  is defined as one third of the trace of the orthogonalized  $U^{ij}$  tensor.

	x	y	z	U(eq)
Co(1)	2849(1)	2035(1)	2493(1)	35(1)
P(1)	1810(1)	1023(1)	3086(1)	49(1)
P(2)	2264(1)	4111(1)	2216(1)	44(1)
P(3)	3010(1)	1283(1)	1201(1)	46(1)
O(1)	3423(1)	3165(1)	3287(1)	39(1)
C(1)	3556(1)	581(2)	2927(1)	44(1)
C(2)	3662(2)	-840(2)	2708(2)	59(1)
C(3)	4258(2)	-1679(2)	3066(2)	68(1)
C(4)	4777(2)	-1125(2)	3667(2)	65(1)
C(5)	4681(1)	222(2)	3927(2)	56(1)
C(6)	4065(1)	1088(2)	3591(1)	43(1)
C(7)	3921(1)	2532(2)	3795(1)	39(1)
C(8)	4264(1)	3355(2)	4485(1)	43(1)
C(9)	4551(2)	2751(3)	5239(2)	58(1)
C(10)	4896(2)	3586(3)	5861(2)	74(1)
C(11)	4964(2)	5003(3)	5761(2)	76(1)
C(12)	4658(2)	5626(3)	5048(2)	66(1)
C(13)	4310(1)	4814(2)	4420(1)	49(1)
C(14)	1818(2)	1313(4)	4227(2)	89(1)
C(15)	1673(2)	-880(3)	3052(2)	84(1)
C(16)	773(2)	1559(3)	2814(2)	75(1)
C(17)	1382(2)	4418(4)	1535(2)	90(1)
C(18)	2983(2)	5412(3)	1813(2)	73(1)
C(19)	1939(2)	5006(3)	3178(2)	69(1)
C(20)	4060(2)	882(4)	903(2)	92(1)
C(21)	2493(2)	-319(3)	843(2)	72(1)
C(22)	2740(3)	2414(4)	307(2)	97(1)

**Table 3.** Bond lengths [Å] and angles [°] for **1**.

---

Co(1)-O(1)	1.9051(12)
Co(1)-C(1)	1.9339(18)
Co(1)-P(1)	2.1786(6)
Co(1)-P(3)	2.1852(6)
Co(1)-P(2)	2.2401(5)
P(1)-C(15)	1.823(2)
P(1)-C(14)	1.830(3)
P(1)-C(16)	1.835(3)
P(2)-C(19)	1.826(3)
P(2)-C(18)	1.828(3)
P(2)-C(17)	1.834(3)
P(3)-C(22)	1.834(3)
P(3)-C(20)	1.835(3)
P(3)-C(21)	1.835(2)
O(1)-C(7)	1.297(2)
C(1)-C(2)	1.405(3)
C(1)-C(6)	1.430(3)
C(2)-C(3)	1.386(3)
C(3)-C(4)	1.385(4)
C(4)-C(5)	1.355(3)
C(5)-C(6)	1.411(3)
C(6)-C(7)	1.430(3)
C(7)-C(8)	1.460(3)
C(8)-C(13)	1.393(3)
C(8)-C(9)	1.407(3)
C(9)-C(10)	1.387(4)
C(10)-C(11)	1.360(4)
C(11)-C(12)	1.372(4)
C(12)-C(13)	1.384(3)
O(1)-Co(1)-C(1)	82.49(7)
O(1)-Co(1)-P(1)	110.84(5)
C(1)-Co(1)-P(1)	90.33(6)
O(1)-Co(1)-P(3)	137.94(5)



C(1)-Co(1)-P(3)	91.46(6)
P(1)-Co(1)-P(3)	110.77(2)
O(1)-Co(1)-P(2)	81.21(4)
C(1)-Co(1)-P(2)	163.52(6)
P(1)-Co(1)-P(2)	97.73(2)
P(3)-Co(1)-P(2)	99.01(2)
C(15)-P(1)-C(14)	100.32(16)
C(15)-P(1)-C(16)	98.84(15)
C(14)-P(1)-C(16)	101.36(17)
C(15)-P(1)-Co(1)	121.54(10)
C(14)-P(1)-Co(1)	110.66(11)
C(16)-P(1)-Co(1)	120.63(10)
C(19)-P(2)-C(18)	99.62(14)
C(19)-P(2)-C(17)	100.60(15)
C(18)-P(2)-C(17)	101.62(17)
C(19)-P(2)-Co(1)	111.90(9)
C(18)-P(2)-Co(1)	112.62(10)
C(17)-P(2)-Co(1)	126.68(11)
C(22)-P(3)-C(20)	98.76(19)
C(22)-P(3)-C(21)	97.70(16)
C(20)-P(3)-C(21)	100.74(16)
C(22)-P(3)-Co(1)	120.21(11)
C(20)-P(3)-Co(1)	115.09(13)
C(21)-P(3)-Co(1)	120.32(10)
C(7)-O(1)-Co(1)	117.51(11)
C(2)-C(1)-C(6)	115.63(18)
C(2)-C(1)-Co(1)	132.40(16)
C(6)-C(1)-Co(1)	111.97(13)
C(3)-C(2)-C(1)	122.7(2)
C(4)-C(3)-C(2)	120.1(2)
C(5)-C(4)-C(3)	119.7(2)
C(4)-C(5)-C(6)	121.4(2)
C(5)-C(6)-C(7)	126.45(18)
C(5)-C(6)-C(1)	120.25(18)
C(7)-C(6)-C(1)	113.08(16)
O(1)-C(7)-C(6)	114.23(16)

Supplementary Material (ESI) for Dalton Transactions  
This journal is (c) The Royal Society of Chemistry 2008

O(1)-C(7)-C(8)	117.51(17)
C(6)-C(7)-C(8)	128.25(17)
C(13)-C(8)-C(9)	116.84(18)
C(13)-C(8)-C(7)	119.94(18)
C(9)-C(8)-C(7)	123.22(19)
C(10)-C(9)-C(8)	120.5(2)
C(11)-C(10)-C(9)	121.2(2)
C(10)-C(11)-C(12)	119.5(2)
C(11)-C(12)-C(13)	120.3(2)
C(12)-C(13)-C(8)	121.6(2)

---

**Table 4.** Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **1**. The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2}U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U <sup>11</sup>	U <sup>22</sup>	U <sup>33</sup>	U <sup>23</sup>	U <sup>13</sup>	U <sup>12</sup>
Co(1)	40(1)	29(1)	34(1)	0(1)	-3(1)	0(1)
P(1)	58(1)	45(1)	43(1)	0(1)	9(1)	-11(1)
P(2)	48(1)	36(1)	49(1)	2(1)	-5(1)	5(1)
P(3)	59(1)	42(1)	38(1)	-3(1)	6(1)	0(1)
O(1)	44(1)	34(1)	40(1)	-1(1)	-6(1)	-1(1)
C(1)	50(1)	34(1)	48(1)	2(1)	-6(1)	4(1)
C(2)	76(2)	39(1)	64(2)	-3(1)	-19(1)	5(1)
C(3)	88(2)	39(1)	76(2)	-1(1)	-8(1)	14(1)
C(4)	69(1)	53(1)	74(2)	10(1)	-16(1)	17(1)
C(5)	54(1)	51(1)	62(1)	7(1)	-12(1)	6(1)
C(6)	43(1)	41(1)	44(1)	5(1)	-4(1)	-1(1)
C(7)	38(1)	40(1)	38(1)	3(1)	-1(1)	-1(1)
C(8)	37(1)	53(1)	37(1)	-4(1)	1(1)	-4(1)
C(9)	61(1)	70(1)	44(1)	5(1)	-6(1)	-1(1)
C(10)	71(2)	106(2)	46(1)	-6(1)	-12(1)	2(2)
C(11)	65(2)	100(2)	63(2)	-37(1)	-9(1)	-10(1)
C(12)	66(2)	66(1)	66(2)	-24(1)	4(1)	-11(1)
C(13)	52(1)	52(1)	43(1)	-9(1)	2(1)	-6(1)
C(14)	117(3)	106(2)	43(1)	3(1)	18(2)	-28(2)
C(15)	102(2)	54(1)	95(2)	3(1)	24(2)	-30(1)
C(16)	45(1)	94(2)	84(2)	-5(2)	11(1)	-10(1)
C(17)	79(2)	89(2)	101(2)	4(2)	-40(2)	26(2)
C(18)	93(2)	46(1)	80(2)	9(1)	6(2)	-11(1)
C(19)	83(2)	52(1)	72(2)	-13(1)	7(1)	21(1)
C(20)	73(2)	107(2)	96(2)	-4(2)	42(2)	3(2)
C(21)	99(2)	60(1)	55(2)	-12(1)	-14(1)	-12(2)
C(22)	176(4)	78(2)	38(1)	7(1)	-4(2)	22(2)

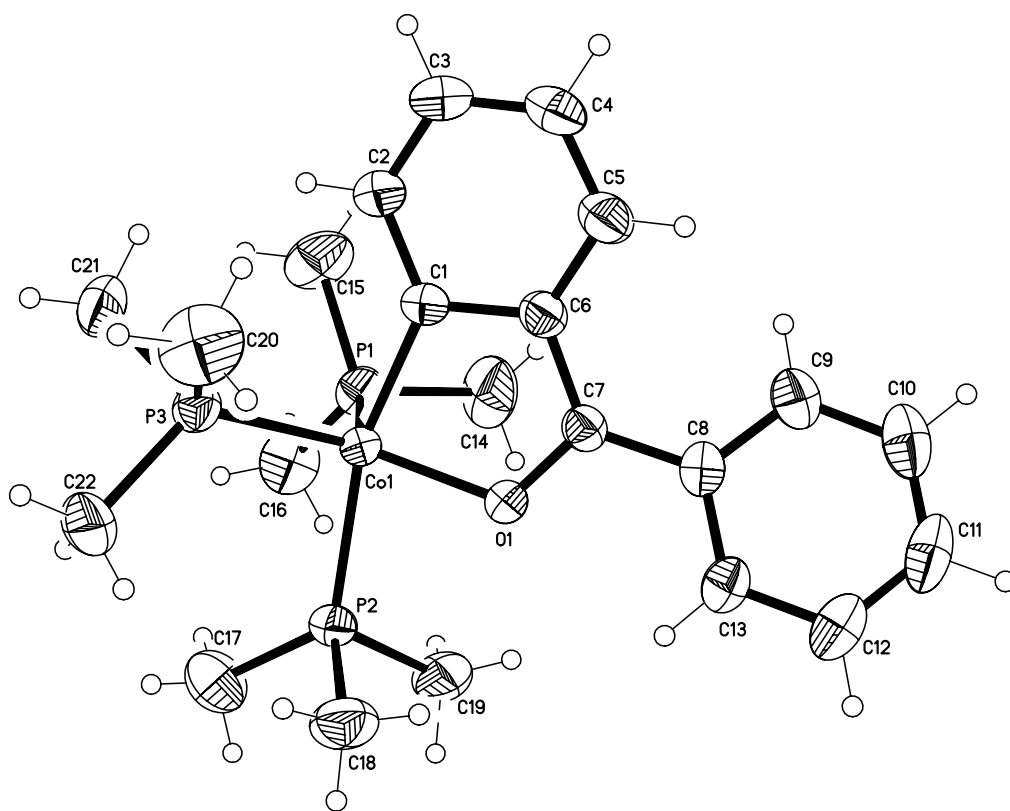
**Table 5.** Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^{-3}$ ) for **1**.

	x	y	z	U(eq)
H(2A)	3319	-1234	2307	71
H(3A)	4308	-2614	2902	81
H(4A)	5191	-1676	3891	78
H(5A)	5028	585	4336	67
H(9A)	4509	1785	5320	70
H(10A)	5084	3170	6354	89
H(11A)	5215	5545	6174	91
H(12A)	4685	6598	4986	79
H(13A)	4101	5254	3943	59
H(14A)	1378	808	4481	133
H(14B)	1760	2300	4342	133
H(14C)	2323	985	4457	133
H(15A)	2159	-1334	3243	125
H(15B)	1558	-1167	2484	125
H(15C)	1229	-1141	3412	125
H(16A)	395	899	3049	112
H(16B)	714	1581	2212	112
H(16C)	668	2479	3040	112
H(17A)	964	3752	1670	135
H(17B)	1538	4308	955	135
H(17C)	1182	5356	1624	135
H(18A)	2712	6298	1739	109
H(18B)	3195	5100	1281	109
H(18C)	3420	5523	2208	109
H(19A)	1790	5958	3047	104
H(19B)	2375	5007	3579	104
H(19C)	1480	4524	3413	104
H(20A)	4080	633	316	138
H(20B)	4255	109	1236	138

Supplementary Material (ESI) for Dalton Transactions  
This journal is (c) The Royal Society of Chemistry 2008

H(20C)	4394	1693	1000	138
H(21A)	1918	-202	902	107
H(21B)	2668	-1103	1178	107
H(21C)	2623	-489	262	107
H(22A)	2917	1982	-209	146
H(22B)	2998	3313	369	146
H(22C)	2162	2538	291	146

---



Molecular structure of **1**

**Table 1.** Crystal data and structure refinement for **3**.

Empirical formula	C <sub>22</sub> H <sub>32</sub> Co F <sub>4</sub> O P <sub>3</sub>	
Formula weight	540.32	
Temperature	150(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P21/c	
Unit cell dimensions	a = 12.7954(16) Å	$\alpha = 90^\circ$ .
	b = 14.830(2) Å	$\beta = 130.890(8)^\circ$ .
	c = 17.434(2) Å	$\gamma = 90^\circ$ .
Volume	2500.9(5) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.435 Mg/m <sup>3</sup>	
Absorption coefficient	0.920 mm <sup>-1</sup>	
F(000)	1120	
Crystal size	0.12 x 0.10 x 0.09 mm <sup>3</sup>	
Theta range for data collection	2.07 to 26.88°.	
Index ranges	-16 ≤ h ≤ 16, -18 ≤ k ≤ 18, -22 ≤ l ≤ 21	
Reflections collected	34404	
Independent reflections	5329 [R(int) = 0.1791]	
Completeness to theta = 26.88°	98.7 %	
Absorption correction	None	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	5329 / 0 / 289	
Goodness-of-fit on F <sup>2</sup>	1.049	
Final R indices [I > 2σ(I)]	R1 = 0.0564, wR2 = 0.1067	
R indices (all data)	R1 = 0.0916, wR2 = 0.1162	
Largest diff. peak and hole	0.525 and -0.623 e.Å <sup>-3</sup>	

**Table 2.** Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **3**.  $U(\text{eq})$  is defined as one third of the trace of the orthogonalized  $U^{ij}$  tensor.

	x	y	z	U(eq)
C(1)	4033(4)	681(2)	-2807(3)	23(1)
C(2)	3606(4)	193(2)	-2372(3)	24(1)
C(3)	2259(4)	142(2)	-2768(3)	26(1)
C(4)	1229(4)	580(3)	-3672(3)	27(1)
C(5)	1559(4)	1048(3)	-4153(3)	26(1)
C(6)	2929(4)	1090(2)	-3774(3)	22(1)
C(7)	3413(4)	1545(2)	-4211(3)	20(1)
C(8)	2596(3)	1919(2)	-5239(3)	20(1)
C(9)	1417(4)	1486(3)	-6093(3)	26(1)
C(10)	700(4)	1853(3)	-7051(3)	31(1)
C(11)	1126(4)	2650(3)	-7181(3)	32(1)
C(12)	2294(4)	3078(3)	-6350(3)	30(1)
C(13)	3034(4)	2713(3)	-5391(3)	25(1)
C(14)	5697(5)	1552(3)	-478(4)	37(1)
C(15)	7879(5)	405(3)	146(3)	40(1)
C(16)	7919(5)	2254(3)	-162(3)	41(1)
C(17)	5551(6)	-1388(3)	-2250(4)	41(1)
C(18)	7890(5)	-880(3)	-1907(4)	47(1)
C(19)	5264(6)	-695(3)	-3864(4)	49(1)
C(20)	9373(4)	1311(3)	-1192(3)	34(1)
C(21)	7413(5)	1229(3)	-3332(3)	37(1)
C(22)	7503(4)	2738(3)	-2354(3)	31(1)
Co(1)	5829(1)	964(1)	-2366(1)	18(1)
F(1)	4556(3)	-252(2)	-1467(2)	32(1)
F(2)	1928(3)	-312(2)	-2274(2)	38(1)
F(3)	-87(2)	563(2)	-4025(2)	40(1)
F(4)	514(2)	1508(2)	-5006(2)	38(1)
O(1)	4744(2)	1617(2)	-3612(2)	19(1)
P(1)	6772(1)	1277(1)	-794(1)	23(1)
P(2)	7543(1)	1516(1)	-2255(1)	22(1)



P(3)	6163(1)	-449(1)	-2536(1)	24(1)
------	---------	---------	----------	-------

---

**Table 3.** Bond lengths [Å] and angles [°] for **3**.

---

C(1)-C(2)	1.393(5)
C(1)-C(6)	1.444(5)
C(1)-Co(1)	1.932(4)
C(2)-F(1)	1.376(4)
C(2)-C(3)	1.378(6)
C(3)-F(2)	1.361(4)
C(3)-C(4)	1.387(6)
C(4)-C(5)	1.352(5)
C(4)-F(3)	1.363(4)
C(5)-F(4)	1.364(4)
C(5)-C(6)	1.413(5)
C(6)-C(7)	1.428(5)
C(7)-O(1)	1.293(4)
C(7)-C(8)	1.470(5)
C(8)-C(9)	1.401(5)
C(8)-C(13)	1.402(5)
C(9)-C(10)	1.386(6)
C(10)-C(11)	1.381(6)
C(11)-C(12)	1.378(6)
C(12)-C(13)	1.383(5)
C(14)-P(1)	1.832(4)
C(15)-P(1)	1.824(4)
C(16)-P(1)	1.830(4)
C(17)-P(3)	1.820(4)
C(18)-P(3)	1.823(5)
C(19)-P(3)	1.825(5)
C(20)-P(2)	1.824(4)
C(21)-P(2)	1.825(4)
C(22)-P(2)	1.817(4)
Co(1)-O(1)	1.907(2)
Co(1)-P(3)	2.1962(11)
Co(1)-P(1)	2.2026(11)
Co(1)-P(2)	2.2273(11)

C(2)-C(1)-C(6)	114.9(3)
C(2)-C(1)-Co(1)	133.1(3)
C(6)-C(1)-Co(1)	112.0(2)
F(1)-C(2)-C(3)	115.0(3)
F(1)-C(2)-C(1)	120.5(3)
C(3)-C(2)-C(1)	124.4(4)
F(2)-C(3)-C(2)	121.1(4)
F(2)-C(3)-C(4)	119.6(3)
C(2)-C(3)-C(4)	119.3(3)
C(5)-C(4)-F(3)	121.6(4)
C(5)-C(4)-C(3)	119.7(3)
F(3)-C(4)-C(3)	118.6(3)
C(4)-C(5)-F(4)	116.8(3)
C(4)-C(5)-C(6)	121.9(4)
F(4)-C(5)-C(6)	121.2(3)
C(5)-C(6)-C(7)	127.6(3)
C(5)-C(6)-C(1)	119.6(3)
C(7)-C(6)-C(1)	112.7(3)
O(1)-C(7)-C(6)	114.2(3)
O(1)-C(7)-C(8)	117.6(3)
C(6)-C(7)-C(8)	128.2(3)
C(9)-C(8)-C(13)	118.0(3)
C(9)-C(8)-C(7)	122.3(3)
C(13)-C(8)-C(7)	119.6(3)
C(10)-C(9)-C(8)	120.2(4)
C(11)-C(10)-C(9)	120.8(4)
C(12)-C(11)-C(10)	119.7(4)
C(11)-C(12)-C(13)	120.2(4)
C(12)-C(13)-C(8)	121.0(4)
O(1)-Co(1)-C(1)	82.35(13)
O(1)-Co(1)-P(3)	113.06(8)
C(1)-Co(1)-P(3)	91.45(11)
O(1)-Co(1)-P(1)	135.02(8)
C(1)-Co(1)-P(1)	94.65(11)
P(3)-Co(1)-P(1)	111.87(4)
O(1)-Co(1)-P(2)	81.58(8)

C(1)-Co(1)-P(2)	163.92(12)
P(3)-Co(1)-P(2)	95.47(4)
P(1)-Co(1)-P(2)	96.20(4)
C(7)-O(1)-Co(1)	118.1(2)
C(15)-P(1)-C(16)	100.1(2)
C(15)-P(1)-C(14)	100.5(2)
C(16)-P(1)-C(14)	96.1(2)
C(15)-P(1)-Co(1)	116.04(15)
C(16)-P(1)-Co(1)	119.09(15)
C(14)-P(1)-Co(1)	120.93(15)
C(22)-P(2)-C(20)	101.6(2)
C(22)-P(2)-C(21)	99.0(2)
C(20)-P(2)-C(21)	101.4(2)
C(22)-P(2)-Co(1)	113.18(13)
C(20)-P(2)-Co(1)	124.30(15)
C(21)-P(2)-Co(1)	113.79(15)
C(17)-P(3)-C(18)	97.8(2)
C(17)-P(3)-C(19)	99.6(2)
C(18)-P(3)-C(19)	101.7(2)
C(17)-P(3)-Co(1)	122.43(15)
C(18)-P(3)-Co(1)	121.33(16)
C(19)-P(3)-Co(1)	110.22(17)

---

Symmetry transformations used to generate equivalent atoms:

**Table 4.** Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **3**. The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [ h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12} ]$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
C(1)	28(2)	19(2)	25(2)	-3(1)	18(2)	-6(2)
C(2)	34(2)	17(2)	26(2)	-1(2)	21(2)	-3(2)
C(3)	42(2)	18(2)	39(2)	-7(2)	36(2)	-10(2)
C(4)	24(2)	24(2)	36(2)	-8(2)	22(2)	-7(2)
C(5)	22(2)	28(2)	27(2)	-2(2)	16(2)	-3(2)
C(6)	22(2)	18(2)	25(2)	-3(2)	15(2)	-2(1)
C(7)	19(2)	17(2)	21(2)	-2(1)	12(2)	0(1)
C(8)	15(2)	20(2)	23(2)	2(1)	12(2)	3(1)
C(9)	24(2)	28(2)	23(2)	-2(2)	14(2)	-1(2)
C(10)	22(2)	43(2)	22(2)	-4(2)	13(2)	-2(2)
C(11)	24(2)	47(3)	22(2)	10(2)	14(2)	10(2)
C(12)	31(2)	30(2)	31(2)	8(2)	21(2)	2(2)
C(13)	23(2)	27(2)	25(2)	3(2)	15(2)	0(2)
C(14)	38(2)	39(2)	42(3)	-13(2)	30(2)	-4(2)
C(15)	44(3)	41(3)	25(2)	5(2)	19(2)	12(2)
C(16)	37(2)	50(3)	30(2)	-14(2)	20(2)	-14(2)
C(17)	66(3)	23(2)	53(3)	-1(2)	48(3)	-3(2)
C(18)	58(3)	27(2)	69(4)	4(2)	47(3)	13(2)
C(19)	78(4)	34(2)	38(3)	-11(2)	40(3)	-15(2)
C(20)	27(2)	32(2)	39(3)	6(2)	20(2)	6(2)
C(21)	45(3)	43(3)	33(2)	3(2)	29(2)	-3(2)
C(22)	25(2)	26(2)	38(2)	5(2)	19(2)	1(2)
Co(1)	18(1)	16(1)	18(1)	2(1)	11(1)	1(1)
F(1)	45(1)	27(1)	30(1)	7(1)	27(1)	0(1)
F(2)	60(2)	29(1)	55(2)	-4(1)	51(2)	-11(1)
F(3)	32(1)	44(1)	60(2)	-10(1)	37(1)	-11(1)
F(4)	22(1)	46(2)	38(2)	11(1)	17(1)	6(1)
O(1)	17(1)	19(1)	19(1)	3(1)	10(1)	1(1)
P(1)	24(1)	23(1)	21(1)	-1(1)	14(1)	3(1)
P(2)	18(1)	21(1)	24(1)	4(1)	13(1)	2(1)
P(3)	36(1)	18(1)	25(1)	0(1)	22(1)	0(1)



**Table 5.** Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **3**.

	x	y	z	U(eq)
H(9A)	1115	950	-6016	31
H(10A)	-78	1558	-7614	37
H(11A)	627	2897	-7825	38
H(12A)	2585	3614	-6434	36
H(13A)	3836	2998	-4839	30
H(14A)	6276	1747	216	55
H(14B)	5186	1028	-569	55
H(14C)	5065	2026	-914	55
H(15A)	7362	-144	-48	60
H(15B)	8206	596	797	60
H(15C)	8654	303	185	60
H(16A)	8256	2299	516	61
H(16B)	7417	2792	-532	61
H(16C)	8684	2184	-140	61
H(17A)	6052	-1404	-1534	61
H(17B)	5694	-1942	-2454	61
H(17C)	4583	-1313	-2612	61
H(18A)	7828	-1499	-2094	71
H(18B)	8447	-838	-1185	71
H(18C)	8305	-531	-2110	71
H(19A)	4298	-563	-4271	73
H(19B)	5379	-1321	-3934	73
H(19C)	5643	-331	-4086	73
H(20A)	9918	1618	-1310	51
H(20B)	9555	676	-1130	51
H(20C)	9610	1533	-578	51
H(21A)	7589	597	-3315	56
H(21B)	8081	1571	-3297	56
H(21C)	6501	1369	-3952	56

Supplementary Material (ESI) for Dalton Transactions  
This journal is (c) The Royal Society of Chemistry 2008

H(22A)	7687	3001	-1773	47
H(22B)	6607	2926	-2959	47
H(22C)	8193	2931	-2382	47

---



**Table 6.** Torsion angles [°] for **3**.

---

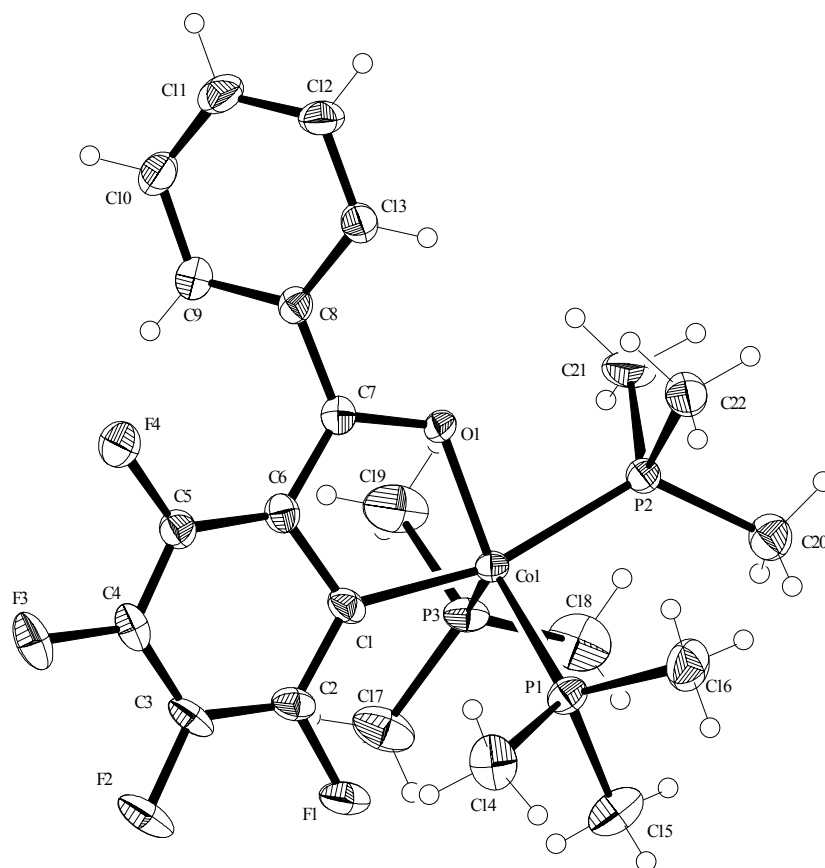
C(6)-C(1)-C(2)-F(1)	177.9(3)
Co(1)-C(1)-C(2)-F(1)	-4.2(6)
C(6)-C(1)-C(2)-C(3)	-4.7(5)
Co(1)-C(1)-C(2)-C(3)	173.1(3)
F(1)-C(2)-C(3)-F(2)	0.8(5)
C(1)-C(2)-C(3)-F(2)	-176.7(3)
F(1)-C(2)-C(3)-C(4)	179.2(3)
C(1)-C(2)-C(3)-C(4)	1.7(6)
F(2)-C(3)-C(4)-C(5)	178.5(3)
C(2)-C(3)-C(4)-C(5)	0.1(6)
F(2)-C(3)-C(4)-F(3)	1.9(5)
C(2)-C(3)-C(4)-F(3)	-176.5(3)
F(3)-C(4)-C(5)-F(4)	0.4(6)
C(3)-C(4)-C(5)-F(4)	-176.1(3)
F(3)-C(4)-C(5)-C(6)	178.0(3)
C(3)-C(4)-C(5)-C(6)	1.5(6)
C(4)-C(5)-C(6)-C(7)	179.0(4)
F(4)-C(5)-C(6)-C(7)	-3.4(6)
C(4)-C(5)-C(6)-C(1)	-4.8(6)
F(4)-C(5)-C(6)-C(1)	172.7(3)
C(2)-C(1)-C(6)-C(5)	6.1(5)
Co(1)-C(1)-C(6)-C(5)	-172.2(3)
C(2)-C(1)-C(6)-C(7)	-177.2(3)
Co(1)-C(1)-C(6)-C(7)	4.5(4)
C(5)-C(6)-C(7)-O(1)	167.5(4)
C(1)-C(6)-C(7)-O(1)	-8.9(4)
C(5)-C(6)-C(7)-C(8)	-13.5(6)
C(1)-C(6)-C(7)-C(8)	170.1(3)
O(1)-C(7)-C(8)-C(9)	140.8(3)
C(6)-C(7)-C(8)-C(9)	-38.2(5)
O(1)-C(7)-C(8)-C(13)	-36.0(5)
C(6)-C(7)-C(8)-C(13)	145.0(4)
C(13)-C(8)-C(9)-C(10)	-1.5(5)

C(7)-C(8)-C(9)-C(10)	-178.3(3)
C(8)-C(9)-C(10)-C(11)	-0.4(6)
C(9)-C(10)-C(11)-C(12)	1.3(6)
C(10)-C(11)-C(12)-C(13)	-0.1(6)
C(11)-C(12)-C(13)-C(8)	-1.8(6)
C(9)-C(8)-C(13)-C(12)	2.6(5)
C(7)-C(8)-C(13)-C(12)	179.5(3)
C(2)-C(1)-Co(1)-O(1)	-177.7(4)
C(6)-C(1)-Co(1)-O(1)	0.2(2)
C(2)-C(1)-Co(1)-P(3)	69.2(4)
C(6)-C(1)-Co(1)-P(3)	-112.9(2)
C(2)-C(1)-Co(1)-P(1)	-42.9(4)
C(6)-C(1)-Co(1)-P(1)	135.0(2)
C(2)-C(1)-Co(1)-P(2)	-175.2(2)
C(6)-C(1)-Co(1)-P(2)	2.7(6)
C(6)-C(7)-O(1)-Co(1)	9.5(4)
C(8)-C(7)-O(1)-Co(1)	-169.6(2)
C(1)-Co(1)-O(1)-C(7)	-5.5(3)
P(3)-Co(1)-O(1)-C(7)	82.8(2)
P(1)-Co(1)-O(1)-C(7)	-94.5(3)
P(2)-Co(1)-O(1)-C(7)	175.2(2)
O(1)-Co(1)-P(1)-C(15)	-173.9(2)
C(1)-Co(1)-P(1)-C(15)	102.3(2)
P(3)-Co(1)-P(1)-C(15)	8.85(19)
P(2)-Co(1)-P(1)-C(15)	-89.60(19)
O(1)-Co(1)-P(1)-C(16)	-54.2(2)
C(1)-Co(1)-P(1)-C(16)	-138.0(2)
P(3)-Co(1)-P(1)-C(16)	128.52(19)
P(2)-Co(1)-P(1)-C(16)	30.07(19)
O(1)-Co(1)-P(1)-C(14)	64.2(2)
C(1)-Co(1)-P(1)-C(14)	-19.7(2)
P(3)-Co(1)-P(1)-C(14)	-113.11(18)
P(2)-Co(1)-P(1)-C(14)	148.44(18)
O(1)-Co(1)-P(2)-C(22)	55.78(18)
C(1)-Co(1)-P(2)-C(22)	53.3(4)
P(3)-Co(1)-P(2)-C(22)	168.33(16)

P(1)-Co(1)-P(2)-C(22)	-78.91(16)
O(1)-Co(1)-P(2)-C(20)	179.5(2)
C(1)-Co(1)-P(2)-C(20)	177.0(4)
P(3)-Co(1)-P(2)-C(20)	-67.90(19)
P(1)-Co(1)-P(2)-C(20)	44.86(18)
O(1)-Co(1)-P(2)-C(21)	-56.25(18)
C(1)-Co(1)-P(2)-C(21)	-58.8(4)
P(3)-Co(1)-P(2)-C(21)	56.30(17)
P(1)-Co(1)-P(2)-C(21)	169.06(17)
O(1)-Co(1)-P(3)-C(17)	-118.2(2)
C(1)-Co(1)-P(3)-C(17)	-35.9(2)
P(1)-Co(1)-P(3)-C(17)	59.7(2)
P(2)-Co(1)-P(3)-C(17)	158.6(2)
O(1)-Co(1)-P(3)-C(18)	116.6(2)
C(1)-Co(1)-P(3)-C(18)	-161.1(2)
P(1)-Co(1)-P(3)-C(18)	-65.5(2)
P(2)-Co(1)-P(3)-C(18)	33.5(2)
O(1)-Co(1)-P(3)-C(19)	-1.8(2)
C(1)-Co(1)-P(3)-C(19)	80.5(2)
P(1)-Co(1)-P(3)-C(19)	176.10(19)
P(2)-Co(1)-P(3)-C(19)	-84.97(19)

---

Symmetry transformations used to generate equivalent atoms:



Molecular structure of **3**.

**Table 1.** Crystal data and structure refinement for **4**.

Identification code	4	
Empirical formula	C <sub>22</sub> H <sub>36</sub> Co F O P <sub>3</sub>	
Formula weight	487.35	
Temperature	150(2) K	
Wavelength	0.71073 Å	
Crystal system	Monoclinic	
Space group	P2 <sub>1</sub> /n	
Unit cell dimensions	a = 9.7220(12) Å	α = 90.000(12)°.
	b = 17.1350(13) Å	β = 101.030(12)°.
	c = 15.4390(11) Å	γ = 90.000(12)°.
Volume	2524.4(4) Å <sup>3</sup>	
Z	4	
Density (calculated)	1.282 Mg/m <sup>3</sup>	
Absorption coefficient	0.887 mm <sup>-1</sup>	
F(000)	1028	
Crystal size	0.12 x 0.10 x 0.08 mm <sup>3</sup>	
Theta range for data collection	4.16 to 26.37°.	
Index ranges	-12 ≤ h ≤ 12, -21 ≤ k ≤ 21, -19 ≤ l ≤ 19	
Reflections collected	30970	
Independent reflections	5112 [R(int) = 0.1130]	
Completeness to theta = 26.37°	99.0 %	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.9324 and 0.9010	
Refinement method	Full-matrix least-squares on F <sup>2</sup>	
Data / restraints / parameters	5112 / 0 / 253	
Goodness-of-fit on F <sup>2</sup>	1.048	
Final R indices [I > 2σ(I)]	R1 = 0.0450, wR2 = 0.1153	
R indices (all data)	R1 = 0.0522, wR2 = 0.1191	
Largest diff. peak and hole	0.717 and -1.107 e.Å <sup>-3</sup>	

**Table 2.** Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **4**.  $U(\text{eq})$  is defined as one third of the trace of the orthogonalized  $U^{ij}$  tensor.

	x	y	z	$U(\text{eq})$
C(1)	6744(2)	2556(1)	4360(2)	20(1)
C(2)	7325(2)	3323(1)	4107(2)	22(1)
C(3)	7365(3)	3577(1)	3252(2)	26(1)
C(4)	7842(3)	4305(2)	3060(2)	37(1)
C(5)	8322(3)	4818(2)	3749(2)	43(1)
C(6)	8305(3)	4595(2)	4612(2)	40(1)
C(7)	7816(3)	3860(2)	4784(2)	30(1)
C(8)	7479(2)	1807(1)	4212(2)	22(1)
C(9)	8181(2)	1674(1)	3514(2)	26(1)
C(10)	8777(3)	950(2)	3400(2)	32(1)
C(11)	8749(3)	352(2)	3998(2)	38(1)
C(12)	8106(3)	485(2)	4714(2)	40(1)
C(13)	7466(3)	1194(2)	4820(2)	31(1)
C(14)	4211(3)	4472(2)	3960(2)	40(1)
C(15)	3604(3)	3703(2)	2315(2)	44(1)
C(16)	1692(3)	3644(2)	3458(2)	39(1)
C(17)	3356(3)	3155(2)	5948(2)	40(1)
C(18)	4153(3)	1579(2)	6033(2)	40(1)
C(19)	1494(3)	2009(2)	5079(2)	41(1)
C(20)	4727(3)	1550(2)	2213(2)	46(1)
C(21)	4200(5)	571(2)	3567(3)	61(1)
C(22)	2077(3)	1565(2)	2656(2)	53(1)
Co(1)	4637(1)	2480(1)	4214(1)	17(1)
F(1)	6844(2)	3098(1)	2548(1)	35(1)
O(1)	6248(2)	2593(1)	5113(1)	23(1)
P(1)	3934(1)	1588(1)	3196(1)	26(1)
P(2)	3332(1)	2302(1)	5230(1)	23(1)
P(3)	3582(1)	3525(1)	3493(1)	26(1)

**Table 3.** Bond lengths [Å] and angles [°] for **4**.

---

C(1)-O(1)	1.343(3)
C(1)-C(8)	1.507(3)
C(1)-C(2)	1.511(3)
C(1)-Co(1)	2.022(2)
C(2)-C(3)	1.397(3)
C(2)-C(7)	1.406(3)
C(3)-F(1)	1.378(3)
C(3)-C(4)	1.382(3)
C(4)-C(5)	1.390(4)
C(5)-C(6)	1.390(5)
C(6)-C(7)	1.389(4)
C(8)-C(9)	1.401(3)
C(8)-C(13)	1.410(3)
C(9)-C(10)	1.394(3)
C(10)-C(11)	1.383(4)
C(11)-C(12)	1.389(4)
C(12)-C(13)	1.388(4)
C(14)-P(3)	1.834(3)
C(15)-P(3)	1.849(3)
C(16)-P(3)	1.839(3)
C(17)-P(2)	1.832(3)
C(18)-P(2)	1.825(3)
C(19)-P(2)	1.828(3)
C(20)-P(1)	1.831(3)
C(21)-P(1)	1.838(3)
C(22)-P(1)	1.838(3)
Co(1)-O(1)	1.8938(17)
Co(1)-P(1)	2.2043(7)
Co(1)-P(2)	2.2189(7)
Co(1)-P(3)	2.2492(7)
O(1)-C(1)-C(8)	115.47(19)
O(1)-C(1)-C(2)	113.38(18)
C(8)-C(1)-C(2)	119.35(19)

O(1)-C(1)-Co(1)	64.87(12)
C(8)-C(1)-Co(1)	115.33(15)
C(2)-C(1)-Co(1)	116.63(15)
C(3)-C(2)-C(7)	115.5(2)
C(3)-C(2)-C(1)	126.5(2)
C(7)-C(2)-C(1)	117.9(2)
F(1)-C(3)-C(4)	117.1(2)
F(1)-C(3)-C(2)	119.1(2)
C(4)-C(3)-C(2)	123.7(2)
C(3)-C(4)-C(5)	119.0(3)
C(6)-C(5)-C(4)	119.7(2)
C(7)-C(6)-C(5)	119.9(3)
C(6)-C(7)-C(2)	122.1(3)
C(9)-C(8)-C(13)	117.2(2)
C(9)-C(8)-C(1)	124.7(2)
C(13)-C(8)-C(1)	118.0(2)
C(10)-C(9)-C(8)	121.1(2)
C(11)-C(10)-C(9)	120.9(3)
C(10)-C(11)-C(12)	118.6(2)
C(13)-C(12)-C(11)	121.1(3)
C(12)-C(13)-C(8)	120.9(2)
O(1)-Co(1)-C(1)	39.93(8)
O(1)-Co(1)-P(1)	133.91(5)
C(1)-Co(1)-P(1)	107.07(7)
O(1)-Co(1)-P(2)	90.04(5)
C(1)-Co(1)-P(2)	129.36(7)
P(1)-Co(1)-P(2)	105.51(3)
O(1)-Co(1)-P(3)	121.06(5)
C(1)-Co(1)-P(3)	111.11(7)
P(1)-Co(1)-P(3)	98.60(3)
P(2)-Co(1)-P(3)	100.99(3)
C(1)-O(1)-Co(1)	75.20(13)
C(20)-P(1)-C(21)	99.70(18)
C(20)-P(1)-C(22)	98.99(16)
C(21)-P(1)-C(22)	100.90(18)
C(20)-P(1)-Co(1)	119.53(10)



Supplementary Material (ESI) for Dalton Transactions  
This journal is (c) The Royal Society of Chemistry 2008

C(21)-P(1)-Co(1)	115.43(11)
C(22)-P(1)-Co(1)	118.78(10)
C(18)-P(2)-C(19)	100.95(14)
C(18)-P(2)-C(17)	100.20(15)
C(19)-P(2)-C(17)	101.23(14)
C(18)-P(2)-Co(1)	109.79(9)
C(19)-P(2)-Co(1)	128.69(10)
C(17)-P(2)-Co(1)	112.07(10)
C(14)-P(3)-C(16)	99.78(14)
C(14)-P(3)-C(15)	100.03(16)
C(16)-P(3)-C(15)	98.80(14)
C(14)-P(3)-Co(1)	115.10(10)
C(16)-P(3)-Co(1)	117.47(10)
C(15)-P(3)-Co(1)	121.87(10)

---

Symmetry transformations used to generate equivalent atoms:

**Table 4.** Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for **4**. The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [ h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12} ]$

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
C(1)	19(1)	23(1)	18(1)	0(1)	1(1)	-2(1)
C(2)	19(1)	23(1)	25(1)	1(1)	6(1)	-1(1)
C(3)	26(1)	26(1)	29(1)	-1(1)	11(1)	3(1)
C(4)	43(2)	28(1)	48(2)	10(1)	28(1)	7(1)
C(5)	42(2)	21(1)	75(2)	3(1)	32(2)	-5(1)
C(6)	32(2)	31(1)	60(2)	-15(1)	16(1)	-10(1)
C(7)	25(1)	32(1)	32(1)	-6(1)	5(1)	-7(1)
C(8)	16(1)	23(1)	24(1)	1(1)	-2(1)	-1(1)
C(9)	20(1)	26(1)	31(1)	-1(1)	3(1)	0(1)
C(10)	23(1)	32(1)	40(2)	-8(1)	4(1)	3(1)
C(11)	28(1)	26(1)	55(2)	-6(1)	-3(1)	7(1)
C(12)	36(2)	29(1)	52(2)	12(1)	3(1)	6(1)
C(13)	29(1)	32(1)	32(1)	8(1)	5(1)	2(1)
C(14)	40(2)	24(1)	56(2)	1(1)	9(1)	5(1)
C(15)	44(2)	56(2)	32(2)	16(1)	7(1)	21(1)
C(16)	32(2)	42(2)	40(2)	-1(1)	3(1)	12(1)
C(17)	48(2)	43(2)	31(1)	-8(1)	14(1)	-3(1)
C(18)	42(2)	44(2)	37(2)	15(1)	17(1)	4(1)
C(19)	27(2)	57(2)	41(2)	-2(1)	11(1)	-9(1)
C(20)	37(2)	66(2)	35(2)	-27(1)	7(1)	0(1)
C(21)	87(3)	26(1)	62(2)	-10(1)	-5(2)	-12(2)
C(22)	24(2)	69(2)	62(2)	-35(2)	-1(1)	-6(1)
Co(1)	18(1)	18(1)	16(1)	-1(1)	2(1)	-1(1)
F(1)	45(1)	37(1)	26(1)	3(1)	9(1)	5(1)
O(1)	21(1)	31(1)	15(1)	-1(1)	1(1)	-5(1)
P(1)	22(1)	28(1)	28(1)	-11(1)	1(1)	-3(1)
P(2)	23(1)	26(1)	22(1)	-1(1)	7(1)	-2(1)
P(3)	26(1)	26(1)	24(1)	4(1)	3(1)	7(1)

**Table 5.** Hydrogen coordinates ( $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ )  
for **4**.

	x	y	z	U(eq)
H(5A)	7840	4449	2479	44
H(11A)	8655	5309	3632	52
H(7A)	8620	4938	5075	48
H(6A)	7813	3719	5365	36
H(2A)	8251	2076	3119	31
H(4A)	9199	868	2916	38
H(9A)	9152	-129	3922	45
H(12A)	8103	92	5130	48
H(10A)	7024	1265	5297	37
H(15A)	5194	4521	3956	48
H(15B)	4061	4509	4555	48
H(15C)	3709	4883	3612	48
H(18C)	3102	4175	2128	53
H(18B)	3166	3273	1969	53
H(18A)	4556	3753	2236	53
H(13A)	1521	3684	4049	46
H(13B)	1199	3200	3172	46
H(13C)	1368	4109	3136	46
H(21A)	3002	3601	5599	48
H(21B)	4300	3255	6246	48
H(21C)	2779	3055	6376	48
H(14A)	5066	1758	6311	48
H(14B)	4238	1093	5738	48
H(14C)	3586	1505	6471	48
H(22C)	1219	1982	5644	50
H(22B)	1376	1507	4800	50
H(22A)	921	2385	4715	50
H(16A)	5718	1466	2387	55
H(16B)	4562	2035	1898	55

Supplementary Material (ESI) for Dalton Transactions  
This journal is (c) The Royal Society of Chemistry 2008

H(16C)	4318	1130	1839	55
H(19A)	3696	476	4034	73
H(19B)	5181	479	3779	73
H(19C)	3863	226	3083	73
H(17C)	1927	1150	2232	64
H(17B)	1827	2054	2363	64
H(17A)	1508	1481	3092	64

---

**Table 6.** Torsion angles [°] for **4**.

---

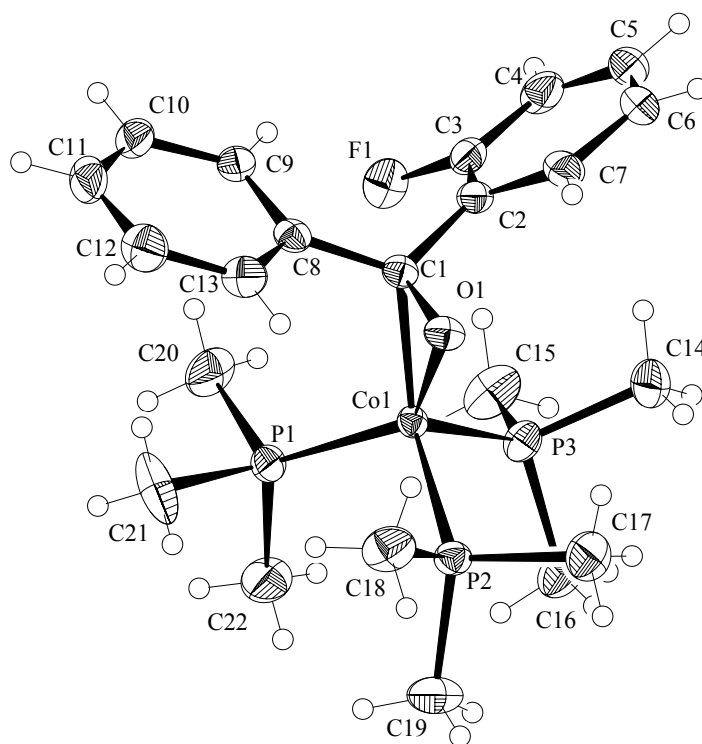
O(1)-C(1)-C(2)-C(3)	-151.2(2)
C(8)-C(1)-C(2)-C(3)	67.5(3)
Co(1)-C(1)-C(2)-C(3)	-78.7(3)
O(1)-C(1)-C(2)-C(7)	25.4(3)
C(8)-C(1)-C(2)-C(7)	-115.9(2)
Co(1)-C(1)-C(2)-C(7)	97.9(2)
C(7)-C(2)-C(3)-F(1)	-177.1(2)
C(1)-C(2)-C(3)-F(1)	-0.5(4)
C(7)-C(2)-C(3)-C(4)	-0.3(4)
C(1)-C(2)-C(3)-C(4)	176.3(2)
F(1)-C(3)-C(4)-C(5)	177.4(2)
C(2)-C(3)-C(4)-C(5)	0.6(4)
C(3)-C(4)-C(5)-C(6)	-0.6(4)
C(4)-C(5)-C(6)-C(7)	0.4(4)
C(5)-C(6)-C(7)-C(2)	-0.2(4)
C(3)-C(2)-C(7)-C(6)	0.2(4)
C(1)-C(2)-C(7)-C(6)	-176.8(2)
O(1)-C(1)-C(8)-C(9)	-174.0(2)
C(2)-C(1)-C(8)-C(9)	-33.4(3)
Co(1)-C(1)-C(8)-C(9)	113.2(2)
O(1)-C(1)-C(8)-C(13)	5.9(3)
C(2)-C(1)-C(8)-C(13)	146.5(2)
Co(1)-C(1)-C(8)-C(13)	-66.9(3)
C(13)-C(8)-C(9)-C(10)	3.3(3)
C(1)-C(8)-C(9)-C(10)	-176.8(2)
C(8)-C(9)-C(10)-C(11)	-3.0(4)
C(9)-C(10)-C(11)-C(12)	0.4(4)
C(10)-C(11)-C(12)-C(13)	1.8(4)
C(11)-C(12)-C(13)-C(8)	-1.5(4)
C(9)-C(8)-C(13)-C(12)	-1.1(4)
C(1)-C(8)-C(13)-C(12)	179.0(2)
C(8)-C(1)-Co(1)-O(1)	107.7(2)
C(2)-C(1)-Co(1)-O(1)	-104.8(2)

O(1)-C(1)-Co(1)-P(1)	-139.76(10)
C(8)-C(1)-Co(1)-P(1)	-32.09(18)
C(2)-C(1)-Co(1)-P(1)	115.46(16)
O(1)-C(1)-Co(1)-P(2)	-11.90(14)
C(8)-C(1)-Co(1)-P(2)	95.77(17)
C(2)-C(1)-Co(1)-P(2)	-116.68(15)
O(1)-C(1)-Co(1)-P(3)	113.61(10)
C(8)-C(1)-Co(1)-P(3)	-138.72(15)
C(2)-C(1)-Co(1)-P(3)	8.83(18)
C(8)-C(1)-O(1)-Co(1)	-107.45(18)
C(2)-C(1)-O(1)-Co(1)	109.66(18)
P(1)-Co(1)-O(1)-C(1)	59.00(13)
P(2)-Co(1)-O(1)-C(1)	170.83(11)
P(3)-Co(1)-O(1)-C(1)	-86.29(12)
O(1)-Co(1)-P(1)-C(20)	-76.02(15)
C(1)-Co(1)-P(1)-C(20)	-40.89(15)
P(2)-Co(1)-P(1)-C(20)	178.42(13)
P(3)-Co(1)-P(1)-C(20)	74.42(13)
O(1)-Co(1)-P(1)-C(21)	42.90(18)
C(1)-Co(1)-P(1)-C(21)	78.04(17)
P(2)-Co(1)-P(1)-C(21)	-62.65(16)
P(3)-Co(1)-P(1)-C(21)	-166.66(16)
O(1)-Co(1)-P(1)-C(22)	162.91(17)
C(1)-Co(1)-P(1)-C(22)	-161.95(17)
P(2)-Co(1)-P(1)-C(22)	57.36(15)
P(3)-Co(1)-P(1)-C(22)	-46.65(15)
O(1)-Co(1)-P(2)-C(18)	-48.67(12)
C(1)-Co(1)-P(2)-C(18)	-41.06(14)
P(1)-Co(1)-P(2)-C(18)	87.38(12)
P(3)-Co(1)-P(2)-C(18)	-170.39(11)
O(1)-Co(1)-P(2)-C(19)	-171.81(15)
C(1)-Co(1)-P(2)-C(19)	-164.20(16)
P(1)-Co(1)-P(2)-C(19)	-35.76(14)
P(3)-Co(1)-P(2)-C(19)	66.47(14)
O(1)-Co(1)-P(2)-C(17)	61.75(12)
C(1)-Co(1)-P(2)-C(17)	69.36(14)

P(1)-Co(1)-P(2)-C(17)	-162.20(11)
P(3)-Co(1)-P(2)-C(17)	-59.97(12)
O(1)-Co(1)-P(3)-C(14)	-9.60(13)
C(1)-Co(1)-P(3)-C(14)	-52.96(13)
P(1)-Co(1)-P(3)-C(14)	-165.08(11)
P(2)-Co(1)-P(3)-C(14)	87.17(12)
O(1)-Co(1)-P(3)-C(16)	-126.66(12)
C(1)-Co(1)-P(3)-C(16)	-170.02(13)
P(1)-Co(1)-P(3)-C(16)	77.86(11)
P(2)-Co(1)-P(3)-C(16)	-29.90(11)
O(1)-Co(1)-P(3)-C(15)	111.59(15)
C(1)-Co(1)-P(3)-C(15)	68.23(16)
P(1)-Co(1)-P(3)-C(15)	-43.89(14)
P(2)-Co(1)-P(3)-C(15)	-151.64(14)

---

Symmetry transformations used to generate equivalent atoms:



Molecular structure of **4**.