

Extended Structures Containing Pt(II)-Tl(I) Bonds. Effect of these interactions on the Luminescence of Cyclometalated Pt(II) Compounds.

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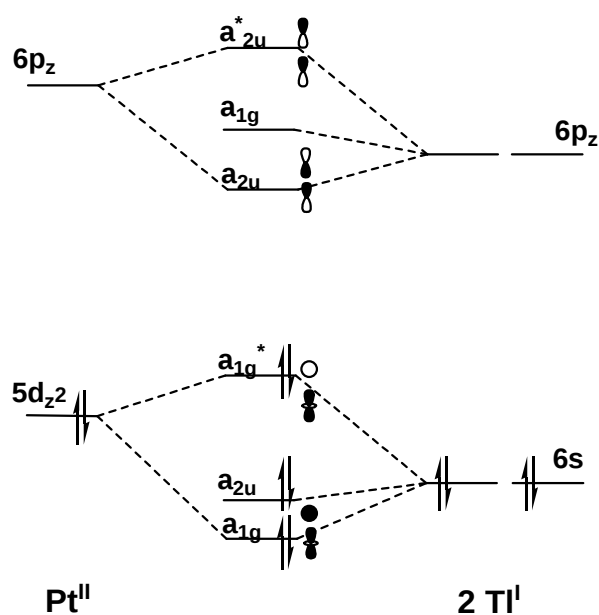


Chart 1: qualitative molecular orbital diagram for the Tl-Pt-Tl unit, proposed by Balch et al. for compound $\text{Tl}_2\text{Pt}(\text{CN})_4$. Ref: Nagle, J. K.; Balch, A. L.; Olmstead, M. M. *J. Am. Chem. Soc.* **1988**, *110*, 319.

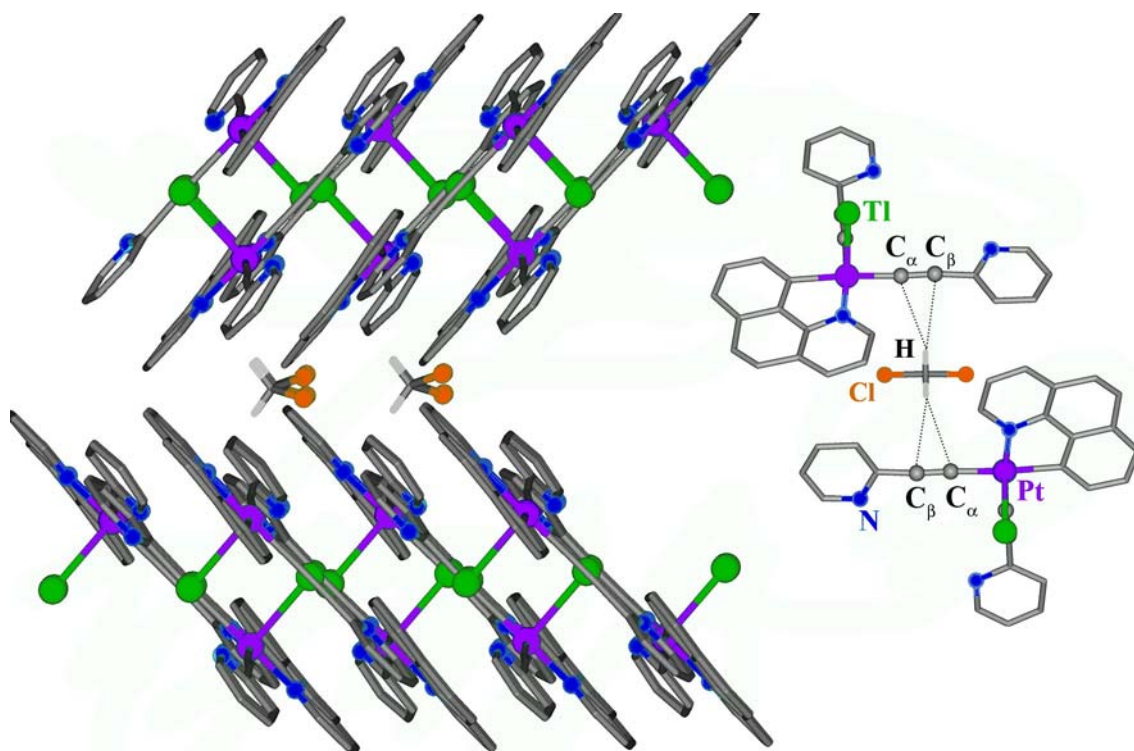


Figure S1: Supramolecular structure of the solvate $2 \text{CH}_2\text{Cl}_2$ showing how each CH_2Cl_2 molecule is connecting two $[\text{PtTl}(\text{bzq})(\text{C}\equiv\text{C}-\text{C}_5\text{H}_4\text{N}-2)_2]_2$ motifs of different layers through weak $\text{C}-\text{H}\cdots\text{C}\equiv\text{C}$ interactions ($\text{CH}\cdots\text{C}_\alpha$ 2.8747; $\text{CH}\cdots\text{C}_\beta$ 2.8234 Å).

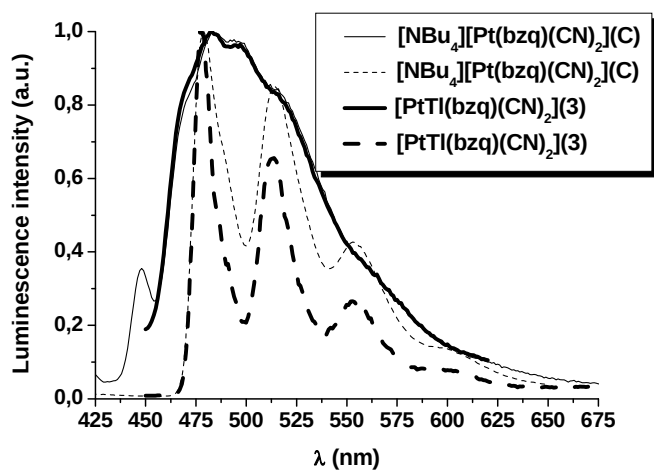


Figure S2. Normalized emission spectra in methanol (10^{-4} M) at 298K (—) and 77K (--) of **C** and **3**.

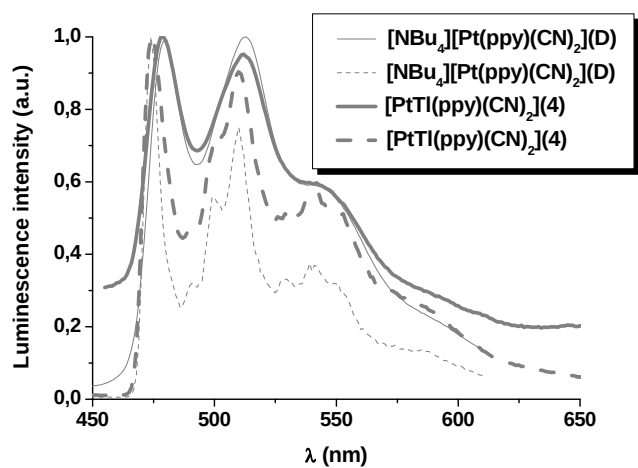


Figure S3. Normalized emission spectra in methanol (10^{-4} M) at 298K (—) and 77K (--) of **D** and **4**.

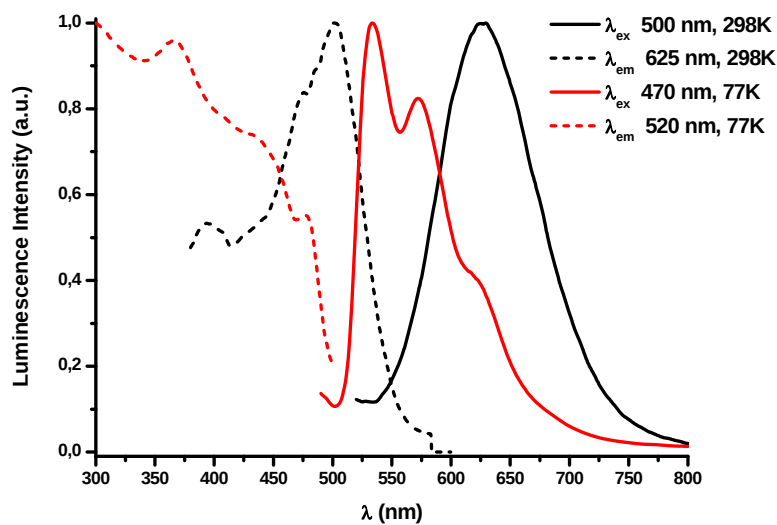


Figure S4. Normalized excitation and emission spectra in solid state of compound **1**.