

Supporting Information

Near-infrared Luminescent Xerogel Materials Covalently Bonded with Ternary Lanthanide [Er(III), Nd(III), Yb(III), Sm(III)] Complexes

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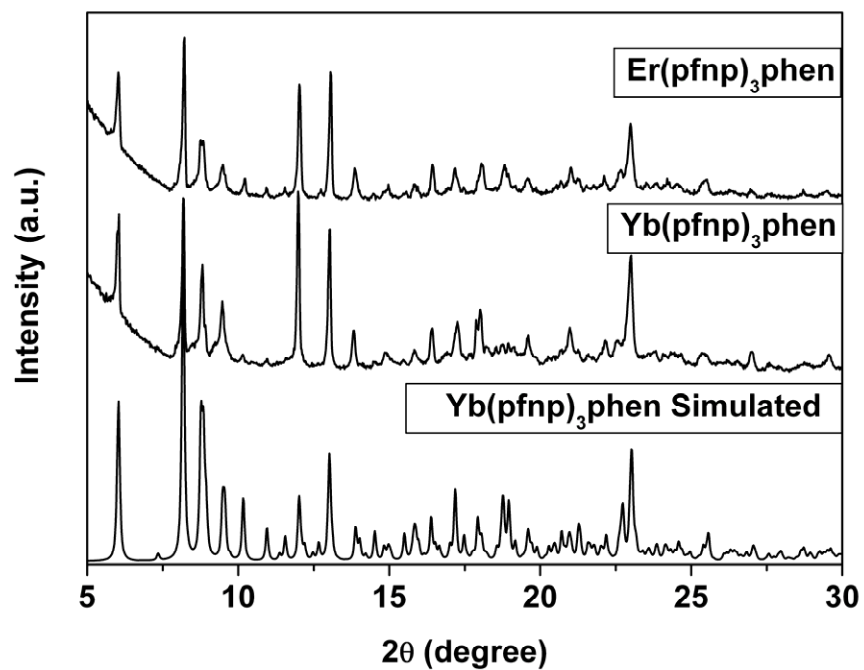


Fig. S1 XRD patterns of $\text{Er(pfnp)}_3\text{phen}$, $\text{Yb(pfnp)}_3\text{phen}$ and $\text{Yb(pfnp)}_3\text{phen}$ simulated.

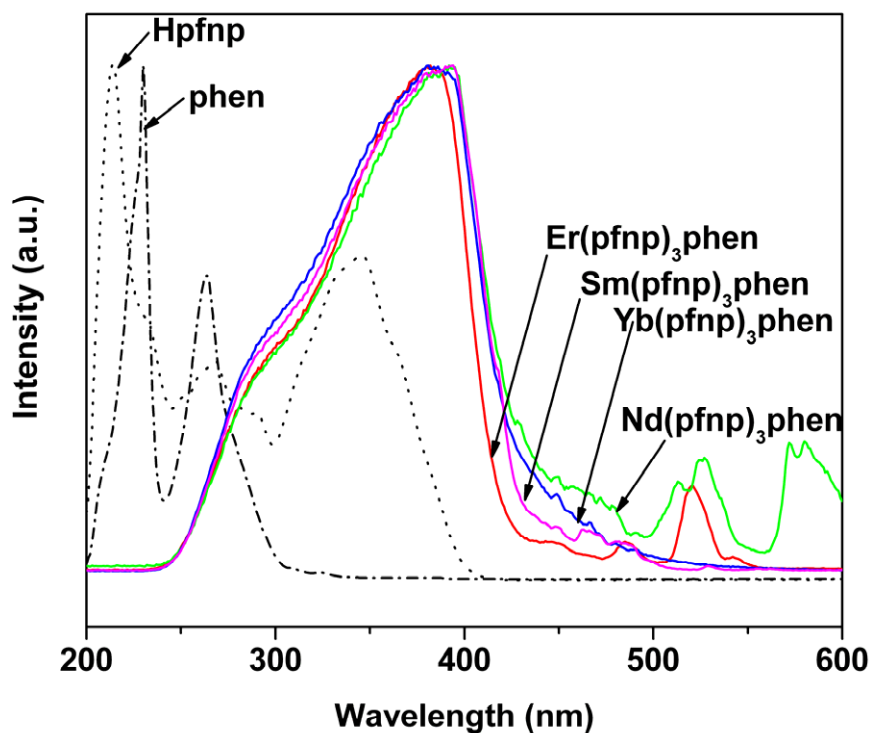


Fig. S2 UV-vis absorption spectra of Hpfnp and phen ligands and excitation spectra of Er(pfnp)₃phen ($\lambda_{em} = 1533$ nm), Nd(pfnp)₃phen ($\lambda_{em} = 1058$ nm), Yb(pfnp)₃phen ($\lambda_{em} = 978$ nm) and Sm(pfnp)₃phen ($\lambda_{em} = 950$ nm) complexes. The ligands are at 5×10^{-4} M in ethanol, and the complexes are in solid state.

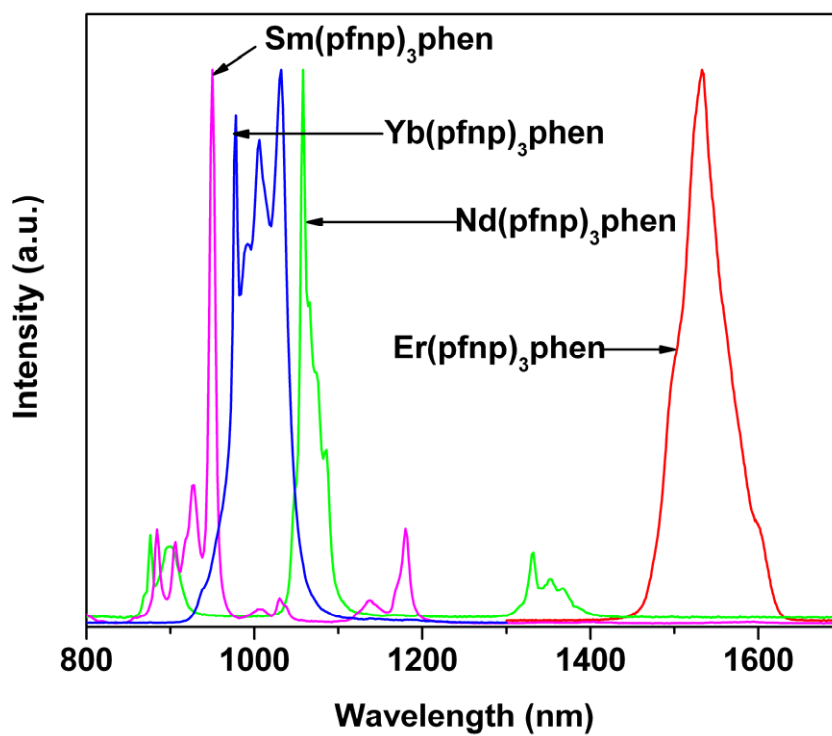


Fig. S3 Emission spectra of Er(pfnp)₃phen ($\lambda_{\text{ex}} = 380$ nm), Nd(pfnp)₃phen ($\lambda_{\text{ex}} = 390$ nm), Yb(pfnp)₃phen ($\lambda_{\text{ex}} = 380$ nm) and Sm(pfnp)₃phen ($\lambda_{\text{ex}} = 385$ nm) complexes.

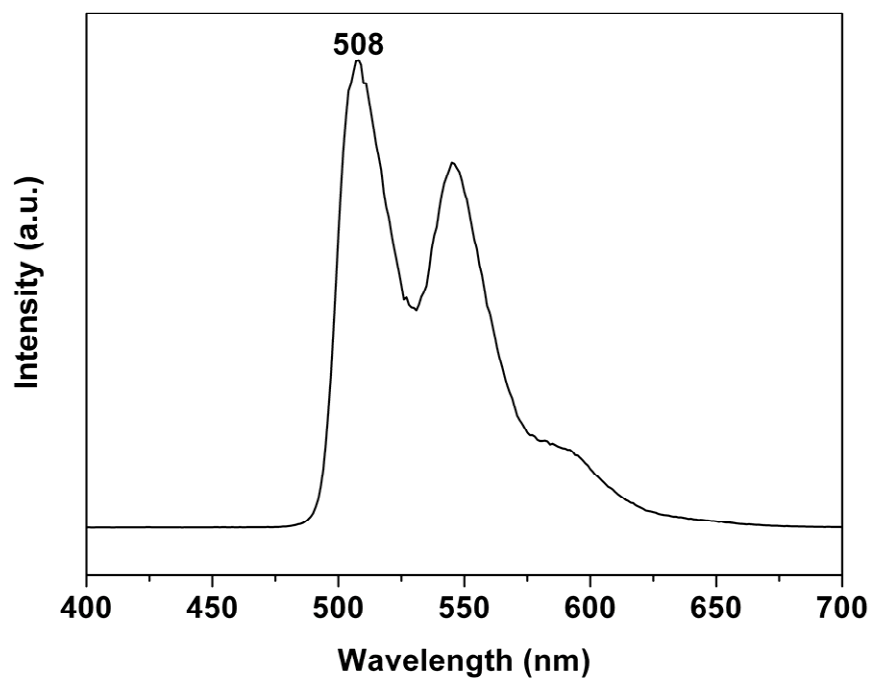


Fig. S4 The phosphorescence spectra of Gd(pfnp)₃(H₂O)₂ complex ($\lambda_{\text{ex}} = 342$ nm), was measured at 77 K after a delay time of 1 s.

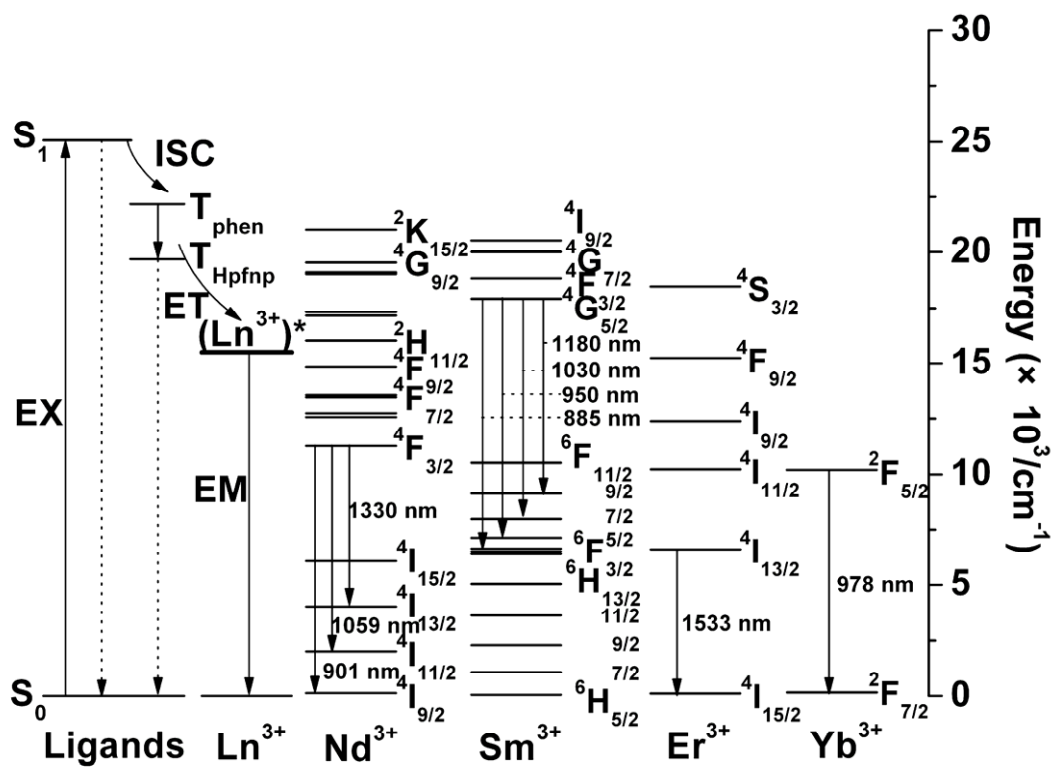


Fig. S5 Energy levels and the energy transfer processes in the Ln-ligand system (Ln = Er, Nd, Yb, Sm). (EX: excitation; EM: emission; ISC: intersystem crossing; ET: energy transfer.)