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#### **Supporting Information for**

# Synthesis, Structural Characterization and Solution Chemistry of Ruthenium(III) Triazole-Thiadiazine Complexes

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**Figure S1.** FAR-IR spectrum of Na[RuCl<sub>4</sub>(L1)(dmso)] (1), CsI disk (above); calculated IR spectra b3lyp/lanl2dz(Ru)-6-31G(C H Cl N O S) (below).<sup>S1</sup> The discrepancy between the Ru-S stretching frequency is due to the difference between the experimental geometry (d(Ru-S)=2.310(3)Å) and the optimized geometry (d(Ru-S)=2.531 Å). According to this, in the optimized geometry the Ru-S bond is weakened with respect to the X-ray structure. The discrepancy between the calculated and experimental geometry for the other coordination distances are less than 0.15 Å.

**Table S1.** Observed frequencies  $(cm^{-1})$  for the Ru-D (D = donor atom) in Ru(III) complexes.

|                                       | v(Ru-N) | v(Ru-S) | v(Ru-Cl) |
|---------------------------------------|---------|---------|----------|
| $[(DMSO)_2H][trans-RuCl_4(DMSO)_2]^a$ | -       | 415     | 345, 329 |
| $Na[RuCl_4(DMSO)(L1)]$ (1)            | 458     | 436     | 364, 343 |
| $Na[RuCl_4(DMSO)(L2)]$ (2)            | 448     | 431     | 356, 332 |
| -), $         -$                      |         |         |          |

a): ref.: S2



Figure S2. Negative ESI-MS spectrum of Na[*trans*-RuCl<sub>4</sub>(dmso)(L1)] (1) dissolved in acetonitrile.





Figure S3. Negative ESI-MS spectrum of Na[*trans*-RuCl<sub>4</sub>(dmso)(L2)] (2) dissolved in acetonitrile.



**Figure S4.** Stack plot of the <sup>1</sup>H NMR spectra of **1** in D<sub>2</sub>O recorded in the 0-180 min interval. \* = coordinated ligand signals. # = water and acetone.



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**Figure S5.** Stack plot of the <sup>1</sup>H NMR spectra of **1** in D<sub>2</sub>O recorded in the 0-180 min interval, together with the free ligand spectrum in D<sub>2</sub>O. \* = coordinated ligand signals, \$ = free ligand signals, \$ = free DMSO signal (2.71 ppm, ref. S3).



**Figure S6.** Stack plot of the <sup>1</sup>H NMR spectra of **2** in D<sub>2</sub>O recorded in the 0-180 min interval. \* = coordinated ligand signals. # = water, acetone and 3-(trimethylsilyl)propane sulfonate as reference.



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**Figure S7.** Stack plot of the <sup>1</sup>H NMR spectra of **2** in D<sub>2</sub>O recorded in the 0-180 min interval, together with the free ligand spectrum in D<sub>2</sub>O. \* = coordinated ligand signals, § = free ligand signals, £ = free DMSO signal (2.71 ppm, ref. S3).



**Figure S8.** Upper diagram: UV-Vis spectra of Na[RuCl<sub>4</sub>(dmso)(L2)] (**2**) in water at 7.5 min interval (0-170 min after dissolution). Lower diagram: UV-Vis spectra of **2** in water at 180 min interval (540-3000 min after dissolution).



**Figure S9.** Experimental (dots) and calculated (lines) absorbances for the hydrolytic reaction of Na[RuCl<sub>4</sub>(dmso)(L2)] (2) in water (0-170 min after dissolution) at 342 and 398 nm.



**Figure S10.** Upper diagram: UV-Vis spectra of Na[RuCl<sub>4</sub>(dmso)(L1)] (1) in buffered water (pH = 7.4) at 60 sec interval (0-30 min after dissolution). Lower diagram: UV-Vis spectra of 1 in buffered water (pH = 7.4) at 120 sec interval (32-180 min after dissolution).



**Figure S11.** Upper diagram: UV-Vis spectra of Na[RuCl<sub>4</sub>(dmso)(L2)] (**2**) in buffered water (pH = 7.4) at 60 sec interval (0-26 min after dissolution). Lower diagram: UV-Vis spectra of **2** in buffered water (pH = 7.4) at 120 sec interval (28-180 min after dissolution).



**Figure S12.** Solvolysis of Na[RuCl<sub>4</sub>(dmso)<sub>2</sub>] in dmso; t = 0, 1, 3, 6, 24 h. Conc. = 2 x 10<sup>-4</sup> M.



**Figure S13.** Solvolysis of Na[*trans*-RuCl<sub>4</sub>(dmso)(L1)] (1) in dmso; t = 0, 1, 3, 6, 24 h. Conc. = 2 x  $10^{-4}$  M. At t = 0,  $\lambda_{max}$  = 410 nm.



**Figure S14.** Cyclic voltammograms of a 5 x  $10^{-4}$  M solutions of **2** dissolved in water. Supporting electrolyte: KNO<sub>3</sub> 0.1 M. Scan rate: 25 mV/s. First scan (black line), second scan (gray line).



Figure S15. Dose-response plot obtained after a 48 h incubation of HT1080 and HF cells with the indicated concentrations of 1 and 2. The effect of *cis*-platin on both cell lines is reported for comparison.

#### References.

S1 Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Montgomery, J. A., Jr.; Vreven, T.; Kudin, K. N.; Burant, J. C.; Millam, J. M.; Iyengar, S. S.; Tomasi, J.; Barone, V.; Mennucci, B.; Cossi, M.; Scalmani, G.; Rega, N.; Petersson, G. A.; Nakatsuji, H.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima,

T.; Honda, Y.; Kitao, O.; Nakai, H.; Klene, M.; Li, X.; Knox, J. E.; Hratchian, H. P.; Cross, J. B.;
Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.;
Pomelli, C.; Ochterski, J. W.; Ayala, P. Y.; Morokuma, K.; Voth, G. A.; Salvador, P.; Dannenberg,
J. J.; Zakrzewski, V. G.; Dapprich, S.; Daniels, A. D.; Strain, M. C.; Farkas, O.; Malick, D. K.;
Rabuck, A. D.; Raghavachari, K.; Foresman, J. B.; Ortiz, J. V.; Cui, Q.; Baboul, A. G.; Clifford, S.;
Cioslowski, J.; Stefanov, B. B.; Liu, G.; Liashenko, A.; Piskorz, P.; Komaromi, I.; Martin, R. L.;
Fox, D. J.; Keith, T.; Al-Laham, M. A.; Peng, C. Y.; Nanayakkara, A.; Challacombe, M.; Gill, P.
M. W.; Johnson, B.; Chen, W.; Wong, M. W.; Gonzalez, C.; Pople, J. A., Gaussian 03, revision
C.02; Gaussian, Inc.: Wallingford CT, 2004.

S2 E. Alessio, G. Balducci, M. Calligaris, G. Costa, W. M. Attia and G. Mestroni, *Inorg. Chem.*, 1991, **30**, 609.

S3 H. E. Gottlieb, V. Kotlyar and A. Nudelman, J. Org. Chem., 1997, 62, 7512.