

Supporting Online Information for

Preparation and characterization of μ -nitrido diiron phthalocyanines with electronwithdrawing substituents: application for catalytic aromatic oxidation.

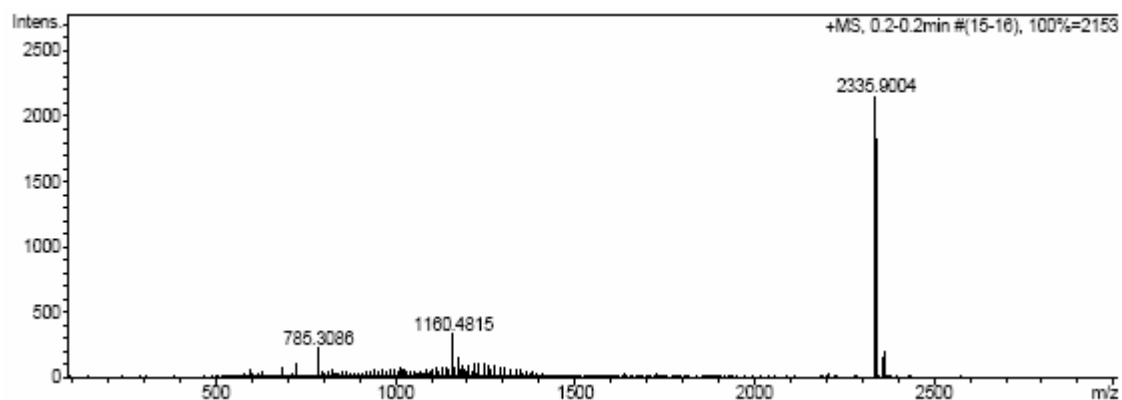
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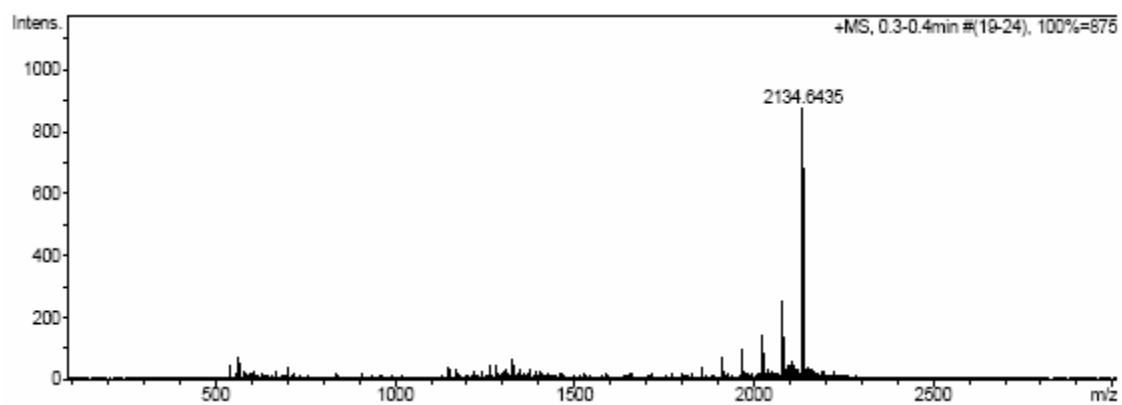
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Supplementary Figures S1 to S6

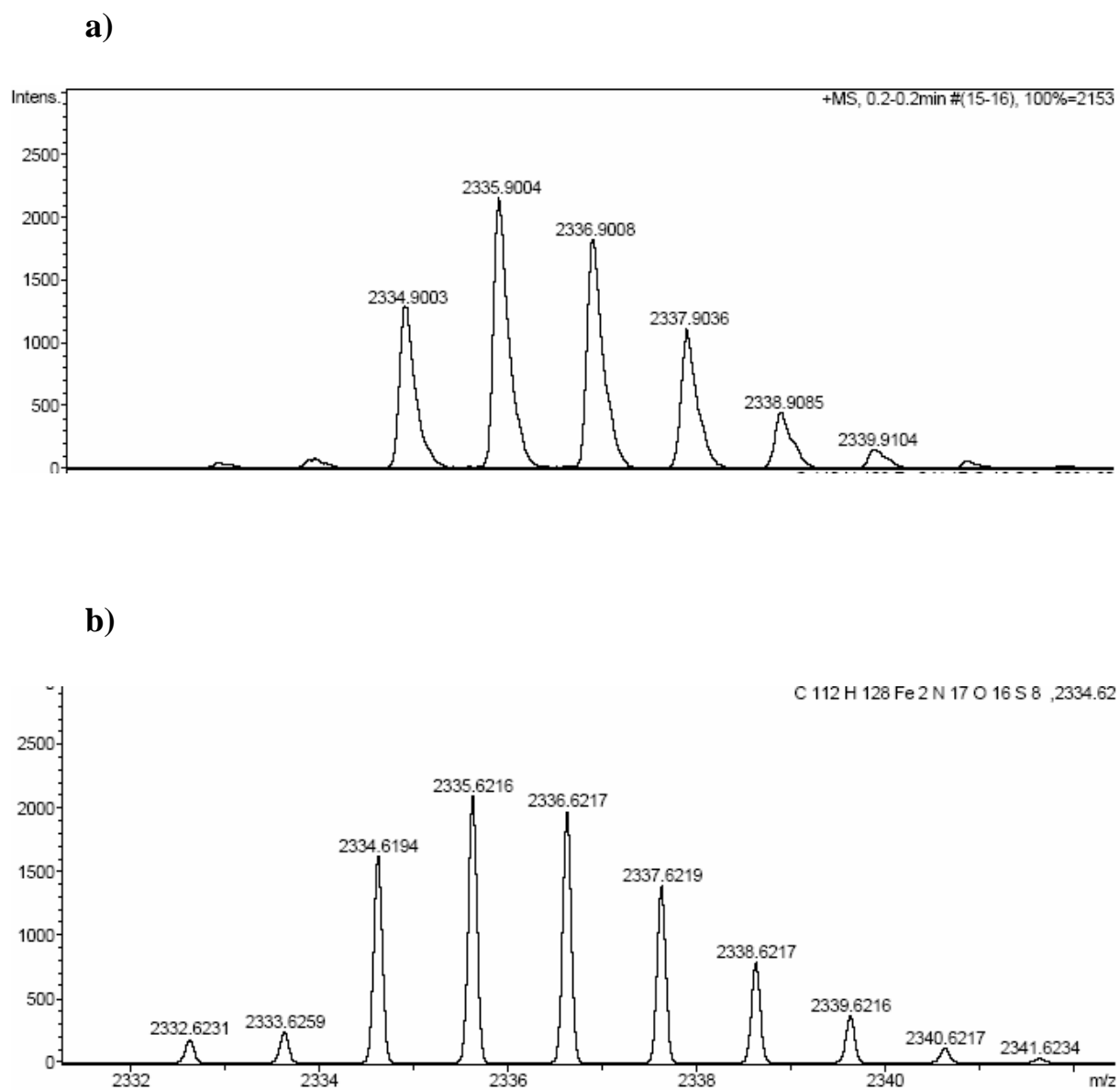
a)



b)

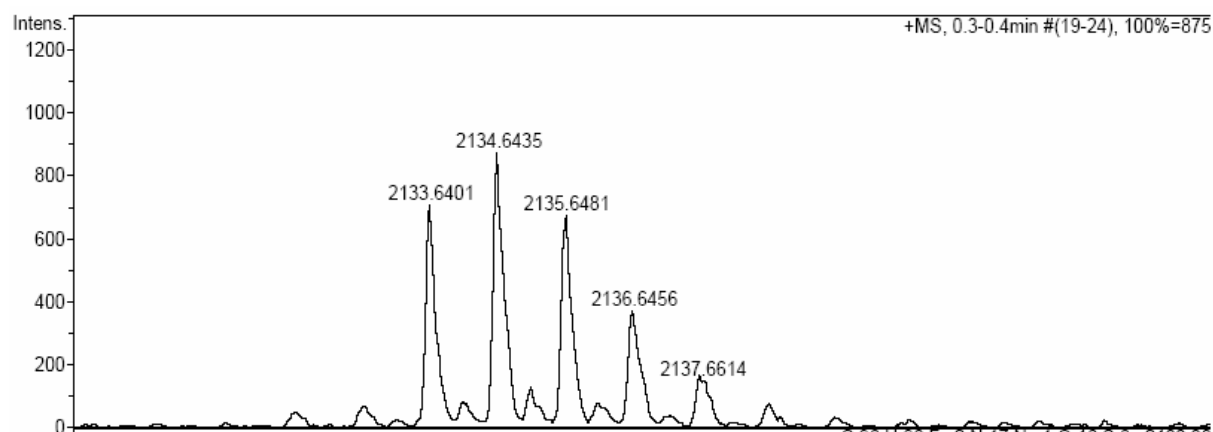


Supplementary Figure 1. ESI-MS spectra of **3a** (a) and **3b** (b).

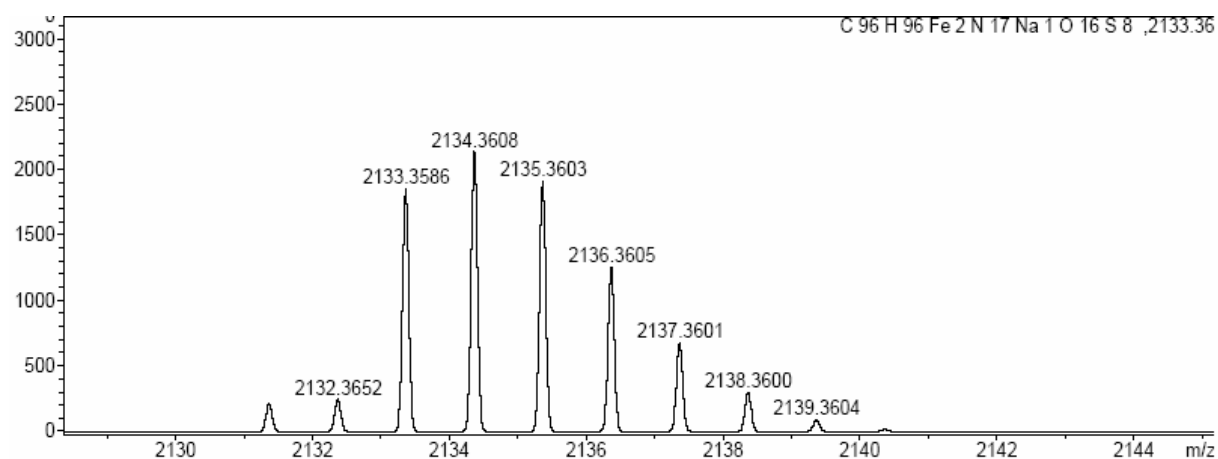


Supplementary Figure 2. Molecular peak cluster of **3a** in ESI-MS spectrum (a) and simulated molecular peak cluster for **3a**, C₁₁₂H₁₂₈N₁₇O₁₆S₈Fe₂ (b).

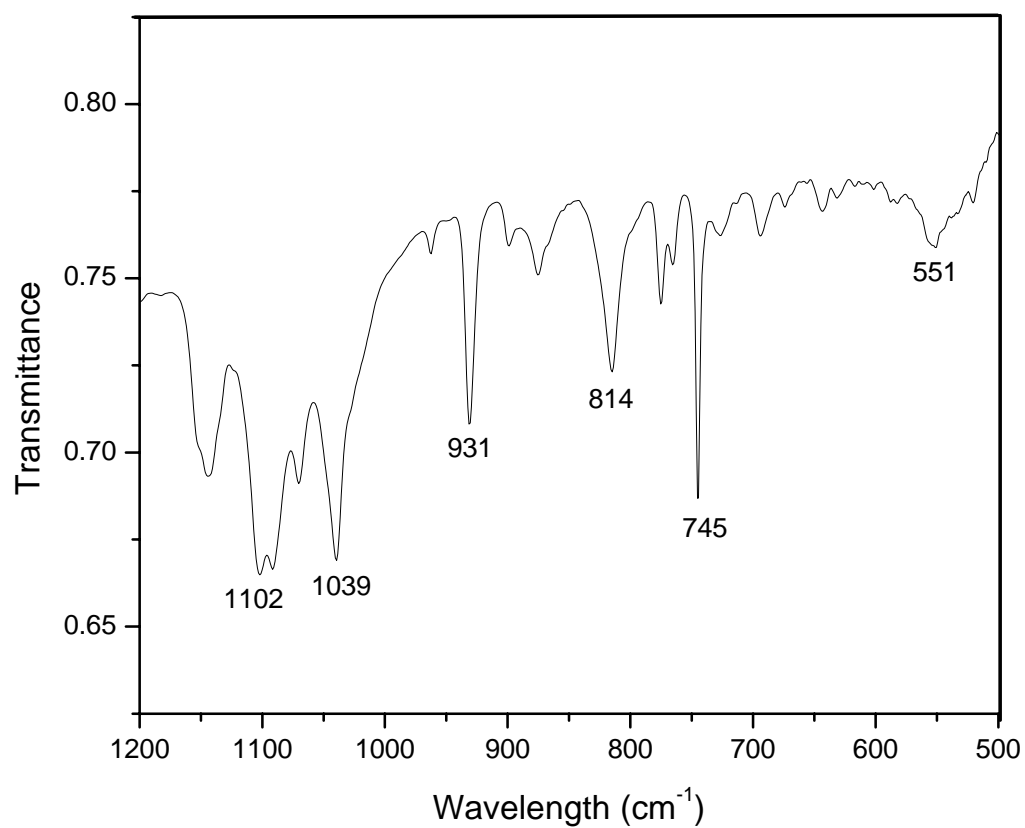
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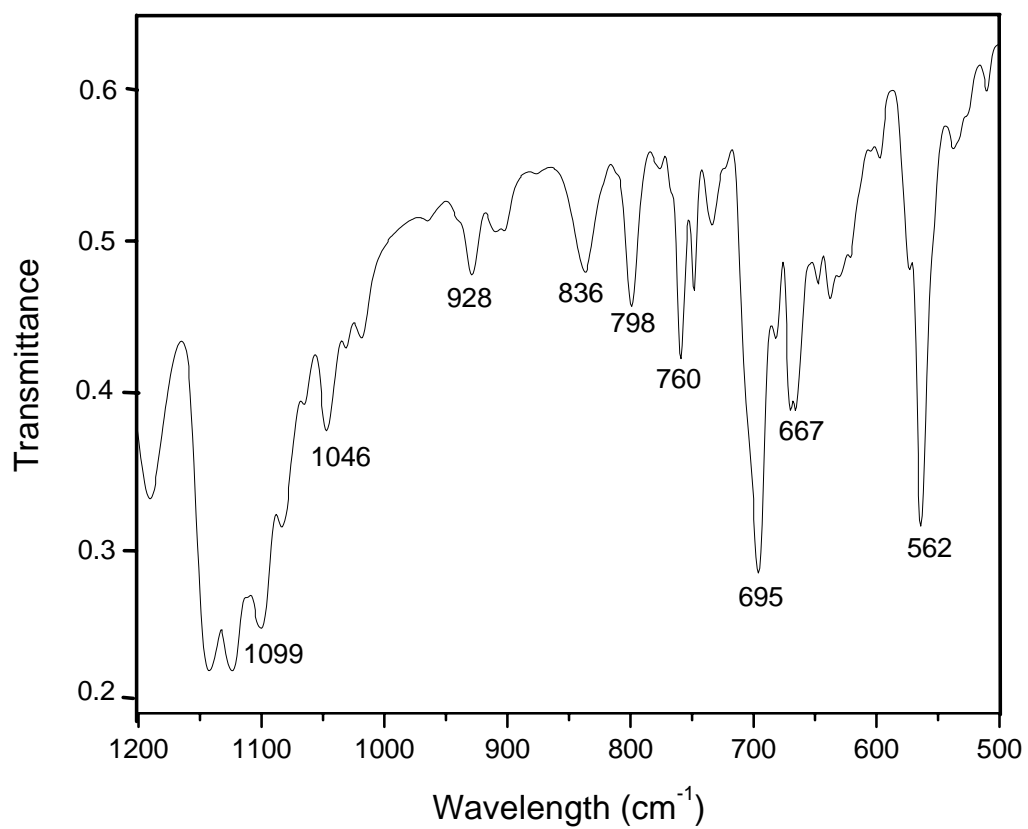
b)



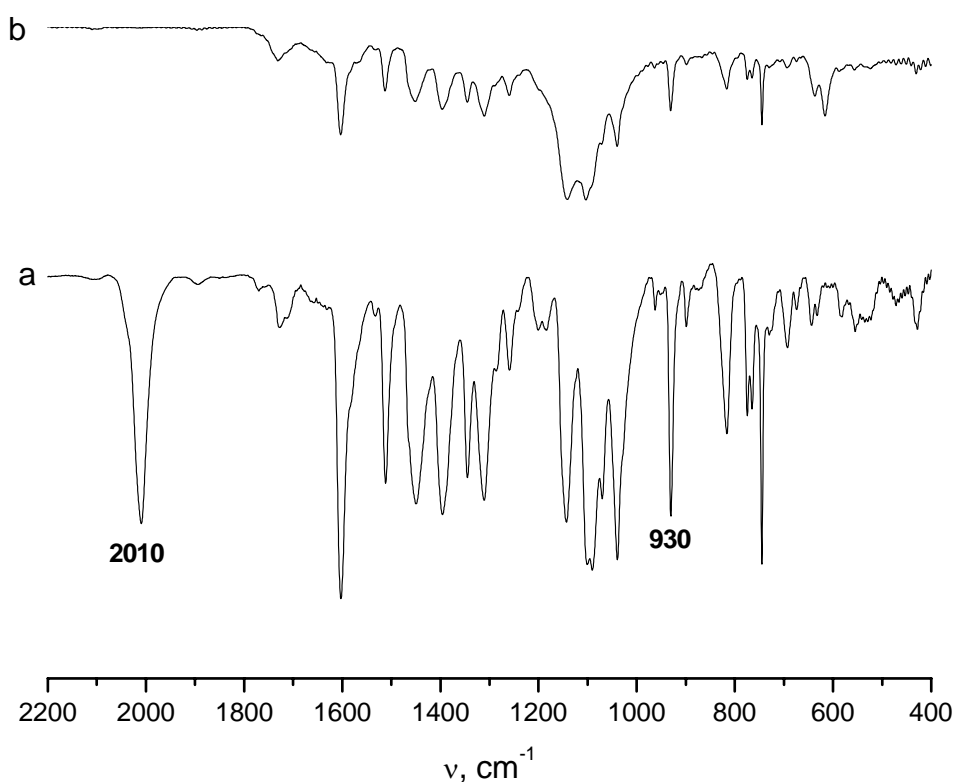
Supplementary Figure 3. Molecular peak cluster of $(\mathbf{3b} + \text{Na})^+$ in ESI-MS spectrum (a) and simulated molecular peak cluster for $(\mathbf{3b} + \text{Na})^+$, C₉₆H₉₆N₁₇O₁₆S₈Fe₂Na (b).



Supplementary Figure 4. IR of 3a.



Supplementary Figure 5. IR of 3b.



Supplementary Figure 6. IR spectrum of **3a** before (a) and after reduction with $\text{Na}_2\text{S}_2\text{O}_4$ (b). Procedure for the reduction of **3a**: 20 mL of 10^{-3} M **3a** solution in CH_2Cl_2 were stirred with 100 mL of deoxygenated 0.02 M $\text{Na}_2\text{S}_2\text{O}_4$ solution for 5 min. Organic layer was separated and washed with water. The solvent was evaporated under reduced pressure. The product was dried.

