

Fig. S1 View of a centrosymmetric Fe₂Cr₂ unit of the crystal structure of **2** showing the second coordination sphere of the iron(II) complex. Colouring of the different atoms: Fe = gold; Cr = magenta; C = brown; H = pink; N = blue; O = red. Thin lines refer to H bonds.

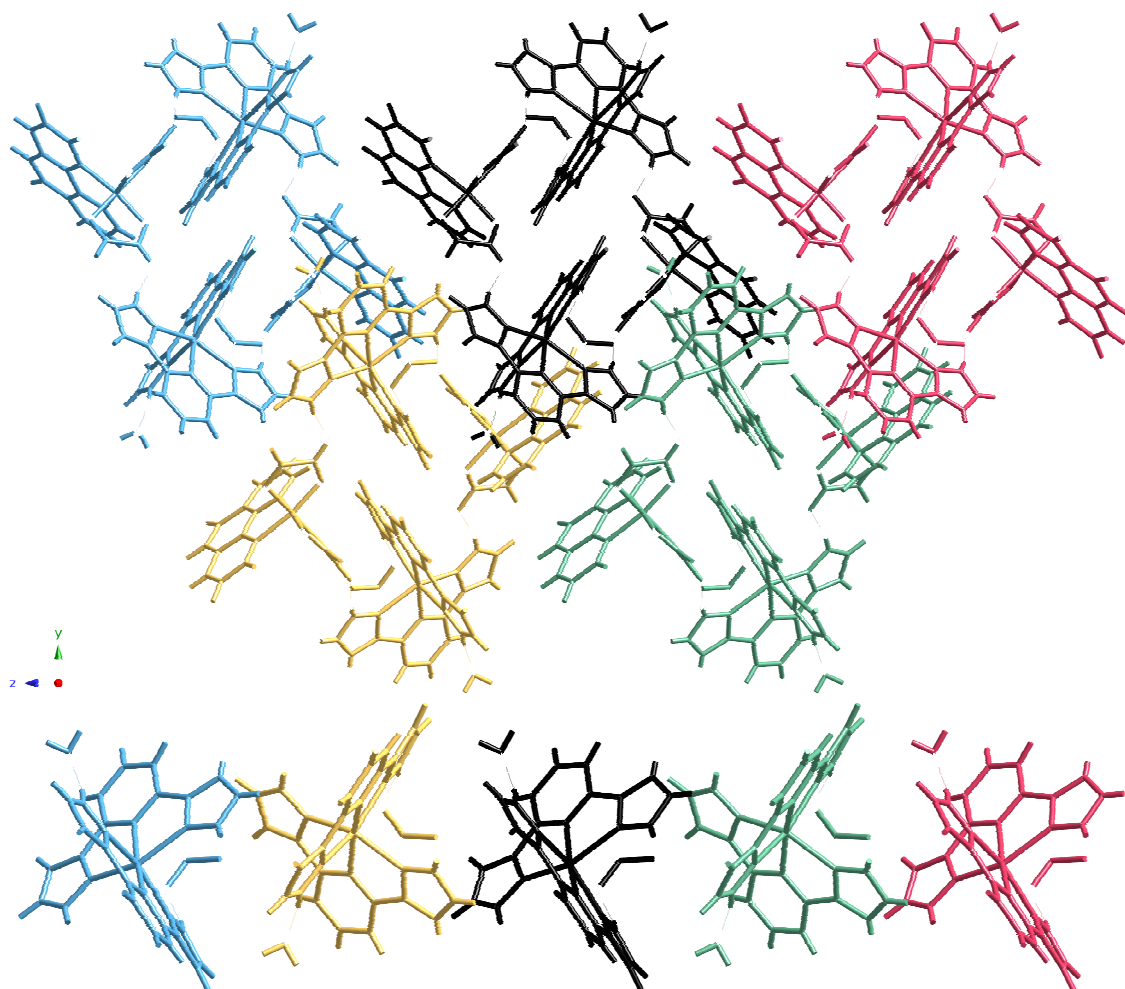


Fig. S2 Projection of the packing of cyclic Fe₂Cr₂ tetramers of **2** onto the *bc* plane. Perchlorate anions are omitted for clarity. A detailed view of the zigzag chains of [Fe(bpp)₂]²⁺ cations is also shown.

Table S1 Intermolecular hydrogen bonds in the crystal structure of **1**.

| D-H | $d_{D-H}/\text{\AA}$ | $d_{H\dots A}/\text{\AA}$ | $d_{D\dots A}/\text{\AA}$ | α° | A |
|-----------|----------------------|---------------------------|---------------------------|----------------|---------------------------------|
| O100-H100 | 0.820 | 1.968 | 2.780 | 170.66 | O7 [$x, -y + 5/2, z + 1/2$] |
| N3-H3 | 0.860 | 1.870 | 2.721 | 170.08 | O8 [$x + 1, y, z + 1$] |
| N7-H7A | 0.860 | 2.041 | 2.824 | 150.92 | O6 [$-x + 2, -y + 2, -z + 2$] |
| N7-H7A | 0.860 | 2.295 | 2.906 | 128.15 | O5 [$-x + 2, -y + 2, -z + 2$] |
| N8-H8A | 0.860 | 1.929 | 2.783 | 171.82 | O2W |
| N12-H121 | 0.860 | 1.936 | 2.778 | 165.69 | O4W [$x, -y + 5/2, z + 1/2$] |
| O4W-H4W2 | 0.861 | | | | |

Table S2 Intermolecular hydrogen bonds in the crystal structure of **2**.

| D-H | $d_{D-H}/\text{\AA}$ | $d_{H\dots A}/\text{\AA}$ | $d_{D\dots A}/\text{\AA}$ | α° | A |
|-----------|----------------------|---------------------------|---------------------------|----------------|--------------------------------------|
| N1-H1N | 0.860 | 1.887 | 2.745 | 175.94 | O3W [$-x + 1, -y + 1, -z + 1$] |
| N5-H5N | 0.860 | 1.906 | 2.726 | 158.87 | O100 [$-x + 1, y - 1/2, -z + 3/2$] |
| N6-H6N | 0.860 | 2.018 | 2.794 | 149.72 | O4 [$-x + 1, y - 1/2, -z + 3/2$] |
| N6-H6N | 0.860 | 2.302 | 2.931 | 130.08 | O3 [$-x + 1, y - 1/2, -z + 3/2$] |
| N10-H10N | 0.860 | 1.905 | 2.727 | 159.46 | O7 [$x + 1, -y + 1/2, z + 1/2$] |
| O100-H100 | 0.820 | 2.043 | 2.844 | 165.30 | O11 [$-x + 1, y + 1/2, -z + 3/2$] |
| O1W-H1W1 | 0.971 | 1.927 | 2.886 | 168.46 | O1 [$x, y, z - 1$] |
| O1W-H1W2 | 0.848 | 2.259 | 2.830 | 124.82 | O3 [$-x + 1, -y + 1, -z + 1$] |
| O2W-H2W1 | 0.656 | 2.345 | 2.998 | 174.42 | O12 [$x, -y + 1/2, z - 1/2$] |
| O2W-H2W2 | 0.830 | 1.984 | 2.804 | 169.26 | O1W |
| O3W-H3W1 | 0.856 | 2.188 | 2.633 | 112.19 | O4W [$-x + 1, y + 1/2, -z + 1/2$] |
| O3W-H3W2 | 0.860 | 2.043 | 2.833 | 152.16 | O2W |

Table S3 Enthalpy and entropy changes associated to thermal processes of **1–2**.

| Compound | Curve ^a | T (K) ^b | $ \Delta H $ (KJ/mol) | $ \Delta S $ (J/mol·K) |
|----------|--------------------|----------------------|-----------------------|------------------------|
| 1 | 1 | 364 | 211.7 | 575.9 |
| | | 316 | 32.4 | 102.7 |
| 2 | 1 | 326 | 89.8 | 274.3 |
| | | 335 | 87.3 | 255.2 |
| | | 356 | 20.6 | 44.0 |
| | | 309 | 87.8 | 284.0 |